Spectra and energy levels of $Gd^{3+}(4f^7)$ **in AlN**

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We report crystal-field splitting investigations of $Gd^{3+}(4f^7)$ energy levels determined from high resolution energy (Stark) level cathodoluminescence (CL) spectroscopy of Gd-implanted 2H-aluminum nitride (AlN) in the wavelength range 250 to 320 nm and at temperatures 12, 77, and 300 K. The CL spectra are interpreted in terms of lattice-sum and crystal-field splitting models that assume implanted Gd replaces Al substitutionally in cationic sites in AlN having C_{3v} point group symmetry. The CL spectra are associated with Gd^{3+} ions in a single site. A rms deviation of 7 cm^{-1} was obtained between 43 observed and calculated Stark levels. The intense UV emission at 318 nm (the ${}^6P_{7/2} \rightarrow {}^8S_{7/2}$ transition) can be understood in terms of multiplet-tomultiplet branching ratios.

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I. INTRODUCTION

Wide band gap semiconductors have considerable potential as components in optoelectronic devices, especially when implanted with trivalent rare earth (RE) ions.^{1–9} Optically characterized by intra- $4f$ electronic transitions, these ions emit electromagnetic radiation at discrete wavelengths ranging from the ultraviolet (UV) to the infrared $({\rm IR})$. $^{10-14}$ As phosphor illuminators, these materials play an important role in the design of white light emitters.^{15,16}

Because of its material and optical properties, 2Haluminum nitride¹⁷ (AlN) has been investigated as a host for light emitters implanted with Er^{3+} , Eu^{3+} , Tb^{3+} , and recently, Gd^{3+} and Tm^{3+} as a blue and UV light emitter.^{1-9,18} AlN has a high thermal conductivity and stability; it is chemically inert in most environments, and has a band gap of more than 6 eV (50000 cm^{-1}) .¹⁷ Such a wide band gap allows for UV emission from RE ion levels that are otherwise unavailable in *w*-GaN or 6H-SiC having band gaps less than 30 000 cm^{-1} .¹⁷ Even many insulator crystal hosts exhibit strong lattice absorption above $35\,000\ \text{cm}^{-1}$.¹¹

In addition to device applications, AlN as a host for RE ions allows for possible detailed investigations of the crystalfield splitting of high-energy $4f^n$ states, $4f^{n-1}5d$ states, and the determination of free-ion parameters of lanthanide configurations, $4f,$ ⁶⁻⁸ whose values are poorly known due to the lack of sufficient experimental energy $(Stark)$ level data.¹⁴ Also closer scrutiny is possible concerning the basic premise that forbidden electric-dipole transitions within the $4f^n$ configuration are due to mixing of states of opposite parity through the odd terms in the crystal-field expansion about the cationic site in the host lattice.^{12,19,20}

Recent crystal-field splitting analyses of lanthanide-doped

GaN^{21–23} based on luminescence data for Pr^{3+} , Sm³⁺, and Tb^{3+} suggest that similar studies can be performed successfully as well on RE ion-doped AlN, and in particular on Gd^{3+} and Tm^{3+} -doped AlN, where the size of the band gap permits direct observations of luminescence from the sextet states of $Gd^{3+}(4f^7)$ and the 3P_J , 1I_6 states of $\text{Tm}^{3+}(4 f^{12})$.^{9,24}

In the present study, the interpretation of the Gd^{3+} cathodoluminescence (CL) spectra that we report, and the Tm^{3+} (CL) spectra, which will appear in a subsequent presentation, are consistant with results obtained from latticesums calculations and crystal-field splitting modeling that assumes that the implanted Gd^{3+} and Tm^{3+} ions in AlN take substitutional Al lattice sites of C_{3v} symmetry. That there appears to be no change in the local cationic symmetry between implanted Gd^{3+} and Tm^{3+} ions is contrary to what has been found between Sm^{3+} , Tb^{3+} , and Er^{3+} in GaN, where the change from C_{3v} symmetry for Sm³⁺ in GaN to D_2 symmetry in Tb³⁺ and Er³⁺ in GaN is postulated as due to local structural changes based on differences in RE ionic radii.23

