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Temperature-dependent pressure-induced softening in Zn(CN)₂

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We study the temperature dependence of the pressure-induced softening in the negative thermal expansion material $Zn(CN)_2$ using neutron powder diffraction and molecular dynamics simulations. Both the simulation and experiment show that the pressure-induced softening only occurs above a minimum temperature and also weakens at high temperatures.

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I. INTRODUCTION

The phenomenon of pressure-induced softening, whereby the elastic stiffness of a material actually decreases on compression, is a counterintuitive mechanical response that is rarely observed in materials, at least over extended pressure ranges. The few known examples from experiment include silica glass¹ and two negative thermal expansion (NTE) framework materials ZrW₂O₈ (Ref. 2) and Zn(CN)₂.³ We have recently predicted the existence of pressure-induced softening in a number of cubic zeolites, all of which show NTE.⁴

The intuition that materials should become stiffer under compression is grounded in the observation that shortening bonds increases the steepness of the corresponding potential energy surface. Thus, there is a strong fundamental physics interest in understanding the apparent violation of this basic mechanical response. It is also likely that the existence of this counterintuitive response in materials that display the equally rare phenomenon of volume NTE is not coincidental.⁴ That pressure-induced softening may signal the existence of a host of unusual thermodynamic responses (e.g., NTE) with important technological applications acts as further motivation to understand both its microscopic origin and its implications for other material properties.

In high-symmetry structures, elastic stiffness can be characterized by the bulk modulus $B = -(\partial \ln V/\partial p)_T^{-1}$, where V is the volume, p is the pressure, and T is the temperature. Typical values of $B_0 = B(p=0)$ lie in the range 30–100 GPa, with larger values corresponding to materials of increasing mechanical stiffness. Experimentally, B_0 is usually measured by fitting the pressure dependence of the crystallographically determined unit cell volume to an appropriate equation of state. The pressure dependence of the bulk modulus is characterized by the dimensionless parameter $B_0' = (\partial B/\partial p)_T \mid_{p=0}$, which for many materials has a value in the vicinity of +4 at ambient temperature; indeed, this is the value to which the widely used second-order Birch-Murnaghan equation of state corresponds. So, in other words, for most materials one

expects an increase of 5%–10% in bulk modulus for each 1 GPa increase in hydrostatic pressure.

In this context, measured *negative* values of $B'_0 = -17$ ($B_0 = 76$ GPa) (Ref. 2) for ZrW_2O_8 , and $B'_0 = -6.0(7)$ [$B_0 = 34.19(21)$ GPa] (Ref. 3) or $B'_0 = -8.6(13)$ [$B_0 = 36.9(22)$ GPa] (Ref. 8) for $Zn(CN)_2$ are strikingly anomalous. In both cases, an applied pressure of 1 GPa causes the material stiffness to *decrease* by $\sim 20\%$. Discussion of the origin of pressure-induced softening has been given in Refs. 2, 9, and 10, the first two corresponding in practice to a model that represents a two-dimensional idealization of $Zn(CN)_2$ (see Appendix A). These works all associated the negative value of B'_0 with dynamical effects associated with low-frequency phonon modes involving rotations of quasirigid polyhedral groups of atoms.

What has not yet been seen at all is the dependence of pressure-induced softening on temperature in any material. In this paper, we study this effect in $Zn(CN)_2$ using both molecular dynamics (MD) simulations and variable-p/T neutron powder diffraction. Our MD simulations reveal a very strong temperature dependence to B_0' , which even involves a change in sign. Based on an interpretation of the relevant fluctuations, we propose an empirical form for the $B_0'(T)$ function. A simple single-particle Hamiltonian of a rigid-unit mode system similar to $Zn(CN)_2$ was analyzed in Ref. 2, but this gives a prediction for the temperature dependence of pressure-induced softening that differs from the results presented in this paper; we discuss this in Appendix A.

This paper begins with a brief discussion of the lattice dynamics of $Zn(CN)_2$, from which we predict qualitatively the form of the $B_0'(T)$ function. We proceed to describe first the MD simulations we have performed, which establish the temperature dependence of B_0' for this material, and second the neutron diffraction experiments that provide some confirmation of the MD results. We conclude with a discussion of the implications of our study for computational and experimental investigations of pressure-induced softening in related systems.

