Adiabatic transformation as a search tool for new topological insulators: Distorted ternary Li2AgSb-class semiconductors and related compounds

Hsin Lin,¹ Tanmoy Das,² Yung Jui Wang,¹ L. A. Wray,^{3,4} S.-Y. Xu,³ M. Z. Hasan,³ and A. Bansil¹

¹*Department of Physics, Northeastern University, Boston, Massachusetts 02115, USA*

²*Theoretical Division, Los Alamos National Laboratory, Los Alamos, New Mexico 87545, USA*

³*Department of Physics, Joseph Henry Laboratories, Princeton University, Princeton, New Jersey 08544, USA*

⁴*Advanced Light Source, Lawrence Berkeley National Laboratory, Berkeley, California 94305, USA*

(Received 26 November 2012; published 12 March 2013)

We demonstrate that the first-principles based adiabatic continuation approach is a very powerful and efficient tool for constructing topological phase diagrams and locating nontrivial topological insulator materials. Using this technique, we predict that the ternary intermetallic series $Li_2M'X$, where $M' = Cu$, Ag, Au, or Cd and $X = Sb$, Bi, or Sn, hosts a number of topological insulators with remarkable functional variants and tunability. We also predict that several III-V semimetallic compounds are topologically nontrivial. We construct a topological phase diagram in the parameter space of the atomic numbers of atoms in $Li₂M'X$ compounds, which places a large number of topological materials presented in this work as well as in earlier studies within a single unified topological framework. Our results demonstrate the efficacy of adiabatic continuation as a useful tool for exploring topologically nontrivial alloying systems and for identifying new topological insulators even when the underlying lattice does not possess inversion symmetry, and the approaches based on parity analysis are not viable.

DOI: [10.1103/PhysRevB.87.121202](http://dx.doi.org/10.1103/PhysRevB.87.121202) PACS number(s): 71*.*20*.*Nr, 73*.*20*.*At

The recent discovery of topological insulators has opened up a new research direction in condensed-matter and materials science communities with tremendous potential for practical applications as well as for conceptual novel ties.^{[1–4](#page-4-0)} The unusual properties of topological insulators become manifest at their surfaces through the appearance of spin-polarized metallic states even though the bulk remains an insulator. These surface states have been predicted to host a large number of interesting topological phenomena. $4-10$ For instance, the electrodynamics of a topological insulator entails additional terms in Maxwell's equations yielding novel Faraday-Kerr effect, magnetoelectric effect, and axion physics.^{[5,6](#page-4-0)} Other exciting possibilities with topological surface states involve exciton condensation, $⁷$ pres-</sup> ence of an image magnetic monopole induced by an electric charge, and Majorana fermions induced by the proximity effect from a superconductor. $8-10$ In this way, topological insulators may be thought of as the materials that unify condensed matter with particle physics and cosmology.⁴

The practical realization of many of the aforementioned opportunities with topological insulators has not been possible to date because the choice of available topological materials is quite limited. The currently known topological insulators mainly belong to the Z_2 class, which includes the $Bi₂Te₃$ family, $11-13$ the ternary half-Heusler compounds, $14,15$ thallium-based chalcogenides,^{[16–18](#page-4-0)} $Ge_{m}Bi_{2n}Te_{m+3n}$ -class,^{[19](#page-4-0)} quaternary chalcogenide compounds, and ternary famatinite compounds.[20](#page-4-0) The search has also been extended to other classes of materials where many-body physics and (or) crystalline symmetries conspire to produce nontrivial topological phases. $21-25$ Despite this progress, however, the realization of wide classes of emergent topological properties has remained a challenge.

