Oxygen hyperstoichiometric hexagonal ferrite CaBaFe₄O_{7+ δ} ($\delta \approx 0.14$): Coexistence of ferrimagnetism and spin glass behavior

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An oxygen hyperstoichiometric ferrite CaBaFe₄O_{7+ δ} ($\delta \approx 0.14$) has been synthesized using "soft" reduction of CaBaFe₄O₈. Like the oxygen stoichiometric ferrimagnet CaBaFe₄O₇, this oxide also keeps the hexagonal symmetry (space group $P6_3mc$), and exhibits the same high Curie temperature of 270 K. However, the introduction of extra oxygen into the system weakens the ferrimagnetic interaction significantly at the cost of increased magnetic frustration at low temperature. Moreover, this canonical spin glass ($T_g \sim 166$ K) exhibits an intriguing crossover from de Almeida–Thouless type to Gabay-Toulouse type critical line in the field temperature plane above a certain field strength, which can be identified as the anisotropy field. Domain-wall pinning is also observed below 110 K. These results are interpreted on the basis of cationic disordering on the iron sites.

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I. INTRODUCTION

The recent discovery of the new series of 114 oxides [the cobaltites, $(Ln,Ca)_1BaCo_4O_7$ (Refs. 1–8), and the ferrites, $(Ln,Ca)_1BaFe_4O_7$ (Refs. 9–11)] have opened up a new field for the investigation of strongly correlated electron systems. These oxides consist of CoO_4 (or FeO_4) tetrahedra sitting in alternating layers of $kagom\acute{e}$ and triangular arrays. ¹⁰ The structure can also be described as the stacking of close-packed $[BaO_3]_{\infty}$ and $[O_4]_{\infty}$ layers whose tetrahedral cavities are occupied by Co^{2+}/Co^{3+} (or Fe^{2+}/Fe^{3+}) species, forming triangular and $kagom\acute{e}$ layers of CoO_4 (or FeO_4) tetrahedra. This structure has been primarily responsible for the wide variety of magnetic states that has been observed in this group of oxides, ranging from a spin glass for cubic $LnBaFe_4O_7$ (Refs. 9 and 10) to a ferrimagnet for orthorhombic $CaBaCo_4O_7$ (Ref. 5) and hexagonal $CaBaFe_4O_7$ (Ref. 9) oxides.

Recent studies of the 114 cobaltites 12-17 have revealed the existence of closely related structures with various crystallographic symmetries, and the possibility of oxygen nonstoichiometry in the range "O₇" to "O_{8.5}" in those systems. This change of oxygen stoichiometry, which induces the variation of the Co²⁺:Co³⁺ ratio in the system, is expected to influence the physical properties of these compounds considerably. This is the case of the oxygen-rich 114 cobaltites YBaCo₄O_{8.1} (Ref. 15) and YbBaCo₄O_{7.2} (Ref. 17), which were shown to be magnetically frustrated rather than magnetically ordered at low temperatures.

In contrast to the cobalt oxides, no report of oxygen hyperstoichiometric 114 ferrites exists to date, probably due to the fact that Fe^{2+} gets too easily oxidized into Fe^{3+} , thereby destabilizing the 114 structure at the benefit of pure "Fe³⁺" oxides. We have thus investigated the possibility to stabilize the mixed valence Fe^{2+}/Fe^{3+} in the 114 oxygen hyperstoichiometric $CaBaFe_4O_{7+\delta}$ ferrite by reducing the fully oxidized compound $CaBaFe_4O_8$ (Ref. 18) at low temperature in an argon-hydrogen atmosphere. We report herein on the magnetic properties of the 114 oxygen hyperstoichiometric $CaBaFe_4O_{7.14}$ hexagonal ferrite. We show that, like the stoichiometric phase $CaBaFe_4O_7$, this oxide also exhibits ferrimagnetism with a T_C of 270 K, but that the competition between ferrimagnetism and magnetic frustration is much

more pronounced than for the stoichiometric phase, as seen from the decrease of the magnetization. More importantly, we observe that CaBaFe₄O_{7.14} is characterized by a canonical spin glass behavior with $T_g \approx 166$ K, and an intriguing crossover from an Ising to a Heisenberg spin glass type behavior in the external magnetic field at low temperature. Besides this competition between ferrimagnetism and spin glass behavior, one also observes domain-wall pinning below 110 K. This very different magnetic behavior of CaBaFe₄O_{7.14} is explained in terms of cationic deficiency and disordering on the iron sites, the "barium-oxygen" hexagonal close packing remaining untouched.