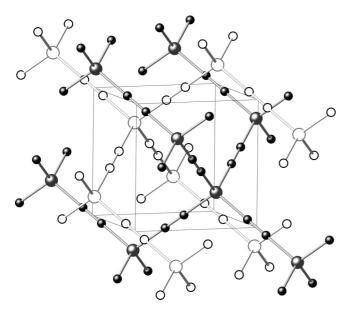


FIG. 1. Crystal structure of Zn(CN)₂ showing the two sublattices where one is shown using photorealistic shading to represent the atoms and the other with the atoms represented by open spheres. The primary features are the tetrahedral coordination of the Zn atoms (larger spheres) to the CN molecular anions (smaller spheres connected by a short bond), and the linear Zn-cyanide–Zn connectivity.

II. LATTICE DYNAMICS AND A PHENOMENOLOGICAL MODEL

From a lattice-dynamics perspective, the key point of interest concerning zinc cyanide has always been its NTE behavior:¹¹ the volume of its cubic unit cell decreases on heating at a rate that is more than double that of longer-established NTE systems such as ZrW₂O₈.¹² The crystal structure of Zn(CN)₂ (Ref. 13) can be described as a network of zinc atoms connected via linear Zn–C–N–Zn linkages as shown in Fig. 1. Each Zn center is coordinated in a tetrahedral fashion by four C or N atoms.¹⁴ It is thought that, to a first approximation, these tetrahedral units are not readily deformed, ^{11,15} so the dominant vibrational motion in zinc cyanide involves flexing of the Zn–C–N–Zn linkages as propagated in a family of low-energy transverse acoustic phonon modes with energies less than 1 THz. ^{16–19}

In the case of amorphous silica, molecular dynamics simulations suggested a relation between the bulk compressibility and the extent of network flexibility as measured by the magnitude of fluctuations involving whole-body rotations of SiO₄ tetrahedra. We can propose a similar explanation for the case of Zn(CN)2, where at some finite temperature the structure is buckled through rotations and translations of the Zn(C/N)₄ tetrahedra and cyanide bridges, the same fluctuations that are responsible for NTE. An initial application of negative pressure to expand the structure will first be accommodated through straightening of the Zn-C-N-Zn linkages and alignment of the Zn(C/N)₄ tetrahedra, which costs relatively little energy. This process will then be followed later by stretching of the Zn-C-N-Zn linkages which costs rather more energy. Thus, the bulk modulus is expected to increase with negative pressure, which implies that $\partial B_0/\partial p$ takes a negative value.

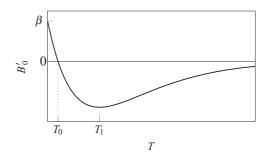


FIG. 2. Schematic curve showing the predicted variation of B'_0 with temperature T according to Eq. (1). This schematic defines the meanings of the parameters β , T_0 , and T_1 .

This explanation suggests that the magnitude of the pressure-induced softening should depend on temperature. At low temperature there will be no fluctuations to accommodate stretching of the structure, and we therefore expect a positive value of B_0' . Indeed, without fluctuations, compression of the structure will also require compression of bonds. Thus, we expect pressure-induced softening to occur only when there are thermal fluctuations, and we expect that on heating B_0' will initially have a positive value, which will reduce and become negative at some finite temperature. At higher temperatures, we might expect not to see the transition between straightening the fluctuations and stretching of bonds, so that B_0 will become relatively constant with pressure and hence B_0' will tend towards zero from its negative value.

We model the variation of B'_0 with temperature T in $Zn(CN)_2$ using the following phenomenological form:

$$B_0' = -\beta \left(\frac{T - T_0}{T_0}\right) \exp\left(-\frac{T}{T_1 - T_0}\right),\tag{1}$$

parametrized by the variables β , T_0 , and T_1 . This is illustrated in Fig. 2. We will find in the following that the temperature dependence of B'_0 in $Zn(CN)_2$ obtained from both molecular dynamics simulations and neutron powder diffraction experiments is described well by this phenomenological model.

