Antiferromagnetic magnetic transition and spin fluctuations in the deformed pyrochlore compound β -Fe₂(OH)₃Cl

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Magnetism of the rhombohedral structure $Fe_2(OH)_3Cl [\beta-Fe_2(OH)_3Cl]$, a member of the geometric frustration series $M_2(OH)_3X$ where the magnetic ions form a deformed pyrochlore lattice, was studied using dc and ac susceptibilities, heat-capacity, neutron powder-diffraction, and muon spin relaxation (μ SR) measurements. Antiferromagnetic transition was observed at T_N =9.0 K with a much larger Curie-Weiss temperature θ_w of approximately -64 K, suggesting that $Fe_2(OH)_3Cl$ is a geometrically frustrated magnet. Neutron-diffraction studies revealed the transition into a long-range antiferromagnetic order below T_N =9.0 K with the Fe²⁺ spins on the triangular lattice planes disordered, and the other ones on the kagome lattice planes having frozen moments of less than 70% of the full Fe²⁺ moment. Spin fluctuations below T_N were verified further using μ SR studies. The magnetically phase-separated feature and the spin fluctuation coexisting with a long-range magnetic order are considered to be of great interest for studying geometric frustration and the behavior of S=2spin systems on prochlore and kagome lattices.

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I. INTRODUCTION

Geometrically frustrated magnetic materials show novel magnetic properties because of their lattice geometry.¹ The lattice geometries responsible for frustration are generally triangular lattice, kagome lattice, and pyrochlore lattice. Of particular recent interest are the rare-earth pyrochlore compounds, in which the magnetic ions form networks of cornershared tetrahedrons and the frustration of spin-spin interactions engenders spin ice states, unconventional spin-glass states, and exotic spin liquid states.^{2–9}

In recent years, we identified unconventional magnetic transitions in a transition-metal hydroxyhalogenide series of deformed pyrochlore compounds $M_2(OH)_3 X$, where M represents a *d*-electron magnetic ion such as Cu⁺², Ni⁺², Co⁺² Fe⁺², and Mn⁺² and X represents the halogen ions of Cl⁻¹, Br⁻¹, or I⁻¹.¹⁰⁻¹⁴ Much of these metal hydroxyhalogenides were found originally in natural minerals. This material category presents a complete series for spins S=1/2 to S=5/2 $(S=1/2 \text{ for } Cu^{+2}, 1 \text{ for } Ni^{+2}, 3/2 \text{ for } Co^{+2}, 2 \text{ for } Fe^{+2}, \text{ and } 5/2$ for Mn⁺²). In particular, clinoatacamite, Cu₂(OH)₃Cl, transitions at $T_{\rm N1}$ =18.1 K with a very small entropy release, $T_{\rm N2}$ =6.4 K and $T_{\rm N3}$ =6.2 K with coexisting order and spin fluctuation. The nature of the $T_{\rm N1}$ =18.1 K phase remains largely unknown and attracts much interest.^{15–17} Furthermore, it is the first example of the S=1/2 Heisenberg quantum spin on a pyrochlore lattice and the parent compound for the substituted "perfect kagome lattice" of herbertsmithite $ZnCu_3Cl_2(OH)_6$, which exhibits spin liquid behavior.^{18,19}

 $Fe_2(OH)_3Cl$ is the S=2 spin system of the transition-metal hydroxyhalogenide series. As exemplified in the Cu²⁺ series of atacamite, clinoatacamite, and botallackite,¹¹ the transition-metal hydroxyhalogenide series usually have multiple polymorphous structures. Three crystal structures of $Fe_2(OH)_3Cl$ have been reported: the hexagonal

α-Fe₂(OH)₃Cl, the rhombohedral β-Fe₂(OH)₃Cl,^{20,21} and the orthorhombic γ-Fe₂(OH)₃Cl (Hibbingite).²² As shown in the present work, the Fe²⁺ ions in β-Fe₂(OH)₃Cl form a deformed pyrochlore lattice like the rhombohedral Co₂(OH)₃Cl, in which we observed geometric frustration.¹² Only a few works in the fields of mineralogy, meteorology, and corrosion science have examined Fe₂(OH)₃Cl compounds. The present work reports the magnetism and the detailed structure information of the deformed pyrochlore lattice β-Fe₂(OH)₃Cl.

