

Impact of Pr doping on superconductivity in $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$

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A systematic study on the nature of Pr doping in $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ ($0.2 \leq x \leq 0.6$) has been made on films grown by laser ablation. These films are a metastable Bi-2212 phase which has not yet been grown in this doping range by other methods. The film with $x=0.2$ shows the highest zero-resistance temperature (T_{cz}) of about 65 K. T_{cz} of the films decreases with increasing Pr content yielding a metal-to-insulator transition for $x > 0.5$. By analyzing x-ray photoelectron spectra we show that Pr is trivalent. From this we are able to demonstrate that the variation of T_{cz} with doping is predominantly due to hole filling induced by the trivalent Pr.

I. INTRODUCTION

Most materials of the $\text{LnBa}_2\text{Cu}_3\text{O}_7$ system (Ln: lanthanide and Y) exhibit superconductivity at about 90 K, and the value of the critical temperature (T_c) does not change dramatically^{1,2} with the change of the element Ln. One exception to this behavior is insulating $\text{PrBa}_2\text{Cu}_3\text{O}_7$, and some experimental results indicate that valence of Pr in $\text{PrBa}_2\text{Cu}_3\text{O}_7$ is trivalent, while others indicate tetravalent (Refs. 3 and 4). To explain its nonsuperconductivity, many models have been proposed: (i) hole filling due to Pr^{4+} ion,⁴ (ii) replacement of Ba site with Pr^{3+} (Ref. 5) and (iii) localization of holes by hybridization of Pr 4*f* and O 2*p* orbitals.^{4,6} Among them, the model of hybridization has been widely accepted, though Blackstead *et al.* pointed out that the model cannot explain the difference in Pr content y , at which superconductivity disappears, and $|dT_c/dy|$ between $\text{Y}_{1-y}\text{Pr}_y\text{Ba}_2\text{Cu}_3\text{O}_7$ and $\text{Y}_{1-y}\text{Pr}_y\text{Ba}_2\text{Cu}_4\text{O}_8$ systems.⁷

Recently, however, bulk superconductivity has been observed in the single crystal of $\text{PrBa}_2\text{Cu}_3\text{O}_7$ prepared by the traveling-solvent floating-zone (TSFZ) method.⁸⁻¹⁰ The results showed: Superconductivity of the sample was very sensitive to the growth condition, the valence of Pr was trivalent in the sample, and the lattice parameter c of the sample was longer than the nonsuperconducting sample. Further, in the $\text{LnBa}_2\text{Cu}_3\text{O}_7$ system (Ln: lanthanide and Y), only the superconducting $\text{PrBa}_2\text{Cu}_3\text{O}_7$ compound showed large T_c enhancement under a high pressure,⁹⁻¹¹ the maximum increase of T_c was by about 50 K and the value of T_c was 105 K. However, neither the reason for the difference between the superconducting and nonsuperconducting $\text{PrBa}_2\text{Cu}_3\text{O}_7$ compounds nor the reason why only the superconducting $\text{PrBa}_2\text{Cu}_3\text{O}_7$ compound has large pressure effect on T_c among the materials of the $\text{LnBa}_2\text{Cu}_3\text{O}_7$ system has yet been understood. Therefore it is interesting to investigate other superconductors containing Pr and to compare them with the $\text{PrBa}_2\text{Cu}_3\text{O}_7$ compound.

The Bi-2212 phase in the Bi-Pr-Ca-Cu-O system is a superconductor containing Pr prepared by solid-state reaction. The crystal structure of the compound was reported to be similar to the Bi-2212 phase of $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+\delta}$, and the sample of $\text{Bi}_2\text{Pr}_{0.5}\text{Ca}_{2.5}\text{Cu}_2\text{O}_{8+\delta}$ (Ref. 12) showed the highest zero resistivity temperature (T_{cz}) of 45 K among the materials of the $\text{Bi}_2\text{Ln}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ systems¹²⁻¹⁷ (Ln=La, Pr, Nd, Sm, Eu, and Gd). However, in the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system, an almost single phase sample could be obtained only for the limited composition of $x=0.5$, and other samples were of the multiphase. On the other hand, the thin-film preparation method allows one to make metastable Bi-2212 phase that cannot be synthesized by solid-state reaction. For example, an almost single Bi-2212 phase sample in the bulk $\text{Bi}_2\text{La}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system was only $x=0.5$ (Ref. 12), while thin-film preparation method has been able to make the Bi-2212 phase with a wide composition range,¹⁸ $0.2 \leq x \leq 0.7$.

