Raman phonons as a probe of disorder, fluctuations, and local structure in doped and undoped orthorhombic and rhombohedral manganites

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We present a rationalization of the Raman spectra of orthorhombic and rhombohedral, stoichiometric and doped, manganese perovskites. In particular, we study $RMnO_3$ (R = La, Pr, Nd, Tb, Ho, Er, Y, and Ca) and the different phases of Ca- or Sr-doped RMnO₃ compounds as well as cation deficient RMnO₃. The spectra of manganites can be understood as combinations of two kinds of spectra corresponding to two structural configurations of MnO_6 octahedra and independently of the average structure obtained by diffraction techniques. One type of spectra corresponds to the orthorhombic Pbnm space group for octahedra with cooperative or dynamic Jahn-Teller distortions, with stretching modes as the main features and whose frequencies correlate to Mn-O distances. The other spectrum is associated to regular but tilted octahedra whose modes can be described in the rhombohedral R3c structure, where only bending and tilt modes are observed. The main peaks of compounds with regular MnO_6 octahedra, such as CaMnO₃, highly Ca-doped LaMnO₃, or the metallic phases of Ca- or Sr-doped LaMnO₃, are bending and tilt MnO_6 octahedra modes which correlate to R-O(1) bonds and Mn-O-Mn angles, respectively. In low and optimally doped manganites, the intensity and width of the broad bands are related to the amplitude of the dynamic fluctuations produced by polaron hopping in the paramagnetic insulating regime. The activation energy, which is proportional to the polaron binding energy, is the measure of this amplitude. This study permits to detect and confirm the coexistence, in several compounds, of a paramagnetic matrix with lattice polaron together with regions without dynamic or static octahedron distortions, identical to the ferromagnetic metallic phase. We show that Raman spectroscopy is an excellent tool to obtain information on the local structure of the different microphases or macrophases present simultaneously in many manganites.

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I. INTRODUCTION

Manganese perovskites $R_{1-x}A_x$ MnO₃ (R = La or rare earth, A = Ca, Sr, Ba, ...) have recently attracted much interest because of their colossal magnetoresistance (CMR) effect¹ that makes these systems promising for magnetic sensors and reading heads devices. In spite of the tremendous amount of published studies on this subject, many experimental facts are not well understood or have to be reinterpreted. Theories do not give yet a quantitative description of, for example, the metal-insulator phase transition, but converge in the necessity to include electron-lattice interactions which are able to localize the carriers into small polarons through their coupling to lattice distortions, disorder, or phonons.² There is also large consensus on the importance of intrinsic inhomogeneities and phase coexistence that, in the optimally doped regime, can be visualized as polaron clusters in the ferromagnetic matrix and metallic clusters in the insulating phase.³

The phonons involved in the theories for these CMR systems are even stretching modes, such as symmetric stretching (SS) (breathing mode) or antisymmetric stretching (AS) mode [Jahn-Teller (JT) mode].⁴ Both normal modes are identical to the lattice distortions achieved by Mn^{3+} ions in order to split the e_g electronic level (JT effect). It is therefore important to identify these phonons and study their behavior through the different structural phases and doping levels.

An anomalous frequency hardening and narrowing of the linewidth of the tilt mode as the temperature decreases

through the magnetic transition has been reported in La_{0.67}Ca_{0.33}MnO₃ compound. This behavior has been explained by a double exchange mechanism with electronphonon coupling.⁵ Liarokapis *et al.*⁶ observed, for $x \ge 0.4$, the disappearance of the high-frequency stretching modes in the La_{1-x}Ca_xMnO₃ series. Granado *et al.*⁷ report, in La_{1-x}Mn_{1-x}O₃ the softening of the high-frequency phonons across *Tc* and explained it to be caused by a strong spinphonon coupling. On the other hand, the Raman background can give information on the electronic excitations. The change of the diffusive electronic Raman scattering in the paramagnetic phase has been attributed to the change from small to large polaron regimes across the paramagnetic-to-ferromagnetic phase transition.^{8,9}

