

Disentangling coordination and alloy effects in transition-metal nanoalloys from their electronic structure

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Accurate prediction of local properties of transition-metal nanoalloys from the electronic structure is a challenge for building new materials with novel properties in a controlled way. To this aim, developing unified descriptions of local electronic states as a function of a minimal set of parameters is the way to disentangle structural and chemical effects. This is achieved here within *sp-d* tight-binding calculations using a self-consistent procedure taking into account both the changes in the structural environment (coordination effect) and in the chemical one (alloying effect). From these calculations, trends in the distributions of energy electronic states are obtained through band shifts and widths allowing one to study in a systematic way ordering tendency and local properties in nanoalloys in a wide range of sizes and structural complexity.

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I. INTRODUCTION

Owing to the recent development of nanoalloys,^{1,2} within application fields like metallurgy, catalysis, magnetism, and optics, a further challenge is to model alloying phenomena in nanosystems from the knowledge of the electronic structure of their pure constituents. This requires that the subtle coupling between electronic, structural, and chemical effects is described and this understanding is used to build new structures with new properties. To this aim, many approaches have been developed, based upon a more or less sophisticated description of the electronic structure. Among them, *ab initio* DFT (density functional theory) -type calculations,³⁻⁷ even the most performing,⁸ are much too time-consuming to allow systematic studies, whereas semiempirical potentials⁹⁻¹⁴ are too crude to establish accurately the link between the electronic, atomic, and chemical structures. In this context, the tight-binding (TB) approximation,^{15,16} which allows handling the electronic structure with a flexible accuracy depending on the addressed problem, is an intermediate suitable method for capturing the essential physics and will be used in the present work.

sp-d TB calculations are performed in this paper by using a self-consistent procedure taking into account both the changes in structural environment (coordination effect) and in the chemical one (alloying effect). The system chosen to illustrate the method is CoPt, which presents a double interest. First it is archetypal of the class of ordering systems having a standard *d*-band shift behavior under alloying (decrease of diagonal disorder), as shown from x-ray photoemission spectroscopy

(XPS) measurements¹⁷ and calculations.¹⁸ In addition, it is widely investigated for both potential applications in magnetic storage media and catalysis.

In this context the following work is dedicated to the determination of the two pertinent parameters characterizing order in an *AB* nanoalloy, namely the *diagonal disorder* parameter $\delta_{d,0} = \varepsilon_d^A - \varepsilon_d^B$ (the difference between gravity centers of the *d* bands, ε_d) and *off-diagonal disorder* parameter $\delta_{nd} = W_d^A - W_d^B$ (the difference between *d*-band widths, W_d). These values have been shown to allow deriving ordering tendencies in all bulk transition-metal alloys.¹⁹ We intend here in addition to derive a law for the evolution of these parameters as a function of the size of clusters and to provide a tool for predicting ordering tendency in nanoalloys as a function of size, which is an essential step for scientists willing to control the atomic arrangements in the fabrication of specific clusters for targeted properties. In a more applied frame, data like ε_d should provide guidance to the interpretation of core level shift measurements using x-ray photoemission spectroscopy (XPS) in nanoalloys, a widely used technique in the field of catalysis, for instance.

II. TIGHT-BINDING METHODOLOGY: FROM ALLOYS TOWARDS NANOALLOYS

For an alloy $A_c B_{1-c}$, the chemical configuration is defined from the set of site occupation factors p_i^a such as $p_i^a = 1$ if site *i* is occupied by an atom of type *a* ($a = A, B$) and $p_i^a = 0$ otherwise. Then the corresponding Hamiltonian is written in

the basis of atomic orbitals λ at sites i , $|i, \lambda\rangle$:

$$H(p_i^a) = \sum_{a=A,B} \sum_{i,\lambda} p_i^a |i, \lambda\rangle \varepsilon_{i\lambda}^a \langle i, \lambda| + \sum_{a,b=A,B} \sum_{i,j \neq i,\lambda, \mu \neq \lambda} p_i^a p_j^b |j, \mu\rangle (\beta_{ij}^{\lambda,\mu})^{ab} \langle i, \lambda|, \quad (1)$$

