Phase stability and structural temperature dependence in sodium niobate: A high-resolution powder neutron diffraction study

S. K. Mishra,¹ R. Mittal,¹ V. Yu. Pomjakushin,² and S. L. Chaplot¹

¹Solid State Physics Division, Bhabha Atomic Research Centre, Mumbai 400085, India

²Laboratory for Neutron Scattering, ETH Zurich and Paul Scherrer Institut, CH-5232 Villigen PSI, Switzerland

(Received 19 November 2010; revised manuscript received 9 February 2011; published 7 April 2011)

We report an investigation of structural phase transitions in the technologically important material sodium niobate as a function of temperature on heating over 300–1075 K. Our high-resolution powder neutron diffraction data show a variety of structural phase transitions ranging from nonpolar antiferrodistortive to ferroelectric and antiferroelectric in nature. A discontinuous jump in lattice parameters is found only at about 680 K. This observation along with the reported thermal hysteresis in the dielectric anomaly indicate that the transition of the orthorhombic *P* to *R* phase is first order in nature, while other successive phase transitions are of second order. Additional superlattice reflections appear at 680 K (*R* phase) and 770 K (*S* phase) that could be indexed using an intermediate long-period modulated orthorhombic structure whose lattice parameter along the $\langle 001 \rangle$ direction is three and six times, respectively, that of the CaTiO₃-like *Pbnm* structure. The correlation of superlattice reflections with the phonon instability is discussed. The critical exponent (β) for the second-order tetragonal to cubic phase transition at about 950 K corresponds to a value $\beta \approx 1/3$, as obtained from the temperature variation of order parameters (tilt angle and intensity of superlattice reflections). It is argued that this exponent is due to a second-order phase transition close to a tricritical point. Based on our detailed temperature-dependent neutron diffraction studies, a phase diagram of sodium niobate is presented that resolves existing ambiguities in the literature.

DOI: 10.1103/PhysRevB.83.134105

PACS number(s): 77.80.-e, 61.05.fm, 77.84.Ek

I. INTRODUCTION

The past few decades have seen a tremendous flurry of research interest in perovskite materials due to their fundamental interest and technological applications.^{1–6} In this class of materials, alkali-metal niobates (like potassium sodium niobate and lithium sodium niobate) with ultralarge piezoresponse comparable to Pb($Zr_{1-x}Ti_x$)O₃ (PZT) have evoked considerable interest as the next-generation ecofriendly lead-free piezoceramics.^{3–6} One of the end members, NaNbO₃, is a well-documented antiferroelectric which finds applications in high-density optical storage, enhancing nonlinear optical properties, as hologram recording materials, etc.^{1,3–6} An understanding of its crystallographic structures and phase transitions is of fundamental interest due to its functional properties such as enhanced ferroelectric and piezoelectric response.

NaNbO₃ exhibits an unusual complex sequence of temperature-, pressure-, and particle-size-driven structural phase transitions.⁷⁻¹⁸ Above 913 K, NaNbO₃ has a paraelectric cubic phase with space group $Pm\bar{3}m$. On cooling, it undergoes a series of antiferrodistortive phase transitions ranging from a paraelectric cubic phase to a ferroelectric rhombohedral phase via intermediate paraelectric tetragonal and orthorhombic phases and an antiferroelectric orthorhombic phase (see Fig. 1 of Ref. 7). These temperature-induced structural phase transitions are driven by the so-called soft-mode mechanism. This concept is built on the assumption that the crystal gets unstable against particular Brillouin zone center and/or boundary phonon vibration modes.¹⁹ The zone boundary phonon vibrations involve rotations of NbO₆ octahedra. The M_3 (q = 1/2, 1/2, 0) mode represents in-phase rotation of the adjacent layers (in the same direction $a^0a^0c^+$ in Glazer notation), while the R_{25} (q = 1/2, 1/2, 1/2) mode represents the antiphase rotation (opposite manner, $a^0a^0c^-$ in Glazer notation).²⁰ In soft-mode formalism, cubic to tetragonal $(T_2 \text{ phase})$ and tetragonal to orthorhombic (T_1) phase transitions are driven by M_3 and R_{25} phonon modes, respectively. The diffuse x-ray scattering at *R* and *M* points into the Brillouin zone is found to be strongly anisotropic and distributed along $\langle 100 \rangle$ reciprocal rods which can be reasonably explained by assuming a contribution of the low- frequency zone boundary phonons along the line *T* (lying between the *M* and *R* points). The phase transitions between the orthorhombic *S*, *R*, and *P* phases^{9–11} have not been investigated yet, and substantial enlargements of the unit cell and complicated octahedral tilting are not understood.

Further, NaNbO₃ exists in a ferroelectric phase at 15 K. The low-temperature transition is driven by zone center and zone boundary (R_{25}) modes. Our recent low-temperature neutron diffraction studies provided unambiguous experimental evidence for the presence of coexisting ferroelectric (R3c) and antiferroelectric phases (Pbcm) over a wide range of temperatures (15–275 K) in NaNbO₃.⁷ The ferroelectric ordering can be stabilized by application of external field, chemical substitution, and reduction of particle size even at room temperature.^{6,8} The high-temperature phases of ferroelectric and antiferroelectric materials are of primary interest since the dynamics of the dipoles in these phases gives indication about the microscopic nature (e.g., order-disorder versus displacive) of the ferroelectricity itself.

In order to understand the complexity of the structural phase transition sequence in NaNbO₃, first-principles calculations were carried out by Vanderbilt and co-workers.^{14–18} The calculations suggest the presence of several competing structural instabilities with similar free energies. The authors included all the distortions involved in the observed structures of NaNbO₃ in their model calculations. The calculations predicted the coexistence of zone center and zone boundary phonon instabilities in the cubic phase. However, they could



FIG. 1. Evolution of the neutron diffraction patterns for NaNbO₃ as a function of temperature. The characteristic superlattice reflections are marked with arrows and labeled as (S1, S2, S3, S4, and S5). Peak marked with *F* is due to the furnace materials.

not fully reproduce the observed sequence of structural phase transitions in NaNbO₃. The structural and dynamical properties, epitaxial strain, and anomalous LO-TO splitting of NaNbO₃ have also been reported.

