Electronic structure, magnetic properties, and magnetostructural transition in Tb₅Si_{2.2}Ge_{1.8} from first principles

Durga Paudyal^{*} and Y. Mudryk

The Ames Laboratory, US Department of Energy, Iowa State University, Ames, Iowa 50011-3020, USA

V. K. Pecharsky and K. A. Gschneidner Jr.

The Ames Laboratory, US Department of Energy, Ames, Iowa 50011-3020, USA and Department of Materials Science and Engineering, Iowa State University, Ames, Iowa 50011-2030, USA (Received 6 May 2011; revised manuscript received 7 June 2011; published 27 July 2011)

The electronic structure and magnetic properties of Tb₅Si_{2.2}Ge_{1.8} have been studied from first principles electronic structure calculations. The total energy of the ferromagnetic (FM) orthorhombic [O(I)] Tb₅Si_{2.2}Ge_{1.8} is lower than the total energy of the FM monoclinic (M) Tb₅Si_{2.2}Ge_{1.8}, indicating that the FM O(I) is the ground state structure of Tb₅Si_{2.2}Ge_{1.8}. Because of a strong 4*f*-5*d* exchange, the splitting of 5*d* bands of Tb atoms in the FM O(I) Tb₅Si_{2.2}Ge_{1.8} is greater than that in the FM M Tb₅Si_{2.2}Ge_{1.8}, giving rise to higher 5*d* moments in the former. The magnetostructural transition temperature, T_M , and the isothermal magnetic entropy change, ΔS_M , have been calculated by coupling the parameters obtained from the first principles to the magnetothermodynamic models. Both T_M and ΔS_M are in good agreement with reported experimental values. The magnetic entropy change increases with decreasing magnetostructural transition temperature, which indicates a pathway toward tuning the magnitude of the magnetocaloric effect.

DOI: 10.1103/PhysRevB.84.014421

PACS number(s): 71.20.Eh, 71.15.Mb, 81.30.Kf

I. INTRODUCTION

Over the last decade, magnetic materials exhibiting giant magnetocaloric effect are at the forefront of research in materials physics because of their potential for applications in magnetic cooling.¹⁻⁴ The magnetic field-induced first-order phase transitions are responsible for the giant magnetocaloric effect.⁵⁻¹⁰ To date, $Gd_5Si_xGe_{4-x}$ compounds are among the best prototypes showing a strong and reversible magnetocaloric effect in an alternating magnetic field when they undergo coupled magnetic and structural transitions. The concurrent change in the magnetization and crystal structure was observed in Gd₅Ge₄ and related materials through bulk magnetization and *in situ* x-ray powder diffraction studies.^{10–13} Among the $Gd_5Si_xGe_{4-x}$ alloys, the largest magnetocaloric effect near room temperature is observed in Gd₅Si₂Ge₂ when the compound changes from the paramagnetic (PM) monoclinic (M) polymorph to the ferromagnetic (FM) orthorhombic [the so-called O(I)] polymorph. This change can be triggered by temperature,^{12,14} magnetic field,^{12,15} and/or hydrostatic pressure.^{16,17} Thus, applied magnetic field above the Curie temperature transforms the compound from the PM M to the FM O(I) state,^{18,19} during which the magnetic field dependence of the isothermal magnetization resembles that of a metamagnetic phase transition in an antiferromagnet.²⁰

The Tb₅Si_xGe_{4-x} compounds also exhibit the giant magnetocaloric effect in the intermediate range of compositions around x = 2.²¹ Initially, it was believed that this system undergoes the coupled magnetostructural phase transformation identical to that found in Gd₅Si₂Ge₂.²² However, unlike in Gd₅Si₂Ge₂, no clear metamagnetic-like behavior was observed in the magnetization of polycrystalline Tb₅Si₂Ge₂, and a follow up neutron powder diffraction study of this compound revealed a decoupling of the structural and magnetic transitions with a separation of ~10 K.²³ This indicates that the monoclinic structure may also adopt a long-range ferromagnetic order in some members of the R₅Si_xGe_{4-x} family, where R is a rare earth element, even though when R =Gd, long-range ferromagnetism is always associated with the O(I) polymorph.^{24,25} Interestingly, the magnetic and structural transitions in Tb₅Si₂Ge₂ can be recoupled by a hydrostatic pressure of 8.6 kbar or higher, consequently enhancing the magnetocaloric effect of the compound by nearly 40% for a magnetic field change from 0 to 50 kOe.²⁶ A neutron diffraction study of a single crystal of Tb₅Si_{2.2}Ge_{1.8}, however, indicates that decoupling of the magnetic and crystallographic transitions, if any, must be smaller than ~ 5 K,²⁷ thus pointing toward a compositional (x) dependence of the degree of decoupling. Because no intermediate magnetic-only transition of the monoclinic phase was detected within 5 K of the crystallographic transition in Tb₅Si_{2.2}Ge_{1.8}, it appears that the extent of the decoupling is decreasing with increasing concentration of Si. The recent measurements of Tb₅Si_{2.2}Ge_{1.8}²⁸ indicate that the decoupling is within the 5 K range.

First principles calculations performed on Gd₅Si₂Ge₂,^{29–36} Gd₅Ge₄,^{37–41} Gd₅Si_{0.5}Ge_{3.5},⁴² Gd₅Sb_{0.5}Ge_{3.5},⁴³ Ho₅Ge₄,⁴⁴ Ho₅Si_{3.2}Ge_{0.8},⁴⁵ and Ho₅Si₄⁴⁶ indicate that chemical bonding and crystallography play a major role in defining the magnetic properties of these compounds. The breaking and reforming of the interslab T-T dimers,¹⁴ which accompanies the O(I)-M and O(I)-O(II) structural transitions [the O(II) phase is an orthorhombic polymorph that is different from O(I), see Refs. 5, 9 for details), and the related nearly 1 Å elongation and contraction, respectively, of the interslab T-T bonds affect both the location of the Fermi level and the effective magnetic exchange coupling, J_0 .^{29,32} Comparing the FM ordered O(I) phase of Gd₅Si₂Ge₂ with the hypothetical FM M Gd₅Si₂Ge₂, the former is characterized by larger 5*d* magnetic moments of the Gd atoms and a larger J_0 than those of the latter, thus explaining the stability of the O(I) structure in the ferromagnetically ordered state. The different

polymorphs (i.e. the monoclinic (M) and orthorhombic O(I) phases in Gd₅Si₂Ge₂), have different T_C s and, therefore, different temperature dependencies of the magnetization.²⁹

When PM M Gd₅Si₂Ge₂ is cooled, its free energy remains lower than that of the FM O(I) Gd₅Si₂Ge₂ below the Curie temperature of the latter. The actual temperature of the magnetostructural transition of $Gd_5Si_2Ge_2$, T_M , is located between $T_C{}^M$ and $T_C{}^{O(I)}$. Hence, when the free energies of the two polymorphs become equal at T_M , the O(I) Gd₅Si₂Ge₂ phase is already well below its $T_C^{O(I)}$. As a result, the polymorphic transition at T_M also involves the ferromagnetic ordering with a large, discontinuous increase of spontaneous magnetization. When a magnetic field is applied just above T_M , it changes the balance of the free energies between the PM M and FM O(I) Gd₅Si₂Ge₂ phases, which in turn triggers a crystallographic transition between the two polymorphs that have different magnetizations, leading to a metamagnetic-like, discontinuous change of the magnetization at various temperature-specific critical fields.

