Characterization of phase structures in semiconducting SnWO₄ powders by Mossbauer and Raman spectroscopies

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Stannous tungstate, SnWO₄, crystallizes in the low-temperature α and high-temperature β phases. α -SnWO₄ powders were prepared by heating an equimolar mixture of SnO and WO_3 either in vacuum or in argon atmosphere at 600 °C for 15 h. The high-temperature β phase was obtained as metastable at room temperature after heating the mixture at 800 °C and rapid quenching. In addition to x-ray-diffraction studies, ^{119}Sn Mössbauer and Raman spectroscopies were used as ''local'' probes for the characterization of the phase structures in the semiconducting and gas-sensitive powders. The composition of the powders was measured by the energy-dispersive spectroscopy of x rays. Mossbauer spectroscopy, especially, as a very local probe for tin atoms gave valuable information of small extra phase(s) in the α - and β -SnWO₄ powders originating in the oxidation of Sn^{2+} ions in the SnWO₄ structures into the Sn⁴⁺ form. The Sn²⁺ Mössbauer doublet from both α and β phases showed some asymmetry not published before. The asymmetry was possible to relate to the Goldanskii-Karyagin effect with calculations based on published results for the atomic positions and thermal displacement parameters in the x-ray temperature factor of the α - and β -SnWO₄ structures. Raman spectra are given together with peak frequencies from a curve fit for both α - and β -phase powders. The symmetries and selection rules of the normal modes at the center of the Brillouin zone are also given for both α and β phases. $[$ S0163-1829(98)00721-8]

I. INTRODUCTION

Jeitschko and Sleight¹ reported the fabrication of $SnWO₄$ from SnO and $WO₃$ powders as two varieties, the lowtemperature α phase and high-temperature β phase, which transform into each other by a fast, but diffusion-controlled phase-transition mechanism at 670 °C. The decomposition of the α -SnWO₄ phase during heating in air starts at temperatures above about 550 °C,² while the metastable β -SnWO₄, after a rapid quenching from above 670 °C, decomposes in air at temperatures above about $450 \degree C$.³ The fabrication of the α phase is possible also in the form of thin films by sputtering.⁴ In the orthorhombic crystal structure of α -SnWO₄ both metal atoms have distorted octahedral oxygen coordinations as in $SnO₂$ and $WO₃$, whereas in the cubic structure of the β phase tungsten atoms sit inside oxygen tetrahedra which are not linked.1,5 However, in contrast to SnO₂, tin appears in the divalent form Sn²⁺ in both α and β structures. There are only a few reports of stannous tungstate in the literature.

Both $SnO₂$ and $WO₃$ are well-known materials in the field of gas-sensing and electrochromic⁷ applications. They are extrinsic *n*-type semiconductors similarly with the α and β phases of $SnWO₄$ which are also gas-sensitive oxides. The electrical conductivity of these oxides changes with concentrations of some gas components in the ambient atmosphere which is the basis for gas sensing with semiconductor gas sensors. However, electrical and gas-response properties of $SnWO₄$ powders are very sensitive to the process and conditions which are used for the fabrication of the material. The easy oxidation of Sn^{2+} into the Sn^{4+} form may generate some extra phase (s) , especially, at the grain surfaces which may have a strong influence on the gas-sensing properties of the material.

Here we use ¹¹⁹Sn Mössbauer and Raman spectroscopies together with x-ray diffraction (XRD) measurements and energy-dispersive spectroscopy (EDS) analyses for the structure characterization of α - and β -SnWO₄ powders fused from SnO and WO_3 powders both in vacuum and in argon atmospheres. In the case of very small crystals, e.g., from an extra phase at the grain surfaces, structural information from XRD and Raman results may seem confusing, since x-raydiffraction probes long-range order whereas Raman is, in principle, a "local" probe.⁸ Mössbauer spectroscopy, on the other hand, is a very useful local probe, based on hyperfine interactions, for the study of possible small extra $phase(s)$ originating in the oxidation of Sn^{2+} ions in the SnWO₄ structures into the Sn^{4+} form.

