Phase relationships, and the structural, magnetic, and thermodynamic properties in the Sm5Si*x***Ge4−***^x* **pseudobinary system**

Kyunghan Ahn, V. K. Pecharsky, and K. A. Gschneidner, Jr.

Ames Laboratory, Iowa State University, Ames, Iowa 50011-3020, USA and Department of Materials Science and Engineering, Iowa

State University, Ames, Iowa 50011-2300, USA

(Received 15 March 2007; revised manuscript received 25 April 2007; published 12 July 2007)

The crystallography, phase relationships, and physical properties of the $\text{Sm}_5\text{Si}_x\text{Ge}_{4-x}$ alloys with $0 \le x \le 4$ have been investigated by using x-ray powder diffraction, dc magnetization, and heat capacity measurements between ~3 K and 350 K in magnetic fields ranging from 0 and 10 T. Similar to $Gd_5Si_xGe_{4-x}$, there are three distinct phase regions in the $Sm_5Si_xGe_{4-x}$ system: the Gd₅Si₄ type for Si-rich compositions, the Gd₅Si₂Ge₂ type for intermediate range of concentrations, and the Sm₅Ge₄ type for Ge-rich alloys. The magnetic properties of the Sm5Si*x*Ge4−*^x* compounds can be well described by considering the temperature-independent Van Vleck term due to a small energy separation between the ground state and the first excited state of $Sm³⁺$ ions. Compared with other light lanthanide $R_5Si_xGe_{4-x}$, magnetic ordering temperatures of $Sm_5Si_xGe_{4-x}$ compounds are much higher than expected from the de Gennes scaling. The change in the magnetic behaviors with the substitution of Ge by Si is also similar to that observed in the Gd₅Si_xGe_{4−*x*} system. The external magnetic fields up to 10 T have no effect on the magnetism of the Sm₅Si_xGe_{4−*x*} alloys.

DOI: [10.1103/PhysRevB.76.014415](http://dx.doi.org/10.1103/PhysRevB.76.014415)

: 75.20. En, 75.30. - m, 61.66. Dk

INTRODUCTION

The R_5 Si₄ and R_5 Ge₄ compounds were discovered by Smith *et al.*, who reported an orthorhombic $Sm₅Ge₄$ -type structure for the germanides with *R*=Nd, Sm, Gd, Tb, Er, and Y and for the silicides with $R = Y$, Tb-Er.¹ The crystal structure of Sm₅Ge₄ (space group symmetry *Pnma*) was described by three layers of atoms, which are the layer G (only Ge atoms), the layer S (only Sm atoms), and the layer C (combination of Sm and Ge atoms), stacked along the *b* axis in a $(GSCSG)$ sequence.² Later, they found that the orthorhombic 5:4 phase is stable for all of the lanthanide germanides R_5Ge_4 except for $R=Pm$, Eu, and Yb, but for silicides there are two different crystal structures depending on the rare earth element: the Zr_5Si_4 -type tetragonal structure for $R = La-Nd$, and the $Sm₅Ge₄$ -type orthorhombic structure for *R*=Y, Sm, Gd-Er except for the monoclinically distorted orthorhombic structure of Lu₅Si₄.^{[3](#page-11-2)} According to Holtzberg *et* $al.$ ^{[4](#page-11-3)} the silicides have relatively high ferromagnetic (FM) ordering temperatures (T_C) , i.e., the respective T_C 's are 336, 225, 140, 76, and 25 K for *R*=Gd, Tb, Dy, Ho, and Er, but the germanides are antiferromagnetic (AFM) with the much lower Néel temperatures (T_N) of 47, 30, 40, 21, and 7 K, respectively. Replacing a small amount of Ge with Si in Gd_5Ge_4 induces ferromagnetism at low temperatures in the $Gd_5Si_xGe_{4-x}$ solid solution.⁴

Since the discovery of the giant magnetocaloric effect (GMCE) in $Gd_5Si_2Ge_2$ $Gd_5Si_2Ge_2$ $Gd_5Si_2Ge_2$ ^{5[,6](#page-11-5)} the $R_5Si_xGe_{4-x}$ systems $(R = rare$ earth element) have been broadly studied to uncover the mechanism of their extraordinary magnetoresponsiveness including the giant magnetoresistance and colossal magnetostriction as well as $GMCE.^{6–8}$ $GMCE.^{6–8}$ $GMCE.^{6–8}$ The phase relationships and crystallography in the Gd₅Si_xGe_{4−*x*} system were reported by Pecharsky and Gschneidner⁹ who identified three structurally distinct phase regions: Gd_5Si_4 -type orthorhombic $(2 < x)$ ≤ 4 . Sm₅Ge₄-type orthorhombic $(0 \le x \le 0.8)$, and $Gd_5Si_2Ge_2$ type, which is a monoclinically distorted derivative of these two closely related orthorhombic structures $(0.96 \le x \le 2)$. Subsequent studies showed that these three solid solution alloys exhibit quite different magnetic behaviors.^{8[,10](#page-11-8)}

In contrast to the crystallographic description reported by Smith *et al.*,^{[3](#page-11-2)} the R_5 Si_{*x*}Ge_{4−*x*} systems can be better described as formed by pseudo-two-dimensional, \sim 7 Å thick slabs that are arranged differently in terms of bonding between the slabs rather than stacking of monolayers.^{11[,12](#page-11-10)} All of the slabs, one-half of the slabs, or none of the slabs are interconnected via short T_2 dimers (T=Si and/or Ge) for the Gd_5Si_4 -type, the $Gd_5Si_2Ge_2$ -type, and the Sm_5Ge_4 -type structures, respectively, e.g., see Fig. 1 in Refs. [8](#page-11-6) and [13.](#page-11-11) The interesting physical properties of Gd₅Si_xGe_{4−*x*} are strongly correlated with the magnetostructural transitions that are characterized by reversible breaking and reforming of the interslab T_2 bonds, which can be controlled by chemical composition, magnetic field, temperature, and pressure. While the interslab bonding differences are preserved in the paramagnetic or antiferromagnetic states depending on the crystal structure type, all of the slabs are always interconnected via the T_2 dimers in the ferromagnetic state, see Fig. 2 in Ref. [8.](#page-11-6)

Until now, the crystallography and phase relationships have been investigated for more than one-half of the possible R_5 Si_xGe_{4−*x*} pseudobinary systems $[R=La,$ ¹⁴ Pr,¹⁵ Nd,^{14[,16](#page-11-14)} Gd,¹⁷ Tb,¹⁸ Dy,¹⁴ Er,¹⁹ Yb,²⁰ Lu,¹⁴ and Y (Ref. [21](#page-11-19))]. Still, there are other R_5 Si_xGe_{4−*x*} systems (R =Ce, Sm, Eu, Ho, and Tm) that have not been studied in detail. The 5:4 phases $(R_5Si_4$ or R_5Ge_4) have not been reported for $R=Pm$ since it is radioactive with the longest half-life of 14 years, and they do not form for *R*=Eu. The latter has been experimentally verified by us: two alloys prepared as described in the next section at $Eu₅Si₄$ and $Eu₅Ge₄$ stoichiometries were clearly twophase materials, each containing $Eu₅Si₃$ and EuSi, and $Eu₅Ge₃$ and EuGe, respectively. For other rare earth metals $[R = Ce^{3,22-24}$ $[R = Ce^{3,22-24}$ $[R = Ce^{3,22-24}$ Sm,^{1-[3](#page-11-2)} Ho,^{3[,25,](#page-11-22)[26](#page-11-23)} and Tm (Refs. 3 and [27](#page-11-24))], only the binary silicides and germanides have been reported

