Electron doping effects in conducting Sr₂FeMoO₆

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Effects of the electron doping on the magnetic and electronic properties have been investigated in the conducting double perovskite Sr_2FeMoO_6 . Judging from the enhanced moment μ_{Mo} at the Mo site, the doped electrons are considered to occupy mainly the down-spin Mo4*d* band. Filling dependence of the saturation magnetization M_s , the spin-wave stiffness coefficient D_s and the γ coefficient are discussed on the basis of a LDA+U band calculation. We further discuss possible origins for the magnetism.

I. INTRODUCTION

The carrier doping by chemical substitution is a powerful tool for investigation of strongly correlated electron systems, especially the perovskitelike 3d and 4d transition-metal oxides. Most of the 3d transition-metal oxides with integer filling, e.g., LaMnO₃ and LaCoO₃, are insulating due to the strong on-site Coulomb repulsion *U*. With substitution of the divalent Sr^{2+} ions for the trivalent La^{3+} ion, however, these compounds become ferromagnetic metal due to the double exchange mechanism.¹ Even in the 4*d*-electron systems with reduced *U* value, notable physical properties have been reported.

Recently, ordered double-perovskite molybdenum oxides²⁻⁶ $A_2 M MoO_6$ (A is a rare-earth metal and M is a 3d transition metal) begin to attract current interest of material scientists in the applicational point of view. Actually, Kobayashi et al.⁷ and Kim et al.⁸ reported room-temperature magnetoresistance in Sr₂FeMoO₆ due to the intergrain tunneling.9 Among the double-perovskite molybdenum oxides, Sr₂FeMoO₆ is most extensively studied, because high quality single crystals can be obtained by the floating-zone method.^{10–12} Sr₂FeMoO₆ is ferrimagnetic and metallic with critical temperature $T_c \approx 400$ K. On the basis of a localdensity approximation (LDA) band calculation,⁷ the conduction band is dominated by the Mo4d level, while Fe3*d*-electrons behave as local spins. The investigation on the double-perovskite molybdenum oxides, however, is only at the starting point. We should clarify the interrelation between the local spins and the conductive 4d electrons to understand the physics (and to optimize the functionality) of the system.

In this paper, we have investigated the electron doping

effects on the magnetic and electronic properties in the conducting double perovskite $(Sr_{2-x}La_x)FeMoO_6$. By means of the neutron powder-diffraction measurements, we have determined the respective magnetic moments, i.e., μ_{Fe} and μ_{M_0} , of $(Sr_{2-x}La_x)$ FeMoO₆. The enhancement of the μ_{M_0} value with doping indicates that the doped electrons occupy mainly the Mo4 d_{\perp} band. With increase of the doping level x, the saturated magnetization M_s , and spin-wave stiffness coefficient D_s decrease, while the γ coefficient (electron contribution in the specific heat) increases. We observed a serious discrepancy between the rigid band model and the experimental results at larger-x region, even though this picture qualitatively reproduces the overall x dependence of M_s , α , and γ . The discrepancy is ascribed to the mixing of the Fe3d_{\perp} state into the conduction band. On the basis of the above arguments, we will discuss possible origins for the magnetism of the conducting double-perovskite molybdenum oxides.

II. EXPERIMENT

A. Crystal growth

A series of crystals, $(Sr_{2-x}La_x)FeMOO_6$, were melt grown by the floating-zone furnace.¹¹ A stoichiometric mixture of commercial SrCO₃, La₂O₃, Fe₂O₃, Mo, and MoO₃ was well ground, and pressed into a rod with a size of 5 mm $\phi \times 100$ mm and sintered at 1200 °C for 2 h in a flow of Ar gas. The crystal is grown at a feeding speed of ~20 mm in an Ar atmosphere. Black and shiny crystals, typically 4–5 mm in diameter and ~20 mm in length, were obtained. The neutron-diffraction experiment has revealed that thus obtained crystals are single domain. The actual chemical composition of the melt grown

