Enhanced tunneling magnetoresistance and high-spin polarization at room temperature in a polystyrene-coated Fe₃O₄ granular system

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Polystyrene-coated $Fe₃O₄$ nanoparticles through surface engineering exhibit an intergranular tunneling magnetoresistance (MR) ratio of 22.8% at room temperature and a maximum MR of 40.9% at 110 K. The drastic enhancement of the MR ratio clearly suggests that there is high degree of spin polarization even at room temperature for half metallic Fe₃O₄. The estimated spin polarization *P* is about 54% and 83% at room temperature and 110 K, respectively.

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I. INTRODUCTION

The half-metal family that includes CrO_2 , Fe_3O_4 , and $La_{2/3}Sr_{1/3}MnO₃$ attracts enormous attention, owing to their speculated 100% spin polarization *P* according to band structure calculations. They are conductors for one spin channel and insulators/semiconductors for the other spin channel and have great application potential in spin electronic devices.^{1,2} On the experimental side, $CrO₂$ best exemplifies the halfmetals with the value of *P* reported in the range of 85–96% at very low temperatures. However *P* rapidly declines with increasing temperature. Other materials fail to show high *P* for a variety of reasons, including disorders and defects, particularly on the surfaces.^{2,3} As far as we know, clear experimental evidence showing high-spin polarization at room temperature is lacking for all known speculated half-metals.4

Ferrimagnetic magnetite $Fe₃O₄$ is a well-known halfmetal with a high Curie temperature T_c of 840 K. It offers the best prospect for spintronic applications since the devices are expected to work around or above room temperature, requiring T_c above 500 K.³ Under ordinary condition, the surface of $Fe₃O₄$ is oxidized and contains $Fe(3+)$ -rich oxide, which is believed to be the reason for the failure to observe high-spin polarization in $Fe₃O₄$. To reveal the true spin polarization of Fe₃O₄, the Fe(3+)-rich oxide on the surface should be removed or avoided. We have selected polymers polystyrene (PS), polycarbonate (PC), and poly(methyl methacrylate) (PMMA) as the coating layer to avoid the surface oxidation of $Fe₃O₄$. These polymers also serve as good insulating barriers between the $Fe₃O₄$ nanoparticles.

II. EXPERIMENTS

Polymers (PS, PC, and PMMA) and α -Fe₂O₃ nanoparticles with various weight ratios were mixed together by first dissolving the polymer in chloroform, then adding $Fe₂O₃$ particles and stirring, and finally evaporating the solvent. The samples were annealed at 200 °C in pure hydrogen flow, and then pressed into pellets. The pellets were again annealed at 200 and 250 °C in pure hydrogen flow.

The structural analysis was done by x-ray diffraction (XRD) and transmission electron microscopy (TEM). The transport properties of the samples were measured using the four-point method and in ac mode. The magnetic measurements were performed with a superconducting quantum interference device (SQUID).

III. RESULTS AND DISCUSSION

XRD spectra indicate that there is a complete phase transformation from α -Fe₂O₃ to Fe₃O₄ after annealing in hydrogen. TEM images show that the $Fe₃O₄$ particles are generally spherical and their size is between 10 and 30 nm [Fig. $1(a)$]. They are dispersed in the polymer matrix. Some particles are close to each other but separated by the polymer [Fig. $1(b)$], which provides the structure for intergranular tunneling of electrons.

The temperature dependence of the resistance (R) follows ln $R \in T^{-1/2}$, and the I-V curves are nonlinear (Fig. 2, insets). These results suggest that the electron transport in the samples is via intergranular tunneling, which has been broadly accepted as exhibiting $exp(T^{-1/2})$ behavior in the resistance.^{5,6} We have tested the exp(T^{-1/4}) form expected for variable range hopping resistance and found that *R* could not be described by such a law. The data presented in Fig. 2 are reproducible. Similar behaviors are seen in other samples.

The magnetoresistance (MR) of the samples, defined as $MR = (R_H - R_0)/R_0$, where R_0 and R_H is the resistance in zero and an applied field of *H*, respectively, is shown in Fig. 2. The MR of the $Fe₃O₄$ samples coated with PC [Fig. 2(a)] is similar to the reported data on pressed $Fe₃O₄$ powders and polycrystalline films,7–10 which have a typical MR ratio of $4-7\%$ or lower at room temperature. The Fe₃O₄ samples coated with PMMA have almost identical results with that coated with PC. On the other hand, the sample coated with PS exhibits a MR ratio of 22.8% in an applied field of 14 T at room temperature. The maximum MR of 40.9% is obtained at $110 K$ [Fig. 2(b)]. The MR curves show rapid change in low fields but are not completely saturated in a

FIG. 1. (a) TEM image for the PS-coated $Fe₃O₄$ sample, indicating that spherical $Fe₃O₄$ nanoparticles are embedded in polymer matrix. (b) High-resolution TEM image showing that the $Fe₃O₄$ particles are separated by a thin polymer layer of a few nanometers, forming a tunnel barrier.

field of 14 T. In the low-field region, butterfly shaped hysteresis in MR curves has been observed, and the coercivities in the MR coincide with that in the magnetization curves for both low and high temperatures [Figs. $2(c)$ and $2(d)$].