Given that Tb₅Si_{2,2}Ge_{1,8} has the same sequence of polymorphs with the same type of crystal structures as Gd₅Si₂Ge₂, the metamagnetic-like transitions in Tb₅Si_{2.2}Ge_{1.8} are expected to have a similar mechanism, but the single-ion anisotropy of the Tb^{3+} ion and the orbital angular momentum quantum number L = 3 are considerable, giving rise to spin orbit and crystalline electric field interactions in Tb₅Si_{2.2}Ge_{1.8} when compared with the negligible single-ion anisotropy and L = 0 of the Gd³⁺ ion in Gd₅Si₂Ge₂. Therefore, the first goal of this work is to evaluate the total energies and magnetic moments, including orbital moments of the O(I) and M structures of Tb₅Si_{2.2}Ge_{1.8}, compute magnetic exchange interactions considering the Heisenberg approach for the nearest neighbor slabs, and then calculate the Curie temperatures utilizing the mean field relation. The next goal is to analyze finite temperature properties, such as the magnetic free energy and magnetic entropy by coupling first principles outputs such as saturated magnetic moments, including orbital moments and Curie temperatures, to the magnetothermodynamic models in order to calculate the magnetostructural transition temperature and magnetic entropy change.

II. CRYSTAL AND MAGNETIC STRUCTURES OF Tb₅Si_{2.2}Ge_{1.8}

From neutron diffraction²⁷ and x-ray diffraction⁴⁷ studies, it has been established that the ground state structure of Tb₅Si_{2.2}Ge_{1.8} is orthorhombic (space group Pnma), Gd₅Si₄ type, also known as O(I).⁴⁸ Above ~107 K, it transforms into the monoclinic (space group P112₁/a) Gd₅Si₂Ge₂-type crystal structure, also known as M.⁴⁸ The Gd₅Si₂Ge₂-type polymorph is a martensitic-like distortion of the Gd₅Si₄-type orthorhombic structure. These structures are shown in Fig. 1.

The two polymorphs of Tb₅Si_{2.2}Ge_{1.8} may be conveniently described in terms of nearly two-dimensional, \sim 7 Å thick slabs formed by strongly bonded Tb, Si, and Ge atoms.^{5,14,21,26} Here, T1 in the O(I) and T1a, T1b in the M structures represent Si and Ge atoms statistically (0.40 Si + 0.60 Ge) mixed in the interslab positions, henceforth referred as the T' sites, whereas T2 and T3 (0.67 Si + 0.33 Ge) in both structures represent the Si and Ge atoms in the intraslab positions



FIG. 1. (Color online) The crystal structures of the monoclinic (left) and orthorhombic (right) phases of $Tb_5Si_{2.2}Ge_{1.8}$. T1 in the orthorhombic and T1a, T1b in the monoclinic structures represent the Si/Ge atoms in the interslab positions (T' sites), and T2, T3 in both phases represent the Si/Ge atoms in intraslab positions (T sites).

(henceforth the T sites). The -Tb-T'-T'-Tb- interslab networks are continuous in the low-temperature O(I) polymorph. But at high temperatures (in the M polymorph), these networks are no longer continuous because of a substantial elongation (bond distances are increasing from 2.81 to 3.45 Å)⁴⁷ of every other T'-T' bond distance. When temperature is reduced, the process is reversible (i.e., the slabs reverse the movements with respect to one another in a shear fashion and the interslab networks become continuous).

The ground state O(I) structure of Tb₅Si_{2.2}Ge_{1.8} is ferromagnetic. Interestingly, above ~107 K, it transforms to the monoclinic polymorph but remains ferromagnetic up to ~112 K, indicating a weakly decoupled magnetostructural transition in this compound.²⁸ Above ~112 K, FM M Tb₅Si_{2.2}Ge_{1.8} transforms to the PM state via a conventional second-order phase transition.

III. THEORETICAL APPROACH

The local spin density approximation including Hubbard U parameter (LSDA + U) approach^{49,50} has been employed to investigate the electronic structure and magnetic properties of the Tb₅Si_{2.2}Ge_{1.8} system. Using LSDA + U, the localized 4*f* electrons of Tb atoms in this system can be suitably treated. The LSDA + U method starts from the LSDA total energy, which is supplemented by an additional intra-atomic Coulomb correlation term *U* and exchange interaction term *J* of multiband Hubbard type, minus a so-called double counting term to subtract the electron-electron interactions already included in LSDA. Our calculations have been performed using the scalar relativistic version (which includes the mass velocity and Darwin correction terms) of the LSDA + U method implemented in the tight-binding linear muffin tin orbital (TB-LMTO).⁵¹ Because the difference between the

occupied energy levels (determined from photoemission spectroscopy) and the unoccupied energy levels (determined from bremsstrahlung isochromatic spectroscopy) of Tb⁵² and Gd⁴⁹ are nearly identical after neglecting spin orbit coupling, we employed U = 6.7 eV and J = 0.7 eV—the well-known values for Gd atoms—also for Tb atoms in Tb₅Si_{2.2}Ge_{1.8}. In these calculations, the conventional von Barth and Hedin parameterization of the LSDA⁵³ has been adopted, which is one of the well-known local exchange correlation functionals. Furthermore, we have used 125 and 170 special k points in the irreducible part of the Brillouin zone for k-space integration in the orthorhombic and monoclinic phases, respectively.

To check the effect of full potential in these O(I) and M structures of $Tb_5Si_{2.2}Ge_{1.8}$, we have also employed the full potential linear augmented plane wave (FP-LAPW) method for the electronic structure calculations. This method yields results similar to TB-LMTO, indicating that the TB-LMTO method with atomic sphere approximation (which is much faster than the FP-LAPW) is also suitable for electronic structure calculations of $Tb_5Si_{2.2}Ge_{1.8}$. Therefore, here, we show only the electronic structure results obtained from the TB-LMTO method and couple these results to the magnetothermodynamic models for estimation of finite temperature properties.