While the outcome of the crystal-field modeling is to establish consistency between calculated and experimental $2S+1$ *L_I* multiplet manifold splittings, an important input parameter for lattice-sum calculations is the lattice location of the RE ions, or, in other words, the position and symmetry of the ion in the unit cell. An attractive direct determination of the ion's lattice location is provided by the emission channeling (EC) technique,²⁵ which uses charged particles emitted in the decay of radioactive ions for lattice location studies. Nearly all lanthanides have a satisfactory decay chain that has made EC studies successful in determining lattice locations in semiconductors like, e.g., Si, InP, GaAs, SiC, GaN, diamond, *c*-BN and AlN.^{8,9,26–30} Recently, in carrying out lattice location studies for Yb^{3+} in AlN,⁹ we showed that RE ions implanted into AlN take substitutional Al lattice sites for annealing temperatures in the range 293 K to 1273 K. This position within the AlN unit cell has C_{3v} symmetry and represents the starting point for our lattice sum calculations. $31-33$ Included in the present investigations are the details of the observed crystal-field splitting of the energy levels of Gd^{3+} in AlN as obtained from an analysis of the observed cathodoluminescence spectra obtained at 12 K, 77 K, and 300 K. Intensities are reported in arbitrary units at different temperatures and show the relatively intense emission between the first excited manifold and the ground state $({}^{6}P_{7/2} \rightarrow {}^{8}S_{7/2})$, which makes Gd³⁺ the ion of choice for implementing quantum cutting devices.^{34,35} With modest adjustment to the crystal-field splitting parameters, B_{nm} , obtained from lattice-sum calculations, an rms deviation of 7 cm^{-1} is obtained between 43 calculated-to-observed Stark levels.

II. SAMPLE PREPARATION AND LUMINESCENCE MEASUREMENTS

Films of AlN, grown on substrates of $6H-SiC (0001)$ by metal-organic chemical vapor deposition (MOCVD), were obtained from commercial sources. Upon receipt, the substrate was cut into small pieces, rinsed in acetone and deionized water, and dried under a flow of nitrogen gas. Rutherford backscattering spectroscopy (RBS) revealed a film thickness of 200 nm . 158 Gd ions were implanted into these films at energies of 100 keV and fluences of 3×10^{13} ions/cm2. Both RBS measurements and implantation techniques are described elsewhere.³⁶ During implantation, samples were tilted 10° to the incident beam to avoid channeling by the incident ions. The mean implantation depth was calculated to be 33 ± 7 nm, assuming an AlN film density of 3.23 $g/cm^{3,37}$ Post-implantation annealing was carried out in a vacuum tube furnace at pressures near 10^{-6} mbar and at a temperature of about 1373 K for 30 min. Further details can be found in Ronning *et al.*³⁸

For CL fluorescence measurements, implanted samples were mounted on the head of a closed-cycle helium refrigerator located inside the vacuum chamber. An electrically controller resistive heater placed at the refrigerator head allows one to adjust the sample temperatures between 12 K and 300 K. The excitation source consisted of a SPECS EQ22 Auger electron gun that provides electrons with energies in the range between 100 eV to 5 keV and beam currents between 0.01 μ A and 150 μ A.

Sample luminescence was passed through a quartz window and collected with a UV-coated achromate lens pair, before reaching the entrance slit of a Czerny-Turnerspectrograph, model Jobin-Yvon 1000M. The light was dispersed using a holographic grating, blazed at 300 nm, with 1200 lines/mm, and detected by a nitrogen-cooled CCD camera, model Jobin-Yvon, UV-enhanced CCD (EEV CCD30-11). Exposure times for the spectra ranged from 1 s to 180 s. Unavoidable cosmic spikes that occur during long exposure times were identified by measuring each of the spectra at

FIG. 1. CL spectra of the ${}^6D_{9/2} \rightarrow {}^8S_{7/2}$ transition of Gd³⁺ in AlN, recorded between 254 and 264 nm at 12, 77, and 300 K.

least twice and were manually removed from the experimental data. The spectrograph was repeatedly calibrated using the air wavelengths of spectral lines from a mercury lamp standard.³⁹ The uncertainty in wavelength measurements was about 0.02 nm. Temperature-dependent measurements were made on transitions at a given wavelength without moving the grating or otherwise disturbing the experimental setup in order to minimize inaccuracies in the measurements.

III. OBSERVED LUMINESCENCE SPECTRA

Figures 1–4 present the CL spectra of the 6D_J , 6I_J , and ${}^{6}P_J$ states of Gd³⁺(4 f^7) observed between 250 nm and 320 nm at temperatures of 12, 77, and 300 K, and under excitation conditions of 5 keV and 0.25 W/cm². With an increase in temperature, higher-energy Stark levels within a given multiplet manifold are populated so that at 300 K all expected $J+1/2$ Stark levels are observed for each Kramers manifold (see, for example, Fig. 1, the ${}^6D_{9/2}$ and Fig. 4, the ${}^6P_{5/2}$ and ${}^6P_{7/2}$ manifolds). The ground-state manifold, ${}^{8}S_{7/2}$, has such a small splitting (predicted to be less than 0.3) cm^{-1}) that its splitting is not observed in the present data.¹⁰

As we pointed out earlier in the analyses of the lumines-

FIG. 2. CL spectra of the ${}^{6}I_{7/2,9/2,17/2,11/2,15/2,13/2}\rightarrow {}^{8}S_{7/2}$ transitions of Gd^{3+} in AlN, recorded between 277 and 287 nm at 12, 77, and 300 K.

