Short-range order in iron-rich Fe-Cr alloys as revealed by Mössbauer spectroscopy

S. M. Dubiel^{*} and J. Cieslak

Faculty of Physics and Applied Computer Science, AGH University of Science and Technology, aleja Mickiewicza 30,

PL-30-059 Krakow, Poland

(Received 26 October 2010; revised manuscript received 4 April 2011; published 19 May 2011)

The distribution of Cr atoms in Fe_{100-x}Cr_x alloys with $x \le 25$ within the first two coordination shells, 1NN–2NN, around probe ⁵⁷Fe atoms was studied using Mössbauer spectroscopy. Clear evidence was found that the distribution is not random but instead characteristic of a given atomic configuration (m,n) (*m* being the number of Cr atoms in 1NN and *n* in 2NN). The behavior was described quantitatively in terms of average short-range order (SRO) parameters $\langle \alpha_1 \rangle$ (for 1NN), $\langle \alpha_2 \rangle$ (for 2NN), and $\langle \alpha_{12} \rangle$ (for 1NN–2NN), as well as in terms of a local SRO parameter $\alpha(m,n)$ for each pair (m,n). A change of sign (inversion) was found both in $\langle \alpha_1 \rangle$ and in $\langle \alpha_2 \rangle$, though going with *x* in opposite directions. No inversion was observed in $\langle \alpha_{12} \rangle$, which was either positive or negative depending on the metallurgical state of the samples. These findings prompt a revision of current interpretation of experimental and theoretical results relevant to the issue.

DOI: 10.1103/PhysRevB.83.180202

PACS number(s): 75.40.-s, 76.80.+y, 81.30.Hd

Among various binary alloys of iron, Fe-Cr alloys occupy a special role for both scientific and technological reasons. They can be treated as good systems for testing models and theories, especially those relevant to magnetism where different phases are exhibited depending on alloy composition.¹ Their crystallographic structure, which for many years was regarded as homogenous bcc over the whole concentration range, turned out to be much more complex when a tetragonal σ phase and a miscibility gap were discovered. The latter two phenomena are, on one hand, of interest per se and have been the subject of intensive study,² and on the other hand, are also of a great importance technologically, namely in the production of important grades of stainless steels³ for which the Fe-Cr alloys are the basic ingredient. Consequently, their useful properties such as a good resistance to high-temperature corrosion and good mechanical properties (toughness, ductility, and welding ability) may be severely degraded if the σ phase precipitates or phase separation into Fe-rich and Cr-rich phases occurs.

Recently there has been increased interest in Fe-Cr alloys. This is driven both by the discovery of giant magnetoresistance in Fe/Cr layers⁴ and also by the potential for Fe-Cr-based steels to be used in the construction of a new generation of power plants (advance fusion and fission reactors or high-power accelerator spallation targets).⁵ In the latter application the materials undergo irradiation damage which can seriously degrade their mechanical properties. On the lattice scale the radiation causes lattice defects and, consequently, a redistribution of Fe/Cr atoms that can result in a short-range order (SRO) or phase decomposition into Fe-rich and Cr-rich phases.

According to previous neutron diffraction (ND) studies,^{6,7} the Cowley SRO parameter $\langle \alpha_{12} \rangle$ was found to change its sign at $x \approx 10-11$. This finding was qualitatively confirmed by a Mössbauer spectroscopic (MS) study,⁸ yet the value of the critical concentration was not determined. Additionally, theoretical calculations predicted the existence of such an inversion, but for different values of x.^{9,10} The aim of the present investigation was to study the issue in more detail using MS since this method applied to the Fe-Cr alloys can provide

precise and relevant information on SRO for each statistically meaningful atomic configuration (m, n), where *m* is the number of Cr atoms in the first-nearest-neighbor shell (1NN) and *n* is the number in the second-nearest-neighbor shell (2NN).^{11,12} Such information would be much more detailed than the one recently found with ND,⁷ where the inversion of the SRO parameter was found as the average over the 1NN–2NN volume.