III. MOLECULAR DYNAMICS SIMULATIONS

A. Method

Molecular dynamics (MD) simulations were performed along isotherms of $Zn(CN)_2$ at selected temperatures with an interatomic potential model developed from *ab initio* calculations.¹⁹ Atomic charges were obtained from a distributed multipole analysis.^{20,21} Morse potentials were used to describe the energies of chemical bonds, and bond angle terms were used to define the $Zn(C/N)_4$ tetrahedra. A linear three-body potential $E_L = K(1 - \cos \varphi)$ was used to describe the transverse vibrations of C and N in the linkages Zn–C(N)–N(C)–Zn. For these potentials, parameters were tuned by fitting to quantum mechanical calculations based on small clusters. The dispersion interactions between carbon and nitrogen atoms were modeled using the Buckingham potentials of Ref. 22. Further details are given in the Supplemental Material²³ and Ref. 19; Ref. 19 also provides comparison of the performance of the model against experimental data.

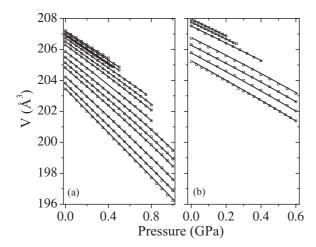


FIG. 3. Simultaneous fit to (a) the isotherm data sets (open squares) at (from up to down in the plot) 1, 10, 25, 30, 50, 75, 100, 150, 200, 250, 300, 400, 500, and 600 K from the MD; (b) the isotherm data sets (open squares) at (from up to down in the plot) 30, 50, 75, 150, 200, 250, and 300 K from the neutron diffraction, using the third-order Birch-Murnaghan equation of state. $V_0(T)$ and $B_0(T)$ in the equation of state were expressed as quadratic polynomials. $B_0'(T)$ was put into the phenomenological form of Eq. (1).

The MD simulations were performed using the DL_POLY code, 24 using a sample described as a $10 \times 10 \times 10$ supercell containing 10 000 atoms. Normal periodic-boundary conditions were used. A constant-stress and constant-temperature ensemble with a Nosé-Hoover thermostat²⁵ was used. The equations of motion were integrated using the leapfrog algorithm with a time step of 0.001 ps.

Each simulation was performed with an equilibration time of 20 ps followed by a run of 40 ps from which an average volume was calculated from writing the instantaneous volume at every 0.02 ps. Altogether, we performed 242 MD simulations.

B. Equation of state from simulation data

Plots of sample volume as a function of pressure for the range of temperatures are given in Fig. 3(a). Values of B_0 and its derivative B_0' for each temperature were obtained by fitting a third-order Birch-Murnaghan equation of state to the data, treating each temperature run separately, and shown in Fig. 4 and provided in Table I for later comparison with experimental data.

As a subsequent task, we refitted the data in Fig. 3(a) by expressing B_0' by the phenomenological form of Eq. (1) and using quadratic polynomials to describe the zero-pressure volume $V_0(T)$ and the zero-pressure bulk modulus $B_0(T)$ as functions of temperature. In this case, we performed an overall fit rather than fitting to each temperature data separately, adjusting the values of the parameters β , T_0 , and T_1 in Eq. (1), together with the parameters in the polynomials of volume and bulk modulus. Figures 3(a) and 4 show the good quality of the fit, and highlight the consistency of the phenomenological model with the MD data.

The fitted values of β , T_0 , and T_1 in Eq. (1) are 5.4(4), 34(2) K, and 168(8) K, respectively. From the simultaneous fit, we also obtained the coefficient of thermal expansion

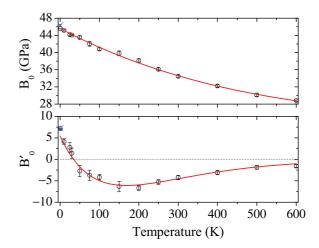


FIG. 4. (Color online) Bulk modulus at zero pressure B_0 and its first derivative B'_0 as functions of temperature, obtained from fitting the third-order Birch-Murnaghan equation of state to the simulated isotherms of $\text{Zn}(\text{CN})_2$. Values at zero temperature calculated in harmonic lattice dynamics using GULP (Ref. 26) are also included in the plot (blue cross). Red curves show the calculated $B_0(T)$ and $B'_0(T)$ using values of the parameters from the simultaneous fit, where $B_0(T)$ was described by a quadratic polynomial and $B'_0(T)$ was expressed by Eq. (1).