There are several models that can be used to calculate the spin polarization *P* from the MR data. Inomura and Maekawa have derived a simple relationship between MR and P for intergranular tunneling,⁶

$$
MR = P^2m^2/(1 + P^2m^2),\tag{1}
$$

where *m* is the relative magnetization of the system and $m^2 = \langle \cos \theta \rangle$. In saturated state, $m = 1$, then

FIG. 2. MR ratio in an applied field of 14 T at 280 and 110 K for (a) PC-coated Fe₃O₄ and (b) PS-coated Fe₃O₄. The weight ratio of $Fe₃O₄$ to polymers is 1:1 for both. The MR ratio of PMMAcoated $Fe₃O₄$ is similar to that of PC-coated samples. (Left insets) resistance as a function of temperature: log *R* versus *T*−1/2 curves exhibit linear relation; (Right insets) I-V curves at room temperature which have nonlinear behavior. These are consistent with the transport mechanism of intergranular tunneling. (c) Magnetic hystereses and (d) butterfly-shaped MR hyestereses in low fields for PS-coated $Fe₃O₄$.

$$
P = [MR_s/(1 - MR_s)]^{1/2},
$$
 (2)

where MR_s is the saturated MR ratio. A modification of the model is to consider a serial connection of the grains in addition to parallel connections used in the model. $11-13$ The difference between these models becomes very small when three-dimensional (3D) nature of the network of grains is considered. This is true even for relatively high-spin polarization $P > 0.5$.^{12,13} Our composite pellet samples fall well within the $3D$ regime, and Eq. (2) was used to calculate the spin polarization. It should be mentioned that the models proposed by Slonczewski¹⁴ and MacLaren¹⁵ suggest the nature of the barrier is a factor influencing the effective spin polarization. Such effect diminishes with increasing barrier height. The polymer barriers used here are in general of very large band gap $({\sim}7 \text{ eV})$; therefore, we believe the effect of the barrier on the spin polarization is small in this case. More importantly, as long as one understands that the *P* obtained from Eq. (2) is the effective spin polarization, the result is valid.

The spin polarization P of $Fe₃O₄$ estimated from the MR values according to Eq. (2) is 83% at low temperature and 54% at room temperature. These values are much higher than the reported experimental results^{16–18} and higher than a recent theoretical calculation after taking into the consideration of modified surface state.¹⁹ Fe₃O₄ indeed belongs to the category of highly spin polarized half-metals. The more than 50% value for spin polarization at room temperature is significant in that this is the first time that such a high-spin polarization has been observed in $Fe₃O₄$ at room temperature, which has both practical and scientific implications. Coulomb blockade effect is believed to be another factor to sometimes contribute to the enhancement of MR; however, it occurs only at very low temperature^{20,21} and should not play a significant role here. We believe these *P* values only set lower limits for $Fe₃O₄$ and its actual values can be higher, especially at room temperature, because

(1) There may exist spin-independent conductance channels owing to the imperfections, defects, and impurities in the barrier in our samples, which reduces the tunneling MR ratio.22

(2) Bulk magnons and surface magnons will reduce the MR via magnon-assisted tunneling.²³ Although they may also reduce the spin polarization itself, 4 theoretical studies indicate that the MR ratio will decrease more rapidly with temperature than *P*. 24

(3) Even an applied field of 14 T may not be high enough to completely align the magnetic moments of the $Fe₃O₄$ particles of 10– 30 nm in size in our samples, especially those on the surface. It should be noted that using MR data taken at 14 T to calculate *P* is justified because the intrinsic magnetoresistance of $Fe₃O₄$ is very small in such a field.⁷

(4) We have assumed that the relative magnetization $m=1$ in the calculation of *P* using Eq. (2), but it takes a reduced value at high temperatures.

It ought to be noted that there are different ways to define spin polarization.³ Some measure the spin polarization of the density of states and some measure that of the transport current density. The spin polarization obtained from our intergranular tunneling experiments is the spin polarization of the tunneling current. We would like to argue that the spin polarization of density of states is also high for $Fe₃O₄$. The difference between the two definitions becomes significant when there are "heavy" (e.g., *d* electrons) and "light" electrons (e.g., *s* electrons) coexisting at the Fermi level, which is not the case for $Fe₃O₄$. For $Fe₃O₄$, $t_{2g}(Fe)$ electrons form small polarons and hop among the *B* sites of the inverse spinel structure in a fully spin-polarized spin-down band.

FIG. 3. (a) Temperature dependence of MR ratio in an applied field of 14 T for a PS-coated sample. (b) ZFC-FC curves with an applied field *H*= 200 Oe, which shows sharp Vervey transition in the range of $110-120$ K in our samples.

Therefore, we believe the two numbers, 54% and 83%, may also represent the approximate values for the lower limits of the spin polarization of the density of states at room temperature and 110 K, respectively.