FIG. 3. CL spectra of the ⁶ $P_{7/2,5/2,3/2}$ ³ $S_{7/2}$ transitions and phonon sidebands of Gd^{3+} in AlN, recorded between 304 and 321 nm at 12, 77, and 300 K.

cence spectra of Pr^{3+} , Sm^{3+} , and Tb^{3+} in GaN,^{21–23} analyzing unpolarized CL spectra for transitions between individual Stark levels is primarily limited to matching repeated energy-level differences, similar to the technique used in analyzing the arc and spark spectra of ions and molecules for their energy-level configurations.⁴⁰ We are aided in the present study, however, by the number of $2S+1L_J$ multiplet manifolds that are separated from one another by sufficient energy so that crystal-field mixing is minimal, and by the fact that the manifold splitting of the ${}^{8}S_{7/2}$ is so small that assignments for each level can be made with relative ease and are listed in Table I. For example, the splitting of the ${}^6D_{9/2}$ manifold (Fig. 1) is represented by transitions 1 through 5, and the splitting of the ${}^6P_{7/2}$ manifold (Fig. 4) is represented by transitions 41 through 44 in Table I, column 2. Both the size of the manifold splitting and the values of the energy levels $\text{ (cm}^{-1}\text{)}$ can be compared to similar data for Gd^{3+} in fluoride, chloride, and oxide insulator host crystals with interesting results that support the nephelauxetic effect observed in rare earth systems.10–13,41

IV. MODELING THE ENERGY LEVELS

A point-charge lattice-sum model described earlier^{21–23} was adapted to analyze the crystal-field splitting of the $2S+1$ _{LJ} energy levels of Gd³⁺(4 $f⁷$) in AlN. From lattice location studies described by Vetter *et al.*⁹ we conclude that implanted Gd^{3+} in AlN take substitutional Al sites in the lattice that have C_{3v} point group symmetry in a hexagonal structure with space group $C_{6v}^4 - P_{63}$. The cell parameters are $a = b = 3.11200 \text{ Å}$, $c = 4.99820 \text{ Å}$, with $\alpha = \beta = 90^{\circ}$, and $\gamma=120^\circ$. The lattice constants are similar to those for hexagonal GaN. The *c*/*a* ratio is 1.601 for AlN and 1.625 for GaN.42,43 Based on ionic radii and volume considerations, Gd^{3+} should be accommodated in the Al sites.^{42–45} The fact that $J + 1/2$ transitions are identified in the spectra to each of the well isolated multiplet manifolds further indicates that Gd^{3+} ions are found in a common site (see Figs. 1 and 4, for example). In AlN the axis of highest symmetry (C_3) is chosen as the z axis (axis of quantization) and corresponds to the c axis in the unit cell.⁴²

FIG. 4. CL spectra of the ${}^6P_{7/2,5/2} \rightarrow {}^8S_{7/2}$ transitions of Gd³⁺ in AlN, recorded between 311 and 321 nm at 12, 77, and 300 K.

The lattice-sum components A_{nm} , expressed in units of $\text{cm}^{-1}/\text{\AA}^n$, are given as

$$
A_{nm} = -e^2 \sum_{j} q_j \frac{C_{nm}(\hat{r}_j)}{r_j^{n+1}},
$$
 (1)

where q_i is the effective charge on the ion, and the sum is taken over all the ions in the lattice. By varying the formal charge of 3 on Al and N between 2.9 and 3.0 in our latticesum calculations, we find the assumption of isovalent impurity (Gd^{3+}) traps to be reasonable and consistent.^{21,42–45}

The crystal-field splitting parameters B_{nm} , are related to the lattice-sum components by the expression $B_{nm} = \rho_n A_{nm}$ where the ρ_n for Gd³⁺ are ρ_2 =0.1668, ρ_4 =0.4656, and ρ_6 $=1.1873^{46}$ The B_{nm} are in units of cm⁻¹ and the units of ρ_n are $(\mathring{A}^n)^{-1}$. The B_{nm} parameters appear in the crystal-field Hamiltonian

$$
H_{CF} = \sum_{n \ even} \sum_{m = -n}^{n} B_{nm}^{*} \sum_{i=1}^{N} C_{nm}(\hat{r}_i)
$$
 (2)

with the normalized one-electron tensor operators, $C_{nm}(\hat{r}_i)$, defined by¹⁹

$$
C_{nm}(\hat{r}_i) = \sqrt{\frac{4\pi}{2n+1}} Y_{nm}(\theta_i, \phi_i)
$$
 (3)

and

$$
C_{n,-m}(\hat{r}_i) = (-1)^m C_{nm}^*(r_i). \tag{4}
$$

In Eq. (3), the Y_{nm} are spherical harmonics, and $n=2, 4$ and 6 with $m=0, \pm 3$, and ± 6 , for $|m| \le n$. The sum on *i* in Eq. (2) runs over the seven $4f$ electrons (the half-filled shell) of Gd^{3+} .

