Dynamic structure factor of liquid and amorphous Ge from *ab initio* simulations

Jeng-Da Chai* and D. Stroud[†]

Department of Physics, The Ohio State University, Columbus, Ohio 43210

J. Hafner and G. Kresse

Institut für Materialphysik and Center for Computational Materials Science, Technische Universität Wien, Sensengasse 8/12,

A-1090 Wien, Austria

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We calculate the dynamic structure factor $S(k,\omega)$ of liquid Ge (*l*-Ge) at temperature T=1250 K, and of amorphous Ge (*a*-Ge) at T=300 K, using *ab initio* molecular dynamics. The electronic energy is computed using density-functional theory, primarily in the generalized gradient approximation, together with a planewave representation of the wave functions and ultrasoft pseudopotentials. We use a 64-atom cell with periodic boundary conditions, and calculate averages over runs of up to about 16 ps. The calculated liquid $S(k,\omega)$ agrees qualitatively with that obtained by Hosokawa *et al.* [Phys. Rev. B **63**, 134205 (2001)] using inelastic x-ray scattering. In *a*-Ge, we find that the calculated $S(k,\omega)$ is in qualitative agreement with that obtained experimentally by Maley *et al.* [Phys. Rev. Lett. **56**, 1720 (1986)]. Our results suggest that the *ab initio* approach is sufficient to allow approximate calculations of $S(k,\omega)$ in both liquid and amorphous materials.

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I. INTRODUCTION

Ge is a well-known semiconductor in its solid phase, but becomes metallic in its liquid phase. Liquid Ge (l-Ge) has, near its melting point, an electrical conductivity characteristic of a reasonably good metal (~ $1.6 \times 10^{-4} \ \Omega^{-1} \ cm^{-11}$), but it retains some residual structural features of the solid semiconductor.¹ For example, the static structure factor S(k)has a shoulder on the high-k side of its first (principal) peak, which is believed to be due to residual tetrahedral shortrange order. This shoulder is absent in more conventional liquid metals such as Na or Al, which have more of a closepacked structure in the liquid state and a shoulderless first peak in the structure factor. Similarly, the bond-angle distribution function just above melting is believed to have peaks at two angles, one near 60° and characteristic of close packing, and one near 108°, indicative of tetrahedral short-range order. This latter peak rapidly disappears with increasing temperature in the liquid state.

These striking properties of *l*-Ge have been studied theoretically by several groups. Their methods fall into two broad classes: empirical and first principles. A typical empirical calculation is that of Yu *et al.*,² who calculate the structural properties of *l*-Ge assuming that the interatomic potentials in *l*-Ge are a sum of two-body and three-body potentials of the form proposed by Stillinger and Weber.³ These authors find, in agreement with experiment, that there is a high-*k* shoulder on the first peak of *S*(*k*) just above melting, which fades away with increasing temperature. However, since in this model *all* the potential energy is described by a sum of two-body and three-body interactions, the interatomic forces are probably stronger and the ionic diffusion coefficient is correspondingly smaller than their actual values.

In the second approach, the electronic degrees of freedom are taken explicitly into account. If the electron-ion interaction is sufficiently weak, it can be treated by linear-response theory.⁴ In linear response, the total energy in a given ionic

configuration is a term which is independent of the ionic arrangement, plus a sum of two-body ion-ion effective interactions. These interactions typically do not give the bondangle-dependent forces which are present in the experiments, unless the calculations are carried to third order in the electron-ion pseudopotential,⁴ or unless electronic fluctuation forces are included.⁵ Such interactions are, however, included in the so-called ab initio approach, in which the forces on the ions are calculated from first principles, using the Hellman-Feynman theorem together with densityfunctional theory⁶ to treat the energy of the inhomogeneous electron gas. This approach not only correctly gives the bond-angle-dependent ion-ion interactions, but also, when combined with standard molecular-dynamics techniques, provides a good account of the electronic properties and such dynamical ionic properties as the ionic self-diffusion coefficients.

This combined approach, usually known as *ab initio* molecular dynamics, was pioneered by Car and Parrinello,⁷ and, in somewhat different form, has been applied to a wide range of liquid metals and alloys, including *l*-Ge,^{8–10} *l*-Ga_xGe_{1-x},¹¹ stoichiometric III-V materials such as *l*-GaAs, *l*-GaP, and *l*-InP,^{12,13} and nonstoichiometric *l*-Ga_xAs_{1-x},¹⁴ *l*-CdTe,¹⁵ and *l*-ZnTe,¹⁶ among other materials which are semiconducting in their solid phases. It has been employed to calculate a wide range of properties of these materials, including the static structure factor, bondangle distribution function, single-particle electronic density of states, dc and ac electrical conductivity, and the ionic selfdiffusion coefficient. The calculations generally agree quite well with data from available experiments.