Composition	Structure type	Unit cell dimensions				
		a, A	b, \AA	c, A	Distance $d_{\text{T3-T3}}$, A	Reference
Sm_5Si_4	Gd_5Si_4	7.5738(7)	14.890(1)	7.8156(7)	2.745(2)	This work
Sm_5Si_4	Sm_5Ge_4	7.57	14.88	7.78		3
Sm_5Si_3Ge	Gd_5Si_4	7.5858(6)	14.911(1)	7.8440(6)	2.722(2)	This work
$Sm_5Si_2Ge_2^a$	$Gd_5Si_2Ge_2$	7.6716(7)	14.945(2)	7.8543(8)	$3.821(4)^{b}$	This work
					$2.831(3)^{b}$	
Sm ₅ SiGe ₃	Sm_5Ge_4	7.7492(5)	14.927(1)	7.8414(6)	3.733(2)	This work
Sm_5Ge_4	Sm ₅ Ge ₄	7.7726(6)	14.947(1)	7.8611(6)	3.754(3)	This work
Sm_5Ge_4	Sm ₅ Ge ₄	7.75	14.94	7.84	3.71(2)	2

TABLE I. Room temperature crystallographic data of Sm₅Si_xGe_{4−*x*} alloys.

^a γ is 93.344(5)° in the monoclinic structure (space group *P*112₁/*a*).

bThere are two kinds of T3-T3 distances in the monoclinic structure; long T3a-T3a and short T3b-T3b distances.

with the exception of the ternary $Ho_5Si_2Ge_2$ compound.²⁸ Here, we report on the phase relationships, the crystal structures, and the magnetic and thermodynamic properties of both binary and several pseudobinary alloys in the Sm₅Si_xGe_{4−*x*} system.

EXPERIMENTAL DETAILS

A total of five alloys in the $Sm_5Si_xGe_{4-x}$ system with *x* varying from 0 to 4 were synthesized by induction melting at \sim 2070 K for 10 min in sealed Ta crucibles. Prior to induction melting, stoichiometric mixtures of the pure components (Sm, Si, and Ge) were loaded into Ta crucibles, and then the crucibles were sealed under a pure helium atmosphere by arc welding in order to avoid the loss of samarium due to its low boiling temperatures $({\sim}2070 \text{ K})$. The Sm metal was prepared by the Materials Preparation Center of the Ames Laboratory²⁹ and it was 99.5 at. % (99.94 wt. %) pure with major impurities in ppm atomic as follows: F, 1040; Ca, 900; O, 864; C, 125; Cl, 110; Mg, 77; N, 32; Zn, 30; Fe, 27; and Yb, 20. The silicon and germanium, which were purchased from a commercial vendor, were 99.999 wt. % pure. The errors in the compositions due to high vapor pressure of Sm at 2070 K estimated using the following assumptions: the empty volume of the crucible was \sim 1.6 cm³, the sample volume was \sim 1 cm³, the vaporized Sm follows ideal gas behavior) were less than 0.2 wt. $% (0.04$ at. $%)$. The partial vapor pressure of Sm at 2070 K was calculated using data of Ref. [30.](#page-11-27) Thus, the alloy compositions were accepted in the as-weighed condition. All alloys were investigated in the ascast condition without additional heat treatment.

The room temperature x-ray powder diffraction was utilized to characterize the crystal structures and phase compositions of the $Sm_5Si_xGe_{4-x}$ alloys. The room temperature x-ray powder diffraction studies were performed on an automated Scintag powder diffractometer using Cu K_{α} radiation. The crystal structures were refined by using full profile Rietveld refinement technique[.31](#page-11-28) Moreover, *in-situ* x-ray powder diffraction measurements of $Sm₅Si₂Ge₂$ were carried out as a function of temperature between 8 and 300 K both on cooling and on heating in a zero magnetic field on a Rigaku TTRAX rotating anode powder diffractometer employing Mo K_{α} radiation. The sample preparation, instrument setup, and the refinement method employed to process the *in-situ* x-ray powder diffraction data were the same as in Refs. [32](#page-11-29)[–34.](#page-11-30) The profile residuals were between 7% and 9%, and the derived Bragg residuals were between 3% and 5%.

Magnetic measurements were performed on a SQUID magnetometer (model MPMS XL). The magnetization was measured as a function of temperature from 1.8 to 300 K in 0.05 T, 0.5 T, and 5 T dc magnetic fields. Isothermal magnetization data were collected only around the magnetic ordering temperatures of the alloys and at 1.8 K in dc magnetic fields varying from 0 to 7 T with 0.2 T steps. The heat capacity of the Sm₅Si_xGe_{4−*x*} alloys was measured using an adiabatic heat-pulse calorimeter³⁵ from \sim 3.5 K to 350 K in dc magnetic fields ranging from 0 to 10 T.

RESULTS AND DISCUSSION

Phase relationships and crystallography

As mentioned above, the crystal structure in the paramagnetic (PM) state in the $R_5Si_xGe_{4-x}$ systems is a critical parameter that usually defines physical properties of individual alloys, especially their magnetism. When nonbonded slabs are present in the PM state, these generally become interconnected via short T_2 dimers at or below the ferromagnetic ordering temperature that may occur as first-order magneticstructural phase transformations.⁸ The magnetic phase transitions, if any, are normally second- order when all slabs are connected in the PM state.

The room temperature x-ray powder diffraction measurements were performed for all five alloys in the Sm₅Si_xGe_{4−*x*} systems in order to investigate their crystallography and phase purity in the PM state. The room temperature crystallographic data including the structure type and lattice parameters are given in Table [I](#page-1-0) and the coordinates of atoms in Table [II.](#page-2-0) Figures [1](#page-3-0) and [2](#page-3-1) show the observed and calculated powder diffraction patterns of the five alloys. $Sm₅Si₂Ge₂$ and $Sm₅Ge₄$ are nearly single phase materials (within the sensitivity of the x-ray powder diffraction technique, which given