II. EXPERIMENT

The precursor CaBaFe₄O₈ (Ref. 18) was prepared by the sol gel method. Stoichiometric amounts of calcium carbonate (Prolabo, 99%) and barium carbonate (Alfa Aesar, 99%) were dissolved in a large excess of melted citric acid monohydrate at $\sim\!200~^\circ\text{C}$. Iron citrate (Alfa Aesar, 20% of Fe) was separately dissolved in hot water leading to a dark brown solution which was poured on the citrate mixture. The water was then evaporated followed by decomposition of the gel. The gel was calcined at 450 $^\circ\text{C}$ under air to obtain an amorphous precursor, which was then pressed into pellets before firing at 1200 $^\circ\text{C}$ to obtain CaBaFe₄O₈.

The oxygen hyperstoichiometric 114 ferrite CaBaFe $_4O_{7+\delta}$ was then obtained by reducing CaBaFe $_4O_8$ under an Ar/H $_2$ 10% mix at 610 °C for 24 h.

The oxygen content of the sample was determined by redox titration. The sample was dissolved in hot HCl (3M) flushed with argon to remove the dissolved oxygen. After cooling down the solution, Fe²⁺ cations were titrated using 2×10^{-2} M cerium (IV) sulfate (Riedel-de Haën) and 1.10-phenantroline iron (II) sulfate (Alfa Aesar) as an indicator under constant argon flow. We obtained $\delta=0.14$.

The x-ray-diffraction patterns were registered with a Panalytical X'Pert Pro diffractometer under a continuous scanning mode in the 2θ range $10^{\circ}-120^{\circ}$ and step size $\Delta 2\theta = 0.017^{\circ}$. The dc magnetization measurements were performed using a superconducting quantum interference device (SQUID) magnetometer with variable temperature cryostat (Quantum

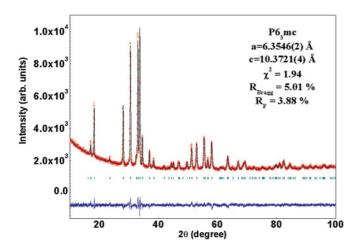


FIG. 1. (Color online) X-ray-diffraction pattern along with the fit for CaBaFe₄O_{7+ δ}.

Design, San Diego, USA). The ac susceptibility, $\chi_{ac}(T)$ was measured with a physical property measurement system (PPMS) from Quantum Design with the frequency ranging from 10 Hz to 10 kHz ($H_{dc}=0$ Oe and $H_{ac}=10$ Oe). All the magnetic properties were registered on dense ceramic bars of dimensions $\sim 4 \times 2 \times 2$ mm³.

III. RESULTS AND DISCUSSION

A. Structural characterization

The x-ray-diffraction pattern (Fig. 1) revealed that CaBaFe₄O_{7.14} stabilized in the same hexagonal symmetry (space group $P6_3mc$) as the O_7 phase. The Rietveld analysis from the x-ray-diffraction (XRD) data was done using the FULLPROF refinement program.¹⁹ The fit is also shown in Fig. 1 (top red curve). The bottom blue curve corresponds to the difference between the observed and calculated diffraction patterns. Satisfactory matching of the experimental data with the calculated profile of the XRD pattern and the corresponding reliability factors $R_F = 3.88\%$ and $R_{\text{Bragg}} = 5.01\%$ confirm that the fit obtained is reasonably accurate. The extracted lattice parameters (a = 6.355 Å, c = 10.372 Å) show a very marginal increase over the O₇ phase—a increases by \sim 0.11% while c remains virtually unchanged. The refinements of the atomic coordinates thus lead to results similar to those previously obtained for CaBaFe₄O₇. The low δ value and the low scattering factor of oxygen do not allow any oxygen excess or cationic deficiency to be detected from x-ray powder-diffraction data. A very careful neutron-diffraction study might perhaps allow the issue to be sorted out, but will really be at the limit of accuracy, and consequently, is not within the scope of this paper.