The knowledge of the lattice vibrations and their correlation to the different phases, local order, and conduction mechanisms is of crucial importance. A clear demonstration of this relation is the frustration of the insulator-to-metallic transition, only by the isotope substitution of O^{16} by O^{18} in a $La_{1/3}Nd_{1/3}Ca_{1/3}MnO_3$ compound.¹⁰ But there is still controversy on the interpretation of the Raman spectra even for stoichiometric *R*MnO₃, whose spectrum corresponds well to the expected normal modes for its orthorhombic structure.^{11–13} The 490-cm⁻¹ peak has been assigned to a bending¹¹ or to an asymmetric stretching.^{6,14} Amelitchev *et al.*¹⁵ studied the dependence of the low-frequency tilt mode with the tolerance factor in many compounds. They assign the broad peaks (around 490 and 610 cm⁻¹) in doped compounds to the second-order Raman scattering. Nevertheless, these authors state as remarkable the absence of the strong phonon lines near 490 and 612 cm⁻¹ that dominate LaMnO₃, while the modes below 350 cm^{-1} are preserved. A detailed Raman study of CaMnO₃ has been published¹⁶ during the publication process of the present work. We do not coincide with this assignment of the tilt and R modes (discussion in Sec. IV B). Björnsson et al.¹⁷ observe two different sets of phonons in La_{0.8}Sr_{0.2}MnO₃, as well as the appearance of new narrow peaks in La_{0.9}Sr_{0.1}MnO₃ at low temperatures. Their explanation for x = 0.2 is the existence of pronounced local orthorhombic distortions, in the rhombohedral structure, which vanish at low temperature. However, they cannot explain the evolution with temperature of the Raman spectra in the x = 0.1 compound. The Raman spectra of doped or nonstoichiometric manganites in their different structural and magnetic phases for all doping levels are not well understood.

In the present work, we study orthorhombic stoichiometric $RMnO_3$ (R=La, Pr, Nd, Tb, Ho, Er, Y, and Ca) as well as A site doped or cation deficient orthorhombic or rhombohedral $RMnO_3$ for different doping levels.

II. EXPERIMENTAL DETAILS

Polycrystalline $RMnO_3$ (R = Pr, Nd, Tb, Ho, Er, and Y) and CaMnO₃ samples were obtained by citrate techniques. For R = Pr and Nd, the precursors were treated at 1100 °C in a N₂ flow for 12 h; annealing treatments in an inert atmosphere were necessary to avoid the formation of nonstoichiometric Pr (Nd) $MnO_{3+\delta}$ phases, containing a significant amount of Mn^{4+} . For R = Tb, the precursor powders were heated at 1000 °C in air for 12 h. Finally, low-temperature treatments were necessary, for the samples with R = Ho, Y, and Er, to increase the yield of the orthorhombic phases, preventing or minimizing the stabilization of competitive hexagonal $RMnO_3$ phases.¹⁸ In the case of stoichiometric LaMnO₃, a single crystal was available; it has been prepared by the floating-zone method.¹⁹ The rhombohedral LaMnO_{3+ δ}, as well as PrMnO_{3+ δ}, samples were prepared in polycrystalline form by a citrate technique as described elsewhere.20,21

Finally, Ca-doped LaMnO₃ compounds were prepared by the classical ceramic method by heating stoichiometric amounts of La₂O₃, MnO₂, and CaCO₃ for 72 h at 1400 °C. Materials so obtained were quenched in air.

Raman spectra were obtained with a Jobin-Yvon HR 460 monochromator coupled to a liquid-nitrogen-cooled chargecoupled device. The excitation light was the 514.5-nm line of a Spectra Physics Ar-Kr laser. The incident and scattered beams were focused using an Olympus microscope. A Kaiser SuperNotch filter was used to suppress the elastic light. The laser power was reduced down to 0.1 mW (depending on the sample) in order to avoid the local heating of the samples at the laser spot. We used a continuous flow Oxford Instrument cryostat CF2102 to perform measurement from 10 K to room temperature (rt).

III. STRUCTURE AND NORMAL MODES

All the samples studied here present the same Pbnm $(D_{2h}^{16}, \text{ with } Z=4)$ orthorhombic structure, except $La_{0.7}Sr_{0.3}MnO_3$ and $LaMnO_{3+\delta}$ that are rhombohedral with the R3c $(D_{3d}^6, \text{ with } Z=2)$ space group.^{18,20,22} The Pbnm orthorhombic RMnO3 compounds are structurally distorted with respect to the cubic perovskite in two ways: the MnO_6 octahedra present a strong Jahn-Teller cooperative distortion due to Mn³⁺ ions, and the octahedra are tilted in order to optimize the R-O bond lengths. CaMnO₃ has the same Pbnm structure as the RMnO3 series but the Jahn-Teller distortion is negligible.²³ As doping RMnO₃ with Ca or Sr, the Jahn-Teller distortion decreases as a consequence of the introduction of Mn4+ cations, and the structure becomes more regular, maintaining the tilting of the octahedra. For a broad range of Sr doping, or for oxygen deficient samples, the reported structure is rhombohedral with tilted octahedra and strictly identical Mn-O bonds.²²

The Raman-active modes of the *Pbnm* structure (the mirror plane "m" is perpendicular to the long *c* axis) are $7A_g + 7B_{1g} + 5B_{2g} + 5B_{3g}$. Mn ions do not participate with any Raman mode, as they are located at inversion centers, while La and O(1) ions display the same kind of movements. These 24 Raman-active modes can be classified into two symmetric and four antisymmetric stretching modes, four bending modes and six rotation and tilt modes, of the octahedra. At last, eight modes are related to *A* site movements.