As stated in the previous section, diffraction experiments⁴⁷ indicate a mixed occupancy of Si and Ge atoms in the T' and T sites of Tb₅Si_{2.2}Ge_{1.8}. The 60% Ge and 40% Si occupy the T' positions, whereas 33% Ge and 67% Si occupy the T positions in both the O(I) and M structures of Tb₅Si_{2.2}Ge_{1.8}. For theoretical calculations, two different mixing approaches were employed to incorporate both Si and Ge in the appropriate sites. The Ge atoms were placed in all T' and T sites as a first model and then Si atoms in all T' and T positions as a second model in the first approach. The 45% of outputs from the first model and 55% outputs from the second model constitute the final outputs. In the second approach, we locate Ge atoms in the T' and Si atoms in the T positions as a third model and then Si in the T' and Ge in the T positions as a fourth model. Then, we take weighted averages of the outputs from the third and fourth models according to the experimentally determined occupancies of the interslab and intraslab sites by Ge and Si atoms.

The magnetic interactions in Tb₅Si_{2.2}Ge_{1.8} are mainly due to indirect (i.e., Ruderman–Kittel–Kasuya–Yosida [RKKY]⁵⁴) Tb 4*f*–Tb 4*f* exchange mediated by the spin-polarized 5*d* electrons. The origin of this 4*f*-4*f* exchange is the 4*f*-5*d* exchange, and, therefore, the changes in the 4*f*-5*d* exchange reflect the changes in the 4*f*-4*f* interactions when one structure is replaced by another. Hence, we estimate the 4*f*-5*d* exchange interactions of Tb atoms in the O(I) and M structures of Tb₅Si_{2.2}Ge_{1.8}. Recently, this approach has been successfully applied to the Gd₅Ge₄ system,³⁸ and for theoretical details and application to other rare earth–based materials, we refer interested readers to Refs. 55–58.

The interactions between orbital states of magnetic ions and crystalline electric fields are typical for rare earth materials. The crystal field has site-matching symmetry, which gives rise to a strong coupling between the 4f orbital moment and the crystal lattice. This coupling may be transferred to the spin moments via spin orbit coupling. As a result, an interesting interplay between spin orbit interactions, crystalline electric

field, and exchange interactions often occurs in lanthanide containing intermetallic compounds. In Gd-based materials, Gd³⁺ ions have negligible single-ion anisotropy; therefore, the indirect 4f-4f exchange, mediated by spin-polarized conduction electrons, dominates over the crystalline electric field-coupled spin orbit mechanism. A study of multiplet effects in the electronic structure of heavy rare earth metals⁵² indicates that the spin orbit constant in Tb metal is 0.24 eV/Tb, which is identical to the value of a single Tb ion.⁵⁹ This may imply that in the metallic Tb, the crystalline electric field effect, which also depends on the orbital moment, has the same value as that of a single Tb^{3+} ion (e.g., Steven's second-order parameter for Tb^{3+} ion is $-1/99^{60}$). The FP-LAPW calculations with the inclusion of spin orbit coupling show the same orbital moments (3.02 $\mu_{\rm B}$ /Tb orbital moment [3 $\mu_{\rm B}$ contributed from 4f and 0.02 $\mu_{\rm B}$ from 5d electrons]) for both the O(I) and M polymorphs of Tb₅Si_{2.2}Ge_{1.8}. The orbital moment contributions from 4f Tb and 5d Tb electrons have not been determined from experimental data, although it may be possible to do so from recent x-ray magnetic circular dichroism measurements on the magnetostructural properties of $\text{Tb}_5(\text{Si}_x\text{Ge}_{1-x})_4$.⁶¹ Because the M structure is a distorted O(I) structure, identical orbital moments in both the O(I) and M structures are not surprising. The results shown below include orbital moments together with the spin moments of Tb atoms in both structures of Tb₅Si_{2.2}Ge_{1.8}.

IV. RESULTS AND DISCUSSION

Experimental observations⁴⁷ show that Tb₅Si_{2.2}Ge_{1.8} transforms from the FM O(I) phase to the FM M variant at ~107 K, but FM M transforms to PM M at ~112 K, indicating a narrowly decoupled magnetostructural transition. To theoretically determine the ground state structure of Tb₅Si_{2.2}Ge_{1.8}, the total energies were calculated in both the O(I) and M structures. The total energy of the FM O(I) is lower than that of the FM M by 3.66 eV/cell from the first mixing approach and 3.63 eV/cell from the second mixing approach (both approaches were discussed in the previous section), which confirms that the FM O(I) is indeed the ground state structure of Tb₅Si_{2.2}Ge_{1.8}.

To probe the relative stabilities of the O(I) and M phases of $Tb_5Si_{2.2}Ge_{1.8}$, the phase formation energies (E_f) have been calculated from the total energies of the alloy and the components using the following relation:

$$E_f = E_{\text{Tb}_5 \text{Si}_{2,2}\text{Ge}_{1,8}} - c_{\text{Tb}}E_{\text{Tb}} - c_{\text{Si}}E_{\text{Si}} - c_{\text{Ge}}E_{\text{Ge}}.$$

Here, *E* and *c* denote total energy and molar fractions, respectively. The difference in the formation energies of the two phases is the energy required to transform one phase into the other. The computed formation energies of both the O(I) and M phases are negative. The E_f of the FM O(I) Tb₅Si_{2.2}Ge_{1.8} is lower compared with the FM M Tb₅Si_{2.2}Ge_{1.8}. But E_f for the PM M Tb₅Si_{2.2}Ge_{1.8} is lower than that of the FM M for Tb₅Si_{2.2}Ge_{1.8} (Table I). These results indicate that the O(I) Tb₅Si_{2.2}Ge_{1.8} is stable in the FM ordered state and that the M Tb₅Si_{2.2}Ge_{1.8} is stable in the magnetically disordered state. Interestingly, the formation energy of the FM O(I) phase of Tb₅Si_{2.2}Ge_{1.8} is higher by 39% than the same of the FM O(I) phase of Gd₅Si₂Ge₂.²⁹ which indicates greater thermodynamic stability of the FM O(I) Gd₅Si₂Ge₂ compared

TABLE I. Phase formation energies calculated using two different mixing approaches. In the first mixing approach, the Ge atoms were placed in all T' and T sites as a first model and then Si atoms in all T' and T positions as a second model. The 45% of outputs from the first model and 55% from the second model constitute the final output. In the second mixing approach, we locate Ge atoms in the T' and Si in the T positions as a furth model and then Si in the T' and Ge in the T positions as a fourth model. Weighted averages of the outputs from the third and fourth models are calculated according to the experimentally determined occupancies of the interslab and intraslab sites by Ge and Si atoms.