We have prepared the Bi-2212 films with the wide composition range of Pr, and have investigated the role of Pr for superconductivity in the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system. In this paper, we report on the preparation and physical properties of Bi-2212 thin films of this system.

II. EXPERIMENT

The films were synthesized by the laser ablation method using $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ compound targets. First, pellets for the target were prepared by solid-state reaction. High purity powders of Bi_2O_3 , Pr_6O_{11} , CaCO_3 , and CuO were mixed to a nominal composition of $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_8$ and pressed to pellets, which were heated at 820 °C for 10 h in air. The pellets for the target were reground, pressed, and sintered at 820 °C for 10 h in air again. The targets were placed in a high-vacuum chamber at a base pressure of 1.0×10^{-7} Torr and ablated by a pulsed KrF excimer laser ($\lambda = 248$ nm, $\tau = 23$ ns, 100 mJ/pulse) in 0.02 Torr oxygen

TABLE I. Pr content x of the target materials, film composition determined by SEM-EDS, substrate temperature, annealing temperature in O_2 , lattice parameter c , T_{cs} , T_{cmid} , T_{cz} , and T_{cm} of the $Bi_2Pr_xCa_{3-x}Cu_2O_{8+\delta}$ films.

Pr(x)	Film composition (SEM-EDS)	Deposition temp.	O_2 temp.	c (Å)	T_{cs} (K)	T_{cmid} (K)	T_{cz} (K)	T_{cm} (K)
0.2	$Bi_{1.79}Pr_{0.20}Ca_{2.64}Cu_{1.59}$	600 °C	760 °C	30.02	80.0	72.5	65.0	75.0
0.3	$Bi_{1.65}Pr_{0.30}Ca_{2.53}Cu_{1.71}$	630 °C	790 °C	29.97	77.0	67.0	57.0	70.0
0.4	$Bi_{1.79}Pr_{0.40}Ca_{2.82}Cu_{1.77}$	650 °C	790 °C	29.95	55.0	37.5	20.0	51.0
0.5	$Bi_{1.91}Pr_{0.50}Ca_{2.43}Cu_{1.81}$	670 °C	820 °C	29.92	45.0	30.0	15.0	
0.6	$Bi_{1.57}Pr_{0.60}Ca_{2.09}Cu_{1.56}$	670 °C	820 °C	29.94				

flow. The energy density of the laser at the target was 1.0 J/cm², and the pulse repetition rate was 4.0 Hz. The ablated materials were deposited on the MgO (100) substrate placed at a distance of 3.0 cm from the target in the chamber. The thickness of the films was 1.5 μ m, and the deposition rate of the films was 3.0 Å per laser shot.

For the $Bi_2Pr_xCa_{3-x}Cu_2O_{8+\delta}$ films, the substrate temperatures were monitored by an optical pyrometer and were kept at 600–670 °C. After the deposition, O_2 annealing treatment was performed under 1 atm O_2 flow for 30 min at the temperature of 760–820 °C. The substrate and annealing temperatures of the films are summarized in Table I.

X-ray-diffraction patterns were obtained with a diffractometer using $CuK\alpha$ radiation. The temperature dependence of electrical resistivity (ρ vs T) was measured by the dc four-probe method. The temperature dependence of magnetization (M vs T) was measured with a dc superconducting quantum interference device (SQUID) magnetometer. The composition of the films was analyzed by scanning electron microscopy–x-ray energy dispersion spectroscopy (SEM-EDS). The x-ray photoelectron spectroscopy (XPS) measurement was performed using $MgK\alpha$ radiation.