A similar *ab initio* approach has also been applied extensively to a variety of amorphous semiconductors, usually obtained by quenching an equilibrated liquid state from the melt. For example, Car, Parrinello, and their collaborators have used their own *ab initio* approach (based on treating the Fourier components of the electronic wave functions as fic-

titious classical variables) to obtain many structural and electronic properties of amorphous Si.^{17,18} A similar approach has been used by Lee and Chang.¹⁹ Kresse and Hafner⁸ obtained both S(k) and g(r), as well as many electronic properties, of a-Ge, using an ab initio approach similar to the one used here, in which the forces are obtained directly from the Hellmann-Feynman theorem and no use is made of fictitious dynamical variables for the electrons, as in the Car-Parrinello approach. A similar calculation for *a*-Si has been carried out by Cooper *et al.*, 20 also making use of a plane-wave basis and treating the electron-density functional in the generalized gradient approximation and other amorphous semiconductors have been carried out by (GGA).²¹ More recently, a number of calculations for *a*-Si Sankey and Niklewsky,²² and by Drabold and collaborators.²³ These calculations use *ab initio* molecular dynamics and electronic density-functional theory, but in a localized basis. A recent study, in which S(k) and g(r) were computed for several *ab initio* structural models of a-Si, has been carried out by Alvarez et al.²⁴

Finally, we mention a third approach, intermediate between empirical and *ab initio* molecular dynamics, generally known as tight-binding molecular dynamics. In this approach, the electronic part of the total energy is described using a general tight-binding Hamiltonian for the band electrons. The hopping matrix elements depend on separation between the ions, and additional terms are included to account for the various Coulomb energies in a consistent way. The parameters can be fitted to *ab initio* calculations, and forces on the ions can be derived from the separation dependence of the hopping matrix elements. This approach has been used, e.g., to treat *l*-Si,²⁵ *a*-Si,²⁶ and liquid compound semiconductors such as *l*-GaAs and *l*-GaSb.²⁷ Results are in quite good agreement with experiment.

In this paper, we extend the method of *ab initio* molecular dynamics (MD) to another dynamical property of the ions: the dynamical structure factor, denoted $S(k,\omega)$. While no fundamentally new theory is required to calculate $S(k,\omega)$, this quantity provides additional information about the timedependent ionic response beyond what can be extracted from other quantities. The present work appears to be the first to calculate $S(k,\omega)$ using *ab initio* molecular dynamics. Here, we will calculate $S(k,\omega)$ for *l*-Ge, where some recent experiments²⁸ provide data for comparison, and also for amorphous Ge (a-Ge). In the latter case, using a series of approximations described below, we are able to infer the vibrational density of states of as-quenched a-Ge near temperature T = 300 K in reasonable agreement with experiment. The calculated $S(k, \omega)$ for the liquid also agrees quite well with experiment, especially considering the computational uncertainties inherent in an *ab initio* simulation with its necessarily small number of atoms and limited time intervals.

The remainder of this paper is organized as follows. A brief review of the calculational method is given in Sec. II. The results are presented in Sec. III, followed by a discussion and a summary of our conclusions in Sec. IV.

II. METHOD

Our method is similar to that described in several previous papers,^{10,11,14} but uses the Vienna *ab initio* simulation package (VASP), whose workings have been extensively described in the literature.²⁹ Briefly, the calculation involves two parts. First, for a given ionic configuration, the total electronic energy is calculated, using an approximate form of the Kohn-Sham free-energy density functional, and the force on each ion is also calculated, using the Hellmann-Feynman theorem. Second, Newton's equations of motion are integrated numerically for the ions, using a suitable time step. The process is repeated for as many time steps as are needed to calculate the desired quantity. To hold the temperature constant, we use the canonical ensemble with the velocity rescaled at each time step. Further details of this approach are given in Ref. 10.

The VASP code uses ultrasoft Vanderbilt pseudopotentials,³⁰ a plane-wave basis for the wave functions, with the original Monkhorst-Pack $(3 \times 3 \times 3)$ k-space meshes³¹ and a total of 21 952 plane waves, corresponding to an energy cutoff of 104.4 eV. We use a finite-temperature version of the Kohn-Sham theory,³² in which the electron-gas Helmholtz free energy is calculated on each time step. This version also broadens the one-electron energy levels to case the k-space sums converge more rapidly. Most of our calculations are done using the GGA (Ref. 21) for the exchangecorrelation energy (we use the particular form of the GGA developed by Perdew and Wang²¹), but some are also carried out using the local-density approximation (LDA).

In our iteration of Newton's laws in liquid Ge (*l*-Ge), we typically start from the diamond structure (at the experimental liquid state density for the temperature of interest), then iterate for 901 time steps, each 10 fs, using the LDA. To obtain S(k) within the GGA, we start from the LDA configuration after 601 time steps, then iterate using the GGA for an additional 1641 10-fs time steps, or 16.41 ps. We calculate the GGA S(k) by averaging over an interval of 13.41 ps within this 16.41 time interval, starting at a time t_2 after the start of the GGA simulation. We average over all t_2 's from 1.0 to 3.0 ps.

For comparison, we have also calculated S(k) within the LDA. This S(k) is obtained by averaging over 601 time steps of the 901 time-step LDA simulation. This 601-step interval is chosen to start a time t_1 after the start of this simulation; the calculated LDA S(k) is also averaged over all t_1 's from 1.0 ps to 3.0 ps.

