

High-energy x-ray reflectivity of buried interfaces created by wafer bonding

F. Rieutord,¹ J. Eymery,¹ F. Fournel,¹ D. Buttard,¹ R. Oeser,¹ O. Plantevin,¹ H. Moriceau,² and B. Aspar²
¹CEA-Département de Recherche Fondamentale Sur la Matière Condensée, 17 rue des Martyrs, 38054 Grenoble Cedex 9, France
²CEA-LETI, Département de Microtechnologies, 17 Rue des Martyrs 38054 Grenoble Cedex 9, France

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The bonding interfaces separating two silicon wafers assembled for making a silicon-on-insulator system are studied using high-resolution high-energy x-ray reflectivity. The evolution of the bond structure upon annealing is investigated, *in situ*. These data directly exhibit water removal and oxide layer structural changes, throughout the temperature sequence.

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Wafer bonding is an important technique for joining materials in microelectronics and micromechanics. It relies on the fact that two surfaces will stick together when brought into contact if they are sufficiently clean, mirror polished, flat, and dust free. Applications include silicon-on-insulator substrates fabrication, which is of large use in microelectronics for low-voltage or high-temperature device manufacturing.¹ The large experience gained in the design of artificial bonds between two materials also allows other applications to be envisioned for this interface fabrication technique.

However, we are still lacking methods to characterize these interfaces buried between thick materials. Standard transmission x-ray diffraction techniques can be used,² but interface-specific techniques are scarce due to the fact that most radiation used in surface studies (e.g., low- and high-energy electrons, light or conventional x rays) are unable to cross millimeters of solid material since they interact too strongly with the embedding bulk material. Scanning probe techniques [scanning tunnel microscopy (AFM)] can at best be used on cross sections of the material.

The method we shall use here meets the seemingly contradictory requirements of a large bulk penetration and a high interface sensitivity. It is based on the use of high-energy x-rays under grazing incidence conditions. Due to their high energy, the x-rays can cross several millimeters of bulk material without a too large absorption while the use of shallow angles allows an increased sensitivity to the interface. We shall concentrate in this paper on high-energy x-ray reflectivity (HEXRR) but off-specular scattering and grazing incidence scattering have been performed in the same conditions, with the same interest, and data will be published in forthcoming papers. To our knowledge, this is the first study of a deeply buried solid interface using this technique. With this technique, we could obtain detailed structural information on the bonding interface, giving access to the physical process occurring at this interface.

The geometry of the experiment is shown in Fig. 1. The samples used were made by bonding a wafer covered by a thin native oxide layer to a wafer on which 400 or 500 Å of thermal oxide were grown. The x rays enter the sample from one edge, are reflected by the interfaces, and exit on the other edge. The length of the sample should be of the order of, or smaller than the attenuation length of the material. However the use of high-energy x rays causes an angular compression

of the reflectivity features and imposes a higher collimation of the beam. In these experiments requiring the crossing of silicon or SiO₂ material, we used an x-rays energy of 27 keV and a sample length L of 4 mm, which is two times the attenuation length for silicon at this energy. The beam size at sample position was 50 μm so that at very small angles the beam was larger than the interface cross section h [$h = L \sin \theta$, θ being the incident angle (see Fig. 1)]. A correction for this geometrical factor was applied. It should be noted that in all the experiments, the distance between the interface and the external surfaces is 0.5 mm (wafer thickness) so that, in the angular range used, the beam never hits the outer surfaces. This is actually the main asset of the technique: as for standard optical reflectivity, reflection comes from index changes (i.e., for x rays, electron density changes). As the radiation does not reach the outer interfaces, the reflection signal is merely due to the change of density in the interfacial zone and the signal is free from any contribution of the strong density gradients at the air (vacuum)-material interfaces. In this respect, a zero background technique as a perfect interface between similar material would just produce no signal: the interfacial zone under study behaves like a free-standing film.³ It should be noted that the incidence of the beam to the sample edge is close to 90°, so that reflection or refraction effects at the entrance or the exit of the beam are negligible. Another point that is worth noting is that this geometry allows reflectivity at both negative and positive incidences to be performed.

