Microscopic theory of the insulating electronic ground states of the actinide dioxides AnO₂ (An = U, Np, Pu, Am, and Cm)

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The electronic states of the actinide dioxides AnO₂ (with An = U, Np, Pu, Am, and Cm) are investigated employing first-principles calculations within the framework of the local density approximation +U (LDA + U) approach, implemented in a full-potential linearized augmented plane-wave scheme. A systematic analysis of the An-5 *f* states is performed which provides intuitive connections between the electronic structures and the local crystalline fields of the *f* states in the AnO₂ series. Particularly the mechanisms leading to the experimentally observed insulating ground states are investigated. These are found to be caused by the strong spin-orbit and Coulomb interactions of the 5 *f* orbitals; however, as a result of the different configurations, this mechanism works in distinctly different ways for each of the AnO₂ compounds. In agreement with experimental observations, the nonmagnetic states of plutonium and curium dioxide are computed to be insulating, whereas those of uranium, neptunium, and americium dioxides require additional symmetry breaking to reproduce the insulator ground states, a condition which is met with magnetic phase transitions. We show that the occupancy of the An-*f* orbitals is closely connected to each of the appearing insulating mechanisms. We furthermore investigate the detailed constitution of the noncollinear multipolar moments for transverse 3*q* magnetic ordered states in UO₂ and longitudinal 3*q* high-rank multipolar ordered states in NpO₂ and AmO₂.

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I. INTRODUCTION

Recently, f electron materials containing actinide elements have drawn considerable interest, stimulated by observations of intriguingly ordered ground states forming at low temperatures. A striking example is the formation of the hidden order phase in URu₂Si₂, where the origin of the arising low-temperature electronic order could not unambiguously be disclosed even after a quarter century (see, e.g., Ref. 1). High-rank multipoles have been proposed recently as possible candidates for the order parameter in the hidden order phase.^{2–7} Another example is the ordered ground state of NpO₂, which after many years could experimentally be established to be due to a high-rank multipolar order, in the absence of any dipolar moment formation.^{8,9} The richness of 5f electron physics can be attributed to their multiple degrees of freedom where entangled spin and orbital moments may occur, activated through the strong spin-orbit (SO) interaction in the actinide elements as well as on-site Coulomb interactions. These conditions are in particular met in the actinide dioxides, which have provided a treasure trove of a rich variety of multiorbital physics over many years.9-12

The actinide dioxides crystallize in the cubic fluorite structure, yet, the ground-state properties of the actinide dioxides are intriguingly diverse. The current knowledge of their ordered ground states stipulates that in UO₂ transverse 3q magnetic dipolar and 3q electric quadrupolar order is realized;^{13,14} NpO₂ is characterized by a 3q ordered magnetic multipole of a high rank, in the absence of any dipole moment formation.^{8,9} AmO₂ undergoes a conspicuous phase transition at around 8.5 K.¹⁵ While a peak structure in the magnetic susceptibility was found,¹⁵ neutron-diffraction measurements could not detect antiferromagnetic order in agreement with the

Mössbauer measurement.^{16,17} PuO_2 is a simple nonmagnetic insulator,¹⁸ whereas CmO_2 has an unexpected paramagnetic moment.¹⁹ Surprisingly, each of these actinide dioxides is an insulator, in spite of the very different emerging orders and physical properties. This suggests that in these dioxides different gap-formation mechanisms are in fact operative. The origin of gap formation can only restrictedly be established through experiments. Conversely, first-principles electronic structure calculations are well suited to study the gap formation mechanisms in relation to the unusually ordered states, but only a few such first-principles calculations, accounting for multipolar order states, have been performed.²⁰

A framework of first-principles calculations allows for a description of the bulk properties even if the investigated system contains some localized f electrons in an open shell. However, it is known that the localized character of f electrons is not reproduced with the basic approximations employed in firstprinciples calculations, like the local density approximation (LDA) or generalized gradient approximation (GGA) in which the two-body correlation between electrons in the open shell is not captured sufficiently. It was actually reported that the LDA approach led to states that are inconsistent with the CEF ground states determined experimentally in the AnO₂.^{21,22} Moreover, the calculations predicted metallic ground states,^{21,23} whereas the AnO₂ compounds with light actinides are known to be large gap semiconductors or insulators with energy gaps around 2 eV.^{24–26} Therefore, first-principles calculations taking into account the strong Coulomb interaction have been applied and have provided insight in the detailed magnetic and electronic character of UO_2 ,²⁷⁻³⁴ of NpO₂,^{20,31,34} and of PuO_2 , 30,31,34-41 although the f characters provided in some of these calculations are different from those of the experimental ground states, especially for the ordered states. As taking strong electron correlations into account is necessary to obtain the energy gap in the first-principles calculations, the insulating ground states of AnO₂ are referred to as Mott insulators.^{29,42,43} Thus, to appropriately describe the ground states in the AnO₂ compounds, it is imperative to include the effects of strong Coulomb interaction, strong spin-orbit coupling, and the multiorbital character of the *f* electrons on equal footing. In this paper we perform such first-principles calculations, focusing particularly on the mechanisms of insulating gap formation. We show that the insulating mechanisms differ substantially from one AnO₂ compound to another, depending on the involved *f* states in AnO₂.

The LDA + U method with SO interaction included has been successful in capturing the ground-state properties of *f*-electron compounds.^{20,44–51} In spite of some limitations, this method is especially suited to describe the local character of f electrons on the same footing with the electronic band description. Moreover, there are two specific properties of the LDA + U framework that make it suitable to study complex ordered phases in the AnO₂: the spin-orbital dependence of the local potential, which is essential since appearing order parameters involve multiple spin and orbital degrees of freedom, and the ability to take into account noncollinearity of local order parameters, because the multipole moments on each An site can be expressed through the local potential only.²⁰ Here we have adopted the LDA + U method with spin-orbit interaction, combined with the framework of the full-potential linearized augmented plane wave (FLAPW) band-structure method. Since a large Coulomb U tends to increase the anisotropic character of the f states, the full-potential treatment is an important ingredient, too, to adequately reproduce the f states in the AnO₂.