In spite of the extensive experimental and theoretical studies as a function of temperature, there are numerous controversies surrounding the phase diagram of NaNbO₃, especially on the existence of ferroelectric ordering and correctness of the structures of the phases in different temperature regions. Darlington and Knight⁹ have questioned the correctness of the structure at room temperature. They have shown that space group and cell dimensions for the high-temperature *R* and *S* phases (stabilized between 633 and 753 K and 753 and 793 K) as reported in the literature^{9–11} are incorrect. Recently, on the basis of work carried out using synchrotron x-ray diffraction, Yuzyuk *et al.*⁶ proposed a different sequence of structural phase transitions in NaNbO₃. However, the correct crystal structures for the two high-temperature *R* and *S* phases were still not determined.

The nature of antiferrodistortive phase transition in materials exhibiting perovskite structure has attracted enormous attention in the literature. Mainly, these phase transitions are driven by *R* point instabilities. We have carried out systematic neutron diffraction measurements, as a function of temperature from 300 to 1075 K, to find the correct crystal structures for high-temperature phases of NaNbO₃ and to understand their phase transition behavior. To the best of our knowledge, studies have not been carried out to investigate the nature of antiferrodistortive phase transitions driven by *M* point instabilities. We have also determined the effective exponent for the second-order cubic to tetragonal phase transition using the intensity of the superlattice reflection and tilt angle. The critical exponent (β) is found to be nearly equal to a value $\beta \approx 1/3$. It is argued that this exponent is due to a nearly tricritical transition.

II. EXPERIMENT

The powder neutron diffraction experiments were carried out at the SINO spallation source of Paul Scherrer Institute (Switzerland) using the high-resolution diffractometer²¹ for thermal neutrons (HRPT) with wavelength 1.4940 Å. The data were collected under high-intensity and high-resolution mode with $\Delta d/d \ge 1.8 \times 10^{-3}$. All the data collections were carried out during heating cycles of the sample. The sample was heated up to 1075 K using a high-temperature tantalum furnace. The structural refinements were performed using the Rietveld refinement program FULLPROF.²² In all the refinements, the background was defined by a sixth-order polynomial in 2θ . A Thompson-Cox-Hastings pseudo-Voigt function with axial divergence asymmetry function was chosen to define the profile shape for the neutron diffraction peaks. Except for the occupancies of the atoms, all other parameters, i.e., scale factor, zero correction, background, and half-width parameters along with mixing parameters, lattice parameters, positional coordinates, and thermal parameters, were refined. All the refinements have used the data over the full angular range of $4^{\circ} \leq 2\theta \leq 164^{\circ}$; although in various figures only a limited range is shown for clarity.

III. RESULTS AND DISCUSSION

A. Evolution of neutron diffraction as a function of temperature

Figure 1 depicts a portion of the powder neutron diffraction patterns of NaNbO₃ as a function of temperature in the range from 300 to 1075 K during the heating cycle. We shall discuss the structures starting from the highest-symmetry cubic phase that occurs at the highest temperature followed by the other phases occurring at lower temperatures. At the highest temperature (T = 1075 K), all the Bragg reflections present in powder diffraction patterns could be indexed as main cubic perovskite reflections. The reflection marked with an arrow and labeled as $F(2\theta = 35.1^{\circ})$ in the diffraction pattern of cubic phase is from the furnace material [see Fig. 1(b)]. Below 950 and 900 K, two different superlattice reflections appear at $2\theta = 35.0^{\circ}$ and 37.0° (near to *F*), respectively. These superlattice reflections are associated with zone boundary Mand R point instabilities, respectively. Further, below 810 K, an additional set of superlattice reflections appears centered at $2\theta = 35.4^{\circ}$ [marked with an arrow and label as S5 in Fig. 1(b)], and some of them diminish followed by enhancement of the intensity of the superlattice reflection at $2\theta = 35.8^{\circ}$ [marked with arrows and labeled as S3 in Fig. 1(a)], which disappears below 680 K. Below 680 K, the superlattice reflections present at $2\theta = 35.0^{\circ}$, 35.8° , and 41.7° [marked with arrows and labeled as S2, S3, and S4 in Fig. 1(a)] vanish and different superlattice reflections appear around $2\theta = 35.6^{\circ}$ and 53.5° [marked with an arrow and labeled as S1 in Fig. 1(a)]. The intensity of these reflections increases gradually with decrease of temperature and reflections are present at room temperature.

 $NaNbO_3$ has an antiferroelectric phase (*Pbcm*) at room temperature. In general, an antiferroelectric phase consists of two or more sublattice polarizations of antiparallel nature,

]	Positional coordinate	es	Thermal parameter (10^{-2} Å^2)			
Atoms	X	Y	Z	<i>U</i> 11	U22	<i>U</i> 33	
Na	0	0	0	7.32(9)	7.32(9)	7.32(9)	
Nb	0.5	0.5	0.5	1.96(4)	1.96(4)	1.96(4)	
0	0.5	0.5	0	7.40(8)	7.40(8)	1.94(6)	

TABLE I. Structural parameters of NaNbO₃ obtained by neutron diffraction at 1075 K. The cubic lattice parameter is 3.9507(2) Å in the space group $Pm\bar{3}m$. The unwritten thermal parameters U_{ij} are zero.

which in turn give rise to superlattice reflections in the diffraction pattern. Superlattice reflections around $2\theta = 35.6^{\circ}$ and 53.5° are some of the strongest antiferroelectric peaks and are marked with arrows and labeled as S1 in Fig. 1(a). Thus, the disappearance and reappearance of superlattice reflections in powder neutron diffraction provide unambiguous evidence for structural phase transitions with temperature in sodium niobate. Superlattice reflections, in different temperature regimes, are also present at higher angles with prominent intensities. The details of symmetry and structure in different temperature regime are discussed in the next section.