B. dc magnetization study

In Fig. 2, we show the zero-field-cooled (ZFC) and field-cooled (FC) magnetization of CaBaFe₄O_{7+ δ} recorded under a magnetizing field of 0.3 T. The sample shows the same increase in magnetization below ~270 K as the oxygen stoichiometric oxide indicating a similar transition to an ordered magnetic state below 270 K. However, a careful

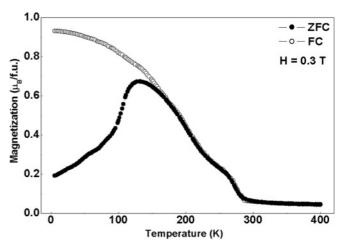


FIG. 2. $M_{\rm ZFC}(T)$ and $M_{\rm FC}(T)$ curves of CaBaFe₄O_{7+ δ} measured at $H=0.3~\rm T.$

look at the magnetization values reached at the lowermost measured temperature (5 K) immediately reveals a striking difference in the magnetic behavior of CaBaFe₄O_{7.14} *vis* à *vis* that of CaBaFe₄O₇. While the FC magnetization of the oxygen stoichiometric compound reaches a value of more than $2.5 \,\mu_B/f.u.$ at T=5 K, the maximum magnetization value of our oxygen-rich sample is only $0.93 \,\mu_B/f.u.$, which is less by more than a factor of $\frac{1}{2}$.

This large difference in the magnetization value at low temperature between the two samples ($\delta = 0$ and $\delta > 0$) prompted us to record the hysteresis curve of our oxygen-rich sample at low temperature (T = 5 K) and compare it with that of the oxygen stoichiometric sample. This is shown in Fig. 3. The magnetization value obtained at the highest measuring field of 5 T ($2.5\mu_B/f.u.$) is again different from the oxygen stoichiometric sample ($3.1\mu_B/f.u.$), as expected. More importantly, a rather striking difference is seen in the shape of the hysteresis loop. The coercive field (H_C) and remanent magnetization (M_T) of our sample at T = 5 K are 0.77 T

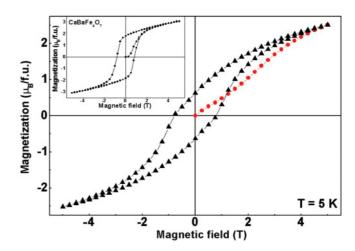


FIG. 3. (Color online) M(H) curve of CaBaFe₄O_{7+ δ} measured at T=5 K. The virgin curve is shown in red circles, while the rest of the hysteresis loop is shown in black triangles. The inset shows the M(H) curve of the oxygen stoichiometric sample (CaBaFe₄O₇) measured at T=5 K.

and $0.63 \, \mu_B/{\rm f.u.}$, respectively. While the value of the coercive field compares well with that of the oxygen stoichiometric sample, the value of the remanent magnetization is much lower than that obtained for CaBaFe₄O₇ ($M_r \sim 1.8 \, \mu_B/{\rm f.u.}$). This results in the overall shape of the hysteresis loop of our sample (Fig. 3) to be very different from that of the oxygen stoichiometric sample (inset of Fig. 3). The degree of magnetic saturation in a sample can be roughly quantified from the M-H loop by calculating $\phi = M(H = 5T)/M_r$. While the oxygen stoichiometric sample had $\phi = 1.7$, our oxygen-rich sample yields $\phi = 4.0$. This higher value of ϕ for the oxygen-rich sample indicates an increased lack of magnetic saturation in the sample, or in other words, a weakening of long-range order.

Another important difference with the oxygen stoichiometric sample is in the virgin curve of the M(H) loop. While the virgin curve of the $\delta = 0$ sample lies entirely within the main loop, our oxygen-rich sample shows an unusual magnetic behavior where a major portion of the virgin curve lies outside the hysteresis loop and meets the main loop only at very high fields.