In connection with the materials discovery effort, adiabatic continuation is an elegant and powerful tool for obtaining a handle on the Z_2 time reversal invariant even when the lattice does not possess a center of inversion symmetry. The idea is to start with a known topological insulator and exploit the fact that if the Hamiltonian of this system can be transformed into that of the new system of interest through a continuous series of transformations without inducing a band inversion, then the new material will also be a topological insulator. In what follows, the adiabatic continuation approach will be seen to be not only a method for identifying the topological property of a material, but to also provide a pathway for finding new classes of topologically interesting materials. We specifically discuss how gray Sn, which is known to be topologically nontrivial, can be mutated to obtain new families of topologically nontrivial phases in zinc-blende-like materials and their superlattice variations. In particular, we predict that distorted- $Li₂AgSb$, a lightweight compound, is adiabatically connected to Sn, and thus harbors a 3D topological insulator state and that its cubic ground state lies near a critical point. In contrast, the related Li₂CuSb-type compounds are found to be topologically trivial. A topological phase diagram in the parameter space of the atomic numbers of the constituent elements in $Li₂AgSb$ structure is also adduced, which places a number of topological materials within a uniform topological framework. The ternary $Li₂M'X$ series (with a number of variants) is thus identified as a new platform for deriving a host of topological compounds, alloys, and nanoscale heterostructures via the flexibility of their lattice parameters and spin-orbit interaction strength. We also predict several III-V semimetallic compounds to be topologically nontrivial.

We start our discussion of band inversion in the electronic structure with reference to gray tin. 26 It is useful to understand first how a nontrivial topological phase emerges in Sn, whereas its isostructural counterparts such as silicon (Si) and germanium (Ge) are topologically trivial, see Fig. [1\(c\).](#page-1-0) As shown in Fig. [1\(d\),](#page-1-0) the *s*-like conduction bands (blue dots) lie well above the *p*-like valence bands in Si and Ge, indicating the

FIG. 1. (Color online) Crystal structure and topological band inversion. (a) Crystal structure of $Li₂AgSb$. Li, Ag, and Sb are denoted by green, red, and blue balls, respectively. Sb and Ag form the zinc-blende sublattice. (b) The zinc-blende structure of InSb. In and Sb are denoted by gold and blue balls, respectively. (c) The diamond structure of gray tin. Diagrams in (d), (e), and (f) illustrate band structures near the Γ point for a trivial insulator, a nontrivial semimetal, and a nontrivial insulator, respectively. Blue dots denote the s -like orbitals at Γ . Band inversion occurs in the nontrivial case where the s -like orbitals at the Γ point fall below the fourfold degenerate $j = 3/2$ states. The degeneracy of the $j = 3/2$ states is lifted by the lattice distortion in the noncubic case.

absence of band inversion in Si and Ge. As the lattice constant increases upon replacing Si or Ge with Sn, the conduction band moves down to touch the valence band, resulting in a band inversion at only the Γ point. This pushes the twofold degenerate *s*-like bands, which were originally lying in the conduction band in Si or Ge, below the valence bands of the fourfold degenerate *p* states of total angular momentum $j = 3/2$, and the Fermi level (E_F) goes through the *p* states as shown in Fig. $1(e)$. As the *s* states are now occupied at Γ point, the Z_2 topological invariant picks up an extra time-reversal factor of −1 compared to Si or Ge. Therefore Sn is a topologically nontrivial zero-gap semiconductor or semimetal in its pristine phase.^{[26](#page-4-0)}

If we define band-inversion strength (BIS) as the energy difference between the $j = 3/2$ states and the twofold degenerate *s* states at the Γ point, compounds with positive values of BIS will be topologically nontrivial, while those with negative values will be topologically trivial. The discussion in the preceding paragraph suggests a general route for realizing a topologically insulating phase where one lifts the fourfold degeneracy of the $j = 3/2$ states at the Γ point by breaking the cubic symmetry of the crystal without inducing a negative band-inversion strength. We explore this possibility by showing how the interplay of effects of finite distortion, heavy element substitution, and (or) formation of a superlattice of cubic structure can either remove the band inversion if the band-inversion strength is not sufficiently large, or conversely, it can enhance the band-inversion strength.

In carrying out adiabatic transformations, it is important to consider the crystal structure first. Here, we begin with the ternary compositions $Li_2M'X$ ($M' = Cu$, Ag, and Au, $X = Sb$

and Bi) whose crystal structure belongs to the space group $F\overline{4}3m$, with the atomic arrangement presented in Fig. 1(a). *M*- and *X* atoms occupy the Wyckoff 4*d* and 4*a* positions, respectively. Li atoms fill the remaining empty space in the Wyckoff 4*b* and 4*c* positions. Because *M*- and *X* atoms form a zinc-blende-type sublattice, these materials resemble InSb if Li atoms are removed [see Fig. $1(b)$]. Note that In and Sb precede and follow Sn in the periodic table and InSb resembles gray Sn with diamond structure as shown in Fig. 1(c). These observations suggest that $Li₂M'X$ compounds could be candidates for Z_2 topological insulators if their electronic band structures resemble the band structure of structurally similar Sn. As already noted above, since there is no spatial inversion symmetry in $Li₂M'X$ or in zinc-blend InSb, parity methods^{[26](#page-4-0)} of computing the Z_2 topological number are not viable.