Compound	Atom/site	x/a	y/b	z/c	$g~(\%)^a$			
Sm_5Si_4	Sm1 in $4(c)$	0.3574(9)	1/4	0.0104(6)				
	Sm2 in $8(d)$	0.0311(5)	0.0972(2)	0.1750(6)				
	Sm 3 in $8(d)$	0.3151(4)	0.8766(2)	0.1866(5)				
	Si1 in $4(c)$	0.225(1)	1/4	0.342(3)	100			
	Si2 in $4(c)$	0.001(3)	1/4	0.926(3)	100			
	$Si3$ in $8(d)$	0.146(3)	0.947(1)	0.479(3)	$100\,$			
Sm_5Si_3Ge	Sm1 in $4(c)$	0.3507(8)	1/4	0.0096(7)				
	Sm2 in $8(d)$	0.0258(5)	0.0967(2)	0.1792(5)				
	Sm3 in $8(d)$	0.3161(4)	0.8781(2)	0.1828(5)				
	T1 in $4(c)$	0.223(2)	1/4	0.339(2)	41.2^{b}			
	T ₂ in $4(c)$	0.997(3)	1/4	0.925(3)	82 ^b			
	T ₃ in $8(d)$	0.187(2)	0.962(1)	0.484(2)	66.4^{b}			
$Sm_5Si_2Ge_2$	Sm1 in $4(e)$	0.3211(1)	0.2544(6)	$-0.0066(9)$				
	Sm2a in $4(e)$	$-0.0117(9)$	0.0986(5)	0.1729(1)				
	Sm2b in $4(e)$	0.0116(1)	0.4009(5)	0.1935(1)				
	Sm3a in $4(e)$	0.3645(1)	0.8850(5)	0.1630(1)				
	Sm3b in $4(e)$	0.3320(1)	0.6221(5)	0.1788(1)				
	T1 in $4(e)$	0.198(3)	0.247(2)	0.366(3)	50 ^c			
	T ₂ in $4(e)$	0.943(3)	0.256(2)	0.908(2)	50 ^c			
	T ₃ a in $4(e)$	0.228(3)	0.956(1)	0.479(3)	50 ^c			
	T ₃ b in $4(e)$	0.150(3)	0.558(1)	0.475(3)	50 ^c			
Sm ₅ SiGe ₃	Sm1 in $4(c)$	0.2883(7)	1/4	0.0002(7)				
	Sm2 in $8(d)$	$-0.0215(4)$	0.0998(2)	0.1817(5)				
	Sm3 in $8(d)$	0.3803(4)	0.8841(2)	0.1633(4)				
	T1 in $4(c)$	0.173(1)	1/4	0.370(2)	25°			
	T2 in $4(c)$	0.915(2)	1/4	0.898(1)	25°			
	T ₃ in $8(d)$	0.221(1)	0.953(4)	0.469(1)	25°			
Sm ₅ Ge ₄	Sm1 in $4(c)$	0.2909(7)	1/4	0.0011(6)				
	Sm2 in $8(d)$	$-0.0187(4)$	0.1000(2)	0.1862(5)				
	Sm 3 in $8(d)$	0.3808(4)	0.8853(2)	0.1647(4)				
	Ge1 in $4(c)$	0.178(1)	1/4	0.362(1)	100			
	Ge2 in $4(c)$	0.929(2)	1/4	0.905(1)	$100\,$			
	Ge3 in $8(d)$	0.222(1)	0.952(1)	0.474(1)	100			

TABLE II. Atomic coordinates and T-site occupancies of Sm5Si*x*Ge4−*^x* compounds.

^aOccupancy by the Si atoms with the remainder occupied by the Ge atoms except for Sm₅Ge₄, where the value is for the site occupancies by the Ge atoms.

bOccupancies of the T sites have been refined with the only imposed constraint that each site has 100% overall occupancy.

c The actual occupancies were not refined—they were assigned based on the as-prepared stoichiometry assuming completely random distribution of the Si and Ge atoms.

the quality of data may be estimated at \sim 98 vol. % pure or better), but there were small amounts of the 5:3 impurity phases in Sm_5Si_4 , Sm_5Si_3Ge , and Sm_5SiGe_3 ; the concentration levels were 5.7, 7.6, and 4.6 vol. %, respectively, as determined from the Rietveld refinements. The lattice parameters as a function of Si content, $x(Si)$, are shown in Fig. [3.](#page-3-2) In the R_5 Si_xGe_{4−*x*} systems, all lattice parameters generally decrease with the replacement of Ge by Si because of the smaller atomic radius of Si. In the Sm₅Si_xGe_{4−*x*} system, there are small discontinuous changes in the *b* and *c* lattice parameters coinciding with the structural changes from the Sm_5Ge_4 -type to the $Gd_5Si_2Ge_2$ -type and the Gd_5Si_4 -type structures, but the greatest change occurs in the lattice parameter *a*. This is similar to the $Gd_5Si_xGe_{4-x}$ system, which exhibits the same phase sequence when Si is substituted for Ge.

Lattice parameters in the known binary 5:4 rare earth silicides, germanides, and ternary $R_5Si_xGe_{4-x}$ compounds with the same crystal structure (except for Yb_5Ge_4 , which unlike all other germanides has the Gd_5Si_4 -type structure) are plotted as a function of the atomic number of the rare earth element in Fig. [4.](#page-4-0) Even though the Sm ions in some compounds may have divalent, or trivalent, or mixed valence state, in the $Sm_5Si_xGe_{4-x}$ system the Sm ion appears to be

FIG. 1. (Color online) The observed (dots) and calculated (lines drawn through the data points) powder diffraction patterns of Sm_5Si_4 , Sm_5Si_3Ge , $Sm_5Si_2Ge_2$, and $Sm₅SiGe₃$ after the completion of Rietveld refinements. The upper sets of vertical bars located just below the plots of the observed and calculated intensities indicate the calculated positions of the Bragg peaks of the majority 5:4 phase, while the lower sets of bars correspond to the calculated positions of the Bragg peaks of the 5:3 impurity. The differences, *Y*_{obs}−*Y*_{calc}, are shown at the bottom of the plot. Several strong Bragg peaks observed between 40° and 50° that do not match the calculated pattern in (c) are experimental artifacts that result from spotty Debye rings.

trivalent because their lattice parameters follow the normal lanthanide contraction with the increasing atomic number.

Magnetic properties

The isofield magnetization as a function of temperature and the isothermal magnetization as a function of magnetic field were measured to characterize the type of magnetic ordering and to derive the magnetic ordering transition tem-

FIG. 2. (Color online) The observed (dots) and calculated (lines drawn through the data points) powder diffraction pattern of Sm5Ge4 after the completion of Rietveld refinements. The calculated positions of the Bragg peaks are shown as vertical bars just below the plots of the observed and calculated intensities. The differences, $Y_{obs} - Y_{calc}$, are shown at the bottom of the plot.

peratures, the effective magnetic moments, and the ordered magnetic moments.

