Electronic structures of doped anatase TiO₂: Ti₁_{ Λ **}***M***_{***x***}O₂ (***M* **= Co, Mn, Fe, Ni)**

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We have investigated electronic structures of a room-temperature-diluted magnetic semiconductor: Codoped anatase $TiO₂$. We have obtained the half-metallic ground state in the local-spin-density approximation $(LSDA)$ and the insulating ground state in the $LSDA+U+SO$ incorporating the spin-orbit interaction. In the stoichiometric case, the low spin state of Co is realized with the substantially large orbital moment. However, in the presence of oxygen vacancies near Co, the spin state of Co becomes intermediate. The ferromagnetisms in the metallic and insulating phases are accounted for by the double-exchange-like and the superexchange mechanisms, respectively. Further, the magnetic ground states are obtained for Mn- and Fe-doped TiO₂, while the paramagnetic ground state is obtained for Ni-doped $TiO₂$.

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Diluted magnetic semiconductors (DMS's) have been studied extensively in the last decade, because of their potential usage of both charge and spin degrees of freedom of carriers in the electronic devices, namely, the spintronics. There have been trials based on two types of DMS families: II-VI, such as Mn-doped CdTe and ZnSe,¹ and III-V, such as Mn-doped GaAs.² Especially, the latter attracts great attention, because it becomes a ferromagnetic (FM) DMS having the Curie temperature $T_c \sim 110$ K. Motivated by the above FM DMS, recent research effort has been focused on developing new FM semiconductors operating at room temperature.^{3,4} It has been reported that the FM DMS's are realized in other types of systems too.^{5–7}

Matsumoto et al.⁶ fabricated Co-doped anatase TiO₂ thin film samples, $Ti_{1-x}Co_xO_2$, using the combinatorial pulsedlaser-deposition (PLD) molecular-beam epitaxy (MBE) technique. A sizable amount of Co, up to $x=0.08$, is soluble in anatase $TiO₂$. Using the scanning superconducting quantum interference device microscope, they observed the magnetic domain structures in Co-doped films characteristic of the FM long-range ordering. The measured saturated magnetic moment per Co ion was $0.32\mu_B$ apparently in the low spin state and T_c was estimated to be higher than 400 K. This sample is conductive at room temperature, but becomes semiconducting at low temperatures. It also exhibits a large positive magnetoresistance of 60% at 2 K in a field of 8 T. Due to transparent property of the system, it can be used in integrated circuits and storage devices with display units.⁸

More recently the Co-doped anatase $TiO₂$ film grown by the oxygen-plasma-assisted (OPA) MBE was reported by Chambers *et al.*⁹ They claimed that magnetic properties of the OPA-MBE grown material are better than those of the PLD-MBE grown material because considerably larger saturated magnetic moment of $1.26\mu_B$ /Co is observed, which seems to be more consistent with the low spin state of Co. The unquenched orbital moment of Co in the asymmetric crystalline field was ascribed to the enhanced magnetic moment. The Co *L*-edge x-ray absorption spectrum of Co-doped $TiO₂$ is similar to that of CoTiO₃, whereby the formal oxidation state of Co^{2+} has been suggested. They also found that the magnetic and structural properties depend critically

on the Co distribution, which varies widely with the growth condition.

Hence the magnetic properties of Co-doped anatase $TiO₂$ are still controversial. To explore these properties, the first essential step is to study the electronic structure of Co-doped anatase $TiO₂$. In this study, we have investigated electronic structures of Co-doped anatase TiO₂: Ti_{1-x}Co_xO₂ (*x* $=0.0625$ and (0.125) using the linearized muffin-tin orbital (LMTO) band method both in the local-spin-density approximation $(LSDA)$ and the $LSDA+U+SO$ incorporating the Coulomb correlation interaction *U* and the spin-orbit interaction.10 For a comparison, we have also investigated electronic structures of other transition-metal-doped $TiO₂$: $Ti_{1-x}M_xO_2$ (*M* = Mn, Fe, Ni).