The crystal field Hamiltonian given in Eq. (2) is diagonalized together with a parametrized free-ion Hamiltonian that includes coulombic, spin-orbit and interconfiguration interaction terms,

Level ^b		λ^d	I^e				
$2S+1$	No. ^c	(nm)	(arb. units)	$E_{\mathit{obs}}^{\quad \ f}$ $\rm (cm^{-1})$	E_{calc} g $\rm (cm^{-1})$	$\mu^{\rm h}$	Percent free-ion mixture
$^6P_{7/2}$	44	318.15	116.38\53.19\75.53	31423	31418	±1/2	99.7 ${}^6P_{7/2}$ + 0.07 ${}^6I_{13/2}$ + 0.07 ${}^6I_{15/2}$
(31449)	43	318.07	20.58\26.68\39.67	31431	31422	$\pm\,1/2$	99.7 ${}^6P_{7/2}$ + 0.08 ${}^6P_{5/2}$ + 0.07 ${}^6I_{17/2}$
	42	317.93	10.51\53.27\97.40	31444	31449	$\pm 3/2$	99.8 ${}^6P_{7/2}$ + 0.05 ${}^6I_{17/2}$ + 0.03 ${}^6I_{11/2}$
	41	317.72	3.06\41.09\92.63	31465	31468	±1/2	99.8 ${}^6P_{7/2}$ + 0.06 ${}^6I_{15/2}$ + 0.03 ${}^6I_{13/2}$
$^6P_{\,5/2}$	40	312.23	$-\frac{6.99}{ }$	32018	32020	$\pm 3/2$	99.6 ${}^6P_{5/2}$ + 0.14 ${}^6I_{17/2}$ + 0.13 ${}^6I_{7/2}$
(32043)	39	312.13	$-\1.58$	32029	32025	±1/2	99.7 ${}^6P_{5/2}$ + 0.09 ${}^6P_{7/2}$ + 0.08 ${}^6I_{17/2}$
	38	311.91	$-\frac{5.3}{3}$	32051	32050	±1/2	99.6 ${}^6P_{5/2}$ + 0.17 ${}^6I_{17/2}$ + 0.15 ${}^6I_{7/2}$
$^6P_{\,3/2}$	37	306.73	$-\117.52$	32593	32585	$\pm\,1/2$	99.6 ${}^6P_{3/2}$ + 0.14 ${}^6I_{15/2}$ + 0.09 ${}^6I_{9/2}$
(32602)	36	306.62	$-\ -4.77$	32604	32596	$\pm 3/2$	99.5 ${}^6P_{3/2}$ + 0.16 ${}^6I_{13/2}$ + 0.14 ${}^6I_{11/2}$
	35	306.47	$-\13.4$	32620			
$^6I_{7/2}$	34	285.34	4.98\2.47\1.49	35036	35040	±1/2	99.8 $^{6}I_{7/2}$ + 0.14 $^{6}I_{9/2}$ + 0.03 $^{6}P_{5/2}$
(35076)	33	285.02	$-1.191.35$	35075	35076	$\pm\,1/2$	99.6 ${}^{6}I_{7/2}$ + 0.27 ${}^{6}I_{9/2}$ + 0.05 ${}^{6}P_{5/2}$
	32	284.94	$-2.182.71$	35085	35079	$\pm 3/2$	99.7 $^{6}I_{7/2}$ + 0.13 $^{6}I_{9/2}$ + 0.12 $^{6}P_{5/2}$
	31	284.74	$-\0.9 - 1.66$	35109	35118	$\pm\,1/2$	99.8 ${}^{6}I_{7/2}$ + 0.11 ${}^{6}P_{5/2}$ + 0.02 ${}^{6}I_{9/2}$
$^6I_{9/2}$	30	282.44		35395	35380	$\pm\,1/2$	99.3 $^{6}I_{9/2}$ + 0.28 $^{6}I_{17/2}$ + 0.24 $^{6}I_{7/2}$
(35420)	29	282.33		35409	35400	$\pm\,1/2$	98.7 ${}^6I_{9/2}$ + 0.78 ${}^6I_{17/2}$ + 0.25 ${}^6I_{11/2}$
	28	282.22		35423	35414	$\pm 3/2$	98.8 $^{6}I_{9/2}$ + 0.53 $^{6}I_{17/2}$ + 0.46 $^{6}I_{17/2}$
	27	282.09		35439	35453	$\pm 3/2$	98.3 ${}^{6}I_{9/2}$ + 1.15 ${}^{6}I_{17/2}$ + 0.20 ${}^{6}I_{11/2}$
	26	282.06		35443	35458	$\pm\,1/2$	96.7 ${}^6I_{9/2}$ + 2.86 ${}^6I_{17/2}$ + 0.20 ${}^6I_{11/2}$
$^6I_{17/2}$	25	281.92		35461	35475	$\pm\,1/2$	99.5 $^{6}I_{17/2}$ + 0.22 $^{6}I_{13/2}$ + 0.14 $^{6}I_{15/2}$
(35479)					35476	$\pm 3/2$	98.4 $^{6}I_{17/2}$ + 0.75 $^{6}I_{9/2}$ + 0.64 $^{6}I_{15/2}$
	24	281.80		35476	35477	$\pm\,1/2$	98.5 $^{6}I_{17/2}$ + 0.66 $^{6}I_{9/2}$ + 0.43 $^{6}I_{15/2}$
					35478	$\pm\,1/2$	98.2 $^{6}I_{17/2}$ + 0.28 $^{6}I_{9/2}$ + 0.20 $^{6}I_{11/2}$