 $\alpha = \partial \ln V_0/\partial T$ and the derivative $\partial B_0/\partial T$; values for T = 300 K are given in Table II for later comparison with the experimental results (below). According to the thermodynamic expressions for α and B combined with the Maxwell relation $\partial^2 V/(\partial T \partial p)_{p,T} = \partial^2 V/(\partial p \partial T)_{T,p}$, the variation of α with pressure can be calculated as

$$\left(\frac{\partial \alpha}{\partial p}\right)_T = \frac{1}{B^2} \left(\frac{\partial B}{\partial T}\right)_p. \tag{2}$$

TABLE I. Simulation (Sim) and neutron powder diffraction (Exp) values of B_0 (units of GPa) and B_0' obtained as described in the text. Most of the simulation results are obtained using the MD method; the exceptions are the values at 0 K, which are taken from a separate simulation of the equation of state using the lattice energy code GULP (Ref. 26).

T(K)	B_0 (Sim)	B_0 (Exp)	B_0' (Sim)	B_0' (Exp)
0	46.37(17)		7.2(3)	
1	45.54(2)		7.1(1)	
10	45.15(17)		4.1(6)	
25	44.15(29)		3(1)	
30	44.07(16)	39.1(7)	1(1)	-2(7)
50	43.5(4)	38.5(9)	-3(1)	-3(6)
75	42.0(5)	38.8(10)	-4(1)	-8(5)
100	40.8(3)		-4.2(6)	
150	39.8(6)	36.8(4)	-6(1)	-7(1)
200	38.2(4)	36.1(3)	-6.7(5)	-8.1(8)
250	36.1(3)	35.6(3)	-5.3(5)	-9.0(7)
300	34.5(3)	33.4(3)	-4.2(5)	-4(1)
400	32.2(3)		-3.1(5)	
500	30.11(28)		-1.8(5)	
600	28.79(18)		-1.6(3)	

-23(2)

-31(21)

MD

Exp

(T = 300 K) from the MD and experimental studies.						
α	$\partial B_0/\partial T$	$\partial \alpha/\partial p$				

-48.9 to -40.9 (10 to 100 K) (Ref. 19)

-30.1(4) (300 K)

-47.7(7) (averaged over 30 to 300 K)

TABLE II. Comparison of the values of α (MK⁻¹), $\partial B_0/\partial T$ (GPa/K) (T=300 K), and $\partial \alpha/\partial p$ (MK⁻¹GPa⁻¹) (T=300 K) from the MD and experimental studies.

The value of $\partial \alpha/\partial p$ at p=0 and T=300 K is also given in Table II.

In the limit of the temperature decreasing to zero, the value of B'_0 from the MD simulations becomes positive. The zero-temperature value given in Table I and Fig. 4 has been computed from a lattice-energy calculation using the GULP code.²⁶ It is clear from Fig. 4 that the MD results are tending towards the zero-temperature lattice-energy result. The ambient-temperature values of B_0 and B'_0 calculated by MD are consistent with those reported in previous experimental studies cited in the Introduction.^{3,8}

It is possible that the values of B_0 and B'_0 obtained from fitting to the MD isotherms might not be directly comparable with experiment at low temperatures because the MD simulations follow the classical equipartition of energy. However, for NTE materials such as Zn(CN)2, most of the flexibility comes from vibrational modes that have low frequencies of the order of ≤ 1 THz. These modes contribute most strongly to NTE through their large negative Grüneisen parameters, which suggests that even at low temperatures such as \sim 50 K, these modes remain populated and hence contribute to the dynamics of the material. This is borne out from measurements of the variation of α with temperature, ¹¹ which show little sign of departure from the classical equipartition of energy down to very low temperatures. Another possible concern is that zero-point motion of the NTE modes is sufficient to allow a bond-bending deformation mechanism even in the limit $T \to 0$ K. Using harmonic lattice dynamics calculations performed using the GULP code, 26 we have estimated the zero-point contribution to B_0' at 0 K to be $\simeq -0.1$ (as shown in Appendix B), and hence a negligible correction to the values given in Fig. 4. For these two reasons, we do not anticipate much discrepancy between the classical MD results and the true quantum picture even at relatively low temperatures.