In the following we provide a complete computational study of the electronic structures of the AnO₂, focusing on the activated spin-orbital degree of freedom in the ground states. First, we give in Sec. II an outline of the known experimental properties of the AnO₂ compounds. In Sec. III we describe the employed computational framework, and in Sec. IV we report the obtained results and provide discussions separately for our calculations of the nonordered states in the AnO₂ and for the ordered states computed for UO₂, NpO₂, and AmO₂.

II. PROPERTIES OF AnO₂ COMPOUNDS

The actinide dioxides crystallize in the cubic fluorite crystal structure. They are characterized by an ionicity with An^{4+} cations and O^{2-} anions, which is strong for the early 5f elements, but is reduced for the late actinide elements.³¹ The dioxides with open 5f shell are further characterized by the strong localized character of the 5f electrons on the An atoms. Note that ThO₂ has no occupied *f* orbitals and is therefore not considered here.

The An-5 *f* states multiplets are classified as a Γ_5 triplet, a Γ_8 quartet, and a Γ_1 singlet in UO₂,^{14,52-54} NpO₂,⁵⁴⁻⁵⁶ and PuO₂,^{54,57,58} respectively, based on the Russell-Saunders LS coupling scheme for their paramagnetic states. In a one electron description, the degenerate 5 *f* orbitals split in $j = \frac{5}{2}$ and $j = \frac{7}{2}$ orbitals under the action of strong SO coupling. The $j = \frac{5}{2}$ orbitals, which have a lower orbital energy than the $j = \frac{7}{2}$ orbitals, further split into a Γ_7 doublet and a Γ_8

Splitting of one electron orbital (schematic)



FIG. 1. (Color online) Schematic description of the actinide one-electron j = 5/2 orbitals in the crystal fields with O_h and D_{3d} point-group symmetry, respectively. The Γ_8 quartet in cubic O_h symmetry splits in two singlets Γ_4 and Γ_5 , degenerate under time-reversal symmetry, and one doublet state $\Gamma_6^{(1)}$, in D_{3d} symmetry.

quartet through the cubic crystalline electric field (CEF), see the schematic description in Fig. 1. Following a simple point charge model, the energy level of the $|j = 5/2, \Gamma_8\rangle$ quartet should be lower than that of the $|j = 5/2, \Gamma_7\rangle$ doublet due to the oxygen anions located in [1 1 1] directions from the actinide cations. If the CEF is large enough to neglect the higher-lying Γ_7 orbitals, the *f* states can be described by states in which two, three, and four electrons occupy the Γ_8 orbitals for U⁴⁺, Np⁴⁺, and Pu⁴⁺ ions, respectively. Although the simple one electron description qualitatively explains the local character of the *f* states on the An single ion in these compounds,^{59,60} it is nonetheless necessary to analyze the electronic structure on the basis of first-principles calculations to reveal the bulk properties that involve the conduction as well as localized electrons, such as, e.g., the insulating behavior of the AnO₂.^{20,30,31}

UO₂ exhibits a clear phase transitions at T = 30.8 K.^{61,62} The low temperature phase of UO₂ is associated with a transverse 3*q* antiferromagnetic structure, ^{13,63,64} which superseded the earlier suggestions of 1*k* (Ref. 65) or 2*k* (Ref. 66) structures, with a dipolar magnetic uranium moment of 1.74 μ_B .^{67,68} The dipolar 3*q* magnetic order in UO₂ is accompanied by a 3*q* ordering of quadrupoles, which corresponds to a distortion of the 5*f* charge density along the direction of dipolar moment.^{13,64} It has been proposed that the superexchange quadrupolar interaction is an important ingredient to stabilize the 3*q* magnetic ground state^{64,69,70} as well as for the magnetization dynamical properties observed for UO₂.^{13,71} Also, recent first-principles studies based on an LDA + *U* method successfully showed the stability of the 3*q* dipolar magnetic ordered state in UO₂.^{29,33}

The phase transition observed^{61,72} at T = 25.4 K in NpO₂ has been an enigmatic issue until not too long ago. At an earlier stage of the research on this issue, the low temperature phase of NpO₂ had been speculated to arise from an antiferromagnetic order due to a similar behavior of the temperature dependent susceptibility to that of UO₂. However, no dipole magnetic moments essential to characterize the phase transition have been observed in neutron scattering^{55,56} and Mössbauer experiments.^{73–75} Intensive experimental and theoretical efforts then were invested to identify the unusual ordered state. Crucial experiments to identify aspects of the ordered state were the muon spin rotation measurement which detected breaking of the time-reversal

symmetry⁷⁶ and resonant x-ray scattering which identified the electric symmetry of the low temperature phase.⁷⁷ Several experiments are consistent with, and have confirmed, the noncollinear 3*q* multipolar order (MPO) with Γ_5 symmetry multipoles.^{55,76–89} Note that the Γ_5 multipoles do not contain a dipolar moment in its tensor elements. In principle, the Γ_5 octupole⁹⁰ moments could be detectable in resonant x-ray measurements,⁹¹ however, direct observation of the octupole moments has not been successful so far. Several recent theories pointed out that a multipolar ordered state with triakontadipole (rank 5) primary order parameter would explain the small weight of the octupole moments.^{8,9,20}