					35480	$\pm 3/2$	98.8 $^{6}I_{17/2}$ + 0.69 $^{6}I_{9/2}$ + 0.33 $^{6}I_{11/2}$
	23	281.72		35486	35483	$\pm\,1/2$	98.5 $^6I_{17/2}$ + 0.20 $^6I_{13/2}$ + 0.13 $^6I_{9/2}$
	22	281.57		35505	35495	$\pm\,1/2$	98.4 $^{6}I_{17/2}$ + 1.03 $^{6}I_{9/2}$ + 0.21 $^{6}I_{15/2}$
	21	281.25		35535	35540	$\pm 3/2$	99.1 $^{6}I_{17/2}$ + 0.22 $^{6}I_{15/2}$ + 0.18 $^{6}I_{11/2}$
	20	281.21		35550	35550	$\pm\,1/2$	97.5 $^{6}I_{17/2}$ + 1.79 $^{6}I_{9/2}$ + 0.26 $^{6}I_{13/2}$
$^6I_{11/2}$	19	280.24		35673	35670	$\pm 3/2$	98.1 $^6I_{11/2}$ + 1.00 $^6I_{15/2}$ + 0.55 $^6I_{9/2}$
(35711)					35675	$\pm\,1/2$	98.8 $^{6}I_{11/2}$ + 0.56 $^{6}I_{15/2}$ + 0.18 $^{6}I_{17/2}$
	18	280.09		35692	35694	$\pm\,1/2$	98.9 $^{6}I_{11/2}$ + 0.64 $^{6}I_{13/2}$ + 0.15 $^{6}I_{15/2}$
					35738	$\pm\,1/2$	95.6 $^{6}I_{11/2}$ + 2.61 $^{6}I_{15/2}$ + 1.32 $^{6}I_{13/2}$
	17	279.75		35736	35739	$\pm 3/2$	97.8 $^{6}I_{11/2}$ + 1.08 $^{6}I_{15/2}$ + 0.40 $^{6}I_{13/2}$
		279.6		35754	35750	$\pm\,1/2$	97.7 $^6I_{11/2}$ + 1.70 $^6I_{15/2}$ + 0.28 $^6I_{17/2}$
$^6I_{15/2}$	16	279.14		35814	35814	$\pm\,1/2$	96.1 $^6I_{15/2}+3.40\ ^6I_{11/2}+0.28\ ^6I_{17/2}$
(35850)					35819	$\pm 1/2$	98.4 $^{6}I_{15/2}$ + 0.82 $^{6}I_{13/2}$ + 0.43 $^{6}I_{17/2}$
$^6I_{13/2}$					35820	$\pm 3/2$	80.2 $^6I_{15/2}$ + 18.4 $^6I_{13/2}$ + 0.63 $^6I_{11/2}$
(35869)	15	279.02		35829	35827	$\pm 3/2$	80.3 $^{6}I_{15/2}$ + 18.4 $^{6}I_{13/2}$ + 0.63 $^{6}I_{11/2}$
	14	279		35832	35831	$\pm\,1/2$	95.3 $^{6}I_{13/2}$ + 3.16 $^{6}I_{15/2}$ + 1.25 $^{6}I_{17/2}$
	13	278.96		35837	35841	$\pm\,1/2$	84.3 $^6I_{13/2}$ + 15.3 $^6I_{15/2}$ + 0.14 $^6I_{11/2}$
	12	278.84		35852	35852	$\pm\,1/2$	85.2 $^6I_{15/2}$ + 14.3 $^6I_{13/2}$ + 0.26 $^6I_{11/2}$
	$11\,$	278.78		35860	35865	$\pm\,1/2$	90.8 $^{6}I_{15/2}$ + 8.01 $^{6}I_{13/2}$ + 0.73 $^{6}I_{11/2}$
	10	278.65		35877	35877	$\pm 3/2$	97.5 $^{6}I_{15/2}$ + 1.35 $^{6}I_{13/2}$ + 0.73 $^{6}I_{11/2}$
					35881	$\pm\,1/2$	83.3 $^6I_{13/2}$ + 16.1 $^6I_{15/2}$ + 0.40 $^6I_{11/2}$
	9	278.57		35887	35887	±3/2	98.5 $^{6}I_{15/2}$ + 0.64 $^{6}I_{13/2}$ + 0.56 $^{6}I_{11/2}$
					35890	$\pm\,1/2$	85.6 $^6I_{15/2}$ + 13.3 $^6I_{13/2}$ + 0.82 $^6I_{11/2}$
	8	278.49		35897	35902	$\pm 3/2$	97.9 $^{6}I_{13/2}$ + 1.35 $^{6}I_{15/2}$ + 0.36 $^{6}I_{11/2}$
	7	278.39		35910	35904	$\pm\,1/2$	97.8 $^6I_{13/2}$ + 1.74 $^6I_{15/2}$ + 0.22 $^6I_{17/2}$
	6	278.34		35917	35912	$\pm\,1/2$	98.9 $^{6}I_{13/2}$ + 0.71 $^{6}I_{15/2}$ + 0.16 $^{6}I_{17/2}$
$^6D_{\,9/2}$	5	259.27	$-\$ 0.11	38558	38573	$\pm 3/2$	99.8 ${}^6D_{9/2}$ + 0.12 ${}^6D_{7/2}$ + 0.14 ${}^6I_{15/2}$
(38671)	4	258.96	$-/-0.09$	38604	38600	$\pm\,1/2$	99.7 ${}^6D_{9/2}$ + 0.14 ${}^6D_{7/2}$ + 0.04 ${}^6I_{15/2}$
	3	258.32	$-\ -0.12$	38700	38692	$\pm 3/2$	99.8 ${}^6D_{9/2}$ + 0.04 ${}^6D_{7/2}$ + 0.03 ${}^6I_{11/2}$