The free trivalent Sm ion has five 4*f* electrons *L*=5 and $S = 5/2$) and in a solid the ⁶*H* levels are split by the spin-orbit interaction into the ground state ${}^{6}H_{5/2}$, the excited states ${}^{6}H_{7/2}$, ${}^{6}H_{9/2}$, ..., ${}^{6}H_{15/2}$ *J* multiplets.³⁶ The consecutive multiplet energy interval between the ground state $({}^{6}H_{5/2})$ and the first excited state $({}^{6}H_{7/2})$ is only ~930 cm⁻¹, so that the Van Vleck theory includes a second term, which is especially important for Sm (and Eu) ions but is negligible for heavy lanthanides.³⁷ Therefore, the Van Vleck term plays a major role in modeling the magnetic susceptibility of Sm com-

FIG. 3. (Color online) Lattice parameters as a function of x in the $Sm_5Si_xGe_{4-x}$ system. Sm₅Ge₄ and Sm₅SiGe₃ have the Sm_5Ge_4 -type structure; $Sm_5Si_2Ge_2$ has the $Gd_5Si_2Ge_2$ -type structure; and Sm_5Si_3Ge and Sm_5Si_4 have the Gd_5Si_4 -type structure.

FIG. 4. (Color online) The unit cell parameters of the isostructural R_5Si_4 (Gd₅Si₄-type structure), R_5Ge_4 (Sm₅Ge₄-type structure except for Yb_5Ge_4 , which has the Gd_5Si_4 -type structure), and $R_5Si_xGe_{4-x}$ (Gd₅Si₂Ge₂-type structure) plotted as a function of the atomic number.

pounds because of the varying population of the excited levels and second-order Zeeman splitting. Normally, the Curie part of the susceptibility in Sm compounds is considerably smaller compared to other lanthanide compounds because the g_J (Landé splitting factor) of the ground state $(J=5/2)$ is only 2/7. Thus, in principle, in the calculation of the susceptibility of the metallic compounds containing Sm^{3+} ions, there are various factors to be considered: the crystalline electric field (CEF) splitting of a ground state and the low level excited states of the *J* multiplet, Ruderman-Kittel-Kasuya-Yosida (RKKY) interaction, the Van Vleck transition between different *J* multiplets, which are caused both by the applied magnetic field and by the exchange fields, the effect of conduction-electron polarization due to 4*f* moments, and the thermal population of higher-lying *J* multiplets. With the consideration of these factors, the susceptibility calculation of metallic tripositive Sm compounds has been reported by several groups. $37-41$ $37-41$ Stewart⁴⁰ reported that when only the conduction electron polarization, interionic Heisenberg exchange interaction, and the thermally populated admixture of the ${}^{6}H_{7/2}$ into the ${}^{6}H_{5/2}$ state are taken into account, the susceptibility $\chi(T)$ can be well described by a simple form

$$
\chi(T) = \chi_0 + C/(T - \theta_p) \tag{1}
$$

without considering the CEF splitting. Here, χ_0 is the temperature-independent Van Vleck term due to the admixture of the first excited angular momentum state $(J=7/2)$ to the unperturbed ground state $(J=5/2)$ when either a magnetic field is applied or a temperature is high enough to allow thermal population. In the Curie-Weiss term, $C/(T - \theta_p)$, arising from the ground state $(J=5/2)$, C is the Curie constant, *T* is the absolute temperature, and θ_n is the Weiss temperature.

The magnetization of $Sm_5Si_xGe_{4-x}$ alloys is shown in Figs. [5](#page-5-0) and [6.](#page-5-1) Irrespective of the dc magnetic fields, including 0.5 T and 5 T data, the magnetizations of Sm_5Si_4 , Sm_5Si_3Ge Sm_5Si_3Ge Sm_5Si_3Ge , and $Sm_5Si_2Ge_2$ shown in Fig. 5 exhibit a broad maximum around the magnetic ordering temperature, and then decrease with the decreasing temperature showing nearly constant values below \sim 50 K. This behavior mimics ferrimagnetic arrangements of spins, but different temperature dependencies of spin and orbital magnetic moments arising from ferromagnetically aligned $Sm³⁺$ ions may also induce such behavior.^{42[–45](#page-12-5)} On the other hand, both Sm_5SiGe_3 and Sm₅Ge₄ seem to order antiferromagnetically around 90 K, but there are upturns in their magnetizations below \sim 50 K, which are uncommon for a conventional AFM, see Fig. [6.](#page-5-1)

The isothermal magnetization data, measured just below the magnetic ordering temperatures that are shown in Figs. [7](#page-6-0)(b), 7(d), and 7(f), confirm that Sm_5Si_4 , Sm_5Si_3Ge , and $Sm_5Si_2Ge_2$ order ferromagnetically with low saturation magnetization (M_S) values, but with a substantial coercivity (H_C) , and hysteresis. At 1.8 K [Figs. [7](#page-6-0)(a), 7(c), and 7(e)], none of the compounds approach saturation and all of them exhibit narrower hysteresis. On the other hand, the isothermal magnetizations of the Ge-rich compounds measured at 80 K and seen in Figs. $7(h)$ $7(h)$ and $7(j)$ are nearly linear functions of the magnetic field between 0 and 7 T; this behavior changes little at 1.8 K. The magnetic behavior in the $\text{Sm}_5\text{Si}_x\text{Ge}_{4-x}$ system is consistent with other $R_5\text{Si}_x\text{Ge}_{4-x}$ systems where the ferromagnetic or ferromagneticlike ground state is found in Si-rich alloys changing over to the AFM state for Ge-rich alloys. The dc magnetic susceptibility of $Sm_5Si_2Ge_2$ in the field-cooled (FC) cooling regime does not

FIG. 5. (Color online) The magnetization of Sm_5Si_4 , Sm_5Si_3Ge , and $Sm₅Si₂Ge₂$ measured in dc magnetic fields of 0.5 T and 5 T. Insets show the nonlinear fits of the magnetic susceptibilities measured in 0.5 T fields in the paramagnetic regime using Eq. (1) (1) (1) .

match that in the zero-field-cooled (ZFC) heating regime around the magnetic ordering temperature exhibiting a Curie temperature difference of \sim 10 K between the heating and cooling branches (Fig. 5), which is consistent with a firstorder magnetic phase transition. There is no measurable hysteresis of T_c in Sm_5Si_4 and Sm_5Si_3Ge nor is there a hyster-

FIG. 6. (Color online) The magnetization of Sm₅SiGe₃ and $Sm₅Ge₄$ measured in dc magnetic fields of 0.5 T and 5 T. Lines drawn through the data points are nonlinear fits of the magnetic susceptibilities measured in 5 T fields using Eq. (1) (1) (1) .

etic behavior around T_N in Sm₅SiGe₃ and Sm₅Ge₄, which is consistent with second-order magnetic phase transitions. Moreover, the $Sm₅Ge₄$ -type compounds show no features consistent with magnetostructural changes despite the absence of the interslab T_2 bonds in the PM state.