which involves two types of parameters, the effective atomic levels $\varepsilon_{i\lambda}^a$ and the hopping integrals $(\beta_{ij}^{\lambda,\mu})^{ab}$. In this framework ε_d^a is the atomic d orbital level for an atom of the type a ($a = A, B$) in its own bulk and the hopping integral between two d orbitals on neighboring sites drives the d -band width (W_d^a).²⁰ The first term gives rise to the so-called *diagonal disorder* effect coming from the difference $\delta_{d,0} = \varepsilon_d^A - \varepsilon_d^B$, whereas the second one accounts for the possible effect of *off-diagonal disorder* due to the difference in d -band width: $\delta_{nd} = W_d^A - W_d^B$. These two parameters $\delta_{d,0}$ and δ_{nd} actually drive the redistribution of the electronic states with respect to those of pure elements and therefore both the new properties of the (nano)alloy and its preference for ordering or phase separation at low temperature. Based on these parameters, well known from compilations in the literature,²¹ systematic studies within the TB approach were previously carried out for deriving ordering tendencies in all bulk transition-metal alloys¹⁹ and general trends in d -band and core level shifts,¹⁸ the latter being tightly related, e.g., to catalytic properties. The next step of the present work is to combine site coordination effects (atomic structure) and alloying effects (chemical structure) in a single model for nanoalloys in the experimental size range (up to thousands of atoms). The essential quantity to study is the local density of states (LDOS), $n(E)$. In all cases $sp-d$ hybridization is taken into account by using the basis of atomic orbitals $\lambda(s, p, d)$. Each partial LDOS $n_{i\lambda}(E)$ is obtained from the continued fraction expansion of the Green function $G(E) = (E - H)^{-1}$,^{20,22,23} whose coefficients are directly related to the moments of the density of states. These coefficients are calculated within the recursion method²⁴ implemented with a self-consistent treatment of charge transfer induced by both coordination changes and alloying effects. The technique makes use of a local charge neutrality rule *per site, per orbital, and per chemical species*, well known for surface effects and recently extended to alloys from DFT calculations²⁵ and further generalized for describing band shifts in bulk transition-metal alloys.¹⁸ Although of course some deviations from this neutrality rule shall be evidenced in some cases, also in experiments,^{26–28} our choice was to join a single law in order to bring the overall behavior of transition-metal alloys under a single description which was indeed shown to be sufficient to describe d -band shifts and obtain a good agreement with photoemission experiments.¹⁸ Here we propose a new application of this TB approach for nanoalloys, where both alloying and structural effects are included in the same procedure. In practice we consider ten pairs of exact coefficients in order to obtain sufficiently detailed LDOS. The main difficulty is then to determine the effective atomic level $\varepsilon_{i\lambda}$ for each inequivalent site i , while ensuring the charge self-consistency on this site. This requires, after indexing all inequivalent sites i and species a , that these levels are shifted for each orbital λ with respect to those in the bulk by a value $\delta\varepsilon_{i\lambda}^a$ as

follows:

$$\varepsilon_{i\lambda}^a = \varepsilon_{0\lambda}^a + \delta\varepsilon_{i\lambda}^a, \quad (2)$$

in order to satisfy a given rule (here the local neutrality rule) on the different band fillings per orbital and per species $N_{i\lambda}^a$, which are obtained by integration of the partial local densities of states up to the Fermi level E_F :

$$N_{i\lambda}^a = \int_{-\infty}^{E_F} n_{i\lambda}^a(E) dE. \quad (3)$$

The total band filling (number of valence electrons) at site i , occupied by an atom of type a , is then determined by summing over all orbitals such as $N_i^a = \sum_{\lambda} N_{i\lambda}^a$.

The remaining parameters of the model are the hopping integrals between orbitals of neighboring sites, $(\beta_{ij}^{\lambda,\mu})^{ab}$. If $a = b$ the values are directly taken from the compilation of Papaconstantopoulos²¹ designed for elemental solids. For the specific case of Co, the parameters for the structural fcc paramagnetic phase were applied, in agreement with the fcc structure of the clusters considered hereafter. If $a \neq b$ the arithmetic average of $\beta^{\lambda,a}$ and $\beta^{\lambda,b}$ is used. Magnetism was not explicitly included in the TB calculations but some considerations about it will be given in the last section of this work.

