Spin gap in malachite Cu₂(OH)₂CO₃ and its evolution under pressure

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We report on the microscopic magnetic modeling of the spin- $\frac{1}{2}$ copper mineral malachite at ambient and elevated pressures. Despite the layered crystal structure of this mineral, the ambient-pressure susceptibility and magnetization data can be well described by an unfrustrated quasi-one-dimensional magnetic model. Weakly interacting antiferromagnetic alternating spin chains are responsible for a large spin gap of 120 K. Although the intradimer Cu–O–Cu bridging angles are considerably smaller than the interdimer angles, density functional theory (DFT) calculations revealed that the largest exchange coupling of 190 K operates within the structural dimers. The lack of the inversion symmetry in the exchange pathways gives rise to sizable Dzyaloshinskii-Moriya interactions which were estimated by full-relativistic DFT + U calculations. Based on available high-pressure crystal structures, we investigate the exchange couplings under pressure and make predictions for the evolution of the spin gap. The calculations evidence that intradimer couplings are strongly pressure dependent and their evolution underlies the decrease of the spin gap under pressure. Finally, we assess the accuracy of hydrogen positions determined by structural relaxation within DFT and put forward this computational method as a viable alternative to elaborate experiments.

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I. INTRODUCTION

Cu-based minerals enjoy close attention of researchers working in the fields of geology and solid-state physics alike. Intricate crystal structures underlie complex arrangements of the spin- $\frac{1}{2}$ Cu²⁺ ions that, in turn, trigger interesting low-temperature quantum effects and exotic ground states.¹ For example, herbertsmithite Cu₃Zn(OH)₆Cl₂, the best available spin- $\frac{1}{2}$ kagomé system, shows a putative spin-liquid ground state. 1-5 Dioptase Cu₆Si₆O₁₈·6H₂O demonstrates unusually strong quantum fluctuations on a nonfrustrated threedimensional spin lattice.⁶ Azurite Cu₃(CO₃)₂(OH)₂ reveals a $\frac{1}{3}$ plateau in the magnetization and presumably hosts a rare magnetic topology of the diamond spin chain.^{7–10} Linarite PbCuSO₄(OH)₂ is an excellent material prototype of the strongly frustrated spin chain. 11,12 Finally, volborthite Cu₃V₂O₇(OH)₂·2H₂O that was originally considered as a kagomé material^{13–15} reveals a more complex and still enigmatic frustrated spin lattice. 16,17

Malachite is arguably the best known Cu secondary mineral typically formed in the oxidation zone of Cu deposits as a weathering product of Cu sulfides. Since antiquity, malachite was the typical source of copper (minerals quarried together with malachite were a convenient flux that facilitated the smelting)¹⁸ and also extensively used as ornamental stone and as a green pigment.¹⁹ The related famous blue Cu-carbonate azurite transforms to malachite by absorption of water and loss of CO₂. This transformation known as "greening" is responsible for greenish instead of blue skies on some historical frescos.²⁰ More recently, malachite and its Zn-substituted versions were recognized as a convenient precursor of mixed CuO–ZnO catalysts.²¹

Regarding this long and prominent history of malachite, surprisingly little is known about its magnetism. Janod *et al.*²² reported the sizable spin gap of about 130 K and proposed a one-dimensional model of bond-alternating spin chains. This model emerges naturally from the crystal structure of

malachite, ²³ where CuO₄ plaquettes form Cu₂O₆ dimers by edge sharing and further link into chains along the [201] direction by corner sharing. According to Janod *et al.*, ²² the stronger coupling should run between the structural dimers because of the larger Cu–O–Cu angle that promotes antiferromagnetic (AFM) superexchange. This suggestion, inferred from the well-known Goodenough-Kanamori-Anderson (GKA) rules, ^{24–26} is in line with many recent studies of Cu²⁺-based compounds, where structural Cu₂O₆ dimers do not match the spin dimers and show weak magnetic couplings only. ^{27–31}

Here we present a detailed density functional theory (DFT)-based microscopic study of malachite and support the ensuing model by magnetization measurements combined with quantum Monte Carlo (QMC) simulations of thermodynamic properties. Contrary to Janod *et al.*,²² we argue that malachite is a rare Cu²⁺ system where structural dimers match the spin dimers. We discuss the origin of this effect, and provide a comprehensive picture of magnetic exchange parameters, including both isotropic and anisotropic exchange couplings. Eventually, we take advantage of available high-pressure crystal structures of malachite³² and investigate the pressure dependence of the exchange couplings and the ensuing spin gap.

The paper is organized as follows. In Sec. II the applied experimental and theoretical methods are presented. The crystal structure of malachite is described in Sec. III. The experimental and computational results for the ambient-pressure malachite are provided in Sec. IV. Our predictions for the evolution of the spin gap under pressure are presented in Sec. V. Finally, discussion and summary are given in Secs. VI and VII, respectively.

II. METHODS

For our experimental studies we used a natural sample of needle-shaped malachite from Tsumeb, Namibia. The sample was investigated by laboratory powder x-ray diffraction (XRD) (Huber G670 Guinier camera, $CuK_{\alpha 1}$ radiation, ImagePlate detector, $2\theta=3^{\circ}-100^{\circ}$ angle range). High-resolution low-temperature XRD data were collected at the ID31 beamline of the European Synchrotron Radiation Facility (ESRF, Grenoble) at a wavelength of about 0.35 Å. The chemical composition was determined by the ICP-OES method.³³ The magnetization was measured with a Quantum Design MPMS SQUID magnetometer in a temperature range of 2–400 K in fields up to 5 T. For measurements up to 14 T, a vibrating sample magnetometer setup of a Quantum Design PPMS was used.

Electronic and magnetic structure calculations were performed within DFT by using the full-potential local-orbital code FPLO9.07-41.³⁴ Local density (LDA)³⁵ and generalized gradient approximations (GGA)³⁶ were used for the exchange-correlation potential together with a well converged k mesh of $5 \times 5 \times 5$ points for the crystallographic unit cell and about 100 points for supercells. For the optimization of hydrogen positions, in addition to FPLO, the Vienna *ab initio* simulation package (VASP5.2)³⁷ was used in combination with LDA, GGA, revPBE,³⁸ DFT-D,³⁹ and HSE06⁴⁰ exchange-correlation functionals. For a full relaxation of all atomic positions in the high-pressure structures, we employed the GGA + U method implemented in VASP.

Strong electronic correlations were included in two different ways: First, by mapping the LDA bands onto an effective one-orbital tight-binding (TB) model. Thereby, the transfer integrals t_{ij} of the TB model are evaluated as nondiagonal matrix elements between Wannier functions (WFs). These transfer integrals t_{ij} are further introduced into the half-filled single-band Hubbard model $\hat{H} = \hat{H}_{\text{TB}} + U_{\text{eff}} \sum_i \hat{n}_{i\uparrow} \hat{n}_{i\downarrow}$, with U_{eff} being the effective on-site Coulomb repulsion. In case of half filling and for the strongly correlated limit $t_{ij} \ll U_{\text{eff}}$, as realized in malachite (see Table II), the Hubbard model can be reduced to the Heisenberg model for the low-energy excitations,

$$\hat{H} = \sum_{\langle ij \rangle} J_{ij} \hat{S}_i \cdot \hat{S}_j, \tag{1}$$

where $\langle ij \rangle$ denotes the summation over all bonds of the spin lattice. Accordingly, the antiferromagnetic (AFM) contributions to the exchange coupling constants can be evaluated in second-order perturbation theory as $J_{ij}^{\text{AFM}} = 4t_{ij}^2/U_{\text{eff}}$. Alternatively, the full exchange couplings J_{ij} , comprising

Alternatively, the full exchange couplings J_{ij} , comprising ferromagnetic (FM) and AFM contributions, can be derived from total energy differences of various collinear spin arrangements evaluated in spin-polarized supercell calculations within the mean-field DFT + U formalism. For the double counting correction, a fully localized limit approximation was used and the on-site Coulomb repulsion and on-site Hund's exchange of the Cu(3d) orbitals are chosen as $U_d = 8.0 \pm 1.0$ eV and $J_d = 1.0$ eV, respectively, similar to parameter sets we have used previously for other cuprates.^{6,10}

The anisotropic exchange was calculated with the full relativistic version of GGA + U provided by VASP with $U_d = 9.5 \,\text{eV}$, $J_d = 1.0 \,\text{eV}$ and the default projector-augmented wave ("PAW-PBE") pseudopotentials⁴¹ on a 4 × 4 × 4 k mesh. For each exchange (J and J') 36 magnetic configurations (four

configurations for each matrix element of the exchange matrix) were calculated. The U_d parameter of GGA + U was chosen so that the isotropic exchanges J_{ij} obtained from VASP agree with those from the FPLO calculations. The 1.5 eV offset in the U_d values arises from the different exchange-correlation functionals and different basis sets used by the two codes.