C. ac magnetic susceptibility study

The temperature dependence of the ac susceptibility of CaBaFe₄O_{7.14} in the temperature range 20–280 K and at four measuring frequencies ranging from 10 Hz to 10 kHz is shown in Fig. 4. The sample shows several interesting features which we will now proceed to discuss separately.

(i) At T=272 K, one can see a sharp peak which is frequency independent, i.e., the position of the peak maximum does not shift with a change in the measuring frequency. This peak corresponds to the paramagnetic to ferrimagnetic (PM-FM) transition occurring in the sample as it is cooled below 272 K. We note here that the oxygen stoichiometric sample also showed a similar peak at around the same temperature. However, in the latter sample, the peak corresponding to the PM-FM transition was the strongest (maximum amplitude) compared to the other peaks. In our sample, this peak is

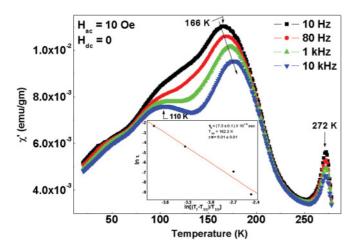


FIG. 4. (Color online) The real (in-phase) component of ac susceptibility for CaBaFe₄O_{7+ δ} as a function of temperature in the frequency range f=10 Hz–10 kHz, at zero static magnetic field ($H_{\rm dc}$) and at a driving ac field ($H_{\rm ac}$) of 10 Oe. The inset shows the plot of $\ln \tau \, {\rm vs} \, \ln(\frac{T_f-T_{SG}}{T_{SG}})$ for the peak at 166 K.

much smaller in magnitude, which corresponds to a significant weakening of the magnetic ordering (or a smaller volume fraction of ferrimagnetic domains), which we had mentioned earlier in connection with the M(H) loop.

(ii) The oxide CaBaFe₄O_{7.14} shows a broader peak at lower temperature (~166 K) which shows pronounced frequency dependence. The peak temperature shifts from 166 K (for a measuring frequency of 10 Hz) to 176 K (for a measuring frequency of 10 kHz). This corresponds to a peak shift of 0.02 per decade of frequency shift ($p = \frac{\Delta T_f/T_f}{\Delta \log f} = 0.02$). This value of the parameter p lies within the range for canonical spin glasses, which indicates that this peak is a signature of the sample undergoing a spin glass transition. We confirm this by analyzing the frequency dependence of this peak using the power-law form $\tau = \tau_0 (\frac{T_f - T_{SG}}{T_{SG}})^{-z\nu}$, where τ_0 is the shortest relaxation time available to the system, T_{SG} is the underlying spin glass transition temperature determined by the interactions in the system, z is the dynamic critical exponent, and ν is the critical exponent of the correlation length. The actual fittings were done using the equivalent form of the power law: $\ln \tau = \ln \tau_0 - z \nu \ln(\frac{T_f - T_{SG}}{T_{SG}})$. The fit parameters ($\tau_0 = 7.3 \times 10^{-10}$ sec, $z\nu = 5.01$, and $T_{SG} = 162.2$ K) give a good linear fit (as can be seen in the inset of Fig. 4), and confirms that this peak does correspond to a spin glass transition in the sample. Whether the magnetic order disappears in the spin glass phase is not clear at the moment. Our data show that the spin glass transition occurs within the ferrimagnetically ordered phase. Whether this transition is accompanied by the destruction of ferrimagnetic long-range order is an open issue as of now. The possibility of coexistence of ferrimagnetic and spin glass orders cannot, however, be ruled out.

(iii) At a lower temperature of ~ 110 K a very broad peak is seen (which broadens out to almost a kind of shoulder at lower measuring frequency). We will come back to the nature of this feature in the following section.