II. CRYSTAL STRUCTURES AND ATOMIC DISPLACEMENTS

Crystal structure is the basis for the interpretation of Raman results (phonon symmetries) and XRD patterns. In addition, a knowledge of the lattice vibrations (atomic displacements) is very important for the interpretation of Mössbauer spectra. Here we use the results of Jeitschko and Sleight^{1,5} for the crystal structure and thermal displacement parameters of α - and β -SnWO₄ at room temperature. α -SnWO₄ has an orthorhombic crystal structure with the space group

FIG. 1. Projection of the α -SnWO₄ structure. The distorted SnO₆ octahedra are shown on the left and the unit cell on the right. The atomic positions together with the *z* coordinates ($\times 10^3$) in the unit cell are from Ref. 5.

 $D_{2h}^6(P_{nna})$ and lattice constants $a=5.6270(3)$ Å, *b* $=$ 11.6486(7) Å, and $c = 4.9973(3)$ Å. The unit cell contains four formula units. Figure 1 shows a projection of the structure in the *z* direction. The positions of the atoms are from Jeitschko and Sleight.⁵ The distorted $SnO₆$ octahedra are also shown in Fig. 1, and the distorted $WO₆$ octahedra $($ not shown in Fig. 1 $)$ are linked at the four corners. In this manner, they form sheets of $[WO₄]^{2–}$ polyanions which are held together by Sn atoms with the formal valency of $+2$. The dark-red color of α -SnWO₄ could be due to the charge transfer involving an electron transfer from Sn^{2+} to W^{6+} (Ref. 5). A value of 3.05 eV was obtained for the band gap *Eg* corresponding to direct allowed transitions from spectral optical constants *n* and *k* computed for an α -SnWO₄ thin film in the visible range from measured spectral transmittance and reflectance data.⁹ No results were found in the literature for the electronic structure of α -SnWO₄ from band-structure calculations.

The temperature factor T_l (**H**) of the *l*th atom in the unit cell at the x-ray reflection **H**, in the expression for the structure factor $F(\mathbf{H})$, has the general expression for a harmonic $crystal¹⁰$

$$
T_l(\mathbf{H}) = \exp(-2\pi^2 \mathbf{H}^T B(l)\mathbf{H}),\tag{1}
$$

where $B(l) = \langle \mathbf{u}(l) \cdot (\mathbf{u}(l))^{T} \rangle$ is the mean-square displacement matrix of the *l*th atom. The element $\langle u_i u_j \rangle$ represents the average value of the atomic displacements along the Cartesian *i* axis multiplied by the displacements along the *j* axis. Usually, least-squares refinement programs yield a list of the anisotropic temperature-factor coefficients β_{ij} for the atoms in the unit cell because the mean position of the atom *l* in the unit cell is not usually introduced in terms of its Cartesian components, but rather as fractional coordinates of the unitcell axes **a**, **b**, and **c**. The temperature factor now takes the form

$$
T_{l}(\mathbf{H}) = \exp\left(-\sum_{i} \sum_{j} h_{i} h_{j} \beta_{ij} \right),
$$
 (2)

where h_i and h_j refer to the reciprocal-lattice vector **H**. β is a symmetric matrix whose dimensionless elements are defined with respect to the unit-cell axes. The relationship between *B* and β is¹⁰ $B = (1/2\pi^2)F^T\beta F$, where the uppertriangular matrix F has the lengths of the unit-cell axes as elements on the main diagonal and all off-diagonal elements are zero in the case of orthorhombic (and cubic) lattice. The anisotropic temperature-factor coefficients β_{ij} are given for Sn, W, O(1), and O(2) atoms (Fig. 1) of the α -SnWO₄ structure at room temperature in Ref. 5 $[\beta_{ij}]$'s are isotropic for $O(1)$ and $O(2)$ atoms].

 β -SnWO₄ has a cubic crystal structure with the space group T^4 ($P2_13$) and a lattice constant $a=7.2989(3)$ Å.¹ The unit cell contains four formula units. Figure 2 shows a projection of the structure in the *z* direction. The positions of the atoms are from Jeitschko and S leight.¹ The distorted $SnO₆$ octahedra around the Sn atoms are also shown in Fig. 2. The structure can be described as an arrangement of $[WO₄]²$ tetrahedra interspersed with $Sn²⁺$ ions. The $[WO₄]²$ tetrahedra are not connected, i.e., they do not share oxygen atoms. The W, Sn, and $O(1)$ atoms in Fig. 2 are situated on the threefold axes (main diagonals) of the unit cell, while the $O(2)$ atoms in the general positions fill the remaining places. The anisotropic temperature-factor coefficients β_{ij} are given for Sn, W, O(1), and O(2) atoms (Fig. 2) at room temperature in Ref. 1 $[\beta_{ij}]$'s are isotropic for O(1) and $O(2)$ atoms].