II. EXPERIMENTAL PROCEDURES

White powders of polycrystalline samples of β -Fe₂(OH)₃Cl [Fe₂(OD)₃Cl for neutron diffraction] were prepared using a hydrothermal reaction of aqueous solutions of 3.5M FeCl₂ 6H₂O [dried FeCl₂ for Fe₂(OD)₃Cl] and 5M NaOH [NaOD for Fe₂(OD)₃Cl] at around 80 °C for several hours in a nitrogen-gas atmosphere. Exposure to air was avoided because β -Fe₂(OH)₃Cl would thereby be oxidized, changing color quickly to brownish in air. The crystal structure was investigated using synchrotron x-ray diffraction at the BL02B2 beam line at the SPring-8 synchrotron using a Debye-Scherer camera with an imaging plate.²³ The incident x-ray beam was monochromatized using a double-crystal monochromator tuned to the wavelength of approximately 0.5 Å (precisely determined using standard sample of CeO₂ to be 0.50217 Å). Powder-diffraction patterns were collected over the 2θ range of $3^{\circ} - 75^{\circ}$ in a step angle of 0.01° . The unit-cell parameters were first calculated roughly from observed d value and hkl peak data of the powder-diffraction data using a least-squares method. Then they were refined using the Rietveld method with a computer program RIETAN-2000.²⁴ The dc and ac susceptibilities measurements were taken using a commercial superconducting quantum in-



FIG. 1. (Color online) Synchrotron x-ray diffraction pattern (filled circles) for $Fe_2(OH)_3Cl$ at 300 K and the result of Rietveld structure refinements showing the calculated (thick solid line) pattern and the difference (thin solid line below the XRD spectra).

terference device magnetometer (MPMS; Quantum Design). The heat capacity was measured using an adiabatic heatpulse method with a ³He cryostat using an amount of 0.4 g of the polycrystalline powder. A neutron powder-diffraction experiment was performed with a wavelength of 1.8264 Å using a Kinken powder diffractometer, HERMES, of the Institute for Materials Research (IMR), Tohoku University, installed at the JRR-3M reactor at the Japan Atomic Energy Research Institute (JAERI), Tokai.²⁵ The collected neutron data were refined using the program FULLPROF suite based on Rietveld refinement.²² Muon spin rotation/relaxation (μ SR) experiments were conducted on M20 port at TRIUMF (Canada) using a positive surface muon beam with a conventional He-gas flow cryostat. The positive muon beam, polarized to the beam direction, is stopped in the specimen and depolarized by its internal field. The time histograms of muon decay positrons are recorded by forward (F) and backward (B) counters as a function of resident time for each μ^+ within the specimen. A positron is emitted preferentially toward the muon spin direction. Therefore, asymmetry a(t)=[F(t)-B(t)]/[F(t)+B(t)] of the two histograms (after correction for solid-angle effects) reflects the time evolution of muon spin polarization, as well as the time evolution and space distribution of the internal field in the specimen.

III. RESULTS AND DISCUSSION

A. Structure characterization and magnetic properties

The rhombohedral Fe₂(OH)₃Cl was synthesized in single phase, as demonstrated by the synchrotron x-ray diffraction pattern and the Rietveld fit presented in Fig. 1. Consistent with Ref. 20, it is a rhombohedral structure in the space group $R\overline{3}m$, the lattice parameters was refined to be a =6.94139(8) Å and c=14.72709(21) Å. The positions of the individual atoms were also determined as presented in Table I. For determination of the hydrogen atoms neutron powder-diffraction data on Fe₂(OD)₃Cl at 12 K was analyzed similarly using the Rietveld fit. The results are presented in Table II. The magnetic ions of Fe^{2+} in β -Fe₂(OH)₃Cl form a three-dimensional network of linked tetrahedrons that can be viewed as alternative stacking layers of kagome and triangular lattice planes in the (001) direction (Fig. 2). A prominent structural feature is a notable distortion in the tetrahedron where the Fe^{2+} - Fe^{2+} distance on the kagome lattice planes

TABLE I. Structure information of synthetic β -Fe₂(OH)₃Cl refined from synchrotron x-ray diffraction data obtained at 300 K.