The phase transitions to the 3q MPO states break translation symmetry of the face centered cubic (fcc) crystal structure but preserve a simple cubic (sc) symmetry containing a four-sublattice unit cell.^{77,92,93} A symmetry analysis of the low temperature phases observed in UO_2 and NpO_2 has already been performed.^{81,94,95} The space group $Fm\bar{3}m$ is lowered to the $Pa\bar{3}$ and the $Pn\bar{3}m$ groups in the ordered phases of UO_2 and NpO₂, respectively.^{77,94} Since the magnetic space group of the 3q-MPO states have no nonunitary part, both electric and magnetic multipoles which belong to the same symmetry can spontaneously appear in these ordered states. The four-sublattice unit cell contains one site of equivalent eight oxygen atoms which has one Wycoff parameter in the $Pa\bar{3}$ and two nonequivalent fixed oxygen sites (cubic 2a site and tetragonal 6d site) in the $Pn\bar{3}m$ space group. These facts are clearly reflected in the measured ¹⁷O-NMR spectra.^{84,96} Although the antiferromagnetically ordered state in UO₂ would allow the oxygen atoms to move with the Wycoff parameter, it was reported that the deformation of the oxygen position is considerably small.⁶⁸

In PuO₂, the magnetic susceptibility is independent of temperature up to 1000 K.⁹⁷ The nonmagnetic property is understood from the CEF analysis for the Pu⁴⁺ ion giving that the ground state is the nonmagnetic singlet and the first excited state, which is 123 meV above the ground state, is a triplet.^{57,58,98} Transport measurements, furthermore, showed that PuO₂ is an insulator with a 1.8-eV activation gap,²⁴ while optical spectroscopy gave a 2.8-eV direct gap.²⁶ A recent Pu-NMR study confirmed the nonmagnetic character of the ground state of PuO₂.¹⁸

In AmO₂ a clear antiferromagnetic-looking phase transition is observed at 8.5 K,¹⁵ however, no magnetic moment has been observed in Mössbauer nor in neutron scattering measurements.^{16,17} The fingerprint of this dioxide is similar to the one obtained in earlier studies of NpO2, and hence, an AF-MPO state is expected in AmO₂. The CEF ground states of the Am⁴⁺ ion in AmO₂ have been believed to be a Γ_7 doublet.^{15,99,100} However, the Γ_7 state has no degree of freedom for the higher rank multipoles and therefore seems to contradict the experiments which observe no magnetic dipole moments. A recent CEF analysis based on the j-j coupling method discussed an instability of the Γ_7 ground state and possibility of stabilization of the Γ_8 ground state, which can induce higher order multipoles without inducing a dipole moment.¹⁰¹ Notably, there are many existing experimental challenges to distinguish the essential bulk contribution of AmO2 due to the strong self-radiation damage caused by alpha decay.^{102–105}

For the next actinide dioxide CmO_2 , only a few experiments have thus far been reported for the detailed constitution of its ground state.^{19,106} Cm⁴⁺ has six electrons in the 5*f* shell, producing a nonmagnetic ⁷ F_0 ground state from Hund's rules. However, a paramagnetic moment has been observed in CmO₂ which is unexpected from the nonmagnetic ground state.¹⁹ Niikura and Hotta explained the magnetic behavior of CmO₂ by assuming the proximity of a magnetic excited state with an excitation energy smaller than the Landé interval rule.¹⁰⁷

III. COMPUTATIONAL METHOD

The LDA + U method^{108–111} provides the one electron Hamiltonian as

$$h_{\text{LDA}+U} = h_{\text{LDA}} + \sum_{\tau} \sum_{\gamma\gamma'} |\tau \ell \gamma \rangle v_{\gamma\gamma'}^{\tau \ell} \langle \tau \ell \gamma' |, \qquad (1)$$

where the on-site +U potential is given by

$$v_{\{ms\}\{m's'\}}^{\tau\ell} = \sum_{m''m'''} \left[\delta_{ss'} \sum_{s''} n_{\{m''s''\}\{m''s''\}}^{\tau\ell} \langle mm''|W|m'm''' \rangle - n_{\{m''s\}\{m''s'\}}^{\tau\ell} \langle mm''|W|m'''m' \rangle \right] - \delta_{mm'} \delta_{ss'} \left[U(n-1/2) - \frac{J}{2}(n-1) \right], \quad (2)$$

where τ and ℓ denote the atoms and angular momenta of the orbitals, respectively, for which the +U potentials are introduced. γ (γ') is an index related to an orbital m (m') and a spin s (s'), or, alternatively, double-valued irreducible representations of the site symmetry of the An site obtained through a unitary transformation. The matrix elements of the full Coulomb interaction of two f electrons is expressed as

$$\langle m_1 m_2 | W | m_3 m_4 \rangle = \delta_{m_1 + m_2, m_3 + m_4} \sum_{k=0}^{0} c^k (\ell m_1; \ell m_3) \\ \times c^k (\ell m_4; \ell m_2) F^k,$$
(3)

where the F^k are the Slater-Condon parameters,^{112,113} and c_k is the Gaunt coefficient.^{114,115} In our calculations, F_0 is taken as $F_0 = U$, and the Hund's coupling parameter J is related with the higher order Slater integrals as $J = (286F_2 + 195F_4 + 250F_6)/6435$ for f electrons. The local potentials within the muffin-tin (MT) spheres are determined from the spin-orbital dependent density matrix,

$$n_{\gamma\gamma'}^{\tau\ell} = \int_{\mathrm{MT}} dr^{\tau} \{r^{\tau}\}^2 \rho_{\gamma\gamma'}^{\tau\ell}(r^{\tau}), \qquad (4)$$

which contains a projection from the band states $|kb\rangle$ to the local basis $|\tau \ell \gamma\rangle$,

$$\rho_{\gamma\gamma'}^{\tau\ell}(r^{\tau}) = \frac{1}{N} \sum_{kb} \langle \tau \ell \gamma | kb \rangle \langle kb | \tau \ell \gamma' \rangle, \tag{5}$$

N being the number of k points and r^{τ} is the radial component of the position vector r^{τ} measured from atom τ . The density matrix as well as the charge density are determined self-consistently.

