Enhanced magnetic entropy in GdNi₂

N. V. Baranov,^{1,2,*} A. V. Proshkin,^{1,2} E. G. Gerasimov,¹ A. Podlesnyak,^{3,4} and J. Mesot³

¹Institute for Metal Physics RAS, 620219 Ekaterinburg, Russia

²Institute of Physics & Applied Mathematics, Ural State University, 620083 Ekaterinburg, Russia

³Laboratory for Neutron Scattering, ETH Zürich & Paul Scherrer Institut, CH-5232 Villigen PSI, Switzerland

⁴Hahn-Meitner-Institut, SF-2, Glienicker Straße 100, D-14109 Berlin, Germany

(Received 26 May 2006; revised manuscript received 30 January 2007; published 12 March 2007)

Measurements of the specific heat have been performed on $Gd_{1-x}Y_xNi_2$ (x=0,0.2) compounds and their nonmagnetic analogs $Lu_{1-y}Y_yNi_2$, which have similar molar masses. It is found that the difference between the entropies of magnetic and nonmagnetic compounds with identical molar masses surpasses substantially (by 14-19%) the theoretical limit for the magnetic contribution $S_m = (1-x)R \ln(8)$ calculated assuming that only Gd ions possess a magnetic moment. This observed enhancement of the magnetic entropy in $Gd_{1-x}Y_xNi_2$ is believed to result from spin fluctuations induced by *f*-*d* exchange in the 3*d* electron subsystem of Ni.

DOI: 10.1103/PhysRevB.75.092402

PACS number(s): 75.20.En, 65.40.Ba, 75.30.Kz

A well known observation for rare earth (R)-3dtransition-metal (T) binary-intermetallic compounds is the reduction of the magnetic moment per 3d metal atom (M_T) with increasing rare earth content.¹ For example, the magnetic moment per Co atom is found to disappear when the Rconcentration exceeds 1/3. The Co 3d subsystem in the boundary RCo₂ compounds reveals an instability and itinerant metamagnetism.² For the R-Ni series, the Ni magnetic moment is observed to vanish even for RNi₅ compounds, while a reappearance of the 3d band splitting was found upon further increase of the *R* content $[Y_2Ni_7 \text{ and } YNi_3 \text{ (Ref.)}]$ 1)]. Until recently, the Ni magnetic moment in compounds with higher R concentration (RNi_2 , RNi, R_3Ni) was taken to be zero. The evolution of the M_T value in R-T intermetallic compounds was explained by a gradual filling of the T 3dband by outer-shell electrons of R atoms.^{2,3} However, recent x-ray magnetic circular dichroism⁴ (MCD) and magnetic Compton profile⁵ experiments have shown that the Ni 3dband in GdNi2 is not fully occupied and the Ni atoms exhibit a magnetic moment of about $(0.2-0.23)\mu_{\rm B}$. These results are in agreement with bulk magnetization measurements.⁶ The presence of a magnetic moment on Ni atoms up to $\sim 0.1 \mu_{\rm B}$ was revealed by MCD measurements even for GdNi.⁷ In all these cases the Ni magnetic moment was found to be antiparallel to the Gd magnetic moment, as in other R-T intermetallics with heavy rare earths. For compounds with a high R content, such as GdNi (Ref. 8) and Gd_3Ni (Ref. 9) the values of the effective magnetic moment per Gd ion were found to be enhanced in comparison with the μ_{eff} value of the free Gd³⁺ ion.

In the present work we have investigated the specific heat and entropy content of the GdNi₂ compound. Due to potential applications for magnetic cooling,^{10,11} the attention on specific heat and magnetocaloric effect studies has recently substantially increased.