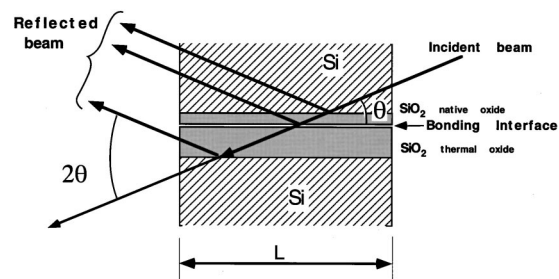


FIG. 1. Sketch of the HEXRR experiment: the beam enters from the side of one of the wafers, is reflected by the interfacial area density gradients, and exits on the other side. Absorption through the system is taken into account via normalization to the transmitted beam.

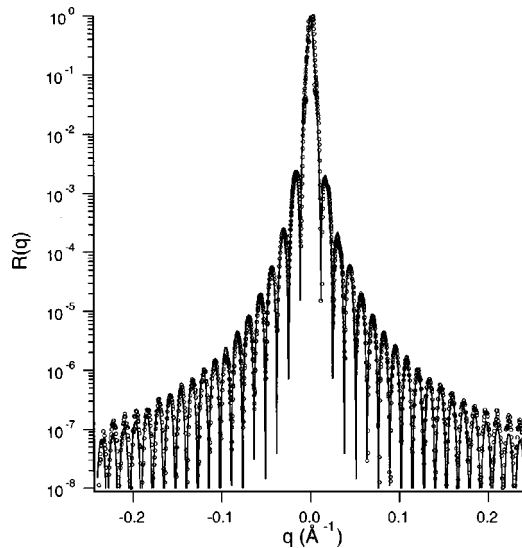


FIG. 2. Reflection curve of a buried silicon oxide film obtained by hydrophilic bonding of a native oxide and thermal oxide covered wafers. The solid line is a fit to data points using a simplest two-parameter box model ($d=499 \text{ \AA}$, $\Delta\rho=-2.5\%$). Data at both positive and negative incidences are shown.

The experiments were performed on the bending magnet beamline of the Collaborative Research Group on Interfaces (CRG-IF) at the European Synchrotron Radiation Facility. This beamline gives a monochromatic beam of the order of 10^{10} ph/s in a 0.3×0.3 mm focal spot with a convergence of $1 \text{ mrad} \times 0.1 \text{ mrad}$. The reflectivity was taken using standard $\theta/2\theta$ scans. Alignment was checked periodically by performing rocking scans (θ alone). At very small angle the detector is very close to the transmitted beam and some background is produced. The background (off-specular scattering) was measured by performing detector scans and keeping the sample at $\theta=0$ and subtracted off the $\theta/2\theta$ signal. The level of this background is negligible as soon as angles not too close to the direct beam are considered. Normalization was performed using the transmitted beam, which includes both incident beam intensity and transmission coefficient.

We shall first describe the results obtained on a system of a thin silicon oxide film embedded between two 0.5 mm silicon wafers. This system was obtained by bonding two silicon wafers, one covered with native oxide (thickness $\sim 7\text{--}8 \text{ \AA}$) and one covered with thermally grown oxide (thickness $\sim 500 \text{ \AA}$). High-temperature ($T > 1000 \text{ }^\circ\text{C}$) annealing was then performed, *ex-situ*, during 3 h. Reflectivity obtained for both negative and positive small angles is shown in Fig. 2. As it can be seen, many well-contrasted fringes are observed, a fact which is not surprising as the two boundary media are identical and very flat.