The calculations have been performed for nonordered states in the series of AnO₂ and for ordered states in UO₂, NpO₂, and AmO₂. In the calculations of nonordered states, we used the fcc unit cell with the space group $Fm\bar{3}m$ (No. 225) and applied a relation $n_{-m-m'-s-s'}^{i\ell} = (-1)^{m+m'+s-s'} n_{mm'ss'}^{i\ell}$, imposed by time-reversal symmetry.^{45,51} The ordered states are calculated with the four sublattice unit cell mentioned above. It is known that the large U introduced in the LDA + U method can induce some metastable states especially in calculations of ordered states and may lead to convergence to an electronic state that is inconsistent with the realistic ground state.^{38,116–119} To avoid this problem, we made use of experimental information concerning the CEF ground states and the order parameters to control the occupations for the initial density matrix and adapted the symmetry preserving the 3*q* structure of the order parameters observed for UO₂ and NpO₂. Doing so, we could confirm that the calculations successfully stabilized the ordered states after convergence of the charge density and the density matrices on the An sites.

The (noncollinear) magnetic multipole moments can be described through the local LDA + U potentials. The expectation values of the local operators $O^{\tau \ell}$ defined on a spinorbital space of the An atoms are calculated with the local basis set { $|\tau \ell \gamma\rangle$ } inside the MT spheres, following the expressions

$$O^{\tau\ell}(\boldsymbol{r}^{\tau}) \equiv \frac{1}{N^2} \sum_{\boldsymbol{k}b} \sum_{\boldsymbol{k}'b'} \sum_{\boldsymbol{\gamma}\boldsymbol{\gamma}'} \langle \boldsymbol{r}^{\tau} | \boldsymbol{k}b \rangle \langle \boldsymbol{k}b | \tau\ell\boldsymbol{\gamma}\rangle O^{\tau\ell}_{\boldsymbol{\gamma}\boldsymbol{\gamma}'} \\ \times \langle \tau\ell\boldsymbol{\gamma}' | \boldsymbol{k}'b' \rangle \langle \boldsymbol{k}'b' | \boldsymbol{r}^{\tau} \rangle \tag{6}$$

and

$$\langle O^{\tau\ell} \rangle = \int_{\mathrm{MT}} d\mathbf{r}^{\tau} O^{\tau\ell}(\mathbf{r}^{\tau})$$

$$= \sum_{\gamma} \sum_{\gamma_1 \gamma_2} \int dr^{\tau} \{r^{\tau}\}^2 \rho_{\gamma\gamma_1}^{\tau\ell}(r^{\tau}) O_{\gamma_1\gamma_2}^{\tau\ell} \rho_{\gamma_2\gamma}^{\tau\ell}(r^{\tau}), \qquad (7)$$

within the space limited to an orbital ℓ . Explicit expressions for the local multipole operators have been listed by Kusunose.¹¹ We used the exchange-correlation functional of Gunnarsson and Lundqvist for the LDA potential.¹²⁰ The Coulomb *U* parameter has been chosen as U = 4 eV and the exchange *J* in the range of 0–0.5 eV. These values have previously been shown to provide an accurate description of measured properties of actinide dioxides.^{39,121,122} The double-counting term has been chosen as in the fully localized limit,¹²³ leaving out the spin dependency of the Hund's coupling part to adapt it for the nonmagnetic LDA part of Eq. (1).

In the basis set we have included the Np 5 f, 6d, 7s, and 6p orbitals as valence states and Np 5d, 6s orbitals as semicore states and for oxygen we treated the 2s and 2p states as valence states. The MT sphere radii were 1.4 Å for An and 0.9 Å for O. The plane-wave cutoffs used in the calculations were about 250 and 900 plane waves at the Γ point for calculations with the fcc normal cell and for the four-sublattice sc unit cell in the multipolar calculations, respectively. In reciprocal space we used for the self-consistent convergence [charge density and density of states (DOS) calculation] 12 × 12 × 12 (24 × 24 × 24) k points in the fcc unit cell and 6×6×6 (12 × 12 × 12) for the four-sublattice sc unit cell.

IV. RESULTS AND DISCUSSION

A. Nonordered state calculations

We first discuss the results obtained from the nonordered state calculations to examine how the anisotropic f character is reproduced with the LDA + U method.

In Figs. 2, 3, and 4 we show the band structures and the DOS, respectively, calculated for the AnO₂ compound by the LDA, and by the LDA + U method, with U = 4 eV and J = 0 and J = 0.5 eV, respectively. For comprehension we show in



FIG. 2. (Color online) Calculated band structures and density of states (DOS) for the nonmagnetic solution (see text) of the AnO_2 , calculated by the LDA. The presence of a gap at the Fermi energy is indicated by the shaded area. The position of the Fermi energy is at 0 eV.





FIG. 3. (Color online) As Fig. 2, but calculated with the LDA + U method (U = 4 eV, J = 0).

Fig. 5 the fcc BZ of the nonordered state, with high-symmetry points indicated. Figure 2 exemplifies that the LDA approach only correctly predicts that nonordered CmO₂ would be an insulator. In the LDA calculations, the $|j = \frac{5}{2}, \Gamma_7\rangle$ and the $|j = \frac{5}{2}, \Gamma_8\rangle$ states are present in the same energy range around the Fermi level, leading to a more or less homogeneous f



FIG. 4. (Color online) As Fig. 2, but calculated with the LDA + U method (U = 4 eV, J = 0.5 eV).

electron occupation for the j = 5/2 orbitals. The computed orbital occupations of the f states are given in Table I.

As a consequence of the homogeneous occupation, the f states predicted by the LDA calculations do not have sufficient anisotropic character as would be expected from the CEF states. In other words, the LDA calculations fail to produce the



FIG. 5. (Color online) The first Brillouin zones of the fcc unit cell used for calculations of nonmagnetic or nonmultipolar ordered actinide dioxides (red lines) and the sc Brillouin zone of the four-sublattice unit cell for 3q-ordered states (blue lines). High-symmetry points are indicated.

CEF ground states in the AnO₂ series as was also previously pointed out,^{21,22} whereas the correct CEF behavior for lighter actinide dioxides is reproduced if only the lower Γ_8 states are taken into account.