IV. NEUTRON POWDER DIFFRACTION

A. Method

In order to confirm the predicted variation of B_0 and B_0' with temperature, we carried out a neutron powder diffraction experiment using the GEM diffractometer at ISIS.²⁷ A polycrystalline $Zn(CN)_2$ sample was contained within a Ti-Zr alloy pressure cell, which produces a featureless background in the diffraction pattern; this was itself contained within a closed-cycle refrigerator, allowing us to control both temperature and pressure. Hydrostatic pressure applied to the sample was generated by an external pressure intensifier unit to an accuracy of 1 bar. The sample temperature was controlled by a closed-cycle refrigerator (CCR) that can operate within a temperature range of 4 to 325 K. Diffraction data were

collected at various pressures from 5 bar to a value limited by the phase diagram of helium gas at temperatures of 30, 50, 75, 150, 200, 250, and 300 K.²³ Measurements at lower temperatures could not be carried out because the helium gas would readily have liquefied at low pressure, making the true sample pressure unknown. For the same reasons, the accessible pressure range is much lower for the very low-temperature data points. This limitation has the unfortunate consequence of increasing the experimental uncertainty in the derived values of B_0 and B_0' for temperatures below 100 K.

-0.027(2)

-0.035(23)

Rietveld refinement of the neutron diffraction patterns using GSAS (Ref. 28) and EXPGUI (Ref. 29) yielded satisfactory fits, as shown in Fig. 5, and enabled determination of the unit-cell volume to an accuracy of 0.001 Å³. The structural model used was that described in Ref. 14, which takes into account the head-to-tail disorder of the cyanide ions. The

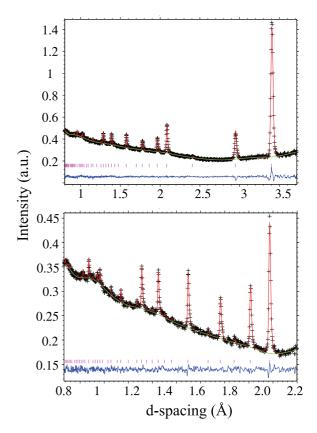


FIG. 5. (Color online) A representative Rietveld fit to the neutron diffraction data collected using the GEM instrument at ISIS. These data correspond to a sample temperature of 300 K and a hydrostatic pressure of 1.2 kbar. Data points are shown as black crosses, the fit as a red line, the background function as a green line, and the residual (data fit) as a blue line.

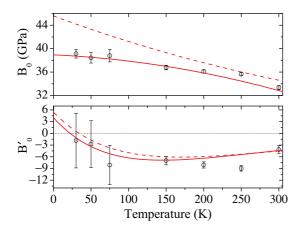


FIG. 6. (Color online) Bulk modulus at zero pressure B_0 and its first derivative B' as functions of temperature, obtained from fitting the third-order Birch-Murnaghan equation of state to the experimental isotherms of $\operatorname{Zn}(\operatorname{CN})_2$. Solid curves show the calculated $B_0(T)$ and $B'_0(T)$ using values of the parameters from the simultaneous fit, where $B_0(T)$ was described by a quadratic polynomial and $B'_0(T)$ was expressed by Eq. (1). The relatively large deviation of the B'_0 curve from the data points beyond 150 K is consistent with the relatively large fitting error for the corresponding isotherms in Fig. 3(b). Dashed curves are the results from the MD for comparison (also seen in Table I).

refined structural parameters were the position of the C/N atoms (constrained by the $Pn\bar{3}m$ crystal symmetry to a single parameter), the anisotropic atomic displacement parameters for the C/N atoms (constrained to be the same for C and N), and the Zn isotropic atomic displacement parameter. A table of refined structural parameters for all data sets is given in the Supplemental Material.²³

B. Equation of state from experimental data

Values of B_0 and B_0' were extracted from the experimental lattice parameter data using the same approach as in the analysis of the MD results, fitting each isotherm to a third-order Birch-Murnaghan equation of states. The corresponding thermal evolution of derived values of B_0 and B' is shown in Fig. 6, and the numerical values of B_0 and B'_0 are listed in Table I for comparison with the results from the MD simulations. We consider that the agreement between the experimental and MD values is reasonable given the difficulties in the experiment and the fact that the intermolecular potential was not tuned against experimental data. We will comment more on the level of agreement below.

