Self-organization of (001) cubic crystal surfaces

Laurent Proville

Groupe de Physique des Solides, UMR 7588-CNRS, Universités Paris 7 & Paris 6, Tour 23, 2 pl. Jussieu 75251, Paris Cedex 05, France (Received 10 May 2001; published 3 October 2001)

Self-organization on crystal surface is studied as a two-dimensional spinodal decomposition in presence of a surface stress. The elastic Green function is calculated for a (001) cubic crystal surface taking into account the crystal anisotropy. Numerical calculations show that the phase separation is driven by the interplay between domain boundary energy and long range elastic interactions. At late stage of the phase separation process, a steady state appears with different nanometric patterns according to the surface coverage and the crystal elastic constants.

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I. INTRODUCTION

Self-organization (SO) on solid surface is an efficient mean for growing nanostructures with regular sizes and spacings. The models proposed by Marchenko¹ and Vanderbilt *et al.*^{2,3} are the basis of the theoretical framework to understand the SO phenomenon. They enhanced the interplay between the long range elastic interaction yielded by the underlying crystal and the domain boundary energy. Indeed, the former is minimum when two surface defaults are separated by a distance as large as possible while the latter is minimum when only one compact domain appears onto the surface. So when these two ingredients are present (see below for experimental descriptions), the surface ground state structure should be a compromise between the aforementioned interactions. The purpose of Refs. 1–3 was to evaluate this structure in different cases.

Assuming that two phases, called A and B, coexisting onto a crystal surface, have different intrinsic stresses, the authors of Refs. 1,2 showed that a state which consists of stripes domains occupied alternatively by A and B lowers the energy with a period selection. This period depends on the crystal stress energy compared with the domain boundary energy and it varies exponentially with the ratio of those two quantities. In order to perform an analytical surface elastic Green function calculation, Marchenko and Vanderbilt both assumed either a crystal anisotropy along one direction of the surface or the anisotropy of the intrinsic stresses. Those works specially addressed the cases of corrugated crystal surfaces and crystal surface reconstruction with broken symmetry (see Refs. 4,5 for recent relevant experimental analysis).

Comparing the stability of different periodic domains in an isotropic two-dimensional dipolar model, Vanderbilt *et al.*³ proved that at intermediate coverage, $\theta_0 > 0.28$, the stripe structure is the optimal candidate for the twodimensional system ground state while at low coverage, θ_0 <0.28, the droplets structure is more stable. As the dipoledipole interaction is similar to the elastic interaction, i.e., it decreases as the inverse of the distance to the power three, these Vanderbilt's results hold for the solid surface SO. It shows that the anisotropy of the intrinsic stress surface, assumed in Refs. 1,2 is not essential in the SO process since the patterning occurs even if the surface is isotropic with no symmetry breaking. The studies of Refs. 1–3 emphasize the main physical ingredients which drive SO, i.e., the interplay between domain boundary energy and elastic interaction energy.

Recent analysis of chemisorbed monolayers on (001)Copper surfaces, via scanning tunneling microscopy^{6–8} (STM) and spot profile analyzing low energy electron diffraction⁹ (SPA-LEED) show mesoscopic morphologies different from droplet or stripe structures. For the N/Cu(001) case, the nitrogen is chemically adsorbed on a (001) Copper crystal surface, and aggregates within square-shaped islands that may arrange either in one-dimensional rafts at low nitrogen coverage or in a two-dimensional array at intermediate coverages. The experimental works mentioned above^{7,9} motivated the present study.

Here we propose to describe the SO kinetics on solid surface as a two-dimensional spinodal decomposition. In this standard theory, we include the stress energy due to the underlying crystal and therefore the surface elastic Green function is calculated. The 2D Cahn-Hilliard equation which drives the surface diffusion, is integrated with computer means. Our method allows us to capture the SO kinetics together with the elastic anisotropy due to the crystal symmetries. The latter feature is proved to play a role in the nanometric arrangements. For simplicity, our study is focused on a (001) cubic crystal surface.

In the last decade, a similar theory has been developed under the name of "phase field" by Khachaturyan *et al.* for the kinetics of phase transition in alloys.¹⁰ The phase field name is also used for models of solidification; see Ref. 11 for recent developments. We choose not to use this ambiguous terminology. In the framework of the SO spinodal theory, the implied coexistent phases may represent either two types of crystal facet, noted A and B or a chemical adsorbed layer A over a clean crystal surface B. In the latter case, we neglect the possible layer thickness.

Our results exhibit a steady state at late stage of the separation process which shows different mesoscopic patterns according to the coverage and to the elastic constants of the materials. Because of the crystal cubic symmetry, the nanometric morphologies of the final state may differ from the one predicted in Refs. 1-3. At low coverage, rafts of either disk or square shaped islands appear while at intermediate coverage a branched stripe structure occurs. The elastic constants of the material are shown to determine the preferential orientations of the rafts and of the stripes. If the surface square symmetry is broken, the kinetics final state is similar to those predicted in Refs. 1,2, i.e., an assembly of regular spaced stripes.