Conversely, the LDA + U calculations (Figs. 3 and 4) do lead to anisotropic f states, in which the $|j = 5/2, \Gamma_8\rangle$ quartet is dominantly occupied for An = U, Np, and Pu in AnO₂ due to a clear splitting between the $|j = 5/2, \Gamma_7\rangle$ and $|j = 5/2, \Gamma_8\rangle$ orbitals.

For PuO₂, an insulator state is obtained with the |j = 5/2, Γ_8 orbitals being fully occupied, see Table I. This state is consistent with the experimentally observed nonmagnetic insulator ground state with a singlet Γ_1 CEF state,^{18,24} as is illustrated in the one-electron CEF states shown in Fig. 6. Meanwhile, a previous first-principles study of PuO₂ reported that magnetic states are energetically more favorable than the



FIG. 6. (Color online) Schematic description of spin-orbit split, nonmagnetic crystal-field ground states in terms of one-electron f orbitals. Numbers in brackets give the degeneracy of the one-electron orbitals.

nonmagnetic state for PuO_2 within LDA + U calculations.⁴⁰ This is probably related to the approximations inherent to the LDA + U method, in which the local exchange interaction for the electron Coulomb interaction is introduced within the Hartree-Fock approximation. The physical property of the experimentally observed nonmagnetic ground state should be captured with the computed nonmagnetic solutions, which reproduce the experimentally observed Γ_1 singlet crystal field ground state. In nonmagnetic LDA + U calculations for PuO_2 , the energy gaps are reduced with increasing J as illustrated in Fig. 4. In this regard the J, which introduces Hund couplinglike electron interaction, appears to make the nonmagnetic insulator ground state less stable in PuO₂. In addition, we mention that it has previously been shown that obtaining the nonmagnetic ground state of δ -Pu with LDA + U calculations depends sensitively on the type of the employed +U functional.⁴⁹ The anisotropic character of the f states in PuO_2 can be visually recognized from the *f*-charge distribution on the Pu sites shown in Fig. 7(a). It can be seen that the charge distribution is extended to the [100] and the equivalent axes, reflecting an f wave function with Γ_8 symmetry.

In UO₂ and in NpO₂ the strong ionization constrains the f states to containing two or three electrons by the strong CEF

TABLE I. Calculated 5 *f* electron occupation numbers per one-electron orbital on the An sites in the AnO₂ (An = U, Np, Pu, Am, and Cm) fluorite-structure compounds. The calculations have been performed assuming the nonmagnetically ordered state, both with the LDA and with the LDA + U (U = 4 eV, and J = 0 and 0.5 eV) approaches.

	Approach	j = 5/2			j = 7/2			
		$\Gamma_7(2)$	$\Gamma_8(4)$	Total	$\Gamma_6(2)$	$\Gamma_7(2)$	$\Gamma_8(4)$	Total
UO ₂	LDA	0.41	1.59	2.00	0.11	0.20	0.23	0.54
	LDA + U, J = 0	0.14	2.00	2.14	0.06	0.13	0.11	0.31
	LDA + U, J = 0.5 eV	0.18	2.02	2.21	0.06	0.13	0.10	0.29
NpO ₂	LDA	0.81	2.18	3.00	0.11	0.26	0.27	0.65
	LDA + U, J = 0	0.16	3.09	3.25	0.06	0.06	0.29	0.41
	LDA + U, J = 0.5 eV	0.23	3.12	3.35	0.06	0.13	0.10	0.28
PuO ₂	LDA	1.16	2.81	3.97	0.12	0.31	0.35	0.78
	LDA + U, J = 0	0.22	3.86	4.09	0.08	0.19	0.20	0.47
	LDA + U, J = 0.5 eV	0.41	3.87	4.29	0.07	0.14	0.13	0.34
AmO ₂	LDA	1.54	3.34	4.88	0.15	0.37	0.44	0.96
	LDA + U, J = 0	1.91	3.44	5.34	0.07	0.26	0.13	0.46
	LDA + U, J = 0.5 eV	1.86	3.72	5.58	0.06	0.13	0.09	0.28
CmO ₂	LDA	1.85	3.80	5.65	0.19	0.46	0.60	1.25
	LDA + U, J = 0	1.92	3.93	5.85	0.13	0.41	0.29	0.83
	LDA + U, J = 0.5 eV	1.97	3.94	5.92	0.11	0.30	0.25	0.66



FIG. 7. (Color online) Charge distributions of the An-5*f* electrons (a) in PuO₂ and (b) in CmO₂ computed with the LDA + U (using U = 4 eV and J = 0.5 eV). The distributions in the left-hand panels are viewed from the [1 0 0] direction, those in the right-hand panels from the [1 1 1] direction. The lines on the outside of the spheres represent the contour lines for the charge distribution (cf. Ref. 124).

effect. As a result, the Γ_8 orbitals being occupied with two or three electrons lead to metallic electronic states in UO₂ and in NpO₂ in the nonordered-state calculations.

In AmO₂, the Γ_7 doublet is populated with about two electrons and the Γ_8 quartet has some hole (see Table I), rendering also AmO_2 metallic in nonordered-state LDA + Ucalculations. Note that in the LDA + U calculations with $J = 0.5 \,\mathrm{eV}$ the majority of empty Am i = 7/2 states is located above 4 eV in Fig. 4. Our results obtained from nonmagnetic calculations imply that the nonordered states in AmO₂ should exhibit a strong susceptibility due to the Γ_8 orbitals. The finding of an incompletely filled Γ_8 orbital is consistent with the recent CEF theory suggesting that multipole order is realized within the Γ_8 CEF ground state in AmO₂.¹⁰¹ Since the Γ_7 CEF ground states, which were suggested on the basis of experimental results, ^{15,99,100} could not bring about any higher order multipoles, our results support the Γ_8 CEF ground states in AmO₂. Furthermore, our results purport that symmetry breaking of some sort is necessary to reproduce the experimentally observed insulator ground states in UO₂, NpO₂, and AmO₂. Especially, the insulator ground states can evidently not be produced with the nonordered-state calculations for NpO_2 and for AmO_2 , since the fcc primitive cells containing an odd number of electrons will always lead to uncompensated metallic states when time-reversal symmetry is present.