In the angular range shown, the data can be fit using a simple two-parameter model. The thickness is measured from the fringe spacing and the fringe intensity is related to the refractive index gradient at both interfaces. For x rays, this is just proportional to the electron-density gradients: far from any absorption edge (our case), the refractive index is related to the electron density ρ_{el} by $n=1-\lambda^2/(2\pi)r_0\rho_{\text{el}}$, where r_0 is the classical electron radius. The simplest model

to express the reflectivity as a function of these parameters is based on the assumption that each interface in the profile ‘‘sees’’ the same incident wave and reflects with a phase depending only on its position along z . This is a kinematical approximation and, in this approximation, the reflectivity is the Fourier transform of the index gradient profile

$$R(q) = \frac{64\pi^4}{\lambda^4 q^4} \left| \int_{-\infty}^{+\infty} \frac{dn(z)}{dz} \exp(iqz) dz \right|^2, \quad (1)$$

where n is the refractive index, λ the wavelength, and $q = 4\pi/\lambda \sin \theta$ is the wave-vector transfer in the geometry. The front factor in $1/q^4$ corresponds to the reflection of a perfect dioptr separating two media with a unity index change (fresnel reflectivity). When this dependence is removed by considering $q^4 R(q)$, this formula gives, for a slab delimited by two boundaries, a sinusoidal dependence with unity contrast:

$$q^4 R(q) = (4\pi r_0)^2 |\Delta\rho [-1 + \exp(iqd)]|^2. \quad (2)$$

The fit to the data yields $|\Delta n| = 1.910^{-8}$ ($|\Delta\rho| = 0.02 \text{ \AA}^{-3}$) corresponding to a density difference of 2.5% between silicon and silicon oxide. The thickness measured on this sample is 499 \AA , very close to the value of 500 \AA expected from the growth kinetics of the thermal oxide.

Telling if the density of the oxide is 2.5% higher or lower than that of silicon is actually not obvious as both values could be physically relevant for this system, which has experienced a high-temperature annealing. Equations (1) or (2) would give a similar result in both cases. However, at very small angles, kinematical approximation breaks down and the actual values of the reflectivities for the two density profiles are very different. They can be calculated using another formalism based on the optical theory of x rays. In this model, one discretizes the profile into a stack of thin slabs of constant index. The reflectivity is then recursively computed from the contribution of each slab.^{4,5} As this optical model takes into account the full set of transmitted and reflected waves at each interface, it corresponds to the dynamical treatment of x-ray diffraction and is correct whatever the angle. If silica were 2.5% denser than silicon, the index change at the silicon-silicon oxide interface would be negative so that total ‘‘external reflection’’ would be expected below an angle given by Snell Descartes law $\theta_c = \sqrt{2|\Delta n|}$. Although located at very small angles and close to the direct transmitted beam, reflection in this angular range is measurable when background subtraction is performed. Comparison of experimental data to the curves calculated using a dynamical theory (optical) model clearly indicates that the situation is that of a positive index gradient, with no total external reflection.

The mass density of the silicon oxide film is thus 2.27 g/cm^3 (electron density 0.683 \AA^{-3}), i.e., less dense than silicon ($\rho_{\text{Si}} = 2.33 \text{ g/cm}^3$). This value is to be compared to the density of tridymite, a variety of silica that forms in a temperature range consistent with the annealing temperature. From the data of Fig. 2 alone, it is not possible to determine a value for the interfacial roughness l_r ($l_r=0$ was used in the fits without any incidence on the results). A maximum value

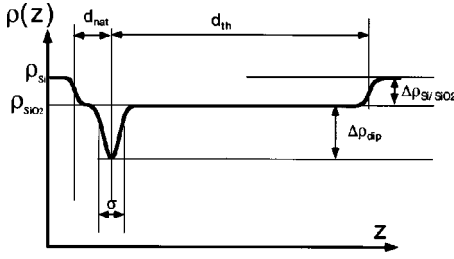


FIG. 3. Electron-density model used to fit the temperature-dependent measurements. The density dip at the bonding interface is modeled by a Gaussian well of width σ and depth $\Delta\rho_{\text{dip}}$.

can be given ($l_r < 2 \text{ \AA}$) and more details could be obtained only if data on a larger q range were considered.