The validity of this TB method must be supported by more fundamental *ab initio* calculations on the basic bulk phase alloys since the description of ordering behavior of complex nanoalloys is necessarily related to the bulk phase diagrams. We shall thereby verify the relative stability of the usual bulk alloy phases against possible competing ones. On the other hand, the method must be proved to provide reliable local densities of states on which all interpretations in terms of local properties (as, for instance, band shifts for catalysis or density at Fermi level for magnetism) are related. For both purposes *ab initio* calculations within density functional theory (DFT) were performed using the SIESTA method.²⁹ SIESTA is based on the standard Kohn-Sham self-consistent DFT, which can be used either in the local-density approximation (LDA) or in the generalized gradient approximation (GGA) for the electron-electron exchange and correlation interactions. The one-particle problem is solved using a linear combination of atomic orbitals (LCAO) and taking as a solution method the diagonalization of the Hamiltonian. In the presented calculations we have used the generalized gradient approximation (GGA) of PBE³⁰ for the exchange and correlation potential. For the magnetic systems we performed spin-polarized calculations. For the ion-electron interactions, the core electrons are replaced by norm-conserving pseudopotentials.³¹ Valence states are described using numerical atomic orbitals (NAO) and double- ζ polarized basis sets. The bulk lattice parameter was relaxed by a variable cell procedure within a conjugate gradient minimization. The low temperature experimental phase diagram of CoPt is characterized by three main ordered phases,³² the fcc $L1_2$ ($\text{Co}_{0.25}\text{Pt}_{0.75}$ and $\text{Co}_{0.75}\text{Pt}_{0.25}$) and the fct (tetragonalized with regards to fcc) $L1_0$ phase at equiconcentration. However, while considering in this work only the fcc crystalline structure, two ordered phases have to be considered at $c = 0.25$ concentration, in what is concerned

TABLE I. Relative stability energy ΔE_{stab} (eV/atom) between several bulk CoPt ordered alloys at different concentrations. DFT calculations were performed with (magnetic) and without (nonmagnetic) spin polarization.

Alloy	Co _{0.75} Pt _{0.25}	Co _{0.5} Pt _{0.5}	Co _{0.25} Pt _{0.75}
ΔE_{stab}	$E_{L1_2} - E_{DO_{22}}$	$E_{L1_0} - E_{B_2}$	$E_{L1_2} - E_{DO_{22}}$
Values in (eV/atom)			
DFT (magnetic)	0.001	-0.247	-0.007
DFT (nonmagnetic)	0.003	-0.337	-0.012
TB (nonmagnetic)	0.028	-2.635	-0.128

relative stability, namely $L1_2$ (Cu₃Au type) and also DO_{22} (Al₃Ti type). For $c = 0.5$ if $L1_0$ has to be the most stable it should be checked that the TB model well predicts this stability against other phases like the bcc B_2 one. In this context we have calculated the relative stability energy ΔE_{stab} of the different ordered structures, at $c = 0.25$ and $c = 0.5$, both with DFT and TB methods. Within the latter only band energy, i.e., the explicit term issuing from the electronic structure, has been considered. The results are presented in Table I.

There is a full consistency between DFT and TB concerning the sign of the obtained values of ΔE_{stab} and the hierarchy of the obtained relative energies. Absolute TB values of ΔE_{stab} are however different by about one order of magnitude than DFT ones. More surprising is the apparent degeneracy between the $L1_2$ and DO_{22} ordered structures obtained both in DFT and TB, although the experimental phase diagram shows only the $L1_2$ structure around $c = 0.25$ (or $c = 0.75$). However, it should be kept in mind that the structural difference between the $L1_2$ and DO_{22} consists only of the insertion of an antiphase boundary between two fcc unit cells leading to a periodic ordering on two fcc cells instead of one, making the two structures rather similar. Therefore, the related energies of these two phases should not differ too much. This is deduced from a description of their density of states in terms of their moments, the DOS of the $L1_2$ and DO_{22} structures having six identical moments.³³ In the same sense, recent almost exhaustive calculations of bulk Co-Pt structures by Chepulskii and Curtarolo resulted in very close values for Co₃Pt $L1_2$ and DO_{22} phases.³⁴

Then the reliability of the method for nanoalloys has been tested by making several DFT calculations on small clusters of different shapes and compositions and comparing the projected DOS with the LDOS obtained using the TB calculations. A good agreement is generally obtained, in this way validating the procedure of self-consistency based on a local neutrality rule per site, orbital, and species. As an illustration we present in Fig. 1 the result of such a conclusive comparison for a 147 atom cluster showing a $L1_0$ order type with alternating Co and Pt (100) planes (see inset in Fig. 1).

Altogether, these comparisons establish well the reliability of our self-consistent $sp-d$ TB model in determining accurately the distribution of electronic states in nanoalloys and to a reasonable extent its reliability to predict the relative stability between structures of different chemical arrangements.