TABLE I. Energy levels of Gd^{3+} $(4f^7)$ in AlN.^a

Level ^b $2S+1L_1$	No. ^c	λ^d (nm)	r^e (arb. units)	E_{obs}^{-1} $\rm (cm^{-1})$	E_{calc} ^g (cm^{-1})	$\mu^{\rm h}$	Percent free-ion mixture
	2	258.09	$-\angle 0.09$	38735	38729	$\pm 1/2$	99.8 ${}^6D_{9/2}$ + 0.05 ${}^6D_{7/2}$ + 0.04 ${}^6I_{15/2}$
		257.90	$0.19\backslash0.13\backslash0.12$	38763	38762	$\pm 1/2$	99.7 ${}^6D_{9/2}$ + 0.10 ${}^6D_{7/2}$ + 0.07 ${}^6D_{1/2}$
$^6D_{1/2}$ (39621)					39603	$\pm 1/2$	83.9 ${}^6D_{1/2}$ + 15.7 ${}^6D_{7/2}$ + 0.25 ${}^6D_{5/2}$
${}^6D_{7/2}$					39679	$\pm 3/2$	83.7 ${}^6D_{7/2}$ + 13.1 ${}^6D_{3/2}$ + 2.97 ${}^6D_{5/2}$
(39713)					39693	$\pm 1/2$	87.8 ${}^6D_{7/2}$ + 11.0 ${}^6D_{3/2}$ + 0.94 ${}^6D_{5/2}$
					39721	$\pm 1/2$	95.4 ${}^6D_{7/2}$ + 2.6 ${}^6D_{1/2}$ + 1.18 ${}^6D_{5/2}$
					39727	$\pm 1/2$	85.2 ${}^6D_{7/2}$ + 12.8 ${}^6D_{1/2}$ + 1.14 ${}^6D_{3/2}$
${}^6D_{3/2}$					39847	±3/2	71.8 ${}^6D_{3/2}$ + 19.1 ${}^6D_{5/2}$ + 9.06 ${}^6D_{7/2}$
(39851)					39871	$\pm 1/2$	74.5 ${}^6D_{3/2}$ + 15.6 ${}^6D_{5/2}$ + 9.63 ${}^6D_{7/2}$
${}^6D_{5/2}$					39952	±3/2	77.8 ${}^6D_{5/2}$ + 15.1 ${}^6D_{3/2}$ + 7.01 ${}^6D_{7/2}$
(39978)					39984	$\pm 1/2$	83.4 ${}^6D_{5/2}$ + 12.4 ${}^6D_{3/2}$ + 3.96 ${}^6D_{7/2}$
					40062	$\pm 1/2$	97.7 ${}^6D_{5/2}$ + 1.96 ${}^6D_{7/2}$ + 0.21 ${}^6D_{3/2}$

TABLE I. (Continued.)

a Spectra obtained at 12, 77, and 300 K.

 $b^{2\bar{S}+1}L$ _{*i*} multiplet manifolds for Gd³⁺ (4f⁷); the number in parenthesis indicates the calculated centroid for each manifold.