First-principles band structure calculations were performed with the linear augmented-plane-wave (LAPW) method using the WIEN2K package²⁷ within the framework of the density functional theory (DFT). The generalized gradient approximation $(GGA)^{28}$ $(GGA)^{28}$ $(GGA)^{28}$ was used to describe the exchange-correlation potential. Spin-orbital coupling (SOC) was included as a second variational step using a basis of scalar-relativistic eigenfunctions.

The computed band structure of $Li₂AgSb$ along highsymmetry lines in the Brillouin zone is presented in Fig. [2\(a\).](#page-2-0) Away from the Γ point, the Fermi level is completely gapped and thus the topological properties can be determined from observations of band structure only near the Γ point. Focusing on the band structure very close to the Fermi level, we find that the orbital angular momentum symmetries of these compounds are identical to those of the low-energy spectrum in Figs. 1(d) and 1(e). The *s*- and *p*-type states are nearly degenerate at the Γ point around which the conduction bands and one of the valence bands have almost linear dispersion, indicating that the system is at the critical point of the topological phase [see Fig. $2(a)$]. With a small expansion of lattice, the *s*-*p* band inversion occurs as shown in Fig. [2\(b\).](#page-2-0) Conversely, the topological band inversion in $Li₂AgSb$ can be removed altogether by uniformly decreasing all lattice constants, demonstrating chemical tunability of the system.

Since $Li₂AgSb$ is at the critical point of the topological phase, it is possible to achieve a 3D topological phase by a finite distortion that lifts the degeneracy of the $j = 3/2$ states and simultaneously induces a band inversion. We find that a rhombohedral distortion with expansion along the hexagonal *ab* plane with $a = a_0 + 3\%a_0$ and $c = c_0$, where a_0, c_0 denote experimental lattice constants, does the job as shown in Fig. [2\(c\).](#page-2-0) The lattice expansion is seen to push the *s* states well below the *p* states, and to also open an insulating gap by lifting the fourfold degeneracy of $j = 3/2$ states through the loss of cubic symmetry. Since the band inversion occurs only at one point in the Brillouin zone, noncubic $Li₂AgSb$ is a 3D strong topological insulator. We return to this point below to show more rigorously that the aforementioned band inversion correctly indicates that the distorted noncubic $Li₂AgSb$ is topologically nontrivial.

Next, we apply the adiabatic continuity principle to a larger range of atomic compositions of $Li₂M'X$ by systematically changing the nuclear charge *Z* of the atoms. Let us assume

FIG. 2. (Color online) Electronic band structures of topological insulating state in Li₂AgSb with lattice distortion. (a) Li₂AgSb with experimental lattice constants. (b) Li₂AgSb with 3% expansion of lattice constants. (c) Rhombohedral Li₂AgSb with hexagonal lattice constants $a = a_0 + 3\% a_0$ and $c = c_0$, where a_0 and c_0 correspond to the experimental values. (d) A sketch of band structure near the *F* point for topologically nontrivial Li2AgSb with a lattice distortion. The *s*-like states are marked with a blue dot. Lattice distortion causes a gap to open at E_F , resulting in a topological insulating state in which surface bands will span the bulk band gap, resembling the dispersion plotted with red lines.

that atoms at the Li, Ag, and Sb positions possess hypothetical nuclear charges $Z_M = 3 - 0.5x + 0.5y$, $Z_{M'} = 47 + x$, and $Z_X = 51 - y$, respectively, where *x* and *y* are adjustable parameters. This choice guarantees that the system remains neutral for all values of *x* and *y*. Note that *x* and *y* need not be integers and that nonintegral nuclear charges are easily accommodated in first-principles all-electron computations in order to change the Hamiltonian continuously, while maintaining charge self-consistency throughout the transformation process. This mapping can start with $x = 0$ and $y = 0$, which corresponds to $Li₂AgSb$, and end with $x = 3$ and $y = 1$, which corresponds to the artificial compound He_2SnSn . He₂SnSn possess inversion symmetry and the wave-function parity analysis can be used to obtain the Z_2 topological invariant.²⁶ Since He is chemically inert, He_2SnSn has similar band structure as Sn and is a topologically nontrivial semimetal with $Z_2 = 1$.