According to previous investigations of the crystal structure of RNi_2 ,¹² which show that single-phase samples can be obtained with some *R* deficiency, we have prepared several GdNi₂ samples with various deviations from the 1:2 stoichiometry (0.96:2; 0.98:2, and 1:2). All compounds were prepared by melting in levitation in an argon atmosphere. The ingots obtained were annealed at 800 °C for 2 weeks. X-ray diffraction and metallographic analysis were used to examine the phase purity of the compounds. The best result was obtained for the compound with 0.96:2 stoichiometry, in agreement with the data of Lindbaum et al.¹² The crystal structure of the sample was identified as a superstructure of the cubic Laves structure (C15).¹² Usually, in order to calculate the lattice contribution $C_{latt}(T)$ to the total specific heat one uses the data of a nonmagnetic isostructural analog. The data are then compared by assuming a correction to the Debye temperature θ_d , which takes into account the difference in molar mass of the two compounds. Such a procedure may lead to an appreciable error, especially in cases where the molar mass difference is substantial or if the lattice specific heat cannot be well described in the frame of a simple Debye model. This last point is precisely relevant to the $C_{latt}(T)$ behavior of RNi₂. As shown in Ref. 13 a good description of the lattice specific heat of LuNi2 can be obtained only under the assumption that θ_d depends on temperature in the range 220-280 K. Therefore, in order to achieve a correct evaluation of the lattice contribution to the specific heat, we prepared using the above-mentioned conditions the $Gd_{1-x}Y_xNi_2$ (x=0,0.2) compounds and their nonmagnetic analogs $Lu_{1-\nu}Y_{\nu}Ni_{2}$ with the same molar masses. Measurements of the specific heat as well as ac magnetic susceptibility were performed using a Quantum Design PPMS-6000 system; the dc magnetization and susceptibility were measured by using a Quantum Design MPMS 5XL magnetometer.

Figure 1(a) shows the temperature dependencies of the specific heat of GdNi₂ and its nonmagnetic analog Lu_{0.794}Y_{0.206}Ni₂. For GdNi₂ a distinct anomaly is observed at $T_c \sim 75$ K while the specific heat for the reference compound increases continuously with increasing temperature. The sharp peak at $T_c \sim 75$ K on the $C_p(T)$ curve of GdNi₂ is associated with the magnetic ordering temperature. As follows from the inset in Fig. 1(a), at the same temperature a sharp fall of the ac susceptibility is observed. The Lu_{0.794}Y_{0.206}Ni₂ compound exhibits a paramagnetic behavior with a weak temperature dependence of the magnetic susceptibility, which is associated with Pauli paramagnetism. The Curie temperature observed for GdNi₂ is close to that ob-



FIG. 1. (Color online) (a) Temperature dependence of the specific heat for GdNi₂ and its nonmagnetic partner Lu_{0.794}Y_{0.206}Ni₂. The solid line corresponds to the nonmagnetic specific heat calculated with γ =5.2 mJ mol⁻² K⁻² and θ_d =241 K. The inset shows the temperature dependence of the ac susceptibility for GdNi₂. (b) Magnetic part of the specific heat for GdNi₂ as a function of temperature (solid line). The dashed line shows the *C_m*(*T*) dependence calculated in the molecular field approximation.

tained in previous works from specific heat measurements [73.7 K (Ref. 14) and 75.3 K (Ref. 15)]. Bulk magnetization and susceptibility data for our GdNi2 were found to be well consistent with those published by Yano et al.⁶ The specific heat of Lu_{0.794}Y_{0.206}Ni₂ includes the lattice and electronic contributions only, $C_p(T) = C_{el} + C_{latt} = \gamma T + C_{latt}$. In view of the closeness of molar masses of GdNi2 and Lu0.794Y0.206Ni2 compounds, we assume that the sum $C_{el} + C_{latt}$ is the same for both compounds. Therefore, the magnetic contribution C_m to the specific heat of GdNi₂ may be obtained by a simple subtraction procedure. The temperature variation of this magnetic contribution is presented in Fig. 1(b) by a solid line. The dashed line corresponds to the $C_m(T)$ dependence calculated in the molecular field approximation assuming that only the Gd subsystem contributes to the magnetic specific heat. It turns out that the magnetic contribution to the specific heat of GdNi₂ persists well above the magnetic ordering temperature. The presence of spin fluctuations and shortrange magnetic order above T_c was also revealed in previous electrical resistivity measurements¹⁶ of GdNi₂.

