Antiferromagnetic ordering states of oxygen-deficient $NdBa_2Cu_3O_{6+x}$ and $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ single crystals

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The paper describes a study of the antiferromagnetic ordering of oxygen-deficient $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ single crystals. In pure, stoichiometric samples, $y=0$, with different oxygen contents x in the Cu(1) plane $(0.02 < x < 0.2)$, the antiferromagnetic I (AFI) phase appears to be stable down to 316 mK. The magnetic ordering within the Cu(2) sublattice of the pure $NdBa_2Cu_3O_{6+x}$ system is therefore similar to that of the pure $YBa₂Cu₃O_{6+x}$ parent compound. With increasing oxygen content, the Ne^el temperature drops significantly and the critical exponent changes from 0.26 for $NdBa_2Cu_3O_{6.09}$ to about 0.5 for $NdBa_2Cu_3O_{6.23}$. Magnetic ordering of the Nd³⁺ moments sets in at 1.7 K with a critical wave vector $\mathbf{q}_{Nd} = (\frac{1}{2} \frac{1}{2} \frac{1}{2})$. Reordering to the AFII phase is observed in a crystal, which has a significant amount $(4%)$ of Nd³⁺ ions substituted on the Ba sites. The relatively high temperature $T_2=95 \text{ K}$ of this reordering suggests that the Nd³⁺ ions on Ba sites are very effective defects forcing the AFI*↔*AFII reordering. The mechanism of reordering is explained in terms of the creation of Cu²⁺ moments within the Cu(1) layer due to the Nd³⁺ ions on Ba sites, which via polarization lead to an effective ferromagnetic coupling between the moments on next-nearest neighboring $Cu(2)$ layers. The Nd^{3+} doping on the Ba^{2+} sites increases the Néel temperature compared to the stoichiometric compound. $[$ S0163-1829(99)02002-0]

I. INTRODUCTION

The NdBa₂Cu₃O_{7- δ} system has attracted considerable attention due to its higher superconducting transition temperature T_c compared with YBa₂Cu₃O_{7- δ}. T_c values of up to 96 K, critical current densities of up to 1.2×10^5 A/cm² at 77 K and 2.5 T, and irreversibility fields exceeding 13 T at 77 K can be achieved,^{1,2} which move NdBa₂Cu₃O_{7- δ} into the range of interest for technical applications.

To understand the differences between the superconducting properties of YBa₂Cu₃O_{7- δ} and NdBa₂Cu₃O_{7- δ} it is important to identify any differences in their magnetic behavior, especially because Nd^{3+} ions can possess a magnetic moment while Y^{3+} ions cannot. A direct consequence of this is that magnetic ordering of the Nd sublattice occurs at low temperatures coexistent with superconductivity. Furthermore, previous investigations of the magnetic structure adopted by the Cu sublattice seemed to indicate that the magnetic ordering scheme in reduced $NdBa_2Cu_3O_{6+x}$ is strictly different from that of $YBa₂Cu₃O_{6+x}$, ^{3–5} even though the crystal structure of the two compounds is virtually identical.

It is well established that pure $YBa₂Cu₃O_{6+x}$ shows a transition from the metallic and superconducting phase to the insulating and antiferromagnetically ordered antiferromagnetic I (AFI) phase upon successive reduction of the oxygen content *x* in the basal plane to below $x \sim 0.4$.^{6–8} For $0 \le x$ < 0.2 the Ne^{el} temperature T_N is slightly above 400 K and almost independent on *x*. This AFI phase is characterized by ordering of the moments on the Cu(2) sites within the CuO₂ planes forming antiferromagnetic bilayers, which are separated by the weak-coupling path along the $Cu(1)$ position occupied exclusively by nonmagnetic Cu^+ for $x=0$. The AFI ordering scheme is characterized by a critical wave vector $\mathbf{q}_{AFI} = (\frac{1}{2} \frac{1}{2} 0)$. An average *ordered* moment on the Cu(2) site of 0.55μ _B with a spin orientation perpendicular to the *c* axis has been reported in a $YBa₂Cu₃O_{6.1}$ crystal,⁹ while no *ordered* magnetic moment is present on the $Cu(1)$ sites in the oxygen-deficient layer. Whereas in very pure $YBa₂Cu₃O_{6+x}$ crystals the AFI phase is stable down to lowest $temperatures$, reordering to a second antiferromagnetic phase AFII with characteristic wave vector $\mathbf{q} = (\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ has been reported in the doped Y system. $3-5,9-22$ This AFII phase has a similar antiferromagnetic arrangement of the spins

within the $Cu(2)$ bilayer as the AFI phase, but alternating AF magnetization of the bilayers stacked along the *c* axis, which is thus doubled with respect to the chemical unit cell.¹³ The AFII phase was found to occur in $YBa_2Cu_{3-z}M_zO_{6+x}$ doped by trivalent ions like $Al^{3+}, ^{9-11}CO^{3+}, ^{14,15}Fe^{3+}, ^{16-22}$ or Ga^{3+} ,²² which are known to substitute predominantly onto the Cu(1) sites. $23-28$

In contrast to the pure Y system, the AFII phase was frequently found in apparently undoped $NdBa_2Cu_3O_{6+x}$.³⁻⁵ Moudden *et al.*⁴ reported that the AFII phase covers the whole antiferromagnetic region in their $NdBa_2Cu_3O_{6.10}$ single crystal. Li *et al.*⁵ observed the reordering from the AFI to the AFII phase at temperatures far below T_N with the reordering temperature T_2 depending on the oxygen content *x*.

One difference between $YBa₂Cu₃O_{6+x}$ and NdBa₂Cu₃O_{6+x} is the fact that Nd³⁺ ions can substitute onto the Ba sites under particular preparation conditions yielding $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$.^{29–31} Furthermore, as mentioned above, Nd^{3+} carries a magnetic moment while Y^{3+} does not. It has been speculated that the AFI*↔*AFII reordering in NdBa₂Cu₃O_{6+x} is due to an *ordering* of Cu²⁺ moments within the $Cu(1)$ layer, which is induced by the substitution of Nd on Ba sites.⁵ For YBa₂Cu_{3-z}M_zO_{6+x} the mechanism of reordering has been attributed to frustration of moments on the Cu(1) sites in the AFI phase.^{13,15–17} However, the nuclear quadrupole resonance (NQR) results of Lütgemeier and Heinmaa22 definitely rule out an *ordered* magnetic moment on Cu(1) sites in YBa₂Cu_{3-z}M_zO_{6+x} doped by different trivalent cations. The AFI*↔*AFII reordering is driven by the magnetic moment of a dopant and also by isolated Cu^{2+} moments created by the excess oxygen coordinating nonmagnetic dopants like Al^{3+} or Ga^{3+} , $9-11,22$ Recently, Andersen and Uimin³² presented a theoretical model that explains the AFI*↔*AFII reordering via intermediate turn-angle phases by the formation of free Cu^{2+} spins in the $Cu(1)$ basal planes. The free spins may be polarized and establish an effective ferromagnetic type of coupling between the adjacent $CuO₂$ layers. It is suggested that trivalent cations on the $Cu(1)$ site as well as the Ba site may give rise to the free Cu^{2+} spins.