We further carried out a simultaneous fit to all the p-V data sets as we did for the MD data. The quality of the fitting is shown in Figs. 3(b) and 6. The values of β , T_0 , and T_1 in Eq. (1) obtained from this simultaneous fit are 4(5), 22(14) K, and 145(29) K, respectively. These are in reasonable agreement with the values obtained by fitting to the MD data (Fig. 4), albeit with rather larger values of the standard deviations on each parameter. The consistency between the MD and neutron diffraction results is highlighted by comparing the fitted curves for B_0 and $B_0'(T)$ in Fig. 6. In the case of B_0 , agreement is within 10%, although the curvature differs at low temperature. This may be a systematic error coming from the fact that at low

temperature we had access to a much more restricted range of pressures. On the other hand, the agreement between the MD and neutron diffraction results for B'_0 is much closer.

Values of α and $\partial B_0/\partial T$ at 300 K extracted from our data are given in Table II, together with the value of $\partial \alpha / \partial p$ obtained from Eq. (2). The value of α we have determined from experiment is consistent with the results reported in other experiments.^{3,11} Altogether, the experimental values given in Table II are in fair agreement with the results from the MD simulations, when account is taken of the relatively large errors for the experimental data. We note that the value of α from experiment agrees well with the values from MD in the temperature range of 10–100 K, but the value of α from the MD simulations decreases faster on heating than in the experiment, suggesting an overestimate of anharmonicity in our model at high temperatures. Despite the different low-temperature curvatures of the fitted $B_0(T)$, the MD values of $\partial B_0/\partial T$ and $\partial \alpha / \partial p$ at 300 K agree to the experiment within the error as shown in Table II.

While we have not been able to measure diffraction data at sufficiently low temperatures to observe a definitive transition from negative to positive values of B_0' , so that Fig. 6 can not be said to confirm the detailed MD results exactly, the agreement between the fitted B_0' curve and the data points nevertheless demonstrates that our data remain consistent with the phenomenological model embodied by Eq. (1).

V. CONCLUSIONS

We have proposed a phenomenological model of pressure-induced softening in zinc cyanide, based on the well-established difference in energy between vibrations that involve sideways buckling of the cyanide ions and those that involve bond stretching. The functional form of Eq. (1), chosen to illustrate this qualitative model, is consistent with both experimental and simulated data.

In this model, the parameter T_0 represents the temperature above which B_0' becomes negative, and as such it is a measure of the temperature at which the structure starts to crumble. In other words, it corresponds to the temperature at which the vibrational modes responsible for negative thermal expansion begin to be substantially occupied. Encouragingly, these modes are known to have an energy less than 1 THz = $k_{\rm B} \times 48~{\rm K}$, $^{16-19}$ which agrees with the values of T_0 obtained from fits to the experimental and MD data.

The importance of revealing the temperature dependence of B_0' transcends the specific case of $Zn(CN)_2$, as important as this particular material is. We have suggested elsewhere that a negative value of B_0' is a common feature of NTE materials, having shown through extensive simulations of many cubic zeolites that almost all exhibit both NTE and pressure-induced softening effects. Indeed, the establishment here of a temperature dependence in which the parameter changes sign on cooling is crucial for the interpretation of simulation studies. We envisage that the striking prediction of negative values of B_0' in many NTE materials will stimulate a growing number of *ab initio* studies, but we caution that without accounting for thermal fluctuations, such calculations may predict incorrectly a positive value of B_0' .

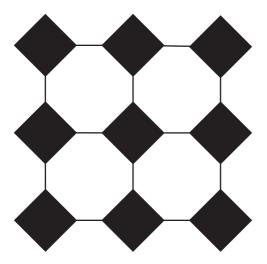


FIG. 7. Representation of the two-dimensional model presented in Ref. 2, where we represented the square containing nonbridging vertex bonds of the original model by a rod that could represent the CN molecular ion of Zn(CN)₂. The model of Ref. 2 has a single rotational variable for each square and, by extension, to each rod.

