

van der Waals dispersion power laws for cleavage, exfoliation, and stretching in multiscale, layered systems

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Layered and nanotubular systems that are metallic or graphitic are known to exhibit unusual dispersive van der Waals (vdW) power laws under some circumstances. In this Brief Report we investigate the vdW power laws of bulk and finite layered systems and their interactions with other layered systems and atoms in the electromagnetically nonretarded case. The investigation reveals substantial difference between “cleavage” and “exfoliation” of graphite and metals where cleavage obeys a C_2D^{-2} vdW power law while exfoliation obeys a $C_3 \log(D/D_0)D^{-3}$ law for graphitics and a $C_{5/2}D^{-5/2}$ law for layered metals. This leads to questions of relevance in the interpretation of experimental results for these systems which have previously assumed more trivial differences. Furthermore we gather further insight into the effect of scale on the vdW power laws of systems that simultaneously exhibit macroscopic and nanoscopic dimensions. We show that, for metallic and graphitic layered systems, the known “unusual” power laws can be reduced to standard or near standard power laws when the effective scale of one or more dimension is changed. This allows better identification of the systems for which the commonly employed “sum of C_6D^{-6} ” type vdW methods might be valid such as layered bulk to layered bulk and layered bulk to atom.

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Layered bulk systems such as graphite and boron nitride have their atoms confined to a series of spatially discrete planes with interplanar distances significantly greater than the intraplanar atomic separations, e.g., 3.34 vs 1.41 Å for graphite. They can exhibit unusual electronic behavior due to the nanometer scale of the layer thickness and macroscopic scale of the other two dimensions. This scale variation means that great care must be taken in investigating subtle physical effects such as dispersion forces as previous examples demonstrate.¹⁻³

All separated electronic systems exhibit long-range attractive potentials arising from instantaneous electron fluctuations correlating via the Coulomb potential. These long-range potentials (often called van der Waals dispersion potentials when electromagnetic retardation is ignored) are typically absent from the commonest *ab initio* calculations such as density-functional theory in the local-density approximation or generalized gradient approximation, or are approximated by pairwise interatomic potentials of the form C_6D^{-6} which are “summed over” in some way (see, e.g., Refs. 4–9) to obtain an effective power law of the form C_nD^{-n} . Here the exponent n is an integer and depends only on the geometry of the system while C_n depends on the individual atoms as well as the geometry.

As summarized in Ref. 1 the exponent of an asymptotic power law in metallic and graphitic systems can depend on both the geometry and the type of material, with metals, graphene, and insulators all differing. For example, with two parallel, nanoscale layers of a metal, graphene and insulator, the power-law exponents are 5/2, 3, and 4, respectively (where insulators do obey a “sum over D^{-6} ” rule). When the number of layers is infinite we will show that the power depends not only on the type of material, but also the way the layers are divided. This will be investigated through three

types of division: equal separation of all layers (“stretching”), division into two sub-bulks (“cleavage”), and removal of one layer from the top of a bulk (“exfoliation”). Furthermore the interaction of layered bulks with atoms will be studied.

In this Brief Report we investigate metallic and graphitic systems under these different types of division (insulators have trivial “sum over atoms” exponents and need no further investigation). Neither metals nor graphitic systems are guaranteed to obey “sum over layer” power laws and special care must be taken to evaluate their long-range correlation effects. As with previous work^{3,10-18} we make use of the adiabatic-connection formula and fluctuation dissipation theorem under the random-phase approximation (RPA) to calculate the leading power laws under these different methods of division. All results in this Brief Report are for the electromagnetically nonretarded case which¹⁹ show to be unimportant in the range of interest for similar systems.