To calculate quantities for amorphous Ge (*a*-Ge), we start with Ge in the diamond structure at T = 1600 K but at the calculated liquid density for that temperature, as given in Ref. 10. Next, we quench this sample to 300 K, cooling at a uniform rate, so as to reach 300 K in about 3.25 ps (in 10-fs time steps). Finally, starting from T = 300 K, we iterate for a further 897 time steps, each of 10 fs, or 8.97 ps, using the LDA. The LDA S(k) is then obtained by averaging over 5.97 of those 8.97 ps, starting at a time t_1 after the system has reached 300 K; we also average this S(k) over all t_1 's from 1.0 to 3.0 ps. To obtain S(k) within the GGA, we start the GGA after 5.7 ps of the LDA simulation, and then iterate using the GGA for an additional 18.11 ps in 10-fs time steps. The GGA S(k) is obtained by averaging over a 15.11-ps time interval of this 18.11-ps run, starting at a time t_2 after the start of the GGA simulation; we also average over all values of t_2 from 1.0 to 3.0 ps.

The reader may be concerned that the 3.25-ps quench time is very short, very unrepresentative of a realistic quench, and very likely to produce numerous defects in the quenched structure, which are not typical of the *a*-Ge studied in experiments. In defense, we note that some of these defects may be annealed out in the subsequent relaxation at low temperatures, which is carried out before the averages are taken. In addition, the static structure factor we obtain agrees well with experiment, and the dynamic structure factor is also consistent with experiment, as discussed below. Thus, the quench procedure appears to produce a material rather similar to some of those studied experimentally. Finally, we note that experiments on *a*-Ge themselves show some variation, depending on the exact method of sample preparation.

We have used the procedure outlined above to calculate various properties of *l*-Ge and *a*-Ge. Most of these calculated properties have been described in previous papers, using slightly different methods, and therefore will be discussed here only very briefly.³³ However, our results for the dynamic structure factor $S(k, \omega)$ as a function of wave vector *k* and frequency ω are new, and will be described in detail. We also present our calculated static structure factor S(k), which is needed in order to understand the dynamical results.

S(k) is defined by the relation

$$S(k) = \frac{1}{N} \left\langle \sum_{i,j} \exp\{i\mathbf{k} \cdot (\mathbf{r}_i(t) - \mathbf{r}_j(t))\} \right\rangle_{t_0} - N \delta_{\mathbf{k},0}, \quad (1)$$

where \mathbf{r}_i is the position of the *i*th ion at time *t*, *N* is the number of ions in the sample, and the triangular brackets denote an average over the sampling time. In all our calculations, we have used a cubic cell with N = 64 and periodic boundary conditions in all three directions. The choices of particle number and cell shape are compatible with any possible diamond-structure Ge within the computational cell.

 $S(\mathbf{k},\omega)$ is defined by the relation (for $\mathbf{k}\neq 0$, $\omega\neq 0$)

$$S(\mathbf{k},\omega) = \frac{1}{2\pi N} \int_{-\infty}^{\infty} \exp(i\,\omega t) \langle \rho(\mathbf{k},t)\rho(-\mathbf{k},0) \rangle dt, \quad (2)$$

where the Fourier component $\rho(\mathbf{k},t)$ of the number density is defined by

$$\rho(\mathbf{k},t) = \sum_{i=1}^{N} \exp[-i\mathbf{k} \cdot \mathbf{r}_{i}(t)].$$
(3)

In our calculations, the average $\langle \dots \rangle$ is computed as

$$\left\langle \rho(\mathbf{k},t)\rho(-\mathbf{k},0)\right\rangle = \frac{1}{\Delta t_1} \int_0^{\Delta t_1} \rho(\mathbf{k},t_1+t)\rho(-\mathbf{k},t_1)dt_1 \quad (4)$$

over a suitable range of initial times t_1 . Because of the expected isotropy of the liquid or amorphous phase, which should hold in the limit of large N, $S(\mathbf{k}, \omega)$ should be a function only of the magnitude k rather than the vector \mathbf{k} , as should the structure factor $S(\mathbf{k})$.

Our calculations are carried out over relatively short times. To reduce statistical errors, we therefore first calculate

$$S(\mathbf{k},\omega,t_{1},t_{2}) = \frac{1}{\pi N} \int_{0}^{t_{2}} dt \rho(\mathbf{k},t_{1}+t) \rho(-\mathbf{k},t_{1}) \exp(i\,\omega t).$$
(5)

For large enough t_2 , $S(\mathbf{k}, \omega, t_1, t_2)$ should become independent of t_2 but will still retain some dependence on t_1 . Therefore, in the liquid, we obtain our calculated dynamic structure factor, $S_{calc}(\mathbf{k}, \omega)$, by averaging over a suitable range of t_1 from 0 to Δt_1 :

$$S_{calc}(\mathbf{k},\omega) = \frac{1}{\Delta t_1} \int_0^{\Delta t_1} dt_1 S(\mathbf{k},\omega,t_1,t_2).$$
(6)

We choose our initial time in the t_1 integral to be 1 ps after the start of the GGA calculation, and (in the liquid) Δt_1 = 6 ps. We choose the final MD time t_2 =16.41 ps. For *a*-Ge, we use the same procedure but t_2 =18.11 ps in our simulations. For our finite simulational sample, the calculated $S(\mathbf{k})$ and $S(\mathbf{k}, \omega)$ will, in fact, depend on the direction as well as the magnitude of **k**. To suppress this finite-size effect, we average the calculated $S(\mathbf{k})$ and $S(\mathbf{k}, \omega)$ over all **k** values of the same length. This averaging considerably reduces statistical error in both $S(k, \omega)$ and S(k).