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APPENDIX A: SOME COMMENTS ON THE SINGLE-PARTICLE HAMILTONIAN OF REF. 2

Reference 2 presents an experimental study of the pressure dependence of the elasticity of ZrW_2O_8 , showing pressure-induced softening. A simple model initially presented in Ref. 30 was used to provide an interpretation. The model is a two-dimensional array of corner-sharing squares, with one quarter of the squares missing, giving a structure with three polyhedra per unit cell of which two have have their bonds not connected to another square. This could be said to represent the existence of the nonbridging W–O bonds in ZrW_2O_8 . However, the squares with two nonbridging bonds could be replaced by a rod connecting the fully connected squares with no changes to the model, as developed in Ref. 9 and shown in Fig. 7. As such, the model is more accurately a two-dimensional representation of $Zn(CN)_2$, albeit with some significant differences. 31

In the analysis of Ref. 2, a model Hamiltonian was constructed. This involves a single variable per square, namely, the rotation angle θ_i , and involves two energy terms. The first

is a pV-like term³²

$$E_{pV} = -p\left(1 - \cos\theta_i\right),\tag{A1}$$

where the volume V is reduced by rotations of the squares. This term automatically lowers the energy for rotations of the square since these lead to a reduction in the volume. The second term is an anharmonic restoring force in even powers of θ_i ,

$$E_{\text{rot}} = \frac{1}{2}K\theta_i^2 + \frac{1}{4}\gamma\theta_i^4 + \frac{1}{6}\delta\theta_i^6 + \cdots$$
 (A2)

and gives a rise in energy for rotations of the squares.

Because there is no coupling between the squares, the phonons will all have frequencies that are independent of wave vector. This is a reasonable approximation for a system in which all phonons are rigid unit modes.³³ However, in the case of Zn(CN)₂ there is a significant coupling term that gives a dependence of frequency on wave vector, which in the two-dimensional model would mean that counter-rotations of nearest-neighbor squares (modes whose wave vectors are at the edge of the two-dimensional Brillouin zone) will have a much lower frequency than the zone center modes where all squares rotate in the same direction.

At temperature T=0 there are two states differentiated by the value of p, namely, the state at low p in which the energy is dominated by $E_{\rm rot}$ leading to the preference for $\theta_i=0$ for all values of i, and the state for p above a critical value in which the energy is dominated by E_{pV} leading to nonzero values of θ_i and a lowering of the volume in equilibrium. The transition between these two states is increasingly blurred at high temperature. The variation of volume around the critical pressure gives a variation in the bulk modulus that changes with pressure. This is not a real phase transition because the Hamiltonian does not contain terms that couple the rotations of neighboring squares.

From Fig. 5 in Ref. 2 we can see (although this is not articulated in Ref. 2) that this model predicts $B'_0 = 0$ at T = 0, with a value of B'_0 that becomes negative and increasingly so with higher temperature. This behavior at p = 0 arises because of the blurring of the transition at higher temperature. Where the model differs from the result presented in this paper is that we observe $B'_0 > 0$ at low T, only becoming negative for temperatures above a particular temperature. This difference arises from the neglect in Ref. 2 of the stiffness of the squares except in an unspecified post hoc manner. Thus, to become consistent with this study, the model should be extended by allowing an explicit noninfinite stiffness of the squares. This would give a finite compressibility at zero temperature, leading to a positive value of B'_0 . We have shown separately (to be submitted) that the model described by Fig. 7 with a finite compressibility of the squares does indeed lead to a variation of B'_0 with temperature that closely follows the phenomenological form of Eq. (1).

APPENDIX B: EFFECT OF ZERO-POINT MOTION

Here, we calculate the effect of zero-point energy on the value of B_0' , which was not included in the lattice dynamics or the MD simulations that gave a positive value of B_0' at T=0. Thermodynamic calculations show that the contribution from the zero-point energy term to the value of B_0' of $Zn(CN)_2$ is a negligibly small negative number.