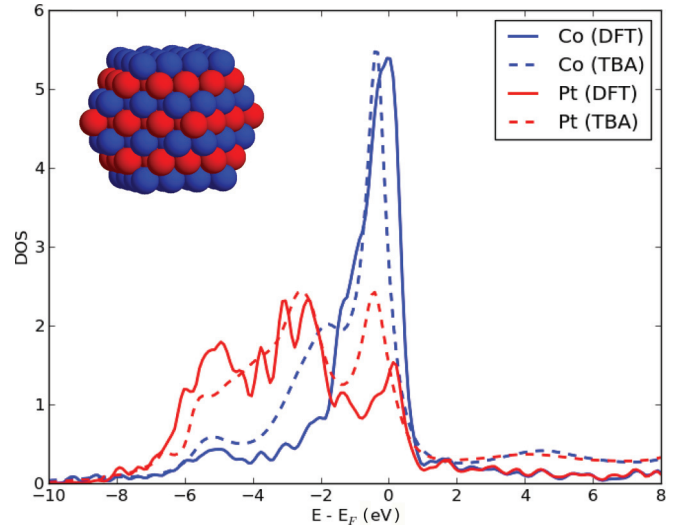


FIG. 1. (Color online) Averaged DOS ($sp-d$, nonmagnetic) in a cuboctahedral CoPt $L1_0$ -like cluster of 147 atoms. Solid lines correspond to DFT calculations and dashed ones to TB calculations. The inset shows the corresponding atomic structure. Color code for both curves and structure in inset is blue for Co and red for Pt.

III. APPLICATION OF THE SELF-CONSISTENT $sp-d$ TIGHT-BINDING METHOD TO COBALT-PLATINUM NANOALLOYS

The usual shape for fcc clusters of sufficiently big size being the cuboctahedral one, we will build a model nanoalloy as a cuboctahedral piece of a bulk fcc $L1_0$ structure. Such a system can be viewed as n shell cuboctahedron presenting at its surface a sufficiently large set of inequivalent sites representative of realistic systems. For the same size of cluster, the use of either Co or Pt centered clusters leads to variable concentrations ($0.46 \leq c_a \leq 0.54$) and surface compositions. In order to differentiate chemical and geometrical effects, the knowledge of pure fcc Co and Pt cluster characteristics is obviously required. The resulting partial d -LDOS for all the inequivalent sites of both a pure Pt and a CoPt cluster, shown in Fig. 2, shows as expected an effective band width which decreases with the site coordination [from facets to edges and vertices; see inset of Fig. 2(a)] and is significantly modified near the Fermi level depending on the site. The values of the centers of gravity of the partial d bands for the different inequivalent sites, $\varepsilon_Z^{\text{Pt}}$ and $\varepsilon_Z^{\text{Co}}$, are plotted against their coordination numbers Z in Fig. 3. Decreasing the size of a pure cuboctahedron from 2869 to 147 atoms does not bring much change in ε_d since the shape and, therefore, the local structural environment around each site remains unchanged. Regarding still pure systems, let us notice also the linear variation of $\varepsilon_Z^{\text{Pt}}$ and $\varepsilon_Z^{\text{Co}}$ as a function of site coordination.

Comparing pure clusters to the mixed ones, the most remarkable feature is the evolution of the center of gravity of these bands, which can be described in terms of geometry and alloying while disentangling both effects in a straightforward manner. Indeed the Z variation of ε_d for the alloy clusters is also nearly linear, though shifted in a rigid way from that in pure clusters. Thus the behavior of the atomic d levels upon alloying in nanoclusters can be estimated by a linear variation