As mentioned above, the dc magnetic susceptibilities of these compounds in the paramagnetic state cannot be described by a simple Curie-Weiss law because of the nonlinearity of $1/\chi$ above the magnetic ordering temperature, but they can be well fitted by the modified Curie-Weiss law Eq. ([1](#page-4-1))]. Especially, in the case of tripositive Sm compounds,

$$
\chi(T) = \chi_0 + C/(T - \theta_p) = \left(\frac{N_A}{k_B}\right) \left(\alpha_J \mu_B^2 + \frac{\mu_{\rm eff}^2}{3(T - \theta_p)}\right) \tag{2}
$$

can be derived, where N_A is Avogadro's number, k_B is Boltzmann constant, μ_B is the Bohr magneton, μ_{eff} is the effective magnetic moment in Bohr magnetons, and

FIG. 7. (Color online) The magnetization as a function of magnetic field measured at 1.8 K and a few Kelvin below the magnetic phase transition temperatures for (a) and (b) Sm_5Si_4 , (c) and (d) Sm_5Si_3Ge , (e) and (f) $Sm_5Si_2Ge_2$, (g) and (h) Sm_5SiGe_3 , (i) and (j) $Sm₅Ge₄$. Insets in (b) and (d) show initial magnetizations in magnetic fields between 0 and 1 T.

$\alpha_J = k_B \left(\frac{(J^2 - (L - S)^2)[(L + S + 1)^2 - J^2]}{6J(2J + 1)(E_{J-1} - E_J)} \right)$	
$+\frac{[(J+1)^2-(L-S)^2][(L+S+1)^2-(J+1)^2]}{6(J+1)(2J+1)(E_{J+1}-E_J)}$ (3)	

is 20/7 Δ , where Δ is the energy separation (expressed in units of K) between $J=7/2$ and $J=5/2$ states.⁴⁶ Thus, the best fitting parameters Δ , μ_{eff} , and θ_p , extracted from the nonlinear least square fits of the dc magnetic susceptibilities are listed in Table [III](#page-7-0) along with other basic magnetic quan-

tities. The root mean square (rms) errors for nonlinear fits were less than 0.7% except for $Sm_5Si_2Ge_2$ where they were 3%. The Van Vleck terms χ_0 in the Sm₅Si_xGe_{4−*x*} system range between 1.04×10^{-3} cm³/Sm mole for Sm₅Ge₄ and 1.38×10^{-3} cm³/Sm mole for Sm₅Si₄, which compares well with the value of 1.18×10^{-3} cm³/mole obtained in Ref. [40](#page-12-3) for the rhombohedral Sm metal.

Even though both Sm_5SiGe_3 and Sm_5Ge_4 are likely to have an AFM ground state, their paramagnetic Curie temperatures, θ_p , are positive (70 and 60 K, respectively), which is one of the most unusual features of all of the other AFM

Composition	T_C (K) ^a	Θ_{P} (K)	$\mu_{\text{eff}}(\mu_B)$	μ at 1.8 K $(\mu_B)^c$	μ at $\sim T_c$ $(\mu_B)^d$	χ_0 (10 ⁻³ cm ³ /Sm mole)	Δ (K) ^e
Sm_5Si_4	220	220	0.34	0.025	0.1	1.38	776
Sm_5Si_3Ge	210	212	0.29	0.02	0.08	1.30	824
$Sm_5Si_2Ge_2$	125	120	0.36	0.02	0.05	1.07	1000
Sm_5SiGe_3	90 ^b	70	0.31	0.024	0.018	1.04	1030
Sm ₅ Ge ₄	90 ^b	60	0.35	0.018	0.018	1.04	1030

TABLE III. The magnetic properties of Sm₅Si_xGe_{4−*x*} compounds.

^a T_C or T_N from the C_P measurement.
^bNéel temperature (T_N) .

^bNéel temperature (T_N) .

c The magnetic moment per Sm atom in a 7 T magnetic field at *T*=1.8 K.

^dThe magnetic moment per Sm atom in a 7 T magnetic field between 10 and 40 K below the magnetic transition temperature.

^e Energy separation between $J=7/2$ and $J=5/2$ states.

 R_5Ge_4 compounds studied to date (i.e., $\theta_P=94$ K for $R=6d$, 80 K for Tb, [4](#page-11-3)3 K for Dy, 16 K for Ho, and 10 K for Er).⁴ The effective magnetic moments of the Sm₅Si_xGe_{4−*x*} alloys range between 0.29 and 0.36 μ_B /Sm atom, which is considerably smaller than the theoretical effective magnetic moment of the free Sm³⁺ ion (0.85 μ_B /Sm atom), while Δ ranges between 776 K and 1030 K; all are lower than \sim 1350 K estimated for free Sm³⁺ ions.³⁷ The reduced effective magnetic moments in the $Sm_5Si_xGe_{4-x}$ alloys are probably due to the effect of CEF splitting of the Hund's rule ground state because crystal fields can also admix the ground and excited levels and thus affect the magnetic susceptibility. The low Δ values have been reported for other Sm-based compounds[.47,](#page-12-7)[48](#page-12-8)

The magnetization values around the Curie temperatures for Sm_5Si_4 , Sm_5Si_3Ge , and $Sm_5Si_2Ge_2$ are ~ 0.091 , 0.076, and $0.046 \mu_B/\text{Sm}$ atom, respectively, obtained in the maximum field of our apparatus (7 T), which are only \sim 12.8%, 10.7%, and 6.5% of the theoretical value of the saturation magnetization $M_{\text{sat}}=g_JJ\mu_B=0.71\mu_B/\text{Sm}$ atom. The possible causes for the low saturation magnetization values at high magnetic fields are the magnetic anisotropy, second-order Zeeman effects, and CEF. Interestingly, high coercive fields H_C of \sim 0.7, 1.4, and 1 T are observed in close proximity of T_c for Sm₅Si₄, Sm₅Si₃Ge, and Sm₅Si₂Ge₂, respectively, see Fig. [7.](#page-6-0) Moreover, the T_c and T_N values for Sm compounds $(T_C = 220 \text{ K}$ for Sm_5Si_4 and $T_N = 90 \text{ K}$ for Sm_5Ge_4) are much higher than expected from the de Gennes scaling in light lanthanide elements (Ce, Pr, Nd, and Sm).

Heat capacity

The heat capacity (C_P) data of all alloys in the Sm₅Si_xGe_{4−*x*} system are shown as a function of temperature between \sim 3.5 K and 350 K in magnetic fields of 0 and 10 T in Figs. [8](#page-7-1) and [9.](#page-8-0) The peak shapes of C_p for Sm_5Si_4 , Sm_5Si_3Ge , Sm_5SiGe_3 , and Sm_5Ge_4 at either T_C or T_N have the typical lambda shape of a material which exhibits a second-order magnetic transformation, but that for $Sm₅Si₂Ge₂$ is indicative of a broad first-order phase transition at T_c =125 K. The magnetic ordering temperatures from C_P data are in good agreement with those from the dc sus-

FIG. 8. (Color online) The heat capacities of Sm_5Si_4 , Sm_5Si_3Ge , and $Sm₅Si₂Ge₂$ measured in 0 and 10 T magnetic fields measured on heating.