	Phase formation energy (meV/atom)			
Phase	First mixing approach	Second mixing approach		
O(I) FM	-819.9	-822.7		
O(I) PM	-714.8	-717.9		
M FM	-717.3	-721.8		
M PM	-724.0	-727.9		

with the FM O(I) Tb₅Si_{2.2}Ge_{1.8}. This difference is consistent with a gradual reduction of the degree of transformation of the M phases into the O(I) phases observed on cooling in the series Gd₅Si₂Ge₂⁶² \rightarrow Tb₅Si_{2.2}Ge_{1.8}⁴⁷ \rightarrow Dy₅Si₃Ge⁶³ \rightarrow Ho₅Si_{3.2}Ge_{0.8}⁴⁵ \rightarrow Er₅Si₄,⁶⁴ which is respectively lowered from ~95% to ~80%, ~50%, 0%, and 0%.

Table II shows that the 5d exchange splitting in the FM O(I) Tb₅Si_{2.2}Ge_{1.8} is higher than the same in the FM M $Tb_5Si_{2,2}Ge_{1,8}$, giving rise to higher 5d moments in the O(I) polymorph. The magnetic moments of Tb atoms in both structures are shown in Table III. Because of the anisotropy in the nearest neighbor distances, the 5d magnetic moments of the nonequivalent sites of Tb atoms are variable in both the O(I) and M structures. The magnetic moments of Tb3 atoms (which are located in the middle of the slabs and play a key role in forming the -Tb-T'-T'-Tb- networks that are continuous in the O(I) and discontinuous in the M $Tb_5Si_{2,2}Ge_{1,8}$) are larger compared with other nonequivalent Tb atoms (which are located on the surfaces of the slabs) in both FM O(I) and FM M Tb₅Si_{2,2}Ge_{1,8}. Hence, the Tb3 atoms occupying 4c sites in both the O(I) and M polymorphs may have a strong influence on the overall magnetism of the compound.

To probe how the structural change affects the electronic structure and magnetism of $Tb_5Si_{2.2}Ge_{1.8}$, the 5*d*-Tb and 4*p*-Ge densities of states (DOS) were calculated, with the FM order imposed in both the O(I) and M structures. Figure 2 shows that the average 5*d* DOS at the Fermi level in the FM

TABLE II. The average 5*d* spin splitting of Tb atoms in the FM O(I) and FM M structures of Tb₅Si_{2.2}Ge_{1.8}.

	5 <i>d</i> Splittin (meV)	g energy /Tb)
Population of T and T' sites	FM O(I)	FM M
Ge in both T' and T sites	511	474
Si in both T' and T sites	506	469
Ge in T' and Si in T sites	505	462
Si in T' and Ge in T sites	509	480

O(I) is higher compared with FM M Tb₅Si_{2.2}Ge_{1.8}. The degree of spin polarization, $\eta = \frac{D^{\uparrow}(E_F) - D^{\downarrow}(E_F)}{D^{\uparrow}(E_F) + D^{\downarrow}(E_F)}$ is also higher by 43% in the O(I) structure. The integrated number of 5*d* electrons up to the Fermi level and spin-up and -down band splitting are also higher by 23% and 8%, respectively, in the O(I) Tb₅Si_{2.2}Ge_{1.8}. Therefore, spin polarization plays a main role in the stability of the O(I) phase in the FM state compared with the M polymorph, in agreement with the total energy analysis.

Figure 3 shows total DOS at E_F assuming different occupations of the interslab T' and intraslab T positions. The DOS (E_F) is always higher in the FM O(I) than in the FM M, regardless of the differences in occupations of the interslab and intraslab T' and T positions. Because the spin-down DOS (E_F) is similar in all cases, the differences are only shown for the spin-up DOS (E_F) , which therefore indicates the differences in the spin polarization with different possibilities of occupation of T and T' sites. Placing Si atoms in the inter- and intraslab positions instead of Ge atoms enhances the number of electrons at the Fermi level, except for the case with the interslab Ge and intraslab Si in the O(I) Tb₅Si_{2.2}Ge_{1.8} phase, in which the density of states falls in a valley of the majority bands instead of a peak. This is peculiar for Tb₅Si_{2.2}Ge_{1.8} because in Gd₅Si_{0.5}Ge_{3.5},⁴² the DOS (E_F) is higher with intraslab Si substitution. Nevertheless the variability of the DOS (E_F) assuming different occupations of Ge and Si in both O(I) and M polymorphs are reflected in the magnetic exchange interactions and the Curie temperatures shown in Table IV.

Figure 4 compares the average DOS of all Tb atoms and DOS for Tb3 atoms (located in the middle of the slabs). Both behave similarly, but the DOS averaged over all Tb atoms is lower than that of the Tb3 atom at the Fermi level in either polymorph. Also the DOS (E_F) of the Tb3 atom is higher in the O(I) than the same in the M polymorph. These results, therefore, provide further evidence that the Tb3 atoms play a key role in the magnetism of Tb₅Si_{2.2}Ge_{1.8}.

Because Tb3 atoms from one slab are connected to Tb3 from the neighboring slabs, either through the short T'-T' pairs in the O(I) structure or through alternatively short and long T'-T' pairs in the M structure, the difference in the overlap between the 5d states of Tb and the 4p/3p states of T' should be responsible for the difference in the interslab coupling in these two structure types. To examine these overlaps, for example, the average 5d DOS of all Tb atoms and average 4pDOS of interslab Ge atoms in both structures are compared in Fig. 5. The spin-up 4p DOS of Ge at E_F is slightly higher than the spin-down 4p DOS of Ge in the O(I) structure, which indicates a small 4p spin polarization induced by 5d electrons of Tb in this polymorph. On the other hand, the spin-up and spin-down 4p DOS (E_F) are identical in the M polymorph, indicating no 4p spin polarization in the M polymorph. The 3p DOS of Si behaves similarly but shows slightly higher spin polarization at the Fermi level compared with that of 4p Ge. There is 70% 5d-Tb and 30% 4p-Ge hybridization at about -0.5 eV in the spin-down bands of the M polymorph, but both spin-down 5d-Tb and 4p-Ge are zero at this energy range in the O(I) structure. These results, therefore, indicate that when the O(I) polymorph transforms to the M polymorph, the spin-down 4p/3p states that are pushed toward the Fermi level because of the weak T'-T' bonding hybridize with 5d

Atom	Magnetic moment (μ_B)					
	Ge in T' and T sites	Si in T' and T sites	Ge in T' and Si in T sites	Si in T' and Ge in T sites	Expt. 27	
FM O(I)						
Tb1	9.31	9.32	9.31	9.33	9.30	
Tb2	9.29	9.27	9.27	9.27	9.20	
Tb3	9.44	9.43	9.42	9.44	9.60	
FM M						
Tb1a	9.24	9.25	9.22	9.26		
Tb1b	9.19	9.18	9.16	9.20		
Tb2a	9.25	9.24	9.22	9.27		
Tb2b	9.19	9.18	9.17	9.20		
Tb3	9.33	9.32	9.31	9.35		

TABLE III. Magnetic moments of Tb atoms in the FM O(I) and FM M polymorphs of Tb₅Si_{2.2}Ge_{1.8}.

states and reduce the magnetic moments of Tb atoms in the M polymorph.