II. DOMAIN BOUNDARY ENERGY

The covering parameter θ describes the A-B coexistent phases at the solid surface. These phases may differ by their composition or their geometry. Let us say that $\theta = 0$ for B and $\theta = 1$ for A. If one assumes that the spatial surface variations of θ are smooth with respect to the atomic scale, a two-dimensional coarse-graining procedure is thus relevant to represent the mesoscopic system state. If *d* is the size of the elementary coarse-grained surfaces, we introduce the mesoscopic quantity $\theta(\mathbf{r})$ which is the θ average, performed over a whole elementary surface, centered at position \mathbf{r} . The $\theta(\mathbf{r})$ variable is the mesoscopic local coverage of the surface since it is the A quantity per unit area which is present in the \mathbf{r} vicinity. The θ average over the entire surface is written θ_0 and it is assumed to be conserved during the system time evolution.

An inhomogeneous mixing of A and B phases involves an energy increasing because of atomic bond breaking at domain boundaries. This energy together with the entropic term due to the surface inhomogeneity are captured in the twodimensional free Ginzburg-Landau (GL) functional energy

$$F_{\text{chem}} = F_0 \int \int_{S} \left\{ \frac{\gamma}{2} \left[\left(\frac{\partial \theta}{\partial x_1} \right)^2 + \left(\frac{\partial \theta}{\partial x_2} \right)^2 \right] + \hat{f}(\theta) \right\} d\mathbf{r}.$$
(2.1)

We introduce here the adimensional free energy density $\hat{f} = 16.\theta^2(1-\theta)^2$ which is a double well potential with minima for $\theta = 0$ and $\theta = 1$. The hat notation points out the adimensional quantities. The \hat{f} form has no direct influence on the mesoscopic structure providing GL functional is invariant with respect to the (001) surface space group. Let us note $\hat{g}(\theta)$, the continuous Hamiltonian including both the 2D gradient term and the free energy density \hat{f} .

The F_0 and γ scalars are respectively the free energy density constant and the amplitude of the gradient term that both must be adjusted to set the model domain boundary energy I to a realistic value, i.e., around 10 meV/Å (see Ref. 2). At phase equilibrium, the I quantity is given by I $=F_0\int_{R}^{A}\hat{g}(\theta)d\vec{l}$ where the integration is performed along a line path that goes from inside a B phase domain to inside an A domain.¹² It is easy to see that the previous path integral is overvalued by the product of the A-B interface width times the \hat{f} maximum value, i.e., $\hat{f}(\theta = 0.5) = 1$. Fixing $\gamma = 20d^2$, which insure the $\theta(\mathbf{r})$ space variation smoothness, the A-B interface width is then around 5d, at equilibrium. Therefore, to obtain $I \approx 10 \text{ meV/Å}$, $F_0 d$ must be fixed to 2 meV/Å (or $3.2 \, 10^{-12}$ J/m). As our investigations are focused over the nanometer scale, we choose to set d=1 nm and thus F_0 $=3.2 \text{ mJ/m}^2$. To study larger space scales as it would be suitable for silicon which may exhibit a 100 nm vicinal period,² then it is sufficient to increase the d parameter and to adjust subsequently F_0 , eventually changing the domain boundary energy *I* if necessary.

For simplicity, we choose to neglect the possible γ variations with respect to the crystal surface direction. Such a feature may be yielded from a crystal step anisotropy but here our study is focused on the elastic anisotropy due to the crystal symmetries.

III. SURFACE GREEN FUNCTION

An important stage of this work consists in the calculation of the cubic crystal surface elastic Green function. Let us note that $\mathbf{P}(\mathbf{r})$ is a surface external force at position \mathbf{r} . In Cartesian coordinates, the (001) surface is defined by $(x_3 = 0)$ and $\mathbf{r} = (x_1, x_2)$. The semi-infinite crystal occupies the half space $x_3 \ge 0$. The surface normal is the \mathbf{n} unit vector. The mechanic equilibrium condition at the surface is given by

$$\sigma_{i,i}(\mathbf{r}, x_3 = 0) \quad n_i = P_i(\mathbf{r}), \tag{3.1}$$

where n_j is a **n** component and the summation over subscript *j* is implicit. The crystal bulk stress, $\sigma_{i,j}(\mathbf{r}, x_3)$ is due to the crystal displacements and these quantities are related to each other by the Hooke law $\sigma_{i,j} = \lambda_{i,j,k,l} \partial u_k / \partial x_l$. The forth order tensor $\lambda_{i,j,k,l}$ gives the crystal elastic constants and for a cubic crystal symmetry, this tensor is composed with three nonzero coefficients,¹⁴ namely, $\lambda_{i,i,i,i} = C_{11}$, $\lambda_{i,i,j,j} = C_{12}$, and $\lambda_{i,j,i,j} = \lambda_{i,j,j,i} = C_{44}$. In the case of copper, those coefficients are $C_{11} = 1.683 \, 10^{11} \, \text{J/m}^3$, $C_{12} = 1.221 \, 10^{11} \, \text{J/m}^3$, and $C_{44} = 0.757 \, 10^{11} \, \text{J/m}^3$.