FIG. 9. (Color online) The heat capacities of Sm₅SiGe₃ and $Sm₅Ge₄$ measured in 0 and 10 T magnetic fields measured on heating.

ceptibilities. Interestingly, the C_P data for 10 T (the highest magnetic field available in our calorimeter) are nearly superimposed on those for zero magnetic field in all the alloys, that is, the external magnetic field has little to no effect on the magnetic entropy. In general, the magnetocaloric effect (MCE) as the isothermal magnetic entropy change, ΔS_{mag} , and the adiabatic temperature change, ΔT_{ad} can be calculated from the C_P data as described by Pecharsky and Gschneidner.⁴⁹ Since $\Delta S_{\text{mag}} = S(B, T) - S(0, T)$ and Gschneidner.⁴⁹ Since $\Delta S_{\text{mag}} = S(B, T) - S(0, T)$ $=\int_0^T \frac{C(B,T)-C(0,T)}{T} dT$, the MCE in the Sm₅Si_xGe_{4−*x*} system is expected to be negligibly small because of roughly overlapping C_P data, i.e., $C(B, T) \cong C(0, T)$.

The heat capacities of nonmagnetic analogs R_5Si_4 , R_5Ge_4 , and $R_5Si_2Ge_2$ for $R=La$ and Lu were used to estimate the electronic and lattice contributions to the heat capacities of magnetic Sm_5Si_4 , Sm_5Ge_4 , and $Sm_5Si_2Ge_2$, and thus determine the purely magnetic heat capacities of these compounds. The molar magnetic entropies of $Sm₅Si₄$, $Sm₅Ge₄$, and $Sm₅Si₂Ge₂$ are plotted in Fig. [10.](#page-8-1) At \sim 350 K they reach 71%, 74%, and 71% of the theoretical maximum molar magnetic entropy of the Sm³⁺ ion $(J=5/2)$ of 5*R* ln 6, i.e., S_{mag} $=$ *R*ln(2*J*+1) J/mol K, respectively (where *R*=8.31 J/mol K

FIG. 10. (Color online) The molar magnetic entropies of Sm_5Si_4 , Sm_5Ge_4 , and $Sm_5Si_2Ge_2$ in zero magnetic field.

is the molar gas constant). In any case the magnetic entropy should attain its full $R\ln(2J+1)$ value for l mole of 4*f* ions at high temperature. With all the rare earths except Gd, the entropy does not reach this maximum value because of CEF splitting. Furthermore, there is a clear indication of a firstorder magnetostructural transition in the molar magnetic entropy of $Sm₅Si₂Ge₂$ in Fig. [10](#page-8-1) and thus the change of the entropy during the structural transformation (ΔS_{str}) can be

 $Tb_5Si_2Ge_2.¹⁸$ $Tb_5Si_2Ge_2.¹⁸$ $Tb_5Si_2Ge_2.¹⁸$

TABLE IV. The Gd₅Si₄-type crystal structure of the low-temperature (LT) $Sm_5Si_2Ge_2$ at 8 K in a zero magnetic field. The space group symmetry is *Pnma*.

easily estimated. It is approximately equal to that of $Gd_5Si_2Ge_2 \ (\Delta S_{str} = 1.08 \ J/g \ at. \ K).⁵⁰$ $Gd_5Si_2Ge_2 \ (\Delta S_{str} = 1.08 \ J/g \ at. \ K).⁵⁰$ $Gd_5Si_2Ge_2 \ (\Delta S_{str} = 1.08 \ J/g \ at. \ K).⁵⁰$

Low-temperature crystallography and preliminary phase diagram

Since both the heat capacity and magnetization of $Sm₅Si₂Ge₂$ point to a first-order transformation, likely coupled with a polymorphic transition, we studied the temperature dependence of the crystal structure of this compound. The crystallographic data and coordinates of atoms of low-temperature (LT) $Sm_5Si_2Ge_2$ at 8 K in a zero magnetic field, which has the Gd_5Si_4 -type structure, are listed in Table [IV.](#page-9-0) The observed x-ray powder diffraction patterns of $Sm₅Si₂Ge₂$, which were collected in a zero magnetic field during cooling from 300 K to 8 K , are shown in Fig. [11.](#page-9-1) There are distinguishable differences in the positions and intensities of Bragg peaks between the low-temperature (Gd₅Si₄-type orthorhombic structure) and high-temperature

7.85 *c* (A) *c* 7.80 $Sm₅Si₂Ge₂$ *a*, *b/2* or 7.75 **Cooling** 7.70 *a* Lattice parameter, Lattice parameter, 7.65 7.60 7.55 7.50 *b/2* 7.45 60 80 100 120 140 160 180 200 7.85 *c* (A) *c* 7.80 *a*, *b/2* or $\rm Sm_{5}Si_{2}Ge_{2}$ 7.75 Heating 7.70 *a*

(Gd₅Si₂Ge₂-type monoclinic structure) patterns, which indicate that a structural phase transformation begins around \sim 115 K on cooling. These crystallographic changes are similar to those observed in $Gd_5Si_2Ge_2$ (Ref. [11](#page-11-9)) and

7.60 7.65

FIG. 11. (Color online) The observed x-ray powder diffraction patterns of $Sm_5Si_2Ge_2$ collected in a zero magnetic field during cooling from 300 K to 8 K. All patterns were collected using Mo K_{α} radiation. Only the range from 13° to 19° 2 θ is shown for clarity. The letter *O* indicates selected characteristic Bragg peaks of the Gd_5Si_4 -type orthorhombic phase.

cell dimensions of the major component during cooling of $Sm_5Si_2Ge_2$ (top) and during heating (bottom) in a zero magnetic field. The error bars are smaller than the size of the symbols.

b/2

FIG. 13. (Color online) The concentration of the orthorhombic Gd_5Si_4 -type phase as a function of temperature determined from Rietveld refinement of the patterns collected during cooling and heating of the $Sm_5Si_2Ge_2$ sample in a zero magnetic field.

The temperature dependencies of the unit cell parameters observed during the temperature-induced structural transformations are illustrated in Fig. [12](#page-9-2) for heating and cooling in a zero magnetic field. As the temperature induces the $Gd_5Si_2Ge_2$ -type to the Gd_5Si_4 -type transformation around 115 K on cooling in the zero magnetic field, the unit cell volume decreases by $\Delta V/V = -0.48\%$ and the lattice parameters change by −0.63%, −0.02%, and +0.01% along the *a*, *b*, and *c* axes, respectively. Reverse structural transformation is completed around 125 K on heating and the unit cell volume increases by $\Delta V/V = +0.55\%$ and the lattice parameters change by +0.66%, +0.01%, and −0.08% along the *a*, *b*, and *c* axis, respectively.