 $TiO₂$ has three kinds of structures, rutile, anatase, and brookite. The space group of anatase structure is tetragonal $I4_1$ /*amd*. The anatase TiO₂ is composed of stacked edgesharing octahedrons formed by six O anions. Ti atoms are in the interstitial sites of octahedrons that are distorted with different bond lengths between the apical (1.979 Å) and the equatorial (1.932 Å) Ti-O bond and with the Ti-O-Ti angle 156.3°. For $Ti_{1-x}Co_xO_2$ ($x=0.0625$), we have considered a supercell containing 16 f.u. in the primitive unit cell by replacing one Ti by Co $(T_{15}Co_{1}O_{32}:a=b=7.570,c$ $=$ 9.514 Å). Sixteen empty spheres are employed in the interstitial sites to enhance the packing ratio for the LMTO band calculation.

We have first calculated the electronic structure of anatase $TiO₂$ without doping elements. The overall band structure of the present LMTO result is consistent with existing results, 11 except that the energy gap is estimated to be a bit larger, \sim 4 eV, as compared to the full-potential-augmented-planewave (FLAPW) result of \sim 2 eV.¹¹ Although the present result is closer to the experimental energy gap of 3.2 eV, it is likely that the energy gap is overestimated due to the open structure of anatase $TiO₂$ and the minimal basis of the LMTO band method. The valence band top and the conduction band bottom correspond to mainly O 2*p* and Ti 3*d* states, respectively. The electronic transport and magnetic experiments on anatase $TiO₂$ indicate peculiar properties, such as shallow donor level and high mobility of the *n*-type carriers due to intrinsic oxygen off-stoichiometry.¹²

FIG. 1. The LSDA total and PLDOS of $Ti_{1-x}Co_xO_2$ (*x* $=0.0625$.

To examine the energetics of $Ti_{1-x}Co_xO_2$ between the FM and antiferromagnetic (AFM) configurations of Co ions, we have performed the LSDA band calculation for anatase $Ti_{1-x}Co_xO_2$ ($x=0.125$). In this case, there are two Co ions in the unit cell $(Ti_{14}Co_2O_{32})$ separated by 5.353 Å. In between two Co ions, there are one Ti and two O ions. As a result, we have obtained that the FM phase has *half-metallic* electronic structure, while the AFM phase has semiconducting electronic structure. Total energies are very close, but the FM phase is lower by \sim 6 mRy than the AFM phase. Hence, in the following discussion, we will consider only the FM configurations of Co ions.

Now, we have performed band calculations for anatase $Ti_{1-x}Co_xO_2$ ($x=0.0625$). Fig. 1 shows the density of states (DOS) obtained from the LSDA band calculation. The energy gap between O 2*p* and Ti 3*d* states is almost unchanged by Co doping and most of Co *d* states are located in the energy-gap region. Noteworthy is the *half-metallic* nature in this system, reminiscent of the Mn-doped GaAs, $13-15$ that is, the conduction electrons at the Fermi level E_F are 100% spin polarized. The carrier types, however, are different in the two. Here the Fermi level cuts the Co t_{2g} states, whereas, in Mn-doped GaAs, the Fermi level cuts mainly the As *p* states since Mn 3*d* states are located far below E_F . The different

carrier types would give rise to the different magnetic mechanisms as discussed below.16 The crystal field splitting between t_{2g} and e_g states is larger than the exchange splitting between t_{2g} states, suggesting the low spin state of Co. The total spin magnetic moment is $1\mu_B$, which comes mostly from Co ions. Ignoring the extended Co *d* states between -5 and -1 eV, which are hybridized bonding states with O 2*p* states, the characters of localized *d* states are mainly t_{2g}^3 spin-up and t_{2g}^2 spin-down states, seemingly corresponding to the ionic valence of $Co⁴⁺$. However, almost two electrons are occupied in the extended Co *d* states, and so the total occupancy of d states amounts to d^7 .