The influence of 4f spins on 5d electrons has been further examined through the calculations of the 4f-5d exchange interactions in both structures. On average, the 4f-5d exchange interactions in the O(I) Tb₅Si_{2.2}Ge_{1.8} are stronger by 10% when compared with that in the M Tb₅Si_{2.2}Ge_{1.8}. This is smaller than the 20% difference in the 4f-5d exchange interactions between the O(I) and M structures of Gd₅Si₂Ge₂. In Gd₅Si₂Ge₂, there is a coupled magnetostructural transition and a 30% increase in the lengths of interslab T'-T' bonds during a transformation from the low-temperature FM O(I) structure to the high-temperature PM M polymorph. A similar 30% increase in the interslab T'-T' bond lengths, but a smaller change in the 4f-5d exchange interactions in Tb₅Si_{2.2}Ge_{1.8} compared with Gd₅Si₂Ge₂, can be related to the competition between magnetic energy and strain energy (the change in the unit cell dimensions in Tb₅Si_{2.2}Ge_{1.8} is highly anisotropic, and the difference in phase volumes exceeds $0.7\%^{47}$). The calculations show that the strain energy associated with the



FIG. 2. (Color online) Comparison of the average 5*d* density of states (DOS) of all Tb atoms in the O(I) and the M structures of Tb₅Si_{2.2}Ge_{1.8}. The DOS at E_F and the exchange splitting of 5*d* bands are higher in the O(I) structure. Here and in all other DOS figures, the vertical arrows indicate the majority (spin-up) and minority (spin-down) bands, respectively.

O(I) to M transformation is higher by $\sim 7\%$ in Tb₅Si_{2.2}Ge_{1.8} than in Gd₅Si₂Ge₂. Here, the strain energy is estimated by calculating the nonmagnetic total energy difference between the O(I) and M polymorphs. The increase in strain energy is not immediately compensated by the magnetic exchange energy, and Tb₅Si_{2.2}Ge_{1.8} remains in the ferromagnetic state in the M Tb₅Si_{2.2}Ge_{1.8} polymorph over a small range of temperatures. This conclusion is also consistent with the lower percentage (80%) of completion of the M to O(I) transformation in Tb₅Si_{2.2}Ge_{1.8} compared with Gd₅Si₂Ge₂ (95%), as mentioned above, and also because the formation energy of FM O(I) in the Tb₅Si_{2.2}Ge_{1.8} is 39% higher than in FM O(I) Gd₅Si₂Ge₂.

Because the 4*f* magnetic moments of Tb are localized, the Heisenberg-type Hamiltonian for the dependence of energy on spin configuration may be applied to $Tb_5Si_{2.2}Ge_{1.8}$ assuming indirect RKKY-type exchange interactions. The spin Hamiltonian with zero external magnetic fields is given by⁶⁵

$$H = -\sum_{i} \sum_{\delta} j_{i,i+\delta} \stackrel{\rightarrow}{\underset{i}{S}} \stackrel{\rightarrow}{\underset{i+\delta}{S}},$$



FIG. 3. Comparison of density of states at the Fermi level with different possibilities of occupation of interslab T' and intraslab T positions.

TABLE IV. Curie temperatures with different occupations of the interslab and intraslab sites by Si and Ge in the FM O(I) and FM M structures of $Tb_5Si_{2.2}Ge_{1.8}$.

	Curie tempe	e temperature (K)	
Site occupation	O(I) FM	M FM	
T' Ge and T Ge positions	119	62	
T' Si and T Si positions	121	70	
T' Ge and T Si positions	99	64	
T' Si and T Ge positions	120	80	

where $j_{i,i+\delta}$ is an exchange coupling constant between the spin \vec{S}_i and its nearest neighbor spin $\vec{S}_{i+\delta}$, which are separated by δ . Then, the Curie temperature of a conventional three-dimensional Heisenberg ferromagnet in the mean field approximation is given by⁶⁶

$$T_C = \frac{2}{3} \frac{E_{\rm AFM} - E_{\rm FM}}{k_B}$$

Here, $E_{AFM} - E_{FM}$ is the energy required to align neighboring slabs ferromagnetically (i.e., the interslab exchange energy), and k_B is Boltzmann's constant. $E_{AFM} - E_{FM}$ is estimated by calculating the total energy difference between the AFM- and FM-aligned nearest neighbor slabs. The AFM structure is constructed in such a way that 10 Tb atoms/cell that belong to the same slab are aligned ferromagnetically, but the remaining 10 Tb atoms/cell that belong to the neighboring slab are aligned ferromagnetically in the opposite direction. In these calculations, the O(I) and M symmetries have been converted into the equivalent triclinic structures with P1 symmetry so that each of the 36 atoms in the unit cell is formally no longer equivalent to any other atom in the same unit cell.³⁸ This helps to assign any configuration of Tb spins.

The calculated Curie temperatures with different possible occupations of the interslab T' and intraslab T sites are shown in Table IV. The Curie temperature increases when the interslab positions are occupied by Si atoms. The greatest increase is observed when switching from Ge in T' and Si

in T to Si in T' and Ge in T. Considering the difference in size of Si and Ge atoms, bringing the neighboring slabs closer together promotes ferromagnetism, which also can be achieved by applying uniaxial pressure along the b-axis.

Finite temperature properties of a material can be calculated by coupling first principles outputs (e.g., magnetic moments and Curie temperatures $[T_{\rm C}]$) to magnetothermodynamic models and solving them self consistently. To determine the magnetostructural transition temperature, we calculate the magnetization and magnetic entropy separately for each of the two phases and then compute the free energies. The temperature at which the free energies are equal for both structures represents a magnetostructural transition temperature. Figure 6 shows the change in magnetic free energy in both the O(I) and M phases of Tb₅Si_{2.2}Ge_{1.8}, calculated using the free energy relation⁶⁷

$$F = -\frac{3}{2}\frac{J}{J+1}RT_C\sigma^2 - TS_M$$

where

$$S_M \approx R \left[\ln(2J+1) - \frac{3}{2} \left(\frac{J}{J+1} \right) \sigma^2 - \frac{9}{20} \frac{(2J+1)^4 - 1}{[2(J+1)]^4} \sigma^4 \right], \text{ and}$$
$$\sigma = \frac{M}{M_S} = \tanh \left[\frac{3J}{J+1} \frac{M}{M_S} \frac{T_C}{T} \right]$$

Here, S_M and σ are magnetic entropy and renormalized magnetization. *R* is the universal gas constant, and *J* is the sum of spin and orbital angular momentum quantum numbers.