In Fig. 2 we plot C/T versus T^2 in the low-temperature region for both GdNi₂ and Lu_{0.794}Y_{0.206}Ni₂ compounds. In this temperature interval $C_p(T)$ can be presented for both compounds as a sum of *T*-linear and T^3 terms. There is clear evidence that these compounds exhibit significantly different values of the *T*-linear specific heat coefficient γ . We find that $\gamma=22$ mJ mol⁻¹ K⁻² for GdNi₂ and $\gamma=5.2$ mJ mol⁻¹ K⁻² for the nonmagnetic analog, respectively. The γ value for Lu_{0.794}Y_{0.206}Ni₂ is close to that obtained experimentally for YNi₂ [5.2 (Ref. 17)] and LuNi₂ [5.4 (Ref. 18) and 6.58 (Ref.



FIG. 2. C_p/T versus T^2 dependencies of GdNi₂ and Lu_{0.794}Y_{0.206}Ni₂.

13)] and to the value 3.67 estimated from density of states $(\tilde{\text{DOS}})$ calculations¹⁹ for YNi₂ (all values are in mJ mol⁻¹ K⁻²). The Gd 4*f* electron states are located far below the Fermi level E_f , and do not contribute to the DOS at E_{f} . Therefore, the difference in the observed γ values may be associated with (i) the change in DOS near E_f due to the 3d band splitting originating from magnetic ordering of GdNi₂; (ii) the presence of an additional contribution from spin fluctuations induced by f-d exchange in the Ni 3d electron subsystem; and (iii) the electron magnon enhancement. The Fermi level in RNi₂ compounds was found to be located on the flat part of the DOS curve N(E) with a relatively low DOS value.²⁰ This, together with the low value of the magnetic moment measured on Ni atoms implies only a weak change of $N(E_f)$ through the magnetic transition. As to the increase of γ caused by the electron-magnon mass enhancement, it was estimated to be about 10% for pure Gd.²¹ This cannot explain such a large difference assuming that the electron-phonon enhancement of γ is the same for GdNi₂ and Lu_{0.794}Y_{0.206}Ni₂. Therefore, one can suggest that the additional contribution to the γ value in GdNi₂, $\Delta \gamma = \gamma_{\text{GdNi}_2}$ $-\gamma_{(Lu,Y)Ni_2}$, results from spin fluctuations induced by $f-d \exp^2 - \gamma_{(Lu,Y)Ni_2}$ change in the 3*d* electron subsystem of Ni.

It is worth noticing that a strong influence of spin fluctuations induced by f-d exchange was observed for other Gd_nNi_m compounds.²² In particular, the γ value for Gd_3T with T=Ni, Co was found to be about one order of magnitude higher ($\gamma = 100 - 170 \text{ mJ mol}^{-1} \text{ K}^{-2}$) than that obtained for isostructural Y_3T compounds (γ $=11-15 \text{ mJ mol}^{-1} \text{ K}^{-2}$.^{23,24} From the slope of the *C*/*T* versus T^2 dependence of Lu_{0.794}Y_{0.206}Ni₂ (Fig. 2) we estimated the Debye temperature to be equal to 241 K. The $C_p(T)$ $=C_{el}+C_{latt}$ dependence calculated within the Debye model and using this θ_d value and $\gamma = 5.2 \text{ mJ mol}^{-1} \text{ K}^{-2}$ is shown by the solid line in Fig. 1. As can be seen, such a procedure is not applicable for the description of the nonmagnetic contribution to the total specific heat of GdNi₂. As to the increased slope of the C/T versus T^2 dependence for GdNi₂, it may result from the T^3 spin wave contribution because of an antiferromagnetic alignment of Gd and Ni magnetic moments.