Chemical formula				Fe ₂ (OH) ₃ Cl						
	Cell setting				Rhombohedral					
	Space group a (Å)				<i>R</i> 3 m (no. 166) 6.94139(8)					
c (Å) V (Å ³)				14.72709(21) 614.526 (13)						
										$R_{wp}(S)$ R_{p}
	4.33									
	x	у	z	g	В	Site	Sym.			
Fe1	0.50000	0.00000	0.00000	1.000	0.807(16)	9e	2/m			
Fe2	0.00000	0.00000	0.50000	1.000	0.807(16)	3b	$\overline{3}m$			
Cl	0.00000	0.00000	0.21620(13)	1.000	0.950(33)	6c	3m			
0	0.19899(17)	0.39799	0.07180(13)	1.000	0.968(66)	18h	m			
		Bond length								
		(Å)								
	$Fe_K - Fe_K$			3.47069						
	Fe_T - Fe_K			3.16858						

Chemical formula Cell setting				Fe ₂ (OD) ₃ Cl Rhombohedral					
	Space grou	ıp	<i>R</i> 3 m (no. 166)						
<i>a</i> (Å)			6.92782(16)						
c (Å)			14.65471(44)						
$R_{ m wp}(\chi^2)$			5.60 (3.37)						
$R_{\rm p}$			4.22						
	$R_{\rm exp}$	3.05							
	x	y	z	g	В	Site	Sym.		
Fe1	0.50000	0.00000	0.00000	1.000	0.499(21)	9e	2/m		
Fe2	0.00000	0.00000	0.50000	1.000	0.499(21)	3b	3m		
Cl	0.00000	0.00000	0.21810(21)	1.000	0.499(21)	6c	3m		
0	0.20056(45)	0.40110(91)	0.06874(20)	1.000	0.499(21)	18h	m		
D	0.12932(46)	0.25864(93)	0.09907(21)	1.000	1.841(63)	18h	m		

TABLE II. Structure information of synthetic β -Fe₂(OD)₃Cl refined from neutron-diffraction data at 12 K.

are 3.47069(2) Å whereas the distance connecting the Fe²⁺ ion on the kagome lattice planes and that on the triangular lattice planes is 3.16858(2) Å.

The temperature dependence of dc magnetic susceptibility is depicted in Fig. 3. The inset plot is the susceptibility measured at a weak field under zero-field-cooled (ZFC) and field-cooled (FC) conditions, respectively. The two curves diverted from the high-temperature end with much higher values for FC, suggesting that there are some filed-induced moments. Antiferromagnetic transition at $T_N=9$ K was observed. The susceptibilities at high temperatures obey the Curie-Weiss law, $\chi^{-1}(T)=(T-\theta_w)/C$, with a Weiss temperature of θ_w of approximately -64 K. The large ratio of θ_w/T_N indicates magnetic frustration. The Curie constant corresponds to an effective moment of $5.3\mu_B$, which is consistent



with the spin S=2 of Fe²⁺. It is worth noting that although the Fe₂(OH)₃Cl has the same crystal structure as Co₂(OH)₃Cl,¹² the two systems show contrasting magnetic properties of antiferromagnetic and ferromagnetic transitions. The ac susceptibilities also confirmed the transition at $T_N=9$ K. The heat capacity C_p as presented in Fig. 4 confirmed the magnetic transition with little dependence on the applied field. Considering the low $T_N=9$ K, a larger field dependence is usually expected. This suggests that the antiferromagnetic state in Fe₂(OH)₃Cl is stable with respect to the external magnetic field. It is worth noting that this behavior differs greatly from the Co₂(OH)₃Cl (Ref. 12) and Co₂(OH)₃Br of the same crystal structure, where external



FIG. 2. (Color online) Crystal structure of $Fe_2(OH)_3Cl[Fe_2(OD)_3Cl]$ featuring the alternatively stacked layers of the kagome and triangular lattice planes along the *c*-axis direction for Fe^{2+} spins.

FIG. 3. (Color online) Temperature dependence of the dc susceptibilities χ (open circles, left axis) and the inverse susceptibilities $1/\chi$ (open squares, right axis) per mole formula, measured at 5 kOe. The solid line obeys the Curie-Weiss law, $\chi^{-1}(T)=(T - \theta_w)/C$, with a Weiss temperature of θ_w of approximately -64 K. The inset plot, respectively, shows the susceptibility curves measured at 100 Oe under FC and ZFC conditions.