For CmO₂, both the LDA and the LDA + U calculations produce the insulator states in the nonmagnetic-state calculations. We can speculate that a reason that the LDA approach is already a good approximation for CmO₂ comes from, first, the strong SO splitting filling the j = 5/2 manifold, and second, the weakly anisotropic character of the f states in the nonmagnetic ground state, which can be visually seen in Fig. 7. The Cm 5 f charge distribution, seen from the [1 0 0] direction [Fig. 7(b)], is much more isotropic than that of Pu [Fig. 7(a)]. Assuming that CmO₂ has a tetravalent Cm⁴⁺ ion just as the An ion in the other actinide dioxides, a singlet ground state with J = 0 is expected. In this regard, the nonmagnetic insulator states obtained in both calculations seem to be a natural consequence of the tetravalent ionized state in CmO₂. We note that energetically more favorable magnetic states in CmO₂ could be present due to the limitation of LDA + U method just as in the case of PuO₂, but such magnetic states would be inconsistent with the expected nonmagnetic Γ_1 singlet ground state. A neutron diffraction experiment reported detection of an effective paramagnetic moment $\mu_{eff} \sim 3.4 \ \mu_{B}$, which would be consistent with the Curie-Weiss behavior observed for the magnetic susceptibility.¹⁹ Niikura and Hotta investigated the possibility of having a magnetic excited state just above the nonmagnetic ground state in the Cm⁴⁺ ion with a small excitation energy, aiming to provide an understanding for the unexpected magnetic behavior in CmO₂.¹⁰⁷

B. Ordered state calculations

Next, we have performed electronic structure calculations for the UO₂, NpO₂, and AmO₂ compounds allowing for selfconsistent convergence to a symmetry-broken ordered ground state. Figure 5 shows the sc Brillouin zone (BZ) for the 3qordered state, which displays a relationship to the fcc BZ of the nonordered state. In the following the results of the orderedstate calculations are discussed in detail for each of these actinide dioxides.

For UO_2 we choose the initial density matrix to correspond to the nonordered state with the Zeeman-type field along [111] equivalent directions, keeping a transverse-3q structure as a symmetry breaking term. The converged electronic structure corresponds to the transverse 3q ordered state with Γ_4 local multipoles. The obtained band structures and DOS are shown in Figs. 8 and 9, for U = 4 eV and two J values, 0 and 0.5 eV, respectively. The energy gaps upon convergence are generated through the splitting of the f states, which is in turn induced through the large U. In nonmagnetic UO₂ the plus and minus j_z components of the $j_z = \pm 5/2, \pm 3/2, \pm 1/2$ orbitals are degenerate; because of the Coulomb U these orbitals split into the lower and upper Hubbard bands in the ordered states as seen in Fig. 9. For U = 4 eV the experimentally observed energy gap^{25} of about 2 eV is well reproduced with the 3qmagnetic order in the calculations.

In the transverse 3q magnetic order, the dipolar magnetic moments are induced along [1 1 1] axis for the U atom at (0,0,0), along the $[-1 \ 1 \ -1]$ axis for the U at (0,1/2,1/2), along the [-1 - 1 1] axis for the U at (1/2, 0, 1/2), and along $\begin{bmatrix} 1 & -1 & -1 \end{bmatrix}$ for the U at (1/2, 1/2, 0). The local magnetic moments on the uranium sites are shown in Fig. 10 as a function of the Hund's coupling parameter J. As seen from the figure, the magnetic moment is dominated by the orbital moment and is gradually enhanced for increasing J. The calculated orbital and spin moments are $\langle L_{111} \rangle =$ $\frac{1}{\sqrt{3}}$ $[|\langle L_x \rangle| + |\langle L_y \rangle| + |\langle L_z \rangle|] = 1.09 \ \mu_B$ and $\langle S_{111} \rangle = \frac{1}{\sqrt{3}}$ $[\langle S_x \rangle + \langle S_y \rangle + \langle S_z \rangle] = -0.14 \,\mu_B \text{ for } J = 0 \text{ and } \langle L_{111} \rangle =$ 1.71 and $\langle S_{111} \rangle = -0.23 \ \mu_B$ for J = 0.5 eV. The total local magnetic moments are $\langle M \rangle = \langle L_{111} \rangle + 2 \langle S_{111} \rangle = 0.80 \, \mu_B$ for J = 0 and 1.26 μ_B for J = 0.5 eV, providing slightly smaller values compared with the experimental value^{25,68} of 1.74 μ_B . Our results show that the complex constitution of the f ground



FIG. 8. (Color online) Computed band structures and density of states in the transverse $3q \Gamma_{4u}$ magnetic MPO state of UO₂, calculated with the LDA + U method [U = 4 eV, J = 0 (top) and J = 0.5 eV (bottom)]. The gap computed at the Fermi energy is depicted by the shaded area.

state, with contributions from the hybridization between U-5 *f* and O-2*p* states as seen in Fig. 9, leads to a reduction of the magnetic moment expected for the Γ_5 CEF ground state, which is slightly higher than 2.0 μ_B .⁶⁹ We further provide in Fig. 10 the *J* dependence of the quadrupole moments $\langle O_{111} \rangle = \frac{1}{\sqrt{3}} [\langle O_{yz} \rangle + \langle O_{zx} \rangle + \langle O_{xy} \rangle]$, calculated from Eq. (7)



FIG. 9. (Color online) The orbital-projected DOS components computed for the multipolar ordered phase of UO₂, for J = 0 (left) and J = 0.5 eV (right).