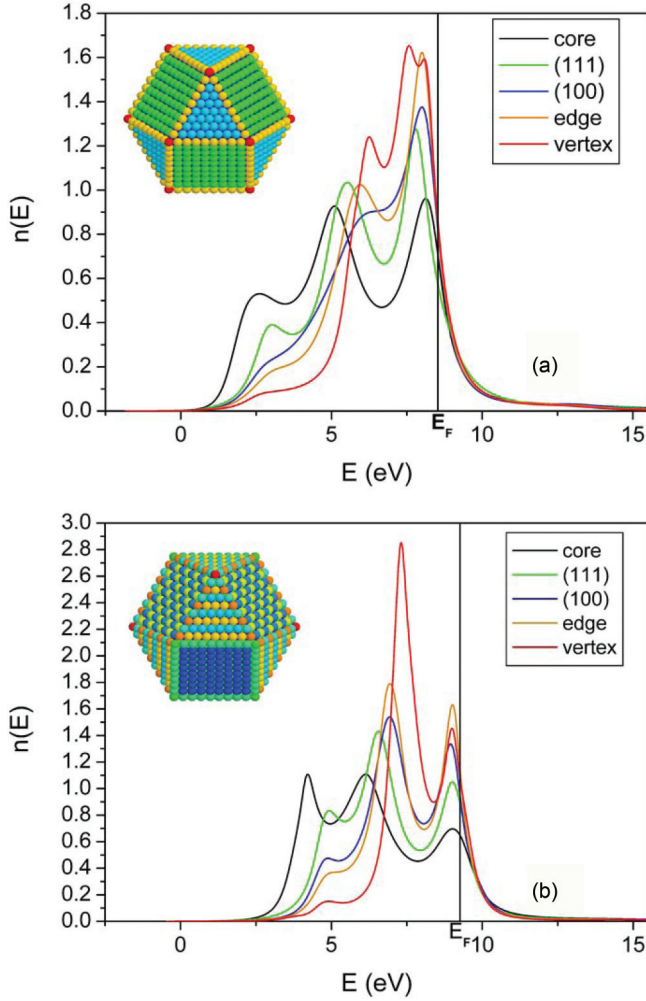


FIG. 2. (Color online) Partial d -LDOS on the geometrically inequivalent sites on a cuboctahedral Pt (a) and CoPt $L1_0$ -like (b) cluster of 2869 atoms. The inset shows the corresponding atomic structures, in which the sites have been colored upon their environment (coordination, chemical).

law as a function of coordination z to which an alloying term estimated by a rigid shift is added, which separates in an unambiguous way the structural and chemical effects. For $a = \text{Co, Pt}$ the atomic d level, ε_Z^a , is then written down as

$$\varepsilon_Z^a = \varepsilon_0^a + \Delta\varepsilon_{\text{alloy}}^a + \Delta\varepsilon_{\text{site}}^a(Z - 12). \quad (4)$$

The linear fit (see Fig. 3) leads to the following values of the constants: for the alloying term, $\Delta\varepsilon_{\text{alloy}}^{\text{Pt}} = +0.6 \pm 0.15$ eV and $\Delta\varepsilon_{\text{alloy}}^{\text{Co}} = -0.3 \pm 0.1$ eV, and for the geometrical term, $\Delta\varepsilon_{\text{site}}^{\text{Pt}} = -0.16$ eV and $\Delta\varepsilon_{\text{site}}^{\text{Co}} = -0.085$ eV.

A rather similar behavior is found in the case of the effective d -band width W_Z^a , defined as the centered second moment of the LDOS for Z -coordinated sites, scaled by the same constant which relates this second moment to the actual band width in the bulk. Thus a same linear equation can be derived just replacing ε^a by W^a , here with $\Delta W_{\text{alloy}}^{\text{Pt}} = -1.0 \pm 0.25$ eV, $\Delta W_{\text{alloy}}^{\text{Co}} = +0.6 \pm 0.1$ eV, $\Delta W_{\text{site}}^{\text{Pt}} = -0.4$ eV, and $\Delta W_{\text{site}}^{\text{Co}} = -0.22$ eV. Further approximation of the alloying terms by the values of the actual bulk alloy, which is justified in view of our