Number of transition as assigned in the figures.

d Experimental wavelength in nanometers.

^eCL intensity in arbitrary units at different temperatures [*I*(12 K)*I*(77 K)*I*(300 K)]. Intensities of the ⁶*P*_{7/2}→8*S*_{7/2} and ⁶*P*_{5/2}→8*S*_{7/2} transitions have to be multiplied by \sim 50.

f Energy in vacuum wave numbers.

^gCalculated splitting based on Gd³⁺ in C_{3v} sites: $B_{20} = -211$, $B_{40} = -1691$, $B_{43} = -1063$, $B_{60} = 655.5$, $B_{63} = -188.4$, and $B_{66} = 293.7$; rms 7 cm^{-1} , between 43 observed and calculated Stark levels.

^hSymmetry label $\mu = \pm 1/2$, $\mu = \pm 3/2$, calculated using the *B_{nm}* parameters in footnote g.

$$
H_{FI} = H_0 + E_{AVE} + \sum_{k=2,4,6} f_k F^k + \zeta \sum_{i=1}^7 \mathbf{s}_i \cdot \mathbf{l}_i + \alpha L^2 + \beta G(G_2)
$$

+ $\gamma G(R_7) + \sum_{i=2,3,4,6,7,8} t_i T^i + \sum_{k=0,2,4} m_k M^k$

$$
+\sum_{k=2,4,6} p_k P^k,\tag{5}
$$

for the $4f^7$ electronic configuration, where the individual terms and definitions are detailed by Görller-Walrand and Binnemans,47 and a starting set of free-ion parameters is given by Carnall *et al.*⁴⁸ for Gd^{3+} . These parameters are used to calculate the reduced matrix elements of U^2 , U^4 , and *U*⁶ between all intermediate-coupled wavefunctions for the electronic configuration. A separate program takes the reduced matrix elements between the free-ion multiplets, sets up the crystal spaces for a given symmetry, and diagonalizes in that space of multiplets the crystal-field splitting Hamiltonian given in Eq. (2). The number of $2S+1$ *L_J* free-ion eigenvectors is 327 for $4f^7$ with more than 2000 doublydegenerate (Stark) energy eigenvalues spanning 180 000 cm^{-1} . A complete diagonalization of the energy matrix consisting of crystal quantum states $\mu = \pm 1/2$ and $\mu = \pm 3/2$ is not particularly useful given that only 19 $2S+1$ *L_J* multiplets lie within the band gap, of which we have data for all but the ${}^{6}G_J$ multiplets. Instead, we diagonalize the complete Hamiltonian for the lowest-energy 25 multiplet manifolds, following the approach we used earlier to reduce truncation errors to a minimum.49 Truncation errors are introduced when the *J*-mixing is large. In Table I the mixing is relatively small and suggests that the approximation made by truncation is reasonable for the levels observed in the CL spectra. In the following fitting scheme, we vary only the six B_{nm} , and the centroid for each multiplet in order to account for the relatively small crystal-field mixing between states. Fitting of the free ion parameters is not possible in our case, because for the observed free ion energy levels corresponding matrix elements are often too small, resulting in large uncertainties. Typical examples are the Slater parameters F^k . The starting set of B_{nm} were obtained from lattice-sum calculations for GaN. With modest adjustment to the initial B_{nm} parameters and centroids, we obtain final agreement between 43 experimental Stark levels and calculated levels with a rms deviation of 7 cm^{-1} . Based on the analysis of the observed spectra, the splitting of other multiplets, such as the ${}^6D_{1/2,7/2,3/2,5/2}$ are predicted and reported in Table I. The ${}^{6}G_J$ multiplets are predicted close to the band gap. Their multiplet splitting may be of interest for energy transfer between the conduction band and the ion, but we lack sufficient data at this time to determine the energy (Stark) levels for these multiplets.

V. CATHODOLUMINESCENCE INTENSITIES

From Figs. 1–4 we obtain qualitative information concerning the emission characteristics that were first reported

FIG. 5. Squared reduced matrix elements in units cm^{-1} for transitions between ground state and first 18 excited $2S+1L_J$ multiplet manifolds of Gd³⁺ (notation ^{2*S*+1}*L*_{2*J*}).

by Vetter *et al.*²⁴ for Gd³⁺:AlN, particularly the intense emission identified from the four Stark levels of the ${}^6P_{7/2}$ to the ground state ${}^{8}S_{7/2}$ observed between 317.72 nm and 318.15 nm (transitions $41-45$, Table I). There is a factor of 50 times greater intensity for these transitions as compared with all other transitions observed.