There are 30 normal modes at the zone center for the R3c(Z=2) rhombohedral structure $A_{1g}+3A_{2g}+2A_{1u}+4A_{2u}$ $+4E_g+6E_u$. Among these, $1A_g+4E_g$ are Raman-active modes, $3A_{2u}+5E_u$ are IR active, and the remaining $2A_{1u}$ $+3A_{2g}$ are silent modes. For this structure, the Raman-active modes can be classified into $1A_{1g}+1E_g$ rotational or tilt modes, $1E_g$ bending and $1E_g$ antistretching of the MnO₆ octahedra, and the remaining E_g is related to a vibration of Aions.²⁴ The symmetric stretching mode is A_{2g} and therefore not observable.

IV. RESULTS AND DISCUSSION

We discuss the assignment of the main Raman peaks of Ca-doped and undoped $RMnO_3$ orthorhombic compounds, and explain the apparently different behavior of their frequencies. We have classified the manganese perovskites depending on the distortion of the Mn-O octahedra. $RMnO_3$ compounds present a strong cooperative JT distortion, a trivalent state for Mn ions, an insulating behavior, high resistivity, and are paramagnetic at rt. A second group corresponds to compounds with a negligible JT distortion either orthorhombic (as CaMnO_3, highly doped La_{1-x}Ca_xMnO₃ $x \ge 0.5$ or the ferromagnetic metallic phase of x=0.33 compound), or rhombohedral as La_{0.67}Sr_{0.33}MnO₃. At last we discuss the paramagnetic phase of the 0 < x < 0.5 compounds as well as nonstoichiometric $RMnO_3$ samples and phase coexistence.



FIG. 1. Raman spectra at rt of *Pbnm* $RMnO_3$ pellets. LaMnO₃ is a single crystal. Same phonons in the spectra are connected with vertical lines. *R* stands for *R* ion modes, *T* stands for the tilt, *B* stands for bending, AS stands for antisymmetric, and SS stands for symmetric stretching.

A. Compounds with strong cooperative Jahn-Teller distortion: *R*MnO₃

We have analyzed the Raman spectra at room temperature of orthorhombic $RMnO_3$ compounds as a function of the chemical pressure by changing the rare-earth ion (Fig. 1). There is an agreement between all authors in the assignment of the peak around 610 cm⁻¹, related to a symmetric stretching of the basal oxygen ions of the octahedra (B_{1g}), and that around 280 cm⁻¹, related to some octahedra tilt; but there are several claims concerning the one around 480 cm⁻¹ and some confusion in the bending, tilt, and rock modes in doped compounds.^{6,11,15} We assigned the 480-cm⁻¹ peak to an antisymmetric stretching (A_g) associated with the JT distortion^{13,14} and give here further evidence. Regarding tilt and bending modes, different combinations of displacements of the in-plane O(2) ions along z axis and of the apical O(1)



FIG. 2. Dependences, vs the unit-cell volume, of the rt frequencies of the symmetric and antisymmetric stretching modes (a), bending modes (b), and tilt mode (c) for $RMnO_3$ (black symbols) and $La_{1-x}Ca_xMnO_3$ series (gray symbols). The numbers in the figure indicate the percentage of Ca doping. The inset shows the linear dependence of the tilt mode frequency with the tilt angle which is defined as (180-[Mn-O(1)-Mn])/2

oxygen perpendicular to it, give rise to four tilt modes $(A_g$ $(B_{2g}+B_{3g})$ and $(B_{2g}+B_{3g})$ and two bending modes (A_g) $+B_{1g}$). We have grouped in brackets the Raman modes that may have very similar frequencies because they correspond to the same movements around x or y axis. Raman spectra corresponding to modes with B_{2g} and B_{3g} symmetries have a very low intensity,¹¹ so we will focus on A_g and B_{1g} modes. The peaks at the lowest energies (marked with R in Fig. 1) correspond to rare-earth movements. Table I collects the measured frequencies together with the values estimated for the three R modes ($\omega_R = \omega_Y \sqrt{M_Y/M_R}$), considering only the effect of the R mass and taking Y compound frequencies as references. The three R modes behave exactly as expected in all measured compounds. The differences between estimated and measured frequencies are lower than 2 cm^{-1} except for $LaMnO_3$ (6 and 9 cm⁻¹), due to the increase of La-O(1) distances compared to Y-O(1) ones.

In Fig. 1, the two most intense peaks (610 and 480 cm⁻¹) vary only slightly (less than 10 cm⁻¹) with the rare-earth one (Fig. 2). The symmetric (around 610 cm⁻¹) and the an-

TABLE I. Observed and estimated *R* modes frequencies for *R*MnO₃ compounds. The estimated values has been obtained with $\omega_R = \omega_Y \sqrt{M_Y/M_R}$. TW stands for "this work." (*) this peak is assigned in Ref. 16 to a *R* mode (A_g symmetry).