Each of the polymorphs, O(I) Tb₅Si_{2.2}Ge_{1.8} and M Tb₅Si_{2.2}Ge_{1.8}, orders magnetically via a conventional second-order phase transition with the $T_C^{O(I)}$ of the O(I), Tb₅Si_{2.2}Ge_{1.8} being considerably higher than the T_C^M of the M polymorph (Table IV; Fig. 8, dotted curves). As shown in Fig. 6, when M Tb₅Si_{2.2}Ge_{1.8} is cooled, its free energy initially remains lower than that of the O(I) Tb₅Si_{2.2}Ge_{1.8}. On further cooling, the free energies of the two polymorphs become equal at $T_M = 96$ K. Below T_M , the free energy of



FIG. 4. (Color online) Comparison of 5d DOS averaged over all Tb atoms and 5d DOS of Tb3 atom in the O(I) (left) and M (right) structures of Tb₅Si_{2.2}Ge_{1.8}. Behavior of the average and individual 5d DOS are similar, but the average DOS is lower than the same of Tb3 at the Fermi level.



FIG. 5. (Color online) Comparison of average 5d DOS of Tb atoms and interslab 4p Ge atoms in the O(I) and M structures of Tb₅Si_{2.2}Ge_{1.8}.

the O(I) polymorph becomes lower than the M polymorph. This polymorphic transition at T_M also involves ferromagnetic ordering, with a large, discontinuous increase of the spontaneous magnetization (Fig. 8). Therefore, theory predicts that at $T_M = 96$ K, there is a magnetostructural transition. Furthermore, when a magnetic field is applied just above T_M , it changes the balance of the free energies between the PM M and FM O(I) phases, which in turn triggers a crystallographic transition between the two polymorphs that have different magnetizations, leading to a metamagnetic-like, discontinuous change of the magnetization at various temperature-specific critical fields.²⁹ It should be noted that the value of the magnetostructural transition temperature, $T_M = 96$ K, has been computed using the first mixing approach when, in one model, T and T' sites were occupied by Si, and in another model, these were occupied by Ge. The second mixing approach results in the magnetostructural transition temperature of 92 K. Hence, both approaches predict nearly identical magnetostructural transition temperatures. Therefore, below, we will only show results of calculations using the first mixing approach.

The free energy analysis points to a simultaneous structural and magnetic transition, whereas experimentally, they are narrowly decoupled. To check whether this theoretical result can be improved, the electronic structure calculations have been repeated with the same O(I) unit cell volume for both the O(I) and M Tb₅Si_{2.2}Ge_{1.8}. This leads to a slight increase of the 5*d* moments of Tb atoms in M Tb₅Si_{2.2}Ge_{1.8}. Hence, the exchange coupling energy in the M Tb₅Si_{2.2}Ge_{1.8} increases and the Curie temperature of the M Tb₅Si_{2.2}Ge_{1.8} increases from 66 to 78 K. The magnetostructural transition temperature becomes 102 K, which is higher than the magnetostructural transition temperature obtained with the use of different unit cell volumes of O(I) and M structures. This shows that the magnetostructural transition temperature is always higher than the Curie temperature of the monoclinic polymorph and lower than that of the O(I) polymorph in Tb₅Si_{2.2}Ge_{1.8}.

Figure 7 shows the variation of the magnetic entropies with temperature in the O(I) and M structures assuming ferromagnetic ordering at their respective T_C . The magnetic entropy change, ΔS_M , at the magnetostructural transition temperature $T_M = 96$ K is -20 J/kg K. We note that as



FIG. 6. Change in the magnetic free energy calculated using magnetic entropies and magnetization obtained from respective saturated magnetic moments and Curie temperature of the O(I) and M phases of $Tb_5Si_{2,2}Ge_{1,8}$.



FIG. 7. (Color online) Magnetic entropy of the O(I) and M structures of $Tb_5Si_{2.2}Ge_{1.8}$ as a function of temperature. The magnetic entropy change at the magnetostructural transition temperature is -20 J/kg K. The magnitude of magnetic entropy change increases with the decrease of magnetostructural transition temperature.



FIG. 8. (Color online) Magnetization as a function of temperature computed using the predicted magnetostructural transition temperature from the free energy analysis. At the magnetostructural transition temperature, T_M , the FM-ordered O(I) Tb₅Si_{2.2}Ge_{1.8} transforms into M Tb₅Si_{2.2}Ge_{1.8}, which must be already PM at T_M , thus resulting in a discontinuous change of the magnetization. The solid line follows the magnetization of the system as temperature varies. The symbols represent the experimental values of magnetization of Tb₅Si_{2.2}Ge_{1.8} in the *a*-axis.

the magnetostructural transition temperature decreases, the magnetic entropy change $(-\Delta S_M)$ becomes larger and larger, potentially reaching -40 to -45 J/kg K if T_C^M and $T_C^{O(I)}$ remain constant but T_M is reduced to ~70 K.

The experimentally observed maximum entropy change for a single crystal with the magnetic field vector parallel to different crystallographic directions *a*, *c*, and *b* are -40 J/kg K, -38 J/kg K, and -3.6 J/kg K.⁴⁷ These differences are due to anisotropy on the magnetocaloric effect of Tb₅Si_{2.2}Ge_{1.8}. The theoretically calculated magnetic entropy change should be compared with the values observed along the easy magnetization direction, which is the *a*-axis. A part of the difference (~10 J/kg K²⁹) between the calculated and experimentally measured entropy change along the easy magnetization direction reflects the contribution from the structural change, which is not accounted for in the magnetic entropy curves shown in Fig. 7 (also see our recent publications describing the difference in the Gd₅Si₂Ge₂ system).^{15,29}

In Fig. 8, magnetization as a function of temperature renormalized to the magnetostructural transition temperature is computed and compared with the experimentally observed

spontaneous magnetic moments extrapolated to zero magnetic field from the isothermal magnetization data corresponding to different temperatures.^{27,47} As follows from Fig. 8, there is a good agreement between theoretically calculated and experimentally observed values of temperature-dependent magnetic moments of Tb atoms in $Tb_5Si_{2.2}Ge_{1.8}$. This shows that the theoretical approach presented here is capable of reproducing magnetic moments at finite temperature.