Finally, we have also incorporated the experimental resolution functions into our plotted values of $S(k,\omega)$. Specifically, we generally plot $S_{av}(k,\omega)/S(k)$, where $S_{av}(k,\omega)$ is obtained from the (already orientationally averaged) $S(k,\omega)$ using the formula

$$S_{av}(k,\omega) = \int_{-\infty}^{\infty} R(\omega - \omega') S(k,\omega') d\omega', \qquad (7)$$

where the resolution function $R(\omega)$ [normalized so that $\int_{-\infty}^{\infty} R(\omega) d\omega = 1$] is

$$R(\omega) = \frac{1}{\sqrt{\pi\omega_0}} \exp(-\omega^2/\omega_0^2).$$
 (8)

In an isotropic liquid, we must have $S(k, -\omega) = S(k, \omega)$, since our ions are assumed to move classically under the calculated first-principles force. Our orientational averaging procedure guarantees that this will be satisfied identically in our calculations, since for every **k**, we always include $-\mathbf{k}$ in the same average. We will nonetheless show results for negative ω for clarity, but they do not provide any additional information.

III. RESULTS

A. S(k) for *l*-Ge and *a*-Ge

In Fig. 1, we show the calculated S(k) for *l*-Ge at T = 1250 K, as obtained using the procedure described in Sec. II. The two calculated curves are obtained using the GGA and the LDA for the electronic energy-density functional; they lead to nearly identical results. The calculated S(k) shows the well-known characteristics already found in pre-



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FIG. 1. (Color online) Static structure factor S(k) for *l*-Ge at T=1250 K, just above the experimental melting temperature. Full curve: present work, as calculated using the generalized gradient approximation (GGA; see text). Dashed curve: present work, but using the local-density approximation (LDA; see text). Open circles: measured S(k) near T=1250 K, as given in Ref. 34.

vious simulations.^{8,10} Most notably, there is a shoulder on the high-*k* side of the principal peak, which is believed to arise from residual short-range tetrahedral order persisting into the liquid phase just above melting. We also show the experimental results of Waseda *et al.*;³⁴ agreement between simulation and experiment is good, and in particular the shoulder seen in experiment is also present in both calculated curves (as observed also in previous simulations).

We have also calculated S(k) for a model of *amorphous* Ge (*a*-Ge) at 300 K. We prepare our sample of *a*-Ge as described in the previous section. For *l*-Ge, we average the calculated S(k) over different **k** vectors of the same length. In Fig. 2, we show the calculated S(k) for *a*-Ge at T=300, again using both the GGA and the LDA. The sample is pre-

pared and the averages obtained as described in Sec. II. In contrast to *l*-Ge, but consistent with previous simulations,^{8,24} the principal peak in S(k) is strikingly split. The calculations are in excellent agreement with experiments carried out on as-quenched *a*-Ge at T = 300 K;³⁵ in particular, the split principal peak seen in experiment is accurately reproduced by the simulations.

We have also calculated a number of other quantities for both *l*-Ge and *a*-Ge, including the pair distribution function g(r), and the electronic density of states n(E). For *l*-Ge, we calculated n(E) using the Monkhorst-Pack mesh with gamma point shifting (one of the meshes recommended in the VASP package). The resulting n(E) is generally similar to that found in previous calculations,^{8,10} provided that an av-



FIG. 2. (Color online) Full curve: Calculated S(k) for *a*-Ge at T=300 K, as obtained using the GGA. Structure is prepared as described in the text. Dashed curve: same as full curve, but calculated in the LDA. Open circles: measured S(k) for *a*-Ge at T=300 K, as given in Ref. 35.