The half-metallic LSDA result for Co-doped $TiO₂$ seems to be compatible with the metallic resistivity behavior above 100 K. Further, in view of the carrier type of Co 3*d*, the FM ground state can be understood based on the *doubleexchange-like* mechanism, e.g., the kinetic-energy gain through the hopping of fully spin-polarized carriers in the half-metallic system. This is contrary to the case of Mndoped GaAs in which As *p* hole carriers mediate the Ruderman-Kittel-Kasuya-Yosida (RKKY)-like exchange interaction. Note, however, that at a lower temperature the system behaves as an insulator.⁶ Since the unfilled t_{2g} states near E_F are very narrow, one can expect that the Coulomb correlation interaction and/or the Jahn-Teller interaction would induce the metal-insulator transition. The Jahn-Teller effect would be relatively weak because the relevant orbitals near E_F are t_{2g} states. Thus we have explored the effect of the Coulomb correlation interaction using the $LSDA+U$ $+$ SO band method. The spin-orbit interaction is taken into account to describe properly atomiclike Co t_{2g} states.

Indeed, the $LSDA+U+SO$ band calculation with parameter values of $U=3.0$ eV and $J=0.87$ eV for Co 3*d* electrons yields the semiconducting ground state in accord with the experiment. The DOS plot in Fig. 2 shows that the t_{2g} spin-down states are separated by the *U* effect with the bandgap size of ~ 0.8 eV. The total spin magnetic moment of $1\mu_B$ and the occupancy of d^7 are also obtained by the LSDA⁺U⁺SO. In the inset of Fig. 2(c), the angular distribution of occupied Co 3*d* spin-down states is plotted based on the orbital occupancies. The main contribution to these states comes from the d_{xy} state mixed partially with d_{yz} and d_{zx} , which is reflected in the shape of the angular distribution. Due to the spin-orbit effect, however, the shape is a bit asymmetric and distorted.

Evidently, atomiclike Co t_{2g} states would yield the unquenched orbital moment. In fact, Co-ion has remarkably large orbital magnetic moment of $0.9\mu_B$, which is as large as that in CoO (\sim 1.0 μ_B).¹⁷ The large orbital moment arises from occupied t_{2g} spin-down states split by the Coulomb correlation and the spin-orbit interaction $(Fig. 2)$. The orbital moment is polarized in parallel with the spin moment, and so the total magnetic moment amounts to $1.9\mu_B$ /Co. The large orbital magnetic moment is in agreement with the expectation by Chambers *et al.*, ⁹ but the total magnetic moment 1.9μ _B is much larger than their experimental value of $1.26\mu_B$.

FIG. 2. The LSDA+U⁺SO total and PLDOS for $Ti_{1-x}Co_xO_2(x=0.0625)$. The angular distribution of occupied Co $3d$ spin-down states [inset (c)].

As mentioned above, oxygen vacancies are easily formed in the anatase $TiO₂$, and so there will also be intrinsic oxygen vacancies in Co-doped TiO₂. We thus examined the effects of oxygen deficiency in the Co-doped TiO₂. By removing one oxygen atom in the supercell $(Ti_{15}Co_1O_{31})$, the formal valence of Co becomes $Co²⁺$ in the ionic picture. We have considered two cases of removing an oxygen atom: (i) from the Ti-contained octahedron, and (ii) from the Co-contained octahedron. For the oxygen vacancy near the Ti site, essentially the same Co 3*d* projected local density of states (PLDOS) is obtained as for the stoichiometric case with the low spin $1.0\mu_B$ /Co and the orbital magnetic moment 0.9μ _B/Co, implying that the Co sites are not affected much by the vacancy.¹⁸ On the other hand, for the oxygen vacancy near the Co site, very different features are revealed: the intermediate (close to the high) spin state is realized rather than the low spin state with the spin magnetic moment of $2.53\mu_B$ /Co. The intermediate spin state in this case results from the reduced crystal field in the pyramidal structure composed of five oxygen anions. Then the d_{z^2} and d_{zx} states become more stabilized than in the case of stoichiometric octahedral structure (see Fig. 3). The configuration of occupied states becomes d^7 with spin-up $t_{2g}^3 d_{z^2}^1$ and extended $d_{x^2-y^2}^{1-\alpha}$ bonding states and spin-down e_g^1 and extended $t_{2g}^{1+\alpha}$

FIG. 3. The LSDA+ U +SO Co PLDOS of oxygen deficient Co-doped TiO₂ (Ti₁₅Co₁O₃₁).