A topological phase diagram can now be constructed based on the band inversion strength as a function of atomic numbers, as shown in Fig. 3 in the region of $0 \le x \le 3$ and $0 \le y \le 1$ at the lattice constant $a = 12.8$ Bohr, which is 3% longer than the experimental value for $Li₂AgSb$. A borderline is drawn at the zero of the band inversion strength (thick black line). A phase transition occurs when one crosses this borderline in that the two regions separated by the borderline are not topologically equivalent. $Li₂AgSb$ with lattice expansion and gray tin are seen to lie on the same side of the thick black line, and are thus adiabatically connected, implying that they are topologically equivalent. Since $Z_2 = 1$ for gray tin (actually, He₂SnSn with lattice expansion or distortion), $Li₂AgSb$ with lattice expansion [see Fig. $2(b)$] or distortion [see Fig. 2(c)] is also topologically nontrivial with $Z_2 = 1$. The borderline itself marks topologically critical phases. It would move toward positive *y* direction if the lattice constants decreased. Li₂AgSb with experimental lattice constants [see Fig. $2(a)$] lies very close to this critical line. Note that the variation of *Z*s can also be considered as a doping effect in the spirit of the virtual crystal approximations. $29-31$ Therefore our topological phase diagram in Fig. 3 could be used also to guide the search of topologically interesting electronic structures through alloying.

To further establish the usefulness of the adiabatic continuation approach in extending the search for topologically nontrivial materials, we carefully examine the phase diagram of Fig. 3 in which we can identify two new candidates for topologically nontrivial materials, namely, InSb and $Li₂CdSn$. InSb is a topologically trivial semiconductor in its native phase with the experimental gap of 0.26 eV.³² Here, we predict that in InSb with adequate lattice expansion, the gap can be closed and a band inversion can occur. It is well known that the band gap in the III-V zinc-blende structure decreases for heavier constituent atoms and the lattice constants are larger. Earlier first-principles calculations have predicted that among heavy-atom III-V compounds, TlP and TlAs are stable

FIG. 3. (Color online) Band inversion strength and topological phase diagram. Band inversion strength is plotted as a function of the atomic numbers of M' and X elements in the zinc-blende sublattice. The calculation is carried out at $a = 12.8$ Bohr, which is 3% greater than the experimental lattice constant of $Li₂AgSb$. The black line, which defines the topological critical points by the zero value of the band inversion strength, separates the trivial and nontrivial topological phases.

FIG. 4. (Color online) Electronic structure of Li₂*M'X* series and III-V binary compounds. Bulk electronic structures of TlP in (a), TlAs in (b), Li_2AgBi in (c), Li_2AuBi in (d), Li_2CdSn in (e), and Li_2CuSb in (f). The size of blue data points is proportional to the probability of *s*-orbital occupation on the anion site. Grey shaded areas in (a)–(e) highlight the band inversion corresponding to the band structure in Fig. [1\(e\).](#page-1-0)

in the zinc-blende structures. 33 We take the total-energy optimized lattice constants 34 and obtain the band structures for TlP and TlAs in Fig. 4. Both compounds possess the band inversion feature at the Γ point. Therefore thallium-based III-V compounds are predicted to be topologically nontrivial. In addition, the II-VI compound CdTe lies at $x = 1$ and $y = -1$ in the phase diagram of Fig. [3.](#page-2-0)While CdTe is topologically trivial, the related compound HgTe is a known topologically nontrivial semimetal.³⁵ Li₂CdSn is also reported to be a cubic crystal.^{[36](#page-4-0)} If we assume Li_2CdSn has the same crystal structure as Li_2AgSb , the optimized lattice constant is obtained as $a_0 = 12.85$ Bohr. Our band structures of $Li₂CdSn$ in Fig. 4 show that a band inversion occurs at the Γ point and the conduction bands are below the E_F around the X point, allowing us to predict that Li₂CdSn is a topologically nontrivial metal.