In the oxygen-deficient $NdBa_2Cu_3O_{6+x}$ system it is still unclear if the reordering to the AFII phase is a general phenomenon or exclusively results from the Nd^{3+} ions on Ba sites. Therefore we studied the structural and magnetic properties of $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ single crystals with stoichiometric composition as well as of crystals with a significant amount of Nd^{3+} on Ba sites.

II. EXPERIMENT

Very pure, stoichiometric $NdBa_2Cu_3O_{6+x}$ single crystals were grown from a BaO/CuO flux in $SnO₂$ crucibles by the slow-cooling method.³³ Powders of CuO, BaCO₃, and $Nd₂O₃$ of a purity better than 99.99% were used. The growth was performed in air at a pressure of 70 mbar. Crystal growth was started at 1283 K, and the cooling rate during the growth was 0.8 K/h. The Nd/Ba-substituted crystals were grown in air atmosphere at 1 bar. For these crystals, Y-stabilized $ZrO₂$ crucibles were used, which due to crucible corrosion led to a slight contamination of Y on the Nd sites. Here the growth started at 1313 K with a cooling rate of 1 K/h.

The as-grown crystals were annealed in 1-bar flowing oxygen atmosphere between 873 K and 673 K during 500– 660 h. The superconducting transition of the oxidized samples was found by ac susceptibility at T_c =90.5 K for the stoichiometric crystals and at 77.5 K for the Nd/Basubstituted crystals, both with a width of about 1.5 K. The depression of T_c is attributed to the Nd³⁺ ions substituted on Ba sites.^{29–31} Comparing this T_c value to the T_c versus *y* diagrams of Refs. 29–31 suggests a Nd content on Ba sites of roughly 10% in our crystals.

For the diffraction experiment the crystals were reduced using a gas volumetric technique.³⁴ For the determination of the oxygen content as well as the structural details, complete nuclear Bragg scattering single-crystal data sets were collected. The crystal $Nd_1Ba_2Cu_3O_{6,23}$ was studied on the TAS2 four-circle diffractometer at the DR3 reactor at Risø National Laboratory with neutron wavelength $\lambda=1.047$ Å. Half a sphere of nuclear Bragg reflections up to sin θ/λ =0.67 Å⁻¹ were collected in the θ -2 θ mode leading to 225 independent observations. Analysis of the nuclear structural data was performed with the $SHELX76$ (Ref. 35) refinement program. The other two crystals $NdBa_2Cu_3O_{6.09}$ and $Nd(Ba_{1.92}Nd_{0.08})Cu_3O_{6.17}$ were measured on the four-circle diffractometer P110/5C2 at the reactor Orphée, CE-Saclay, 36 using neutrons with λ =0.83 Å, employing the ω -scan technique. 908 reflections were collected up to sin θ/λ $=0.75 \text{ Å}^{-1}$, resulting in 242 symmetry independent observations. The structural analysis of these data sets was performed with the Prometheus program package. Numerical absorption and extinction corrections employing the Becker-Coppens model for secondary extinction of type I were applied.

Nonpolarized magnetic neutron-diffraction experiments were performed on the TAS1 triple-axis spectrometer at the DR3 reactor at Risø National Laboratory using incident neutrons of energy 13.7 meV $(\lambda = 2.42 \text{ Å})$ selected by a graphite monochromator. Higher-order neutrons were depressed by a pyrolytic graphite filter. The crystals were oriented with their $\lceil 110 \rceil$ axis vertical, so that the (hhl) reflections were in the horizontal scattering plane. Details of the experimental setup are the same as those described in Ref. 10. For most of the work a standard ⁴He cryostat was used for temperatures between 2 K and 300 K. For temperatures below 2 K, an Oxford Instruments Heliox $He³$ cryostat was employed. The studies above room temperature were performed using a resistively heated furnace can.

The NQR spectra were recorded using a standard spinecho technique with quadrature detection. The NQR spectra were measured by a frequency sweep. For measurements at 4.2 K and 1.2 K the crystals were mounted in an Oxford Instruments CF1200 He cryostat.

III. RESULTS

A. Nuclear structure

The significant 48% difference in neutron-scattering lengths of Nd $(Ref. 37)$ and Ba $(Ref. 38)$ makes neutron

Site		Position	Stoichiometric crystal $NdBa_2Cu_3O_{6.09}$	Stoichiometric crystal $NdBa_2Cu_3O_{6.23}$	Nd/Ba-substituted crystal $Nd_{1.08}Ba_{1.92}Cu_{3}O_{6.17}$	
$\rm Nd$	1d	Occupancy	1.023(7)	1.00(2)	1.023(7)	
		U_{11}	0.0032(3)	0.0040(8)	0.0044(3)	
		U_{33}	0.0062(3)	0.0073(11)	0.0083(5)	
Ba	2h	Occupancy	0.997(7)	1.00(1)	1.020(7)	
		\mathcal{Z}	0.19246(9)	0.19262(21)	0.19160(9)	
		U_{11}	0.0076(4)	0.0099(9)	0.0089(4)	
		U_{33}	0.0072(5)	0.0099(12)	0.0111(6)	
Cu(1)	1 _a	Occupancy	1.022(8)	0.98(2)	0.972(7)	
		U_{11}	0.0111(4)	0.0142(10)	0.0181(5)	
		U_{33}	0.0086(5)	0.0083(11)	0.0084(5)	
Cu(2)	2g	$\ensuremath{\mathnormal{Z}}$	0.35374(6)	0.35353(14)	0.35362(6)	
		U_{11}	0.0034(2)	0.0044(5)	0.0040(2)	
		U_{33}	0.0086(3)	0.0094(7)	0.0110(3)	
O(1)	2g	Occupancy	0.998(6)	0.98(2)	0.973(6)	
		Z.	0.15210(9)	0.15189(22)	0.15311(10)	
		U_{11}	0.0148(3)	0.0284(10)	0.0226(4)	
		U_{33}	0.0096(5)	0.0112(12)	0.0111(6)	
O(2)	4i	\overline{z}	0.37181(6)	0.37099(14)	0.37194(7)	
		U_{11}	0.0058(2)	0.0054(7)	0.0074(3)	
		U_{22}	0.0038(2)	0.0085(7)	0.0047(3)	
		U_{33}	0.0096(3)	0.00109(7)	0.0124(3)	
O(4)	2f	Occupancy	0.090(7)	0.23(2)	0.175(8)	
		U_{11}	0.025(14)	0.047(17)	0.039(10)	
		U_{22}	0.003(10)	0.016(12)	0.031(9)	
		U_{33}	0.026(16)	0.015(12)	0.016(8)	
		a	3.898 (6) Å	3.900(2) \AA	3.890(4) \AA	
		\boldsymbol{c}	$11.882(24)$ Å	11.790(7) Å	$11.814(25)$ Å	
		\boldsymbol{R}	2.4%	2.8%	2.6%	
		R_W	2.8%	3.8%	2.9%	