Quantum Monte Carlo (QMC) simulations were performed using the codes LOOP⁴³ and DIRLOOP_SSE⁴⁴ from the software package ALPS-1.3.⁴⁵ Magnetic susceptibility and magnetization of the two-dimensional model were simulated on finite lattices comprising up to N=1024 spins, using periodic boundary conditions. For simulations in zero field, we used 200 000 sweeps for thermalization and 2 000 000 sweeps after thermalization. For finite-field simulations, 40 000 and 400 000 sweeps were used, respectively.

III. CRYSTAL STRUCTURE

Malachite crystallizes in the monoclinic space group $P2_1/a$ with the lattice constants a=9.5020 Å, b=11.9740 Å, c=3.240 Å, and the monoclinic angle $\beta=98.75^{\circ}.^{23}$ Nearly planar CuO_4 plaquettes form doubly bridged Cu_2O_6 dimers by edge sharing (Fig. 1). The dimers themselves share common corners and form slightly twisted chains running along the [201] direction. The Cu-O-Cu bridging angles within the

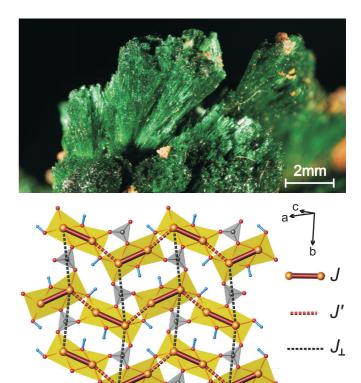


FIG. 1. (Color online) Natural malachite specimen from Tsumeb, Namibia, with aggregates of needle-shaped crystals (top). Bottom: The ambient pressure crystal structure and the microscopic magnetic model of malachite. The structural sketch shows Cu_2O_6 dimers (yellow), CO_3 triangles (gray), and OH bonds (blue). Leading magnetic exchange couplings are shown as thick cylinders (legend in the right panel). Alternating spin chains are formed by J and J'.

TABLE I. Distances (Å) and angles (deg) of hydrogen obtained by structural optimization using the FPLO and VASP codes in combination with various exchange-correlation functionals. "dev" is the averaged deviation (in %) between theoretical and experimental²³ data for distances and angles, respectively.

	VASP					FP		
	PBE	PBEsol	revPBE	DFT-D	HSE06	LDA	PBE	Experiment
H1-O4	0.995	1.002	0.991	0.996	0.976	1.006	0.995	0.975(2)
H2-O5	0.991	0.998	0.987	0.991	0.972	1.004	0.992	0.969(2)
H2-H2	1.916	1.932	1.912	1.918	1.899	1.943	1.918	1.892(2)
dev	1.9	2.6	1.5	1.9	0.3	3.2	1.9	
H1-O4-Cu1	115.1	115.8	114.9	115.2	115.7	115.90	114.92	115.21(13)
H1-O4-O2	157.9	158.5	157.8	158.3	158.7	158.45	157.62	158.17(11)
H2-O5-Cu2	104.1	103.5	104.2	104.4	104.7	103.27	104.00	104.99(14)
H2-O5-O2	140.9	140.5	140.6	141.7	140.3	140.75	140.68	140.64(16)
dev	0.3	0.6	0.3	0.4	0.3	0.6	0.4	

dimers are rather different 94.7° and 106.4° , respectively, resulting in an average bridging angle of 100.5° . The angle between the dimers amounts to 122.1° . Carbonate groups link the chains to planes parallel ($\overline{201}$) which are separated from each other by about 2.37 Å⁴⁶ and are responsible for the perfect cleavage of malachite. Owing to the short interplane distance, however, the Mohs hardness of this mineral, 3.5–4, is significantly higher than for other layer-structured Cu minerals as, e.g., clinoclase (2.5–3) or posnjakite (2–3).⁴⁷ There is another fair cleavage on (010), i.e., orthogonal to planes and parallel to chains, that breaks the bonds between dimers and the CO_3 groups.

Malachite represents one of the rare cases of Cu minerals for which accurate hydrogen positions are available from the experiments. The positions of hydrogen atoms are essential for deriving reliable microscopic magnetic models. Though H is not directly involved in the magnetic superexchange process, in particular its bonding angle has a crucial effect on the type and strength of the exchange couplings. 31,48 The standard experimental technique for determining H positions is neutron scattering which is, however, difficult, expensive, and requires big single crystals (>1 mm³), preferably of deuterium-enriched samples. For these reasons, H positions remain undetermined for most minerals. Alternatively, the hydrogen positions could be obtained by structural optimization within DFT. However, to establish such a procedure, first a careful analysis and comparison with experimental data are mandatory. Malachite thus provides an excellent opportunity to test several different DFT functionals and compare the results with the neutron data (Table I).

Standard LDA and GGA functionals should provide a reasonable description of the covalent O–H bonds. However, they may have deficiencies in describing weak dispersion effects,³⁹ in particular longer O···H hydrogen bonds that determine the direction of the covalent O–H bond. Since there is no functional particularly designed for an accurate description of hydrogen bonds, we test different functionals that either incorporate empirical corrections for van der Waals interactions (DFT-D) or contain different adjustments providing an improved description of lattice parameters, cohesive energies, and different bond lengths (PBEsol, revPBE, HSE06).

Although footing on very different levels of sophistication, all tested functionals supply excellent results, in particular, for the relevant bond angles (Table I). The bond lengths are generally overestimated, with the typical deviation of 1%–2%, the largest deviation of 3% for LDA, and the smallest deviation of less than 0.5% for HSE06. It thus seems that HSE06 provides the best results. However, the structural data for malachite are obtained at room temperature, whereas DFT results pertain to the crystal structure at 0 K. The experimental bond lengths are thus longer because of thermal expansion. Additionally, room temperature data are affected by the libration, a strong rocking vibration of the O-H bond that shortens the apparent O-H distance. This effect is well seen in the oblate thermal ellipsoids of hydrogen atoms that are stretched in the directions transverse to the O–H bonds.²³ These competing but unquantified effects prevent a clear distinction of a certain functional. However, the deviations between theory and experiment shown in Table I are quite small and demonstrate that any functional can be used for optimizing hydrogen positions without significant loss of accuracy. These results show the capability of DFT to provide accurate hydrogen positions in Cu²⁺ minerals and, potentially, in other hydrogen-containing compounds. DFT, therefore, represents a valuable alternative to elaborate experiments. The reliable determination of hydrogen positions further enables us to calculate the missing H position in the high-pressure structures of malachite³² and to assess the magnetic behavior of malachite under pressure.

IV. MALACHITE AT AMBIENT PRESSURE

A. Sample characterization

Powder XRD measurements confirmed the purity of our malachite sample. The high sample quality was additionally supported by chemical analysis yielding 56.8(1)% Cu, 0.1% Pb, 0.1% Ca and <0.1% S. All other detectable elements, including transition metals, are below 0.03%. The lead impurity most likely stems from trace amounts of cerussite (PbCO₃), whereas sulfur may be present as elementary sulfur. Both PbCO₃ and S are found in the specimen matrix. Calcium may be attributed to calcite (CaCO₃). The low content of these

formally nonmagnetic phases should have no disturbing effect on the magnetization measurements.