Along the preceding lines, we can theoretically predict still other topological insulators. If we replace $Li₂AgSb$ with heavier atoms, the band inversion can be achieved at the Γ point (see Fig. 4). We predict thus that $Li₂AgBi$ and $Li₂AuBi$ are nontrivial topological metals in their pristine phases whereas $Li₂CuSb$ is a trivial band insulator. Another novel way of breaking the cubic symmetry of the zinc-blende structure to attain the topological insulating state is to double the unit cell, which leads to a tetragonal crystal structure. The chalcopyrite semiconductors of I-III-VI₂ and II-IV-V₂ group compositions, which are described by this tetragonal structure, are shown to be nontrivial topological insulators in Ref. [37,](#page-4-0) which is consistent with our predictions. Furthermore, we have found that quaternary chalcogenides and ternary famatinites, which are tetragonal with zinc-blende sublattice, are candidates for 3D topological insulators.²⁰ We note that the adiabatic continuation method used in the present calculations also predicts that the half-Heusler compounds are topologically nontrivial.^{[14](#page-4-0)}

Concerning the practical realization of topologically nontrivial phases in $Li₂AgSb$ class of compounds, as shown in Fig. [2\(c\),](#page-2-0) the nontrivial topological insulating phase is predicted to exist when a uniaxial tensile strain is applied. This could be achieved by growing the materials on suitable semiconducting or insulating substrates. Notably, all the materials discussed in the present study display electronic structures similar to those in HgTe/CdTe, which were used to achieve the two-dimensional topological insulating phase or the quantum spin Hall (QSH) phase in a quantum well system.^{[35](#page-4-0)} Therefore our work also expands the list of candidate materials for realizing QSH phases.

In conclusion, we have predicted new classes of topological insulator materials in III-V semiconductors and Li-based compounds Li₂M'X. Small-gap semiconductors such as InSb could be turned into nontrivial phases with a lattice expansion. With the example of $Li₂AgSb$, we show how the nontrivial insulating phase could be obtained by a uniaxial strain. The exciting opportunities present within the ternary $Li₂M'X$ series (with a number of variants) offer a new platform for realizing multifunctional topological devices for spintronics and fault-tolerant quantum computing applications. Undergirding our study is the demonstration that a first-principles approach based on arguments of adiabatic continuation provides a powerful tool for materials discovery in search for new topologically interesting materials encompassing centrosymmetric as well as noncentrosymmetric crystal structures. The topological phase diagrams of the sort we have obtained for the Li-based compounds in this work, which encode global materials characteristics extending from single element to binary, ternary, and possibly quaternary compositions would provide systematic opportunities for materials discovery and tunability between trivial and nontrivial topological phases, and serve to motivate further research for both fundamental and applied purposes.

The work at Northeastern and Princeton is supported by the Division of Materials Science and Engineering, Basic Energy Sciences, U.S. Department of Energy Grants DE-FG02- 07ER46352, DE-FG-02-05ER46200 and AC02-05CH11231, and benefited from theory support at the Advanced Light Source, Berkeley, and the allocation of supercomputer time at NERSC and Northeastern University's Advanced Scientific Computation Center (ASCC). M.Z.H. is supported by NSF-DMR-1006492 and DARPA-N66001-11-1-4110.