	Ca (40)			Y (89)		La (139)		Nd (144)		Tb (159)		Ho (165)		Er (167)	
Assignment	Estimated	TW	Ref. 15	TW	Ref. 11	Estimated	TW	Estimated	TW	Estimated	TW	Estimated	TW	Estimated	ΤW
$\overline{A_g/B_{1g}}$	220		242/259	148	151/151	118	109	116.5		110.5	111	108.8	109	108	
A_g	274		278	184	188	147	141	144.9	143	137	138	135.2	136	134	134
B_{1g}	325	322	322*	218	220	174		171.6		162.5	160	160.3	160	159.1	157



FIG. 3. Dependences, vs the unit-cell volume, of mean Mn-O distances (a), mean R-O(1) distances (b), and the octahedra tilt angle (c) for RMnO₃ (structural data from Refs. 18 and 19) and La_{1-x}Ca_xMnO₃ series (structural data from Refs. 6 and 26–28). The numbers in the figure indicate the percentage of Ca doping.

tisymmetric, or Jahn-Teller (around 480 cm⁻¹) stretching modes involve nearly pure Mn-O bond stretching therefore a simple $(dMn-O)^{-1.5}$ dependence of these phonon frequencies is expected²⁵ and observed. This is an indication that the MnO₆ octahedron volume and bonds are insensitive to quite large changes in the tolerance factor in perovskites when comparing compounds with the same Mn valence state (in this case 3^+). On the contrary, the frequencies of the peaks labeled with "B" and "T" increase strongly (over 100 cm^{-1}) when the cell volume shrinks [Figs. 2(b) and 2(c)]. The behavior of these phonons is parallel to that of the mean value of the two shortest R-O(1) bonds and to the static octahedra tilt (deviation from the cubic structure defined as (180-[Mn-O(1)-Mn)]/2), respectively [Figs. 3(b) and 3(c)]. We assign the "B" peaks (two peaks can be distinguished) to bending modes $(A_g + B_{1g})$ and the "T" peak to $(A_g + B_{1g})$ tilt modes. The tilt and bending modes with A_g symmetry, which correspond to rotations around y axis, are plotted in Figs. 4(a) and 4(b), respectively. The B_{1g} modes correspond to the same atomic movements but around x axis, therefore A_g and B_{1g} modes are expected to be very similar in frequency. The displacement of O(1) in these modes is, in fact, a stretching of R-O(1) bonds [Fig. 4(c)]. Each O(1) ion is surrounded by four R ions with two short and two long distances. Figure 3(b) displays the evolution of the mean value of the two short R-O(1) bonds. Comparing Figs. 2(b) and 3(b), the correlation between the bending mode frequency and the R-O bond length becomes obvious.

In summary, the present data show that stretching mode frequencies correlate to Mn-O bond distances, while the



FIG. 4. (a) A_g Octahedra tilt mode, (b) and (c) A_g octahedra bending mode. Both in the *Pbnm* structure.

R-O bond and octahedra tilt angle dominate bending and tilt modes, respectively. These are the expected behaviors for such normal modes and must be kept in mind when trying to assign the spectra of doped or related compounds.

B. Insulating paramagnetic $La_{1-x}Ca_xMnO_3$ from x=0 to 1

CaMnO₃ is also described with the *Pbnm* space group but with important differences compared to the previous $RMnO_3$ compounds, which are the clues to understand their very different Raman spectra [Fig. 5(c), 0% and 100% spectra]. The Mn valence state is 4+ therefore neither static nor dynamic JT distortions are present since the $3d e_g$ Mn orbital is empty. CaMnO₃ cannot be described with the cubic perovskite structure because the tilt of the octahedra remains. The two pairs of peaks (around 240 cm^{-1} and 460 cm^{-1}) cannot correspond to stretching modes (the dominant ones in RMnO₃). Taking into account the much shorter Mn-O distances, their frequencies would be around 520 and 660 cm^{-1} . We assign these quite sharp peaks to the bending $(A_g = 487 \text{ cm}^{-1}, B_{1g} = 466 \text{ cm}^{-1})$ and tilt $(A_g = 245 \text{ cm}^{-1})$, $B_{1g} = 260 \text{ cm}^{-1})$ modes $(A_g \text{ modes are plotted in Fig. 4})$. The frequencies and relevant structural parameters of Cadoped series are presented in Figs. 2 and 3 making obvious their parallel behavior and confirming the mode assignment.