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 $R_{\rm I}~(\%)$ a (Å) c (Å) x $d_{\rm Fe-O}/d_{\rm Fe-Fe}$ $d_{\rm Fe-O}/d_{\rm Fe-Fe}$ R_{wp} (%) 0.00 5.5711(2)0.244(1)7.8996(4)0.238(3)2.777.89 0.10 5.5714(1)0.246(2)7.9087(3) 0.244(4)2.63 7.05 0.20 3.09 5.5729(1)0.243(3)7.9071(2)0.245(5)5.85 0.30 5.5751(2)0.241(3)7.9078(3) 0.241(6)3.50 4.45

TABLE I. Lattice constants and relative length of the Fe-O bond $(d_{\text{Fe-O}}/d_{\text{Fe-Fe}})$ of $(\text{Sr}_{2-x}\text{La}_x)$ FeMoO₆ at 300 K. The crystal symmetry is tetragonal (*I*4/*mmm*; Z=2).

 $(Sr_{1.7}La_{0.3})$ FeMoO₆ crystal was determined by the inductively coupled plasma (ICP) method, and is almost the same as the ideal values; Sr:La:Fe:Mo=1.692 (1.7: prepared value):0.316 (0.3):1.024 (1.0):0.972 (1.0).

B. Sample characterization

In order to characterize thus obtained crystals, synchrotron radiation x-ray powder experiments were carried out at SPring-8 BL02B2 to obtain the powder patterns with good counting statistics and high angular resolution. The inner region of the crystal ingot was crushed into a fine powder and were sealed in a 0.2 mm ϕ quartz capillary. Precipitation method was adopted¹³ in order to get a fine powder, which gives a homogeneous intensity distribution in a Debye-Scherrer powder ring. The wavelength of the incident x ray was ≈ 0.5 Å, and exposure time was for 5–10 min.

The crystal symmetry at room temperature ($\leq T_c$) is tetragonal (*I4/mmm*; *Z*=2). We have analyzed thus obtained x-ray patterns with RIETAN-97 β program,¹⁴ and listed the lattice parameters in Table I. We found no impurity peaks in the x-ray patterns. The final refinements are satisfactory, in which R_{wp} and R_I (reliable factor based on the integrated intensity) are fairly reduced ($R_{wp}=2.7-3.5\%$, R_I =4.5-7.9%). We further have estimated the mixing ratio *s* of the *B* site with putting a virtual atom (1-*s*)Fe³⁺ +*s*Mo⁵⁺ [*s*Fe³⁺+(1-*s*)Mo⁵⁺] at [0,0,0] ([0,0,1/2]). The *s* value is nearly independent of *x* except for *x*=0.3: *s* =16(1)% at *x*=0.0,15(1)% at 0.1, 14(1)% at 0.2, and 21(2)% at 0.3.

C. Resistivity

For four-probe resistivity measurements, the sample was cut into a rectangular shape, typically of $2 \times 1 \times 1$ mm³, and electrical contacts were made with a heat-treatment-type silver paint. Heat treatment has been done in a flow of Ar. Figure 1 shows temperature variation of resistivity ρ of $(Sr_{2-x}La_x)FeMoO_6$. Overall feature of the ρ - T^2 curve of the doped crystals is essentially the same as that of the parent Sr₂FeMoO₆ (thick curve). The resistivity well scales T^2 in the temperature range from ~50 to ~350 K. Such a T^2 law has been ascribed to the electron-electron scattering or the one-magnon scattering.^{11,15}

III. RESULTS AND DISCUSSION

A. Doping effects on the magnetic structure

Figure 2 shows the magnetization curve of $(Sr_{2-x}La_x)FeMoO_6$ measured at 10 K. The *M* value of Sr_2FeMoO_6 rapidly increases with an external field, and then

saturates at $M_s \approx 3.3 \mu_B$ per an Fe site. The magnitude of M_s systematically decreases with increase of x, and becomes $\approx 2.4 \mu_B$ at x = 0.3. Here, we should be careful on the effects of the *B*-site disorder, which suppresses the absolute magnitude of M_s .¹⁰ Since the mixing ratio *s* is nearly *x* independent, however, the doping-induced suppression of the net magnetization is intrinsic. To determine the critical temperature T_c for the magnetic transition, we further have measured temperature variation of the ac susceptibility.¹⁶ In the inset of Fig. 3, we plotted T_c against *x*. T_c is nearly independent of *x*, and is ≈ 400 K.