The slight deficiency of Cu with respect to the expected value of 57.5% may reflect trace amounts of amorphous impurities, which are not seen in XRD but affect the overall composition measured by chemical analysis. Our high-resolution XRD data and the ensuing structure refinement ⁴⁶ rule out any Cu deficiency in the crystalline phase of malachite. Additionally, we performed a structure refinement at 80 K and confirmed that the structure does not change upon cooling. Our data perfectly match earlier room-temperature neutron data. ²³

B. Thermodynamical measurements

The temperature dependence of the magnetic susceptibility $\chi(T)$ shows a paramagnetic behavior at high temperatures and no sign of long-range magnetic ordering down to 2 K (Fig. 2). The Curie-Weiss fit $\chi(T)=\chi_0+C/(T+\theta)$ to the high-temperature part $(T>250~{\rm K})$ yields $\chi_0=5\times10^{-6}$ emu (mol Cu) $^{-1}$, C=0.4802 emu K (mol Cu) $^{-1}$, and $\theta=120.5~{\rm K}$. The resulting effective magnetic moment $\mu_{\rm eff}=1.960~\mu_{\rm B}$ exceeds the spin-only value of $S=\frac{1}{2}~(\mu_{\rm eff}\simeq1.73~\mu_{\rm B})$, yielding the g factor of 2.26, which is within the typical range for Cu $^{2+}$ compounds. 6,49 The positive value of the Weiss temperature θ and the broad maximum in the magnetic susceptibility around 111 K evidence that leading exchange couplings and spin correlations in malachite are AFM.

Following Ref. 22, we fit the experimental curve with the parametrized solution for an alternating Heisenberg chain⁵⁰ (Fig. 2) and obtain leading exchange couplings of 189 and 89 K, and the g factor of 2.18, in agreement with Ref. 22. The spin gap can be estimated as⁵¹

$$\Delta \approx J(1-\alpha)^{\frac{3}{4}}(1+\alpha)^{\frac{1}{4}},\tag{2}$$

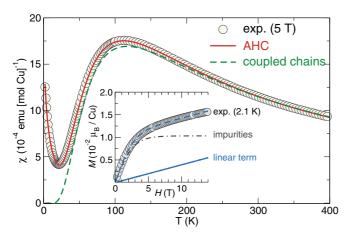


FIG. 2. (Color online) Magnetic susceptibility of malachite at ambient pressure (circles). The solid line shows the parametrized solution of the alternating chain model (AHC), adopted from Ref. 50, supplemented with a Curie impurity term. The dashed line is the intrinsic contribution of the magnetic planes, as yielded by QMC simulations for the full microscopic model. Inset: Experimental magnetization isotherm (circles) and its fit (dashed line) using two terms: paramagnetic impurity contribution (dash-dotted line), described by the Brillouin function, and a linear term (solid line).

where $\alpha = J'/J$. In this way we obtain a large spin gap of $\Delta = 129$ K. The low-temperature upturn can be reproduced by a paramagnetic impurity with $C_{\rm imp} = 0.00808$ emu K (mol Cu)⁻¹ which corresponds to a fraction of about 2% spin- $\frac{1}{2}$ impurities. The temperature-independent contribution is 3.20×10^{-6} emu (mol Cu)⁻¹.

Despite the excellent agreement of the alternating chain model with the experimental $\chi(T)$, small interchain couplings are inevitably present in malachite. Although these couplings do not suffice to stabilize a long-range order (LRO), they can substantially affect the value of the spin gap. To understand the nature of interchain couplings, we evaluate a DFT-based microscopic magnetic model.

C. Microscopic magnetic model

As a first step we performed LDA calculations that provide valuable information about the crucial exchange pathways, though yielding a wrong metallic ground state. The width of the LDA valence band block of about 9.5 eV is typical for cuprates (see Fig. 3). The blocks between -2 and -1 eV and between -0.8 and -0.5 eV, with a sizable $\mathrm{Cu}(3d)$ character, belong to antibonding $pd\pi^*$ and $pd\sigma^*$ orbitals, respectively. The $pd\sigma^*$ block can be separated into bands with dominating $\mathrm{Cu}(3d_{z^2-r^2})$ character, between -0.8 and -0.4 eV, and partially filled bands, crossing the Fermi level, with $\mathrm{Cu}(3d_{x^2-y^2})$ character. The orbitals are defined with respect to a local coordinate

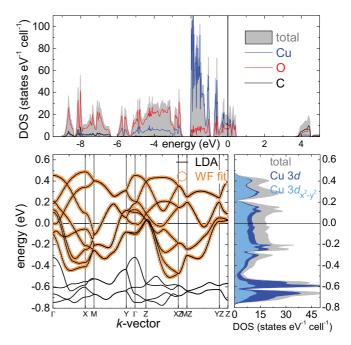


FIG. 3. (Color online) Density of states (DOS) and the LDA valence bands of malachite. The top panel shows the contributions from the $\operatorname{Cu}(3d)$, $\operatorname{O}(2p)$, and C valence states to the total DOS. The Fermi level is at zero energy. In the bottom left panel, the LDA bands around the Fermi level are displayed and compared with bands derived from a fit using an effective one-band tight-binding model based on Cu-centered Wannier functions (WFs) projected on local $\operatorname{Cu}(3d_{x^2-y^2})$ orbitals. The bottom right panel shows that the partial $\operatorname{Cu}(3d)$ -DOS at the Fermi level is basically of $\operatorname{Cu}(3d_{x^2-y^2})$ character, justifying our construction of the WFs.

TABLE II. Transfer integrals t_{ij} (in meV) and the AFM exchange contributions $J_{ij}^{\rm AFM}=4t_{ij}^2/U_{\rm eff}$ (in K) for $U_{\rm eff}=4.5$ eV. The total exchange couplings J_{ij} are calculated with the LSDA + U method and the parameters $U_d=8.0\pm1.0$ eV, $J_d=1.0$ eV.

	Cu-Cu distance (Å)	t_{ij}	$J_{ij}^{ m AFM}$	J_{ij}
J	3.06	-143	211	193 ± 40
J'	3.34	-99	101	109 ± 20
J_{\perp}	6.03	-97	96	50 ± 10

system with the x axis parallel to one of the Cu–O bonds and the z axis orthogonal to the CuO₄ plaquette plane. The small separation between the two types of $pd\sigma^*$ bands arises from the relatively short distance of about 2.37 Å between Cu2 and the apical oxygen, which lifts the energy of the Cu($3d_{z^2-r^2}$) orbital.

The partially filled bands are sufficient to describe the low-lying magnetic excitations and exchange couplings. Their projection onto a TB and subsequently onto a Hubbard model yields the transfer integrals t_{ij} and the corresponding AFM contributions to J_{ij} as given in Table II. Based on the GKA rules, basically describing the dependency of the t_{ij} on the bridging angle, Janod et al.²² expected the J coupling to be smaller than J' according to the smaller bridging angles of 94.7° and 106.4° for J versus 122.1° for J'. Our results, however, show exactly the opposite, with J^{AFM} being twice as large as J'^{AFM} . There are two reasons for this behavior. First, the sizable intradimer transfer t is indeed not unusual and can also be found in several other Cu compounds featuring doubly bridged Cu₂O₆ dimers. For example, in clinoclase the respective transfer amounts to 191 meV (101.9°),³¹ in $Cu_2As_2O_7$ $t = 170 \text{ meV } (101.7^\circ)^{52}$ and in $Li_3CuB_3O_7$ t is 235 meV (100.4°).⁵³ The values in parentheses are the average intradimer bridging angles, which are very similar to that in malachite (100.5°).

The bridging angles alone do not explain why $J^{\prime {\rm AFM}}$ is so much smaller than J^{AFM} while featuring a much larger Cu-O-Cu bridging angle. Thus, second, the twisting of the chains has to be considered. The planes of the neighboring dimers are rotated against each other by about 20°. This reduces the overlap of the WFs and, thus, the transfer integral t'. A similar effect of out-of-plane angles was discussed for clinoclase.³¹ The fact that the long-range interchain transfer t_{\perp} is of similar strength as t' can, on one hand, be attributed to this reduction of t' and, on the other hand, to the polarization of the WFs by the carbonate group (see Fig. 4) that might significantly increase the overlap of the WFs despite the long Cu-Cu distance (1.8 times longer than that for t'). The short interlayer distance in malachite is responsible for a number of further sizable transfer integrals between the planes, 46 which are, however, strongly reduced by FM contributions. All interlayer couplings are below 10 K and, thus, play a minor role for the microscopic magnetic model.