One of the advantages of the technique of HEXRR is that it can be performed *in-situ* during the annealing that is used to increase the bond strength. We have performed this study using a Meca2000 UHV-compatible furnace installed in a custom-made vacuum chamber equipped with beryllium windows (vacuum was of the order of 10^{-6} mbar during experiment). The sample considered was similar to the previous one, i.e., Si/SiO₂ (native 7–8 Å) bound to a SiO₂ (thermal, 400 Å)/Si wafer. The reflection curve at ambient temperature shows two main features.

First, the reflection level is much higher than for the annealed system, typically by two orders of magnitude (see Fig. 4). As the intensity reflected by an interface is proportional to the magnitude of the electron-density gradient at this interface [see Eq. (1) or (2)], this indicates that large density gradients are now present in the system. As the silicon-silicon oxide interfaces have only weak density changes, these large index gradients are due to the lack of electrons at the bonding interface.

Second, fringes are still visible but their contrast C [defined as $C = (I_{\text{max}} - I_{\text{min}})/(I_{\text{max}} + I_{\text{min}})$] is low. This again is due to the fact that these fringes are due to interference between the large density gradients at the interface and the small ones at the Si/SiO₂ interfaces. The interference between the native oxide-silicon interface and the bond interface is not visible: as these two interfaces are close to each other, they would give in q space a long period fringe that is not separable from the contribution due to the shape of the interfacial electron density gap.

To model quantitatively the reflection data, we used, as shown in Fig. 3, an electron-density profile where the dip at the bonding interface is represented by a ‘‘Gaussian well’’ of width σ and depth $\Delta\rho_{\text{dip}}$. $\rho(z) = -\Delta\rho_{\text{dip}} \exp(-z^2/2\sigma^2)$. With the notation of Fig. 3, the Fourier transform (1) becomes

$$q^4 R(q) = (4\pi r_0)^2 |\Delta\rho_{\text{SiO}_2/\text{Si}} \{-1 + \exp[iq(d_{\text{nat}} + d_{\text{th}})]\} - \Delta\rho_{\text{dip}} iq \sqrt{2\pi} \sigma \exp(-q^2 \sigma^2/2) \exp(iq d_{\text{nat}})|^2, \quad (3)$$

where d_{nat} and d_{th} are the thicknesses of the native and thermal oxide layers. This expression shows directly how to interpret the experimental data of Fig. 4. The influence of the different parameters can be best understood using an expan-

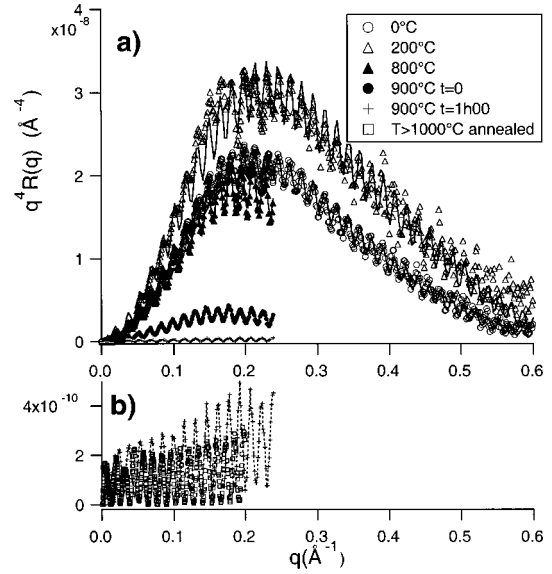


FIG. 4. (a) Normalized $q^4 R(q)$ reflection curves of a wafer bonding assembly as a function of annealing temperature. Very large intensity variations are seen due to the evolution of the interfacial bond gap structure. (b) enlarged part (scale $\times 100$) of $T = 900^\circ\text{C}$, 1h00 data together with high-temperature annealed system (squares) [i.e., data of Fig. 2, in $q^4 R(q)$ representation].

sion of Eq. (3): When the density gap $\Delta\rho_{\text{dip}}$ at the bonding interface is large, the terms including it dominate those involving the small SiO₂/Si density gradients ($\Delta\rho_{\text{SiO}_2/\text{Si}}$) and Eq. (3) approximates under the form

$$q^4 R(q)/(4\pi r_0)^2 = (\Delta\rho_{\text{dip}} \sigma)^2 2\pi q^2 \exp(-q^2 \sigma^2) + 2(\Delta\rho_{\text{dip}} \sigma) \Delta\rho_{\text{SiO}_2/\text{Si}} q \sqrt{2\pi} \times \exp(-q^2 \sigma^2/2) \sin(q d_{\text{th}}). \quad (4)$$

In this expression, the first term on the right-hand side reproduces the hump of the curves in the $q^4 R(q)$ representation, while the second term is the interference term responsible for the fringes.