B. Rietveld analysis of powder neutron diffraction data

1. Cubic (U) phase (above T > 950 K)

The powder neutron diffraction patterns above 950 K are analyzed on the basis of an ideal cubic perovskite structure with the space group $Pm\bar{3}m$. The detailed structural parameters and goodness of fit for the cubic phase of sodium niobate at 1075 K, as obtained from neutron diffraction data, are given in Table I. The fit between the observed and calculated profiles is satisfactory [Fig. 2(a)]. Refinement with anistropic atomic displacement parameters gives a better fit. Strongly anisotropic thermal vibrations of oxygen atoms, with the site symmetry 4/mmm, are observed. In the cubic phase, as seen in Table I, the oxygen atom at (0.5,0.50) has $U_{11} = U_{22} \gg U_{33}$, $U_{ij} = 0$, and $U_{33} \approx U(Nb)$. This signifies large-amplitude libration motion of the NbO₆ octahedra, and suggests structural phase transitions driven by octahedral librational M_3 and R_{25} modes at lower temperatures.²³

2. Tetragonal (T_1) and orthorhombic (T_2) phases (temperature range 825 < T < 950 K)

The presence of superlattice reflections at $2\theta = 35.0^{\circ}$ and 37.0° below 950 and 865 K, respectively, in powder neutron diffraction patterns indicates the structural phase transitions from the cubic phase. These superlattice reflections arise due to condensation of *M* and *R* zone boundary phonons and characteristics of the tetragonal and orthorhombic phases, respectively. Thus, for the temperature ranges 865 < T



FIG. 2. (Color online) Observed (open circles), calculated (continuous line), and difference (bottom line) profiles obtained after the Rietveld refinement of NaNbO3 using (a) cubic $(Pm\bar{3}m)$, (b) tetragonal (P4/mbm), and (c) orthorhombic (Cmcm) phases. Peak marked with F is due to the furnace materials. In the cubic phase, middle and lower vertical tick marks above the difference profiles correspond to the furnace and thermocouple materials, respectively. The R factors obtained by Rietveld refinement of neutron diffraction data are $R_B = 4.79$; $R_p = 4.74$; $R_{wp} = 6.56; R_{expt} = 2.89; \chi^2 = 5.15$ for tetragonal (P4/mbm) and $R_B =$ 4.93; $R_p = 4.24$; $R_{wp} = 6.72$; $R_{expt} =$ 2.93; $\chi^2 = 5.26$ orthorhombic (*Cmcm*) phases.

TABLE II. Structural parameters of NaNbO₃ obtained by Rietveld refinement of neutron diffraction at 770 (S phase) and 680 K (R phase) using *Pbnm* space group.

	(a) Temperature = $770 \text{ K} (S \text{ phase})$					(b) Temperature = $680 \text{ K} (R \text{ phase})$				
	Positional coordinates			Thermal parameter		Positional coordinates			Thermal parameter	
Atoms	X	Y	Ζ	B (Å) ²	Atoms	X	Y	Ζ	B (Å) ²	
Na1	0.5260(2)	-0.0073(7)	0.0410(7)	0.710(1)	Na1	0.5138(1)	0.4984(1)	0.2500	2.185(3)	
Na2	0.5299(9)	-0.0315(7)	0.1182(9)	2.026(6)	Na2	0.9863(2)	0.0109(3)	0.0872(1)	2.990(1)	
Na3	0.5107(5)	-0.0146(5)	0.2092(2)	3.132(6)	Nb1	0.5000	0.0000	0.0000	1.013(1)	
Nb1	0.0000	0.0000	0.0000	0.844(4)	Nb2	0.4981(1)	0.0041(2)	0.3323(2)	0.868(1)	
Nb2	0.0068(8)	0.0065(2)	0.0822(2)	0.878(3)	01	0.5315(1)	-0.0267(1)	0.2500	1.512(2)	
Nb3	-0.0100(5)	-0.0005(1)	0.1680(6)	1.022(4)	O2	0.0597(2)	0.4824(2)	0.0835(1)	2.684(4)	
Nb4	0.0017(3)	-0.0012(4)	0.25000	1.082(3)	O3	0.7553(1)	0.2442(3)	0.0040(2)	2.210(5)	
01	0.0520(4)	-0.0007(7)	0.0415(1)	3.225(3)	O4	0.7070(3)	0.2932(1)	0.3391(4)	0.453(3)	
O2	0.0368(4)	0.0557(6)	0.1243(1)	2.636(4)	05	0.7151(2)	0.2885(3)	0.6734(3)	0.573(4)	
O3	-0.0207(5)	0.0005(4)	0.2083(3)	0.773(3)						
O4	0.1990(7)	0.2903(6)	0.0028(6)	0.001(1)						
05	0.2663(2)	0.2358(1)	0.0817(8)	1.811(2)						
06	0.7290(1)	0.7636(3)	0.0827(1)	1.977(2)						
O7	0.2028(1)	0.2851(1)	0.1692(4)	0.541(1)						
08	0.7992(2)	0.6980(5)	0.1692(3)	1.062(2)						
09	0.2055(3)	0.2737(5)	0.2500	2.482(3)						
O10	0.7792(5)	0.7163(1)	0.2500	0.668(3)						
Lattice parameters (Å) A = 5.5555(2) (Å); $B = 5.5556(2)$ (Å)					Lattice parameters (Å) A = 5.5459(1) (Å); $B = 5.5505(1)$ (Å)					
C = 47.1489(9): volume = 1455.230(8) (Å) ³					C = 23.5229(3); volume = 723.087(5) (Å) ³					
$R_B = 5.84; R_p = 10.2; R_{wp} = 10.4; R_{expt} = 4.01;$ $\chi^2 = 6.77$						$R_B = 6.85; R_p = 10.3; R_{wp} = 10.35; R_{exp} = 4.05; \chi^2 = 6.62$				

< 950 K and 810 < T < 865 K we refined the powder neutron diffraction data using tetragonal symmetry (space group P4/mbm) and orthorhombic symmetry (space group Cmcm), respectively. The results of Rietveld refinements are shown in Figs. 2(b) and 2(c). We note that the contamination due to the furnace material decreased in the lower temperature range.