FIG. 2. Projection of the β -SnWO₄ structure. The distorted $SnO₆$ octahedra are shown on the left and the unit cell on the right. The atomic positions together with the *z* coordinates ($\times 10^2$) in the unit cell are from Ref. 1.

III. EXPERIMENT

The fabrication of stannous tungstate powder began by mixing equimolar quantities of SnO and $WO₃$ powders and by ball milling in acetone for 1 h. For the synthesis of α -SnWO₄ powder the mixture was sealed in a silica tube in vacuum and heated at 600 °C for 15 h. Another method to obtain fusion was to heat the mixture in an argon atmosphere at 600 °C for 15 h. The high-temperature β phase was obtained as metastable at room temperature by a rapid quenching of the mixture sealed in an evacuated silica tube after heating at 800 °C for 15 h and slow cooling to 700 °C. ^{119}Sn Mössbauer spectroscopy studies were carried out at room temperature with a Mössbauer spectrometer working in the acceleration mode with sources on each side of the vibrator. The velocity calibration spectrum was measured with a 57 CoRh source and an α -Fe absorber at room temperature. The source used for the ¹¹⁹Sn spectrum was Ca¹¹⁹SnO₃ (2) mCi).

Mössbauer spectra in the transmission geometry were measured from samples with the amount of Sn of around 30 mg cm^{-2} in the radiation direction. To spread out the stannous tungstate powder over the whole measuring area the sample was mixed with BN to form a homogeneous mass. This was, in turn, pressed into a 2-mm-thick plate kept in the holder by a thin aluminium layer on each side of the plate. To avoid texture effects the measurement was conducted at the "magic angle" (54.7°) .¹¹

To obtain more information about the α -SnWO₄ structure two further experiments were conducted. The samples used contained the α -SnWO₄ powder fused in vacuum, since this powder had narrower line width [full width at half maximum $(FWHM)$] than the one fused in the argon atmosphere. An experiment was carried out on a cube (1 cm^3) which was gently filled with BN and 13 mg Sn per cm³. Measurements were made in the *x*, *y*, and *z* directions in order to check possible texture effects. The second experiment was conducted on a plate with only 3 mg Sn per $cm²$ to minimize saturation effects.

Raman-scattering measurements were performed at room temperature using the Jobin Yvon T-6400 triple Raman spectrometer equipped with a CCD detector. An Ar-ion laser with a wavelength of 514.532 nm was employed and the measurements were carried out with the backscattering configuration. The micro-Raman utility was used to measure the spectra under a microscope. The Raman spectra were measured from discs of pressed powders with a diameter of 1 cm.

X-ray diffraction (XRD) and EDS were used for the characterization of the crystal structure and composition of the stannous tungstate powders, respectively. EDS measurements were performed using the JEOL JSM-6400/LINK AN 10-85 spectrometer and the Philips PW 1353/00 diffractometer with Cu $K\alpha$ radiation was used for the recording of XRD patterns.

For electrical and gas-response measurements some thickfilm samples were made from various $SnWO₄$ powders by screen printing on alumina substrates having preprinted gold electrodes and a Pt-heating resistor on the reverse side of the substrate.

FIG. 3. XRD patterns from α -SnWO₄ powders fused (a) in vacuum and (b) in argon, and (c) from metastable β -SnWO₄ powder fused in vacuum.

IV. RESULTS AND DISCUSSION

A. XRD and EDS experiments

X-ray-diffraction patterns measured from α - and β -SnWO₄ powders are shown in Fig. 3. Figures 3(a) and 3(b) show the patterns from the α -SnWO₄ powders fused in vacuum and in argon atmosphere, respectively. The identification of the α -SnWO₄ phase is not ambiguous since practically all reflections in both patterns belong to this phase. In addition, the patterns are very identical to each other and it is impossible to see any differences in the structure of the two powders. An estimate for the average grain size of the powders is possible to obtain from the patterns by the Scherrer's formula. It gave the same value of about 40 nm for the average grain size in both α -SnWO₄ powders in Fig. 3. However, there were considerable differences in the electrical and gas-response properties between the two powders.²

The XRD pattern from the metastable β -SnWO₄ powder fused in vacuum is shown in Fig. $3(c)$. The identification of the high-temperature β phase is also not ambiguous. Some very small reflections, e.g., from the rutile structure of $SnO₂$ at the 2θ angles of 26.6° , 33.9° , and 51.8° , are seen in the pattern in Fig. 3(c). The average grain size of the β -SnWO₄ powder was also about 40 nm. The EDS results for the composition of both α - and β -phase powders and screen-printed thick films agreed with the chemical formula unit $SnWO₄$.