D. Magnetic-field dependence of the spin glass freezing temperature

In Fig. 5, ac susceptibility at 10 kHz driving frequency is plotted as a function of temperature for different external magnetic fields $H_{\rm dc}$ ranging from 0 to 0.3 T. As can be seen from the figure, χ' is suppressed by the magnetic field. First, we focus our discussion on the evolution of the spin glass freezing temperature (marked by a black arrow in the figure), which occurs at \sim 176 K at the lowest applied field. This peak temperature shows a continual shift toward lower temperature as the external magnetic field is increased, and reaches a value of 155 K at an external magnetic field of 0.3 T. Concomitantly, the peak amplitude keeps decreasing as the external magnetic field is increased from 0 to 0.3 T. A further increase of the external magnetic field should eventually suppress the spin glass transition completely.

The purpose of exploring how the freezing temperature responds to external magnetic field is to check the stability of the spin glass system. This is done by examining the field versus temperature phase diagram obtained from the ac susceptibility measurement as shown in Fig. 6. As was

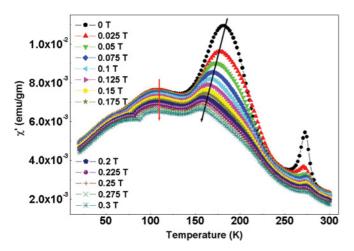


FIG. 5. (Color online) The real (in-phase) component of ac susceptibility for CaBaFe₄O_{7+ δ} as a function of temperature. The driving frequency was fixed at f=10 kHz and $h_{\rm ac}=10$ Oe. Each curve was obtained under different applied static magnetic field ($H_{\rm dc}$) ranging from 0 to 0.3 T.

mentioned above, the spin glass freezing temperature is suppressed by increasing the external magnetic field.

From a theoretical perspective, de Almeida and Thouless²⁰ studied the Ising spin glass system, and predicted that the spin freezing temperature (T_g) depends on H. In the low H range, T_g follows the so-called de Almeida-Thouless (AT) line, expressed as $H = H_0[1 - \frac{T_g(H)}{T_g(0)}]^{3/2}$. In addition, Gabay and Toulouse²¹ investigated the H dependence of the spin freezing temperature for the Heisenberg spin glass system. This led to the so-called Gabay-Toulouse (GT) line, expressed as $H = H_0[1 - \frac{T_g(H)}{T_g(0)}]^{1/2}$. The AT line and the GT line are the two critical lines predicted in the presence of field on the H-T plane, which mark the phase transition. The first one occurs for an anisotropic Ising spin glass while the second is valid for an isotropic Heisenberg spin glass.

Our sample shows a very interesting behavior. At low field values ($H_{\rm dc}$ < 0.15 T), T_g follows the AT line. This can be seen in Fig. 6, where the red line denotes the AT line. Since

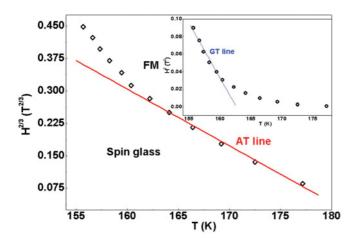


FIG. 6. (Color online) Field vs temperature phase diagram of CaBaFe₄O_{7+ δ}. In order to show the AT line, we have plotted $H^{2/3}$ vs T_g . The inset shows H^2 vs T_g and the GT line.

the AT line predicts that $T_g \propto H^{2/3}$, so we have plotted $H^{2/3}$ in the H-T phase diagram. However, with an increase in the field (H > 0.15 T), we find deviation from the AT line. Remarkably, it is found that at high field, the variation of $T_g(H)$ agrees with the GT line. This can be seen in the inset of Fig. 6, where the blue line denotes the GT line. Since the GT line predicts that $T_g \propto H^2$, so the H-T phase diagram in the inset is plotted with H^2 in the Y axis.

These experimental results can be explained using the theoretical calculation by Kotliar and Sompolinsky,²² who have predicted that in the presence of random anisotropy, the critical behavior for a spin glass in fields lower than the anisotropy field is close to Ising type following the AT line, and crosses over to Heisenberg behavior in high fields. The fact that we see a crossover in critical lines on the H-T plane for our sample indicates the existence of magnetic anisotropy in the system. At higher applied fields, the system behaves like a Heisenberg spin glass, where the spins can freeze along any direction with respect to the applied magnetic field. However, when the applied field is lower than the anisotropy field, the spins are forced to be aligned along the local anisotropy axis. The preference of the spin alignment adds an Ising character to the associated spin cluster.