Since the total entropy of $GdNi_2$ includes the electronic, lattice, and magnetic contributions

$$S = S_{el} + S_{latt} + S_m, \tag{1}$$

the S_m value can be obtained by subtraction of the temperature dependencies of the entropy for $GdNi_2$ and



FIG. 3. (a) Temperature dependencies of entropies for $GdNi_2$ and $Lu_{0.794}Y_{0.206}Ni_2$. (b) Magnetic contribution to the entropy of $GdNi_2$. Dashed line indicates the additional electronic contribution to the entropy $\Delta S_{el} = \Delta \gamma T$; the chain line shows the temperature dependence of the magnetic entropy after subtraction of the additional contribution. Horizontal solid line corresponds to the maximal theoretical value $S_m^{theor} = R \ln(8)$ associated with Gd.

 $Lu_{0.794}Y_{0.206}Ni_2$: $S = S_{GdNi_2} - S_{(Lu,Y)Ni_2}$. The S(T) curves calculated by integration of $\vec{C_p}(T)/T$ versus T dependencies for both compounds and the difference $S_m(T)$ are displayed in Figs. 3(a) and 3(b). Unlike previous specific heat studies, ^{14,15} we obtain that the magnetic entropy of GdNi2 above the Curie temperature exceeds substantially the maximal theoretical value $S_m^{theor} = R \ln(2J+1) = 17.29 \text{ J mol}^{-1} \text{ K}^{-1}$, which was calculated assuming that only Gd ions have a magnetic moment. The S_m value obtained for GdNi₂ reaches $\sim 20.6 \text{ J} \text{ mol}^{-1} \text{ K}^{-1}$ with increasing temperature up to 200 K. Bearing in mind the presence of the above-mentioned discrepancy in the γ values, such an enhanced difference between the entropies for GdNi₂ and its nonmagnetic partner observed above T_c may be associated with an additional contribution to the first term in Eq. (1), i.e., $S_{el} = (\gamma_0 + \Delta \gamma)T$. Here we suppose that the γ_0 value for GdNi₂ is equal to $\gamma_{(Lu,Y)Ni_2}$ and it is determined by the DOS value at the Fermi level. Assuming that the value of $\Delta \gamma$ does not depend on the temperature we plotted this additional contribution to the entropy $\Delta S_{el} = \Delta \gamma T$ in Fig. 3(b) by a dashed line. As it turns out, the magnetic entropy obtained after subtraction of ΔS_{el} from the $S_m(T)$ becomes close to the expected value S_m^{theor} = $R \ln(8)$ above T_c [shown by solid line in Fig. 3(b)]. Therefore, one may conclude that in GdNi2 the excess of the magnetic entropy above T_c originates from additional spinfluctuation contributions induced by f-d exchange. It should be noted however, that $C_p(T)$ for both magnetic and nonmagnetic compounds become identical at temperatures larger than 150 K [see Fig. 1(a)] which implies for GdNi₂ a decrease of $\Delta \gamma$ with increasing temperature.

In order to confirm these results we have also measured the specific heat of $Gd_{0.8}Y_{0.2}Ni_2$ and its nonmagnetic analog $Lu_{0.635}Y_{0.365}Ni_2$, which have nearly the same molar masses. $C_p(T)$ for these compounds are presented in Fig. 4(a). As



FIG. 4. (a) Temperature dependence of the specific heat for $Gd_{0.8}Y_{0.2}Ni_2$ and its nonmagnetic partner $Lu_{0.635}Y_{0.365}Ni_2$. Inset shows the magnetization per Gd ion as a function of magnetic field at T=2 K. (b) Magnetic contribution to the entropy of $Gd_{0.8}Y_{0.2}Ni_2$. Inset shows the C_p/T vs T^2 dependencies for $Gd_{0.8}Y_{0.2}Ni_2$ (full circles) and $Lu_{0.635}Y_{0.365}Ni_2$ (open circles). Horizontal line corresponds to the appropriate S_m^{theor} value for $Gd_{0.8}Y_{0.2}Ni_2$.