3. Orthorhombic S phases (temperature range 755 < T < 825 K)

The main feature of this phase is the appearance of additional superlattice reflections beyond those in the T_2 phase at $2\theta = 35.4^{\circ}$ and 35.8° . These superlattice reflections, marked with arrows and label as S_5 in Fig. 1(b), could not be indexed using the symmetry and space group of the orthorhombic (T_2) phase. The absence of splitting in the main perovskite reflections and the presence of different superlattice reflections suggest that the multiplicity of the cell is increased with respect to the orthorhombic T_2 phase. The x-ray diffraction measurements by Ahtee *et al.*¹⁰ showed that the *S* phase has orthorhombic symmetry with *Pnmm* space group and cell dimension $2 \times 2 \times 2$ with respect to the elementary perovskite cubic cell. However, Rietveld refinement of neutron diffraction data using the model proposed by Ahtee *et al.*¹⁰ did not account for the superlattice reflection at $2\theta = 35.4^{\circ}$.

In order to index the additional superlattice reflection, we explored various possibilities and found that further multiplication of the lattice parameter along [001] of the orthorhombic R phase (see next section) indexed all the reflections. Thus the

S phase has orthorhombic structure with space group *Pbnm* and cell dimensions $\sqrt{2} \times \sqrt{2} \times 12$ with respect to the elementary perovskite cell. This unit cell indicates that the structure of the *S* phase of NaNbO₃ is a modulated structure in which the lattice parameter along [001] is twice the *C*₀ lattice parameter of the *R* phase.

In the space group *Pbnm*, for the description of the crystal structure, we used the following axes: $\mathbf{A}_0 = -\mathbf{a}_p + \mathbf{b}_p$, $\mathbf{B}_0 = \mathbf{a}_p + \mathbf{b}_p$, and $\mathbf{C}_0 = 12\mathbf{c}_p$, where a_p, b_p , and c_p correspond to the parent cubic phase perovskite. The orthorhombic unit cell contains 24 formula units. The detailed structural parameters and goodness of fit for the *S* phase of sodium niobate at 770 K, as obtained from neutron diffraction data, are given in Table II(a). The fit between the observed and calculated profiles is quite satisfactory and includes the weak superlattice reflections were also seen by Darlington and Knight⁹ using high-resolution neutron data in term of *d* spacing (Å) [see Fig. 3(b)].

4. Analysis of R phase (temperature range 650 < T < 770 K)

In the *R* phase, the characteristic superlattice reflections appear at $2\theta = 35.0^{\circ}$, 35.8° , 37° , and 41.7° , etc. We note that the superlattice reflection in the *S* phase at 35.4° does not occur in the *R* phase. Moreover, the intensity of the reflection at 35.8° is enhanced in the *R* phase compared to that in the



FIG. 3. (Color online) Observed (open circles), calculated (continuous line), and difference (bottom line) profiles obtained after the Rietveld refinement of NaNbO₃ using long-period modulated orthorhombic structure with space group *Pbnm* at 770 K. (b) In order to have an easy comparison with the Darlington and Knight work,⁹ the data have been plotted again in term of *d* spacing (Å). Inset shows the accountabilities of the characteristic superlattice reflections appearing in the *S* phase at $2\theta = 35.4^{\circ}$ and 35.8° (in *d*-spacing terms d = 2.454 and 2.431 Å).

S phase. The superlattice reflections in the *R* phase have (h,k/6,l); (h/2,k/2,l), and (h/2,k/2,l/2) indices (in terms of the elementary perovskite cell). The reflections appear due to condensation of *DT* (q = 0,1/6,0), *M* (q = 1/2,1/2,0), and *R* (q = 1/2,1/2,1/2) soft modes. The structure of the *R* phase is not yet well investigated using neutron diffraction data. Sakowski has studied the structure of the *R* phase; however, the structural parameters were not refined.⁹ The author has shown that the *R* phase has orthorhombic structure with space group *Pnmm* with cell dimension $2 \times 6 \times 2$ times as large as that of the elementary perovskite cell.

The Miller indices of the superlattice reflections based on an elementary perovskite cell give information about the nature of the octahedral tilts in the structure. Superlattice reflections with all-odd integered indices ("odd-odd-odd" i.e., "000" type in Gazer notation) and two-odd and one-even integered indices result from antiphase (– tilt) and in-phase (+ tilt) tilting of the adjacent oxygen octahedron. This is due to structural phase transitions driven by softening and freezing of the phonons at R (q = 1/2, 1/2, 1/2) and M (q = 1/2, 1/2, 0) points of the cubic Brillouin zone, respectively. The presence of superlattice (h,k/6,l), (h/2,k/2,l), and (h/2,k/2,l/2) type reflections in

powder neutron diffraction pattern confirms that these phase transition involve *DT*, *M*, and *R* soft modes.