It is necessary to study the temperature dependence of MR since $Fe₃O₄$ undergoes a Verwey transition, which is characterized by an increase in the resistivity by about two orders of magnitude at the transition temperature $Tv \sim 120$ K. This transition is associated with an orderdisorder transition from a charge-ordered state of the Fe ion on the *B* sites at low temperature to a statistical distribution at high temperature. A sharp narrow negative MR peak is normally observed at the Verwey point in single crystal $Fe₃O₄$.^{25,26} In our samples, the MR ratio continuously increases with decreasing temperature before the Verwey transition [Fig. $3(a)$]. After the transition, the MR ratio exhibits a plateau between 80 and 120 K. We cannot acquire MR data below 80 K because the resistance of the samples becomes too high to measure with our setup. According to the zerofield-cooled (ZFC) and field-cooled (FC) magnetization curves [Fig. 3(b)], the Verwey transition is quite sharp and occurs in a relative narrow temperature range of 110– 120 K in our samples. This suggests that MR observed in our samples can be used to calculate the spin polarization because it is not part of the sharp peak ordinarily associated with the Verwey transition. The latter does not arise from the spin polarization but is a critical phenomenon at the phase transition and thus cannot be used for deriving spin polarization. The Verwey transition does not significantly change the tunneling MR and the spin polarization, consistent with reported results.^{16,17} To understand the temperature dependence of MR ratio in our samples, we propose the following model. We think there exist two channels of conductance.

FIG. 4. Al $K\alpha$ -exited Fe 2p core-level photo emission spectra for (I) pure $Fe₃O₄$ powder sample, (II) PC-coated $Fe₃O₄$, (III) PMMA-coated Fe₃O₄, and (IV) PS-coated Fe₃O₄. The arrow indicates the characteristic shake-up satellite associated with $Fe(3+)$ ion photoemission at a binding energy of \sim 719 eV.

One is the intergranular spin-dependent channel and the other is a spin-independent channel due to thermal excitation or inelastic hopping through localized states due to imperfections in the barrier, etc. Above the Verwey transition, the current of the spin-independent channel rapidly decreases with decreasing temperature, whereas the current of the spindependent channel decreases relatively slower. This results in the enhancement of tunneling MR ratio with the decrease of temperature. Below the transition, the resistance of $Fe₃O₄$ increases rapidly with decreasing temperature. The number of carriers available for tunneling decreases accordingly, which greatly diminishes the spin-dependent tunneling current. At the same time, the spin-independent current decreases with temperature as well. The plateau in the MR ratio below the Verwey point is the combined effects of these two channels.

It is widely accepted that the surface state of $Fe₃O₄$ plays a key role in magnetotransport properties, and x-ray photoelectron spectroscopy (XPS) is one of the most powerful tools to obtain information about the electronic structure of a solid's surface. Figure 4 shows the XPS Fe 2*p* core-level

spectra for pure $Fe₃O₄$ powders and PS-, PMMA-, and PCcoated $Fe₃O₄$ samples, which contain contributions from the top 15 layers of the surface. Different from the polymercoated Fe₃O₄ samples, the lineshape of the pure Fe₃O₄ sample exhibits shake-up satellite at a binding energy of ~719 eV and a little narrow peak of the Fe $2p_{3/2}$ peak, which is characteristic of Fe $(3+)$ oxide.^{27,28} This clearly demonstrates that the surface of $Fe₃O₄$ is $Fe(3+)$ oxide once it is exposed to air. The lineshapes of PS-, PMMA-, and PC-coated samples reveal the characteristics of $Fe₃O₄$ and there is no obvious difference among them, suggesting that the coating prevented the oxygen infusion into the $Fe₃O₄$ particles. However, the very top layers of the surface may be different from the top 15 layers, the latter of which are probed by XPS, depending on the coating materials. Studies have indicated that the top two layers of the surface are rich in Fe(3+) compared to the top 15 layers in Fe₃O₄ films molecular-beam epitaxy (MBE) grown in ultrahigh vacuum.29,30 Polymer PMMA and PC contain oxygen whereas PS is oxygen free. For the PMMA- and PC-coated samples, it is possible to form bonding between the oxygen in polymer and $Fe(2+)$ ion on the surface, which will result in the presence of $Fe(3+)$ -rich oxide on the top one or two layers of $Fe₃O₄$ surface, which will greatly diminish the spin polarization as it does in pure $Fe₃O₄$ powders or polycrystalline film. Since PS contains no oxygen element, the $Fe₃O₄$ state can be preserved and the high-spin polarization can survive on the surface. This greatly enhances the spindependent tunneling MR.

IV. CONCLUSIONS

Drastic enhancement of the MR ratio suggests that there is a high degree of spin polarization at both low and room temperatures for half-metallic $Fe₃O₄$. The improvement was achieved by controlling the surface of $Fe₃O₄$ through surface engineering using oxygen-free insulating barriers. Based on our result, it is possible that a simple tunnel junction made of Fe₃O₄ exhibits large MR in a relatively small field. Fe₃O₄ may also be used as an effective spin injector. Knowing that our results provided only a lower limit on the spin polarization of $Fe₃O₄$, it has the potential to play an important role in spintronic devices.

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