expected, the substitution of Y for Gd decreases the magnetic ordering temperature. For Gd_{0.8}Y_{0.2}Ni₂ the specific heat maximum which corresponds to the Curie temperature is observed at 60 K. The saturation magnetization of Gd_{0.8}Y_{0.2}Ni₂ calculated per Gd ion [see inset in Fig. 4(a)] is found to be $\sim 6.8 \mu_{\rm B}$, which implies the presence of a small Ni magnetic moment in the opposite direction to that in the nondiluted GdNi₂. It should be noted that the total magnetization may also include the contribution from polarized conduction electrons and 5d electrons of Gd ions. Above 150 K, $C_p(T)$ becomes very close to that of Lu_{0.635}Y_{0.365}Ni₂. The magnetic part of the entropy calculated as in the previous case reaches a value of about 15.8 J mol⁻¹ K⁻² which surpasses substantially the expected value of 13.8 J mol⁻¹ K⁻² [see Fig. 4(b)]. The Gd_{0.8}Y_{0.2}Ni₂ compound exhibits also an enhanced value of the coefficient $\gamma = 34$ mJ mol⁻¹ K⁻² in comparison with the nonmagnetic reference 5.1 mJ mol⁻¹ K⁻² [shown in the inset in Fig. 4(b)]. These data support the results obtained for the nondiluted GdNi₂ compound.

In conclusion, the study of the specific heat of $Gd_{1-x}Y_xNi_2$ compounds with x=0 and 0.2 and their nonmagnetic analogs $Lu_{1-y}Y_yNi_2$ with y=0.206 and 0.365 has revealed that the difference between the entropies of magnetic and nonmagnetic compounds with the same molar masses surpasses substantially (by 14–19 %) the theoretical limit for the magnetic contribution $S_m = (1 - x)R \ln(8)$ calculated under the assumption that only Gd ions have a magnetic moment. However, this additional magnetic contribution to the entropy seems to be underestimated since the retention of short-range magnetic order above the Curie temperature was not taken into account. Analysis of the specific heat behavior of some Gd compounds with nonmagnetic partners²⁵ has shown that at the magnetic ordering temperature, the magnetic entropy of Gd compounds reaches a substantially lower value (about 85%) than its expected limit. The same was observed for Gd metal.²⁶ At T=360 K, i.e., well above the Curie temperature ($T_c \sim 294$ K), the magnetic entropy of Gd metal reaches only 94% of its full value. These data allow us to suggest that in GdNi₂ the additional contribution to the magnetic entropy which originates in the Ni subsystem may reach 25–30 %. The values of the coefficient γ of the *T*-linear specific heat for Gd_{1-x}Y_xNi₂ are observed to exceed significantly (by 4–6 times) the γ values for Lu_{1-v}Y_vNi₂.

All these results suggest the presence of additional contributions arising from the Ni 3*d* electron subsystem and do not support a charge transfer model which implies the full filling of the 3*d* band in *R*-*T* intermetallics at high rare earth content. The enhanced magnetic entropy and γ value observed for Gd_{1-x}Y_xNi₂ originate presumably from additional spin-fluctuation contributions induced by *f*-*d* exchange.

It should be emphasized that, in order to be convinced by the results obtained for $GdNi_2$ we have, in the present work, synthesized the $Gd_{0.8}Y_{0.2}Ni_2$ compound and its nonmagnetic analog with the same molar mass and performed specific heat measurements for these samples. We consider the enhanced magnetic entropy obtained for $Gd_{0.8}Y_{0.2}Ni_2$ as an additional confirmation of our data for $GdNi_2$. The results obtained in our work could not be explained by existing theories. We hope that our work will stimulate further theoretical investigations of the entropy content in rare earth– transition-metal compounds.

This work was supported by the Russian Foundation for Basic Research (Grant No. 04-02-96060), by the Program 2.1.1.6945 of the Russian Ministry for Education and Science, and by the Swiss National Science Foundation (SCOPES, Project No. IB7420-110849).