There are 63 different atomic configurations possible for the bcc structure within such volume. Although for a random distribution the probability of most of them [P(m,n)] is very small, all those with $P(m,n) > \sim 0.01$ are measurable using MS. This improvement over the information available by ND techniques means that MS can be used as a more adequate basis for quantitative verification of different theoretical models pertinent to the issue.^{9,10,13,14}

⁵⁷Fe Mössbauer spectra were recorded at 295 K in transmission mode on four series of $Fe_{100-x}Cr_x$ alloys (I, II, IIIa, and IIIb) with different histories and composition using a standard spectrometer with a sinusoidal drive and a ⁵⁷Co/Rh source of 14.4 keV γ rays.

Samples of series I, with $0 \le x \le 15$, were 40 years old. They were prepared as follows: Armco iron and 4N-purity chromium were melted in a vacuum induction furnace. After melting they were kept in a liquid state for about 10 min and cooled down to an ambient temperature. The ingots were next forged into flat bars (8-mm thick) which were subsequently cold rolled into 2-mm-thick tapes. The tapes were annealed in a vacuum at 840 °C for 1 h and then cooled in a furnace. The 2-mm-thick tapes were next rolled down to a thickness of 0.1 mm from which 20–30- μ m-thick foils were obtained again by cold rolling. Samples of series II, 3 years old and with $15 \le x \le 25$ were prepared in a similar way. Examples of the spectra recorded on these samples are shown in Fig. 1.

The series IIIa and IIIb samples were EFDA/EURATOM model Fe-Cr alloys that had been prepared in 2007. They were delivered in the form of bars 10.9 mm in diameter in a recrystallized state after cold reduction of 70% and then heat treated for 1 h under pure Ar flow at the following



FIG. 1. Examples of ⁵⁷Fe Mössbauer spectra recorded at 295 K on Fe-Cr alloys with various Cr content in at % (3 = 3.25, 5 = 4.85, 10 = 10.25, 15 = 14.9, 21 = 21.0, 25 = 25.0). The solid lines are equations of best fit.

temperatures: 750 $^{\circ}$ C for Fe_{94.4}Cr_{5.6}, 800 $^{\circ}$ C for Fe_{89.75}Cr_{10.25}, and 850 °C for Fe₈₅Cr₁₅ followed by air cooling. For the MS measurements a slice \sim 1-mm thick was cut off from each bar using a diamond saw, and was subsequently cold rolled down to a final thickness of 20–30 μ m. Samples obtained in this way constituted the series IIIa. Some of the latter in the form of 20-mm-diameter circular foils were annealed at 800 °C for 4 h under Ar flow followed by a liquid nitrogen quenching. The samples that underwent this heat treatment constituted series IIIb. All the spectra were analyzed in the same way, that is, with the two-shell model. It was assumed that only Cr atoms situated within the 1NN and 2NN neighbor shells caused measurable changes in spectral parameters, that is, the hyperfine field B and the isomer shift IS. It was also assumed that the changes both in B and in IS were additive, that is, $X(m,n;x) = X(0,0;x) - m\Delta X_1 - n\Delta X_2$, where X = B or IS and ΔX_{12} stands for the change in X due to one Cr atom in 1NN (index 1) or in 2NN (index 2).

This procedure proved to be successful in the analysis of the spectra of various Fe-rich Fe-X alloys (e.g., Ref. 15). Each spectrum was treated as composed of a number of subspectra N corresponding to a particular atomic configuration (m,n). Its relative spectral area was equal to the probability of the atomic configuration associated with the spectrum P(m,n).

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FIG. 2. (Color online) Probabilities of various atomic configurations P(m,n) around the probe ⁵⁷Fe atoms as calculated for a random distribution (solid lines) and as derived from the spectra (open triangles series I and II, open circles series IIIa, and full circles series IIIb).