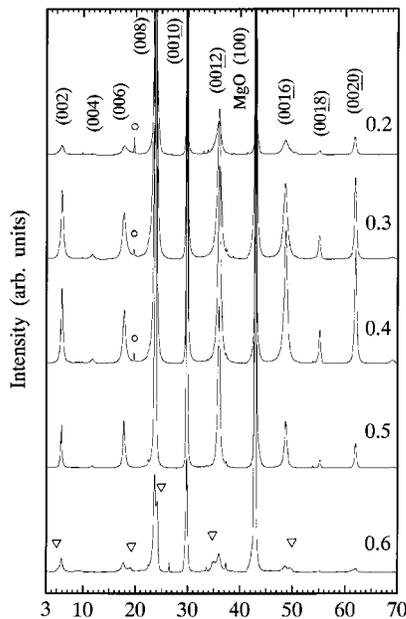


FIG. 1. X-ray-diffraction patterns of $Bi_2Pr_xCa_{3-x}Cu_2O_{8+\delta}$ films with $0.2 \leq x \leq 0.6$. Circles, impurity phase of Bi-Ca-O compound; triangles, Bi-2223 phase in the Bi-Pr-Ca-Cu-O system.

III. RESULTS

X-ray-diffraction patterns of the Bi-2212 thin films in the $Bi_2Pr_xCa_{3-x}Cu_2O_{8+\delta}$ system ($0.2 \leq x \leq 0.6$) are shown in Fig. 1. From the figure, we can see that most of the diffraction peaks can be attributed to the c -axis oriented Bi-2212 phase. However, the impurity phase of Bi-Ca-O is mixed to a larger extent for smaller Pr content x . Though results for the films with $x < 0.2$ are not shown in the figure, these films have been found to be of the multiphase consisting of the Bi-2212 phase, Bi-Ca-O and other unknown impurity phases. The result indicates that Pr is necessary to stabilize the 2212 structure.

On the other hand, the film with $x = 0.6$ contains Bi-2212 and Bi-2223 phases of the Bi-Pr-Ca-Cu-O system together with unknown phases. The fact that the film contains the Bi-2223 phase shows the existence of the Bi-2223 phase in the Bi-Pr-Ca-Cu-O system. The detailed result of the Bi-2223 phase will be published elsewhere.¹⁹ When x exceeds 0.6, the samples become of the mixed phase consisting of the present Bi-2212 phase, the Bi-2201 phase of the Bi-Pr-Ca-Cu-O system²⁰ and impurity phases. These results indicate that the composition range of Pr to stabilize the 2212 structure is $0.2 \leq x \leq 0.6$. This result is in contrast to the bulk Bi-2212 phase of $Bi_2Pr_xCa_{3-x}Cu_2O_{8+\delta}$, as only for $x = 0.5$ an almost single phase was obtained.¹² From these results, we can emphasize that the present thin-film processing method is one of the most useful methods to prepare new metastable phases.

M vs T for the Bi-2212 films in the $Bi_2Pr_xCa_{3-x}Cu_2O_{8+\delta}$ system is shown in Fig. 2. The measurement was performed on heating after cooling within a magnetic field of 200 G, which was applied parallel to the c axis of the film. In the

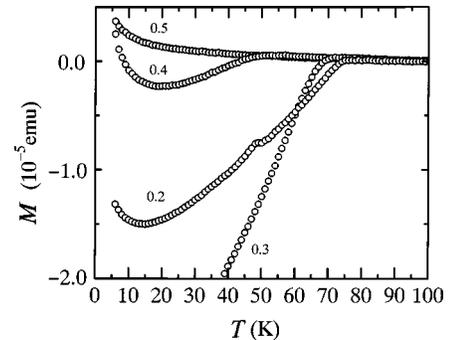


FIG. 2. M vs T of the $Bi_2Pr_xCa_{3-x}Cu_2O_{8+\delta}$ films with $0.2 \leq x \leq 0.5$.