At mechanic equilibrium, the bulk displacements fulfill the Lamé equation

$$\lambda_{i,j,k,l} \frac{\partial^2 u_k}{\partial x_i \partial x_l} = 0.$$
(3.2)

Except constants, the displacement functions are fully determined by the set of equations (3.1), (3.2). As proposed in Ref. 15 for the isotropic case, these equations may be solved by writing the displacements as two-dimensional Fourier transforms

$$u_i(\mathbf{r}, x_3) = \int \int \exp(i\mathbf{Q} \cdot \mathbf{r}) \widetilde{u}_i(\mathbf{Q}, x_3) d\mathbf{Q}.$$
(3.3)

Note that the Fourier components $\tilde{u}_i(\mathbf{Q}, x_3)$ depend on both the wave vector $\mathbf{Q} = (q_1, q_2)$ and x_3 . Once the set of Eqs. (3.1), (3.2) is expressed using Eq. (3.3), the subsequent differential equations involving \tilde{u}_i and their derivatives with respect to x_3 turn out to be linear, so that $\tilde{u}_i(\mathbf{Q} \neq 0, x_3)$ have an exponential dependence on the bulk penetration length x_3 :

$$\widetilde{u}_{j}(\mathbf{Q}\neq 0,x_{3}) = \sum_{l} \beta_{j,l} \exp(-\alpha_{l}x_{3}), \qquad (3.4)$$

where both $\beta_{j,l}$ and α_l depend on the wave vector **Q**. As the whole sample is at rest, the **P** average is assumed to be zero which implies $\tilde{u}_j(\mathbf{0}, x_3) = 0$. The Eq. (3.2) is then reduced to a linear equation $M(\mathbf{Q}, \alpha_l)(\beta_{j,l}) = 0$, where $M(\mathbf{Q}, \alpha_l)$ is the 3×3 matrix

$$M(\mathbf{Q},\alpha_l) = \begin{pmatrix} C_{11}q_1^2 + C_{44}q_2^2 - C_{44}\alpha_l^2 & (C_{12} + C_{44})q_1q_2 & (C_{12} + C_{44})iq_1\alpha_l \\ (C_{12} + C_{44})q_1q_2 & C_{11}q_2^2 + C_{44}q_1^2 - C_{44}\alpha_l^2 & (C_{12} + C_{44})iq_2\alpha_l \\ -(C_{12} + C_{44})iq_1\alpha_l & -(C_{12} + C_{44})iq_2\alpha_l & C_{11}\alpha_l^2 - C_{44}(q_1^2 + q_2^2) \end{pmatrix}.$$
(3.5)

As usual, the nontrivial solutions are such as det[$M(\mathbf{Q}, \alpha_l)$]=0. This yields a third degree polynomial equation for the α_l^2 parameters that we solve numerically. The only three α_l values with positive real parts are physically acceptable. At least one of the α_l is real, the two other roots may be either real or complex conjugated depending on the sign of the χ parameter:

$$\chi = C_{11} - C_{12} - 2C_{44}. \tag{3.6}$$

This combination of the elastic constants is related to the elastic anisotropy of the cubic crystal (see Ref. 14). When $\chi = 0$, the crystal is isotropic and the three α_j are degenerate. This case was addressed in Refs. 14,15. One retains as examples that the copper and gold χ 's are negative ($\chi_{Cu} = -1.0, \chi_{Au} = -0.5$) and the chromium and niobium χ 's are positive ($\chi_{Cr} = +1.8, \chi_{Nb} = +0.5$).

To each α_j corresponds a unique set of three coefficients $(\beta_{j,l})$ that are determined by inverting Eq. (3.1). Noting \tilde{P}_j the Fourier transform of the force component P_j , we write the Fourier counterpart of Eq. (3.1):

$$\sum_{l=(1,2,3)} -\alpha_{l}\beta_{l,1} + iq_{1}\beta_{l,3} = \tilde{P}_{1},$$

$$\sum_{l=(1,2,3)} -\alpha_{l}\beta_{l,2} + iq_{2}\beta_{l,3} = \tilde{P}_{2},$$

$$\sum_{l=(1,2,3)} C_{12}(iq_{1}\beta_{l,1} + iq_{2}\beta_{l,2}) - C_{11}\alpha_{l}\beta_{l,3} = \tilde{P}_{3}.$$
 (3.7)

Combining the first and second rows with the third rows of the matrix $M(\mathbf{Q}, \alpha_l)$ [see Eq. (3.5)], one easily gets both equations

$$\beta_{l,1} = iq_1\beta_{l,3}\Gamma_{l,1}/\alpha_l,$$

$$\beta_{l,2} = iq_2\beta_{l,3}\Gamma_{l,2}/\alpha_l,$$
 (3.8)

where we write

$$\Gamma_{l,1} = \frac{(C_{11} - C_{12} - C_{44})\alpha_l^2 - C_{44}(q_1^2 + q_2^2)}{(C_{11} - C_{12} - C_{44})q_1^2 + C_{44}(q_2^2 - \alpha_l^2)},$$

$$\Gamma_{l,2} = \frac{(C_{11} - C_{12} - C_{44})\alpha_l^2 - C_{44}(q_1^2 + q_2^2)}{(C_{11} - C_{12} - C_{44})q_2^2 + C_{44}(q_1^2 - \alpha_l^2)},$$
(3.9)

provided the denominators of $\Gamma_{i,1}$ and $\Gamma_{i,2}$ are nonzero. If not, which occurs for either $q_1=0$ or $q_2=0$ then it is easy to show that $\beta_{l,3}=0$ and the calculation can be performed as well. In addition let us note

$$\Gamma_{l,3} = -C_{11}\alpha_l - C_{12}(q_1^2\Gamma_{l,1} + q_2^2\Gamma_{l,2})/\alpha_l. \quad (3.10)$$