E. Domain wall pinning at lower temperature

In this section, we discuss the third feature seen in the $\chi'(T)$ curve—the broad peak at ~ 110 K (Fig. 4). This peak at 110 K does not shift (i.e., the peak maximum occurs at the same temperature) with a change in the external magnetic field $H_{\rm dc}$ (see the red line in Fig. 5). Based on this behavior, we attribute the origin of the feature seen at ~ 110 K to enhanced domain wall pinning. The signature of this domain-wall pinning can also be seen in Fig. 7, where we plot the variation of the coercivity (H_C) with temperature. As is clear from the figure, the coercivity is significantly enhanced below 110 K, which occurs due to the domain-wall pinning. A close look at the high-temperature region, which is enlarged and shown in the inset of Fig. 7, reveals that the coercivity also shows an

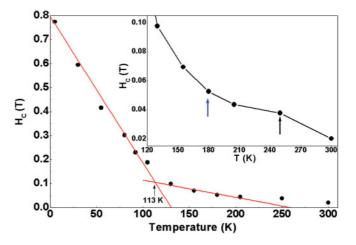


FIG. 7. (Color online) Temperature dependence of coercive field for CaBaFe₄O_{7+ δ}. The inset is an enlarged version of the high-temperature region.

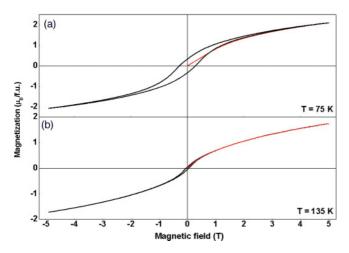


FIG. 8. (Color online) M(H) loops of CaBaFe₄O_{7+ δ} at (a) T=75 K and (b) T=135 K.

enhancement below the paramagnetic to ferrimagnetic phase transition temperature (shown by a black arrow), and another enhancement below the spin glass freezing temperature [shown by a gray (blue) arrow], as expected.

At this stage, we need to go back to our earlier observation of an unusual initial magnetization curve in the M(H) loop measured at low temperature (Fig. 3). Such unusual magnetic hysteresis behavior, with the virgin curve lying outside the main hysteresis loop, was earlier associated with irreversible domain-wall motion in spinel oxides.²³ Thus this unusual magnetization curve is an additional confirmation of the domain-wall pinning at \sim 110 K that we had mentioned earlier. In fact, we find that the virgin curve lies outside the main M(H) loop for temperatures below 110 K, but above 110 K, it lies completely inside the main hysteresis loop. This is shown in Fig. 8, where we plot the M(H) loops at temperatures slightly below [Fig. 8(a)], and slightly above [Fig. 8(b)] 110 K. In the figures, the virgin curves are shown in gray (red) for the sake of clarity.

F. Origin of the competition between ferrimagnetism and spin glass behavior

In order to understand the different magnetic behavior of CaBaFe₄O_{7,14} with respect to CaBaFe₄O₇, we must keep in mind that the oxygen excess in the former induces an increase of the Fe³⁺ content in the structure, i.e., the Fe³⁺:Fe²⁺ ratio increases from 1 in the stoichiometric phase to 1.32 in the oxygen hyperstoichiometric phase. As a consequence, the Fe³⁺-Fe³⁺ antiferromagnetic interactions increase in the oxygen-rich phase, and may decrease the ferrimagnetism in the structure. Bearing in mind the model previously proposed by Chapon et al..4 to explain the competition between one-dimensional (1D) ferromagnetism and 2D magnetic frustration in the cobaltite YBaCo₄O₇ which has the same hexagonal structure, we must consider the iron framework of our compound. The latter consists of corner-sharing [Fe₅]_∞ bipyramids running along "c" interconnected through Fe₃ triangles (Fig. 9). In other words, in both oxides, CaBaFe₄O₇ and CaBaFe₄O_{7,14}, we can expect, similarly to the hexagonal cobalt oxides LnBaCo₄O₇, that the system exhibits a unidimensional mag-