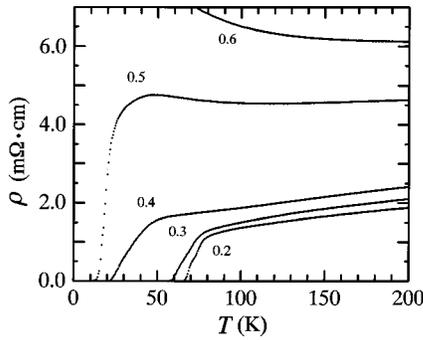


FIG. 3. ρ vs T of the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ films ($0.2 \leq x \leq 0.6$).

figure, the film with $x=0.2$ shows the highest starting temperature of diamagnetic signal (T_{cm}), and its value decreases with increasing Pr content x . The film with $x=0.5$ does not show the diamagnetic signal. The T_{cm} value of each film is shown in Table I. The result indicates that the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ film changes from a superconductor to a nonsuperconductor with increasing x .

ρ vs T for the Bi-2212 films in the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system is shown in Fig. 3. From the figure, we can see that the film with $x=0.2$ shows the highest zero resistivity temperature (T_{cz}) at 65 K in the system. This value is higher by 20 K than that of the bulk sample.¹² Moreover, the starting temperature of resistivity drop (T_{cs}), the middle temperature between T_{cs} and T_{cz} (T_{cmid}) and T_{cz} of the films decreased with increasing Pr content x . When x exceeded 0.5, metal-insulator transition occurred and the film with $x=0.6$ showed semiconductorlike temperature dependence with no trace of resistivity drop.

From the magnetic and resistivity measurements of the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ films, it is clear that T_c of the film decreases with increasing Pr content x , and the film changes from a superconductor to an insulator. From the results mentioned in the previous section, we have found that T_c 's of Bi-2212 films in the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system decrease with increasing Pr content x . In order to clarify the origin of this decrease in T_c , we need to address the valence of Pr. From the experimental results described below, we are able to show that Pr is trivalent.

Figure 4 shows XPS spectra in the binding-energy range of 100 to 140 eV for the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ films ($0.2 \leq x \leq 0.6$) as well as the bulk compounds Pr_6O_{11} and Pr_2CuO_4 . In the case of the Pr_6O_{11} compound, the peak of Pr 4d is observed at the binding energy of about 115 eV; and an additional structure denoted by $4f^1$ at the energy of about 126 eV. The former shows the existence of Pr^{3+} , and the latter, the existence of Pr^{4+} (Ref. 3). Since the valence of Pr is trivalent in the Pr_2CuO_4 compound, the peak of Pr 4d (115 eV) is observed without the $4f$ (Ref. 3) peak. On the other hand, in the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system, the peaks of Pr 4d and Cu 3s are present³ at 115 and 121 eV and the peak intensity of Pr 4d increases with increasing Pr content x . However, the $4f$ (Ref. 3) peak which shows the existence of Pr^{4+} is absent. The Pr 4d spectra of the films is similar to the Pr_2CuO_4 compound than the Pr_6O_{11} compound though there is a shoulder on the low-energy side of the Pr 4d peak. From

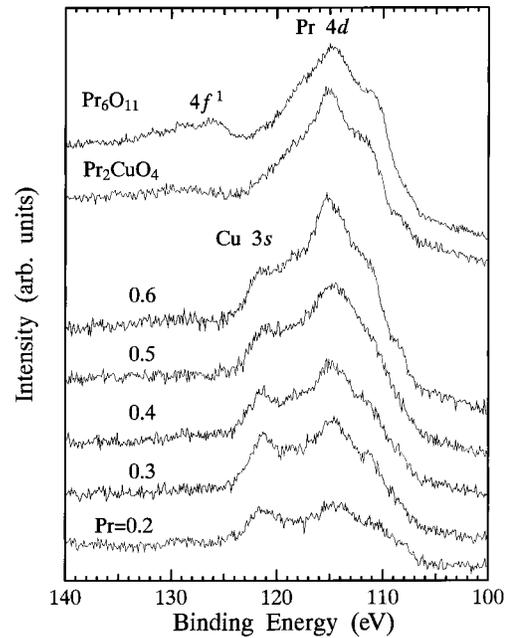


FIG. 4. XPS spectra for the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ films ($0.2 \leq x \leq 0.6$) and the reference samples of Pr_6O_{11} and Pr_2CuO_4 compounds.