FIG. 4. (Color online) Temperature dependence of total specific heat C(T) per mole of Fe₂(OH)₃Cl under a zero field, 10 kOe, and 50 kOe, respectively. The small upturn at around 14 K was caused by the inclusion of a polymorphous phase of Fe₂(OH)₃Cl, which undergoes an antiferromagnetic transition at T_N =14 K as we previously reported in Ref. 13. It is hard to subtract the lattice contribution from the specific heat, therefore, the magnetic entropy is not estimated.

magnetic field induced a ferromagnetic state. The stability of the antiferromagnetism in $Fe_2(OH)_3Cl$ is considered to be related to its magnetic spin structure, as is discussed below.

B. Possible magnetic structures from neutron diffraction

Neutron-diffraction patterns obtained with the powder sample of Fe₂(OD)₃Cl at 12 K and 3.5 K, respectively, are depicted in Fig. 5. Long-range antiferromagnetic order developed below the transition temperature. The magnetic reflections can be indexed by $k=(0\ 0\ 3/2)$, as marked in Fig. 5. Consistent with the susceptibility measurements, a broad



FIG. 5. (Color online) Neutron-diffraction patterns obtained, respectively, for $Fe_2(OD)_3Cl$ at 12 and 3.5 K. The difference between 12 and 3.5 K is presented on the bottom depicting the magnetic reflection peaks. The inset is an enlarged plot for the low-angle part.



FIG. 6. (Color online) [(a) and (b)] Proposed magnetic structures with spin order only occurring on the kagome lattice planes. (c) The calculated neutron-diffraction pattern using either of the magnetic structures for $Fe_2(OD)_3Cl$.

peak (the inset plot) is apparent at 12 K and higher temperatures, suggesting a short-range spin correlation well above the $T_{\rm N}$.

Possible magnetic structures were explored using a representation analysis technique.²⁶ The intensities of the magnetic reflection peaks are relatively low as compared with the full moment of Fe²⁺. Good fit was obtained by assuming the spin structures illustrated in Figs. 6(a) and 6(b), respectively, in which the Fe²⁺ spin order occurs on the kagome lattice planes but with the spin directions aligned differently. Equivalent goodness of fitting was obtained for the two spin configurations, therefore, further determination only can be made with a single-crystal experiment. The results of the Rietveld refinement of the 3.5 K diffraction pattern for one of them are shown in Fig. 6(c). Another type of candidate structures is shown in Figs. 7(a) and 7(b). The Fe²⁺ on the triangular lattice planes also shows a frozen moment while the Fe²⁺ ions on the kagome lattice planes keep any of the two configurations shown in Figs. 6(a) and 6(b). The Rietveld refinement results are shown in Fig. 7(c). For the two models portrayed in Figs. 6(a) and 6(b) only the three Fe²⁺ ions on the kagome lattice plane are ordered with two of them having magnetic moments of $3.70\mu_B$ and the third one $3.36\mu_{\rm B}$. For the models presented in Figs. 7(a) and 7(b) the Fe²⁺ ion on the triangular lattice plane has a small ordered moment of $1.05\mu_{\rm B}$, and the two Fe²⁺ ions on the kagome lattice plane have moments of $3.57 \mu_{\rm B}$ and the third one $3.62\mu_{\rm B}$. As a matter of fact, the two models in Figs. 7(a) and 7(b) can be viewed as slightly modified version of those



FIG. 7. (Color online) [(a) and (b)] Proposed magnetic structures with spin order occurring both on the kagome lattice planes as well as the triangular lattice planes. (c) The calculated neutron-diffraction pattern using the proposed magnetic structures for $Fe_2(OD)_3Cl$.

shown in Figs. 6(a) and 6(b), the moment on the triangular lattice plane compensated for the decreased moments on the kagome lattice plane. As the refinement plots show, it is difficult to determine which one is the correct magnetic structure from the powder-diffraction data. Nevertheless, in all models, the ordered magnetic moments are mostly residing on the kagome lattice planes. Furthermore, they are remarkably smaller than the full moment of Fe²⁺ (5.3 $\mu_{\rm B}$, as estimated from the magnetic susceptibilities). This fact suggests almost disordered Fe²⁺ on the triangular lattice planes, as well as remaining spin fluctuation for the ordered Fe^{2+} on the kagome lattice planes. The image of weakly coupled kagome lattice planes is drastically different from that of Co₂(OH)₃Cl (Ref. 12) and $Co_2(OH)_3Br$ which share the same crystal structure as $Fe_2(OH)_3Cl$. In $Co_2(OH)_3Cl$ the spins on the triangular lattice planes are ferromagnetically ordered while in antiferromagnetic Co₂(OH)₃Br external magnetic field induced transitions into ferromagnetic states. The stability of the antiferromagnetic state in $Fe_2(OH)_3Cl$, as shown by the specific heat in Fig. 4, may be related to its Heisenberg-type two-dimensional magnetic structure.