V. SUMMARY

The lower value of the total energy in the FM O(I) phase compared with the FM M phase confirms the stability of the FM O(I) Tb₅Si_{2.2}Ge_{1.8} phase at low temperature. The calculated magnetic moments of Tb atoms in the FM O(I) Tb₅Si_{2.2}Ge_{1.8} are larger than in the FM M Tb₅Si_{2.2}Ge_{1.8}. The magnetic exchange coupling energy of the orthorhombic phase is higher than that of the monoclinic phase, giving rise to a higher Curie temperature in the O(I) phase than in the M phase. The calculations show that the strain energy due to structural change is not immediately compensated by the magnetic exchange energy; hence, the system may remain in the ferromagnetic state over a small range of temperature, even after the structural change, which is observed experimentally. The magnetic moment as a function of temperature calculated using the Brillouin function with inputs from first principles agrees with the corresponding experimental values. The magnetostructural transition and magnetocaloric effect in $Tb_5Si_{2.2}Ge_{1.8}$ have been estimated by coupling the electronic structure with the magnetothermodynamic model. This approach closely reproduces the magnetostructural transition temperature observed experimentally. The magnetic entropy change at the magnetostructural transition temperature is smaller than the experimental value along the easy magnetization direction, which is related to the unaccounted difference in the lattice entropies of O(I) and M Tb₅Si_{2.2}Ge_{1.8}. The magnetic entropy change increases with a decrease of the magnetostructural transition temperature, which indicates a pathway toward tuning the magnitude of the magnetocaloric effect.

ACKNOWLEDGMENTS

The Ames Laboratory is operated by Iowa State University of Science and Technology for the US Department of Energy under Contract No. DE-AC02-07CH11358. This work was supported by the Office of Basic Energy Sciences, Materials Science and Engineering Division of the Office of Science.

^{*}durga@ameslab.gov

 ¹A. M. Tishin and Y. I. Spichkin, *The Magnetocaloric Effect and its Applications* (Institute of Physics, Bristol and Philadelphia, 2003).
 ²K. A. Gschneidner Jr., V. K. Pecharsky, and A. O. Tsokol, Rep.

Prog. Phys. **68**, 1479 (2005), and references therein. ³V. K. Pecharsky and K. A. Gschneidner Jr., J. Magn. Magn. Mater.

²⁰⁰, 44 (1999).

⁴E. Bruck, J. Phys. D **38**, R381 (2005).

⁵V. K. Pecharsky and K. A. Gschneidner Jr., Adv. Mater. (Weinheim) **13**, 683 (2001).

⁶F. X. Hu, B. G. Shen, J. R. Sun, Z. H. Cheng, G. H. Rao, and X. X. Zhang, Appl. Phys. Lett. **78**, 3675 (2001).

⁷H. Wada and Y. Tanabe, Appl. Phys. Lett. **79**, 3302 (2001).

ELECTRONIC STRUCTURE, MAGNETIC PROPERTIES, ...

- ⁸W. Chen, W. Zhong, C. F. Pan, H. Chang, and Y. W. Du, Acta Phys. Sin. **50**, 319 (2001).
- ⁹V. K. Pecharsky and K. A. Gschneidner Jr., Pure Appl. Chem. **79**, 1383 (2007).
- ¹⁰V. K. Pecharsky, A. P. Holm, K. A. Gschneidner Jr., and R. Rink, Phys. Rev. Lett. **91**, 197204 (2003).
- ¹¹Ya. Mudryk, A. P. Holm, K. A. Gschneidner Jr., and V. K. Pecharsky, Phys. Rev. B **72**, 064442 (2005).
- ¹²L. Morellon, P. A. Algarabel, M. R. Ibarra, J. Blasco, and B. García-Landa, Phys. Rev. B 58, R14721 (1998).
- ¹³L. Morellon, J. Blasco, P. A. Alrarabel, and M. R. Ibarra, Phys. Rev. B **62**, 1022 (2000).
- ¹⁴W. Choe, V. K. Pecharsky, A. O. Pecharsky, K. A. Gschneidner Jr., V. G. Young Jr., and G. J. Miler, Phys. Rev. Lett. **84**, 4617 (2000).
- ¹⁵V. K. Pecharsky and K. A. Gschneidner Jr., in *Magnetism and Structure in Functional Materials*, edited by A. Planes, L. Manosa, and A. Saxena, Vol. 79 Springer Series in Materials Science (Springer, Heidelberg, 2005).
- ¹⁶Ya. Mudryk, Y. Lee, T. Vogt, K. A. Gschneidner Jr., and V. K. Pecharsky, Phys. Rev. B **71**, 174104 (2005).
- ¹⁷C. Magen, L. Morellon, P. A. Algarabel, M. R. Ibarra, Z. Arnold, J. Kamarad, T. A. Lograsso, D. L. Schlagel, V. K. Pecharsky, A. O. Tsokol, and K. A. Gschneidner Jr., Phys. Rev. B **72**, 024416 (2005).
- ¹⁸E. M. Levin, V. K. Pecharsky, and K. A. Gschneidner Jr., Phys. Rev. B **62**, R14625 (2000).
- ¹⁹H. Tang, A. O. Pecharsky, D. L. Schlagel, T. A. Lograsso, V. K. Pecharsky, and K. A. Gschneidner Jr., J. Appl. Phys. **93**, 8298 (2003).
- ²⁰V. K. Pecharsky and K. A. Gschneidner Jr., Phys. Rev. Lett. 78, 4494 (1997).
- ²¹L. Morellon, C. Magen, P. A. Algarabel, M. R. Ibarra, and C. Ritter, Appl. Phys. Lett. **79**, 1318 (2001).
- ²²C. Ritter, L. Morellon, P. A. Algarabel, C. Magen, and M. R. Ibarra, Phys. Rev. B **65**, 094405 (2002).
- ²³L. Morellon, C. Ritter, C. Magen, P. A. Algarabel, and M. R. Ibarra, Phys. Rev. B 68, 024417 (2003).
- ²⁴V. K. Pecharsky and K. A. Gschneidner Jr., Appl. Phys. Lett. **70**, 3299 (1997).
- ²⁵A. O. Pecharsky, K. A. Gschneidner Jr., V. K. Pecharsky, and C. E. Schindler, J. Alloys Compd. **338**, 126 (2002).
- ²⁶L. Morellon, Z. Arnold, C. Magen, C. Ritter, O. Prokhnenko, Y. Skorokhod, P. A. Algarabel, M. R. Ibarra, and J. Kamarad, Phys. Rev. Lett. **93**, 137201 (2004).
- ²⁷V. O. Garlea, J. L. Zarestky, C. Y. Jones, L.-L. Lin, D. L. Schlagel, T. A. Lograsso, A. O. Tsokol, V. K. Pecharsky, K. A. Gschneidner Jr., and C. Stassis, Phys. Rev. B **72**, 104431 (2005).
- ²⁸M. Zou, V. K. Pecharsky, K. A. Gschneidner Jr., D. L. Schlagel, and T. A. Lograsso, Phys. Rev. B **78**, 014435 (2008).
- ²⁹D. Paudyal, V. K. Pecharsky, K. A. Gschneidner Jr., and B. N. Harmon, Phys. Rev. B **73**, 144406 (2006).
- ³⁰V. K. Pecharsky, G. D. Samolyuk, V. P. Antropov, A. O. Pecharsky, and K. A. Gschneidner Jr., J. Solid State Chem. **171**, 57 (2003).
- ³¹G. D. Samolyuk and V. P. Antropov, J. Appl. Phys. **91**, 8540 (2002).
- ³²G. D. Samolyuk and V. P. Antropov, J. Appl. Phys. **97**, 10A310 (2005).