In order to determine the magnetic moments for the respective sites, i.e., $\mu_{\rm Fe}$ and $\mu_{\rm Mo}$, we have measured neutron powder-diffraction patterns of Sr₂FeMoO₆ and $(Sr_{1,7}La_{0,3})$ FeMoO₆ with the Kinken powder diffractometer for high efficiency and high-resolution measurements, HERMES, installed at the JRR-3M reactor in Japan Atomic Energy Research Institute, Tokai, Japan.¹⁷ Neutrons with wavelength 1.819 Å were obtained by the (331) reflection of Ge monochromater, and 12'-B-Sample-18' collimation. Melt-grown crystal ingots (~ 5 g) were crushed into a fine powder and were sealed in a vanadium capsule with helium gas, and mounted at the cold head of the closed-cycle He-gas refrigerator.

Figure 3 shows neutron powder patterns of (a) Sr_2FeMoO_6 and (b) $(Sr_{1.7}La_{0.3})FeMoO_6$ at 15 K ($\leq T_c$). Magnitudes of the respective moments are determined by the

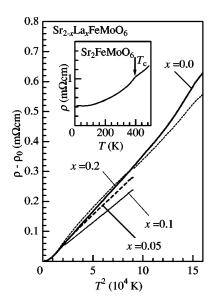


FIG. 1. Temperature variation of resistivity ρ of $(Sr_{2-x}La_x)FeMoO_6$. Note that residual resistivity ρ_0 was subtracted. Inset shows the ρ -*T* curve of Sr₂FeMoO₆. *T*_c is the critical temperature for the magnetic transition.

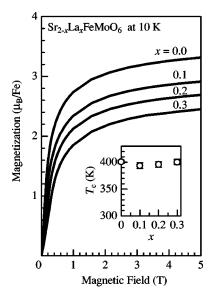


FIG. 2. Magnetization curve at 10 K of $(Sr_{2-x}La_x)$ FeMoO₆. The inset shows doping dependence of the critical temperature T_c for the magnetic transition.

Rietveld fitting (RIETAN-97 β , Ref. 14), in which we assume a collinear magnetic structure with spin direction along c.¹⁸ Several impurity peaks, which are inevitably introduced from the skin region of the crystal ingot, were observed and were removed from the pattern. The obtained moments are $\mu_{\rm Fe}$ =4.2(2) $\mu_{\rm B}$ and $\mu_{\rm Mo}$ =-0.5(2) $\mu_{\rm B}$ at x=0.0, and $\mu_{\rm Fe}$

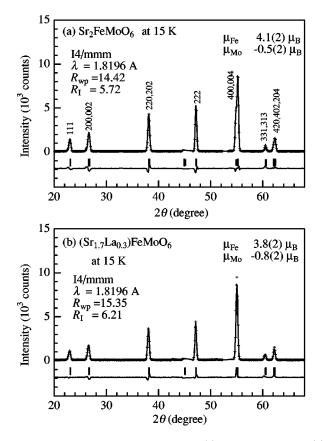


FIG. 3. Neutron powder profile of (a) Sr_2FeMoO_6 and (b) $(Sr_{1.7}La_{0.3})_2FeMoO_6$ at 15 K. Crystal structures of both the compounds are tetragonal (*I4/mmm*; *Z*=2). Solid curve is the results of the Rietveld refinement.