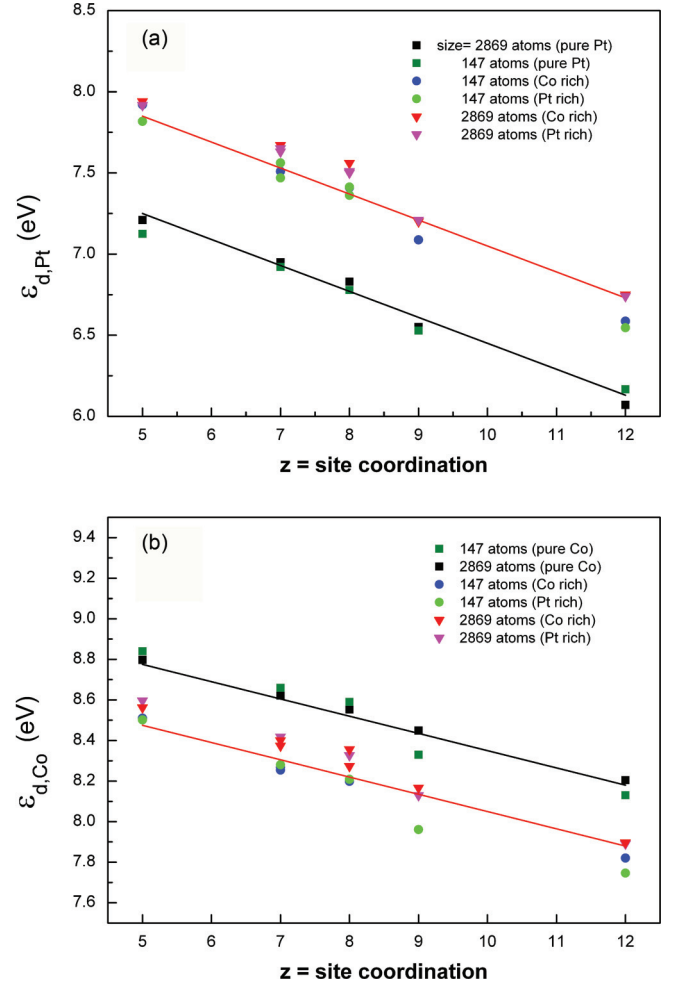


FIG. 3. (Color online) Center of gravity of partial energy d band for the geometrically inequivalent Pt (a) and Co (b) sites of cuboctahedral CoPt $L1_0$ -like clusters compared respectively to pure Pt and Co clusters. Two sizes are considered: $N_{\text{tot}} = 147$ and $N_{\text{tot}} = 2869$. Straight lines correspond to linear fits (see text). Label “Co rich” (“Pt rich”) refers to concentration in Co (Pt) > 0.5 .

results, should allow a generalization of these formulations to any bimetallic nanoalloy. Then, for an n shell cuboctahedron, one is able to write an analytic formula for the dependence of the average atomic d levels as a function of the cluster size:

$$\varepsilon_n^a = \varepsilon_0^a + \Delta\varepsilon_{\text{alloy}}^a + \Delta\varepsilon_{\text{site}}^a \sum_{Z=5,7,8,9} x_Z^a \frac{N_Z}{N_{\text{tot}}} (Z - 12), \quad (5)$$

where $N_{\text{tot}} = [1 + \frac{n}{3}(10n^2 + 15n + 11)]$ is the total number of atoms, and the summation over Z means summing over the different undercoordinated sites, the respective numbers of which are N_Z : $N_{Z=5} = 12$ vertices, $N_{Z=7} = 24(n-1)$ edge atoms, $N_{Z=8} = 6(n-1)^2$ atoms for the (100) facet, and $N_{Z=9} = 4(n-1)(n-2)$ atoms for the (111) facet. $x_Z^a = c_Z^a/c^a$ is the ratio between the partial a concentration at a site of coordination Z and c^a the global a concentration in the whole cluster. Here, again, a similar equation can be used to describe the size dependence of the effective d -band width by just replacing ε^a by W^a .

IV. ORDERING TRENDS IN NANOALLOYS

From the results of the previous section, we are now able to return to the question of the ordering tendency in nanoalloys, the ultimate goal being to be able to predict this tendency for any transition-metal nanoalloy as a function of its size. Following the previous work on bulk alloy,¹⁹ the ordering tendency for a cluster of order n will be linked to the differences $\delta_d^n = \varepsilon_n^{\text{Co}} - \varepsilon_n^{\text{Pt}}$ and $\delta_{nd}^n = W_n^{\text{Co}} - W_n^{\text{Pt}}$ obtained either by averaging all the Pt and Co LDOS in the cluster, respectively, or using the previously derived linear law. Averaging over all possible configurations in each case gives the law of variation for δ_d^n as a function of the size of the cluster, which is written analytically as

$$\delta_d^n = \delta_{d,0} + \Delta\varepsilon_{\text{alloy}}^{\text{Co}} - \Delta\varepsilon_{\text{alloy}}^{\text{Pt}} + \sum_{Z=5,7,8,9} (x_Z^{\text{Co}} \Delta\varepsilon_{\text{site}}^{\text{Co}} - x_Z^{\text{Pt}} \Delta\varepsilon_{\text{site}}^{\text{Pt}}) \frac{N_Z}{N_{\text{tot}}} (Z - 12), \quad (6)$$