During the referee process, a study of the Raman spectra of CaMnO₃ appeared.¹⁶ These authors assigned the doublet at 242 and 258 cm⁻¹ to A_g and B_{2g} modes involving O(1) and Ca vibrations in the *x*-*z* plane (in the *Pnma* representation that correspond to A_g and B_{1g} modes and *x*-*y* plane in *Pbnm* one), and the 184 cm⁻¹ peak to the A_g tilt mode. Looking at Table I, the *R* modes in CaMnO₃ are expected at 220, 274, and 325 cm⁻¹ and, in fact, two peaks of the cor-



FIG. 5. (a) Cooperative static JT distortion, from diffraction data in Refs. 16 and 20–22, for $La_{1-x}Ca_xMnO_3$ series; (b) Raman spectra of FM metallic Ca-doped (50 K) and Sr-doped (300 K) LaMnO_3 with x=0.33; (c) rt Raman spectra of $La_{1-x}Ca_xMnO_3$ samples, as function of Ca content. The arrows point to the narrow peaks corresponding to the tilt and bending modes.

rect symmetries are observed at 278 (Ref. 16) and 322 cm⁻¹ [Fig. 5(c) and Ref. 16]. The Ca-O(1) distances are similar to Y-O(1) ones, consequently, the rule that has been found to be valid for all RMnO₃ series should be valid for Ca compound. Therefore, the peaks at 242 and 259 cm⁻¹, which show differences compared to the expected values over 22 cm⁻¹ (Table I), are not the *R* modes. These peaks correspond to oxygen tilt modes. In a first approximation, as the unit-cell volume decreases, the frequencies of all the modes are expected to increase because the bonds shrink. The reason for the tilt modes to decrease is due to the reduction of the tilt angle as shown in the inset of Fig. 2.

Figure 5(c) collects the Raman spectra of $La_{1-x}Ca_xMnO_3$ compounds for several doping levels. Very significant changes are observed. Compared to stoichiometric LaMnO₃, for doping levels as low as 10%, the width of the stretching modes increases and their spectral weight decreases dramati-



FIG. 6. *Pbnm* orthorhombic (a) and rhombohedral R3c (b) $RMnO_3$ structures. The rhombohedral axes have been rotated in order to better visualize the similarities between the octahedra network in both structures. The A_{1g} tilt mode is shown.

cally with doping. At 50% Ca doping and above, no stretching modes appear and the spectra resembles strongly to the $CaMnO_3$ one.

It is important to note that the paramagnetic CaMnO₃, highly doped compounds [Fig. 5(c)], and the ferromagnetic metallic phase of La_{0.67}Ca_{0.33}MnO₃ [Fig. 5(b)] present very similar Raman spectra. These quite different systems have in common that the cooperative JT distortion is negligible [Fig. 5(a)] and that they do not present the stretching modes. We can conclude that a JT distortion, or probably a distortion engendering significantly different Mn-O bond lengths, is necessary to produce a measurable intensity of the stretching modes. The allowed modes by global symmetry considerations are identical to the previous $RMnO_3$ compounds, but the change in the polarizability due to a particular mode, which determines its Raman intensity, is found to vanish for the stretching modes of the regular octahedra.

On the other hand, these spectra are nearly identical to that corresponding to rhombohedral $La_{0.7}Sr_{0.3}MnO_3$ (LSMO) [Fig. 5(b)]. The observable Raman-active modes of the R3cstructure correspond to bending and tilt modes of the oxygen octahedra, whose atomic displacements are very similar to the bending and tilt modes of the orthorhombic structure [see tilt mode in Figs. 4(a) and 6(b)]. From R3c to Pbnm the number of expected peaks increases. Only one bending E_g and two tilt modes $(A_{1g} + E_g)$ are expected in the R3c rhombohedral structure, while four bending and four tilts modes correspond to the orthorhombic *Pbnm* one. As shown in Fig. 6, the octahedra network in the Pbnm and R3c structures are very similar. Both present tilted octahedra with similar angles (about 160° for R3c), but *Pbnm* structure can present up to three different Mn-O distances, while all Mn-O bond lengths are identical in the rhombohedral one. But highly doped Ca compounds, as well as the metallic phases of Ca- and Sr-doped compounds, do not present co-



FIG. 7. Raman spectra at rt of nonstoichiometric $RMnO_3$ (R = La and Pr) as well as 33% Ca-doped compounds. Bottom: Raman spectrum of LaMnO₃ at T > 800 K above the JT transition. Small arrows point to the narrow peaks related to the R3c-type phase and the star's position to the modes associated with the *Pbnm*-type phase with dynamic and/or cooperative JT distortion.

operative or dynamic JT distortions. In these cases, the octahedra in *Pbnm* and $R\overline{3}c$ space groups are almost identical. Even if the stretching modes are allowed by symmetry in the *Pbnm* space group, the origin of the close similarity of their spectra, in particular the absence of high-frequency modes, becomes clear.