The temperature dependencies of the molar concentrations of the Gd_5Si_4 -type $Sm_5Si_2Ge_2$ phase derived from the Rietveld refinement of the x-ray patterns are shown in Fig. [13.](#page-10-0) The structural transformation from the Gd_5Si_4 type to the $Gd_5Si_2Ge_2$ type on heating is nearly complete, but the transformation on cooling is incomplete. Even at 8 K, the concentration of the monoclinic $Gd_5Si_2Ge_2$, type phase amounts to 40%. A similar phenomenon was also reported in *in-situ* x-ray powder diffraction studies of Gd_5Ge_4 , with ~6.5% of the high-temperature phase retained at low temperatures, which was explained by existence of microstructure imperfections, such as impurity and defects. 33 The much greater degree of incompleteness of the monoclinic-to-orthorhombic transformation in $Sm₅Si₂Ge₂$ compared to closely related compounds with Gd (Ref. 33) and Tb (Refs. [18](#page-11-16) and 51) is consistent with a reduced spin magnetic moment *S*=5/2 in Sm, while it is 7/2 and 6/2 in Gd and Tb, respectively) and the resulting weakening of the effective magnetic exchange interactions. This finding is consistent with a recent report by Nirmala *et al.*^{[52](#page-12-12)} who showed that the monoclinic Dy_5Si_3Ge to orthorhombic Dy_5Si_3Ge transformation, which occurs around $T=50$ K on cooling, is only $\sim 50\%$ complete (the spin magnetic moments of Sm and Dy are identical).

FIG. 14. (Color online) The tentative magnetic and crystallographic phase diagram of the $Sm_5Si_xGe_{4-x}$ system. Boundaries drawn using dashed lines are imprecise and should be considered as estimates only. Ferromagnetic $Sm_5Si_2Ge_2$ exists in a phase separated state adopting two different types of crystal structure, i.e., the monoclinic $Gd_5Si_2Ge_2$ and orthorhombic Gd_5Si_4 types.

The $Gd_5Si_2Ge_2$ -type to Gd_5Si_4 -type phase transformation begins at $T_{st} \cong 115$ K on cooling in a zero field, while the reverse transformation is completed at $T_{st} \approx 125$ K on heating. There is a difference of \sim 10 K between the structural phase transformation on cooling and on heating in a zero magnetic field, which is consistent with that observed in the dc magnetic susceptibility in both ZFC and FC conditions for $Sm_5Si_2Ge_2$ $Sm_5Si_2Ge_2$ $Sm_5Si_2Ge_2$ (see Fig. 5). Moreover, $T_{st} \cong 125$ K on heating is nearly equal to the Curie temperature (125 K) determined from the heat capacity peak on heating in a zero magnetic field, which indicates that there is the coupling of the magnetic and crystallographic phase transitions in $Sm₅Si₂Ge₂$.

A tentative magnetic and crystallographic phase diagram of the Sm5Si*x*Ge4−*^x* system is shown in Fig. [14.](#page-10-1) Even though the diagram is preliminary, its similarity to those of the $Gd_5Si_xGe_{4-x}$ and $Tb_5Si_xGe_{4-x}$ systems is clear. We also note that this diagram is simpler than that of the $Tb_5Si_xGe_{4-x}$ system 18 because of the absence of spin-reorientation transitions. A magnetostructural phase transformation is only observed in the monoclinic phase region, because no ferro magnetic or ferrimagnetic state is reached in the $Sm₅Ge₄$ -type phase region. Because this study is based on the examination of only five alloys, phase transition boundaries drawn using dashed lines should be considered only preliminary because their location may change when more data becomes available.

CONCLUSIONS

In summary, there are three distinct phase regions for Sm₅Si_xGe_{4−*x*} alloys as *x* varies from 0 to 4, which are the Gd_5Si_4 -type, the $Gd_5Si_2Ge_2$ -type, and the Sm_5Ge_4 -type structure; this behavior is consistent with the $Gd_5Si_3Ge_{4-r}$ system. The unit cell dimensions of both Sm_5Si_4 and Sm_5Ge_4 compounds follow normal lanthanide contraction rule when compared to those of other lanthanides, thus indicating that Sm ions are in the trivalent state. The dc magnetic susceptibilities for Sm₅Si_xGe_{4−*x*} alloys can be well described with the consideration of the temperature-independent Van Vleck term because of the narrow energy separation between *J* $=5/2$ and $J=7/2$ multiplet states of Sm³⁺ ions. The reduced effective magnetic moment and saturated magnetic moment of Sm5Si*x*Ge4−*^x* alloys are likely due to CEF splitting. The magnetic behaviors with the replacement of Ge by Si in Sm₅Si_xGe_{4−*x*} alloys are similar to those observed in Gd₅Si_xGe_{4−*x*} alloys (FM for Si-rich regions, first-order FM for intermediate regions, and AFM for Ge-rich regions). The magnetic transition temperatures of Sm_5Si_4 (T_C =220 K) and Sm_5Ge_4 ($T_N=90 \text{ K}$) are higher than expected from de Gennes factor scaling in light lanthanide elements. An external magnetic field of 10 T does not suppress the heat capacity peaks, which are clearly of magnetic origin in the Sm₅Si_xGe_{4−*x*} system, which is in contrast to the behavior observed in other *R*5Si*x*Ge4−*^x* systems such as when *R*=Gd or Tb, which are significantly affected (either reduced and broadened or shifted) by high magnetic fields. Thus, the magnetocaloric effect values are negligible in the Sm₅Si_xGe_{4x} system. *In-situ* x-ray powder diffraction measurements as a function of temperature indicate that there is a coupling of the magnetic and structural phase transitions for $Sm_5Si_2Ge_2.$

ACKNOWLEDGMENTS

This work was supported by the Office of Basic Energy Sciences of the Office of Sciences of the U. S. Department of Energy under Contract No. DE-AC02-07CH11358 with Iowa State University of Science and Technology. The authors thank Ya. Mudryk for help with collecting temperature dependent x-ray powder diffraction data.