The leading total exchange couplings J_{ij} , including AFM as well as FM contributions, are calculated with the LSDA + U method yielding the correct insulating ground state of malachite. According to the small bridging angles, one could expect a sizable reduction of J due to FM contributions. However, $J^{\rm FM}$ estimated as $J^{\rm FM} = J - J^{\rm AFM}$, does not exceed 60 K

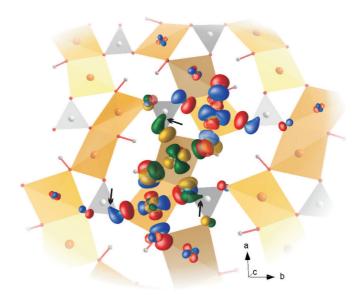


FIG. 4. (Color online) Wannier functions on the Cu1 (yellow-green) and Cu2 sites (red-blue). The polarization (marked by arrows) by the CO₃ groups is responsible for the large t_{\perp} transfer integral.

so that J is still 1.5 times larger than J' and, thus, remains the strongest coupling in this compound. J_{\perp} running via a carbonate group, exhibits sizable FM contributions which is very unusual since the FM contributions to long-range couplings are in general small. However, similar effects have also been observed for the Cu-carbonate azurite 10 and CuSe_2O_5 , 54 the latter featuring pyramidal SeO_3 groups. By contrast, for exchange couplings mediated by tetrahedral PO_4 and AsO_4 ligands FM contributions play a minor role. 31,55 This interesting effect deserves a closer look, which is, however, beyond the scope of the present work but will be pursued in future studies.

Now we are in a position to simulate the temperature dependence of the magnetic susceptibility for the full model of J-J' alternating Heisenberg chains coupled by the interchain exchange J_{\perp} . QMC yields the temperature dependence of a reduced magnetic susceptibility χ^* , 50 which is scaled to fit the experimental data by adjusting the exchange coupling J and the g factor. The second and third terms in Eq. (3) account for spin- $\frac{1}{2}$ impurities and the temperature-independent contributions, respectively.

$$\chi(T) = \frac{N_A g^2 \mu_B^2}{k_B J} \times \chi^* \left(\frac{T}{k_B J}\right) + \frac{C_{\text{imp}}}{T + \theta_{\text{imp}}} + \chi_0.$$
 (3)

J' and J_{\perp} enter χ^* as ratios J'/J and J_{\perp}/J , respectively. Our attempts to get a good fit by taking the ratios provided by LSDA + U, J'/J = 0.56 and $J_{\perp}/J = 0.26$, were not successful. To improve the agreement with the experimental $\chi(T)$, we, thus, varied the ratios in a reasonable range. The best fit can be obtained by taking J'/J = 0.45, $J_{\perp}/J = 0.15$, which yields J = 191 K for the intradimer coupling and the g factor of 2.21 [the intrinsic contribution to $\chi(T)$ is depicted as a dashed line in Fig. 2].

The absolute value of J perfectly agrees with the LSDA + U estimate and also nearly coincides with the estimate from the alternating chain fit. The second largest

exchange J' is somewhat smaller than the value supplied by LSDA + U, while the interchain coupling J_{\perp} is substantially smaller, with the difference exceeding typical error bars. We argue that it could represent a general tendency of DFT to overestimate the superexchange via CO₃ groups. For instance, in the closely related compound, the natural mineral azurite Cu₃(CO₃)₂(OH)₂, DFT also largely overestimates the interchain coupling running through a CO₃ group. 10,56 Further studies should shed light on this issue as we have already stressed before.

The low-temperature part of the intrinsic magnetic susceptibility conforms to an activated behavior:

$$\chi \propto \exp\left(-\frac{\Delta}{T}\right).$$
 (4)

A fit of the intrinsic $\chi(T)$ up to 23 K yields the spin gap $\Delta=119$ K. Compared with the one-dimensional alternating chain model (Sec. IV B), yielding $\Delta=129$ K, the interchain couplings reduce the spin gap in malachite by about 8%.

D. Dzyaloshinskii-Moriya couplings

Since malachite features a large spin gap of $\Delta=119$ K, its low-field uniform magnetization is expected to be zero up to the critical field $H_c = (g \cdot \mu_B)^{-1} \Delta \simeq 80$ T, where the spin gap is closed. In contrast, the experimental magnetization isotherm reveals that the magnetization is nonzero and grows at least up to 14 T (Fig. 2, inset). We also attempted a pulsed-field measurement in higher fields up to 60 T, but no visible signal could be detected. This confirms that H_c lies above 60 T, whereas the signal below H_c remains quite weak (<3.5 × 10⁻² μ_B /Cu) and stays below the sensitivity limit of our pulsed-field experiment. However, in a static-field experiment this weak nonzero signal can be detected.

The low-field range is typically affected by paramagnetic impurities, whose behavior is described by the Brillouin function. We can reproduce the experimental data by the sum of the Brillouin function (the impurity contribution is 1.8%, its g factor is 2.29, and the measurement temperature is T=2.1 K) and a linear $M=\gamma H$ term (Fig. 2, inset). Although the temperature-independent contribution χ_0 would also lead to a linear increase in M, its magnitude is way too small to explain our data: Compare $\gamma=3.9\times10^{-4}~\mu_B/T$ to the slope related to χ_0 , which is on the order of $10^{-5}~\mu_B/T$. Therefore, the nearly linear growth of magnetization above 5 T is likely of intrinsic origin.

This experimental behavior can indicate the presence of anisotropic Dzyaloshinskii-Moriya couplings. ^{57,58} Such couplings break the SU(2) invariance of the Heisenberg Hamiltonian [Eq. (1)] and in the simplest case of an isolated dimer mix the singlet state $\frac{1}{\sqrt{2}}(|\uparrow\downarrow\rangle-|\downarrow\uparrow\rangle)$ with the zero-momentum triplet component $\frac{1}{\sqrt{2}}(|\uparrow\downarrow\rangle+|\downarrow\uparrow\rangle)$. In magnetic field this mixing gives rise to a finite magnetization and yields a linearly increasing M(H). ⁵⁹

To estimate the Dzyaloshinskii-Moriya couplings in malachite, we perform full-relativistic GGA + U calculations using VASP, and map the resulting total energies onto a generic

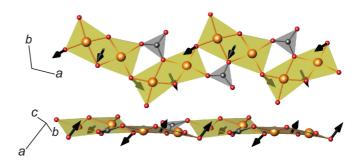


FIG. 5. (Color online) Dzyaloshinskii-Moriya vectors within \mathbf{D} and between the dimers \mathbf{D}' of the alternating spin chains of malachite. Note that the neighboring vectors of the same type (\mathbf{D} or \mathbf{D}') are not identical: The signs of the x and z components alternate along the chain, while the y component retains its sign.

bilinear exchange model:

$$\hat{H} = \sum_{i>j} \sum_{\alpha,\beta} \mathbf{M}_{\alpha,\beta} \hat{S}_i^{\alpha} \hat{S}_j^{\beta}, \quad \alpha,\beta = x, y, z,$$
 (5)

where M is a 3×3 matrix. Three independent components of its antisymmetric part define the respective Dzyaloshinskii-Moriya vector \mathbf{D}_{ij} :

$$\hat{H}_{\rm DM} = \sum_{i>j} \mathbf{D}_{ij} \cdot (\mathbf{S}_i \times \mathbf{S}_j). \tag{6}$$

For the two dominant couplings, we find $\mathbf{D} = [-11.7, -11.3, 10.0]$ K and $\mathbf{D}' = [-16.3, -7.5, 11.9]$ K, with the Cu–Cu bond vectors [-0.266, 0.105, -0.505] and [0.234, -0.181, -0.505], respectively. The components are given in the crystallographic coordinate system with the axes a, b, and c. As can be seen in Fig. 5, the orientation of \mathbf{D} on the neighboring dimers is different: only the y component is preserved, while the x and z components of \mathbf{D} change sign. As a result, the \mathbf{D} vectors on the neighboring dimers are at a skew angle to each other. The same trend holds for the \mathbf{D}' vectors (Fig. 5) and emerges naturally from the Moriya rules.⁵⁸

A standard estimate for the magnitude of Dzyaloshinskii-Moriya coupling is the $|\mathbf{D}_{ij}|/J_{ij}$ ratio. In malachite, these ratios amount to 0.11 and 0.26 for J and J', respectively. Since magnetic dimers are formed on the J bonds, the low-energy behavior is governed by $|\mathbf{D}|/J = 0.11$. Qualitatively, any nonzero D/J conforms to the experimental M(H) behavior. Quantitative estimates can be done by simulating the full microscopic model with Heisenberg and Dzyaloshinskii-Moriya couplings, and a subsequent averaging in order to reproduce the behavior of a powder sample. However, such an analysis is beyond the scope of the present paper.