We conclude therefore that there are two types of spectra, one corresponding to compounds with a cooperative JT distortion (LaMnO₃ type) that is dominated by the stretching phonons, and the other without these modes corresponding to compounds without JT distortion, such as CaMnO₃, metallic *Pbnm* phases, or rhombohedral $La_{1-x}Sr_xMnO_3$.

C. Paramagnetic phases of low to optimally doped manganites

At low doping levels, for example La_{0.9}Ca_{0.1}MnO₃ in Fig. 5(c) or RMnO_{3- δ} (R = La or Pr) in Fig. 7, the spectra are similar to LaMnO₃ one but with wider peaks. The dynamic fluctuations of the octahedra caused by the lattice polaron hopping increase very substantially their width. But La_{0.66}Ca_{0.33}MnO₃ (LCMO) or La_{0.9}MnO₃ (Fig. 7) present some spectral weight in the stretching frequency range (AS and SS) together with narrower peaks (indicated with arrows). In order to check whether these peaks have the same origin as the broad high-frequency bands, we compare their spectra to the undoped LaMnO₃ at T>800 K (lower spec-

trum in Fig. 7) above the so-called JT transition. At these temperatures, the cooperative JT distortion is melted but dynamical JT fluctuations remain.^{14,19} This is a very similar scenario to what occurs in the paramagnetic phase of LCMO, therefore similar Raman spectrum would be expected. Figure 7 shows that above 800 K the intensity has decreased drastically compared to rt spectrum (Fig. 5), but some spectral weight is observable in the high-frequency region corresponding to the stretching modes while tilt and bending modes have almost vanished. It is evident that dynamic distortions of the MnO₆ octahedra cannot give rise to the quite narrow peaks around 230 and 420 cm⁻¹, but are the origin of the broad stretching bands.

Doping LaMnO₃ gives rise to Mn valence mixing and local static distortions around the doping ion. Simultaneously, in the paramagnetic phase, the hopping electrons trapped as lattice polarons induce lattice distortions. This disorder is of dynamic character and is expected to change at the insulator-to-metallic phase transition and has been described as a crossover from small-to-large polaron regimes.⁸ Diffraction techniques give us valuable information on the mean interatomic distances but not on instantaneous or randomly distributed atomic positions which are important for the Raman-scattering effect. The analysis of the diffuse x-ray scattering²⁹ and reverse Monte Carlo simulations of neutrondiffraction patterns^{30,31} have shown that the local octahedron distortions are much larger, around and above Tc, than the ones obtained by standard diffraction analysis. Diffraction shows very slight structural changes at the ferromagnetic metallic transition, while Raman spectra above [Fig. 5(c)] and below [Fig. 5(b)] Tc are quite different. Above Tc, the features similar to high temperature LaMnO₃ correspond to the paramagnetic matrix dynamically distorted by polaron hopping, and the ones similar to CaMnO₃ spectrum indicate that the part of the sample has a structure alike the metallic ferromagnetic phase. It might correspond to the magnetic clusters or magnetic polarons that have been observed by different techniques.³² The structure of these entities is unknown but present ferromagnetic correlations. In the present knowledge of phase segregation or inhomogeneous intrinsic phase, present in most of the manganites, it is realistic to think that the second spectrum corresponds to ferromagnetic metallic droplets in the paramagnetic matrix with polaronic (lattice polarons) conduction. At low temperatures, well in the metallic regime, only the narrow peaks, similar to CaMnO₃, remain.

The coexistence of the two types of spectra, related to two phases or microphases, is observed in several compounds in Figs. 5(c) and 7 and, for example, in Fig. 1(b) of Ref. 16. In fact, in the present scenario we can understand the temperature dependence of spectra from Ref. 17. The compounds with x=0.2 and 0.1 are both paramagnetic insulators at high temperatures. The x=0.2 one becomes ferromagnetic (FM) and metallic at 280 K, while x=0.1 transforms in a FM insulator (FMI) at 200 K. The temperature dependence of La_{0.8}Sr_{0.2}MnO₃ spectrum is "identical" to the changes observed in La_{0.67}Ca_{0.33}MnO₃, in spite of their different aver-



FIG. 8. Normalized resistance of samples with several Ca content. Lines are fits to the paramagnetic state resistance with $R = R_0 \exp(E_a/K_BT)$. The inset shows the obtained activation energies E_a vs Ca content.

age structures ($R\overline{3}c$ and Pbnm, respectively). In the insulating paramagnetic phases, dynamically distorted octahedra (Pbnm type) coexist with regular octahedra (R3c type), while in the ferromagnetic metallic regime only regular octahedra remain. On the contrary, in x=0.1 Sr-doped $LaMnO_3$, above Tc, only the broad features are observed corresponding to the Pbnm structure with dynamic distortions related to the polaron hopping. But, below Tc, the appearance of narrows peaks, in coexistence with the broad bands, indicates that metallic, and probably ferromagnetic, clusters in the insulating matrix are formed. In the discussion about the nature of the FMI phase of this x = 0.1 compound, on whether it is a canted ferromagnetic or phase segregation is occurring, Raman spectrum is consistent with the second hypothesis. Nevertheless, a more detailed analysis is necessary.