- 1M. Z. Hasan and C. L. Kane, [Rev. Mod. Phys.](http://dx.doi.org/10.1103/RevModPhys.82.3045) **82**, 3045 [\(2010\).](http://dx.doi.org/10.1103/RevModPhys.82.3045)
- 2X.-L. Qi and S.-C. Zhang, [Rev. Mod. Phys.](http://dx.doi.org/10.1103/RevModPhys.83.1057) **83**, 1057 [\(2011\).](http://dx.doi.org/10.1103/RevModPhys.83.1057)
- 3M. Z. Hasan and J. E. Moore, [Ann. Rev.: Condens. Matter Phys.](http://dx.doi.org/10.1146/annurev-conmatphys-062910-140432) **2**, [5578 \(2011\).](http://dx.doi.org/10.1146/annurev-conmatphys-062910-140432)
- 4F. Wilczek, [Nature \(London\)](http://dx.doi.org/10.1038/458129e) **458**, 129 (2009).
- 5X.-L. Qi, T. L. Hughes, and S.-C. Zhang, [Phys. Rev. B](http://dx.doi.org/10.1103/PhysRevB.78.195424) **78**, 195424 [\(2008\).](http://dx.doi.org/10.1103/PhysRevB.78.195424)
- 6A. M. Essin, J. E. Moore, and D. Vanderbilt, [Phys. Rev. Lett.](http://dx.doi.org/10.1103/PhysRevLett.102.146805) **102**, [146805 \(2009\).](http://dx.doi.org/10.1103/PhysRevLett.102.146805)
- 7B. Seradjeh, J. E. Moore, and M. Franz, [Phys. Rev. Lett.](http://dx.doi.org/10.1103/PhysRevLett.103.066402) **103**, [066402 \(2009\).](http://dx.doi.org/10.1103/PhysRevLett.103.066402)
- 8X.-L. Qi, R. Li, J. Zhang, and S.-C. Zhang, [Science](http://dx.doi.org/10.1126/science.1167747) **323**, 1184 [\(2009\).](http://dx.doi.org/10.1126/science.1167747)
- 9L. Fu and C. L. Kane, Phys. Rev. Lett. **100**[, 096407 \(2008\).](http://dx.doi.org/10.1103/PhysRevLett.100.096407)
- 10L. A. Wray *et al.*, [Nat. Phys.](http://dx.doi.org/10.1038/nphys1762) **6**, 855.
- 11Y. Xia *et al.*, Nat. Phys. **5**[, 398 \(2009\).](http://dx.doi.org/10.1038/nphys1274)
- 12D. Hsieh *et al.*, Phys. Rev. Lett. **103**[, 146401 \(2009\).](http://dx.doi.org/10.1103/PhysRevLett.103.146401)
- 13H. Lin *et al.*, New J. Phys. **13**[, 095005 \(2011\).](http://dx.doi.org/10.1088/1367-2630/13/9/095005)
- 14H. Lin *et al.*, Nat. Mater. **9**[, 546 \(2010\).](http://dx.doi.org/10.1038/nmat2771)
- 15S. Chadov *et al.*, Nat. Mater. **9**[, 541 \(2010\).](http://dx.doi.org/10.1038/nmat2770)
- 16H. Lin *et al.*, Phys. Rev. Lett. **105**[, 036404 \(2010\).](http://dx.doi.org/10.1103/PhysRevLett.105.036404)
- 17B. Yan *et al.*, [Europhys. Lett.](http://dx.doi.org/10.1209/0295-5075/90/37002) **90**, 37002 (2010).
- 18S.-Y. Xu *et al.*, Science **332**[, 560 \(2011\).](http://dx.doi.org/10.1126/science.1201607)
- 19S.-Y. Xu *et al.*, arXiv[:1007.5111.](http://arXiv.org/abs/1007.5111)
- 20Y. J. Wang *et al.*, New J. Phys. **13**[, 085017 \(2011\).](http://dx.doi.org/10.1088/1367-2630/13/8/085017)
- 21M. Dzero, K. Sun, V. Galitski, and P. Coleman, [Phys. Rev. Lett.](http://dx.doi.org/10.1103/PhysRevLett.104.106408) **104**[, 106408 \(2010\).](http://dx.doi.org/10.1103/PhysRevLett.104.106408)
- 22R. S. K. Mong, A. M. Essin, and J. E. Moore, [Phys. Rev. B](http://dx.doi.org/10.1103/PhysRevB.81.245209) **81**, [245209 \(2010\).](http://dx.doi.org/10.1103/PhysRevB.81.245209)
-