Stretched graphitic systems (“staphite”) have already been studied in Ref. 3 where it was shown that the dispersion potential for an infinite number of graphene layers, each with interlayer spacing D , follows a C_3D^{-3} asymptotic power law at $T=0$ K where $C_3=0.80$ eV Å³. We may make use of the same basic approach employed for graphitic layered systems to calculate the dispersion for metallic layered systems (we believe that graphite-metal intercalates may be examples of this type of system). For brevity we define an intralayer Coulomb potential multiplied by a density-density response function

$$C(q,u) = \bar{\chi}(q,u)\bar{w}(q), \quad (1)$$

where q is the wave number parallel to the plane, $\bar{\chi}(q,u)$ is the noninteracting electron density-density response in a

TABLE I. Comparison of the system to bilayer ratio of the per-layer vdW coefficients for graphitic and metallic systems, for trilayered and stretched systems. Column 3 is obtained by summing over the individual layers and takes the form $\frac{2}{3}(2+2^{-n})$ for tri/bi and $2\zeta(p)$ for stretched/bi.

System	$C_n^{\text{system}}/C_n^{\text{bi}}$	Predicted	Error (%)
Tri-graphene	1.3632	1.4167	3.6
Stretched graphite	2.1157	2.4041	14
Tri-metal	1.3496	1.4512	7.5
Stretched metal	2.0628	2.6830	30

layer, and $\bar{w}(q)$ is the Coulomb potential. When the system is metallic this takes the form $\mathcal{C}(q, u) = -\omega_{\text{p2D}}(q)^2/u^2$ for $q \rightarrow 0$ where $\omega_{\text{p2D}}(q) = (\frac{n_0 e^2 q}{2\epsilon_0 m_e})^{1/2}$, $n_0 = N_{e\text{-layer}}/A_{\text{layer}}$ is the two-dimensional (2D) electron density of each layer, and m_e is the mass of an electron. We may thus utilize Eqs. 4 and 7 of Ref. 3 and change variables to $\theta = qD$ and $\eta = \sqrt{D}\omega_{\text{p2D}}(q)^{-1}u$ to show that the difference in the correlation energy per layer from the infinitely separated case takes the form

$$U_{\text{vdW}} \sim \frac{1}{D^{5/2}} \int_0^1 d\lambda \int_0^\infty \theta^{3/2} d\theta \int_0^\infty d\eta \mathcal{G}(\theta, \eta, \lambda) \quad (2)$$

demonstrating a $C_{5/2}D^{-5/2}$ dispersion power law for layered metals. Numerical evaluation gives $C_{5/2} = 9.26\hbar(\frac{n_0 e^2}{2\epsilon_0 m_e})^{1/2}$. As with graphitic systems this power law (although not the constant prefactor) is universal to all multilayered metallic systems with an infinite stack of isotropically stretched layers.

Given the universality of the van der Waals exponent for isotropic stretching of a given material it is worth exploring the validity of a “sum over $C_n D^{-n}$ ” rule for layered systems. In Table I we present the ratio of the C_n coefficient for a trilayered or stretched system to the bilayered system for metals and straphite as well as a the “sum over C_n ” prediction for this ratio (here n is 3 for straphite and $5/2$ for metals). The prediction is somewhat sound for straphite with an overprediction of 14% for straphite but is much less so for metals where it leads to a 30% overprediction of the potential. This suggests that, even using a correct power law, rules which effectively sum the coefficients may prove troublesome.

“Cleavage” represents another means of division of a layered system. Here the system is split between a single pair of layers to form two new layered systems. We refer to the new systems as half bulks as opposed to the original full bulk. For homogeneous, infinite layered systems the two are mirror images of each other.

Separating a layered bulk into two smaller half bulks is equivalent to keeping all but one layer at an interlayer spacing d while increasing the remaining one to $D \gg d$ as in Fig. 1 (we ignore any relaxation of the layers or layer spacing at the newly created surfaces). Here we are interested in the total dispersion energy per unit area rather than that per layer per area as in the previous case.