The reason why the Fe²⁺ spins on the triangular lattice planes have stronger fluctuation than those on the kagome lattice planes can be found in its lattice structure. As seen in Fig. 2, the three Fe^{2+} ions (denoted by Fe_K) on the kagome lattice planes are in a highly symmetric position surrounded by nearby O⁻ and Cl⁻ ions but the Fe²⁺ ion (denoted by Fe_T) on the triangular lattice plane is only surrounded by O⁻. The distance of the Fe_{K} -O bonding is 2.123 Å and the bonding angle of Fe_K -O-Fe_K is 109.7° while the Fe_T-O bonding has a longer distance of 2.135 Å and a smaller angle of 96.17° for Fe_T-O-Fe_K. If we assume superexchange interactions for the antiferromagnetism through the Fe-O-Fe bonding, apparently one can expect stronger coupling for the Fe_K ions on the kagome lattice planes. Moreover, from our previous studies on the hydroxyhalogenide series we learned that larger halogen ions effectively enhanced the superexchange coupling. For example, in botallackite-structure $Cu_2(OH)_3X$ (X



FIG. 8. (Color online) LF-µSR asymmetries for polycrystalline Fe₂(OH)₃Cl at 100 K, 20 K, and 1.8 K, respectively.

=Cl,Br,I) [although their crystal structure is different from the present $Fe_2(OH)_3Cl$, the Cu-X bonding has a similar coordination as the Fe-Cl in $Fe_2(OH)_3Cl$], the T_N increased from $T_N = 7.2$ K for $Cu_2(OH)_3Cl$ to $T_N = 10$ K for $Cu_2(OH)_3Br$ and $T_N=14$ K for $Cu_2(OH)_3I$, respectively.¹⁴ The enhanced antiferromagnetic coupling by larger halogen ions was also demonstrated in the case of pyrochlore Co₂(OH)₃Cl and Co₂(OH)₃Br, which show the same crystal structure as the present Fe₂(OH)₃Cl. Replacing Cl with Br leads to transformation from a partial ferromagnetic order in $Co_2(OH)_3Cl$ into an antiferromagnetic order in $Co_2(OH)_3Br$ [Ref. 13 and unpublished data]. All of the above-mentioned results suggest strong superexchange interactions through the halogen ions. Therefore, we can safely conclude that strong coupling through the Fe_K-Cl-Fe_K bonding dominates in the present material Fe₂(OH)₃Cl.

C. Investigation of spin dynamics with μ SR

As many μ SR studies have demonstrated, in addition to being a direct probe showing magnetic order, the positive muon μ^+ is a sensitive probe to investigate spin fluctuation in the frustrated magnetic materials.^{6,12,18,27-30} In the magnetically ordered state of a polycrystalline sample, the longitudinal polarization function $P_Z(t)$ is expected to be the weighted sum of longitudinal and transverse components: $P_Z(t)$ =[exp($-\lambda_Z t$)+2 exp($-\lambda_X t$)cos($\gamma_{\mu} \langle B_{loc} \rangle t$)]/3. Therein, λ_Z and λ_X , respectively, denote the spin-lattice and spin-spin relaxation rates, γ_{μ} is the muon gyromagnetic ratio ($\gamma_{\mu}=2\pi$ ×135.5 MHz/T), and $\langle B_{loc} \rangle$ stands for the mean value of the local field at the muon site.

Therefore, we further studied the spin dynamics using zero-field (ZF) and longitudinal field (LF) muon spin relaxation (μ SR) measurements. In Fe₂(OH)₃Cl, four possible stopping sites exist for the positive muons: one near the Cl and three near the three O ions. The positive muons feel the local fields produced by the nuclei moments and the electron spins of Fe²⁺. In a paramagnetic state, when the magnetic coupling of the electron spins is not present, the field of the electron spins is fluctuating too fast to be detected by the muons. Therefore, the muons only detect those nuclei fields which look static to the muons. The muon spin depolarization is expressed using a static Kubo-Toyabe function. It is decoupled in a weak external field. Some typical ZF and LF μ SR asymmetries at representative temperatures are presented in Fig. 8. Even at 100 K, muon spin depolarization was observed, which remained in a strong magnetic field of 2400 G, suggesting the existence of dynamic field, i.e., the existence of Fe²⁺-Fe²⁺ spin coupling, at high temperatures. The decoupled proportion decreased to approximately 20% at 20 K (Fig. 8), which is attributed to the muons on the Cl site distant from the Fe²⁺ spins (refer to the crystal structure in Fig. 2).

