Hybridization in PrBa₂Cu₃O₇ and PrBa₂Cu₄O₈

Howard A. Blackstead

Department of Physics, University of Notre Dame, Notre Dame, Indiana 46556

John D. Dow

Department of Physics, Arizona State University, Tempe, Arizona 85287-1504

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Comparative studies of $RBa_2Cu_3O_7$ and $RBa_2Cu_4O_8$, with $R = (R'_{1-y}Pr_y)$ and with $R' = Y$, Yb, Nd, or Gd, exclude the possibility that hybridization of Pr's 4f states with cuprate-plane oxygen 2p states is responsible for the observed degradation of critical temperatures T_c with increasing Pr content *y*. $[$ S0163-1829(98)07409-8]

 $RBa₂Cu₃O₇$ (*R*123-7) materials have been fabricated for $R = Y$, Cm, and all rare earths except Ce, Tb, and Pm. Almost all of these compounds superconduct with critical temperatures $T_c \approx 90 \text{ K}$.¹ The one noncontroversial exception is Cm123-7, which does not superconduct.² The other exception is controversial: Pr123-7 is the insulator of choice for many Josephson-junction technologies, and is widely believed to not superconduct as conventionally grown, although several authors have recently succeeded in synthesizing *superconducting* $Pr123-7,3-11$ using sophisticated methods that minimize the number of Pr-on-Ba-site defects (Pr_{Ba}) . Since a few of these demonstrations of superconductivity involve only granular superconductivity, $4-6$ some skeptical researchers have claimed that, even in the cases of bulk superconductivity,7–10 the observed superconductivity is *extrinsic* to Pr123-7, namely, some impurity phase that has nearly the same $T_c \approx 90$ K but necessarily with a new superconducting crystal structure.

Currently there appear to be three general viewpoints of why Pr123-7 is different from most R 123-7 homologues: (i) Pr123-7 is an *intrinsic nonsuperconductor* and does not superconduct, primarily due to *hybridization* of the large-radius Pr^{+3} ions's 4f electronic state with a 2p state of an adjacent oxygen ion in a cuprate plane;¹² (ii) Pr123-7 is an *intrinsic insulator* because of *hole filling*, namely, the Pr ion is in the Pr^{+4} state;¹³ and (iii) Pr123-7, when synthesized with the sophisticated methods, is an *intrinsic superconductor*, whose failure to superconduct is attributable to a defect formed during conventional synthesis, most likely Pr-on-a-Ba-site (Pr_{Ba}) , which forms easily because of Pr^{+3} 's large size.^{3,7} This paper deals with hybridization, the first of these viewpoints, which has been the most popular reason given for the failure of Pr123-7 to superconduct.

This hybridization depends critically on the $4f-2p$ overlap, namely, on the Pr-O bond length and the spatial extent of the $4f$ electron of the Pr^{+3} . To verify that this effect is responsible for the failure of Pr123-7 to superconduct, Kim *et al.*¹⁴ executed an experiment on \approx 90 K superconducting Nd123-7, based on the fact that Nd^{+3} has a $4f$ radius (and hence Nd $4f$ -oxygen $2p$ overlap) only slightly smaller than Pr^{+3} 's [0.56 Å vs 0.58 Å (Ref. 15)]: Applying pressure, they squeezed Nd123-7 until the Nd-O bond length contracted by an amount comparable with both the difference in the Pr^{+3} and Nd⁺³ radii, and the difference in 4 *f* radii, \sim 0.02 Å, and *expected the critical temperature for superconductivity to drop, due to increased* $4f-2p$ *hybridization.* Surprisingly, T_c *increased* with pressure. This result has never been satisfactorily explained within the context of any cuprate-plane model of superconductivity.¹⁶

The main concern with the Kim *et al.* experiment is that it was performed on pressurized Nd123-7, which is different from Pr123-7, both in the size of its $4f$ radius and in its energies. The hybridization mixing coefficient *M* depends on an energy denominator ΔE as well as on a 4*f*-2*p* matrix element *V*,

$$
M = -V/\Delta E,
$$

and both *V* and ΔE are different for Nd and Pr. Not only does *V* depend strongly on the $4f - 2p$ overlap (which is different), but also the chemical difference of Nd^{+3} and Pr^{+3} is reflected in the energy denominator ΔE , and either difference might play a role in determining whether T_c increases or decreases with pressure. However, the facts that (i) Cm123-7 does not superconduct,² and that (ii) Ce destroys superconductivity in both $(Nd_{1-u}Ce_u)123-7$ (Ref. 17) and (iii) in $(Y_{1-u}Ce_u)$ 123-7 (Ref. 17) with increasing Ce content *(to the extent that these materials have been formed), sug*gest that large *rare-earth* or Cm *sizes*, not *chemical differences* are related to the destruction of the superconductivity. To clarify this point, one would like to have an experiment similar to the Kim *et al.* pressure experiment that involves the same rare-earth cation, rather than the different Pr and pressurized Nd cations—and hence eliminates any differences of the mixing coefficients *M*.