We may make use of the second-order perturbation for-

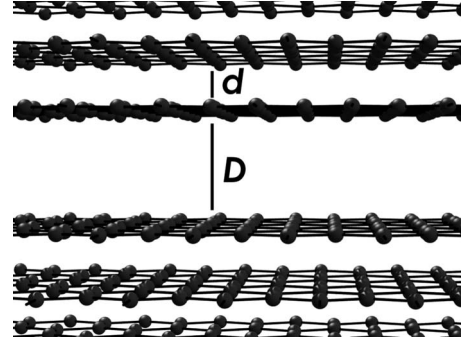


FIG. 1. Layers after “cleavage.”

mula for the dispersion energy between the half bulks while treating that within them to all orders (sometimes called the Zaremba-Kohn²⁰ formula). Thus

$$U_{\text{vdW}} = \frac{-\hbar}{4\pi^2} \int_0^\infty du \int_0^\infty dq q \mathcal{F}(q, u; D), \quad (3)$$

where

$$\mathcal{F}(q, u; D) \approx e^{-2qD} \text{Tr}\{[\chi_{\lambda=1}(q, u; d)w^b(qD)]^2\}, \quad (4)$$

where $w_{ij}^b = \bar{w}e^{-qd(i+j)}$ governs the interaction only *between* the two separate half bulks while χ_{ij} is the full interacting response of the density in layer i to a potential perturbation at layer j *within* a single half bulk.

Expanding the trace gives

$$\mathcal{F} = e^{-2qD} \sum_{ijkl} \chi_{ij} w_{jk}^b \chi_{kl} w_{li}^b = e^{-2qD} \left(\sum_i \delta n_i p^i \right)^2, \quad (5)$$

where δn_i is the interacting electron density response of layer i in the half bulk to an external potential perturbation $\delta v_i = \bar{w}p^i$ with $p = e^{-qd}$.

We may calculate δn_i through the RPA equation $\delta n_i = C p^i + C \sum_{j \geq 0} p^{i-j} \delta n_j$ where $C = C(q, u)$ is as defined in Eq. (1). This gives the following recursion relationship:

$$\delta n_{i+2} + \delta n_i = [(p + p^{-1}) - C(p - p^{-1})] \delta n_{i+1}, \quad (6)$$

where $\delta n_0 = C(1 + \sum_{j \geq 0} p^j \delta n_j)$ and we require $\delta n_\infty < \infty$. Writing a formal power series generating function $\mathcal{M}(x; C) = \sum_{i \geq 0} \delta n_i x^i$ transforms Eq. (6) into

$$(1 + x^2)\mathcal{M}(x; C) - \delta n_0 - x \delta n_1 = x[(p + p^{-1}) + C(p - p^{-1})] \times [\mathcal{M}(x; C) - \delta n_0], \quad (7)$$

so that (with correct series asymptotics)

$$\mathcal{M}(x; C) = \frac{C(p - p^{-1})}{p(1 - r_-x)(1 - r_+p^{-1})}, \quad (8)$$

where $r_\pm = \zeta \pm \sqrt{\zeta^2 - 1}$, $\zeta = \frac{1}{2}(p + p^{-1}) + \frac{1}{2}C(p - p^{-1})$. Noting that $\sum_i p^i \delta n_i = \mathcal{M}(p)$ lets us write Eq. (5) as

$$\mathcal{F}(q, u; d, D) = e^{-2qD} \mathcal{M}[e^{-qd}; C(q, u)]^2. \quad (9)$$

For graphite¹ shows that $C(q, u) = \kappa[1 + u^2/(v_0 q)^2]^{-1/2}$ where $v_0 = 5.0 \times 10^5 \text{ ms}^{-1}$, $\kappa = \frac{e^2}{4\epsilon_0 \hbar v_0} = 12.1$. Defining $\alpha = d/D$, $\theta = qD$ and $\eta = ud/v_0$ lets us expand $\mathcal{F}(\theta, \eta)$

$=\mathcal{F}(qD, ud/v_0; \alpha, 1)$ in powers of α . To leading order this gives

$$\mathcal{F}(\theta, \eta) = e^{-2\theta} \frac{\kappa^2}{(\kappa + \eta + \sqrt{\eta^2 + 2\kappa\eta})^2} \quad (10)$$

for graphite. A similar change of variables yields a similar leading order expansion for metals. For these cases Eq. (3) gives a leading power law of exponent $n=2$ as with insulators.