This expression shows that the depth and width of the electron deficit at the bonding interface are readily determined from the maximum amplitude and width of the reflection curves: the $q^4 R(q)$ signal displays a hump, which is the Fourier transform of the dip profile. As the bond gap fills in, this hump disappears.

Using the second term in Eq. (4), one can work out an expression for the fringes contrast:

$$C = \frac{2\Delta\rho_{\text{SiO}_2/\text{Si}}}{\Delta\rho_{\text{dip}} \sigma \sqrt{2\pi} q \exp(-q^2 \sigma^2)} = \frac{2\Delta\rho_{\text{SiO}_2/\text{Si}}}{\sqrt{\langle I(q) \rangle}}, \quad (5)$$

where $\langle I(q) \rangle$ is the reflection intensity averaged over a fringe period. From this expression one can readily obtain the silicon oxide silicon density gradient.

Data were collected at ambient, 200, 800, and 900 °C (see Fig. 4). Due to the kinetics at high temperature only a limited part of q range could be scanned. Curves 900-1 and 900-2

TABLE I. Fit parameters used to describe the reflectivity data of Fig. 4 using the model of Eq. (3).

Temperature	Thermal oxide thickness d_{th} (Å)	Surface area S of the dip (Å ⁻²)	Width of the dip σ (Å)	Depth of the dip $\Delta\rho_{dip}$ (Å ⁻³)	Thermal oxide density ρ_{SiO_2} (Å ⁻³)
Ambient	388	2.6	4.4	0.23	0.683
200 °C	387	2.9	4.1	0.28	0.683
800 °C	395	2.5	5.5	0.18	0.662
900 °C $t=0$	398	1.0	6	0.07	0.662
900 °C $t=1h00$	404 ^a	0.15	7	0.01	0.662

^aThis value includes the native oxide thickness as the bond gap is very weak for this last set of data.

refers to two sets of data taken at 900 °C with a one-hour time interval. The parameters obtained from fit of Eq. (3) to the data are shown in Table I.

At ambient temperature, the surface area S of the density “well,” which is directly related to both the level of reflection and the width of the hump, is 2.6 \AA^{-2} . This parameter is actually the first data that can be extracted from the reflection data (e.g., from low q data), as it is the first moment of the density distribution at the interface.⁶ The width parameter σ of the dip, which is the second moment of the electron-density distribution, is 4.2 \AA . This corresponds to a full width at half maximum of $w_{1/2} = \sigma \sqrt{(2 \text{Ln}2)} = 10 \text{ \AA}$. The depth of electron-density deficit is obtained from the height of the hump [first term in Eq. (4) for $q\sigma=1$] or through $\Delta\rho_{dip} = S/\sqrt{(2\pi)}/\sigma$. The electron density in the well ($\rho_{el} = \rho_{SiO_2} - \Delta\rho_{dip}$) can be determined from this data. The value found is consistent with the idea that the bond gap may actually be filled with water, assuming water keeps its bulk electron density. Our data indicate that 3–4 monolayers of water are present at the interface and this should be viewed as a maximum since part of the gap may just be empty, thus reducing the amount of water necessary to achieve the measured electron deficit. This observation is consistent with the values observed using infrared absorption spectroscopy.^{7–9} We believe this number of layers is intimately linked to the presence of bound water layers at the surfaces of hydrophilic wafers and the physics of dewetting involved during the propagation of the bond wave.