This equation simplifies for a completely disordered alloy, or for a perfect $L1_0$ system if one averages Co and Pt centered clusters (as done here), since in these cases $x_Z^a = 1$. One is then able to draw a diagram of this analytical law and to compare the resulting curve with the values obtained by averaging respectively all the Pt and Co LDOS in ordered clusters of different sizes. This is plotted in Fig. 4, in which we can see that bulk values are definitely reached around 2000 atoms (or in terms of diameter as used in Fig. 4, $N^{\frac{1}{3}} \approx 12$). The presented analytical model is based on approximations implying that only d -band shifts of the surface atoms are modified and where only prefactors are fitted to local values. Therefore, an overall quantitative agreement with the values resulting from the LDOS averaging cannot be expected as it would be the case of an actual fit of these values. It is worth noticing that there is no basis to consider the increase of diagonal disorder parameter with the cluster size a general rule. Indeed the sign of the Z -dependent term in Eq. (6) is driven by the difference

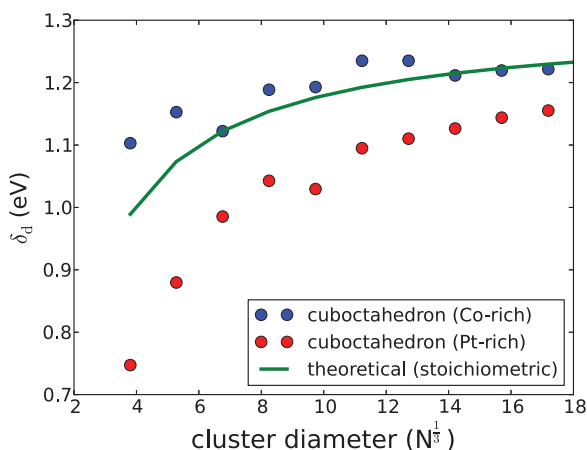


FIG. 4. (Color online) Variation of the *diagonal disorder* parameter in nanoalloys, δ_d^n , as a function of $N^{\frac{1}{3}}$ (representative of the diameter of a cluster containing N atoms). The continuous line represents the theoretical values given by Eq. (6). Red and blue circles represent values deduced from averaged LDOS respectively for the so-called Pt rich and Co rich ordered cuboctahedral clusters (see Fig. 3).

$\Delta\varepsilon_{\text{site}}^{\text{Co}} - \Delta\varepsilon_{\text{site}}^{\text{Pt}}$, which depends on the system under study, and can be deduced for any nanoalloy from the simple calculation of this dependence for both pure components. Finally, a similar equation can be used to describe the n dependence of the *off-diagonal disorder* parameter, replacing δ_d by δ_{nd} and ε^a by W^a , the variation of which will also depend on the nature of the alloy through the sign of $\Delta W_{\text{site}}^{\text{Co}} - \Delta W_{\text{site}}^{\text{Pt}}$.

Once the behavior of the two main parameters ($\delta_d^n, \delta_{nd}^n$) which drive the electronic structure of a nanoalloy of size n is known, let us show how the mapping previously derived to predict ordering tendency for bulk alloys¹⁹ can be extended to the case of nanoalloys. A relevant way to visualize the alloying effects in clusters vs bulk effects is to build a mapping of ordering and demixing domains in the parameter space spanned by δ_{nd} and d -band electronic filling N_e for different values of δ_d and for the concentration of interest, here $c = 0.5$. Results are presented in Fig. 5. Contrary to Ref. 19, here we focus on a single set of values corresponding to an average value of d -band width related to CoPt and equal to 6 eV. The most significant effect is that for $\delta_d = 2.0$ eV (black curve) only one large and centered domain appears, meaning in particular that the corresponding alloys with $N_e > 8$ and $N_e < 2$ should show strictly demixing tendency. However, reducing δ_d to at least 1.0 eV reveals new ordering domains at higher electronic filling allowing again order tendency for late transition-metal alloys. It remains to introduce the cluster points in this bulk map, by postulating that it can be done by just moving the bulk point according to the values of $\delta_{nd}^n, \delta_d^n$