V. MALACHITE AT HIGH PRESSURES

According to our calculations, malachite is one of the rare cases where structural and magnetic dimers coincide. Other examples are Cu₂As₂O₇,⁵² SrCu₂(BO₃)₂,^{60,61} and TlCuCl₃.^{62,63} While the first compound exhibits magnetic LRO below 10 K, the latter two remain magnetically disordered and feature small spin gaps of 34 and 7.7 K, respectively, that can considerably be affected by pressure. Experimentally, pressure leads to the closing of the spin gap and LRO. On the model side, this effect

TABLE III. Pressure dependence of the magnetic properties and relevant structural parameters in malachite. $\operatorname{ang}(J)$ gives the evolution of the nearest-neighbor bridging angles (deg) under pressure (GPa). The subscripts < and > denote the smaller and larger bridging angles involved in the intradimer coupling J. Transfer integrals t_{ij} (in meV) the total exchange couplings J_{ij} (in K), calculated with LSDA + U and $U_d = 8.0$ eV, and the QMC-simulated spin gap Δ (in K), are given for different pressures. The t_{ij} (opt) are obtained with relaxed atomic positions. For an extended list, including error bars for the bridging angles, see the Supplemental Material.

Pressure	0.0	1.02	2.03	3.10	5.17
$ang_{<}(J)$	94.75	95.53	93.23	90.16	91.23
$\operatorname{ang}_{>}(J)$	106.41	105.56	105.47	107.00	105.67
ang(J')	122.13	123.19	120.95	123.33	121.62
t	-143	-135	-127	-114	-111
t'	-99	-91	-83	-83	-94
t_{\perp}	-97	-88	-91	-86	-68
J	193	164	159	136	114
J'	109	110	105	97	129
J_{\perp}	50	40	25	40	11
Δ	77	51	58	27	17
t(opt)	_	-137	-135	-125	-101
t'(opt)	_	-93	-96	-88	-75
$t_{\perp}(\text{opt})$	_	-82	-71	-59	-37

implies the increase in the \bar{J}'/J ratio, where \bar{J}' is a sum of all interdimer couplings which, in case of malachite, basically include J', J_{\perp} , and the interlayer couplings.

Different microscopic mechanisms affecting the \bar{J}'/J ratio have been discussed on an empirical level (see Sec. VI). However, as long as the reliable high-pressure structural data are missing, no analysis of the pressure effects on the microscopic magnetic model of a dimer compound could be performed. Results of a recent DFT study on the pressure dependence of exchange couplings of CuO^{64} cannot be applied because of the different crystal structure of CuO.

Therefore, we take advantage of existing high-pressure XRD data for malachite up to 5.17 GPa³² to investigate effects of pressure on the individual exchange couplings. Missing hydrogen positions were obtained by structural optimization within GGA. The transfer integrals t_{ij} and exchange couplings J_{ij} for different pressures together with the evolution of the bridging angles are condensed in Table III. An extended table is provided in the Supplemental Material.⁴⁶

Regarding the crystal structure, the applied pressure has its strongest effect on the interlayer distance, according to the weak bonding between the structural layers in malachite. Thus, the a and c lattice constants are reduced from 9.502 to 9.114 Å and 3.240 to 3.057 Å, respectively, and the enclosed monoclinic angle decreases by about 3.7° when increasing the pressure from 0 to 5.17 GPa. This entails a diminishing Jahn-Teller distortion on the Cu2 site, where the apical Cu–O distance decreases from 2.37 to 2.15 Å and approaches the in-plane distances of 1.9–2.1 Å. This results in a nearly octahedral coordination, which is highly unfavorable for a $3d^9$ ion. At pressures above 6 GPa, the system, therefore, undergoes a phase transition to the rosasite structure. Cu2 is

then again fourfold coordinated, but with a modified CuO_4 plaquette plane lying perpendicular to its orientation in the malachite structure. This transformation involves an abrupt increase in the longest lattice parameter by about $0.6~\text{Å}.^{32}$ The rosasite structure consists of planar chains of edge-sharing CuO_4 plaquettes, running along the c axis and linked by Cu1 monomers. The magnetic behavior of this high-pressure phase definitely deserves a closer examination, which is, however, beyond the scope of this work.

Within the structural layers of malachite, the most prominent effect of pressure is the sizable reduction of the smaller intradimer Cu2–O–Cu1 bridging angle by about 3.5° (see Table III) which is accompanied by a large increase of the Cu2–O distance by about 0.1 Å. The larger intradimer angle as well as the Cu–O–Cu interdimer angle of the J^\prime exchange pathway vary both unsystematically by about 2°, only.

The effects of the applied pressure on the exchange couplings are most pronounced for the intradimer coupling J which is reduced by more than 40%. This can be directly related to the evolution of t which, according to the GKA rules, decreases in terms of its absolute value with the decreasing intradimer bridging angle. J' exhibits an unsystematic variation within ± 20 K that cannot be perfectly related to the evolution of the corresponding bridging angle. It seems that also changes of the twisting angle, the angular dependence of the ferromagnetic contribution, and may be even more subtle effects determine the pressure dependence of J'. The same is true for the interchain coupling J_{\perp} : Though one could argue that an expansion of the exchange pathway by about 0.1 Å is responsible for the decreasing coupling strength, it is most likely also affected by distortions of the carbonate group and slightly enhanced buckling of the structural planes. A more detailed analysis of the pressure effects and a comparison with SrCu₂(BO₃)₂ and TlCuCl₃ will be given in Sec. VI.

The LSDA + U estimates for the leading exchange integrals allow for tracing the evolution of the spin gap under pressure. Using QMC, we simulate the field-dependent magnetization, which is zero for a gapped state and nonzero otherwise. QMC simulations cannot be performed for T=0, hence we compute magnetization isotherms at sufficiently low temperature of 0.01 J (corresponds to 1.9 K). To correct for finite size effects, we evaluate magnetization for different finite lattices of Nspins and extrapolate to the $N \to \infty$ limit. This way we obtain the values listed in Table III. The general decrease of the spin gap can safely be established, yet the absolute values of the spin gap are not perfectly reliable. For instance, the simulated ambient-pressure spin gap is 77 K, which is substantially smaller than the experimental value (119 K). This difference originates from inaccuracies in the values of the leading exchanges: LSDA + U overestimates J_{\perp} and also J'. Thus, LSDA + U values lead to an underestimation of the spin gap at ambient pressure. An additional source of inaccuracies is the restriction to a 2D magnetic model, where a coupling between the magnetic layers is neglected. With increasing pressure these couplings, however, become more effective⁴⁶ and will further reduce the spin gap.

Though the overall trend clearly shows a reduction of the spin gap, there is an unexpected sudden increase at 2.03 GPa (Table III). We had a closer look at bond angles and distances but could not find any obvious reason for this behavior, although in general pressure evolution of the bond lengths and distances is somewhat nonmonotonic. As structure determination under pressure may be less accurate than in ambient conditions, we made an additional test by performing a full relaxation of all atomic positions of the high-pressure crystal structures (only the lattice parameters were fixed to their experimental values). Exact results of such a relaxation will, to some extent, depend on the exchange-correlation functional, U_d parameter, spin arrangement, and other details of the calculation. Here we chose the GGA + U method, as implemented in VASP, as a suitable reference and considered the AFM spin arrangement. The leading transfer integrals obtained from the relaxed structures are displayed in Table III. All t_{ij} (opt) show a smooth pressure dependence without any peculiarities at 2.03 GPa. These results, thus, establish the aforementioned trend for the evolution of the exchange couplings and clearly indicate a substantial decrease of the spin gap under pressure. Such a reduction should be well visible experimentally. A more detailed analysis of the deviations between experimental and relaxed structures as well as a revisiting of the experimental structure of malachite, at least at 2.03 GPa, are left to future studies.

VI. DISCUSSION

In the present work we have discussed a microscopic magnetic model for the famous Cu-mineral malachite. Despite the layered crystal structure, the magnetization data at ambient pressure can be described by weakly coupled AFM alternating chains. Intrachain exchange couplings of $J=191~\rm K$ and $J'=86~\rm K$ open a large spin gap, which is slightly reduced (8%) to 119 K by interchain exchanges conveyed by carbonate groups. The unfrustrated couplings of dimerized spin chains via polyanions typically facilitate LRO as, e.g., in Cu₂P₂O₇²⁷ and Cu₂As₂O₇. In contrast, no signs of the LRO have been observed in malachite at least down to 2 K.