To find the most probable space group we coupled DT, M_3^+ , and R_4^+ irreducible representations (irreps) using the software package ISOTROPY 2000²⁴ and obtained many space groups. The P phase of NaNbO₃ is orthorhombic (Pbcm) and undergoes a first-order phase transition to the R phase. Thus we imposed the constraints that the R phase has higher symmetry than the P phase (*Pbcm*), is primitive in nature and a subgroup of *Cmcm*, and does not follow a group-subgroup relation for Pbcm. This constraint reduced the number of possible space groups for the R phase. We find that the orthorhombic structure with space group Pnma (different setting of Pbnm) with cell dimensions $\sqrt{2} \times 6 \times \sqrt{2}$ with respect to the elementary perovskite cell accounts for all the reflection in the *R* phase (see Fig. 4). In the space group Pbnm, for the description of the crystal structure, we used the following axes: $\mathbf{A}_0 = -\mathbf{a}_p + \mathbf{b}_p \mathbf{B}_0 = \mathbf{a}_p + \mathbf{b}_p$, and $C_0 = 6c_p$, where a_p , b_p , and c_p are elementary perovskite cell parameters. The orthorhombic unit cell contains 12 formula units. The detailed structural parameters and goodness of fit for the R phase of sodium niobate at 680 K, as obtained from neutron diffraction data, are given in Table II(b). The



FIG. 4. (Color online) Observed (open circles), calculated (continuous line), and difference (bottom line) profiles obtained after the Rietveld refinement of NaNbO₃ using orthorhombic *Pbnm* space groups at 680 K. The arrow shows the accountability of the additonal superlattice reflection. (b) In order to have an easy comparison with the Darlington and Knight work,⁹ the data have been plotted again in terms of *d* spacing (Å). Inset shows the accountabilities of the characteristic superlattice reflections appearing in the *R* phase.



FIG. 5. (Color online) Observed (open circles), calculated (continuous line), and difference (bottom line) profiles obtained after the Rietveld refinement of NaNbO₃ using orthorhombic *Pbcm* space groups at 300 K. (b) Variation of the integrated intensity of one of the strongest antiferroelectric reflections appearing around $2\theta = 35.6^{\circ}$ with temperature. The *R* factors obtained by Rietveld refinement of neutron diffraction data for the orthorhombic antiferroelectric phase are $R_B = 4.24$; $R_p = 4.24$; $R_{wp} = 5.33$; $R_{expt} = 3.67$; $\chi^2 = 2.11$.

fit between the observed and calculated profiles is quite satisfactory and includes the weak superlattice reflections. For comparison, we have also carried out Rietveld refinement using the *Pnmm* space group with cell dimension $2 \times 6 \times 2$. We found higher χ^2 with the *Pnmm* space group compared to the orthorhombic *Pbnm* space group although the former involved more parameters in the refinement.

5. Analysis of P phase (temperature range 300 < T < 650 K)

The powder neutron diffraction data of the P phase (300– 650 K) include superlattice reflections which assume the indices (h,k,l/p), where h, k, and l are integers, and p (four in this case) gives the multiple along the [001] direction with respect to the equivalent cubic perovskite cell. These superlattice reflections are characteristic of the antiferroelectric phase. The powder neutron data have been analyzed using the orthorhombic structure with space group Pbcm (different setting of Pbma). The size of the orthorhombic unit cell is $\sqrt{2} \times \sqrt{2} \times 4$ times as large as that of the high-temperature cubic cell (elementary perovskite cell). Figure 5(a) depicts the observed, calculated, and difference profiles obtained by Rietveld analysis of the powder neutron data at 300 K. The fit between the observed and calculated profiles is quite good and includes all the weak superlattice reflections. This confirms the orthorhombic structure with space group Pbcm of NaNbO₃ in the temperature range from 300 to 650 K (Fig. 1). Based on synchrotron x-ray diffraction diffraction studies, Yuzyuk et al.⁶ proposed that the P phase (from 250 to 650 K) consists of three phases: monoclinic (Pm) between 250 and 410 K, incommensurate between 410 and 460 K, and orthorhombic (Po) between 460 and 633 K. In view of this, we have collected data in a small temperature interval and carried out careful Rietveld analysis. We have determined the structure using neutron diffraction, which offers advantages over x-ray diffraction, especially in the accurate determination of oxygen positions that are found to be responsible for subtle changes in structures. However, we did not find any signature for the presence of the above phase during Rietveld analysis of powder neutron data. The intensity of the characteristic antiferroelectric reflection is found to decrease with increasing temperature and finally vanish at 680 K [see Fig. 5(b)].

C. Variation of lattice parameters

Variation of lattice parameters with temperature obtained from the Rietveld refinements is plotted in Fig. 6. The [110], $[1,\overline{1},0]$, and [001] directions of the cubic phase correspond to the [100], [010], and [001] directions, respectively, in all the orthorhombic P (space group *Pbcm*), R and S (space group Pbnm) phases, and tetragonal (space group P4/mbm) phases. On the other hand, the lattice vectors of the orthorhombic T_1 phase (space group *Cmcm*) coincide with those of the original (100) cubic phase vectors. For the sake of easy comparison with the corresponding cell parameters of the various phases of sodium niobate, we have plotted equivalent elementary lattice parameters instead of lattice parameters corresponding to the different phases in Fig. 6. It is evident from the figure that in the antiferroelectric orthorhombic phase (P phase), the a_p and c_p parameters increase while the b_p parameter decreases with increase in temperature. We observe a discontinuous jump of lattice parameters at about 680 K. Shirane et al.²⁵ also observed thermal hysteresis in the dielectric anomaly around



FIG. 6. (Color online) The variation of the pseudocubic lattice parameters as a function of temperature obtained after Rietveld refinement for NaNbO₃.

this temperature. These two observations suggest a first-order phase transition.

Around 680 K, all the lattice parameters come close to each other and increase continuously with increase of temperature. The absence of abrupt change in the lattice parameters at the R to S phase transition temperature suggests that the phase transition is of second order in nature. As we further increase the temperature, the a_p , b_p , and c_p lattice parameters of the orthorhombic (Cmcm phase) phase increase and the a_p parameter approaches b_p around 865 K. At this temperature, the sample undergoes an orthorhombic to tetragonal phase transition. Further, as temperature increases, the c_p lattice parameter decreases and finally approaches the a_p parameter, and the sample transforms into the cubic phase. The continuous change in the lattice parameters reveals that the successive phase transformations from orthorhombic R to high-temperature cubic phase are not of a strong first order in nature.