The ZF- μ SR asymmetries at typical temperatures are presented in Fig. 9. The depolarization accelerates quickly below around 30 K. The ZF- μ SR asymmetry for $T > T_N$ are



FIG. 9. (Color online) ZF- μ SR asymmetries for polycrystalline Fe₂(OH)₃Cl at typical temperatures. The solid lines are calculated to fit as described in the text.



FIG. 10. (Color online) Temperature dependences of the fitted parameters. The inset plot shows the estimated local fields in the dynamic Kubo-Toyabe function.

best fit by two exponential functions, $a_0P_Z(t)=a_1 \exp(-\lambda_1 t)$ + $a_2 \exp(-\lambda_2 t)$, wherein the two depolarization rates differ remarkably, as presented in Fig. 10. When the temperature approaches $T_N=9$ K, both depolarization rates increased rapidly, implying a critical slowing toward T_N .

For $T < T_N$, the best fit was obtained with the sum of two exponential functions and a dynamic Kubo-Toyabe function as

$$a_0 P_Z(t) = \exp(-\lambda_0 t) + a_s \exp(-\lambda_s t) + a_D G^{DGKT}(t, \Delta, \nu)$$

where $\lambda_{\rm q}$ denotes a quick depolarization and $\lambda_{\rm s}$ a slow one. Here, $G^{\rm DGKT}(t,\Delta,\nu)$ is the dynamic Kubo-Toyabe relaxation function, where Δ is a measure of the local field by $\Delta = \sqrt{B_{\rm loc}^2}$, ν is a reciprocal of $\tau_{\rm c}$, which is the interval of time during which the muons detect the change in the local magnetic field.³¹ When Δ/ν is approximately 1, $G^{\rm DGKT}(t,\Delta,\nu)$ is approximately expressed by the exponential function of $\exp(-\Delta^2 t/\nu)$. A small Δ of approximately 50 G was estimated as shown in the inset plot in Fig. 10. Based on the depolarization proportion as well as the small Δ , we presume that the dynamic Kubo-Toyabe function represents the muon spin depolarization on the Cl sites. The other two exponential functions therefore reflect the strong local fields near the O sites immediately outside of the Fe²⁺ tetrahedron.

It is worth noting that although the long-range magnetic order is clearly demonstrated, no oscillations responding to it were observed, which differs from $Co_2(OH)_3Cl$, for which oscillations of 50 MHz were observed.¹² Instead, the μ SR data suggest spin fluctuation on different time scales. The oscillation signals might be absent because they are wiped by the dynamical fields or the magnetic order is a novel "dynamical long-range order," as proposed for the "ordered spin ice:" Tb₂Sn₂O₇.^{29,30} The μ SR result is consistent with the proposed models for the magnetic structure by the neutron-diffraction study, which suggests a remaining spin fluctuation of drastically different spins (spins with zero or a small mo-

ment of $1.05\mu_{\rm B}$ on the triangular lattice planes, and spins with $3.36\mu_{\rm B}-3.70\mu_{\rm B}$ on kagome lattice planes). Apparently, the spin fluctuation on the triangular planes is quicker than those on the kagome lattice planes. The fact that the quick depolarization rate $\lambda_{\rm q}$ for $T < T_{\rm N}$ is continuous with the $\lambda_{\rm 1}$ in the paramagnetic phase, implies a paramagneticlike state for the Fe²⁺ spins on the triangular planes. Therefore, the models illustrated in Figs. 6(a) and 6(b) are favored.