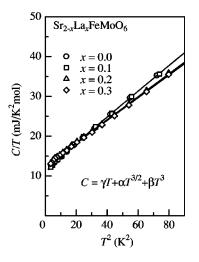


FIG. 4. Temperature dependence of the low-temperature specific heat *C* of $(Sr_{2-x}La_x)$ FeMoO₆. Solid curves are the best fitted results taking into account of contributions from electrons, spin-waves and phonons: $C = \gamma T + \alpha T^{3/2} + \beta T^3$.

=3.8(2) $\mu_{\rm B}$ and $\mu_{\rm Mo}$ =-0.8(2) $\mu_{\rm B}$ at x=0.3. At x=0.0, both the moments are suppressed as compared with the ideal values (5.0 $\mu_{\rm B}$ and -1.0 $\mu_{\rm B}$ for the Fe³⁺ and Mo⁵⁺ ions, respectively). This is partly due to the disorder of the *B*-site ions. With increase of x, absolute magnitude of $\mu_{\rm Mo}$ increases from -0.5(2) $\mu_{\rm B}$ at x=0.0 to -0.8(2) $\mu_{\rm B}$ at x = 0.3. This enhancement of $\mu_{\rm Mo}$ indicates that the doped electrons occupy mainly the Mo4 d_{\downarrow} band, consistently with the LDA band calculation⁷ as well as the LDA+U calculation (*vide infra*).

B. Doping dependence of physical properties

The electron doping has significant effects on the lowtemperature specific heat *C*. The temperature variation of *C* was measured by the relaxation method, and plotted in Fig. 4 against T^2 . In the magnetic compound, the total specific heat is composed of three parts, namely, contribution from the conduction electrons, the spin waves, and the phonons:

$$C = \gamma T + \alpha T^{3/2} + \beta T^3. \tag{1}$$

The solid curves in Fig. 4 are the best-fitted results. Thus obtained parameters, that is, α and γ , are plotted in Fig. 5

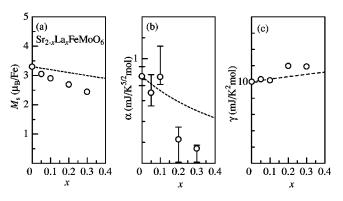


FIG. 5. Filling dependence of (a) saturation magnetization M_s , (b) α (spin-wave), and (c) γ (electron) coefficients in the low-temperature specific heat of $(Sr_{2-x}La_x)$ FeMoO₆. Broken curves are results of a model calculation (see text).

against *x*. The parameters of Sr₂FeMoO₆ are nearly consistent with the values reported by Tomioka *et al.*¹⁰ The thermal effective mass $m_{\text{eff}} [\equiv \gamma/\gamma_{\text{free}};$ where $\gamma_{\text{free}} (=5.9 \text{ mJ/K}^2 \text{ mol})$ is for the free electrons per Mo site] is estimated to be ≈ 1.7 . Such an enhancement of m_{eff} is perhaps ascribed to the electron-correlation effect. Incidentally, the magnitude of β ($\propto \Theta^{-3}$; where Θ is the Debye temperature) slightly decreases from $\approx 0.56 \text{ mJ/K}^4$ mol at x = 0.0 to $\approx 0.53 \text{ mJ/K}^4$ mol at x = 0.3.

The contribution of the spin waves is expressed as

$$\alpha = 0.113k_{\rm B}(k_{\rm B}/D_{\rm s})^{3/2},\tag{2}$$

where $k_{\rm B}$ and $D_{\rm s}$ are the Boltzmann constant and the effective spin stiffness coefficient for the acoustic branch, respectively. With use of α (=0.8 meV Å^{5/2}) at x=0.0, $D_{\rm s}$ is estimated to be \approx 140 meV Å². Taking into account of the ferrimagnetic structure, $D_{\rm s}$ is further expressed as

$$D_{\rm s} = 4|J|a^2 S_{\rm Fe} S_{\rm Mo} / (S_{\rm Fe} - S_{\rm Mo}), \tag{3}$$

where *J* and *S_i* are the exchange coupling between the Fe and Mo spins, and the spin quantum number $(=\mu_i/2)$ at the *i* site, respectively. *a* (=3.95 Å) is the averaged lattice constant.