We rewrite Eq. (3.7) as a linear equation $N_{j,l}\beta_{l,3} = \tilde{P}_j$ where the 3×3 matrix N is defined as follows:

$$N(\mathbf{Q}) = \begin{pmatrix} iq_1(1 - \Gamma_{1,1}) & iq_1(1 - \Gamma_{2,1}) & iq_1(1 - \Gamma_{3,1}) \\ iq_2(1 - \Gamma_{1,2}) & iq_2(1 - \Gamma_{2,2}) & iq_2(1 - \Gamma_{3,2}) \\ \Gamma_{1,3} & \Gamma_{2,3} & \Gamma_{3,3} \end{pmatrix}.$$
(3.11)

Provided $q_1 \neq q_2$, this matrix (3.11) can be inverted which gives the $\beta_{l,3}$'s as linear functions of the $P_j(\mathbf{r})$ Fourier transforms $\beta_{l,3} = N_{l,j}^{-1} \tilde{P}_j$. One must distinguish the case $q_1 = q_2$ for which details are not presented but are simple to deal with. Using Eqs. (3.8), one gets the whole set of $\beta_{l,j}$'s and thus $\tilde{u}_k(\mathbf{Q}, x_3 = 0) = G_{k,j}(\mathbf{Q}) \tilde{P}_j(\mathbf{Q})$ where $G_{k,j}$ is the surface elastic Green function that we write as follows:

$$G_{1,j}(\mathbf{Q}) = iq_1 \sum_{l} N_{l,j}^{-1} \Gamma_{l,1} / \alpha_l,$$

$$G_{2,j}(\mathbf{Q}) = iq_2 \sum_{l} N_{l,j}^{-1} \Gamma_{l,2} / \alpha_l,$$

$$G_{3,j}(\mathbf{Q}) = \sum_{l} N_{l,j}^{-1}.$$
(3.12)

The latest function varies as $1/|\mathbf{Q}|$. The total elastic energy of the system is given at mechanical equilibrium according to Eqs. (3.1), (3.2) by

$$E_{el} = -1/2 \int_{x_3=0} P_i u_i dS \tag{3.13}$$

which gives an analytical expression of E_{el} in the Fourier space:

$$E_{el} = -1/2 \int_{x_3=0} \tilde{P}_i^* [G_{i,l}] \tilde{P}_l d\mathbf{Q}.$$
 (3.14)

The total energy is in fact given by the sum $F = F_{chem} + E_{el}$ [Eqs. (2.1) and (3.2)].

The **P** force is induced by the phases intrinsic stress misfit. Let us note σ_A^0 and σ_B^0 the intrinsic stress tensor of the A and B phases, respectively. At position **r**, the intrinsic stress is thus given by $\sigma^0(\mathbf{r}) = \sigma_B^0 + \delta \sigma^0 \theta(\mathbf{r})$ where we introduce the tensor $\delta \sigma^0 = \sigma_A^0 - \sigma_B^0$. The induced force is simply obtained by deriving σ^0 with respect to the surface coordinates, it gives

$$P_i = \sum_{j=1,2} \,\delta\sigma^0_{ij} \frac{\partial\theta}{\partial x_j}.\tag{3.15}$$

Here we propose to distinguish whether there is a strong directional anisotropy such as for vicinal surface or no broken surface symmetry. In the former case, if a negligible stress variation is induced along a direction which angle with respect to [100] is noted α , the **P** vector must be zero if θ gradient points this direction. The $\delta\sigma^0$ tensor coefficients are developed to satisfy this condition and one may write **P** as follows:

$$P_{i} = \Lambda_{i} \left[\sin(\alpha) \frac{\partial \theta}{\partial x_{1}} - \cos(\alpha) \frac{\partial \theta}{\partial x_{2}} \right], \qquad (3.16)$$

where we introduce three constants $\{\Lambda_i\}$. We restrict our study to a uniaxial stress with only one nonzero Λ_i which is $\Lambda_3 = \Lambda$. According to our tests, nonzero Λ_1 and Λ_2 involve no qualitative change in our results.

If the surface square symmetry is preserved, the $\delta\sigma^0$ tensor is invariant under the square symmetry group operators. So $\delta\sigma_{12}^0 = \delta\sigma_{21}^0$, $\delta\sigma_{31}^0 = \delta\sigma_{32}^0$, and $\delta\sigma_{11}^0 = \delta\sigma_{22}^0$. Again we introduce the Λ parameter setting that $\delta\sigma_{11}^0 = \delta\sigma_{22}^0 = \Lambda$. The surface is assumed to be a perfect plane so the P_3 component is negligible, i.e., $\delta\sigma_{31}^0 = \delta\sigma_{32}^0 = 0$. Nonzero coefficients $\mu = \delta\sigma_{12}^0 = \delta\sigma_{21}^0$ implies a shear that may be induced by the internal structures of the coexistent phases. To a first step, we choose to ignore such a possibility, i.e., $\mu \ll \Lambda$. Thus the force **P** is simply proportional to the θ gradient.