Thus for graphite and metals we can write

$$U_{\text{vdW}}(D \rightarrow \infty) \approx \frac{C_2}{D^2}, \quad (11)$$

where

$$C_2 = \frac{-\hbar \Gamma(2)}{4\pi^2} \int_0^\infty d\eta T(\eta). \quad (12)$$

Here $T(\eta) = \frac{v_0}{d} \kappa^2 (\kappa + \eta + \sqrt{\eta^2 + 2\kappa\eta})^{-2}$ for graphite and $T(\eta) = \frac{\omega_{\text{p3D}}}{\sqrt{2}} (1 + \eta^2 + \eta\sqrt{\eta^2 + 2})^{-2}$ for metals with $\omega_{\text{p3D}} = (\frac{n_0 e^2}{\epsilon_0 m_e d})^{1/2}$. For graphite we find $C_2 = 0.13d^{-1} \text{ eV } \text{\AA}^3$ and for metals $C_2 = 1.6886 \times 10^{-3} \hbar \omega_{\text{p3D}}$. The latter result agrees exactly with continuous but anisotropic models of half-bulk metals where electron movement is restricted to be parallel to the surface, as is expected if the $d \rightarrow 0$ limit is well defined.

These results differ significantly from those expected by a simple sum-of-layers approach where we would expect graphite to obey a $C_1 D^{-1}$ power law, and metals to obey a $C_{1/2} D^{-1/2}$ power law. The screening in these layered systems seems to cancel the different correlation effects of the individual layers so that they act as pseudoinulating bulks.

Another means of dividing graphene (“exfoliation”) is to peel a single layer of graphene from the top of a bulk. This represents yet another method of division where one system is a layered half bulk and the other a single layer. We restrict our investigation to the case where the removed layer plane is always stiff and parallel to the planar surface of the bulk.

To model exfoliation we use a similar perturbative approach to cleavage but in Eq. (3) set $\mathcal{F} = e^{-2qD} \mathcal{M}(e^{-qd}; C) \frac{C}{1+C}$ the product of the interacting response of a half bulk and a single layer. For graphene this complicates the problem as the $\alpha = d/D \rightarrow 0$ limit works well for \mathcal{M} but not for $\frac{C}{1+C}$. Here we keep α in the single-layer response *only*. Under transformation of variables $\theta = qD$ and $\eta = \frac{\alpha}{v_0} u$ (which already gives a D^{-2} outside the integrals) we find

$$\mathcal{F} = \frac{v_0}{D} \kappa^2 \theta e^{-2\theta} (\kappa + \eta + \sqrt{\eta^2 + 2\kappa\eta})^{-1} \times (\sqrt{\eta^2 + (\alpha\theta)^2} + \kappa\alpha\theta)^{-1}. \quad (13)$$

Setting $\theta=1$ (where $\theta^2 e^{-2\theta}$ takes its maximum), approximating $\sqrt{\eta^2 + \alpha^2}$ by $\alpha + \eta$ and $\kappa + \eta + \sqrt{\eta^2 + 2\kappa\eta}$ by $2(\kappa + \eta)$ allows us to approximate the η integral to show a leading $\alpha \log(\alpha)$ term. This is equivalent to a power law of the form

$$U_{\text{vdW}}(D \rightarrow \infty) \approx \frac{C_3 \log(D/D_0)}{D^3} \quad (14)$$

and numerical calculation of Eq. (3) validates this assumption. For graphene (where $\kappa=12.1$) we find $D_0=0.16d$ and $C_3=0.07 \text{ eV } \text{\AA}^3$.

A similar analysis of metals shows a $C_{5/2} D^{-5/2}$ power law with $C_{5/2} = 6.42 \times 10^{-3} \sqrt{d} \hbar \omega_{\text{p3D}} = 9.07 \times 10^{-3} \hbar \omega_{\text{p2D}} (q=1)$. The 5/2 power law is the same as that of a metal layer interacting with a continuum model of a metallic half bulk as is, again, predicted by the limit of $d \rightarrow 0$.