Here, let us argue the electron doping effects on the magnetic and electronic properties, i.e., M_s , α , and γ , in terms of the rigid Mo4 d_{\perp} band. We summarized in Fig. 5 x dependence of (a) M_s , (b) α , and (c) γ . The broken curve in Fig. 5(a) is the calculated magnetization, with assuming the decreasing μ_{Mo} [=-(1.0+x) μ_{B}] and the constant μ_{Fe} $(=+4.3\mu_{\rm B})$. We further have calculated the α value using Eqs. (2) and (3). Magnitude of J(=-6.4 meV) is so determined that the D_s value at x=0.0 becomes 140 meV Å². On the other hand, increase of the γ coefficient is qualitatively reproduced by the free-electron picture, i.e., $\gamma \propto n^{1/3}$ $=(1+x)^{1/3}$ [see a broken curve in Fig. 5(c)]. Thus the rigidband picture qualitatively reproduces the overall x-dependent feature of M_s , α , and γ . We, however, observe a serious deviation of the experimental data from the calculation at larger-x region.

C. Comparison with the band calculation

Now, let us examine the electronic structure of parent Sr_2FeMoO_6 . We show in Fig. 6 the density of state (DOS) for the up- and down-spin state of Sr₂FeMoO₆. We have calculated the electronic structure with the full-potential linearized augmented plane-wave (FLAPW) method¹⁹ within the LDA+U scheme^{20,21} with the effective U parameters of 2.0 eV for Fe and 1.0 eV for Mo. The actual tetragonal lattice parameters at 300 K were used (see Table I). The plane-wave cutoff energies are 12 Ry for the wave function, and 48 Ry for the potential and the charge density. We have performed the self-consistent calculation with 39 k points in the irreducible Brillouin zone for the tetragonal lattice, which correspond to 19 k points for the fcc lattice. The calculated $M_{\rm s}(=3.99\mu_{\rm B})$ and $\gamma(\approx 4.9 \text{ mJ/mol K}^2)$ are nearly consistent with the experimental results $(M_s = 3.3 \mu_B \text{ and } \gamma$ = 10.0 mJ/mol K^2). The generic feature is consistent with the calculation done by Kobayashi et al.,7 who assumed the fcc lattice $(d_{\rm Fe-O}/d_{\rm Fe-Fe} \equiv 1/4)$. One may notice very small

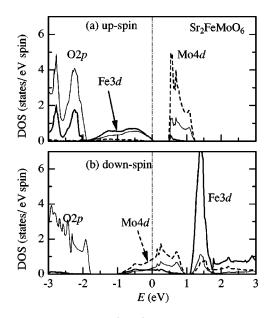


FIG. 6. Density of state (DOS) of Sr_2FeMoO_6 . The electronic structure was calculated with the full-potential linearized augmented plane-wave (FLAPW) method within the LDA+U scheme with the effective U parameter of 2.0 eV for Fe and 1.0 eV for Mo. Actual tetragonal lattice parameters at 300 K were used.

but finite DOS at $E_{\rm F}$ even for the up-spin state. The partial DOS of the Fe3*d* state is significantly affected by the oxygen position between the Mo and Fe ions, because the shorter Fe-O bond enhances the Fe3*d*(t_{2g})-O2*p* hybridization.

As seen in Fig. 6(b), the down-spin conduction band near $E_{\rm F}$ is dominated by the Mo4 $d_{\downarrow}(t_{2g\downarrow})$ state, but has a considerable amount of the Fe3 $d_{\downarrow}(t_{2g\downarrow})$ components. This is because the Mo4 $d_{\downarrow}(t_{2g\downarrow})$, Fe3 $d_{\downarrow}(t_{2g\downarrow})$, and O2 p_{\downarrow} states can be hybridized via the π bonding, and can gain the kinetic energy with forming a wide conduction band. [Contrary to the down-spin band, mixing of the Mo4 $d_{\uparrow}(t_{2g\uparrow})$ and Fe3 $d_{\uparrow}(e_{g\uparrow})$ states is prohibited due to the different symmetry.] The mixing of the Fe3 d_{\downarrow} state can explain the deviation of the experimental data from the simple model shown in Fig. 5. In addition, the mixing is consistent with the suppressed $\mu_{\rm Fe}$ and $\mu_{\rm Mo}$.