IV. CONCLUSION

In summary, the rhombohedral structure Fe₂(OH)₃Cl $[\beta$ -Fe₂(OH)₃Cl] was synthesized in single phase and the crystal lattice was refined. The magnetic ions of Fe²⁺ in β -Fe₂(OH)₃Cl form a three-dimensional network of linked tetrahedra that can be viewed as alternatively stacked layers of kagome and triangular lattice planes along the (001) direction. Antiferromagnetic transition was observed at $T_{\rm N}$ =9.0 K with a much larger Curie-Weiss temperature of $\theta_{\rm w}$ of approximately -64 K, suggesting that $Fe_2(OH)_3Cl$ is a geometrically frustrated magnet. Neutron-diffraction studies verified the transition into a long-range antiferromagnetic order below $T_{\rm N}$ =9.0 K, for which possible magnetic structures are proposed. Of the four Fe²⁺ spins on a tetrahedron the one spin on the triangular lattice plane are disordered while the other three spins on the kagome lattice plane have frozen moments of less than 70% of the full Fe^{2+} moment. The most important results are the magnetically phase separation and the spin fluctuations coexisting with the long-range order. It is interesting that although the β -Fe₂(OH)₃Cl compound has a three-dimensional pyrochlore crystal structure, magnetically it can be viewed as an S=2 spin system with twodimensional kagome correlations. The chemical bonding for the Fe²⁺ on the triangular lattice planes are different from those of the Fe²⁺ on the kagome lattice planes, which implies that selective substitution by nonmagnetic ions is possible. Such efforts are in progress to realize a chemically twodimensional kagome lattice $Fe_3Zn(OH)_6Cl_2$ and $Fe_3Mg(OH)_6Cl_2$. It is worth noting that the transition occurs in a readily accessible range of temperatures, which is expected to facilitate future studies of spin correlation in geometrically frustrated systems.

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- ¹For recent reviews see *Frustrated Spin Systems*, edited by H. T. Diep (World Scientific, Singapore, 2004); see also J. F. Sadoc and R. Mosseri, *Geometrical Frustration* (Cambridge University Press, Cambridge, England, 1999), reedited 2007.
- ²S. T. Bramwell and M. J. P. Gingras, Science **294**, 1495 (2001).
- ³M. J. Harris, S. T. Bramwell, D. F. McMorrow, T. Zeiske, and K. W. Godfrey, Phys. Rev. Lett. **79**, 2554 (1997); M. J. Harris, S. T. Bramwell, P. C. W. Holdsworth, and J. D. M. Champion, *ibid.* **81**, 4496 (1998); S. T. Bramwell and M. J. Harris, J. Phys.: Condens. Matter **10**, L215 (1998).
- ⁴A. P. Ramirez, A. Hayashi, R. J. Cava, R. Siddharthan, and B. S. Shastry, Nature (London) **399**, 333 (1999).
- ⁵J. S. Gardner, B. D. Gaulin, S.-H. Lee, C. Broholm, N. P. Raju, and J. E. Greedan, Phys. Rev. Lett. **83**, 211 (1999).
- ⁶J. S. Gardner, S. R. Dunsiger, B. D. Gaulin, M. J. P. Gingras, J. E. Greedan, R. F. Kiefl, M. D. Lumsden, W. A. MacFarlane, N. P. Raju, J. E. Sonier, I. Swainson, and Z. Tun, Phys. Rev. Lett. **82**, 1012 (1999).
- ⁷J. R. Stewart, J. S. Gardner, Y. Qiu, and G. Ehlers, Phys. Rev. B 78, 132410 (2008).
- ⁸H. D. Zhou, C. R. Wiebe, J. A. Janik, L. Balicas, Y. J. Yo, Y. Qiu, J. R. D. Copley, and J. S. Gardner, Phys. Rev. Lett. **101**, 227204 (2008).
- ⁹I. Mirebeau, I. Goncharenko, H. Cao, and A. Forget, Phys. Rev. B **80**, 220407 (2009).
- ¹⁰X. G. Zheng and E. S. Otabe, Solid State Commun. **130**, 107 (2004); X. G. Zheng and C. N. Xu, *ibid.* **131**, 509 (2004).
- ¹¹X. G. Zheng, T. Mori, K. Nishiyama, W. Higemoto, H. Yamada, K. Nishikubo, and C. N. Xu, Phys. Rev. B **71**, 174404 (2005); X. G. Zheng, T. Kawae, Y. Kashitani, C. S. Li, N. Tateiwa, K. Takeda, H. Yamada, C. N. Xu, and Y. Ren, *ibid.* **71**, 052409 (2005); X. G. Zheng, M. Hagihala, T. Kawae, and C. N. Xu, *ibid.* **77**, 024418 (2008).
- ¹²X. G. Zheng, H. Kubozono, K. Nishiyama, W. Higemoto, T. Kawae, A. Koda, and C. N. Xu, Phys. Rev. Lett. **95**, 057201 (2005); X. G. Zheng, T. Kawae, H. Yamada, K. Nishiyama, and C. N. Xu, *ibid.* **97**, 247204 (2006).
- ¹³M. Hagihala, X. G. Zheng, T. Toriyi, and T. Kawae, J. Phys.: Condens. Matter **19**, 145281 (2007).
- ¹⁴X.-G. Zheng, and K. Nishiyama, Physica B **374-375**, 156 (2006); X. G. Zheng, M. Hagihala, K. Nishiyama, and T. Kawae, *ibid.* **404**, 677 (2009); X. G. Zheng, T. Yamashita, M. Hagihala,