The parameter Λ is a constant that fixes the amplitude of the surface force and it may vary from an experimental case to another. In our model, for each different set of crystal elastic coefficients, Λ is adjusted, i.e., our numerical calculations are performed with $\Lambda = 40 \text{ mJ/m}^2$ (or Λ = 0.25 eV/Å²) for the copper, gold, and niobium crystal cases and $\Lambda = 57 \text{ mJ/m}^2$ (or $\Lambda = 0.35 \text{ eV/Å}^2$) for the chromium case.

IV. KINETICS OF PHASE SEPARATION

To modelize the phase separation kinetics, we use a standard spinodal decomposition theory which is precisely described in Ref. 12 for the three-dimensional case. Extension to the two-dimensional systems is straighfoward. The time evolution of the conserved quantity $\theta(\mathbf{r})$ is then driven by the Cahn-Hilliard equation (also known as model B):

$$\frac{\partial \theta(\mathbf{r},t)}{\partial t} = M \triangle \frac{\delta F}{\delta \theta(\mathbf{r},t)} + \boldsymbol{\epsilon}(\mathbf{r},t), \qquad (4.1)$$

where ϵ is a Langevin stochastic term which simulates thermal fluctuations as proposed by Cook in Ref. 13:

$$\langle \boldsymbol{\epsilon}(\mathbf{r},t) \rangle = 0, \tag{4.2}$$

$$\left\langle \boldsymbol{\epsilon}(\mathbf{r},t)\boldsymbol{\epsilon}(\mathbf{r}',t')\right\rangle = -2k_{B}TM\Delta\,\delta(\mathbf{r}-\mathbf{r}')\,\delta(t-t'). \tag{4.3}$$

In Eq. (4.1), M is a mobility constant which value may be related to the Fick diffusion constant D_{Fick} . As known, for a

given chemical species the parameter D_{Fick} strongly depends on both temperature and coverage (see Ref. 16). For simplicity, we assum here that $D_{\text{Fick}} \approx 10^{-6} \text{ cm}^2/\text{s}$ (see Ref. 16). Neglecting the elastic term E_{el} , a linear expansion of Eq. (4.1) around a uniform coverage $\theta(\mathbf{r})=0.5$ shows that Eq. (4.1) is equivalent to a diffusion equation for short wavelength perturbations (see Ref. 12). Then, we establish $MF_0(d^2\hat{f}/d\theta^2) = D_{\text{Fick}}$ which fixes approximately the mobility since $(d^2\hat{f}/d\theta^2)(\theta_0=0.5)=16$, in the present case. As both D_{Fick} and $d^2\hat{f}/d\theta^2$ strongly decrease with a decreasing θ_0 , the mobility variations with respect to the average coverage cannot be estimated in the general context of our work. For simplicity, we choose to keep it constant, i.e., it is equal to the mobility estimated at half coverage.

Regarding the kinetics of the coarse-grained surface, it is simply derived from the space and time discretization of Eq. (4.1). The previous linear development of Eq. (4.1) is still valid and it comes that the diffusion constant D_{Fick} is rescaled as D_{Fick}/d^2 due to the discretized Laplacian operator. It corresponds to the intuitive idea according to which the larger the coarse graining is, i.e., the larger is *d*, the slower each unit cell of the coarse graining evolves and thus the slower is the global kinetics. One guesses that the larger is a B domain, the longer is the time needed to transform its whole area in a A phase region. Thus we are able to propose a rough estimation of the surface kinetics time scale. Our computer calculations work with a time unit which is equivalent to $\delta t = d^2/D_{\text{Fick}} \approx 10^{-8}$ s. To perform the time integration of Eq. (4.1), we use a time increment of $(10^{-3} \times \delta t)$.

Equation (4.1) is integrated starting from a uniform initial state for which $\theta(\mathbf{r}) = \theta_0$. Spontaneous formation of domains is observed for a coverage θ_0 lying between $0.25 < \theta_0 < 0.75$. This range is imposed by the classical spinodal region defined by $\partial^2 \hat{f} / \partial^2 \theta < 0$ (see Ref. 12). These limits are indeed slightly modified in presence of elastic interactions. The temperature dependency may be activated in the functional F_{chem} , in the diffusion coefficient D_{Fick} , and in the amplitude of the Langevin noise. The external parameters are the material elastic constants that fix χ , the average coverage θ_0 , the surface stress Λ , and the α angle for both isotropic and anisotropic stress cases.

Our numerical findings confirm the analytical results of Refs. 1–3. The phase separation is frustrated by long range interaction due to the crystal surface stress, i.e., the kinetics yields a final steady state. As model input parameters, one may retain that the domain boundary energy is of order I = 10 meV/Å and the stress amplitude is either $\Lambda = 40 \text{ mJ/m}^2$ (or $\Lambda = 0.25 \text{ eV/Å}^2$) for the copper, gold, and niobium crystal cases or $\Lambda = 57 \text{ mJ/m}^2$ (or $\Lambda = 0.35 \text{ eV/Å}^2$) for the chromium case.

