High-pressure evolution of Fe₂O₃ electronic structure revealed by x-ray absorption

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We report the high-pressure measurement of the Fe K edge in hematite (Fe₂O₃) by x-ray absorption spectroscopy in partial fluorescence yield geometry. The pressure-induced evolution of the electronic structure as Fe₂O₃ transforms from a high-spin insulator to a low-spin metal is reflected in the x-ray absorption pre-edge. The crystal-field splitting energy was found to increase monotonically with pressure up to 48 GPa, above which a series of phase transitions occur. Atomic multiplet, cluster diagonalization, and density-functional calculations were performed to simulate the pre-edge absorption spectra, showing good qualitative agreement with the measurements. The mechanism for the pressure-induced electronic phase transitions of Fe₂O₃ is discussed and it is shown that ligand hybridization significantly reduces the critical high-spin/low-spin transition pressure.

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I. INTRODUCTION

An archetypal 3d transition-metal oxide and important geological compound, α -Fe₂O₃ (hematite) undergoes a series of structural and electronic transitions at high pressure. At ambient conditions, Fe₂O₃ is an antiferromagnetic insulator and adopts the corundum structure. This structure is maintained until approximately 50 GPa whereupon it transforms to a Rh₂O₃(II)-type structure, accompanied by a 10% drop in volume. The structural transition is associated with changes in magnetic and electronic structures. X-ray K_{β} emission at ambient pressure and 72 GPa show that the magnetic moment drops from high spin (HS) to low spin (LS) at high pressure.2 Conductivity measurements indicate that an insulator to metal transition occurs between 40 and 60 GPa.³ Mössbauer spectroscopy up to 82 GPa (Ref. 3) and synchrotron Mössbauer spectroscopy at 70 GPa (Ref. 4) imply the collapse of the magnetic moments and a nonmagnetic nature of the HP phase.

The nature of these transitions has been a popular research topic over the past decade. Based on their structural study of the Rh₂O₃-II phase, Rozenberg *et al.*¹ have suggested that the charge-transfer gap closure is responsible for metallization and concurrent spin moment transition. Combined local-density approximation and dynamical meanfield theory calculations by Kuneš *et al.*⁵ have implied that the reduction in the Mott gap with pressure drives the volume collapse and structure change. This idea appears to be at odds with experimental observations of a metastable state in which the HS and high-pressure structure occur

simultaneously.⁶ Thus, despite many studies of the transitions in Fe₂O₃, the nature of the evolution of the electronic structure with pressure remains unresolved. In this paper, we implemented experimental method and theoretical approaches bringing valuable information to the problem.

A number of spectroscopic techniques have been applied to investigate the electronic configuration of 3d transition-metal compounds. Photoemission and x-ray L-edge absorption provide useful information on the 3d levels of transition metals but unfortunately, these probes cannot penetrate the high pressure cells. X-ray absorption spectroscopy (XAS) at the K edge of 3d transition elements, however, operates in the hard x-ray regime, allowing the study of the electronic structure at high pressure.

The pre-edge region of the K-edge absorption spectrum can be used to investigate 3d electrons of transition-metal compounds. In Fe-bearing compounds, the pre-edge spectra contain information about the oxidation state and local coordination. However, limited by the 1s core-hole lifetime broadening, the energy resolution using a transmission geometry is not sufficient to resolve the detailed structure of the pre-edge region. Therefore we use the partial fluorescence yield method for measuring absorption. Instead of collecting the transmitted x-ray, the $K_{\alpha 1}$ emission line is measured. This method thus has a 2p core-hole lifetime broadening of about 0.3 eV, resulting in much higher energy resolution.

Here we present the first high-pressure XAS measurement in partial fluorescence yield on Fe_2O_3 up to 64 GPa. The improved resolution of the resulting spectra shows the evolution of the Fe^{3+} 3d electronic structure as the material un-

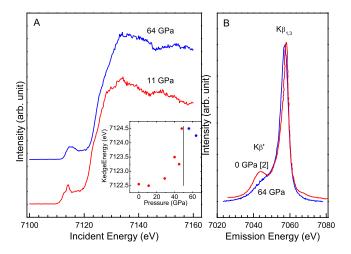


FIG. 1. (Color online) (A) X-ray K-edge absorption spectra of Fe₂O₃ in partial fluorescence yield geometry at 11 and 64 GPa; Inset: Fe K-edge position at different pressures. The edge is determined by the maximum of the first derivative of the absorption spectra. (B) X-ray K_{β} emission spectra of Fe₂O₃ at 64 and 0 GPa from Ref. 2, showing the reduction in the spin moment. Red: highspin state and blue: low-spin state.