FIG. 3. Calculated ratio of dynamic structure factor $S(k, \omega)$ to static structure factor S(k) for *l*-Ge at T = 1250 K for several values of *k*, plotted as a function of ω , calculated using *ab initio* molecular dynamics with a MD time step of 10 fs. For clarity, each curve has been vertically displaced by 0.05 units from the curve below. In each case, the plotted curve is obtained by averaging both the calculated $S(\mathbf{k}, \omega)$ and the calculated $S(\mathbf{k})$ over all values of \mathbf{k} of the same length. We also incorporate a Gaussian resolution function of half width $\hbar \omega_0 = 2.5$ meV, as in Eqs. (7) and (8). This value of ω_0 is chosen to equal the quoted experimental resolution (Ref. 28).

erage is taken over at least five to ten liquid state configurations. Our n(E) for a-Ge [calculated using a shorter averaging time than that used below for $S(k,\omega)$] is also similar to that found previously.⁸ Our calculated g(r)'s for both *l*-Ge and a-Ge, as given by the VASP program, are similar to those found in Refs. 8 and 10. The calculated number of nearest neighbors in the first shell is 4.18 for a-Ge measured to the first minimum after the principal peak in g(r). For *l*-Ge, if we count as "nearest neighbors" all those atoms within 3.4 Å of the central atom (the larger of the cutoffs used in Ref. 8) we find approximately 7.2 nearest neighbors, quite close to the value of 6.9 obtained in Ref. 8 for that cutoff. Finally, we have recalculated the self-diffusion coefficient D(T) for *l*-Ge at T = 1250 K, from the time derivative of the calculated mean-square ionic displacement; we obtain a result very close to that of Ref. 10.

B. $S(k, \omega)$ for *l*-Ge and *a*-Ge

1. l-Ge

Figure 3 shows the calculated ratio $S(k,\omega)/S(k)$ for *l*-Ge at T=1250 K, as obtained using the averaging procedure described in Sec. II. We include a resolution function [Eqs.



FIG. 4. Same as in Fig. 3, but without the resolution function.

(7) and (8)] of width $\hbar \omega = 2.5$ meV, the same as the quoted experimental width.²⁸ In Fig. 4, we show the same ratio, but without the resolution function (i.e., with $\omega_0 = 0$). Obviously, there is much more statistical noise in this latter case, though the overall features can still be distinguished.

To interpret these results, we first compare the calculated $S(k, \omega)$ in *l*-Ge with hydrodynamic predictions, which should be appropriate at small *k* and ω . This prediction takes the form (see, for example, Ref. 36)

$$2\pi \frac{S(k,\omega)}{S(k)} = \frac{\gamma - 1}{\gamma} \left[\frac{2D_T k^2}{\omega^2 + (D_T k^2)^2} \right] + \frac{1}{\gamma} \left[\frac{\Gamma k^2}{(\omega + c_s k)^2 + (\Gamma k^2)^2} + \frac{\Gamma k^2}{(\omega - c_s k)^2 + (\Gamma k^2)^2} \right].$$
(9)

Here $\gamma = c_P/c_V$ is the ratio of specific heats, constant pressure, and constant volume; D_T is the thermal diffusivity; c_s is the adiabatic sound velocity; and Γ is the sound attenuation constant. D_T and Γ can in turn be expressed in terms of other quantities. For example, $D_T = \kappa_T/(\rho c_P)$, where κ_T is the thermal conductivity and ρ is the atomic number density. Similarly, $\Gamma = \frac{1}{2}[a(\gamma - 1)/\gamma + b]$, where $a = \kappa_T/(\rho c_V)$ and *b* is the kinematic longitudinal viscosity (see, for example, Ref. 36, pp. 264–266).

Equation (9) indicates that $S(k,\omega)$ in the hydrodynamic regime should have two propagating peaks centered at $\omega = \pm c_s k$, and a diffusive peak centered at $\omega = 0$ of width determined by D_T . The calculated $S(k,\omega)/S(k)$ for the three smallest values of k in Fig. 3 does show the propagating peaks. We estimate peak values of $\hbar \omega \sim 10$ meV for k= 5.60 nm⁻¹, $\hbar \omega \sim 11$ meV for k = 7.92 nm⁻¹, and (somewhat less clearly) $\hbar \omega \sim 13$ meV for k = 9.70 nm⁻¹. The value of c_s estimated from the lowest-k value is $c_s \sim 2.7 \times 10^5$ cm/sec. (The largest of these three k values may already be outside the hydrodynamic, linear-dispersion regime.)

These predictions agree reasonably well with the measured $S(k,\omega)$ obtained by Hosokawa *et al.*,²⁸ using inelastic x-ray scattering. For example, the measured sound-wave peaks for $k = \pm 6 \text{ nm}^{-1}$ occur near 10 meV, while those k $= \pm 12 \text{ nm}^{-1}$ occur at $\hbar \omega = 17.2 \text{ meV}$. Furthermore, the integrated relative strength of our calculated sound-wave peaks, compared to that of the central diffusion peak, decreases between $k=7.92 \text{ nm}^{-1}$ and 12.5 nm^{-1} , consistent with both Eq. (9) and the change in experimental behavior²⁸ between $k=6 \text{ nm}^{-1}$ and 12 nm^{-1} .

Because $S(k,\omega)$ in Fig. 3 already includes a significant Gaussian smoothing function, a quantitatively accurate half width for the central peak, and hence a reliable predicted value for D_T , cannot be extracted. A rough estimate can be made as follows. For the smallest-k value of 5.6 nm⁻¹, the full width of the central peak at half maximum is around 7.5 meV. If the only broadening were due to this Gaussian smoothing, the full width would be around $2\hbar\omega_0 = 5.5$ meV. Thus, a rough estimate of the intrinsic full width is $\approx \sqrt{7.5^2 - 5.5^2} = 5$ meV $\approx 2\hbar D_T k^2$. This estimate seems reasonable from the raw data for $S(k,\omega)$ shown in Fig. 4. Using this estimate, one obtains $D_T \approx 1.3 \times 10^{-3}$ cm²/sec.