TABLE I. Results of the structural refinement of three neutron Bragg data sets. The Debye-Waller factor has the form $\exp-2\pi^2[a^{\star2}(h^2U_{11}+k^2U_{22})+l^2c^{\star2}U_{33}]$, where the U_{ii} are the mean-square displacements.

diffraction a suitable tool for determining the content of Nd on Ba sites. Table I gives the results of our structure refinements. For the crystals grown under reducing conditions, the refined occupancy of the Ba site is very close to unity. Wavelength dispersive x-ray microprobe (WDX) investigations on crystals of the same batch also showed no deviation from the Nd:Ba ratio of 1:2. 39 We conclude that under reducing growth conditions the Nd^{3+} ions do not enter the Ba sites. The oxygen occupancy on the $O(4)$ site of these two stoichiometric crystals was refined to $x=0.09$ and $x=0.23$, respectively. Although the apical oxygen $O(1)$ shows a slight deficiency, in the following we will call these crystals $Nd_1Ba_2Cu_3O_{6.09}$ and $Nd_1Ba_2Cu_3O_{6.23}$, respectively, according to the oxygen content *x* within the $Cu(1)$ layer.

In contrast, the crystal prepared in 1 bar air shows a refined Ba occupancy significantly larger than 1. Taking into account the scattering length of both elements, one finds a Nd content of 4% on Ba sites. The occupancy *x* of the oxygen $O(4)$ within the Cu(1) layer is 0.17, so that in the following we will call this crystal $Nd(Ba_{1.92}Nd_{0.08})Cu_{3}O_{6.17}$. The Nd:Ba ratio obtained from the structural refinement is also consistent with careful WDX analysis.³⁹

B. Magnetic properties of stoichiometric $Nd_1Ba_2Cu_3O_{6+x}$

In all our reduced stoichiometric $Nd_1Ba_2Cu_3O_{6+x}$ crystals, the spin system orders into the AFI structure below T_N , as indicated by magnetic superlattice peaks with wave vector $\mathbf{q}_{AFI} = (\frac{1}{2} \cdot \frac{1}{2} 0)$. The superlattice peaks are sharp in all directions implying three-dimensional (3D) and long-range order.

Figure 1 illustrates the temperature dependence of the integrated intensity of the magnetic $(\frac{1}{2}, \frac{1}{2})$ peak of $Nd_1Ba_2Cu_3O_{6.09}$. The transition from the paramagnetic phase to AFI sets in at $T_N = 376(1)$ K. The integrated intensity shows a power-law behavior $I \sim I_0(T_N-T)^{2\beta}$ with a critical exponent $\beta=0.25(1)$, fitted between 300 K and T_N . From 11 accessible magnetic superlattice peaks, the ordered magnetic moment on $Cu(2)$ at 4.2 K was refined to $0.59(5)\mu_B$, oriented within the (001) plane.

Below about 1.7 K additional magnetic peaks with **q** $=$ $(\frac{1}{2}, \frac{1}{2})$ appear. While the intensity of the $(\frac{1}{2}, \frac{1}{2}, \frac{3}{2})$ peak strongly increases with decreasing temperature below 1.7 K, the $(\frac{1}{2}, \frac{1}{2})$ peak remains unaffected to within experimental error. Two different origins of $\mathbf{q} = (\frac{1}{2} \frac{1}{2} \frac{1}{2})$ ordering are

FIG. 1. Temperature dependence of the integrated intensity of the $(\frac{1}{2}\frac{1}{2}2)$ magnetic reflection for NdBa₂Cu₃O_{6.09}. The AFI ordering sets in at the Néel temperature T_N =376 K. Below 1.7 K peaks of the type $(h + \frac{1}{2}, k + \frac{1}{2}, l + \frac{1}{2})$ occur, which are due to an antiferromagnetic ordering of the Nd^{3+} moments; the three datapoints (full circles) were obtained at 1.7 K, 1.5 K, and 1.2 K. Note that the Nd^{3+} magnetic ordering at wave vector $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ does not affect the ordering of the Cu²⁺ moments at $(\frac{1}{2}\frac{1}{2}0)$.

known: (i) reordering of the copper spins to the AFII phase and (ii) ordering of the spins on the rare-earth site. As can be seen from Fig. 1, the AFI $\mathbf{q} = (\frac{1}{2} \frac{1}{2} 0)$ ordering component is not affected by the $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$ component. Thus the AFII ordering of the $Cu(2)$ moments can be excluded as the origin of the $\mathbf{q} = (\frac{1}{2} \frac{1}{2} \frac{1}{2})$ intensities in the present case, as AFII is necessarily competitive to AFI and depresses the $\mathbf{q} = (\frac{1}{2} \cdot \frac{1}{2} 0)$ intensities when the AFII order parameter increases.^{13,10,17,11,32} Further, the Cu(1) NQR line at 1.2 K shows no sign of line splitting or broadening, which it would if AFII ordering or the intermediate turn-angle phase were present.¹¹ As AFII ordering is thus ruled out, the $\mathbf{q} = (\frac{1}{2} \frac{1}{2} \frac{1}{2})$ magnetic ordering in our stoichiometric $Nd_1Ba_2Cu_3O_{6,10}$ is most probably related to ordering of the rare-earth sites. Antiferromagnetic ordering of the rare-earth moments occurs typically at Néel temperatures T_N^{RE} of 1–2 K.^{40–44} For Nd₁Ba₂Cu₃O_{6+x} ordering of the Nd^{3+} moments has been reported with $T_N^{\overline{R}E}$ = 1.7 K for *x* < 0.2.^{43,44} Clinton *et al.*⁴⁰ postulated a longrange, 3D antiferromagnetic structure formed by the Nd^{3+} moments with an average ordered moment $\langle \mu_z \rangle$ of 0.85 μ_B and the moment direction tilted by 45° with respect to the *c* axis. Details of the ordering of the Nd^{3+} moments in our sample will be reported elsewhere.⁴⁵