The latter is the relevant quantity for a quantitative description of a real atoms distribution over lattice sites in a given sample. Theoretically the probabilities for the random case $P_r(m,n;x)$ can be calculated from the formula

$$P_r(m,n;x) = \binom{8}{m} \binom{6}{n} x^{m+n} (1-x)^{14-m-n}.$$
 (1)

Most of the possible 63 configurations have vanishingly small probabilities and can therefore be neglected. In practice one usually takes into account only the most probable ones in order to fulfill the condition $\sum P(m,n;x) > 0.99$. This condition significantly reduces N from 63 to, for example, N = 4(x = 1), N = 10 (x = 10), N = 14 (x = 15). Using the above described procedure, we have successfully fitted all the measured spectra with the following values of the spectral parameters: $\Delta B_1 = 31.0 \pm 0.5$ kOe, $\Delta B_2 = 21.3 \pm 0.6$ kOe, $\Delta IS_1 =$ -0.022 ± 0.001 mm/s, and $\Delta IS_2 = -0.009 \pm 0.001$ mm/s which agree well with those previously reported.^{11,12} The values of P(m,n) determined from the analysis for the most significant six configurations are presented in Fig. 2 together with the corresponding ones calculated from formula (1). It is clear that the actual distribution is, in general, not random and the degree and direction of deviation from randomness is characteristic of a given atomic configuration. In particular, the P(0,0) values are close to the $P_r(m,n)$ ones for all x values, though those determined for series IIIb show a systematic



FIG. 3. (Color online) SRO parameters $\alpha(m,n)$ for various atomic configurations vs Cr content *x* as calculated using formula (2). Open triangles stand for series I and II, open circles for series IIIa, and full circles for series IIIb.

deviation. The distribution of atoms is partly random for (0,1), (1,0), and (2,0) configurations, and nonrandom for (1,1) and (0,2) configurations. In order to quantitatively describe the actual departure from the randomness for (m,n), we introduce the following measure for the short-range order:

$$\alpha(m,n) = \frac{P(m,n)}{P_r(m,n)} - 1.$$
 (2)

This SRO parameter can be regarded as an adequate measure for the departure of the actual distribution of atoms from the random distribution, and its value can be easily determined based on the spectral parameters. When the actual probability of finding an (m,n) atomic configuration around the probe Fe atom is higher than the one for the random distribution (atomic short range order) $\alpha(m,n) > 0$, and $\alpha(m,n) < 0$ otherwise (clustering). In our opinion this definition is simple and justified, at least from the viewpoint of MS, as it ascribes a positive value of α to a larger number of Cr atoms within the 1NN–2NN shell, as seen by the probe Fe atoms, than the one expected for the random case and a negative value otherwise.

Values of $\alpha(m,n)$ obtained from formula (2) for all four series are presented in Fig. 3. Here it is clearly evident that the $\alpha(m,n)$ are characteristic of a given (m,n) and that they also depend on the samples' histories. It is evident then that the actual distribution of atoms in the Fe-Cr system is much more complex than the one obtained from the ND experiments.^{6,7}





FIG. 4. (Color online) The average SRO parameters for the 1NN shell $\langle \alpha_1 \rangle$, for the 2NN shell $\langle \alpha_2 \rangle$, and that for the 1NN–2NN shells $\langle \alpha_{12} \rangle$ vs Cr concentration *x* as calculated using formula (3) (left-hand panel) and using formula (4) (right-hand panel). Open triangles stand for series I and II, open circles for series IIIa, and full circles for series IIIb.

The inversion of the SRO parameter at $x \approx 10$ can locally [i.e., in terms of (m,n)] only be seen for the following configurations: (1,0) series I and IIIa, (2,0) series IIIa and IIIb, and perhaps (0,0) series IIIa. The opposite inversion takes place at $x \approx 3$ in $\alpha(0,2)$. On the other hand, $\alpha(1,1) < 0$ over the whole concentration range showing a saturation behavior. It is also worth noticing that $\alpha(0,0) \approx 0$ for $x < \sim 10$, but only for the cold-rolled samples.

An important issue is the effect of heat treatment on the distribution. To get some insight one can compare the $\alpha(m,n)$ calculated for series IIIa and IIIb. Interestingly, for some configurations viz. (0,0) and (1,0) they are significantly different, while for other viz. (0,1), (2,0), and (1,1) they are similar. In these circumstances it seems reasonable to also introduce average values of α to the description of the actual distribution of Cr atoms in the studied samples. Thus the average α for the 1NN shell $\langle \alpha_1 \rangle$, that for the 2NN shell $\langle \alpha_{22} \rangle$, and also the average for the 1NN–2NN shells $\langle \alpha_{12} \rangle$ can be defined as follows:

$$\langle \alpha_i \rangle = \frac{\langle k \rangle}{\langle k_r \rangle} - 1,$$
 (3)

where k = m, n, m + n for i = 1, 2, 12, respectively, and $\langle m \rangle$ is the average number of Cr atoms in 1NN, $\langle n \rangle$ is that in 2NN, and $\langle m + n \rangle$ is that in 1NN–2NN as determined from

analysis of the Mössbauer spectra. The three symbols with subscript *r* represent the same quantities but calculated for the random distribution. A graphical illustration of $\langle \alpha_1 \rangle$, $\langle \alpha_2 \rangle$, and $\langle \alpha_{12} \rangle$ is displayed in Fig. 4 on the left-hand side of the panel. Alternatively, following Wittle and Campbell¹⁶ and staying with the 1NN–2NN model and bcc structure, one can define $\langle \alpha_1 \rangle$, $\langle \alpha_2 \rangle$, and $\langle \alpha_{12} \rangle$ as follows:

$$\langle \alpha_i \rangle = \frac{\langle k \rangle - lx}{l(1-x)} - 1,$$
 (4)

where k = m, n, m + n and l = 8, 6, 14 for i = 1, 2, 12, respectively. It should be remembered that both definitions of the average α differ in sign in comparison to that introduced by Cowley.

The average SRO parameters obtained using formula (4) are presented on the right-hand side panel of Fig. 4. Quantitative agreement between the corresponding average SRO parameters obtained with the two approaches can be readily seen. Concerning the crucial question of the inversion, one can definitely observe its existence both in $\langle \alpha_1 \rangle$ and $\langle \alpha_2 \rangle$, especially in the samples of series IIIa and IIIb. However, the inversions go in opposite directions: on increasing x one observes a change from ordering to clustering in the former and a change from clustering to ordering in the latter. The critical concentration at which the inversion occurs depends on the samples' histories. As a consequence of such behavior the SRO parameter averaged over the 1NN–2NN volume $\langle \alpha_{12} \rangle$ does not show any inversion: the one for series IIIa is positive, hence revealing the short-range ordering with a maximum at $x \approx 10$, while that for series IIIb is negative, hence indicating the clustering effect. Similar effect was revealed for a series of Au-Fe where the alloys were found to exhibit either clustering of Fe atoms or atomic short-range ordering depending on their metallurgical state and heat treatment.¹⁷ The presently reported behavior is completely different than the one found with ND.^{6,7}

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To summarize, the distribution of Cr atoms in the 1NN-2NN shells was studied quantitatively using Mössbauer spectroscopy on the level of atomic configurations in four series of Fe-Cr having different metallurgical states. Clear evidence was found that the actual distribution is much more complex than the one accepted to date from ND.^{6,7} In particular, it was shown here that the change of the SRO-parameter sign observed with ND at 10-11 at % Cr and regarded as an experimental confirmation of various theoretical calculations including SRO itself,¹⁰ is exclusively a feature of the 1NN shell. The average SRO parameter for the 2NN shell was also found to exhibit inversion but in the opposite direction, that is, from clustering to ordering. Consequently, the SRO parameter averaged over the two shells $\langle \alpha_{12} \rangle$ does not show any inversion and its actual value depends on the metallurgical state of the samples: for the cold-rolled ones its is positive, hence indicative of the atomic short-range ordering, whereas for the quenched samples it is negative, indicating the existence of clustering. In other words, the actual distribution of atoms in the Fe-Cr alloys is very sensitive to their metallurgical state. This in turn reflects the fact that the initial state of these alloys is metastable. Upon heating the alloys decompose into Fe-rich and Cr-rich phases. The degree of the decomposition, hence the actual distribution of atoms (and values of the SRO parameters) depends on samples' metallurgical histories as experimentally revealed in this study, and theoretically demonstrated by performing Monte Carlo atomistic simulations.¹⁸

This work, supported by the European Communities under the contract of Association between EURATOM and IPPLM, was carried out within the framework of the European Fusion Development Agreement, and it was also supported by the Ministry of Science and Higher Education, Warszawa. J. Aguilar is thanked for his help in improving the language.

- *Corresponding author: dubiel@novell.ftj.agh.edu.pl
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