The hydrodynamic expression for $S(k, \omega)/S(k)$ was originally obtained without consideration of the electronic degrees of freedom. Since *l*-Ge is a reasonably good metal, one might ask if the various coefficients appearing in Eq. (9) should be the full coefficients, or just the ionic contribution to those coefficients. For example, should the value of D_T which determines the central peak width be obtained from the full c_P , c_V , and κ_T , or from only the ionic contributions to these quantities? For *l*-Ge, the question is most relevant for κ_T , since the dominant contribution to c_P and c_V should be the ionic parts, even in a liquid metal.⁴ However, the principal contribution to κ_T is expected to be the electronic contribution.

We have made an order-of-magnitude estimate of D_T using the experimental liquid number density and the value $C_P = (5/3)k_B$ per ion, obtaining the electronic contribution to κ_T from the Wiedemann-Franz law³⁷ together with previously calculated estimates of the electronic contribution.¹⁰ This procedure yields $D_T \sim 0.1 \text{ cm}^2/\text{sec}$, about two orders-of-magnitude greater than that extracted from Fig. 3, and well outside the possible errors in that estimate. We conclude that the D_T which should be used in Eq. (9) for *l*-Ge (and by inference, other liquid metals) is the ionic contribution only.

In support of this interpretation, we consider what one expects for $S(k,\omega)$ in a simple metal such as Na. In such a metal, ionic motions are quite accurately determined by effective pairwise screened ion-ion interactions.⁴ Since the ionic motion is determined by such an interaction, the $S(k,\omega)$ resulting from that motion should not involve the contribution of the electron gas to the thermal conductivity. Although *l*-Ge is not a simple metal, it seems plausible that its $S(k,\omega)$ should be governed by similar effects, at least in the hydrodynamic regime. This plausibility argument is supported by our numerical results.

For k beyond around 12 nm⁻¹, the hydrodynamic model should start to break down, since the dimensionless parameter $\omega \tau$ (where τ is the Maxwell viscoelastic relaxation time) becomes comparable to unity. At these larger k's, both our calculated and the measured²⁸ curves of $S(k,\omega)/S(k)$ continue to exhibit similarities. Most notable is the existence of a single, rather narrow peak for k near the principal peak of S(k), followed by a reduction in height and broadening of this central peak as k is further increased. This narrowing was first predicted by de Gennes.³⁸ In our calculations, it shows up in the plot for $k = 20.9 \text{ nm}^{-1}$, for which the half width of $S(k,\omega)/S(k)$ is quite narrow, while at k=28.5 and 30.7 nm^{-1} , the corresponding plots are somewhat broader and lower. By comparison, the measured central peak in $S(k,\omega)/S(k)$ is narrow at $k=20 \text{ nm}^{-1}$ and especially at k = 24 nm⁻¹, while it is broader and lower at k = 28 nm⁻¹.²⁸

The likely physics behind the de Gennes narrowing is straightforward. The half width of $S(k, \omega)$ is inversely proportional to the lifetime of a density fluctuation of wave number k. If that k coincides with the principal peak in the structure factor, a density fluctuation will be in phase with the natural wavelength of the liquid structure, and should decay slowly, in comparison to density fluctuations at other wavelengths. This is indeed the behavior observed both in our simulations and in experiment.

In further support of this picture, we attempt to describe these fluctuations by a very oversimplified Langevin model. We suppose that the Fourier component $\rho(\mathbf{k},t)$ [Eq. (3)] is governed by a Langevin equation

$$\dot{\rho}(\mathbf{k},t) = -\xi \rho(\mathbf{k},t) + \eta(t). \tag{10}$$

Here the dot is a time derivative, ξ is a constant, and $\eta(t)$ is a random time-dependent "force" which has ensemble average $\langle \eta \rangle = 0$ and correlation function $\langle \eta(t) \eta^*(t') \rangle = A \,\delta(t - t')$. Equation (10) can be solved by standard methods (see, e.g., Ref. 36 for a related example), with the result (for sufficiently large t)

$$\langle \rho(\mathbf{k},t)\rho^*(\mathbf{k},t+\tau)\rangle = \frac{\pi A}{\xi} \exp(-|\tau|\xi).$$
 (11)

According to Eq. (2), $S(k,\omega)$ is, to within a constant factor, the frequency Fourier transform of this expression, i.e.,

$$S(k,\omega) \propto \int_{-\infty}^{\infty} (\pi A/\xi) \exp(i\omega\tau) \exp(-|\tau|\xi) d\tau, \quad (12)$$

or, on carrying out the integral,



FIG. 5. Calculated $S(k,\omega)/S(k)$ for *a*-Ge at T=300 K at $k \le 35$ nm⁻¹, plotted as a function of ω . Again, each curve has been vertically displaced by appropriate amounts from the one below it, as evident from the figure, and both $S(k,\omega)$ and S(k) have been plotted after an average over all **k**'s of the same length. We also incorporate a Gaussian resolution function of half width $\hbar \omega_0 = 2$ meV. This value is chosen to give the best results for $n_{vib}(\omega)$ as measured by Ref. 39. The time step here is 10 fs.