IV. PHASE TRANSITIONS

Distinct changes or vanishing of the intensity of superlattice reflections observed in the temperature dependence of neutron diffraction patterns of NaNbO₃ clearly reveal that it undergoes a series of phase transitions as a function of temperature, and consistent with previous observations from the literature.^{9–11} Based on powder synchrotron x-ray diffraction studies, Yuzyuk *et al.*⁶ claimed existence of three phases over the temperature range 250 to 633 K, namely, monoclinic (*Pm*) between 250 and 410 K, incommensurate between 410 and 460 K, and orthorhombic (*Po*) between 460 and 633 K. We are able to refine all the diffraction patterns over 300 to 633 K with the orthorhombic phase and do not find any evidence of a phase transition over the temperature range. However, it may be noted from Fig. 7 that around 500 K, the



FIG. 7. Logarithmic plots of the variation of the integrated intensity (*I*) (in arbitrary units) of (310) reflection and tilt angle (α) (in degrees) with temperature. The symbols represent experimental values while the continuous line corresponds to the least-squares fit in the temperature range $0.89T_c \leq T < 0.99T_c$. The transition temperature (T_c) is found to be 953 K.

plots of the lattice parameters a_p and c_p cross each other. The microscopic origins of various phases that stabilized over 10 to 1075 K are discussed below.

Recently, Izumi et al.12 carried out a detailed inelastic neutron scattering study in the cubic phase of NaNbO₃. The measurements show gradual softening of transverse acoustic (TA) phonon modes at the zone boundary M(1/2, 1/2, 0) and R (1/2, 1/2, 1/2) points, indicating the instabilities of the in-phase and out-of-phase rotations of the oxygen octahedron about the [001] direction. The softening of these modes suggests a low-lying flat transverse acoustic phonon-dispersion relation along the zone boundary M-R line (T line). The sequence of phase evolution depends on the condensation sequence of the soft modes. The R_{25} mode is threefold degenerate and the M_3 mode is nondegenerate. The triply degenerate R_{25} soft mode could be treated as made up of three components corresponding to the rotations of the octahedra around three separate [001] axes. If only one component condensed at the transition point, the resulting structure would be tetragonal I4/mcm. On the other hand, coupled condensation of all the three components would give a rhombohedral R3c structure. Similarly, depending on the number of condensed M_3 phonons in the **q** star (successively one, two, and three), the symmetry-related second-order phase transition from the cubic perovskite could be to $Pm\bar{3}m$, P4/mbm, I4/mmm, and Im3, respectively.

In the case of NaNbO₃, in contrast to other perovskites like SrTiO₃, CaTiO₃, KNbO₃, etc., the M_3 phonon condenses first on cooling, followed by the R_{25} phonon on further cooling. The first structural transformation is from cubic $Pm\bar{3}m$ to tetragonal $(T_1) P4/mbm$ structure, doubling the unit cell in the plane perpendicular to the rotation axes though one M_3 mode condensation. Further, condensation of the R_{25} phonon leads to the orthorhombic Cmcm (T_2) phase. This is an example of the M_3 mode instability in the cubic to tetragonal phase transition in a crystal of perovskite structure. Further, the observation of (1) a low-lying flat transverse acoustic phonon dispersion relation along the zone boundary M-R line and (2) strongly anisotropic diffuse scattering¹¹ that is distributed along (100) reciprocal rods at R and M point strongly indicate that the low-frequency zone boundary phonons lying between the M and R points also contribute to the transitions. These phonons play an important role in stabilization of different (P, S, and R) phases in NaNbO₃. The successive phase transitions from orthorhombic $Cmcm(T_1)$ to other phases may be explained as follows. Comparison of neutron diffraction patterns of orthorhombic phases (P, R, S, and T_2 phases) reveals that P, R, and S phases have additional superlattice reflections with respect to the orthorhombic T_2 (*Cmcm*) phase. We propose that these additional superlattice reflections arise due to condensation of the zone boundary phonons lying on the T(M-R) line $(\mathbf{q} = 1/2, 1/2, g)$ as a function of temperature. The orthorhombic structure of S, R, and P phases result from condensation of phonon modes ($\mathbf{q} = 1/2, 1/2, g$) with g = 1/12, 1/6, and 1/4, respectively. In other words, the structures of these various orthorhombic phases result from modulation of the high-symmetry cubic phase associated with phonon modes at $\mathbf{q} = (1/2, 1/2, g)$. Further, freezing of all the R_{25} modes along with a zone center phonon stabilizes the low-temperature ferroelectric rhombohedral phase.

The continuous variation of lattice parameters with temperature for the orthorhombic T_1 , S, and R phases suggests that the successive phase transitions are of second order in nature. However, the phase transition from the orthorhombic R to Pphase is of first order in nature. We recall that the space groups for orthorhombic T_1 and antiferroelectric P phases are *Cmcm* and *Pbcm*, respectively. Since the *Cmcm* to *Pbnm* transition follows the group-subgroup relation while *Pbnm* to *Pbcm* does not, we propose that the space group for both the orthorhombic S and R phases is *Pbnm* but with different unit cell modulation which accounts for all the superlattice reflections observed in neutron diffraction patterns.