While this analysis is not valid in the small D regime it is worth noting that exfoliation and cleavage exhibit different power laws for graphite. This suggests that sum of C_6 models for converting experimental results from one to the other such as those employed in Ref. 21 may need reexamination. Unfortunately accurate calculation of the dispersion energy of such systems for $D \approx d_0$ (where d_0 is the layer spacing of graphite) is as yet intractable.

In nanoscale systems there are often combinations of molecules, layers, and bulks. A simple example is an atom interacting with the surface of a layered metal or a molecule interacting with a graphene surface. Here the power law could be affected by the layering and electronic properties of the material.

In the coordinate system used for the layered models the interacting response function of an infinitely small “atom” located at $Z\hat{z}$ can be written as

$$\chi_A(\mathbf{q}, \mathbf{q}', z, z'; u) = \alpha(iu) [\mathbf{q} \cdot \mathbf{q}' + \partial_z \partial_{z'}] \delta(z - Z) \delta(z' - Z), \quad (15)$$

where \mathbf{q} and \mathbf{q}' are reciprocal lattice vectors in the plane and $\alpha(iu)$ is the interacting dipole polarizability of the atom at imaginary frequency u . This formula, used in Eqs. (3) and (4), correctly reproduces the $C_6 D^{-6}$ power law for interacting atoms.

We can use Eq. (15) to calculate the interaction of an atom with a layered bulk (metallic or graphitic) by making use of Eqs. (3) and (4). This gives a power-law exponent $n=3$ in agreement with the prediction of a sum over $C_6 D^{-6}$ potentials. This result strongly suggests that the unusual power laws exhibited by layered systems result from the interaction between long-range fluctuations in *both* systems and that removing them from one reduces the systems to “typical” dispersive behavior.

As has been seen here and in other work¹⁻³ the asymptotic power-law behavior of layered systems can be anything but simple. Both graphitic and metallic systems exhibit vastly different dispersive power laws to insulators so that “sum of C_6 ” approximations such as those typically employed cannot be used in the asymptotic region. It seems unlikely that, with such varied asymptotes, the cohesive energies and other similar measurables can be investigated using simple models.

To illustrate these discrepancies we present in Table II a summary of the various power laws studied here. The insulator result represents the “classic” sum over atomic power-

TABLE II. Asymptotic power laws for various systems demonstrating both the material and structural dependence.

	Graphite	Metal	Insulator
Stretching	D^{-3}	$D^{-5/2}$	D^{-4}
Cleavage	D^{-2}	D^{-2}	D^{-2}
Exfoliation	$\log(\frac{D}{D_0})D^{-3}$	$D^{-5/2}$	D^{-3}
Atom-bulk	D^{-3}	D^{-3}	D^{-3}

laws behavior of each system and any difference from its exponent represents “unusual” behavior.

While uniform stretching results in different power laws for metals, graphitics and insulators, cleavage removes this variation and involves the same exponent for all materials. This suggests that the interlayer screening induced by the Coulomb potential dominates the local response of each layer in the van der Waals energy for such systems converting their behavior into that of nonlayered or insulating bulks. This is further demonstrated by the fact that both the power law *and* coefficient of cleaved layered metals are the same as that of cleaved bulk metals with electron movement restricted to the plane.

By contrast, keeping a finite number of layers asymptotically isolated, as in exfoliation, or all layers asymptotically separated, as in stretching, returns different power laws for different systems. In these cases at least one layer can be considered infinitesimally thin which we believe to be a requirement for the unusual power laws. Replacing the isolated layer by an atom, however, returns the classical results which suggests that at least one large dimension is required for unusual vdW dispersion power laws as postulated in Refs. 1 and 2.

Overall, as this work and references¹⁻³ demonstrate, the dispersion forces of systems with a mix of nanometer and macroscopic length scales are more complex than classic Lifshitz theory predicts. As the differing power laws for cleavage and exfoliation demonstrate we must take great care in using indirectly derived cohesive energies from experiment.

These unusual van der Waals power laws may also have profound effects on the behavior of many nanosystems. For certain systems it may be necessary to adapt molecular dynamics and other semiempirical and approximate *ab initio* simulation methods to account for these differences in order to best replicate experiment.

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