dergoes its complex pressure-induced transitions. Previously, Caliebe *et al.* applied this technique to Fe_2O_3 , and assigned the double-peak structure of the pre-edge to the t_{2g} and e_g components of the 3d band⁸ as suggested previously.⁹ Similar methods have been used to study orbital hybridization and spin polarization of Fe_2O_3 (Ref. 10) and pre-edges of other Fe-containing compounds.¹¹

II. EXPERIMENT

 Fe_2O_3 powder was loaded in a hydrostatic pressure transmitting medium (He or Ne) in an x-ray transparent Be gasket. Ruby fluorescence was used for pressure calibration. High-pressure XAS spectra of Fe_2O_3 were collected at two-third generation synchrotron facilities. In both setups, monochromatic x-rays focused by Kirkpatrick-Baez mirrors were directed through a panoramic diamond-anvil cell, and the analyzer was fixed at 90° from the incident beam.

In the SPring-8 XAS experiment conducted at BL12XU, we scanned the incident x-ray energy from 7110 to 7145 eV with a step size of 0.1 eV and over the smaller range of 7112–7115 eV at 0.05 eV step size. In the APS setup at HPCAT 16-IDD, the entire edge was scanned from 7100 to 7160 eV with a step size of 0.25 eV. The pre-edge was scanned from 7108 to 7118 eV (7109 to 7119 eV for 56 and 64 GPa) with a step size of 0.2 eV. For both measurements, the partial fluorescence yield was collected with the analyzers set at the Fe $K_{\alpha 1}$ energy (6405.6 eV).

Figure 1(A) shows the representative Fe K-edge XAS spectra for Fe₂O₃. The partial fluorescence yield geometry allows us to resolve the pre-edge features. At the highest pressure in our study, we collected the K_{β} emission spectrum of the sample shown in Fig. 1(B). Compared with the 0 GPa spectrum of Badro *et al.*, there is a dramatic reduction in the K'_{β} satellite peak intensity in the 64 GPa spectrum, indicating a LS ground state.^{2,12}

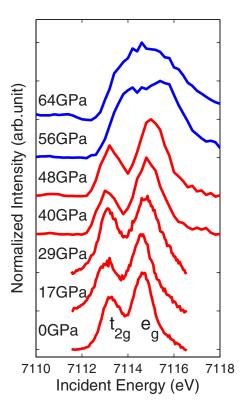


FIG. 2. (Color online) X-ray K-edge pre-edge of $\mathrm{Fe_2O_3}$ at 0, 17, 29, 40, 48, 56, and 64 GPa. The bottom three spectra are from SPring-8 using high-resolution monochromator and the top four spectra are from APS using diamond monochromator.

As shown in Fig. 1(A) inset, it is also observed that the K-edge blueshifts with pressure until the phase-transition region and remain approximately constant thereafter. This shift of K edge with pressure is also observed in other 3d transition-metal oxides, 13 a result of the increase in electron density upon compression.

Figure 2 shows the Fe K-edge pre-edge spectra of the sample from ambient pressure to 64 GPa. The tail of the main absorption edge was subtracted for each spectrum by removing the K-edge absorption spectrum of Fe in the Be gasket. The pre-edge features at ambient pressure are associated with excitations to t_{2g} and e_g orbitals, split by the octahedral crystal field. Our ambient pressure data can be fit with a crystal-field splitting energy (CFSE) of 1.4 eV, consistent with previous observation. ^{8,9} The two-peak feature in the pre-edge persists until 48 GPa, just before the phase transitions occur. By fitting the pre-edge spectra we estimate a monotonic increase in the CFSE to 1.85 eV at 48 GPa, as shown in Table I. This increase is expected as the FeO₆ octahedra shrink with pressure, and the shorter metal-ligand distance elevates the e_g level relative to the t_{2g} level.

The pre-edge spectra above the phase transitions (i.e., above 48 GPa) are more complicated to interpret. The full width at half maximum of the pre-edge features significantly broadens and a simple assignment in terms of single particle t_{2g} and e_g transitions is inconsistent; at such pressures, Fe₂O₃ is in the LS state in which e_g should be empty and five of the six t_{2g} states occupied. Such a single-particle configuration should lead to relatively small (large) $t_{2g}(e_g)$ amplitudes, un-

TABLE I. Crystal-field splitting energy (CFSE) of Fe₂O₃ as a function of pressure.