D. Possible origins for the magnetic transition

Finally, let us discuss possible origins for the magnetism of Sr₂FeMoO₆. Historically, the ferrimagnetism of this system has been understood in terms of antiferromagnetic superexchange interaction between the Mo⁵⁺(d^1) spin and Fe³⁺(d^5) spins.^{3–5} However, Moritomo *et al.*⁶ have indicated a strong interrelation between the conductivity σ_{300K} at room temperature and T_c in Sr₂MMoO₆ (M = Cr, Mn, Fe, and Co). This implies that the conduction electrons mediate the exchange interaction between the local Fe³⁺ spins. In addition, the picture of the local Mo⁵⁺ spins contradicts the reduced μ_{Mo} value of Sr₂FeMoO₆ [see Fig. 3(a)].

As discussed in the previous subsection, the conduction electrons have a considerable amount of the Fe3 d_{\downarrow} component. Then, the carriers can mediate the exchange interaction between the local Fe³⁺(d^5) spins via the double exchange mechanism.¹ Consistently, a characteristic cusp structure is observed at $\approx T_c$ in the ρ -T curve of Sr₂FeMoO₆ (see the inset of Fig. 1), indicating interrelation between resistivity and magnetic ordering. The same mechanism is applicable for Sr₂FeReO₆ ($T_c \ge 400$ K), whose band structure⁷ is similar to that of Sr₂FeMoO₆. Note that the exchange coupling between the local Fe³⁺(d^5) spins and the carriers is *antiferromagnetic*. Then, the strong hybridization between the Mo4 $d_{\downarrow}(t_{2g\downarrow})$ and Fe3 $d_{\downarrow}(t_{2g\downarrow})$ states causes negative spin polarization of the Mo4d electrons.

Quite recently, Kanamori and Terakura²² has proposed an alternative model for the magnetism of Sr_2FeMoO_6 . If the Fe³⁺ spins are ferromagnetically ordered, the hybridization between the Fe3*d* and Mo4*d* states pushes up (down) the Mo4*d*_↑ (Mo4*d*_↓) states located between the Fe3*d*_↑ and Fe3*d*_↓ levels. Then, the electrons near E_F transfer from the Mo4*d*_↑ state to the Mo4*d*_↓ state. Resultant kinetic energy gain stabilizes the ferromagnetic state, as compared with the antiferromagnetic (or the paramagnetic) states.

In the latter two pictures, the magnetic moment at the Mo site is merely induced by the Fe magnetic moment through the hybridization between the Fe3d and Mo4d states. In this sense, the present system is *not ferrimagnetic but ferromagnetic*. Our experimental results are consistent with these pictures.