The nature of the antiferrodistortive phase transition in materials exhibiting perovskite structure has attracted enormous attention^{26–28} in the literature. Mainly, these phase transitions are driven by *R* point instabilities. To the best of our knowledge, no attempt has been made to investigate the nature of the antiferrodistortive phase transition driven by M point instabilities. With this aim, we have determined the effective exponent using the intensity of the superlattice reflection and tilt angle for sodium niobate of the tetragonal to cubic phase transition. The general relation between the order parameter (p) and the intensity (I) of these superlattice reflections is $I \propto p^2$. Figure 7 depicts the temperature dependence of the integrated intensity of the (310) superlattice reflection (associated with M point instabilities) and also the tilt angle (α) as a function of temperature in the tetragonal phase. The data are fitted to the power laws $I \propto (T_c - T)^{2\beta}$ and $\alpha \propto (T_c - T)^{\beta}$, which yields a critical exponent of $\beta \approx 0.32 \pm 0.03$ and transition temperature $(T_c) = 953$ K. This behavior as seen in SrTiO₃, CaTiO₃, Sr_xCa_{1-x}TiO₃, and certain other perovskites^{26–28} is typical of tricritical phase transitions. The critical exponent (β) for the cubic to tetragonal phase transition, as obtained from the temperature variation of the primary order parameter, is found to have a value of 1/3. Thus, it is argued that this exponent is also a nearly tricritical transition.

Before we close, we would like to mention that by combining our present neutron diffraction data and earlier work⁷, we have modified the phase diagram proposed by pervious workers,² which was mainly based on x-ray diffraction studies. We now proceed to discuss the different features of our phase diagram given in Fig. 8. At high temperatures, the structure of NaNbO3 is cubic. First-principles calculations predicted the coexistence of zone center and zone boundary phonon instabilities in the cubic phase. On lowering the temperature, it undergoes successively transitions to tetragonal (T_2 phase) and orthorhombic (T_1 phase) phases as a result of the condensation of zone boundary M and R point instabilities, respectively. Further, on lowering the temperature, the appearance of additional superlattice reflections in orthorhombic S, R, and P phases with respect to the orthorhombic (T_1) phase is linked to freezing of zone boundary phonons along the line T (lying between the M and R points) with $\mathbf{q} = (1/2, 1/2, g)$ with g = 1/12, 1/6, and 1/4, respectively. The inelastic neutron scattering and x-ray diffuse scattering experiments also suggest phonon instabilities along the T line. Finally, NaNbO₃ undergoes a rhombohedral phase transition due to condensation of both zone center and zone boundary (R_{25}) modes. It is also important to state that for the orthorhombic S and R phases, we have also explored the possibilities for



FIG. 8. (Color online) The phase diagram of NaNbO₃ as a function of temperature after neutron diffraction studies. The structure and space group of S and R phases correspond to our own present work; while the existence of two phases below room temperature correspond to previous work.

the incommensurate-commensurate modulation using JANA software,²⁹ as observed for solid solutions of Ba or other perovskites.^{30,31} We refined the diffraction data of the *S* and *R* phases with incommensurate modulated structures using JANA software; we find that the refined modulation vector does not vary significantly with temperature and the refined structure in both cases is quite close to the corresponding commensurate structure we reported above. Therefore, we do not consider the structures of the *S* and *R* phases as incommensurate, which is also consistent with the transmission electron study by Chen *et al.*¹³ Single-crystal diffraction data are highly desirable for further confirmation.

V. CONCLUSIONS

In summary, we have established the structures and space groups of the technologically important sodium niobate up to 1075 K. Distinct changes or vanishing of the intensity of superlattice reflections, as observed in the temperature dependence of neutron diffraction patterns, clearly revealed that NaNbO₃ undergoes a series of phase transitions as a function of temperature, ranging from nonpolar antiferrodistortive to ferroelectric and antiferroelectric in nature. These phase transitions are mainly driven by zone boundary phonon instabilities at the *R* and *M* points and along the line *T* (connecting the *M* and *R* points). Additional superlattice reflections are found to appear at 680 K (*R* phase) and 770 K (*S* phase), which could be indexed by using an intermediate long-period modulated orthorhombic structure whose lattice parameter is three and six times the lattice parameter of the CaTiO₃ type *Pbnm* structure along (001) directions. The correlation of superlattice reflections with the instability of zone boundary phonons is discussed. The discontinuous jump in lattice parameters at about 680 K and thermal hysteresis in the dielectric anomaly observed by Shirane *et al.*²⁵ suggest that the transition from the orthorhombic *P* to *R* phase is of first order in nature, while other successive phase transitions are of second order. The phase diagram of NaNbO₃ has been reviewed based on our detailed temperature-dependent neutron diffraction studies that resolved certain ambiguities in the literature.