To investigate the role of interchain couplings in malachite, we simulate the magnetization isotherm of the 2D $J-J'-J_{\perp}$ model using QMC (Fig. 6). As expected for a gapped system, magnetization remains zero up to the critical field h_c . The

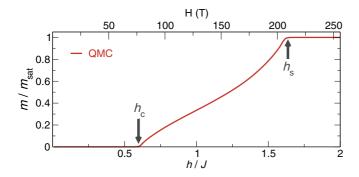


FIG. 6. (Color online) Magnetization isotherm for the $J-J'-J_{\perp}$ model with $J:J':J_{\perp}=1:0.45:0.15$, simulated using quantum Monte Carlo (QMC). The steep increase of m above h_c (closing of the spin gap) and below the saturation field (h_s) are characteristic features of quasi-one-dimensional systems.

m(h) behavior right above h_c is ruled by the dimensionality of the system. Thus, in a 1D model at zero temperature, magnetization behaves as $m \propto \sqrt{h^2 - \Delta^2}$, 65 leading to a steep increase right above h_c (i.e., for $h \simeq \Delta$), which eventually transforms to the linear dependence $m \propto h$ for $h > \Delta$. Two-dimensional models show a remarkably different behavior: the m(h) slope is linear, with logarithmic corrections in the vicinity of h_c . 66 The m(h) dependence in Fig. 6 is closer to the former scenario, indicating a quasi-one-dimensional behavior.

Another characteristic feature is the transition to the fully polarized state (saturation). A common feature of one-dimensional magnets is a steep increase of m right below the saturation field h_s .⁶⁷ Again, malachite shows a clear resemblance to the one-dimensional scenario: A pronounced upward bending below h_s is clearly visible (Fig. 6). Thus we can conclude that the interchain couplings in malachite play a minor role and this mineral is a quasi-one-dimensional magnet.

A possible explanation for the quasi-1D character of malachite can be found by comparing the spin lattices of different spin-dimer compounds. In malachite, there are two couplings J_{\perp} at each Cu2 site and no interchain couplings at the Cu1 site, thus yielding one interchain coupling per Cu site in average. In Cu₂P₂O₇ and Cu₂As₂O₇, each Cu atom interacts with four Cu atoms from the neighboring chains.²⁷ Therefore, the tendency of these compounds to the LRO is much stronger compared to malachite.

The DFT results further allow locating the J and J'exchanges in the crystal structure. In a former experimental study²² the strongest coupling J has empirically been located between the structural dimers where, according to the GKA rules, the large bridging angle of 122.1° should lead to a strong coupling. Our results, however, unambiguously reveal that the intradimer coupling is strongest. Our explanation for this is twofold: First, sizable exchange coupling for an averaged intradimer bridging angle of about 100° is indeed not uncommon in Cu compounds, as we have shown in Sec. IV C. Second, we stress that the effect of nonplanar arrangement of neighboring WFs has to be considered in the empirical modeling. The tilting of the interacting CuO₄ plaquettes by about 20° for J' is responsible for the rather small interdimer coupling. Accordingly, malachite is one of the rare cases where magnetic and structural dimers

The coinciding magnetic and structural dimers have been previously observed in TlCuCl₃ and $SrCu_2(BO_3)_2$, which attracted large attention because of their small spin gaps of 7.7^{68} and $34~K^{60}$ and the quantum phase transitions that were observed by applying pressure or magnetic fields. The possibility of directly observing magnetic ordering processes and quantum phase transitions is of enormous importance for understanding collective quantum phenomena. Which mechanism actually closes/reduces the spin gap under pressure was controversially discussed in the literature. Structural dimers are often considered rigid and unaffected by pressure. Therefore, the reduction of the spin gap was ascribed to the increased interdimer couplings. On the other hand, the bridging angles for the intradimer coupling J are close to the range of $95^{\circ}-98^{\circ}$, where a transition from AFM to FM

coupling can be expected. Therefore, the dimers could be very sensitive even to small structural changes. The latter scenario is supported by a work of Johannsen *et al.* Who analyzed susceptibility and magnetostriction data for TlCuCl₃. However, these authors set the pressure dependence of the interdimer couplings to zero and used a very simplified magnetic model as a basis for their analysis of pressure effects. Though the spin gap of malachite is much larger than in TlCuCl₃ and SrCu₂(BO₃)₂, the mechanisms and effects on the exchange couplings induced by pressure should be similar. We thus used existing high-pressure structural data of malachite for the first DFT-based microscopic analysis of pressure effects on exchange couplings in a dimer compound and the typical pressure induced decrease of the spin gap.

According to our DFT results, external pressure reduces the intradimer coupling J by about 40%, which can be attributed to the bridging angles decreasing from 94.7° to 91.2° and from 106.4° to 105.7°, respectively. The interdimer coupling J'varies unsystematically within ± 20 K which is driven by small changes in the bridging angle (\approx 2°) and the bonding distance $(\approx 0.03 \text{ Å})$. The long-range interchain coupling J_{\perp} is reduced from about 50 to 12 K. However, as our QMC simulations for the ambient pressure data have suggested, LSDA + Uoverestimates J_{\perp} so that this coupling is in fact even smaller and thus of minor importance, in particular at high pressures. Therefore, the evolution of the spin gap is basically driven by the intradimer coupling, which confirms the results of Ref. 71. These authors have found dJ/dp = 22 K/GPa and $d\Delta/dp =$ 14 K/GPa for the pressure dependence of the intradimer coupling and the spin gap, respectively. The latter value stems from experimental data, while the former one is derived from a simple magnetic model. In the case of SrCu₂(BO₃)₂, different methods provided different estimates for $d\Delta/dp$ in the range 6–11 K/GPa.^{61,72} For malachite, DFT calculations in combination with QMC simulations supplied the average ratios dJ/dp = 12 K/GPa and $d\Delta/dp = 11$ K/GPa.⁷³ We thus can conclude that the spin gap in dimer compounds is generally reduced by applying pressure. The changes of Δ and J are thereby similar for the different compounds and, thus, suggest that the same microscopic mechanisms are effective. We have demonstrated that pressure has its main impact on the intradimer coupling, which is crucially responsible for the closing/reduction of the spin gap. However, as our analysis revealed, interdimer couplings in fact cannot a priori be regarded as constant and/or negligible. Thus, in general lots of subtle details have to be taken into account for a quantitative analysis of pressure-induced effects.

Some of the spin-dimer compounds lack inversion symmetry in the middle of the spin dimer, which gives rise to anisotropic Dzyaloshinskii-Moriya interactions. In $SrCu_2(BO_3)_2$, these interactions mixing singlet and triplet states are strong enough to invalidate the description of field-induced transition in terms of the Bose-Einstein condensation of magnons.⁷⁴ The Dzyaloshinskii-Moriya vectors we estimated for malachite, $|\mathbf{D}|/J = 0.11$ and $|\mathbf{D}'|/J' = 0.26$, are much stronger than those in $SrCu_2(BO_3)_2$ ($|\mathbf{D}|/J \simeq 0.05$, $|\mathbf{D}'|/J' < 0.02$).⁷⁵ Moreover, $|\mathbf{D}'|/J' \gg |\mathbf{D}|/J$, which is opposite to the situation in $SrCu_2(BO_3)_2$. Considering the large DM anisotropy in malachite, we suggest that experimental studies of the magnetic excitation spectrum by electron

spin resonance and/or inelastic neutron scattering could be insightful. For example, we envisage a peculiar splitting of the triplet band, similar to the recent observation of two nearly parallel bands in the frustrated-spin-ladder compound BiCu₂PO₆. Experimental studies on the magnetism of malachite under pressure should be interesting as well. Our work provides solid microscopic basis for such studies.

VII. SUMMARY

In summary, we have performed a combined theoretical and experimental study on the famous Cu-mineral malachite at ambient pressure and have made predictions for the evolution of its spin gap for pressures up to 5.17 GPa. For the magnetic modeling of the high-pressure structures, we first determined the hydrogen positions, missing in the presently available XRD data, by structural optimizations within DFT. The reliability of this method has been tested on ambient pressure malachite for which accurate neutron data exist. The results are in excellent agreement with the experimental data, so that we propose the determination of hydrogen positions by DFT as a highly valuable, fast, and cheap alternative to experiments.