In the case of a broken surface symmetry with uniaxial stress, the kinetics yields nonbranched stripes structures with spacial periodicity along a given direction as shown in Fig. 1 (for $\alpha = \pi/4$). The selected period strongly decreases when Λ increases but the formula proposed in Refs. 1,2 which gives the period versus the model parameters, has not been confirmed by our computations because it required many long time computer runs. It is the purpose of a forthcoming



FIG. 1. Phase separation kinetics of a (001) cubic crystal surface $(\chi = \chi_{Au} \approx -0.5)$ with a stress directional anisotropy $\alpha = \pi/4$ (defined in the text). Pictures occurrence time is indicated at the low left hand corners: time unit is around 10 ns. Kinetics is started at initial time t=0 with a uniform surface state. The direction [010] is indicated.

paper. For a coverage $\theta_0 = 0.3$ (see second row in Fig. 1), the time needed to reach the final stripe structure is longer than for $\theta_0 = 0.5$. As soon as the nucleation occurs, it appears some distorted stripes length of which increases with coverage. During the coalescence regime, at $\theta_0 = 0.5$, the stripes branch to each other before their connections move along the stripe sides such that it eliminates the domain which separates the primary stripe couple. The mecanism is very different at low coverage (see second row of Fig. 1), since the short domains move to align themselves with longer stripes.

For a preserved surface symmetry, the kinetics computer calculations results are presented in Fig. 2. In Fig. 2, the time evolution of a half covered surface with and without induced stress is shown. One notes that for a negligible elastic interaction, i.e., $\Lambda = 0$, the domains never stop growing because of the Ostwald's ripening.^{12,18} We had to stop the simulation



FIG. 2. Phase separation kinetics on a (001) cubic crystal surface with negligible stress, i.e., $\Lambda = 0$ (defined in the text) first row, and with $\Lambda = 40 \text{ mJ/m}^2 \chi = \chi_{\text{Cu}} = -1$, second row and $\Lambda = 57 \text{ mJ/m}^2$ and $\chi = \chi_{\text{Cr}} = 1$ third rows.



FIG. 3. Time evolution of the total free energy *F* for negligible stress, i.e., $\Lambda = 0$, and for a stressed surface, i.e., $\Lambda = 40 \text{ mJ/m}^2$ for different coverages: $\theta_0 = 0.25$ (dashed lines) and $\theta_0 = 0.5$ (solid lines). Time unit is 10 ns and the free energy is divided by the constant F_0S (defined in the text).

when the size of the domains is of the order of the whole sample. For $\Lambda \neq 0$, the kinetics is radically different since the system is driven to a "self-organized" final state. Beyong an evolving stage which duration decreases as Λ increases, the surface reaches a steady structure as shown by the two last pictures of the second and the third rows in Fig. 2. The specific size and the distance between the neighboring domains are determined by the value of Λ . The larger is Λ , the smaller are the domain sizes. According to our calculations, the final state is composed of branched stripes with tips as shown by the six last pictures in Fig. 2. As in Fig. 1, the SO appears as soon as the nucleation occurs. At this stage, the domains already arrange along preferential orientations. The structure evolves such that the thinnest branches disappear to the benefit of the thick stripes by first consuming the tips. The stripe orientations depend clearly on the sign of χ : for $\chi > 0$ as for chromium, stripes are along either [110] or $[1\overline{1}0]$ while for $\chi < 0$ as for copper, stripes are along either [100] or [010]. According to the few experimental results about SO of chemisorbed monolayer, the adsorption of nitrogen on a (001) chromium surface has been analyzed with STM by Pinczolits et al. in Ref. 19. Unfortunately, none of their results allows one to conclude about the possible arrangement of nitrogen atoms along [110] or $[1\overline{10}]$ directions. The experimental results obtained in Refs. 6,7 about (001) copper surface are in a good agreement with our calculations.

In Fig. 3, the plot of the ratio $F/(F_0S)$ where S is the total surface size, shows how the free energy gain $\Delta F = F - F_s$ evolves with time. The constant F_s is the free energy of a totally phase separated surface with only two compact domains. The stressed surface decomposition kinetics generates a steady state with lower energy because of a negative elastic energy E_{el} . After a nucleation regime where the total energy decreases strongly as time increases, the energy reaches an asymptotic value. The asymptotic state free energy clearly



FIG. 4. First row: steady states of a (001) cubic crystal surface with $\Lambda = 40 \text{ mJ/m}^2$, $\chi = \chi_{Cu} = -1$, $\theta_0 = 0.21$ (on the left), and $\theta_0 = 0.35$ (on the right). Second row: steady state of a (001) cubic crystal surface with $\Lambda = 57 \text{ mJ/m}^2$, $\chi = \chi_{Cr} = 1$, $\theta_0 = 0.21$ (on the left), and $\theta_0 = 0.35$ (on the right).

depends on the coverage θ_0 and the difference $F(\theta_0=0.5)$ - $F(\theta_0=0.25)$ increases in absolute value with Λ . The same plot with no surface stress shows that the surface evolves in order to reduce its positive domain boundary energy. In such a case, the asymptotic energy difference of two surfaces with different coverages, i.e., $F(\theta_0=0.5)-F(\theta_0=0.25)$ becomes negligible.