- ³³B. N. Harmon and V. N. Antonov, J. Appl. Phys. **91**, 9815 (2002).
- ³⁴B. N. Harmon and V. N. Antonov, J. Appl. Phys. **93**, 4678 (2003).
- ³⁵H. Tang, V. K. Pecharsky, G. D. Samolyuk, M. Zou, K. A. Gschneidner Jr., V. P. Antropov, D. L. Schlagel, and T. A. Lograsso, Phys. Rev. Lett. **93**, 237203 (2004).
- ³⁶G. Skorek, J. Deniszczyk, and J. Szade, J. Phys. Condens. Matter 14, 7273 (2002).
- ³⁷D. Paudyal, V. K. Pecharsky, K. A. Gschneidner Jr., and B. N. Harmon, Phys. Rev. B **75**, 094427 (2007).
- ³⁸D. Paudyal, V. K. Pecharsky, and K. A. Gschneidner Jr., J. Phys. Condens. Matter **20**, 235235 (2008).
- ³⁹D. Paudyal, Ya. Mudryk, V. K. Pecharsky, and K. A. Gschneidner Jr., Phys. Rev. B 82, 144413 (2010).
- ⁴⁰Y. Mudryk, D. Paudyal, V. K. Pecharsky, K. A. Gschneidner Jr., S. Misra, and G. J. Miller, Phys. Rev. Lett. **105**, 066401 (2010).
- ⁴¹D. Paudyal, Ya. Mudryk, V. K. Pecharsky, S. Misra, G. J. Miller, and K. A. Gschneidner Jr., J. Appl. Phys. **107**, 09A908 (2010).
- ⁴²Ya. Mudryk, D. Paudyal, V. K. Pecharsky, and K. A. Gschneidner Jr., Phys. Rev. B 77, 024408 (2008).
- ⁴³A. S. Chernyshov, Ya. Mudryk, D. Paudyal, V. K. Pecharsky, K. A. Gschneidner Jr., D. L. Schlagel, and T. A. Lograsso, Phys. Rev. B 80, 184416 (2009).
- ⁴⁴N. K. Singh, D. Paudyal, Ya. Mudryk, V. K. Pecharsky, and K. A. Gschneidner Jr., Phys. Rev. B **79**, 094115 (2009).
- ⁴⁵N. K. Singh, D. Paudyal, Ya. Mudryk, V. K. Pecharsky, and K. A. Gschneidner Jr., Phys. Rev. B **81**, 184414 (2010).
- ⁴⁶N. K. Singh, D. Paudyal, V. K. Pecharsky, and K. A. Gschneidner Jr., J. Appl. Phys. **107**, 09A921 (2010).
- ⁴⁷M. Zou, Ya. Mudryk, V. K. Pecharsky, K. A. Gschneidner Jr., D. L. Schlagel, and T. A. Lograsso, Phys. Rev. B **75**, 024418 (2007).
- ⁴⁸V. K. Pecharsky and K. A. Gschneidner Jr., J. Alloys Compd. 260, 98 (1997).
- ⁴⁹V. I. Anisimov, F. Aryasetiawan, and A. I. Lichtenstein, J. Phys. Condens. Matter. 9, 767 (1997).
- ⁵⁰B. N. Harmon, V. P. Antropov, A. I. Lichtenstein, I. V. Solovyeb, and V. I. Anisimov, J. Phys. Chem. Solids **56**, 1521 (1995).
- ⁵¹O. K. Andersen and O. Jepsen, Phys. Rev. Lett. **53**, 2571 (1984).
- ⁵²S. Lebègue, A. Svane, M. I. Katsnelson, A. I. Lichtenstein, and O. Eriksson, J. Phys. Condens. Matter **18**, 6329 (2006).
- ⁵³U. von Barth and L. Hedin, J. Phys. C 5, 1629 (1972).
- ⁵⁴R. M. White, *Quantum Theory of Magnetism*, 3rd ed. (Springer-Verlag, Berlin, Heidelberg, 2007), Chap. 9, p. 286.
- ⁵⁵M. S. S. Brooks and B. Johansson, in *Hand Book of Magnetic Materials*, edited by K. H. J. Buschow (Elsevier Science, North-Holland, 1993), Vol. 7, Chap. 3, p. 139.
- ⁵⁶J. Jensen and A. R. Mackintosh, *Rare Earth Magnetism Structures and Excitations* (Clarendon Press, Oxford, 1991), Chap. 1, p. 45.
 ⁵⁷M. Richter, J. Phys. D: Appl. Phys. **31**, 1017 (1998).
- ⁵⁸R. Ahuja, S. Auluck, B. Johansson, and M. S. S. Brooks, Phys. Rev.
- B **50**, 5147 (1994). ⁵⁹F. Gerken and Schmidt-May, J. Phys. F: Met. Phys. **13**, 1571 (1983).

⁶⁰K. W. H. Stevens, Proc. Phys. Soc. A **65**, 209 (1952).

- ⁶¹Y. C. Tseng, H. J. Ma, C. Y. Yang, Y. Mudryk, V. K. Pecharsky, K. A. Gschneidner Jr., N. M. Souza-Neto, and D. Haskel, Phys. Rev. B 83, 104419 (2011).
- ⁶²A. P. Holm, V. K. Pecharsky, and K. A. Gschneidner Jr. (unpublished).
- ⁶³R. Nirmala, Ya Mudryk, V. K. Pecharsky, and K. A. Gschneidner Jr., Phys. Rev. B **76**, 104417 (2007).
- ⁶⁴V. K. Pecharsky, A. O. Pecharsky, Y. Mozharivskyj, K. A. Gschneidner Jr., and G. J. Miller, Phys. Rev. Lett. **91**, 207205 (2003).
- ⁶⁵A. B. Shick, W. E. Pickett, and C. S. Fadley, Phys. Rev. B **61**, R9213 (2000).
- ⁶⁶Ph. Kurz, G. Bihlmayer, and S Blügel, J. Phys. Condens. Matter 14, 6353 (2002).
- ⁶⁷C. P. Bean and D. S. Rodbell, Phys. Rev. **126**, 104 (1962).