FIG. 10. (Color online) Top: Computed magnetic dipole moments in UO₂ as a function of the Hund's exchange *J* parameter. Shown are the spin moment $\langle S_{111} \rangle$, the orbital moment $\langle L_{111} \rangle$, and the total moment $\langle M_{111} \rangle = \langle L_{111} \rangle + 2\langle S_{111} \rangle$ (see text). Bottom: The computed quadrupole moments $\langle O_{111} \rangle = \frac{1}{\sqrt{3}} [\langle O_{yz} \rangle + \langle O_{zx} \rangle + \langle O_{xy} \rangle]$ in UO₂ as a function of *J* applied to the *f* electrons.

for the uranium f electrons. We find that the quadrupole moments also develop with increasing J, changing the sign around J = 0.03 eV. The finite contribution from the quadrupole moments is consistent with recent experiments.^{13,64}

In Fig. 11 the spatial shape of the uranium 5f-wave function is displayed by plotting its magnetic moment distribution projected to the [1 1 1] local axis on an isodensity surface of the calculated *f*-charge density for U = 4 eV and two *J* values, 0 and 0.5 eV. It is seen that an overall, dipolar magnetic moment exists along the [1 1 1] axis and that the local moments are enhanced with the effect of *J*. Also, reflecting the local site symmetry of the ordered state, the calculated charge and magnetic distributions preserve the C_{3i} symmetry for UO₂.

Next, we consider NpO₂. The obtained band structures and DOS are shown in Figs. 12 and 13, for U = 4 eV and two J values, 0 and 0.5 eV, respectively. For this actinide dioxide the computed electronic structure corresponds to the longitudinal 3q ordered state with Γ_5 local multipoles.

A schematic picture of how the one-electron f orbital levels will split, depending on the symmetry at the Np sites in NpO₂, has been provided in Fig. 1. The Γ_5 multipole order lowers the O_h local symmetry to the D_{3d} symmetry on the Np sites. Thus, the $|j = 5/2, \Gamma_8\rangle$ quartet splits into the two singlets Γ_4 and Γ_5 and one doublet $\Gamma_6^{(1)}$ in the MPO states. The two singlets Γ_4 and Γ_5 are degenerated under time-reversal invariance but not in the *magnetic* MPO states, which break the time-reversal



FIG. 11. (Color online) Charge and magnetic distributions of the U-5*f* electrons in UO₂, computed with (a) U = 4 eV and J = 0 and (b) U = 4 eV and J = 0.5 eV. The distributions in the left-hand panels are viewed from the [1 0 0] direction, and those in the right-hand panels from the [1 1 1] direction, corresponding to the threefold axis. The charge distributions are depicted by the isodensity surface, and the distributions of magnetic moments $M_{[111]}(\mathbf{r}) = \frac{1}{\sqrt{3}} \{M_x(\mathbf{r}) + M_y(\mathbf{r}) + M_z(\mathbf{r})\}$ (shown by the color code) are calculated from Eq. (6) and plotted on the isosurface of the charge density.



FIG. 12. (Color online) Computed band structures and density of states in the longitudinal $3q \Gamma_{5u}$ magnetic MPO state of NpO₂, calculated by the LDA + U method (U = 4 eV, J = 0 or J = 0.5 eV).



FIG. 13. (Color online) The orbital-projected DOS components computed for the multipolar ordered phase of NpO₂, for J = 0 (left) and J = 0.5 eV (right).

symmetry. The two doublets derived from the Γ_7 and Γ_8 orbitals belong to the same symmetry and hybridize with each other in the MPO states. A Γ_5^- multipole is expected to be produced with the three 5 f electrons occupying the Γ_4 singlet and the $\Gamma_6^{(1)}$ doublet, which are split from the Γ_8 quartet in the paramagnetic state. Accordingly, we thus choose, as an initial density matrix, the one corresponding to the local f states in which the Γ_4 singlet and the $\Gamma_6^{(1)}$ doublet are occupied by three f electrons. In the longitudinal 3q structure, the induced Γ_5 multipoles obey a relation, for instance, $\langle O_{yz} \rangle = \langle O_{zx} \rangle =$ $\langle O_{xy} \rangle$ for the Np (0, 0, 0) ion, $\langle O_{yz} \rangle = -\langle O_{zx} \rangle = -\langle O_{xy} \rangle$ for the Np ion at $(0, 1/2, 12), -\langle O_{yz} \rangle = \langle O_{zx} \rangle = -\langle O_{xy} \rangle$ for the Np (1/2, 0, 1/2) ion, and $-\langle O_{yz} \rangle = -\langle O_{zx} \rangle = \langle O_{xy} \rangle$ for the Np (1/2, 1/2, 0) ion. Reflecting the D_{3d} local symmetry, the Np-f orbital components in the DOS still keep the degeneracy for the Γ_6 doublets in the exotic magnetic multipole ordered states as shown in Fig. 13.

The top-right panel of Fig. 14 shows the calculated J dependence of the multipole moments $\langle O \rangle$ normalized to the multipole moments for the initial f occupation $\langle O \rangle_0$, as mentioned above. The noncollinear multipole moments in the ordered state of NpO₂ are strongly affected by the value of the Hund's coupling J, which also affects the occupation difference of f orbitals, see the top-left panel. At J = 0, the Γ_4 singlet and the $\Gamma_6^{(1)}$ doublet are fully occupied. As J increases, the f electron on the $\Gamma_6^{(1)}$ doublet starts to transfer to the $\Gamma_6^{(2)}$ doublet through hybridization. This transfer strongly enhances the triakontadipole moment and suppresses the octupole moment in NpO₂. The occupation of the one-electron CEF orbitals and its relation to the multipolar order is schematically illustrated in Fig. 15. Thus, the Γ_5 -triakontadipole moment can be the leading order parameter in NpO₂.²⁰

The charge and magnetic distributions of the neptunium 5f-wave function are plotted in Fig. 16 for U = 4 eV and two J



FIG. 14. (Color online) Left panels: Calculated one-electron orbital occupation numbers as a function of exchange J at U = 4 eV, for the MPO states of NpO₂ (top) and AmO₂ (bottom). Right panels: Calculated expectation values of the multipolar (quadrupolar, octupolar, hexadecapolar, and triakontadipolar) order parameters in NpO₂ and AmO₂ as a function of J.

values, 0 and 0.5 eV. In contrast to the equivalent distributions of the U 5 f wave function in Fig. 11, the magnetic distribution corresponds to a vanishing atomic magnetic dipole moment along the [1 1 1] local axis after the space integration. The calculated charge and magnetic distributions preserve the D_{3d} symmetry of NpO₂. The enhancement of the triakontadipole moment with increasing J is reflected in the enhanced local magnetic moments on the charge isosurface in Fig. 16(b).