The Ne^{el} temperature T_N of AFI ordering depends on the oxygen content x within the $Cu(1)$ layer. While the $Nd_1Ba_2Cu_3O_{6.09}$ crystal has a T_N of 376 K, T_N drops to 289 K for the Nd₁Ba₂Cu₃O_{6.23} crystal and the critical exponent β becomes 0.53(2) (cf. Fig. 2). Further, in the $x=0.23$ sample the AFI order parameter goes through a maximum at about 70 K and decreases almost linearly with decreasing temperature below 70 K. As refined from 11 accessible magnetic superlattice peaks, the average ordered magnetic moment on the Cu(2) site at 70 K is $0.35(3)\mu_B$. As down to 2 K no $(\frac{1}{2}, \frac{1}{2})$ -type reflections were observed, the decrease of the AFI order parameter is not related to a turn-angle reordering to AFII. For the pure YBa₂Cu₃O_{6+x} system with $x \ge 0.2$ a reentrant behavior of the AFI order parameter has frequently been reported and has been related to a small amount of localized oxygen $2p$ holes transferred to the Cu(2) layers, which build up static disorder between the antiferromagnetic bilayers.^{46,47}

C. Magnetic properties of Nd/Ba-substituted $Nd(Ba_{1.92}Nd_{0.08})Cu_{3}O_{6.17}$

The Nd/Ba-substituted crystal $Nd(Ba_{1.92}Nd_{0.08})Cu_3O_{6.17}$ shows a completely different magnetic behavior compared to

have been assumed to be perpendicular to the c axis. AFI $(T = 100 \text{ K})$ AFII $(T=4.2 \text{ K})$							
hkl	κ (\AA^{-1})	F_{obs}^2 (10 ³ b)				F_{calc}^2 (10 ³ b) hkl κ (\AA^{-1}) F_{obs}^2 (10 ³ b)	F_{calc}^2 (10 ³ b)
$rac{1}{2}$ $rac{1}{2}$ 0	1.140	\leq 1	$0.0\,$	$rac{1}{2}$ $rac{1}{2}$ 0	1.140	\leq 1	0.0
$rac{1}{2}$ $rac{1}{2}$ 1	1.258	19.5	20.3	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	1.171	7.7	7.8
$rac{1}{2}$ $rac{1}{2}$ 2	1.559	34.9	34.3	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{3}{2}$	1.391	45.3	47.2
$\frac{1}{2}$ $\frac{1}{2}$ 3	1.960	5.0	4.4	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{5}{2}$	1.751	28.6	28.5
$rac{1}{2}$ $rac{1}{2}$ 4	2.412	12.9	12.7	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{7}{2}$	2.182	1.1	1.0
$\frac{1}{2}$ $\frac{1}{2}$ 5	2.892	37.8	38.1	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{9}{2}$	2.650	a	42.4
$\frac{1}{2}$ $\frac{1}{2}$ 6	3.387	12.7	13.6	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{11}{2}$	3.138	a	42.4
$rac{1}{2}$ $rac{1}{2}$ 7	3.391	\leq 1	2.1	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{13}{2}$	3.638	1.9	2.0
$rac{3}{2}$ $rac{3}{2}$ 1	3.461	5.0	5.2	$\frac{3}{2}$ $\frac{3}{2}$ $\frac{1}{2}$	3.431	2.3	2.3
$\frac{3}{2} \frac{3}{2} 2$	3.582	7.7	7.6	$\frac{3}{2}$ $\frac{3}{2}$ $\frac{3}{2}$	3.512	10.5	10.9
$\frac{3}{2} \frac{3}{2} 3$	3.774	\leq 1	1.0	$\frac{3}{2}$ $\frac{3}{2}$ $\frac{5}{2}$	3.669	6.1	6.4
$R = 2.8\%$				$R = 2.9\%$			
$\mu_{Cu(2)} = 0.47(1) \mu_B$				$\mu_{Cu(2)} = 0.54(1) \mu_B$			
				$\mu_{\text{Cu(1)}} = 0.007(10)\,\mu_B$			

TABLE II. Comparison between observed and calculated square of the structure factors *F*² for the AFI and AFII magnetic Bragg reflections for $Nd_{1.08}Ba_{1.92}Cu_3O_{6.17}$. The data collections for the two phases have

FIG. 2. Temperature dependence of the integrated intensity of the $(\frac{1}{2}\frac{1}{2}2)$ Bragg peaks for NdBa₂Cu₃O_{6.23}. The transition from the paramagnetic phase to the AFI phase occurs at 289 K. Although the AFI order parameter decreases below 70 K with decreasing temperature, no $(h + \frac{1}{2}, k + \frac{1}{2}, l + \frac{1}{2})$ peaks occur down to 2 K, excluding the presence of the AFII phase.

the two stoichiometric crystals. Between T_N =385 K and 95 K only magnetic peaks of the $\mathbf{q} = (\frac{1}{2} \cdot \frac{1}{2} 0)$ type are observed, indicating the presence of the AFI phase. Below T_2 =95 K magnetic superlattice peaks appear with $q = (\frac{1}{2} \frac{1}{2} \frac{1}{2})$. Between 95 K and 35 K both types of magnetic peaks coexist, while below 35 K only the $\mathbf{q} = (\frac{1}{2} \frac{1}{2} \frac{1}{2})$ type of ordering is present. The magnetic Bragg reflections in this low-temperature phase are also resolution limited, implying 3D long-range ordering. Table II reports the refinement results for the magnetic intensities of $Na_{1.08}Ba_{1.92}Cu₃O_{6.17}$ and Table III gives the magnetic moments in comparison to those of other $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ crystals. In Fig. 3 the temperature dependence of the integrated intensity of the $(\frac{1}{2}, \frac{1}{2}, 2)$ and $(\frac{1}{2}, \frac{1}{2}, \frac{3}{2})$ peaks is displayed. The AFI ordering component follows a power-law with a critical exponent β of 0.25(1), which is identical to that found for the stoichiometric $Nd_1Ba_2Cu_3O_{6.09}$ crystal. In the temperature range between 95 K and 35 K, however, the intensity of the $(\frac{1}{2}\frac{1}{2}2)$ peak decreases monotonically, while the $(\frac{1}{2}, \frac{1}{2}, \frac{3}{2})$ peak increases monotonously.

FIG. 3. Temperature dependence of the integrated intensity of the $(\frac{1}{2}\frac{1}{2}2)$ and $(\frac{1}{2}\frac{1}{2}\frac{3}{2})$ magnetic Bragg peaks for $Nd_{1.08}Ba_{1.92}Cu_3O_{6.1}$. The AFI ordering sets in at T_N =385 K. The order parameter follows a power law with a critical exponent β $=0.25(1)$. The transition to the turn-angle phase AFI \oplus AFII occurs at T_2 =95 K. Reordering to the AFII phase is completed at 35 K.