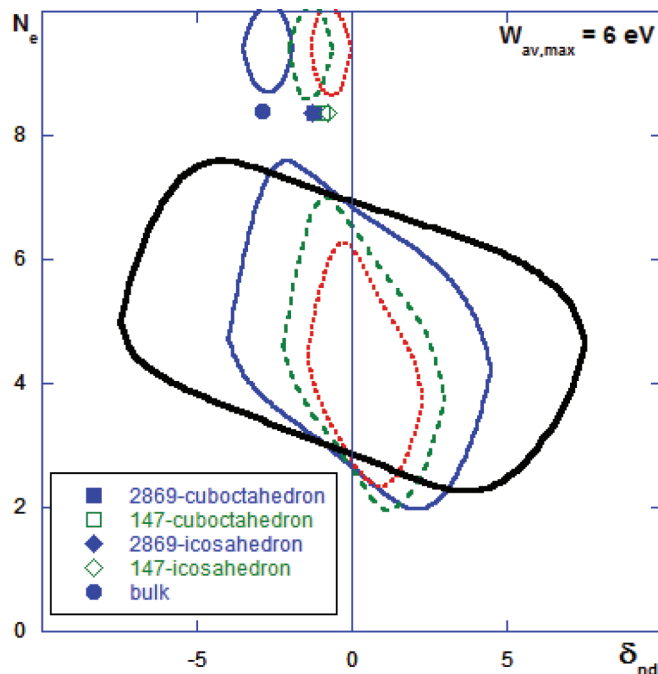


FIG. 5. (Color online) (N_e, δ_{nd}) ordering domains represented in the parameter space spanned by δ_{nd} and N_e for different values of δ_d and for alloys at equiconcentration. Ordering domains are delimited by black, blue, dotted green, and dotted red lines corresponding respectively to $\delta_d = 2.0, 1.0, 0.5,$ and 0.25 eV, the area outside these domains corresponding to demixing tendency. This plot is related to an average band width of pure constituents of $W_{\text{av,max}} = 6.0$ eV. The points correspond to bulk and 2869 and 147 atom clusters.

issued from the previous laws. In addition, consistent with the charge neutrality rule, these points will be always positioned at the electronic filling of $N_e = 8.5$ characteristic of CoPt at equiconcentration. In order to cover both possible size and morphological effects we have considered two limiting sizes (147 and 2869 atoms) and two morphologies adding icosahedra to the previously investigated cuboctahedra. As can be seen from the corresponding specific points in Fig. 5, all the systems fall at the frontier of the upper ordering domains.

A general behavior of the evolution of electronic structure in CoPt nanoalloys can be tentatively deduced from these results. Actually, a global evolution is obtained when going from bulk alloy to clusters with a large decrease of the absolute value of *off-diagonal disorder*. Further decrease of *diagonal disorder* is found when decreasing the size of the clusters from 2869 atoms to 147 atoms. Since icosahedral and cuboctahedral shapes are leading to similar results it can be concluded that the structure has a weak effect on the ordering tendency. It is also worth noticing that even though the points are not falling directly in the upper ordering domains their displacements as a function of size strictly follow the shift towards lower absolute values of δ_{nd} , indicating that order tendency does not change for any CoPt nanoalloy. The possible role of magnetism was not yet invoked although, as recently proposed by Karoui and co-workers,³⁵ in a spin polarized description of the CoPt LDOS, the majority spin up band is completely full and does not participate to the cohesion of the alloy so that an effective average electronic filling of 6.6 *d* electrons should be attributed to CoPt systems. In this case all the points of Fig. 5 would be shifted down into the larger ordering domains, meaning that once again the bulk ordering tendency is not only preserved but even enhanced in the cluster.

V. CONCLUSION

In summary, we have presented an accurate way to describe and predict the main local characteristics of nanoalloys, namely their mixing behavior and redistribution of energy electronic states from the single knowledge of the electronic structure of their pure constituents, using a self-consistent TB approach in the direct space. Local effects can be quantified through linear laws as a function of site coordination, by disentangling structural and chemical effects in a straightforward manner. Such a methodology is extendable to any other alloy, since it has been shown to give a unified description of local electronic structure both at surfaces of pure materials and in pure bulk alloys. Thus analytic laws have been derived which give the variation of both *d* atomic level and effective *d*-band width on one hand, *diagonal* and *off-diagonal disorder* parameters on the other hand, as a function of cluster size. Ordering trends in nanoalloys can then be estimated from bulk stability maps by just shifting the corresponding (δ_d, δ_{nd}) points according to these analytic laws. This method has no limit either in size or in structural asymmetric features, allowing the presence of surfaces, low coordinates sites, defects like vacancies, or stacking faults.

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