When Ca content increases the binding energy of the lattice polarons decreases, this indicates the reduction of the activation energy for conduction in the paramagnetic phase (Fig. 8). The reduction of the polaron energy implies the collapse of the amplitude of the dynamic distortions it produces in the lattice and, therefore, of the spectral weight of the LaMnO₃-type spectrum. The width of the instantaneous distribution of Mn-O distances is the origin of the width of the Raman peaks. For doping concentrations of 50% and above the lattice is in fact, instantaneously, much more regular than at low doping. We recover the CaMnO₃- or LSMO- type spectrum. In the metallic regime, the hopping carriers become delocalized and do not induce lattice distortions. Sharp phonon peaks are observed below 450 cm⁻¹ independently of the crystallographic structure (compare metallic *Pbnm* LCMO and metallic R_3c LSMO at rt [Fig. 5(b)] and at low temperature.³³) Therefore, the width and intensity of the stretching modes are a measure of the amplitude of the dynamical JT distortions, which correlates to the activation energy for polaronic conduction.

At last, the main characteristic of the Raman spectra of most charge ordered phases, that is the recovery of the stretching phonon peaks, is understood as being caused by the orbital order and the concomitant cooperative JT distortion.

V. CONCLUSIONS

The analysis of the Raman spectra of RMnO₃ Pbnm compounds that present strong cooperative JT distortion show that the stretching modes (symmetric and antisymmetric) correlate to Mn-O bonds, while bending modes are determined by the R-O(1) mean distance and tilt modes by Mn-O-Mn angle. The compounds without cooperative or dynamic distortions of the MnO₆ octahedra, independently of their mean crystallographic structure, do not present stretching modes and the characteristics of their spectra can be approached to the R3c space group vibrations. The bending and tilt normal modes are equivalent to the Pbnm ones and the measured frequencies follow the same rules as $RMnO_3$ ones. This explains the similarities between insulating $CaMnO_3$ (*Pbnm*) and the metallic phases of orthorhombic La_{0.67}Ca_{0.33}MnO₃ and rhombohedral La_{0.67}Sr_{0.33}MnO₃. The activation energy for conduction in the insulating phases, which is proportional to the polaron binding energy, is a measure of the amplitude of the dynamic distortion it produces in the lattice. Therefore, as the Ca content increases, the activation energy decreases as well as the amplitude of the dynamic distortions. The width of the peaks corresponding to RMnO₃-type spectrum increases and their intensity decreases as the Ca content rises up to about 50%, where these peaks vanish and only the CaMnO₃-type spectrum is observed. The identification of these types of spectra and their correspondence with particular octahedron configurations are the keys to understand the Raman spectra of most manganese perovskite. Moreover, this allows to detect the simultaneous presence of several different phases and to obtain insight in their local structure.

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- ¹S. Jin, T.H. Tiefel, M. McCormack, R.A. Fastnacht, R. Ramesh, and L.H. Chen, Science **264**, 413 (1994).
- ²M. Ziese, Rep. Prog. Phys. **65**, 143 (2002).
- ³E. Dagotto, T. Hotta, and A. Moreo, Phys. Rep. 344, 1 (2001).
- ⁴Philip B. Allen and Vasili Perebeinos, Phys. Rev. B **60**, 10 747 (1999).
- ⁵J.C. Irwin, J. Chrzanowski, and J.P. Franck, Phys. Rev. B **59**, 9362 (1999).
- ⁶E. Liarokapis, Th. Leventouri, D. Lampakis, D. Palles, J.J. Neumeier, and D.H. Goodwin, Phys. Rev. B **60**, 12 758 (1999).
- ⁷E. Granado, A. García, J.A. Sanjurjo, C. Rettori, I. Torriani, F. Prado, R.D. Sánchez, A. Caneiro, and S.B. Oseroff, Phys. Rev. B 60, 11 879 (1999).
- ⁸S. Yoon, H.L. Liu, G. Schollerer, S.L. Cooper, P.D. Han, D.A. Payne, S.W. Cheong, and Z. Fisk, Phys. Rev. B 58, 2795 (1998).
- ⁹H.L. Liu, S. Yoon, S.L. Cooper, S.W. Cheong, P.D. Han, and D.A. Payne, Phys. Rev. B 58, 10 115 (1998).
- ¹⁰M.R. Ibarra, Guo-meng Zhao, J. M De Teresa, B. García-Landa, Z. Arnold, C. Marquina, P.A. Algarabel, H. Keller, and C. Ritter, Phys. Rev. B **57**, 7446 (1998).
- ¹¹M.N. Iliev, M.V. Abrashev, H.G. Lee, V.N. Popov, Y.Y. Sun, C. Thomsen, R.L. Meng, and C.W. Chu, Phys. Rev. B **57**, 2872 (1998).
- ¹²V.B. Podobedov, A. Weber, D.B. Romero, J.P. Rice, and H.D. Drew, Phys. Rev. B 58, 43 (1998).
- ¹³L. Martín-Carrón, A. de Andrés, M.T. Casais, M.J. Martínez-Lope, and J.A. Alonso, J. Alloys Compd. **323-324**, 494 (2001).
- ¹⁴L. Martín-Carrón and A. de Andrés, Eur. Phys. J. B 22, 11 (2001);
 J. Alloys Compd. 323-324, 417 (2001).
- ¹⁵ V.A. Amelitchev, B. Güttler, O.Yu. Gorbenko, A.R. Kaul, A.A. Bosak, and A.Yu. Ganin, Phys. Rev. B 63, 104430 (2001).
- ¹⁶M.V. Abrashev, J. Bäckström, L. Börjesson, V.N. Popov, R.A. Chakalov, N. Kolev, R.-L. Meng, and M.N. Iliev, Phys. Rev. B 65, 184301 (2002).
- ¹⁷ P. Björnsson, M. Rübhausen, J. Bäckström, M. Käll, S. Eriksson, J. Eriksen, and L. Börjesson, Phys. Rev. B **61**, 1193 (2000).