This characteristic temperature dependence indicates the presence of the turn-angle phase $\widehat{AFI} \oplus \Pi$,¹¹ via which the reordering to the AFII phase takes place.³² Figure 4 shows the NQR spectrum at 4.2 K in the frequency range between 24 MHz and 35 MHz for that crystal. The NQR line of the 63Cu isotope shows a Zeeman splitting which is typical for AFII ordering.^{11,20–22} The hyperfine field at the $Cu(1)$ site equals 0.2 T, which is very similar to those values found in $YBa_2Cu_{3-z}Fe_2O_{6+x}$ ceramics^{21,22} as well as in $YBa₂Cu_{3-z}Al_zO_{6+x}$ single crystals in the AFII phase.¹¹

IV. DISCUSSION

Our study on $NdBa_2Cu_3O_{6+x}$ has shown that substitution of Nd^{3+} on Ba sites in this material occurs only if the crystal growth takes place under oxidizing conditions^{30,31} while the substitution is prevented if growth is performed under reducing atmosphere. Further, we showed that substitution of Nd^{3+} for Ba^{2+} is accompanied by the uptake of oxygen on the $O(4)$ site in the Cu(1) layer to maintain charge balance in

TABLE III. Comparison between the magnetic transitions of the $NdBa_2Cu_3O_{6+x}$ single crystals with data from literature. Moudden *et al.* (Ref. 4) reported that their $NdBa_2Cu_3O_{6.10}$ crystal orders directly from the paramagnetic state into the AFII phase. The averaged *ordered* moments on the Cu(1) and Cu(2) sites have been determined at different temperatures T_{meas} .

Sample	T_N (K)	T_2 (K)	$\mu_{Cu(2)} (\mu_B)$	$\mu_{Cu(1)} (\mu_B)$	T_{meas} (K)
$NdBa_2Cu_3O_{6.09}^a$	376		0.59(5)		4.2
$NdBa_2Cu_3O_{6.23}^a$	289		0.35(3)		70
$Nd_{1.08}Ba_{1.92}Cu_3O_{6.17}^a$	385	95	0.47(1)		100
$Nd_{1.08}Ba_{1.92}Cu_3O_{6.17}^a$	385	95	0.54(1)	0.007(10)	4.2
$NdBa2$	385	385	0.40(2)	0.04(2)	80
$NdBa_2Cu_3O_{6.10}^{\circ}$	430	80	0.83	0.32	20
$NdBa_2Cu_3O_{6.35}^{\nc}$	230	10	0.27	0.23	5

a This work.

^bMoudden *et al.* (Ref. 4).

 c Li *et al.* (Ref. 5).

FIG. 4. Cu(1) NQR spectrum of $Nd_{1.08}Ba_{1.92}Cu_3O_{6.17}$ at 4.2 K indicating Zeeman splitting of the NQR lines of the two isotopes. The Zeeman splitting reveals a hyperfine field at the $Cu(1)$ site and thus indicates the presence of the AFII phase. For $Nd_1Ba_2Cu_3O_{6.09}$ and $Nd_1Ba_2Cu_3O_{6.23}$, there is no Zeeman splitting, while the lines are equally sharp.

the structure. Consequently, the Nd^{3+} ions are able to enter the Ba sites only if the oxygen partial pressure is high enough during growth. Excess oxygen in the basal layer is also reflected by the observed orthorhombic-tetragonal phase transition with increasing amount of Nd on Ba sites. $5,30$ The smaller ionic size of Y^{3+} (89.2 pm) (Ref. 48) compared to Nd^{3+} (99.5 pm) may explain, why the former does not substitute for Ba²⁺ (136 pm) in YBa₂Cu₃O_{6+x}.

In both stoichiometric $Nd_1Ba_2Cu_3O_{6+x}$ crystals investigated here, the spin system orders from the paramagnetic state to the AFI or $\mathbf{q} = (\frac{1}{2} \frac{1}{2} 0)$ structure, and the AFI arrangement is stable to lowest temperatures forming the magnetic ground state with respect to the Cu spin system, which is complemented by $\mathbf{q} = (\frac{1}{2} \frac{1}{2} \frac{1}{2})$ ordering of the Nd³⁺ moments. T_N is slightly depressed compared to the $YBa_2Cu_3O_{6+x}$ compound: while we find T_N =376 K for $x=0.09$ and 289 K for $x=0.23$, respectively, the expected T_N values are about 410 K and 380 K for $YBa₂Cu₃O_{6+x}$ with the same values of x^{46} The smaller T_N may be due to the slightly larger cell of $Nd_1Ba_2Cu_3O_{6+x}$ compared to $YBa_2Cu_3O_{6+x}$ at the same *x*, which reduces the coupling of the Cu^{2+} moments between the antiferromagnetic $Cu(2)$ bilayers. Moreover, a lowering of T_N is expected because the tendency for the oxygen to order in chains and thereby to dope the $Cu(2)$ layers with charge carriers is smaller in $Nd_1Ba_2Cu_3O_{6+x}$ than in $YBa₂Cu₃O_{6+x}$.^{49,50}

While the AFI ground-state ordering of the Cu^{2+} moments in stoichiometric NdBa₂Cu₃O_{6+x} is similar to that of pure YBa₂Cu₃O_{6+x} crystals, the slight Nd/Ba substitution in our $Nd(Ba_{1.92}Nd_{0.08})Cu₃O_{6.17}$ crystal induces reordering to the AFII phase via the turn-angle phase, setting in at T_2 =95 K. This magnetic reordering, in turn, is similar to that
occurring in M^{3+} (=Co³⁺,Fe³⁺,Al³⁺,...)-doped occurring in M^{3+} $(=C_0^{3+},Fe^{3+},Al^{3+},\dots)$ -doped $YBa₂Cu₃O_{6+x}$.^{9–11,14–22}

AFII ordering has been reported in nominally stoichiometric NdBa₂Cu₃O_{6+x},^{3–5} and it has been suggested that AFII ordering is an intrinsic property of the NdBa₂Cu₃O_{6+x}

FIG. 5. Crystallographic and magnetic structure of $Nd_1Ba_2Cu_3O_{6+x}$ and $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$. In the stoichiometric compound the AFI phase exists. In the Nd/Ba-substituted compound, from charge balance considerations substitution of Ba^{2+} by Nd^{3+} results in one O^{2-} ion built in at the $O(4)$ site, which oxidizes one of its Cu(1) neighbors to Cu²⁺. The associated free spin within the $Cu(1)$ layer establishes an effective ferromagnetic coupling that frustrates the arrangement of spins on $Cu(2)$ sites in the AFI phase and thus induces the reordering to the AFII phase.

compound. Li *et al.*⁵ explained the AFI*↔*AFII reordering in their $Nd_1Ba_2Cu_3O_{6+x}$ single crystals with an ordering of moments within the $Cu(1)$ layer, which in their opinion are created by substitution on either the $Cu(1)$ sites or the Ba sites. However, in our refinement of the AFII structure we did not observe any *ordered* magnetic moment on $Cu(1)$ sites within experimental error. Furthermore, our NQR measurements also exclude an *ordered* magnetic moment on $Cu(1)$ sites.