$$S(k,\omega) \propto \frac{\pi A}{\omega^2 + \xi^2}.$$
 (13)

This is a Gaussian function centered at $\omega = 0$ of half width ξ . On the other hand, the static structure factor

$$S(k) \propto \lim_{\tau \to 0} \langle \rho(\mathbf{k}, t) \rho^*(\mathbf{k}, t+\tau) \rangle = \frac{\pi A}{\xi}.$$
 (14)

Thus, if the constant A is independent of **k**, the half width ξ of the function $S(k,\omega)$ at wave number **k** is inversely proportional to the static structure S(k). This prediction is consistent with the "de Gennes narrowing" seen in our simulations and in experiment.²⁸

To summarize, there is overall a striking similarity in the shapes of the experimental and calculated curves for $S(k,\omega)/S(k)$ both in the hydrodynamic regime and at larger values of *k*.

2. a-Ge

We have also calculated the dynamic structure for our sample of *a*-Ge at T=300 K. The results for the ratio $S(k,\omega)/S(k)$ are shown in Figs. 5 and 6 for a range of k values, and, over a broader range of ω , in Fig. 7. Once again, both $S(k,\omega)$ and S(k) are averaged over different values of **k** of the same length, as described above. We have incorporated a resolution function of width $\hbar\omega_0=2$ meV into $S(k,\omega)$. This width is a rough estimate for the experimental



FIG. 6. Same as Fig. 5, but at $k \ge 35$ nm⁻¹.

resolution function in the measurements of Maley *et al.*,³⁹ we assume it to be smaller than the liquid case because the measured width of the central peak in $S(k,\omega)/S(k)$ for *a*-Ge is quite small.

Ideally, our calculated $S(k, \omega)$ should be compared to the measured one. However, the published measured quantity is not $S(k, \omega)$ but is, instead, based on a modified dynamical structure factor, denoted $G(k, \omega)$, and related to $S(k, \omega)$ by³⁹

$$G(k,\omega) = \left(\frac{C}{k^2}\right) \left[\frac{\hbar\omega}{n(\omega,T)+1}\right] S(k,\omega).$$
(15)

Here *C* is a *k*- and ω -independent constant, and $n(\omega,T) = 1/[e^{\hbar\omega/k_BT} - 1]$ is the phonon occupation number for phonons of energy $E = \hbar \omega$ at temperature *T*. The quantity plotted by Maley *et al.*³⁹ is an average of $G(k,\omega)$ over a range of *k* values from 40 to 70 nm⁻¹. These workers assumed that this average is proportional to the vibrational density of states $n_{vib}(\omega)$. The measured $n_{vib}(\omega)$ as obtained in this way³⁹ is shown in Fig. 8 for two different amorphous structures, corresponding to two different methods of preparation, and having differing degrees of disorder.

In order to compare our calculated $S(k,\omega)$ to experiment, we use Eq. (15) to infer $G(k,\omega)$, then average over a suitable range of k. However, in using Eq. (15), we use the classical form of the occupation factor, $n(\hbar\omega)+1 \approx k_B T/\hbar\omega$. This choice is justified because we have calcu-



FIG. 7. (Color online) Calculated $S(k,\omega)/S(k)$ as in Figs. 5 and 6 but including higher frequencies ω . At $\hbar \omega = 60$ meV, the curves are arranged vertically in order of increasing frequency. Each curve is vertically displaced by 0.0005 units from the one below it.

lated $S(k,\omega)$ using classical equations of motion for the ions. We thus obtain for the calculated vibrational density of states

$$n_{vib}(\omega) \approx \left(\frac{C\hbar^2}{k_BT}\right) \left(\frac{\omega^2}{k^2}\right) S(k,\omega).$$
 (16)

In Fig. 8 we show two such calculated plots of $n_{vib}(\omega)$, obtained by averaging Eq. (16) over two separate groups of k's, as indicated in the caption⁴⁰. For comparison, we also show $n_{vib}(\omega)$ for *a*-Ge as calculated in Ref. 8 directly from the Fourier transform of the velocity-velocity autocorrelation function.

The calculated plots for $n_{vib}(\omega)$ in Fig. 8 have some distinct structure, which arises from some corresponding highfrequency structure in $S(k,\omega)$. The plot of $n_{vib}(\omega)$ for the group of smaller *k*'s has two distinct peaks, near 8 meV and 29 meV, separated by a broad dip with a minimum near 18 meV. The plot corresponding to the group of larger *k*'s has similar structure and width, but the dip is less pronounced. The two experimental plots also have two peaks separated by a clear dip. The two maxima are found around 10 and 35 meV, while the principal dip occurs near 16 meV. In addition, the overall width of the two densities of states is quite similar.