	0		1.7	20	40	40
Pressure (GPa)	0	6	17	29	40	48
CFSE (eV)	1.41	1.44	1.59	1.73	1.82	1.85

like the features observed in the pre-edge spectra at 56 and 64 GPa.

III. THEORETICAL INTERPRETATION

To understand the pressure dependence of the XAS, we first used crystal-field atomic multiplet theory to calculate the electronic structure. The relevant parameters are the atomic t_{2g} - e_g energy-level spacing 10Dq (Ref. 14) and the "Racah parameters" B and C associated with d-d interactions. The fix Racah B=0.075 eV and C=0.346 eV appropriate for solid-state Fe³⁺ systems, the fixed perform calculations for a range of 10Dq. The lowest two eigenenergies for the $(1s)^2(3d)^5$ configuration are shown in Fig. 3(a) from which a HS-LS transition is evident near 10Dq=2.2 eV. For low pressure (low Dq) the ground state has 6A_1 character (HS) and crosses over at high pressure to a state of 2T_2 character (LS). 17,18

While the critical value of 10Dq determined by the atomic multiplet calculation is larger than that suggested by the experimental t_{2g} - e_g peak splitting in Fig. 2, it is well known that the critical 10Dq for the HS-LS transition is reduced by the Fe-O charge-transfer processes. We perform calculations on a FeO₆ octahedral cluster that explicitly includes multiplets, ligand hybridization and charge-transfer via the Slater-Koster matrix elements, ^{19,20} Racah parameter A, and chargetransfer gap energy Δ . At ambient pressure, the values of the parameters are (in units of electron volt): $V_{pd\sigma} = -1.13$, $V_{pd\pi}$ =0.65, $V_{pp\sigma}$ =0.56, and $V_{pp\pi}$ =-0.16, A=5.0, 10Dq=0.96, and Δ =2.7.19 We have used the smaller value of $V_{pd\sigma}$ from Ref. 19. The lowering of the critical 10Dq is illustrated in Fig. 3(a), which shows the energies of the HS and LS states calculated in the FeO₆ cluster compared to atomic multiplet theory as a function of 10Dq. The HS to LS transition occurs at smaller 10Dq since the hybridization most strongly couples the d^5 LS state with the d^6L LS state, lying lower in energy than the d^6L HS state.

These parameters yield the ambient pressure spectra shown in Fig. 3(c), which is in good agreement with experiment (cf. Fig. 2 and Table I). The two spectral peaks separated by $\sim 1.4\,$ eV correspond to excitations into the t_{2g} and e_g orbitals, respectively, and indicate a HS ground state, with the observed CFSE coming from 10Dq plus a 0.45 eV covalent contribution. Thus while the critical 10Dq is reduced by the Fe-O charge-transfer processes, the ligand field splitting due to covalency pushes up the spectral t_{2g} - e_g peak separation of the XAS spectra. 8

With parameters set to reproduce ambient spectra, we consider the pressure evolution of the HS-LS transition and the XAS spectra. As the pressure increases, both 10Dq and the hopping integrals increase, respectively having $\sim d^{-5}$ and $\sim d^{-4}$ Fe-O bond-length dependence. ^{14,19} The combined effect of pressure-dependent hopping and 10Dq is explained in

the phase diagram of Fig. 3(b). We consider several variations in $V_{pd\sigma}$ with d as shown in Fig. 3(b), which all indicate that the critical pressure occurs between 52 and 55 GPa. Although variation in the exponent of $V_{pd\sigma}$ induces a variation on the order of 5% in the predicted critical pressure it is striking to observe that the experimentally observed limits on the critical pressure are in general agreement with theoretical predictions.

Figures 3(c)-3(e) show the calculated pre-edge XAS spectra from the FeO₆ cluster at various pressures. The spectrum at 48 GPa [Fig. 3(d)] shows a clear two peak structure in the HS state, with a t_{2g} - e_g peak separation of ~1.6 eV. The calculated CFSE is ~15% smaller in energy than experiment, which may be in part due to structural deviations from