On the other hand we have strong indications that the AFI←→AFII reordering is related to the Nd³⁺ ions substituted on Ba²⁺ sites. Since the reordering temperature T_2 for $Nd_{1.08}Ba_{1.92}Cu_{3}O_{6.17}$ is relatively high, the magnetic effect of Nd^{3+} on the Ba site is very large. There are two possibilities to explain how the Nd^{3+} ions on Ba sites may induce the AFII ordering. The first possibility is the magnetic moment of the Nd^{3+} ions themselves. It is quite conceivable that the Nd^{3+} ions on the Ba sites may couple magnetically to the moments on $Cu(2)$ sites, resulting in an effective ferromagnetic coupling of the adjacent $Cu(2)$ layers, and thus precipitating AFII ordering. However, the coupling mechanism itself not is at all obvious.

The other possible mechanism is related to the creation of free spins within the $Cu(1)$ layer by the excess oxygen, which is introduced at the $O(4)$ site to charge balance the Nd^{3+} substituted for Ba^{2+} (cf. Fig. 5). This balance requires one negative elementary charge. The second elementary charge on the O^{2-} is balanced by the oxidation of one neighboring Cu⁺ on a Cu(1) site, creating a Cu²⁺ with a free spin. If these local moments couple magnetically to the moments on neighboring $Cu(2)$ sites, there will be an effective ferromagnetic coupling between the adjacent antiferromagnetic $Cu(2)$ layers, which triggers the reordering to the AFII phase. This postulated mechanism for the Nd compound is very similar to that postulated for the Al-doped system, $10,11$ and has also been treated in the theoretical work by Andersen and Uimin.³² Note that in pure, stoichiometric $Nd_1Ba_2Cu_3O_{6+x}$ or YBa₂Cu₃O_{6+x} the excess oxygen on the O(4) site oxidizes *two* neighboring $Cu⁺$ to $Cu²⁺$ that are spin-paired by superexchange, and thus no free spins are created.

Although we are suggesting a similar mechanism for the reordering, it must be borne in mind that the T_2 values observed in the $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ system are much higher than those of the Al-doped $YBa₂Cu_{3-z}Al_zO_{6+x}$ system. While the onset of AFII order induced by Al^{3+} -doping ever exceeds 25 K,^{10,11} T_2 values reported for nominally stoichiometric NdBa₂Cu₃O_{6+x} scatter between 0 and 80 K (Table III) and Moudden $et al.⁴$ even found a transition from the paramagnetic phase directly into the AFII phase at 385 K for their nominal $NdBa_2Cu_3O_{6.10}$. While the large values and also the large scatter in T_2 are quite unlikely due to M^{3+} contamination, we believe that the methods employed by these authors to determine the elemental composition of their crystals are not sensitive enough to detect a small Nd/Ba nonstoichiometry, 51 to which we attribute the AFII ordering. Further, we believe that the distribution of the free spins on Cu^{2+} ions within the Cu(1) layer has a significant influence on T_2 .^{10,28} For $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ variations in the growth procedure are likely to produce differences in the distribution of Nd^{3+} ions on Ba^{2+} sites, which in turn may have a strong influence on the stabilization of the AFII phase. If two neighboring Ba sites within the (001) plane are substituted by Nd^{3+} , the amount of free spins in the Cu(1) layer and thus T_2 is expected to be lower than in the case of randomly distributed, isolated Nd^{3+} ions. Indeed, Kramer *et al.*³¹ recently reported that also the superconducting transition temperature T_c strongly depends on the degree of Nd^{3+} pairing and that the width of the transition reflects the uniformity of the distribution.

Lütgemeier²¹ found in his studies on $YBa₂Cu_{3-z}Fe_zO_{6+x}$ ceramics, that the AFII ordering is suppressed by an excess of oxygen in the $Cu(1)$ layer. As such a mechanism could also have suppressed AFII order in the undoped, stoichiometric crystals with $x=0.09$ and $x=0.23$ studied here, we reduced another crystal of the same batch under conditions to get the oxygen content as low as possible. The reduction was performed at 800 °C at an oxygen pressure of 2.8 $\times 10^{-4}$ bars, for which, according to the oxygen isotherms published by Lee *et al.*⁵² the oxygen content should be approximately $x=0.04$. From the intensity of the 24 MHz NQR line, which is directly related to threefold coordinated $Cu(1)$ ions,⁵⁰ i.e., to Cu ions with one neighboring O^{2-} ion within the Cu(1) layer, we estimated the oxygen content *x* within the $Cu(1)$ layer for that crystal as less than 0.02. Furthermore, at 1.2 K the Cu-NQR spectrum between 27 MHz and 31 MHz gave extremely sharp lines for both isotopes without any sign of Zeeman splitting, implying that the AFI phase is present. Therefore, we conclude that in the pure, stoichiometric Nd compound, the AFI ordering scheme of the Cu moments is stable down to lowest temperature and, supplemented by ordering of the Nd^{3+} spins, forms the groundstate phase. This means that the magnetic ordering of the Cu moments in the stoichiometric Nd system is very similar to that of pure, oxygen-deficient $YBa₂Cu₃O_{6+x}$.

V. CONCLUDING SUMMARY

We have demonstrated that Nd/Ba-substituted $Nd_{1+v}Ba_{2-v}Cu_3O_{6+x}$ crystals are obtained when growth takes place under oxidizing conditions, while a reduced oxygen fugacity leads to stoichiometric $NdBa_2Cu_3O_{6+x}$. Magnetic ordering in the reduced pure stoichiometric $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ system occurs from the paramagnetic state to the antiferromagnetic AFI structure of the Cu^{2+} moments, which remains the stable configuration down to lowest temperatures, where it is supplemented by $\mathbf{q}_{Nd} = (\frac{1}{2} \frac{1}{2} \frac{1}{2})$ ordering of the Nd^{3+} moments below 1.7 K. The nonstoichiometric $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ system, however, shows a reordering from the AFI to the antiferromagnetic AFII phase via the intermediate turn-angle phase AFI \oplus AFII below T_2 595 K. We attribute the AFI*↔*AFII reordering in $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ to the creation of *isolated* magnetic Cu^{2+} ions within the Cu(1) layer, which are formed by the one excess Q^{2-} ion for each Nd³⁺ ion substituted for Ba²⁺. Coupling of these free Cu^{2+} spins to the moments on the neighboring $Cu(2)$ layers results in an effective ferromagnetic coupling between the adjacent antiferromagnetic $Cu(2)O₂$ layers, which forces the AFII reordering via an intermediate turn-angle phase upon decreasing temperature. The suggested mechanism is similar to that proposed for the Al-doped $YBa_2Cu_{3-z}Al_zO_{6+x}$ system.^{10,11,32} The higher reordering temperature T_2 and thus the larger coupling strength in the $Nd_{1+y}Ba_{2-y}Cu_3O_{6+x}$ system compared to $YBa₂Cu_{3-z}Al_zO_{6+x}$ may indicate a more random distribution of Cu^{2+} ions within the Cu(1) layer. An alternative or additional ferromagnetic coupling via the Nd^{3+} moments on Ba sites also cannot be ruled out. Since the AFI*↔*AFII reordering is induced already by a small amount of Nd^{3+} on Ba sites, we suggest that the observation of the magnetic reordering can be used as a sensitive probe to detect Nd^{3+} on Ba sites. Apart from the Nd^{3+} magnetic ordering, however, the pure, stoichiometric Nd compound shows very similar magnetic AFI ordering as the pure Y compound.