The magnetic structure of malachite at ambient pressure is well described by the model of alternating antiferromagnetic chains with the couplings $J=191~\rm K$ and $J'=86~\rm K$. Interchain couplings slightly reduce the resulting spin gap by 8% to 119 K. The evolution of the exchange couplings and the spin gap under pressure has been investigated by DFT calculations and QMC simulations. The results have been compared with the dimer compounds $TlCuCl_3$ and $SrCu_2(BO_3)_2$ for which different mechanisms for the closing of the spin gap have been proposed. In this study we have explicitly demonstrated that the reduced intradimer coupling is the driving force for closing the spin gap under pressure. Furthermore, Dzyaloshinskii-Moriya interactions were estimated and assigned to be responsible for the linear increase in the magnetization at low fields.

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- ¹L. Balents, Nature (London) **464**, 199 (2010).
- ²P. Mendels and F. Bert, J. Phys. Soc. Jpn. **79**, 011001 (2010), and references therein.
- ³J. S. Helton, K. Matan, M. P. Shores, E. A. Nytko, B. M. Bartlett, Y. Qiu, D. G. Nocera, and Y. S. Lee, Phys. Rev. Lett. **104**, 147201 (2010).
- ⁴D. E. Freedman, T. H. Han, A. Prodi, P. Mller, Q.-Z. Huang, Y.-S. Chen, S. M. Webb, Y. S. Lee, T. M. McQueen, and D. G. Nocera, J. Am. Chem. Soc. **132**, 16185 (2010).
- ⁵T. Han, S. Chu, and Y. S. Lee, Phys. Rev. Lett. 108, 157202 (2012).
 ⁶O. Janson, A. A. Tsirlin, M. Schmitt, and H. Rosner, Phys. Rev. B 82, 014424 (2010), and references therein.
- ⁷H. Kikuchi, Y. Fujii, M. Chiba, S. Mitsudo, T. Idehara, T. Tonegawa, K. Okamoto, T. Sakai, T. Kuwai, and H. Ohta, Phys. Rev. Lett. **94**, 227201 (2005).
- ⁸K. C. Rule, A. U. B. Wolter, S. Süllow, D. A. Tennant, A. Brühl, S. Köhler, B. Wolf, M. Lang, and J. Schreuer, Phys. Rev. Lett. **100**, 117202 (2008); M. C. R. Gibson, K. C. Rule, A. U. B. Wolter, J.-U. Hoffmann, O. Prokhnenko, D. A. Tennant, S. Gerischer, M. Kraken, F. J. Litterst, S. Süllow, J. Schreuer, H. Luetkens, A. Brühl, B. Wolf, and M. Lang, Phys. Rev. B **81**, 140406(R) (2010); K. C. Rule, D. A. Tennant, J.-S. Caux, M. C. R. Gibson, M. T. F. Telling, S. Gerischer, S. Süllow, and M. Lang, *ibid.* **84**, 184419 (2011).
- ⁹F. Aimo, S. Krämer, M. Klanjŝek, M. Horvatić, C. Berthier, and H. Kikuchi, Phys. Rev. Lett. **102**, 127205 (2009); F. Aimo, S. Krämer, M. Klanjšek, M. Horvatić, and C. Berthier, Phys. Rev. B **84**, 012401 (2011).
- ¹⁰H. Jeschke, I. Opahle, H. Kandpal, R. Valenti, H. Das, T. Saha-Dasgupta, O. Janson, H. Rosner, A. Bruhl, B. Wolf, M. Lang, J. Richter, S. Hu, X. Wang, R. Peters, T. Pruschke, and A. Honecker, Phys. Rev. Lett. 106, 217201 (2011).
- ¹¹A. U. B. Wolter, F. Lipps, M. Schäpers, S.-L. Drechsler, S. Nishimoto, R. Vogel, V. Kataev, B. Büchner, H. Rosner, M. Schmitt, M. Uhlarz, Y. Skourski, J. Wosnitza, S. Süllow, and K. C. Rule, Phys. Rev. B 85, 014407 (2012); Y. Yasui, M. Sato, and I. Terasaki, J. Phys. Soc. Jpn. 80, 033707 (2011).
- ¹²B. Willenberg, M. Schäpers, K. C. Rule, S. Süllow, M. Reehuis, H. Ryll, B. Klemke, K. Kiefer, W. Schottenhamel, B. Büchner, B. Ouladdiaf, M. Uhlarz, R. Beyer, J. Wosnitza, and A. U. B. Wolter, Phys. Rev. Lett. 108, 117202 (2012).
- ¹³Z. Hiroi, M. Hanawa, N. Kobayashi, M. Nohara, H. Takagi, Y. Kato, and M. Takigawa, J. Phys. Soc. Jpn. **70**, 3377 (2001); A. Fukaya, Y. Fudamoto, I. M. Gat, T. Ito, M. I. Larkin, A. T. Savici, Y. J. Uemura, P. P. Kyriakou, G. M. Luke, M. T. Rovers, K. M. Kojima, A. Keren, M. Hanawa, and Z. Hiroi, Phys. Rev. Lett. **91**, 207603 (2003); S. Yamashita, T. Moriura, Y. Nakazawa, H. Yoshida, Y. Okamoto, and Z. Hiroi, J. Phys. Soc. Jpn. **79**, 083710 (2010).
- ¹⁴F. Bert, D. Bono, P. Mendels, F. Ladieu, F. Duc, J.-C. Trombe, and P. Millet, Phys. Rev. Lett. **95**, 087203 (2005).
- ¹⁵M. Yoshida, M. Takigawa, H. Yoshida, Y. Okamoto, and Z. Hiroi, Phys. Rev. Lett. **103**, 077207 (2009); M. Yoshida, M. Takigawa, S. Krämer, S. Mukhopadhyay, M. Horvatić, C. Berthier, H. Yoshida, Y. Okamoto, and Z. Hiroi, J. Phys. Soc. Jpn. **81**, 024703 (2012).
- ¹⁶O. Janson, J. Richter, P. Sindzingre, and H. Rosner, Phys. Rev. B 82, 104434 (2010).
- ¹⁷G. J. Nilsen, F. C. Coomer, M. A. de Vries, J. R. Stewart, P. P. Deen, A. Harrison, and H. M. Rønnow, Phys. Rev. B 84, 172401 (2011).