Next, we consider AmO₂. The order parameter in the ground state of AmO₂ has not been determined experimentally yet. The recent theoretical study¹⁰¹ of Hotta suggested the Γ_8 CEF ground state to explain the experimental facts.¹⁰² It is hence plausible to have order parameters which are similar to those of NpO₂. We thus choose initial density matrices, so



FIG. 15. (Color online) Schematic picture showing the correspondence of the contributing j = 5/2 orbitals and the active multipoles in the Γ_5 multipolar order.



FIG. 16. (Color online) Charge and magnetic distributions of the Np-5 f electrons in NpO₂, computed with (a) U = 4 eV and J = 0 and (b) U = 4 eV and J = 0.5 eV. Details of the plots are as given in Fig. 11.



FIG. 17. (Color online) Computed band structures and density of states in the longitudinal $3q \Gamma_{5u}$ magnetic MPO state of AmO₂, calculated by the LDA + U method (U = 4 eV and J = 0 or J = 0.5 eV).

as to have finite Γ_5 multipole moments, corresponding to the local 5 *f* state which has filled $\Gamma_6^{(1)}$ and $\Gamma_6^{(2)}$ doublets as well as the Γ_4 singlet in the calculations for ordered states in AmO₂. Then, our calculations lead to the solution of the longitudinal 3*q* ordered state with Γ_5 -local multipoles only for small *J* as shown in Fig. 14. The obtained band structures and DOS are shown in Figs. 17 and 18 for U = 4 eV and again two *J*



FIG. 18. (Color online) As Fig. 13, but for the multipolar ordered phase of AmO_2 .



FIG. 19. (Color online) Charge and magnetic distributions of the Am-5 f electrons in AmO₂, computed with (a) U = 4 eV and J = 0 and (b) by U = 4 eV and J = 0.5 eV. Details of the plots are as given in Fig. 11.

values, 0 and 0.5 eV, respectively. The charge and magnetic distributions of the americium 5f-wave function are plotted in Fig. 19 for U = 4 eV and two J values, 0 and 0.5 eV. These are considerably different from the equivalent distributions computed for the Np 5f wave function in Fig. 16, reflecting the substantially different constitution of the high-rank multipoles in AmO₂ from that of NpO₂, see Fig. 14.

We have found that the multipolar ordering is responsible for stabilizing the insulating ground state for small values of J in AmO_2 , like in NpO₂; the resulting gap is indicated by the shaded area in Fig. 17. Meanwhile, the constitution of the high-rank multipoles in AmO₂ differs substantially from the one of NpO₂, see Fig. 14. The Γ_5 multipoles in AmO₂ are rather insensitive for small J values (bottom-right panel), whereas they decrease for J values larger than 0.1 eV and show an abrupt disappearance of the multipole order for Jvalues slightly smaller than 0.3 eV. This striking difference of the J dependence of the multipoles in AmO_2 from those of NpO₂ stems from the fact that the $|j = 5/2, \Gamma_7\rangle$ orbitals are completely occupied in the f^5 state of the Am⁴⁺ ion and there is no state available to couple to it through the Hund's coupling J. Furthermore, the calculated energy gap for AmO_2 is found to depend strongly on the Hund's coupling J, see Fig. 17. Clearly, with increasing J a magnetically ordered solution becomes more favorable. A similar sensitivity was not observed for UO_2 nor for NpO_2 . As can be seen from Fig. 17, the energy gap in AmO_2 decreases with increasing J and disappears simultaneously with the disappearance of the multipole order as discussed above. This implies that a large Hund's rule J can make the insulator solution as well as the longitudinal 3q ordered state of Γ_5^- multipoles unstable in AmO_2 .

V. CONCLUSIONS

We have investigated the origin of the gap formation in the actinide dioxides. The origin of the insulating gap formation is found to lie in the strong on-site Coulomb repulsion of the 5 f electrons and strong spin-orbit interaction in the relevant forbitals in the AnO₂ compounds. LDA + U calculations for a non-long-range ordered state reproduce well the energy gaps following the singlet CEF ground states in PuO₂ and CmO₂. On the other hand, the insulating ground states in UO_2 , NpO_2 , and AmO_2 are obtained only by allowing for the symmetry lowering that can give rise to the ordered states. Thus, the strong correlation is necessary to describe the anisotropic f ground states in AnO₂. The energy gaps and magnetic properties are correctly reproduced within the accepted range of the parameters, U and J, by taking the proper magnetic space symmetries in the calculations. Especially, using values of J in the acceptable range produces magnetic moments which are reduced from the one expected from the CEF ground states, yet with the appropriate experimental energy gap in UO₂. In addition, the contribution from the electric quadrupole moments is enhanced with increasing J. We also showed that the active multipoles in these ordered states are closely related to the orbital occupation, and the higher rank Γ_5 multipole

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ordered states in AmO_2 have quite different constitution of the multipoles from NpO₂. Whereas Hund's coupling J enhances

the energy gap in NpO2 together with changing the constitution

of high-rank multipoles, it makes the insulator states with 3q

longitudinal Γ_5 multipole order unstable in AmO₂. Further

experimental investigations are required and encouraged to

verify the here-computed ground state properties of AmO₂

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