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- ¹Th. Wolf, A.-C. Bornarel, H. Küpfer, R. Meier-Hirmer, and B. Obst, Phys. Rev. B 56, 6308 (1997).
- ²Th. Wolf, H. Küpfer, and H. Wühl, in *Proceedings of the 8th International Workshop on Critical Currents in Superconductors, Kitakyushu, Japan, 1996*, edited by T. Matshushita and K. Yamafuji (World Scientific, Singapore, 1996), p. 411.
- ³ J. W. Lynn, W. H. Li, H. A. Mook, B. C. Sales, and Z. Fisk, Phys. Rev. Lett. 60, 2781 (1988).
- 4A. H. Moudden, G. Shirane, J. M. Tranquada, R. J. Birgeneau, Y. Endoh, K. Yamada, Y. Hidaka, and T. Murakami, Phys. Rev. B **38**, 8720 (1988).
- 5 W. H. Li, J. W. Lynn, and Z. Fisk, Phys. Rev. B 41, 4098 (1990).
- ⁶ J. M. Tranquada, A. H. Moudden, A. I. Goldman, P. Zolliker, D. E. Cox, G. Shirane, S. K. Sinha, D. Vaknin, D. C. Johnston, M. S. Alvarez, A. J. Jacobson, J. T. Lewandowski, and J. M. Newsam, Phys. Rev. B 38, 2477 (1988).
- $⁷P$. Burlet, C. Vettier, M. J. G. M. Jurgens, J. Y. Henry, J. Rossat-</sup> Mignod, H. Noël, M. Potel, P. Gougeon, and J. C. Levet, Physica C 153-155, 1115 (1988).
- ⁸ J. Rossat-Mignod, L. P. Regnault, C. Vettier, P. Burlet, J. Y. Henry, and G. Lapertot, Physica B 169, 58 (1991); J. Rossat-Mignod, L. P. Regnault, P. Bourges, P. Burlet, C. Vettier, and J. Y. Henry, *ibid.* **192**, 109 (1993).
- ⁹H. Casalta, P. Schleger, E. Brecht, W. Montfrooij, N. H. Andersen, B. Lebech, W. W. Schmahl, H. Fuess, Ruixing Liang, W. N. Hardy, and Th. Wolf, Phys. Rev. B 50, 9688 (1994).
- 10E. Brecht, W. W. Schmahl, H. Fuess, H. Casalta, P. Schleger, B. Lebech, N. H. Andersen, and Th. Wolf, Phys. Rev. B **52**, 9601 $(1995).$
- 11 E. Brecht, W. W. Schmahl, H. Fuess, S. Schmenn, H. Lütgemeier, N. H. Andersen, B. Lebech, and Th. Wolf, Phys. Rev. B 56, 940 (1997).
- 12S. Shamoto, M. Sato, J. M. Tranquada, B. J. Sternlieb, and G. Shirane, Phys. Rev. B 48, 13 817 (1993).
- ¹³H. Kadowaki, M. Nishi, Y. Yamada, H. Takeya, H. Takei, S. M. Shapiro, and G. Shirane, Phys. Rev. B 37, 7932 (1988).
- ¹⁴P. Zolliker, D. E. Cox, J. M. Tranquada, and G. Shirane, Phys. Rev. B 38, 6575 (1988).
- 15P. F. Miceli, J. M. Tarascon, L. H. Greene, P. Barboux, M. Giroud, D. A. Neumann, J. J. Rhyne, L. F. Schneemeyer, and J. V. Waszczak, Phys. Rev. B 38, 9209 (1988); P. F. Miceli, J. M. Tarascon, P. Barboux, L. H. Greene, B. G. Bagley, G. W. Hull, M. Giroud, J. J. Rhyne, and D. A. Neumann, *ibid.* **39**, 12 375 $(1989).$
- ¹⁶ J. L. García-Muñoz, J. Rodríguez-Carvajal, S. H. Kilcoyne, C. J. Boardman, and R. Cywinski, J. Magn. Magn. Mater. **104-107**, 555 (1992); J. L. García-Muñoz, R. Cywinski, S. H. Kilcoyne, and X. Obradors, Physica C 233, 85 (1994).
- ¹⁷ I. Mirebeau, C. Bellouard, M. Hennion, G. Jehanno, V. Caignaert, A. J. Dianoux, T. E. Philips, and K. Moorjani, Physica C **184**, 299 (1991); I. Mirebeau, C. Bellouard, M. Hennion, V. Caignaert, and E. Suard, J. Appl. Phys. **73**, 5689 (1993); I. Mirebeau, E. Suard, V. Caignaert, and F. Bourée, Phys. Rev. B 50, 3230 $(1994).$
- 18D. Hechel, I. Nowik, E. R. Bauminger, and I. Felner, Phys. Rev. B 42, 2166 (1990).
- ¹⁹ I. S. Lyubutin, V. G. Terziev, S. V. Luchko, A. Ya. Shapiro, A. M. Balagurov, and G. A. Bonch-Osmolovskiy, Physica C **199**,

296 (1992); I. S. Lyubutin, V. G. Terziev, T. V. Dmitrieva, S. V. Luchko, and A. Ya. Shapiro, Solid State Commun. **86**, 651 $(1993).$