- 23S. Raghu, X.-L. Qi, C. Honerkamp, and S.-C. Zhang, [Phys. Rev.](http://dx.doi.org/10.1103/PhysRevLett.100.156401) Lett. **100**[, 156401 \(2008\).](http://dx.doi.org/10.1103/PhysRevLett.100.156401)
- 24T. Das, Sci. Rep. **2**[, 596 \(2012\).](http://dx.doi.org/10.1038/srep00596)
- 25L. Fu, Phys. Rev. Lett. **106**[, 106802 \(2011\).](http://dx.doi.org/10.1103/PhysRevLett.106.106802)
- 26L. Fu and C. L. Kane, Phys. Rev. B **76**[, 045302 \(2007\).](http://dx.doi.org/10.1103/PhysRevB.76.045302)
- 27P. Blaha *et al.* WIEN2K, An augmented plane wave plus local orbitals program for calculating crystal properties (Karlheinz Schwarz, Techn. University Wien, Austria, 2001).
- 28J. P. Perdew, K. Burke, and M. Ernzerhof, [Phys. Rev. Lett.](http://dx.doi.org/10.1103/PhysRevLett.77.3865) **77**, 3865 [\(1996\).](http://dx.doi.org/10.1103/PhysRevLett.77.3865)
- ²⁹H. Lin, S. Sahrakorpi, R. S. Markiewicz, and A. Bansil, *[Phys. Rev.](http://dx.doi.org/10.1103/PhysRevLett.96.097001)* Lett. **96**[, 097001 \(2006\);](http://dx.doi.org/10.1103/PhysRevLett.96.097001) A. Bansil, Phys. Rev. B **20**[, 4025 \(1979\);](http://dx.doi.org/10.1103/PhysRevB.20.4025) **20**[, 4035 \(1979\).](http://dx.doi.org/10.1103/PhysRevB.20.4035)
- ³⁰It will be interesting to examine effects of alloying and doping using first-principles approaches beyond the virtual crystal or rigid band type approximations, see Ref. 31.
- 31A. Bansil, R. S. Rao, P. E. Mijnarends, and L. Schwartz, [Phys. Rev.](http://dx.doi.org/10.1103/PhysRevB.23.3608) B **23**[, 3608 \(1981\);](http://dx.doi.org/10.1103/PhysRevB.23.3608) P. E. Mijnarends and A. Bansil, *ibid.* **13**[, 2381](http://dx.doi.org/10.1103/PhysRevB.13.2381) [\(1976\);](http://dx.doi.org/10.1103/PhysRevB.13.2381) L. Huisman, D. Nicholson, L. Schwartz, and A. Bansil, *[ibid.](http://dx.doi.org/10.1103/PhysRevB.24.1824)* **24**[, 1824 \(1981\);](http://dx.doi.org/10.1103/PhysRevB.24.1824) L. Schwartz and A. Bansil, *ibid.* **10**[, 3261 \(1974\).](http://dx.doi.org/10.1103/PhysRevB.10.3261)
- 32M. van Schilfgaarde, A. Sher, and A.-Ban Chen, [Appl. Phys. Lett.](http://dx.doi.org/10.1063/1.109523) **62**[, 1857 \(1993\).](http://dx.doi.org/10.1063/1.109523)
- 33Y. O. Ciftci, K. Colakoglu, and E. A. Deligoz, [Cent. Eur. J. Phys.](http://dx.doi.org/10.2478/s11534-008-0109-y) **6**[, 802 \(2008\).](http://dx.doi.org/10.2478/s11534-008-0109-y)
- 34N. Saidi-Houat, A. Zaoui, and M. Ferhat, [J. Phys.: Condens. Mat.](http://dx.doi.org/10.1088/0953-8984/19/10/106221) **19**[, 106221 \(2007\).](http://dx.doi.org/10.1088/0953-8984/19/10/106221)
- 35B. A. Bernevig, T. L. Hughes, and S.-C. Zhang, [Science](http://dx.doi.org/10.1126/science.1133734) **314**, 1757 [\(2006\).](http://dx.doi.org/10.1126/science.1133734)
- 36H.-U. Schuster, [Naturwissenschaften](http://dx.doi.org/10.1007/BF00621874) **53**, 361 (1966).
- 37W. Feng, D. Xiao, J. Ding, and Y. Yao, [Phys. Rev. Lett.](http://dx.doi.org/10.1103/PhysRevLett.106.016402) **106**, 016402 [\(2011\).](http://dx.doi.org/10.1103/PhysRevLett.106.016402)