At low coverage, as it was predicted in Ref. 3, an assembly of dots appear in the final stage. As for stripes, the arrangement of the dots also depends on the sign of χ . The dots are arranged in an assembly of one-dimensional rafts: the raft orientation is either along [110] or $[1\overline{1}0]$ for $\chi > 0$ (first row of Fig. 4), and either along [100] or [010] for χ <0 (second row of Fig. 4). In Fig. 4, the dots may be square shaped because of the relaxation of their elastic energy and the underlying crystal symmetry. Comparison of the total free energy of a single A phase dot on a B surface, imposing different shapes to this dot, shows that the square shape is optimal, provided the area covered by the dot is larger than a critical size. For smaller sizes, the droplets are disk shaped. This critical size depends on both the elastic constants and Λ . As a consequence, in the first stage of the growth, when the domains have small area, all of them are disk shaped. In the case of copper, before the surface reaches its steady state, most of the islands undergo a shape tranformation passing from a disk shape to a square shape (see the first row of Fig. 4). In such a case, the domain size is clearly larger than the critical shape transformation size. In the left picture of the first row of Fig. 4, performed with chromium elastic coefficients, it is the opposite case, since the disk shape domains never transform. In the right picture of the same row, increasing the surface coverage θ_0 yields larger domains. Some of them overpass the critical size to becomes square with [110] and $[1\overline{1}0]$ oriented sides.



FIG. 5. Profile of the displacement field inside the bulk, for $\Lambda = 40 \text{ mJ/m}^2$, $\chi = \chi_{Cu} = -1$ and $\theta_0 = 0.21$. The arrows' length is proportional to the displacement (multiplied by one hundred). A distance d=1 nm separates neighboring sites of the coarse-grained lattice. The directions [010] and [001] are indicated.

For a coverage lying between $\theta_0 = 0.25$ and half coverage, we find that the nanostructure of the final stage is composed of a mixing between short length stripes and dots (see Fig. 4). So, we deduce that the surface SO shows a crossover with respect to the coverage θ_0 , i.e., going from droplet structure at low coverage to a stripe structure at half coverage, passing through mixed structures.

Figure 5 shows the displacement field involved by the presence of the adsorbate. Nearby the surface, vortices are induced inside the bulk and extend over a depth of few ten nanometers. The maximum amplitude of the displacements is of few 0.1 Å. With both a molecular dynamics study and a linear elasticity analysis of the specific N/Cu (001) case, those structures are also found by Croset *et al.*¹⁷ who first pointed them out to us. Moreover they showed that the bulk displacements due to vortices contribute to x-ray diffraction and they deduce a surface force amplitude measurement 2.2×10^{-9} N at $^{-1}$ (or 0.3 eV/Å²). In our model, the Λ amplitude has the same order of magnitude.

For the (001) copper surface case, an induced shear is tested, i.e., a nonzero $\mu = \delta \sigma_{12}^0 = \delta \sigma_{21}^0$ (defined above). Both the island shape and the arrangement directions are tilted with an angle that depends on the ratio μ/Λ (see Fig. 6).

The previous computations are performed with no thermal fluctuations. Putting on the Langevin noise in Eq. (4.1), with an amplitude which is related to a temperature by Eq. (4.3), we performed the same simulations and obtained similar final states with the same structures. This demonstrates the stability of both the kinetics and the final state. At half coverage and for the copper elastic coefficients, the amplitude of the thermal noise may be increased up to the melting point of the copper metal. Our method does not simulate the desorp-



FIG. 6. Steady state of a (001) cubic crystal surface with $\Lambda = 40 \text{ mJ/m}^2$, $\chi = \chi_{\text{Cu}} = -1$, $\theta_0 = 0.3$, and a shear component $\delta \sigma_{12}^0 = \delta \sigma_{12}^0 = \Lambda/2$.

tion process which could be activated at lower temperature than the crystal melting point. Moreover the white noise amplitude is calculated to satisfy the fluctuation dissipation theorem which is strictly valid around the thermodynamical equilibrium. Nevertheless the use of the Langevin noise allows us to estimate qualitatively the surface structure ability to stand up to thermal fluctuations. Here the nanometric selforganized surface structure are proved to be very strong.

Similar results have been obtained with elastic constants of gold ($\chi < 0$) and with niobium elastic constants ($\chi > 0$) with a smaller amplitude Λ , indeed (see above for details).

V. CONCLUSION AND PERSPECTIVES

In the present paper, we investigated the patterns that are yielded from the (001) cubic crystal surface SO. If the surface square symmetry is preserved, the nanometric morphologies are proved to depend on both the coverage and the elastic coefficients of the crystal. One-dimensional rafts of dots (with disk shape or square shape) occur at low coverage and branched stripes appear at half coverage. Rafts and stripes are aligned with either [100] or [010] for a negative anisotropy factor and with either [110] or $[1\bar{1}0]$ for a positive anisotropy factor. Our computations have been performed with different cubic metal elastic coefficients, e.g., gold, copper, chromium, and niobium. Obviously, it is possible to extend our calculations to the surfaces of either alloys or ionic crystal with cubic symmetry. In case of metal,

this study should be extended to other surface symmetries which exibits interesting properties such as the (111) gold surface, e.g., the gold herringbone reconstruction (see, e.g., Refs. 4,20), and as the (110) copper surface, e.g., the oxide adsorption (see Ref. 21). Then, in the framework of the spin-odal theory for SO, only the elastic Green function should be modified.