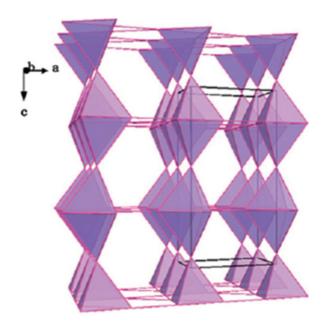


FIG. 9. (Color online) Schematic representation of the $[Fe_4]_{\infty}$ tetrahedral framework of hexagonal CaBaFe₄O_{7+ δ} showing the Fe₅ bipyramids sharing corners with Fe₃ triangular groups (adapted from Ref. 11).

netic order in the bipyramidal rows along c, whereas the triangular geometry of the iron lattice in the (001) plane induces magnetic frustration as soon as the iron species are coupled antiferromagnetically. Such a model can account for the competition between 1D ferrimagnetism and 2D magnetic frustration in both oxides, CaBaFe₄O₇ and CaBaFe₄O_{7.14}, and explain that the magnetic frustration may be larger in the latter owing to the appearance of larger short-range antiferromagnetic interactions in the (001) plane.

Nevertheless, the valency effect alone is not sufficient to explain the appearance of the spin glass behavior. Two hypotheses can be considered to explain this particular behavior. The first scenario deals with the fact that CaBaFe₄O_{7.14} contains interstitial oxygen in spite of the apparent closepacked character of the structure, leading to a local puckering of the O₄ and BaO₃ layers. As a result, the distribution of iron in the cationic sites would be locally disordered, leading to a spin glass behavior. This local distortion would also change the crystal field and would be responsible for the domain-wall pinning. The second scenario deals with the fact that the "barium oxygen" framework remains close packed, but that the compound exhibits a cationic deficiency according to the formula Ca_{0.98}(Ba_{0.98}O_{0.02})Fe_{3.93}O₇. Such an effect would be similar to that observed for "oxidized" spinels γ -Fe₂O₃ and Co_{3-x}O₄, which do not contain interstitial oxygen, but were found to be iron or cobalt deficient.^{24,25} This second scenario would explain the magnetic behavior of this phase, which is close to that observed for CaBaFe_{4-x}Li_xO₇.²⁶ In both systems, the doping of the Fe sites with lithium or vacancies, respectively, introduces disordering on the Fe sites, which is in turn, at the origin of the appearance of spin glass behavior at lower temperature. Thus the competition between 1D ferrimagnetism and spin glass behavior appears normal. Subsequently, the competition between anisotropic (Ising) and isotropic (Heisenberg) spin glass can be understood from the peculiar geometry of the $[Fe_4]_\infty$ lattice. Finally, the iron vacancies would change the nature of the crystal field in the structure, playing the role of pinning centers. This explains both the broad peak at 110 K and the enhanced coercivity below this temperature, which are the signatures of domain-wall pinning.

The small deviation from the stoichiometry does not allow us to distinguish the possibility of interstitial oxygen *vis à vis* that of cationic deficiency from a structural study. Attempts are being made to synthesize similar hexagonal ferrites with larger oxygen excess in order to answer this question.

IV. CONCLUSIONS

This study illustrates the extraordinarily rich physics of the 114 CaBaFe₄O_{7+ δ} ferrite, in connection with its ability to accommodate oxygen excess, similar to what is observed for the spinel family, Fe₃O₄– γ -Fe₂O₃. The remarkable feature of this 114 oxide deals with the competition between ferrimagnetism and spin glass behavior that can be induced by varying the

oxygen content, without changing the hexagonal symmetry of the structure. Such a behavior can be explained, like for the 114 cobaltites, as due to the competition between 1D magnetic ordering along c and 2D magnetic frustration in the triangular (001) lattice. Nevertheless, CaBaFe₄O₇ differs significantly from CaBaCo₄O₇, the latter's ferrimagnetism originating mainly from a lifting of its 2D geometrical frustration through a strong orthorhombic distortion of its initial hexagonal lattice. We believe that the scenario of cation disordering on iron sites is the key for understanding the magnetism of these materials. Further investigations, especially using neutron-diffraction and x-ray synchrotron have to be performed in order to further understand this phenomenon.

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