To derive B'_0 , we start from the pressure of an insulating crystal³⁴

$$p = -\frac{\partial \Phi}{\partial V} + \sum_{s} \left(\frac{\hbar \omega_s}{V} \gamma_s n_s \right), \tag{B1}$$

where Φ is the lattice energy of the crystal at zero temperature. The sum is over all the phonon modes $s = j, \mathbf{k}$ in the system with the angular frequency ω_s and Grüneisen parameter γ_s . The average phonon occupation number of each mode is

$$n_s = \frac{1}{\exp(\hbar\omega/\tau) - 1} + \frac{1}{2}$$
 (B2)

with $\tau = k_{\rm B}T$ the temperature in units of energy. At zero temperature, $n_s = \frac{1}{2}$ and

$$p|_{\tau=0} = -\frac{\partial \Phi}{\partial V} + \frac{1}{2} \sum_{s} \frac{\hbar \omega_{s}}{V} \gamma_{s} = -\frac{\partial \Phi}{\partial V} + \pi_{0},$$
 (B3)

where

$$\pi_0 = \frac{1}{2} \sum_s \frac{\hbar \omega_s}{V} \gamma_s \tag{B4}$$

is the contribution from zero-point energy. According to thermodynamic relations, the bulk modulus at zero temperature is

$$B|_{\tau=0} = -V \frac{\partial p|_{\tau=0}}{\partial V} = V \frac{\partial^2 \Phi}{\partial V^2} - \frac{\partial \pi_0}{\partial \ln V}$$
 (B5)

and the first derivative of bulk modulus at zero temperature is

$$B'|_{\tau=0} = \frac{\partial B|_{\tau=0}}{\partial p}$$

$$= -\frac{V}{B|_{\tau=0}} \left[\frac{\partial^2 \Phi}{\partial V^2} + V \frac{\partial^3 \Phi}{\partial V^3} \right] + \frac{V}{B|_{\tau=0}} \frac{\partial}{\partial V} \left(\frac{\partial \pi_0}{\partial \ln V} \right)$$

$$= B' \Big|_{\tau=0}^{LD} + B' \Big|_{\tau=0}^{ZP}$$
 (B6)

with V the volume of the crystal. In Eq. (B6),

$$B'\big|_{\tau=0}^{LD} = -\frac{V}{B\big|_{\tau=0}} \left[\frac{\partial^2 \Phi}{\partial V^2} + V \frac{\partial^3 \Phi}{\partial V^3} \right]. \tag{B7}$$

TABLE III. The values of cell volume and π_0 [Eq. (B4)] calculated at different pressures using the Zn(CN)₂ potential model (Ref. 19). These values were further used to calculate $B'|_{\tau=0}^{ZP}$ using Eqs. (B7) and (B8).

P (GPa)	V (Å ³)	$\pi_0(\times 10^{-3})$ (GPa)	$\partial \pi_0 / \partial \ln V$ (GPa)	$B' _{\tau=0}^{ZP}$
0.0	207.2201	2.0	0.12	-0.13
0.2	206.3314	1.5	0.14	
0.4	205.4938	0.92		

This term can be obtained from a harmonic lattice dynamics calculation. Here,

$$B'\Big|_{\tau=0}^{ZP} = \frac{V}{B|_{\tau=0}} \frac{\partial}{\partial V} \left(\frac{\partial \pi_0}{\partial \ln V} \right)$$
 (B8)

is the contribution from the zero-point energy. Using Eq. (B4), we estimated this term with the phonon frequencies at different volumes using our $Zn(CN)_2$ potential model. ¹⁹ The derivatives were approximated by

$$\frac{\partial \pi_0}{\partial \ln V} \approx V_1 \frac{\pi_0 (V_2) - \pi_0 (V_1)}{V_2 - V_1}$$
 (B9)

and

$$\frac{\partial}{\partial V} \left(\frac{\partial \pi_0}{\partial \ln V} \right)$$

$$\approx V_2 \frac{\pi_0 (V_3) - \pi_0 (V_2)}{(V_3 - V_2) (V_2 - V_1)} - V_1 \frac{\pi_0 (V_2) - \pi_0 (V_1)}{(V_2 - V_1)^2},$$
(B10)

where $V_3 < V_2 < V_1$ are equilibrium volumes at different pressures, and V_1 corresponds to T=0 and p=0. Table III lists the data from the lattice dynamics of the potential model. The calculated $B'|_{\tau=0}^{ZP}$ using these data is -0.13. Thus, for $\operatorname{Zn}(\operatorname{CN})_2$, the contribution from the zero-point energy to the value of B'_0 is negative, and the obtained values from the lattice dynamics and MD in the paper should be corrected accordingly, although this correction is too small to have any significant effect on the results.

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