- ¹G. S. Smith, A. G. Tharp, and Q. Johnson, Nature (London) 210, 1148 (1966).
- 2G. S. Smith, Q. Johnson, and A. G. Tharp, Acta Crystallogr. **22**, 269 (1967).
- 3G. S. Smith, A. G. Tharp, and Q. Johnson, Acta Crystallogr. **22**, 940 (1967).
- 4F. Holtzberg, R. J. Gambino, and T. R. McGuire, J. Phys. Chem. Solids 28, 2283 (1967).
- 5V. K. Pecharsky and K. A. Gschneidner, Jr., Phys. Rev. Lett. **78**, 4494 (1997).
- 6L. Morellon, P. A. Algarabel, M. R. Ibarra, J. Blasco, B. García-Landa, Z. Arnold, and F. Albertini, Phys. Rev. B **58**, R14721 $(1998).$
- 7L. Morellon, J. Stankiewicz, B. García-Landa, P. A. Algarabel, and M. R. Ibarra, Appl. Phys. Lett. **73**, 3462 (1998).
- 8V. K. Pecharsky and K. A. Gschneidner, Jr., Adv. Mater. Weinheim, Ger.) 13, 683 (1997).
- 9V. K. Pecharsky and K. A. Gschneidner, Jr., J. Alloys Compd. 260, 98 (1997).
- 10V. K. Pecharsky and K. A. Gschneidner, Jr., Appl. Phys. Lett. **70**, 3299 (1997).
- 11W. Choe, V. K. Pecharsky, A. O. Pecharsky, K. A. Gschneidner, Jr., V. G. Young, Jr., and G. J. Miller, Phys. Rev. Lett. **84**, 4617 $(2000).$
- 12V. K. Pecharsky, A. O. Pecharsky, and K. A. Gschneidner, Jr., J. Alloys Compd. 344, 362 (2002).
- 13W. Choe, A. O. Pecharsky, M. Worle, and G. J. Miller, Inorg. Chem. **42**, 8223 (2003).
- 14K. A. Gschneidner, Jr., V. K. Pecharsky, A. O. Pecharsky, V. V. Ivtchenko, and E. M. Levin, J. Alloys Compd. 303, 214 (2000).
- 15H. F. Yang, G. H. Rao, W. G. Chu, G. Y. Liu, Z. W. Ouyang, and J. K. Liang, J. Alloys Compd. 339, 189 (2002).
- 16H. F. Yang, G. H. Rao, G. Y. Liu, Z. W. Ouyang, W. F. Liu, X. M. Feng, W. G. Chu, and J. K. Liang, J. Alloys Compd. **346**, 190 $(2002).$
- 17A. O. Pecharsky, K. A. Gschneidner, Jr., V. K. Pecharsky, and C. E. Schindler, J. Alloys Compd. 338, 126 (2002).
- 18C. Ritter, L. Morellon, P. A. Algarabel, C. Magen, and M. R. Ibarra, Phys. Rev. B 65, 094405 (2002).
- 19A. O. Pecharsky, K. A. Gschneidner, Jr., V. K. Pecharsky, D. L. Schlagel, and T. A. Lograsso, Phys. Rev. B 70, 144419 (2004).
- 20K. Ahn, A. O. Tsokol, Yu. Mozharivskyj, K. A. Gschneidner, Jr., and V. K. Pecharsky, Phys. Rev. B 72, 054404 (2005).
- 21A. O. Pecharsky, V. K. Pecharsky, and K. A. Gschneidner, Jr., J. Alloys Compd. 379, 127 (2004).
- 22M. V. Bulanova, P. N. Zheltov, K. A. Meleshevich, P. A. Saltykov, and G. Effenberg, J. Alloys Compd. 345, 110 (2002).
- 23S. Bobev and E. D. Bauer, Acta Crystallogr., Sect. E: Struct. Rep. Online **61**, i73 (2005).
- 24L. M. Wu, S. H. Kim, and D. K. Seo, J. Am. Chem. Soc. **127**, 15682 (2005).
- ²⁵ P. Schobinger-Papamantellos and A. Niggli, J. Phys. (Paris), Colloq. **C5**, 156 (1979).
- 26V. N. Eremenko, V. E. Listovnichii, S. P. Lusan, Yu I. Buyanov, and P. S. Martsenyuk, J. Alloys Compd. 219, 181 (1995).
- 27V. N. Eremenko, K. A. Meleshevich, Yu. I. Buyanov, and P. S. Martsenyuk, Sov. Powder Metall. Met. Ceram 28, 543 (1989).
- 28N. P. Thuy, Y. Y. Chen, Y. D. Yao, C. R. Wang, S. H. Lin, J. C. Ho, T. P. Nguyen, P. D. Thang, J. C. P. Klaasse, N. T. Hien, and L. T. Tai, J. Magn. Magn. Mater. 262, 432 (2003).
- 29Materials Preparation Center, Ames Laboratory US-DOE, Ames, IA, USA, www.mpc.ameslab.gov
- 30A. Desiredi, V. Piacente, and S. Nobili, J. Chem. Eng. Data **18**, 140 (1973).
- 31B. A. Hunter, Rietica-A visual Rietveld program. International Union of Crystallography Commission on Powder Diffraction, Newsletter No. 20 (Summer, 1998) http://www.rietica.org
- 32V. K. Pecharsky, A. P. Holm, K. A. Gschneidner, Jr., and R. Rink, Phys. Rev. Lett. 91, 197204 (2003).
- 33Ya. Mudryk, A. P. Holm, K. A. Gschneidner, Jr., and V. K. Pecharsky, Phys. Rev. B 72, 064442 (2005).
- 34A. P. Holm, V. K. Pecharsky, K. A. Gschneidner, Jr., R. Rink, and M. Jirmanus, Rev. Sci. Instrum. **75**, 1081 (2004).
- 35V. K. Pecharsky, J. O. Moorman, and K. A. Gschneidner, Jr., Rev.

Sci. Instrum. 68, 4196 (1997).

- ³⁶N. W. Ashcroft and N. D. Mermin, *Solid State Physics* (Holt, Reinhart and Winston, New York, 1976).
- ³⁷ J. H. Van Vleck, *The Theory of Electric and Magnetic Suscepti*bilities (Oxford University Press, London, 1932).
- ³⁸ A. Frank, Phys. Rev. **48**, 765 (1935).
- 39H. W. De Wijn, A. M. Van Diepen, and K. H. J. Buschow, Phys. Rev. 161, 253 (1967).
- ⁴⁰ A. M. Stewart, Phys. Rev. B **6**, 1985 (1972).
- ⁴¹Z. Liu, Phys. Rev. B **64**, 144407 (2001).
- ⁴² H. Adachi, H. Ino, and H. Miwa, Phys. Rev. B **56**, 349 (1997).
- ⁴³ H. Adachi, H. Ino, and H. Miwa, Phys. Rev. B **59**, 11445 (1997).
- ⁴⁴ H. Adachi and H. Ino, Nature (London) **401**, 148 (1999).
- 45H. Adachi, H. Kawata, H. Hashimoto, Y. Sato, I. Matsumoto, and Y. Tanaka, Phys. Rev. Lett. 87, 127202 (2001).
- ⁴⁶D. Wagner, *Introduction to Theory of Magnetism* (Pergamon, Oxford, 1972).
- 47H. C. Hamaker, L. D. Woolf, H. B. MacKay, Z. Fisk, and M. B. Maple, Solid State Commun. 32, 289 (1979).
- 48W. M. Yuhasz, N. A. Frederick, P.-C. Ho, N. P. Butch, B. J. Taylor, T. A. Sayles, M. B. Maple, J. B. Betts, A. H. Lacerda, P. Rogl, and G. Giester, Phys. Rev. B 71, 104402 (2005).
- 49V. K. Pecharsky and K. A. Gschneidner, Jr., J. Appl. Phys. **86**, 565 (1999).
- 50V. K. Pecharsky and K. A. Gschneidner, Jr., in *Magnetism and Structure in Functional Materials*, edited by A. Planes, L. Mañosa, and A. Saxena (Springer-Verlag, Berlin, 2005), Chap. 11, p. 199.
- 51M. Zou, Ya. Mudryk, V. K. Pecharsky, K. A. Gschneidner, Jr., D. L. Schlagel, and T. A. Lograsso, Phys. Rev. B **75**, 024418 $(2007).$
- 52R. Nirmala, Ya. Mudryk, V. K. Pecharsky, and K. A. Gschneidner, Jr. (unpublished).