To do this, we compared the effects of various rare-earth dopants R' on R 123-7 and on R Ba₂Cu₄O₈ (R 124-8) conventionally grown with $R = R'_{1-v} Pr_v$. Here the difference being exploited is the extra chain layer of the *R*124-8 compound: *R*123-7 and *R*124-8 are virtually identical structurally in the vicinity of the rare-earth ion *R* and in the adjacent cuprate planes and Ba-O layers, and differ only in that *R*124-8 has an extra Cu-O chain. Even the critical temperatures for superconductivity (which are believed to be extremely sensitive to differences in structure) are not too dif-

ferent, being ≈ 90 K for *R*123-7 and ≈ 82 K for *R*124-8.¹⁸ Moreover, the extra Cu-O chain of *R*124-8 is remote from the rare-earth site, and so we expect that the hybridization of Pr with cuprate-plane oxygen will be virtually the same in $(R_{1-y}^{\prime} Pr_{y})$ 123-7 and in $(R_{1-y}^{\prime} Pr_{y})$ 124-8. Consequently, for any rare-earth R' that occupies only the rare-earth sites, the suppression of T_c in R 123-7 and R 124-8 should be the same for $R = R'_{1-y}Pr_y$, if hybridization is responsible for the suppression and for the failure of Pr123-7 to superconduct:

$$
|dT_c/dy|_{123-7} = |dT_c/dy|_{124-8}.
$$

If there is any difference in such hybridization and suppression of T_c , the difference in $\left| dT_c/dy \right|$ should be slight and $|dT_c/dy|$ should be larger in the material with the smaller bond length between the rare-earth site and the cuprate-plane oxygen. (There is no difference in ΔE , and only a difference in *V* related to the larger $4f-2p$ overlap associated with the shorter bond length.) But the difference in the Pr-O bond lengths in Pr123-7 and Pr124-8 is only about 0.04 \AA ,¹⁹ quite a small difference, with Pr124-8 having the *shorter* bond. Therefore, if hybridization is responsible for the failure of Pr123-7 and Pr124-8 to superconduct, then a plot of T_c vs *y* for $(R_{1-v}^{\'} Pr_y)$ 123-7 and $(R_{1-v}^{\'} Pr_y)$ 124-8 should produce lines with the same slopes $|dT_c/dy|$, or a slope for $(R_{1-v}^{\prime} Pr_{v})$ 124-8 that is only slightly larger than $\left|dT_{c}/dy\right|$ for $(R'_{1-v}Pr_v)$ 123-7: the suppression of T_c by hybridization should be the same for the same local crystal geometry, independent of the choice of $R⁷$.

Such experiments have been reported by Horii *et al.*²⁰ for $R' = Y$, Yb, Nd, and Gd, and by a number of authors for $R' = Y$, in both the 123-7 and 124-8 materials. Typical results for $Y_{1-v}Pr_vBa_2Cu_3O_7$ (Ref. 21) and $Y_{1-v}Pr_vBa_2Cu_4O_8$ (Ref. 22) are presented in Fig. 1^{23-26}

The predicted similarities of the slopes due to hybridization are not evident in the data (Fig. 1): dT_c/dy for $(R_{1-v}^{\prime} Pr_{v})$ 123-7 and $(R_{1-v}^{\prime} Pr_{v})$ 124-8 are not nearly the same; and $|dT_c/dy|$ is not slightly larger for the shorter Pr-O bond length of $(R'_{1-v}Pr_v)124-8$; but dT_c/dy for the 123-7 compounds is roughly two times as large as for the 124-8 compounds, for all choices of R' we have examined: $R' = Y$, Yb, Gd, and Nd.²⁰ This is completely contrary to what is expected of a hybridization mechanism of T_c suppression.

Independent confirming evidence of the unimportance of hybridization in these Pr-based compounds is provided by measurements²⁷ that show that the normal-state resistivity is almost independent of pressure in Pr124-8, indicating that the electronic states participating in transport are insensitive to pressure-induced changes in the hybridization.

Based on comparative analyses of the T_c suppression in $(R_{1-v}^{\prime} Pr_{v})Ba_{2}Cu_{3}O_{7}$ and in $(R_{1-v}^{\prime} Pr_{v})Ba_{2}Cu_{4}O_{8}$, we conclude that *hybridization is not responsible for the failure of*

FIG. 1. Onset critical temperatures T_c for $(R'_{1-v}Pr_v)$ 123-7 (Ref. 21) and $(R'_{1-y}Pr_y)124-8^{22}$ vs Pr content *y*, for $R' = Y$, a small ion that is known to not dissolve on Ba sites. These data are typical of many measurements, which cover quite a range (Refs. 23 and 24). In both cases, we have fit the data to an Abrikosov-Gor'kov pairbreaking curve (Ref. 25), as a matter of convenience, and recognizing its limitations (Ref. 26).

conventionally grown Pr123-7 *and* Pr124-8 *to superconduct*. That leaves hole-filling (disputed by many authors^{3,22,28}) and pair breaking by Pr_{Ba} (which requires that the primary superconducting condensate not lie in the cuprate planes³) as the only well-known remaining current explanations of the anomalous behavior of these materials. Future efforts to understand why Pr123-7 and Pr124-8 do not superconduct when synthesized conventionally should focus on proving or disproving one of these two mechanisms, or on developing a new mechanism.

If, as appears to be the case, these materials are both *intrinsic* superconductors when synthesized with the sophisticated techniques, then one must understand how Pr_{Ba} destroys superconductivity, but Pr_{Pr} does not—especially if (as many workers believe) the superconducting condensate is in the plane that lies about halfway between the two sites.³

Note added in proof. Recent measurements of the effect of pressure on *superconducting* PrBa₂Cu₃O₇ show that T_c *increases* with pressure, roughly 2.5 K/GPa, exceeding 105 K at 10 GPa. These data provide some of the strongest evidence confirming the ideas discussed here. [J. Ye, Z. Zou, A. Matsushita, K. Oka, Y. Nishihara, and T. Matsumoto (unpublished)].

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