We can make a rough approximation of expected transition probabilities for electric- and magnetic-dipole transitions to the ground state using the parametrization methods developed by Axe,⁵⁰ and Krupke and Gruber,⁵¹ which later evolved into the Judd-Ofelt approximation methods. The radiative transition probabilities are of the form 12,52

$$
W = \frac{64\pi^4 e^2 \langle \nu \rangle^3 n (n^2 + 2)^2}{27h[J]}
$$

$$
\times \sum_{\lambda = 2,4,6} \Omega_{\lambda} |\langle^{2S+1} L_J || U^{(\lambda)} ||^8 S_{7/2} \rangle|^2
$$

$$
\equiv \sum_{\lambda = 2,4,6} A^{\prime(\lambda)} \Omega_{\lambda}, \qquad (6)
$$

where n is the index of refraction at the wavelength of the transition, assuming an isotropic medium, $\langle v \rangle$ is the energy of the radiation in cm⁻¹, $\langle (||U^{(\lambda)}||) |^{2}$ are the squared reduced matrix elements, and Ω_2 , Ω_4 , and Ω_6 are the intensity parameters. For comparison, values of Ω_2 , Ω_4 , and Ω_6 are reported by Sytsma *et al.*⁵³ based on analyses of emission data from Gd^{3+} in other hosts.⁴¹ The values of the squared reduced matrix elements for this study are given in Fig. 5 for manifold transitions between the ground state ${}^{8}S_{7/2}$, and excited 6P_J , 6I_J , 6D_J , and 6G_J states.

In addition to radiative relaxations between Gd^{3+} ion states, we need to consider nonradiative deexcitation due to multiphonon relaxation as well. The relatively small manifold splittings, and the closely spaced multiplets for many of the states reported in Table I, suggest that multiphonon relaxations will reduce the actual lifetime and observed emission from most states, with the possible exception of ${}^6D_{9/2}$, ${}^6I_{7/2}$, and ${}^6P_{7/2}$, the latter which is separated from the ground state by more than $31,000 \text{ cm}^{-1}$.

Since all emission from the ${}^6P_{7/2}$ goes directly to the ground state ${}^{8}S_{7/2}$, it is worthwhile to explore branching ratios from states higher in energy than ${}^6P_{7/2}$ to see if we can explain the intense emission observed at 318 nm reported by Vetter *et al.*²⁴ Since both even and odd crystal-field components *Anm* are obtained from lattice-sum calculations, and Stark levels and corresponding wave functions have been identified for Gd^{3+} in AlN, we can model branching ratios following an approach outlined in our analysis of the CL spectra of Sm^{3+} in GaN.²²

Line-to-line branching ratios are computed as a function of temperature by including a partition function representing the population in individual Stark levels as

$$
\beta_{ij} = \frac{z_i W_{ij}}{\sum_{ij} z_i W_{ij}},\tag{7}
$$

where

$$
z_i = \frac{\exp(E_i/k_B T)}{\sum_i \exp(E_i/k_B T)},
$$
\n(8)

and *i* represents the emitting Stark level. Multiplet branching ratios corresponding to the Judd-Ofelt method $⁵³$ are obtained</sup> by summing contributions from line-to-line transitions, each weighted equally $(z_i=1.0$ for all *i*). By calculating multiplet branching ratios from ${}^6G_{7/2} \rightarrow {}^6D_J$, 6I_J , 6P_J , ${}^8S_{7/2}$; ${}^{\bar{6}}D_J$ \rightarrow ⁶*I*_{*J*}, ⁶*P*_{*J*}, ⁸*S*_{7/2}</sub>; ⁶*I*_{*J*} \rightarrow ⁶*P*_{*J*}, ⁸*S*_{7/2}, we find substantial probability of decay to the ${}^{6}P_J$ states. Rapid relaxation among 6P_J states leads to the prediction that the ${}^6P_{7/2}$ \rightarrow ⁸S_{7/2} transition at 318 nm is the most likely emission by several orders of magnitude. Based on analyses of the fluorescence spectra of Gd^{3+} in other hosts and our modeling studies, we conclude that the ${}^6P_{7/2} \rightarrow {}^8S_{7/2}$ transition observed at 318 nm clearly dominates other radiative transitions possible between excited states not involving the ground state.

VI. SUMMARY

In summary, we report a detailed crystal-field splitting analysis of Gd^{3+} implanted in AlN. Lattice-sum calculations, and modeling of the crystal-field splitting supports our recent work indicating that the Gd^{3+} ions substantially replace Al in lattice sites having C_{3v} point group symmetry in hexagonal AlN. All the CL spectra are associated with Gd^{3+} in a single site. Modest adjustments to six crystal field parameters B_{nm} , and to the centroid for each multiplet manifold, $2S+1I$ look to the section of the section o ${}^{1}L_{I}$, leads to a rms deviation of 7 cm⁻¹ for 43 observedto-calculated Stark levels. The intense UV emission at 318 nm (the ${}^6P_{7/2} \rightarrow {}^8S_{7/2}$ transition) can be understood in terms of multiplet-to-multiplet branching ratios based on a Judd-Ofelt analysis of the observed spectra.

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