- 20 R. A. Brand, Ch. Sauer, H. Lütgemeier, and P. M. Meufels, Hyperfine Interact. **55**, 1229 (1990).
- 21 H. Lütgemeier, Hyperfine Interact. **61**, 1051 (1990).
- ²²H. Lütgemeier and I. Heinmaa, in *Proceedings of the XXVI Zakopane School on Physics, Zakopane, 1991*, edited by J. Stanek and A. T. Pedziwiatr (World Scientific, Singapore, 1991), p. 264; H. Lütgemeier and I. Heinmaa, in Proceedings of the *Workshop on Phase Separation in Cuprate Superconductors, Erice, 1992*, edited by K. A. Müller and G. Benedek (World Scientific, Singapore, 1992), p. 243.
- ²³ J. M. Tarascon, P. Barboux, P. F. Miceli, L. H. Greene, G. W. Hull, M. Eibschutz, and S. A. Sunshine, Phys. Rev. B **37**, 7458 $(1988).$
- 24Y. Xu, M. Suenaga, J. Tafto, R. L. Sabatini, and A. R. Moodenbaugh, Phys. Rev. B 39, 6667 (1989); Y. Xu, R. L. Sabatini, A. R. Moodenbaugh, Y. Zhu, S. G. Shyu, M. Suenaga, K. W. Dennis, and R. W. McCallum, Physica C 169, 205 (1990).
- 25W. W. Schmahl, A. Putnis, E. Salje, P. Freeman, A. Graeme-Barber, R. Jones, K. K. Singh, J. Blunt, P. P. Edwards, J. Loram, and K. Mirza, Philos. Mag. Lett. **60**, 241 (1989).
- 26 R. Sonntag, D. Hohlwein, A. Hoser, W. Prandl, W. Schäfer, R. Kiemel, S. Kemmler-Sack, S. Lösch, M. Schlichenmaier, and A. W. Hewat, Physica C 159, 141 (1989).
- 27T. Siegrist, L. F. Schneemeyer, J. V. Waszczak, N. P. Singh, R. L. Opila, B. Batlogg, L. W. Rupp, and D. W. Murphy, Phys. Rev. B 36, 8365 (1987).
- 28E. Brecht, W. W. Schmahl, M. Rodewald, G. Miehe, H. Fuess, N. H. Andersen, J. Hanßmann, and Th. Wolf, Physica C **265**, 53 $(1996).$
- 29S. Li, E. A. Hayri, K. V. Ramanujachary, and M. Greenblatt, Phys. Rev. B 38, 2450 (1988).
- 30K. K. Singh, D. E. Morris, and A. P. B. Sinha, Physica C **224**, 231 (1994).
- 31M. J. Kramer, S. I. Yoo, R. W. McCallum, W. B. Yelon, H. Xie, and P. Allenspach, Physica C 219, 145 (1994).
- ³²N. H. Andersen and G. Uimin, Phys. Rev. B **56**, 10 840 (1997).
- ³³Th. Wolf, W. Goldacker, B. Obst, G. Roth, and R. Flükiger, J. Cryst. Growth 96, 1010 (1989).
- 34N. H. Andersen, B. Lebech, and H. F. Poulsen, Physica C **172**, 31 $(1990).$
- 35G. M. Sheldrick, computer code SHELX76, University of Cambridge, England, 1976.
- 36Laboratoire commun CEA-CNRS.
- 37 V. F. Sears, Neutron News 3, 29 (1992) .
- $38P$. Schweiss (unpublished).
- 39 P. Schätzle (unpublished).
- 40T. W. Clinton, J. W. Lynn, J. Z. Liu, Y. X. Jia, T. J. Goodwin, R. N. Shelton, B. W. Lee, M. Buchgeister, M. B. Maple, and J. L. Peng, Phys. Rev. B 51, 15 429 (1995).
- ⁴¹H. Drößler, H.-D. Jorstarndt, J. Harnischmacher, J. Kalenborn, U. Walter, A. Severing, W. Schlabitz, and E. Holland-Moritz, Z. Phys. B 100, 1 (1996).
- ⁴² K. Nehrke and M. W. Pieper, Phys. Rev. B **51**, 12 618 (1995); K. Nehrke, M. W. Pieper, and T. Wolf, *ibid.* **53**, 229 (1996).
- 43A. Longmore, A. T. Boothroyd, C. Changkang, H. Yongle, M. P. Nutley, N. H. Andersen, H. Casalta, P. Schleger, and A. N. Christensen, Phys. Rev. B 53, 9382 (1996).
- 44A. T. Boothroyd, A. Longmore, N. H. Andersen, E. Brecht, and

Th. Wolf, Phys. Rev. Lett. **78**, 130 (1997).

- 45A. T. Boothroyd, J. M. Reynolds, E. Brecht, N. H. Andersen, and Th. Wolf (unpublished).
- ⁴⁶ J. Rossat-Mignod, P. Burlet, M. J. Jurgens, C. Vettier, L. P. Regnault, J. Y. Henry, C. Ayache, L. Foro, H. Noël, M. Potel, P. Gougeon, and J. C. Levet, J. Phys. (Paris), Colloq. 49, C-8, 2119 $(1988).$
- ⁴⁷ I. Ya. Korenblit and A. Aharony, Phys. Rev. B **49**, 13 291 (1994).
- 48R. D. Shannon and C. T. Prewitt, Acta Crystallogr., Sect. B: Struct. Crystallogr. Cryst. Chem. **B25**, 925 (1969).
- 49H. Shaked, B. W. Veal, J. Faber, Jr., R. L. Hitterman, U. Balachandran, G. Tomlins, H. Shi, L. Morss, and A. P. Paulikas, Phys. Rev. B 41, 4173 (1990).
- ⁵⁰H. Lütgemeier, S. Schmenn, and I. Heinmaa, J. Low Temp. Phys. **105**, 693 (1996).
- 51 Due to the well-known influence of magnetic M^{3+} dopants in the $YBa₂Cu₃O_{6+x}$ system, the possibility of contamination by impure chemicals or by crucible corrosion may be considered as

the origin of the T_2 differences of nominally NdBa₂Cu₃O_{6+x} crystals (Table III). However, to explain the high- T_2 values of the NdBa₂Cu₃O_{6+x} in Refs. 4 and 5, a Fe contamination of $y_{>0.06}$ would be necessary according to the $z-T_2$ phase diagrams of Mirebeau *et al.* (Ref. 17) and Lyubutin *et al.* (Ref. 19). Such a degree of contamination with Fe^{3+} or Co^{3+} would have been easily detected by the electron-probe microanalysis and Rutherford backscattering employed by these authors. However, the precision in determining the Nd/Ba ratio by microprobe techniques requires excellent standards and is hampered by an overlap of the *K* lines of Nd and Ba, so that the weaker *L* lines have to be taken. The Rutherford backscattering edges of Nd and Ba can be separated only with difficulties, as the difference in atomic mass is only about 5%. Thus we can show that these methods are hardly sensitive enough to determine Nd doping on Ba sites unless very special measures are taken.

⁵²D. N. Lee and B. J. Lee, J. Am. Ceram. Soc. **76**, 1609 (1993).