the result, we conclude that the valence of Pr is trivalent in the films.

Finally, we discuss the role of Pr^{3+} for superconductivity in the system. In Fig. 5, the variations, with Pr content x and hole concentration p , of T_{cmid} in the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system are shown (closed triangles). The data of the $\text{Bi}_2\text{Sr}_2\text{Pr}_x\text{Ca}_{1-x}\text{Cu}_2\text{O}_{8+\delta}$ (open triangles) is cited from Ref. 21 for comparison. The hole concentration p of the systems has been calculated by the formula

$$p = 0.3 - x/2 \quad (1)$$

by assuming the following experimental facts of the $\text{Bi}_2\text{Sr}_2\text{Pr}_x\text{Ca}_{1-x}\text{Cu}_2\text{O}_{8+\delta}$ system: (i) the oxygen content of the system is a constant independently of Pr content x , (ii) Pr is trivalent, (iii) one Pr^{3+} atom substituted for a Ca^{2+} atom diminishes one hole, and (iv) the hole concentration p in

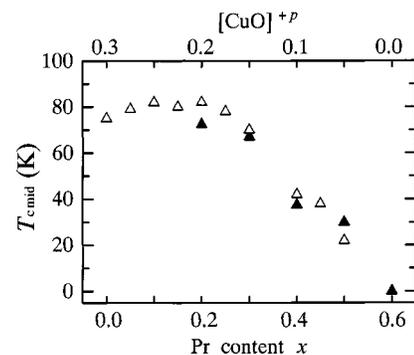


FIG. 5. Variations of T_{cmid} in the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system with $0.2 \leq x \leq 0.6$ as a function of Pr content x and hole concentration p (closed triangle). The data of the $\text{Bi}_2\text{Sr}_2\text{Pr}_x\text{Ca}_{1-x}\text{Cu}_2\text{O}_{8+\delta}$ system (open triangle) (Ref. 21) are given for comparison. The hole concentration p of the systems was calculated by the formula $p = 0.3 - x/2$ (see the text).

$[\text{CuO}]^{+p}$ is 0.3 for $x=0.0$. As shown in the figure, the two systems show the parabolalike correlation of $T_{c\text{mid}}$ with p , that is, $T_{c\text{mid}}$ increases with increasing hole concentration p , takes on a maximum at $p=0.2$ and then decreases. Therefore the decrease of T_c in the system is interpreted in terms of hole filling due to the substitution of Pr^{3+} for Ca^{2+} .

IV. CONCLUSION

We have synthesized Bi-2212 thin films of the $\text{Bi}_2\text{Pr}_x\text{Ca}_{3-x}\text{Cu}_2\text{O}_{8+\delta}$ system prepared by the laser ablation method. For this system, the Bi-2212 films were obtained in a wide composition range, $0.2 \leq x \leq 0.6$. The films in this composition range were of the metastable Bi-2212 phase that have not been produced by solid-state reaction except the film with $x=0.5$. The film with $x=0.2$ showed the highest values of T_{cm} and T_{cz} at about 75 and 65 K in the system. The value of T_{cz} was higher by about 20 K than that previ-

ously reported for the bulk Bi-2212 sample. Further, T_c of the films decreased with increasing Pr content x , and metal-insulator transition occurred when x exceeded 0.5. The variation of T_c was explained in terms of hole filling due to the replacement of Ca^{2+} with Pr^{3+} .

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