- ¹P.W. Anderson and H. Hasegawa, Phys. Rev. **100**, 675 (1955).
- ²M.T. Anderson, K.B. Greenwood, G.A. Taylor, and K.R Poeppelmeier, Prog. Solid State Chem. **22**, 197 (1993).
- ³F. Galasso, F.C. Douglas, and R. Kasper, J. Chem. Phys. **44**, 1672 (1966).
- ⁴F.K. Patterson, C.W. Moeller, and R. Wald, Inorg. Chem. 2, 196 (1963).
- ⁵J. Longo and R. Wald, J. Am. Chem. Soc. 83, 2816 (1961).
- ⁶Y. Moritomo, Sh. Xu, A. Machida, T. Akimoto, E. Nishibori, M. Takata, and M. Sakata, Phys. Rev. B **61**, R7827 (2000).
- ⁷K.-I. Kobayashi, T. Kimura, H. Sawada, K. Terakura, and Y. Tokura, Nature (London) **395**, 677 (1998); K.-I. Kobayashi, T. Kimura, Y. Tomioka, H. Sawada, K. Terakura, and Y. Tokura, Phys. Rev. B **59**, 11 159 (1999).
- ⁸T.H. Kim, M. Uehara, S.-W. Cheong, and S. Lee, Appl. Phys. Lett. **74**, 1737 (1999).
- ⁹H.-Y. Hwang, S-W. Cheong, P.G. Radaelli, M. Marezio, and B. Batlogg, Phys. Rev. Lett. **75**, 914 (1995).
- ¹⁰Y. Tomioka, T. Okuda, Y. Okimoto, R. Kumai, K.-I. Kobayashi, and Y. Tokura, Phys. Rev. B 61, 422 (2000).
- ¹¹Y. Moritomo, S. Xu, A. Machida, T. Akimoto, E. Nishibori, M. Takata, M. Sakata, and K. Ohoyama, J. Phys. Soc. Jpn. **69**, 1723 (2000).

IV. SUMMARY

We have investigated the electron doping effects on the magnetic and electronic properties in the conducting doubleperovskite molybdenum oxides $(Sr_{2-x}La_x)FeMoO_6$. The LDA+*U* band calculation indicates that the down-spin conduction band near E_F has a considerable amount of the $Fe3d_{\downarrow}(t_{2g\downarrow})$ component, reflecting the strong hybridization between the Fe3*d* and Mo4*d* states. This is consistent with the *x* dependence of the magnetic and electronic properties as well as the suppressed μ_{Fe} and μ_{Mo} To clarify the significant role of the Mo4*d*-Fe3*d* hybridization on the magnetism, spectroscopic investigations are indispensable.

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- ¹²Y. Moritomo, H. Kusuya, T. Akimoto, and A. Machida, Jpn. J. Appl. Phys., Part 2 **39**, L360 (2000).
- ¹³M. Takata, E. Nishibori, K. Kato, M. Sakata, and Y. Moritomo, J. Phys. Soc. Jpn. **68**, 2190 (1999).
- ¹⁴F. Izumi *The Rietveld Method*, edited by R.A. Young (Oxford University Press, Oxford, 1993), Chap. 13; Y.-I. Kim and F. Izumi, J. Ceram. Soc. Jpn. **102**, 401 (1994).
- ¹⁵I. Mannari, Prog. Theor. Phys. 22, 335 (1959).
- ¹⁶Y. Moritomo, A. Asamitsu, and Y. Tokura, Phys. Rev. B **51**, 16 491 (1995).
- ¹⁷K. Ohoyama, T. Kanouchi, K. Nemoto, M. Ohashi, T. Kajitani, and Y. Yamaguchi, Jpn. J. Appl. Phys., Part 1 37, 3319 (1998).
- ¹⁸Determination of the spin direction is difficult, due to the quasicubic structure.
- ¹⁹O.K. Andersen, Phys. Rev. B **12**, 3060 (1975); T. Takeda and J. Kubler, J. Phys. F **9**, 661 (1979).
- ²⁰P. Hohenberg and W. Kohn, Phys. Rev. **136**, B864 (1964); W. Kohn and L.J. Sham, *ibid*. **140**, A1133 (1965); S.H. Vosko, L. Wilk, and M. Nusair, Can. J. Phys. **58**, 1200 (1980).
- ²¹V.I. Anisimov, J. Zaanen, and O.K. Andersen, Phys. Rev. B 44, 943 (1991); V.I. Anisimov, F. Aryasetiawan, and A.I. Lichtenstein, J. Phys.: Condens. Matter 9, 767 (1997); I. Solovyev, N. Hamada, and K. Terakura, Phys. Rev. B 53, 7158 (1996).
- ²²J. Kanamori and K. Terakura (unpublished).