*Corresponding author. Email address: nikolai.baranov@usu.ru

- ¹H. Kirchmayr and C. A. Poldy, in *Handbook on Physics and Chemistry of Rare Earths*, edited by K. A. Gschneidner, Jr. and L. Eyring (North-Holland, Amsterdam, 1979), p. 55.
- ²N. H. Duc and P. E. Brommer, in *Handbook of Magnetic Materials*, edited by K. H. J. Buschow (Elsevier Science, Amsterdam, 1999), Vol. 12.
- ³K. Taylor, Adv. Phys. **20**, 603 (1971).
- ⁴M. Mizumaki, K. Yano, I. Umehara, F. Ishikawa, K. Sato, A. Koizumi, N. Sakai, and T. Muro, Phys. Rev. B 67, 132404 (2003).
- ⁵K. Yano, Y. Tanaka, I. Matsumoto, I. Umehara, K. Sato, H. Adachi, and H. Kawata, J. Phys.: Condens. Matter **19**, 6891 (2006).
- ⁶K. Yano, I. Umehara, T. Miyazawa, Y. Adachi, and K. Sato, Physica B **367**, 81 (2005).
- ⁷K. Yano, I. Umehara, K. Sato, and A. Yaresko, Solid State Commun. **136**, 67 (2005).
- ⁸R. Mallik, P. L. Paulose, E. V. Sampathkumaran, S. Patil, and V. Nagarajan, Phys. Rev. B **55**, 8369 (1997).
- ⁹E. Talik, Physica B **193**, 213 (1994).
- ¹⁰ K. A. Gschneidner, Jr., V. K. Pecharsky, and A. O. Tsokol, Rep. Prog. Phys. **68**, 1479 (2005).
- ¹¹ P. J. von Ranke, D. F. Grangeia, A. Caldas, and N. A. de Oliveira, J. Appl. Phys. **93**, 4055 (2003).
- ¹²A. Lindbaum, E. Gratz, and S. Heathman, Phys. Rev. B 65, 134114 (2002).
- ¹³J. Ćwik, T. Palewski, K. Nenkov, G. S. Burkhanov, O. D. Chistyakov, N. Kolchugina, and H. Mädge, Physica B **358**, 323

(2005).

- ¹⁴J. A. Cannon, J. I. Budnick, R. S. Craig, S. G. Sankar, and D. A. Keller, in *Magnetism and Magnetic Materials—1972*, edited by C. D. Graham, Jr. and J. J. Rhyne, Conf. AIP Proc. No. 10 (AIP New York, 1973), Vol. 2, p. 905.
- ¹⁵ J. J. Melero, R. Burriel, and M. R. Ibarra, J. Magn. Magn. Mater. 15, 841 (1995).
- ¹⁶M. P. Kawatra, S. Skalski, J. Mydosh, and J. Budnick, Phys. Rev. Lett. 23, 83 (1969).
- ¹⁷D. Bloch, D. Camphausen, J. Voiron, J. Aasse, A. Berton, and J. Chaussy, C.R. Seances Acad. Sci., Ser. A **275**, 601 (1972).
- ¹⁸A. F. Deutz, H. B. Brom, C. D. Wentworth, W. J. Huiskamp, L. J. de Jongh, and K. H. J. Buschow, J. Magn. Magn. Mater. **78**, 176 (1989).
- ¹⁹K. Nakada, H. Shimizu, and H. Yamada, Physica B **329–333**, 1129 (2003).
- ²⁰H. Yamada, Physica B & C **149**, 390 (1988).
- ²¹D. M. Eagles, J. Magn. Magn. Mater. **28**, 117 (1982).
- ²²N. V. Baranov *et al.* (unpublished).
- ²³N. V. Baranov, A. A. Yermakov, P. E. Markin, U. M. Possokhov, H. Michor, B. Weingartner, and G. Hilscher, J. Alloys Compd. 329, 22 (2001).
- ²⁴G. Hilscher, H. Michor, N. V. Baranov, P. E. Markin, and A. A. Yermakov, Acta Phys. Pol. B **34**, 1205 (2003).
- ²⁵M. Bouvier, P. Lethuillier, and D. Schmitt, Phys. Rev. B 43, 13137 (1991).
- ²⁶F. Jelinek, B. Gerstein, M. Griffel, R. Skochdopole, and F. Spedding, Phys. Rev. **149**, 489 (1966).