FIG. 4. Raman spectra from α -SnWO₄ powders fused (a) in vacuum and (b) in argon, and (c) from metastable β powder fused in vacuum.

B. Raman experiments

The characteristic Raman spectra of the α -SnWO₄ powders fused in vacuum and in argon atmosphere, respectively, in Figs. $4(a)$ and $4(b)$ are very similar, except for the intensities. In the case of the vacuum-fused powder, the intensity of the Raman peaks is higher and the spectrum is more detailed with narrower peak widths. The spectra were fitted with a curve fit program in order to determine the frequencies of the peaks. The results from the peak fit are seen in Fig. 4 for the main peaks. The α -SnWO₄ structure has a strong Raman mode at 779 cm^{-1} and a narrow one at 59 cm^{-1} . Sixteen peaks were obtained from the fit of the spectra for the α phase.

The Raman spectrum from the β -SnWO₄ powder fused in vacuum is shown in Fig. $4(c)$ together with peak frequencies from the curve fit. The β -SnWO₄ structure shows more resolved peaks than the α phase. It shows two characteristic narrow peaks, one at a high frequency of 955 cm^{-1} and another at a low frequency of 54 cm^{-1} . There were 16 peaks that were obtained from the fit of the spectrum. Raman spectra from the initial SnO and $WO₃$ powders together with a spectrum from the $SnO₂$ cassiterite structure (rutile-type) are shown in Fig. 5, for comparison. The Raman peaks in the spectra from the tin oxides in Fig. 5 are very weak as com-

FIG. 5. Raman spectra from WO_3 , SnO, and SnO₂ powders. The intensity values of the WO_3 spectrum are multiplied by a factor of 1/10.

pared with the peaks in the spectrum of the WO_3 powder. Therefore, the main peaks in the Raman spectra of α - and β -SnWO₄ may originate in the phonons which are closely related to the vibrations of tungsten atoms.

No data was found in the literature for the Raman spectra and phonon symmetries in the $SnWO₄$ phases. Therefore, we determined the symmetries for the normal modes at the center Γ of the Brillouin zone in α - and β -SnWO₄. Now, the group theory considerations are restricted to those normal modes transforming as the irreducible representations of the point group of the space group, which is isomorphic to the little group at $k=0$. For the determination of the phonon symmetries at Γ it is possible to use the mechanical or Δ representation at $\mathbf{k} = 0$. It is a $3r \times 3r$ matrix representation $\Delta^{(\Gamma)}$, where *r* is the number of atoms in the crystal basis $(r=4\times6=24$ in both α - and β -SnWO₄).

The reduction of the representation $\Delta^{(\Gamma)}$ into a direct sum of allowable irreducible representations of the point group of the space group determines the specific representations related to the symmetry of all normal modes which occur at the zone center Γ , where the little group is the factor group of the space group with respect to the crystal translation group. The number of times a_j , the *j*th irreducible representation, appears in the direct-sum reduction of $\Delta^{(\Gamma)}$ is given by the equation

$$
a_j = h^{-1} \sum_{\kappa=1}^m N_{\kappa} [\chi^{(j)}(C_{\kappa})]^* \chi^{(\Delta)^{(\Gamma)}}(C_{\kappa}), \tag{3}
$$

where *h* is the order of the point group, *m* is the number of classes in the point group, N_k is the number of elements in the class C_K , and $\chi^{(j)}(C_K)$, and $\chi^{(\Delta)^{(\Gamma)}}(C_K)$ are the characters of the elements in the class C_k in the *j*th irreducible representation of the point group and mechanical representation $\Delta^{(\Gamma)}$, respectively. Values of $\chi^{(j)}(C_{\kappa})$ for different *j*

TABLE I. Character table together with normal modes at $\mathbf{k} = \mathbf{0}$ and selection rules for α -SnWO₄. Bases of the irreducible representations of the point group D_{2h} are also given. R and IR indicate Raman-active and IR-active modes, respectively. $t_0 = (000)$; $t_1 = (a/2\ 00)$; t_2 $= (0 b/2 0); t_3 = (a/2 0 c/2); t_4 = (0 b/2 c/2).$