When temperature is raised to 200 °C, the reflected intensity is higher than at ambient temperature (by 50%) which means that the deficit of electron density has increased (by 25%). From the measure of both the dip surface area change and the width change, one can estimate the amount of water that has left the interface to one monolayer, with a remaining quantity of 2–3 monolayers. This partial removal of water can be associated to the observed increase in surface energy observed in this temperature range (typically from 0.15 J/m^2 to 1.0 J/m^2) due to formation of siloxane bonds.

At 800 °C, the reflected intensity starts to decrease and comes back below the ambient temperature data. The overall value of the electron-density deficit at the interface is still important, but the width starts to increase, indicating that the gap gets shallower. Note that the range of our data is smaller as we limited the duration of the experiment to a few minutes to limit the evolution of the material during the annealing. Indeed, this temperature of 800 °C is known to correspond to

the beginning of SiO₂ chemical modification. The interesting point is that the fringe spacing indicates that the thickness of the thermal oxide has increased by 7–8 Å. Correlatively, the fringe amplitudes indicate that the density gradient between silicon and silicon oxide has increased. The fit to the data indicate a decrease of the density of silicon oxide by another 2% to a value of $\rho_{SiO_2} = 2.20 \text{ g/cm}^3$ (0.662 \AA^{-3}), which is the standard for amorphous silica. The thermal expansion of the oxide being low ($<0.5 \text{ \AA}$), this increase in thickness correlated to a reduction in density is certainly due to some phase transformation taking place in the thermal oxide under heating.

The kinetic evolution of the bond gap is most visible on the data taken at 900 °C. On the first curve taken 2 hours after the 800 °C data, the gap has reduced by a factor of 4 in surface area. One hour later, it has almost disappeared and the data show unity contrast fringes as observed on the high-temperature annealed system described in the first part of the paper (Fig. 2). This evolution is in-line with the possibility for the oxide to flow at these high temperatures, thus allowing a fill up of the gap between the two oxide surfaces. The thickness deduced from fringe spacing on the last curve has now increased by another 9 Å compared to the 800 °C value. This increase is the result of two factors. First, the incorporation of the native oxide thickness. The fringe spacing no longer yields the distance between the dip and the thermal oxide-silicon interface but the distance between the two Si/SiO₂ interfaces. This distance includes the thickness of the native oxide [i.e., approximation (4) is no longer valid in this case]. Second, the gap closing. Assuming no material is lost, the closure of the gap induces a reduction of the distance between the two Si/SiO₂ interfaces by about 4 Å, obtained from the ratio of the surface area of the electron density gap (2.5 \AA^{-2}) to the silicon oxide electron density (0.66 \AA^{-3}). These data means that the thickness of the native oxide was close to 13 Å. It should be noted that as long as the electron-density gap at the bonding interface is large, the thickness of the native oxide could not be extracted from the data and this parameter value was kept fixed to a value of 8 Å.

Finally it is interesting to compare the last data of the annealing sequence to the previously described fully annealed system, studied *ex-situ* (Fig. 4b). Apart from the difference in fringe period due to the different oxide layer thickness used in the *ex-situ* sample (500 Å instead of 400 Å), the fringe amplitude and contrast are similar. This shows that the

electron density dip was almost filled up after 1 h at this temperature. Only the slight increase in intensities of fringe minima and maxima for the 900 °C sample is indicative of a remaining small electron deficit at the bonding interface, a deficit that is no longer measurable on the ex-situ annealed sample.

High-energy x-ray reflectivity is one of the few techniques available to characterize bonding interfaces buried between thick materials. It allows a direct determination of the electron-density profile of the interface with an angstrom resolution. The technique allows a direct control of molecu-

lar reorganization taking place during the formation of the bond interface. It readily gives the closing temperature of the gap and allows one to follow the associated growth and density changes of the oxide layers. We have seen here some examples on the most classical bonding procedures used industrially (room-temperature bonding in air, followed by annealing) but the method is suited to test the influence of various parameters on the bonding (e.g., chemical and physical surface treatment, surface roughness, ion implantation). Other systems and bonds between other materials (e.g., metallic films) can also be studied by this technique.

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