In some specific cases, the crystal surface SO displays experimental features that are not included in the present model. The STM analysis of N/Cu(001) performed by Ellmer et al.⁷ demonstrates that the surface morphologies are not equivalent for coverage $\theta_0 = (0.5 + x)$ and for $\theta_0 = (0.5 + x)$ -x), as one would expect from the spinodal decomposition theory we proposed. In addition, attractive interactions between nitrogen islands seems to be dominant at very low coverage. We also note a remarkable result of Fishlock et al.⁸ on Br/Cu(001) where STM analysis shows bromine islands organized in a chessboard at half coverage while we would expect branched stripes. We believe that our simple method can be improved in order to investigate those specific cases even if it is clear that a continuous approach is limited to the mesoscopic scale analysis, e.g., it seems hopeless to include certain atomic scale features as surface dislocations, or adatom missing row.

In the present work, the chemisorbed layers thickness is neglected. As shown in Ref. 22 for the silicon quantum dots, this thickness plays a role in the arrangement of dots but for the moment no idea emerges to tackle this problem on the basis of our approach.

For simplicity, we ignored the possible internal structures of each phases. If not it may involve both anisotropy and shear that are expected to modify the mesoscopic patterns. Moreover, some translational and orientational variants may be yielded from the different way the adatoms arrange on the underlying surface crystal lattice. Those variants can be treated in the present model by developing the free energy density with respect to the so-called long range order parameters. This extension is well known within the phase field method for alloy physics.¹⁰ As a consequence of the coexistent variants, some antiphase boundaries should appear between domains and play a role in the pattern growth. In summary, the model we present here is a promising start to describe and also to predict the nanostructures induced by self-organization of a crystal surface.

- ¹V.I. Marchenko, Sov. Phys. JETP **54**, 605 (1981).
- ²O.L. Alerhand, David Vanderbilt, R.D. Meade, and J.D. Joannopoulos, Phys. Rev. Lett. **61**, 1973 (1988).
- ³Kwok-On Ng and David Vanderbilt, Phys. Rev. B **52**, 2177 (1995).
- ⁴V. Repain, J-M. Berroir, B. Croset, and S. Rousset, Phys. Rev. Lett. 84, 5367 (2000).
- ⁵S. Song, M. Yoon, S.G.J. Mochrie, G.B. Stephenson, and S.T. Milner, Surf. Sci. **372**, 37 (1997).
- ⁶F.M. Leibsle, C.F.J. Flipse, and A.W. Robinson, Phys. Rev. B 47,

15 865 (1993).

- ⁷H. Ellmer, V. Repain, S. Rousset, B. Croset, M. Sotto, and Peter Zeppenfeld, Surf. Sci. 476, 95 (2001).
- ⁸T.W. Fishlock, J.B. Pethica, and R.G. Egdell, Surf. Sci. **445**, 47 (2000).
- ⁹M. Sotto and B. Croset, Surf. Sci. 461, 78 (2000).
- ¹⁰A. G. Khachaturyan, *Theory of Structural Transformations in Solids* (Wiley-Interscience, New-York, 1983), Chap. 7; L.Q. Chen and A.G. Khachaturyan, Phys. Rev. Lett. **70**, 1477 (1993); Yunzhi Wang, L. Q. Chen, and A. G. Khachaturyan, in *Com*

puter in Material Science, NATO ASI Series No. 308, edited by H. O. Kirchner, L. P. Kubin, and V. Pontikis (Plenum, New York, 1996), p. 325.

- ¹¹K. Kassner and C. Misbah, Europhys. Lett. **46**, 217 (1999); A. Karma, Y.H. Lee, and M. Plapp, Phys. Rev. E **61**, 3996 (2000).
- ¹²J. D. Gunton, M. M. San Miguel, and P. S. Sahni, *Phase Transitions and Critical Phenomena*, edited by C. Domb and J. L. Lebowitz (Academic, New York, 1983), p. 269.
- ¹³H.E. Cook, Acta Metall. **18**, 297 (1970).
- ¹⁴E. M. Lifshitz and L. D. Landau, *Theory of Elasticity*, Vol. 7 of *Theoretical Physics*, 3rd ed. (Butterworth-Heinemann, London, 1995), pp. 54-55.
- ¹⁵J. Villain and A. Pimpinelli, *Physique de la Croissance Cristalline*

(Eyrolles, Saclay, 1995), Appendix I.

- ¹⁶A. Zangwill, *Physics at Surfaces* (Cambridge University Press, Cambridge 1988), p. 380.
- ¹⁷B. Croset *et al.*, Phys. Rev. Lett. (to be published).
- ¹⁸I.M. Lifshitz and V.V. Slyozov, J. Phys. Chem. Solids **19**, 35 (1961); C. Wagner, Z. Elektrochem. **65**, 581 (1961).
- ¹⁹M. Pinczolits et al., Surf. Sci. 396, 78 (1998).
- ²⁰S. Narasimhan and D. Vanderbilt, Phys. Rev. Lett. **69**, 1564 (1992).
- ²¹Klaus Kern et al., Phys. Rev. Lett. 67, 855 (1991).
- ²²J. Tersoff, C. Teichert, and M.G. Lagally, Phys. Rev. Lett. **76**, 1675 (1996).