- ¹M. E. Lines and A. M. Glass, Principles and Application of Ferroelectrics and Related Materials (Clarendon, Oxford, 1977); Xu. Yuhuan, Ferroelectric Materials and Their Applications (North-Holland/Elsevier Science, Amsterdam, 1991); L. G. Tejuca and J. L. G. Fierro, Properties and Applications of Perovskite-Type Oxides (Dekker, New York, 1993).
- ²E. Bousquet, M. Dawber, N. Stucki, C. Lichtensteiger, P. Hermet, S. Gariglio, J. M. Triscone, and P. Ghosez, Nature (London) **452**,
- 723 (2008); T. Choi, Y. Horibe, H. T. Yi, Y. J. Choi, Weida Wu, and S.-W. Cheong, Nature Mater. **9**, 253 (2010).
- ³Y. Saito, H. Takao, T. Tani, T. Nonoyama, K. Takatori, T. Homma,
- T. Nagaya, and M. Nakamura, Nature (London) 423, 84 (2004).
- ⁴E. Cross, Nature (London) **432**, 24 (2004).
- ⁵E Valdez, C. B. de Araujo, and A. A. Lipovskii, Appl. Phys. Lett. **89**, 031901 (2006); E. Hollenstein, M. Davis, D. Damjanovic, and Nava Setter, *ibid.* **87**, 182905 (2006); E. L. Falcão-Filho, C. A. C. Bosco, G. S. Maciel, L. H. Acioli, Cid B. de Araújo, A. A. Lipovskii, and D. K. Tagantsev, Phys. Rev. B **69**, 134204 (2004); G. S. Maciel, N. Rakov, Cid B. de Araujo, A. A. Lipovskii, and
- D. K. Tagantsev, Appl. Phys. Lett. 79, 5 584 (2001).
- ⁶Yu. I. Yuzyuk, P. Simon, E. Gagarina, L. Hennet, D. Thiaudiere, V. I. Torgashev, S. I. Raevskya, I. P. Raevskii, L. A. Reznitchenko, and J. L. Sauvajol, J. Phys.: Condens. Matter 17, 4977 (2005); Yu. I. Yuyuk, E. Gagarina, P. Simon, L. A. Reznitchenko, L. Hennet, and D. Thiaudiere, Phys. Rev B 69, 144105 (2004); S Tripathi, D. Pandey, S K Mishra, and P. S. R. Krishna, *ibid.* 77, 052104 (2008); A. V. Ulinzheev, A. V. Leiderman, V. G. Smotrakov, V. Yu. Topolov, and
- O. E. Fesenko, Phys. Solid State **39**, 972 (1997).
- ⁷S. K. Mishra, N. Choudhury, S. L. Chaplot, P. S. R. Krishna, and R. Mittal, Phys. Rev B **76**, 024110 (2007), and reference therein.
- ⁸Y. Shiratori, A. Magrez, J. Dornseiffer, F. H. Haegel, C. Pithan, and R. Waser, J. Phys. Chem. B **43**, 20122 (2005); Y. Shiratori, A. Magrez, W. Fischer, C. Pithan, and R. Waser, *ibid.* **111**, 18493 (2007); Y. Shiratori, A. Magrez, M. Kato, K. Kasezawa, C. Pithan, and R. Waser, *ibid.* **112**, 9610 (2008).
- ⁹C. N. W. Darlington and K. S. Knight, Physica B **266**, 368 (1999); Acta Crystallogr. Sect. B **55**, 24 (1999); C. N. W. Darlington and H. D. Megaw, *ibid.* **29**, 2171 (1973); C. N. W. Darlington, Solid State Commun. **29**, 307 (1979); L. E. Cross and B. J. Nicholson, Philos. Mag. **46**, 212 (1955); A. C. Sakowski-Cowley, Ph.D. Thesis, University of Cambridge, 1967.
- ¹⁰M. Ahtee, A M. Glazer, and H. D Megaw, Philos. Mag. **26** 995 (1972); A. M. Glazer and H. D. Megaw, *ibid.* **25** 1119 (1972); Acta Crystallogr. Sect. A. **29**, 489 (1973); A. C. Sakowski-Cowely,

K. Lukazewicz, and H. D. Megaw, Acta Crystallogr. Sect. B 25, 851 (1969);

- ¹¹I. Lefkowitzk, K. Lukazewicz, and H. D. Megaw, Acta. Crystllogr. **20**, 670 (1966); F. Denoyer, M. Lambert, and R. Comes, Solid State Commun. **18**, 441 (1976), and references therein.
- ¹²I. Tomeno, Y. Tsunoda, K. Oka, M. Matsuura, and M. Nishi, Phys. Rev. B **80** 104101 (2009); K. Ishida and G. Honj, J. Phys. Soc. Jpn. **34**, 1279 (1973).
- ¹³J. Chen and D. Feng, Phys. Status Solidi A **109**, 171 (1988); **109**, 427 (1988).
- ¹⁴W. Zhong and D. Vanderbilt, Phys. Rev. Lett. **74**, 2587 (1995).
- ¹⁵O. Diéguez, K. M. Rabe, and D. Vanderbilt, Phys. Rev. B **72**, 144101 (2005).
- ¹⁶D. Vanderbilt and W. Zhong, Ferroelectrics 181, 206 (1998).
- ¹⁷W. Zhong, R. D. King-Smith, and D. Vanderbilt, Phys. Rev. Lett. **72**, 3618 (1994).
- ¹⁸R. D. King-Smith and D. Vanderbilt, Phys. Rev. B **49**, 5828 (1994). ¹⁹Throughout this paper, we use the nomenclature of the specific points and lines in the Brillouin zone of the parent cubic perovskite phase (space group $pm \bar{3}m$) only. The term "superlattice reflection" is used with respect to the parent cubic phase.
- ²⁰A. M. Glazer, Acta Crystallogr., Sect. B **28**, 3384 (1972); Acta Crystallogr. **31**, 756 (1975).
- ²¹P. Fischer *et al.*, Physica B **276–278**, 146 (2000); [http://sing.web.psi.ch/hrpt]
- ²²J. Rodriguez-Carvajal Physica B 192, 55 (1993).
- ²³J. Hutton, R. J. Nelms, G. M. Meyer, and V. R. Eiriksson, J. Phys. C 12, 5393 (1979).
- ²⁴H. T. Stokes, D. M. Hatch, and B. J. Campbell, computer code ISOTROPY (2007) [http://stokes.byu.edu/isotropy.html].
- ²⁵G. Shirane, R. Newnham, and R. Pepinsky, Phys. Rev. **96**, 581 (1954).
- ²⁶A. D. Bruce and R. A. Cowely, Adv. Phys. **29**, 219 (1980).
- ²⁷S. A. T. Redfern, J. Phys.: Condens. Matter 8, 8267 (1996); B. J. Kennedy, C. J. Howard, and B. C. Chakoumakos, *ibid.* 11, 1479 (1999).
- ²⁸S K Mishra and D Pandey, Phys. Rev. B 79, 174111 (2009).
- ²⁹V Petříček, M. Dušek, and L Palatinus, computer code JANA2000 (Institute of Physics, Praha, Czech Republic).
- ³⁰G. Baldinozziy, D. Grebillez, Ph. Sciauy, J.-M. Kiatyx, J. Moretk, and J-F Berar, J. Phys.: Condens. Matter **10**, 6461 (1998); J. M. Kia, G. Calvarin, and J. Schneck, Phys. Rev. B **49**, 776 (1994).
- ³¹H. Z. Cummins, Phys. Rep. **185**, 211 (1990).