- ¹⁸J.A. Alonso, M.J. Martínez-Lope, M.T. Casais, and M.T. Fernández-Díaz, Inorg. Chem. **39**, 917 (2000).
- ¹⁹J. Rodríguez-Carvajal, M. Hennion, F. Moussa, A.H. Moudden, L. Pinsard, and A. Revcolevschi, Phys. Rev. B 57, 3189, (1998).
- ²⁰J.A. Alonso, M.J. Martínez-Lope, M.T. Casais, J.L. MacManus-Driscoll, P.S.I.P.N. de Silva, L.F. Cohen, and M.T. Fernández-Díaz, J. Mater. Chem. 7, 2139 (1997).
- ²¹J.A. Alonso, Philos. Trans. R. Soc. London, Ser. A **356**, 1617 (1998)
- ²²A. Urushibara, Y. Moritomo, T. Arima, A. Asamitsu, G. Kido, and Y. Tokura, Phys. Rev. B **51**, 14 103 (1995).
- ²³K.R. Poeppelmeier, M.E. Leonowicz, J.C. Scanlon, J.M. Longo, and W.B. Yelon, J. Solid State Chem. 45, 71 (1982).
- ²⁴M.V. Abrashev, A.P. Litvinchuk, M.N. Iliev, R.L. Meng, V.N. Popov, V.G. Ivanov, R.A. Chakalov, and C. Thomsen, Phys. Rev. B **59**, 4146 (1999).
- ²⁵ A. de Andrés, S. Taboada, J.L. Martínez, A. Salinas, J. Hernádez, and R. Sáez-Puche, Phys. Rev. B **47**, 14 898 (1993).
- ²⁶P. Dai, J. Zhang, H.A. Mook, S-H. Liou, P.A. Dowben, and E.W. Plummer, Phys. Rev. B **54**, R3694 (1996).
- ²⁷P.G. Radaelli, D.E. Cox, M. Marezio, and S-W. Cheong, Phys. Rev. B 55, 3015 (1997).
- ²⁸S. Faaland, K.D. Knudsen, M.A. Einarsrud, L. Rørmark, R. Høier, and T. Grande, J. Solid State Chem. **140**, 320 (1998).
- ²⁹S. Shimomura, N. Wakabayashi, H. Kuwahara, and Y. Tokura, Phys. Rev. Lett. 83, 4389 (1999).
- ³⁰D. Louca and T. Egami, Phys. Rev. B **59**, 6193 (1999).
- ³¹M. García-Hernádez, A. Mellergard, F.J. Mompean, D. Sánchez, A. de Andrés, R.L. McGreevy, and J.L. Martínez, cond-mat/0201436 (unpublished).
- ³²J.M. De Teresa, M.R. Ibarra, P.A. Algarabel, C. Ritter, C. Marquina, J. Blasco, J. García, A. del Moral, and Z. Arnold, Nature (London) **386**, 256 (1997).
- ³³E. Granado, N.O. Moreno, A. García, J.A. Sanjurjo, C. Rettori, I. Torriani, S.B. Oseroff, J.J. Neumeier, K.J. McClellan, S.W. Cheong, and Y. Tokura, Phys. Rev. B 58, 11 435 (1998).