(mainly d_{zx}) bonding states. Due to decreased t_{2g} characters near E_F , the orbital magnetic moment is reduced to $0.28\mu_B$. Comparing the total energies between above two cases, the oxygen vacancy near the Ti site is more stable than the oxygen vacancy near the Co site. Therefore oxygen vacancies tend to be formed mostly in the Ti-contained octahedrons without affecting the Co spin state. However, the possible oxygen vacancies near Co sites formed during the nonequilibrium MBE growth would influence drastically the magnetic properties in Co-doped TiO₂ film. This explains the observation that the magnetic properties depend critically on the film growth condition.

For comparison, we have also examined electronic structures of other transition-metal-doped anatase $TiO₂$: $Ti_{1-x}M_xO_2$ (*M* = Mn, Fe, Ni) for *x* = 0.0625. The same *U* and *J* parameters were employed for Mn and Fe 3*d* electrons. Figure 4 shows that Ni-doped TiO₂ has the paramagnetic ground state, whereas, Mn- and Fe-doped systems have the magnetic ground states with local magnetic moments of $3.0\mu_B$ and $3.7\mu_B$, respectively. Ignoring the extended bonding d states, the apparent nominal valences look like Mn^{4+} $(d³)$ and Fe⁴⁺ $(d⁴)$. Including the extended states, however, the electron configuration for the Mn-doped case becomes d^5 with spin-up $t_{2g}^3 e_g^1$ states and spin-down bonding $t_{2g}^{\alpha} e_g^{1-\alpha}$ states. Likewise, for the Fe-doped case, the electron configu-

FIG. 4. The LSDA+U DOS of Mn, Fe-doped DOS, and the LSDA DOS of Ni-doped TiO₂($x=0.0625$).

ration becomes d^6 with spin-up $t_{2g}^3 e_g^{2-\alpha}$ states and spindown bonding $t_{2g}^{\beta}e_{g}^{1+\alpha-\beta}$ states. Therefore, for both cases, the intermediate states close to the high spin states are realized even without oxygen defects.

No evidence of FM behavior has been observed in Mn- or Fe-doped $TiO₂$ film yet. The different magnetic natures in

these systems are presumably due to their different electronic structures. Note that the characters of unoccupied states near E_F are different: t_{2g} for the Co-doped TiO₂ (Fig. 2), whereas e_g for the Mn- and Fe-doped TiO₂. For Mn-doped case, even the LSDA yields the insulating electronic structure, and so, by considering only the localized states, the superexchange via occupied $\text{Mn}(t_{2g})$ -O(p_{σ})-Mn(t_{2g}) orbitals would lead to the nearest-neighbor AFM interaction. On the other hand, for Fe-doped case, the LSDA yields the half-metallic electronic structure, as for Co-doped case. The Jahn-Teller effects would be more operative in this case due to the e_g characters near E_F , which will drive the structural distortion and the concomitant metal-insulator transition. Then the AFM phase is more likely to be stabilized. In contrast, the Co-doped $TiO₂$ even in its insulating phase would have the nearest-
neighbor FM superexchange interaction via superexchange interaction via $Co(t_{2g})-O(p_{\pi})-Co(t_{2g})$ kinetic-exchange energy gain. Therefore, the FM phases in the metallic and insulating states are accounted for by the double-exchange-like and the superexchange mechanisms, respectively. In reality, however, the situation may not be that simple, since there could be some effects of extra carriers originating from the intrinsic or extrinsic O vacancies. The present study serves to provide basic band-structure information for understanding the magnetic mechanisms in these systems.

In conclusion, the LSDA yields the *half-metallic* ground state for Co-doped TiO₂ with the carrier type of mainly Co 3*d* states. In contrast, the LSDA+ U +SO yields the insulating ground state and the low spin state with $1\mu_B$ spin moment and 0.9μ ^B orbital moment per Co ion. The possible oxygen vacancies near Co sites substantially affect the magnetic properties: the intermediate spin state of Co with 2.53μ _B spin moment is realized and the orbital moment is reduced to $0.28\mu_B$. We have also found that Mn- and Fedoped $TiO₂$ have the magnetic ground states, while Ni-doped $TiO₂$ has the paramagnetic ground state.

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