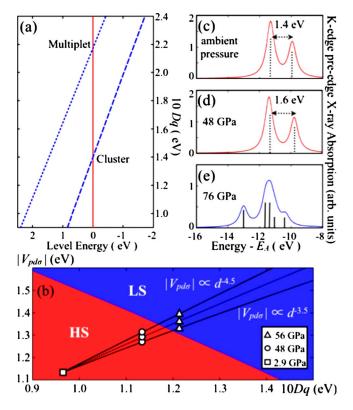


FIG. 3. (Color online) (a) Energy of LS state for the single atom multiplet calculation (dotted line) compared with the FeO₆ cluster diagonalization (dashed line) relative to the HS state (solid line). (b) HS-LS phase diagram for Fe₂O₃. The dotted line shows the probable trajectory of $(10Dq, V_{pd\sigma})$ with increasing pressure (see text). [(c)-(e)] K-edge pre-edge XAS spectra from the FeO₆ cluster calculation at various pressures; E_A is the Fe K-edge absorption energy. (c) At ambient pressure, the spectrum shows distinct t_{2g} - e_g absorption peaks separated by 1.4 eV, indicating a high-spin ground state. (d) At 48 GPa, the peak separation is 1.6 eV, and the ground state still resides in the high-spin sector. (e) At 76 GPa, the spectrum shows broad, multiple peaks, indicating a low-spin ground state. All the spectra were broadened with a 0.3 eV Lorentzian.

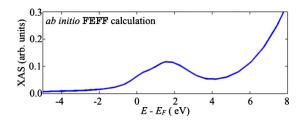


FIG. 4. (Color online) *Ab initio* calculations with the FEFF software of the pre-edge region of the Fe K-edge XAS spectra for the high-pressure metallic Rh₂O₃(II)-type structure of Fe₂O₃. E_F is the Fermi level.

octahedral symmetry giving inequivalent Fe-O bonds not included in the cluster calculation, 1 as well as the overall uncertainty in cluster parameters. Figure 3(e) shows the calculated XAS spectra at 76 GPa. The high-pressure spectra have multiple-peak features indicating a LS ground state; this qualitative change in character of the ground state is reflected as a qualitative change in the calculated spectra. It is the simple transformation properties (A_1) of the HS state that allow the XAS to be interpreted in terms of single-particle t_{2g} and e_g levels; the final state, with one additional d electron, transforms as $A_1 \otimes (T_2 \oplus E) = T_2 \oplus E$ mimicking the single-particle t_{2g} and e_g levels. On the other hand, addition of a d electron to the LS state yields $T_2 \otimes (T_2 \oplus E) = A_1 \oplus E \oplus T_1 \oplus T_2 \oplus T_1 \oplus T_2$ resulting in more peaks than would be expected based on a single-particle interpretation.

While an insulator-metal transition is not necessarily concomitant with a change in the local-spin configuration (and vice versa), a low-spin metallic state is always expected at a high enough pressure. In this regime, we use the all-electron FEFF code^{21–23} to calculate the high-pressure Fe *K*-edge XAS for a large cluster of 152 atoms in the high-pressure structure. Figure 4 shows the calculated pre-edge XAS, having broad pre-edge features in qualitative agreement with the experiment at and above 56 GPa.

We last turn to the electronic phase-transition mechanism. Badro $\it et al.$ have shown the coexistence of HS and Rh_2O_3 -II structure indicating that the electronic transition cannot drive the structural transition. Kuneš $\it et al.$ divided the electronic transition into a Mott gap closing and a HS-LS gap closing, and estimated the respective regimes of stability via a local "density-based interaction." Here we have indicated the importance of atomic multiplets and ligand hybridization. Our results indicate the location of the HS-LS transition can be well described within the charge-transfer multiplet-

hybridization cluster approach and reasonable choices for the pressure dependence of the cluster parameters. The reduction in the critical pressure for the HS-LS transition in comparison with atomic multiplet theory due to ligand hybridization is seen to be significant. These results lead to the prediction that the critical pressure occurs between 52 and 55 GPa, at values of 10Dq much smaller than would be expected from atomic multiplet theory based on the experimental spectra. While our cluster calculation cannot address in detail the closing of a bulk Mott gap, the observed reduction in the HS-LS transition pressure leads us to suggest that the physics of a local HS-LS transition should be strongly reconsidered as the key ingredient giving the evolution of spectral features observed in the pre-edge XAS spectra with pressure.

In summary, we measured x-ray absorption spectra of Fe₂O₃ up to 64 GPa, and experimentally resolved the crystal-field splitting and its pressure dependence through the metal-insulator transition. The CFSE increases from 1.41 eV at ambient conditions to 1.85 eV at 48 GPa. The pre-edge features change drastically at higher pressures corresponding to the range where a number of electronic and structural transitions have been reported. We constructed the phase diagram for Fe₂O₃ which shows that the changes in multiplet structure and hybridization are important for a quantitative estimate of the critical pressure. Based on considerations of local cluster physics, excellent agreement between the observed pressure dependence of the experimental and calculated spectra were obtained.

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