The reasonable agreement between the calculated and measured $n_{vib}(\omega)$ suggests that our *ab initio* calculation of $S(k,\omega)$ for *a*-Ge is reasonably accurate. The noticeable differences probably arise from several factors. First, there are several approximations involved in going from the calculated and measured $S(k,\omega)$'s to the corresponding $n_{vib}(\omega)$'s, and



FIG. 8. (Color online) Full curve: calculated vibrational density of states $n_{vib}(\omega)$, in units of 10^{-3} states/meV. $n_{vib}(\omega)$ is obtained from the resolution broadened $S(k,\omega)$ and S(k) of Fig. 7 using the formula $n_{vib}(\omega) = \langle G_{calc}(k, \omega) \rangle$, where $G_{calc}(k,\omega)$ is given by the right-hand side of Eq. (16), and the averaging is carried out over the three magnitudes of k near 40 nm⁻¹ for which we have computed $S(k,\omega)$. Dashed curve: same as full curve, but calculated by averaging over the six magnitudes of k near 90 nm⁻¹ for which we have computed $S(k,\omega)$. The open circles and open stars denote the measured $n_{vib}(\omega)$, as reported in Ref. 39 for two forms of a-Ge. Finally, the open diamonds denote $n_{vib}(\omega)$ as calculated in Ref. 8 from the ionic velocity-velocity autocorrelation function (dot-dashed curves). In all plots except that of Ref. 8, $n_{vib}(\omega)$ is normalized so that $\int_0^{\omega_{max}} n(\omega) d(\hbar \omega) = 1$. ω_{max} is the frequency at which $n_{vib}(\omega) \rightarrow 0$, and is estimated from this figure by extrapolating the right-hand parts of the solid and dashed curves linearly to zero.

these may be responsible for some of the discrepancies. Secondly, there may actually be differences between the particular amorphous structures studied in the experiments, and the quenched, then relaxed structure considered in the present calculations. (However, the similarities in the *static* structure factors suggest that these differences are not vast.) Finally, our calculations are carried out over relatively short times, using relatively few atoms; thus, finite-size and finite-time effects are likely to produce some additional errors. Considering all these factors, agreement between calculation and experiment is quite reasonable.

Previous ab initio calculations for a-Ge (Ref. 8) have also obtained a vibrational density of states, but this is computed directly from the ionic velocity-velocity autocorrelation function rather than from the procedure described here. The calculations in Ref. 8 do not require computing $S(k,\omega)$. In the present work, by contrast, we start from $S(k,\omega)$ (which is calculated here in an *ab initio* calculation for *a*-Ge), and we work backwards to get $n_{vib}(\omega)$. In principle, our $S(k,\omega)$ includes all anharmonic effects on the vibrational spectrum of a-Ge, though in extracting $n_{vib}(\omega)$ we assume that the lattice vibrates harmonically about the metastable atomic positions. In Fig. 8, we also show the results of Ref. 8 for $n_{nib}(\omega)$ as obtained from this correlation function. They are quite similar to those obtained in the present work, but have a somewhat deeper minimum between the two principal peaks.

The quantity $n_{vib}(\omega)$ could, of course, also be calculated directly from the force-constant matrix, obtained by assuming that the quenched configuration is a local energy minimum and calculating the potential energy for small positional deviations from that minimum using *ab initio* molecular dynamics. This procedure has been followed for *a*-GeSe₂, for example, by Cappelletti *et al.*⁴¹ These workers have then obtained $S(q, \omega)$ versus *q* from their $n_{vib}(\omega)$ at selected values of ω , within a one-phonon approximation. However, as

- *Present address: Institute for Physical Science and Technology, University of Maryland, College Park, Maryland 20742. Email address: jdchai@wam.umd.edu
- [†]Corresponding author. FAX: (614)292-7557. Email address: stroud@mps.ohio-state.edu
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noted above, the present work produces the full $S(k,\omega)$ and thus has, in principle, more information than does $n_{vib}(\omega)$.

IV. DISCUSSION AND CONCLUSIONS

The present results show that *ab initio* molecular dynamics can be used to calculate the dynamic structure factor $S(k,\omega)$ for both liquid and amorphous semiconductors. Although the accuracy of the calculated $S(k,\omega)$ is lower than that attained for static quantities, such as S(k), nonetheless it is sufficient for comparison to most experimental features. This is true even though our calculations are limited to 64atom samples and fewer than 20 ps of elapsed real time.

We have presented evidence that the calculated $S(k,\omega)/S(k)$ in *l*-Ge agrees qualitatively with that measured by inelastic x-ray scattering,²⁸ and that the one calculated for *a*-Ge leads to a vibrational density of states qualitatively similar to the quoted experimental one.³⁹ Since such calculations are thus shown to be feasible, our work should spur further numerical studies, with longer runs on larger samples, to obtain even more detailed information. Furthermore, we can use these dynamical simulations to probe the underlying processes at the atomic scale which give rise to specific features in the measured and calculated $S(k,\omega)$.

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