and T. Kawae, *ibid.* **404**, 680 (2009); M. Fujihala, M. Hagihala, X.-G. Zheng, and T. Kawae, *ibid.* **404**, 674 (2009); M. Hagihala, X.-G. Zheng, and T. Kawae, *ibid.* **404**, 671 (2009); X. G. Zheng, M. Hagihala, K. Nishiyama, and T. Kawae, J. Magn. Magn. Mater. **310**, 1288 (2007); X. G. Zheng, M. Hagihala, M. Fujihala, and T. Kawae, J. Phys.: Conf. Ser. **145**, 012034 (2009).

- ¹⁵S.-H. Lee, H. Kikuchi, Y. Qiu, B. Lake, Q. Huang, K. Habicht, and K. Kiefer, Nature Mater. 6, 853 (2007).
- ¹⁶A. S. Wills and J.-Y. Henry, J. Phys.: Condens. Matter **20**, 472206 (2008); J. Phys.: Conf. Ser. **145**, 012056 (2009).
- ¹⁷S. Maegawa, A. Oyamada, S. Sato, M. Nishiyama, T. Itou, and X. G. Zheng, J. Phys.: Conf. Ser. **145**, 012018 (2009).
- ¹⁸P. Mendels, F. Bert, M. A. de Vries, A. Olariu, A. Harrison, F. Duc, J. C. Trombe, J. S. Lord, A. Amato, and C. Baines, Phys. Rev. Lett. **98**, 077204 (2007).
- ¹⁹M. A. de Vries, J. R. Stewart, P. P. Deen, J. O. Piatek, G. J. Nilsen, H. M. Rønnow, and A. Harrison, Phys. Rev. Lett. **103**, 237201 (2009).
- ²⁰H. R. Oswald and W. Feitknecht, Helv. Chim. Acta 47, 272 (1964).
- ²¹G. Springer, Can. Mineral. 27, 311 (1989).
- ²²B. Saini-Eidukat, H. Kucha, and H. Keppler, Am. Mineral. 79, 555 (1994).
- ²³E. Nishibori, M. Takata, K. Kato, M. Sakata, Y. Kubota, S. Aoyagi, Y. Kuroiwa, M. Yamakata, and N. Ikeda, Nucl. Instrum. Methods Phys. Res. A **467-468**, 1045 (2001).
- ²⁴F. Izumi and T. Ikeda, Mater. Sci. Forum **321-324**, 198 (2000).
- ²⁵ K. Ohoyama, T. Kanouchi, K. Nemoto, M. Ohashi, T. Kajitani, and Y. Yamaguchi, Jpn. J. Appl. Phys. **37**, 3319 (1998).
- ²⁶A. S. Wills, Phys. Rev. B **63**, 064430 (2001).
- ²⁷P. Dalmas de Réotier, P. C. M. Gubbens, and A. Yaouanc, J. Phys.: Condens. Matter 16, S4687 (2004).
- ²⁸J. Lago, T. Lancaster, S. J. Blundell, S. Bramwell, F. L. Pratt, M. Shirai, and C. Baines, J. Phys.: Condens. Matter **17**, 979 (2005).
- ²⁹P. Dalmas de Réotier, A. Yaouanc, L. Keller, A. Cervellino, B. Roessli, C. Baines, A. Forget, C. Vaju, P. C. M. Gubbens, A. Amato, and P. J. C. King, Phys. Rev. Lett. **96**, 127202 (2006).
- ³⁰F. Bert, P. Mendels, A. Olariu, N. Blanchard, G. Collin, A. Amato, C. Baines, and A. D. Hillier, Phys. Rev. Lett. 97, 117203 (2006).
- ³¹R. S. Hayano, Y. J. Uemura, J. Imazato, N. Nishida, T. Yamazaki, and R. Kubo, Phys. Rev. B **20**, 850 (1979).