- ¹⁸S. J. Fleming and C. P. Swann, Nucl. Instrum. Methods Phys. Res., Sect. B **75**, 440 (1993).
- ¹⁹S. Bruni, F. Cariati, F. Casadio, and L. Toniolo, Spectrochim. Acta A 55, 1371 (1999); K. Eremin, J. Stenger, and M. L. Green, J. Raman Spectrosc. 37, 1119 (2006); L. Burgio, R. J. H. Clark, and R. R. Hark, Proc. Natl. Acad. Sci. USA 107, 5726 (2010).
- ²⁰K. Wehlte, Werkstoffe und Techniken der Malerei (Christophorus, Wiesbaden, Germany, 2001).
- ²¹M. Behrens, F. Girgsdies, A. Trunschke, and R. Schlögl, Eur. J. Inorg. Chem. **2009**, 1347 (2009); M. Behrens and F. Girgsdies, Z. Anorg. Allg. Chem. **636**, 919 (2010).
- ²²E. Janod, L. Leonyuk, and V. Maltsev, Solid State Commun. 116, 513 (2000).
- ²³F. Zigan, W. Joswig, H. U. Schuster, and S. A. Mason, Z. Kristallogr. 145, 412 (1977).
- ²⁴J. B. Goodenough, Phys. Rev. **100**, 564 (1955).
- ²⁵J. Kanamori, J. Phys. Chem. Solids **10**, 87 (1959).
- ²⁶P. W. Anderson, Solid State Phys. **14**, 99 (1963).
- ²⁷A. A. Tsirlin, O. Janson, and H. Rosner, Phys. Rev. B 82, 144416 (2010); O. Janson, A. A. Tsirlin, J. Sichelschmidt, Y. Skourski, F. Weickert, and H. Rosner, *ibid*. 83, 094435 (2011).
- ²⁸O. Mentré, E. Janod, P. Rabu, M. Hennion, F. Leclercq-Hugeux, J. Kang, C. Lee, M.-H. Whangbo, and S. Petit, Phys. Rev. B 80, 180413(R) (2009); A. A. Tsirlin, I. Rousochatzakis, D. Kasinathan, O. Janson, R. Nath, F. Weickert, C. Geibel, A. M. Läuchli, and H. Rosner, *ibid.* 82, 144426 (2010).
- ²⁹H. Ben Yahia, E. Gaudin, J. Darriet, D. Dai, and M.-H. Whangbo, Inorg. Chem. **45**, 5501 (2006).
- ³⁰ J. Deisenhofer, R. M. Eremina, A. Pimenov, T. Gavrilova, H. Berger, M. Johnsson, P. Lemmens, H. A. Krug von Nidda, A. Loidl, K. S. Lee, and M. H. Whangbo, Phys. Rev. B 74, 174421 (2006).
- ³¹S. Lebernegg, A. A. Tsirlin, O. Janson, and H. Rosner, Phys. Rev. B 87, 235117 (2013).
- ³²M. Merlini, N. Perchiazzi, M. Hanfland, and A. Bossak, Acta Crystallogr. Sect. B 68, 266 (2012).
- ³³ICP-OES (inductively coupled plasma optical emission spectrometry) analysis was performed with the Vista instrument from Varian.
- ³⁴K. Koepernik and H. Eschrig, Phys. Rev. B **59**, 1743 (1999).
- ³⁵J. P. Perdew and Y. Wang, Phys. Rev. B **45**, 13244 (1992).
- ³⁶J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996).
- ³⁷G. Kresse and J. Furthmüller, Comput. Mater. Sci. 6, 15 (1996); Phys. Rev. B 54, 11169 (1996).
- ³⁸Y. Zhang and W. Yang, Phys. Rev. Lett. **80**, 890 (1998).
- ³⁹S. Grimme, Wiley Interdisciplinary Rev. Comput. Mol. Sci. 1, 211 (2011).
- ⁴⁰J. Heyd, G. E. Scuseria, and M. Ernzerhof, J. Chem. Phys. **118**, 8207 (2003).
- ⁴¹G. Kresse and D. Joubert, Phys. Rev. B **59**, 1758 (1999).
- ⁴²A. A. Tsirlin, O. Janson, I. Rousochatzakis, and H. Rosner (unpublished).
- ⁴³S. Todo and K. Kato, Phys. Rev. Lett. **87**, 047203 (2001).
- ⁴⁴F. Alet, S. Wessel, and M. Troyer, Phys. Rev. E 71, 036706 (2005), and references therein.
- ⁴⁵A. Albuquerque, F. Alet, P. Corboz, P. Dayal, A. Feiguin, S. Fuchs, L. Gamper, E. Gull, S. Gürtler, A. Honecker, R. Igarashi, M. Körner, A. Kozhevnikov, A. Läuchli, S. R. Manmana, M. Matsumoto, I. P. McCulloch, F. Michel, R. M. Noack, G. Pawlowski, L. Pollet, T. Pruschke, U. Schollwöck, S. Todo, S. Trebst, M. Troyer, P. Werner, and S. Wessel, J. Magn. Magn. Mater. 310, 1187 (2007).

- ⁴⁶See Supplemental Material at http://link.aps.org/supplemental/ 10.1103/PhysRevB.88.224406 for a figure of the crystal structure, an extended table of hopping parameters, and exchange constants, bonding angles, and distances as a function of pressure, interlayer t_i , and J_i as a function of pressure, as well as details of the structure refinement.
- ⁴⁷www.mindat.org (2013).
- ⁴⁸E. Ruiz, P. Alemany, S. Alvarez, and J. Cano, Inorg. Chem. 36, 3683 (1997).
- ⁴⁹A. A. Tsirlin, O. Janson, S. Lebernegg, and H. Rosner, Phys. Rev. B **87**, 064404 (2013).
- ⁵⁰D. C. Johnston, R. K. Kremer, M. Troyer, X. Wang, A. Klümper, S. L. Bud'ko, A. F. Panchula, and P. C. Canfield, Phys. Rev. B 61, 9558 (2000).
- ⁵¹T. Barnes, J. Riera, and D. A. Tennant, Phys. Rev. B **59**, 11384 (1999).
- ⁵²Y. C. Arango, E. Vavilova, M. Abdel-Hafiez, O. Janson, A. A. Tsirlin, H. Rosner, S.-L. Drechsler, M. Weil, G. Nénert, R. Klingeler, O. Volkova, A. Vasiliev, V. Kataev, and B. Büchner, Phys. Rev. B 84, 134430 (2011).
- ⁵³N. V. Kuratieva, M. Bànki, A. A. Tsirlin, J. Eckert, H. Ehrenberg, and D. Mikhailova, Inorg. Chem. (2013), doi:10.1021/ic4015724.
- ⁵⁴O. Janson, W. Schnelle, M. Schmidt, Y. Prots, S.-L. Drechsler, S. K. Filatov, and H. Rosner, New J. Phys. 11, 113034 (2009).
- ⁵⁵M. D. Johannes, J. Richter, S.-L. Drechsler, and H. Rosner, Phys. Rev. B **74**, 174435 (2006).
- ⁵⁶J. Kang, C. Lee, R. K. Kremer, and M.-H. Whangbo, J. Phys.: Condens. Matter 21, 392201 (2009).
- ⁵⁷I. Dzyaloshinsky, J. Phys. Chem. Solids 4, 241 (1958).
- ⁵⁸T. Moriya, Phys. Rev. **120**, 91 (1960).
- ⁵⁹S. Miyahara, J.-B. Fouet, S. R. Manmana, R. M. Noack, H. Mayaffre, I. Sheikin, C. Berthier, and F. Mila, Phys. Rev. B 75, 184402 (2007).
- ⁶⁰H. Kageyama, M. Nishi, N. Aso, K. Onizuka, T. Yosihama, K. Nukui, K. Kodama, K. Kakurai, and Y. Ueda, Phys. Rev. Lett. 84, 5876 (2000).

- ⁶¹T. Sakurai, M. Tomoo, S. Okubo, H. Ohta, K. Kudo, and Y. Koike, J. Phys.: Conf. Ser. **150**, 042171 (2009).
- ⁶²M. Matsumoto, B. Normand, T. M. Rice, and M. Sigrist, Phys. Rev. B **69**, 054423 (2004).
- ⁶³A. Oosawa, M. Fujisawa, T. Osakabe, K. Kakurai, and H. Tanaka, J. Phys. Soc. Jpn. **72**, 1026 (2003).
- ⁶⁴X. Rocquefelte, K. Schwarz, and P. Blaha, Sci. Rep. 2, 759 (2012).
- ⁶⁵G. I. Dzhaparidze and A. A. Nersesyan, JETP Lett. **27**, 334 (1978).
- ⁶⁶S. Sachdev, T. Senthil, and R. Shankar, Phys. Rev. B **50**, 258 (1994).
- ⁶⁷M. Schmidt, C. Gerhardt, K.-H. Mütter, and M. Karbach, J. Phys.: Condens. Matter **8**, 553 (1996).
- ⁶⁸W. Shiramura, K. Takatsu, H. Tanaka, K. Kamishima, M. Takahashi, H. Mitamura, and T. Goto, J. Phys. Soc. Jpn. **66**, 1900 (1997).
- ⁶⁹C. Rüegg, N. Cavadini, A. Furrer, H.-U. Güdel, K. Krämer, H. Mutka, A. Wildes, K. Habicht, and P. Vorderwisch, Nature 423, 62 (2003).
- ⁷⁰M. Braden, G. Wilkendorf, J. Lorenzana, M. Aïn, G. J. McIntyre, M. Behruzi, G. Heger, G. Dhalenne, and A. Revcolevschi, Phys. Rev. B 54, 1105 (1996).
- ⁷¹N. Johannsen, A. Vasiliev, A. Oosawa, H. Tanaka, and T. Lorenz, Phys. Rev. Lett. **95**, 017205 (2005).
- ⁷²H. Kageyama, N. V. Mushnikov, M. Yamada, T. Goto, and Y. Ueda, Physica B **329–333**, 1020 (2003).
- ⁷³The 2.03 GPa data is neglected in this estimate according to the ambiguities in the crystal structure we had found at this pressure.
- ⁷⁴K. Kodama, S. Miyahara, M. Takigawa, M. Horvatic, C. Berthier, F. Mila, H. Kageyama, and Y. Ueda, J. Phys.: Condens. Matter 17, L61 (2005).
- ⁷⁵V. V. Mazurenko, S. L. Skornyakov, V. I. Anisimov, and F. Mila, Phys. Rev. B **78**, 195110 (2008).
- ⁷⁶K. W. Plumb, Z. Yamani, M. Matsuda, G. J. Shu, B. Koteswararao, F. C. Chou, and Y. J. Kim, Phys. Rev. B **88**, 024402 (2013).