Space group D_{2h}^6				Group operations					Modes at $k=0$			
Irreps (and $\Delta^{(\Gamma)}$	Basis	$\{E t_0\}$	${C_{2z}} t_0\}$	${C_{2y}} t_2\}$	${C_{2x} t_0}$	$\{I t_0\}$	$\{\sigma_z t_1\}$	$\{\sigma_y t_3\}$	$\{\sigma_x t_4\}$	Acoustic	Optic	R/IR
A_g	x^2, y^2, z^2										8	R
B_{1g}	xy			-1	-1			- 1	-1		9	R
B_{2g}	xz		-1		-1		-1		-1		10	R
B_{3g}	yz		-1	-1			-1	- 1			9	R
A_u	xyz					$-$	$ \overline{}$	$ \overline{}$	-1		8	
B_{1u}	$\ensuremath{\mathnormal{Z}}$			$\qquad \qquad -$	-1	-1	- 1				8	IR
B_{2u}	\mathcal{Y}		-1		-1	-1		-1			9	IR
B_{3u}	\boldsymbol{x}		-1	$-$		-1			-1		8	IR
$\Delta^{(\Gamma)}$		72	-4	θ	-4	$\overline{0}$	θ	$\mathbf{0}$	θ			

and κ are given in standard character tables. The character (trace) of the representation $\Delta^{(\Gamma)}$ for the operation $\{\varphi|t\}$ is given by the equation 12

$$
\chi^{(\Delta)^{(\Gamma)}}(C_{\kappa})(\{\varphi|t\}) = \pm (1 + 2 \cos \varphi) \sum_{\kappa=1}^{r} \delta_{\kappa, \kappa_{\varphi}}, \qquad (4)
$$

where each of the basis indices κ and κ_{φ} refer to one of the *r* basis atoms and

$$
\delta_{\kappa,\kappa_{\overline{\varphi}}} = \begin{cases} 1, & \text{if } \kappa \text{ and } \kappa_{\overline{\varphi}} \text{ are identical basis nuclei} \\ & \text{belonging to the same Bravais sublattice.} \\ 0 & \text{otherwise,} \end{cases}
$$

The sum gives the number of basis particles left unchanged, or sent into equivalent particles in the same Bravais sublattice. The other factor in Eq. (4) arises from the trace of the rotation matrix where φ is the angle of rotation of the operation $\{\varphi|0\}$. The \pm is for the proper/improper element, respectively. In the case of symmorphic structure of the space group, it is possible to write Eq. (4) at $\mathbf{k} = \mathbf{0}$ in the simple form

$$
\chi^{(\Delta)^{(\Gamma)}}(\{\varphi|0\}) = \pm n(\varphi)(1+2\cos\varphi),\tag{5}
$$

where $n(\varphi)$ is the number of basis atoms left unchanged by the element $\{\varphi|0\}$. The use of character tables^{13,14} together with Eq. (3) and characters of $\Delta^{(\Gamma)}$, calculated either with Eqs. (4) or (5) , makes it possible to reduce the mechanical representation into a direct sum of the irreducible representations of the point group of the space group.

The space group of the orthorhombic α phase of SnWO₄ is the nonsymmorphic D_{2h}^6 with the point group D_{2h} . The character table of the point group D_{2h} together with the characters of the mechanical representation $\Delta^{(\Gamma)}$, calculated by Eq. (4) , are shown in Table I. The direct-sum reduction of the mechanical representation $\Delta^{(\Gamma)}$ is $8A_g \oplus 9B_{1g} \oplus 10B_{2g}$ \oplus 9*B*_{3*g*} \oplus 8*A_u* \oplus 9*B*_{1*u*} \oplus 10*B*_{2*u*} \oplus 9*B*_{3*u*}, as calculated from the characters in Table I with Eq. (3) . The group operations given in Table I are performed with respect to different origins which are given in the International Tables for x-ray crystallography. Even (gerade) representations only are Raman active, and three acoustic modes correspond to a representation $B_{1u} \oplus B_{2u} \oplus B_{3u}$. Optical modes B_{1u} , B_{2u} , and B_{3u} are IR active and eight optical A_u modes are silent. In all, we have the direct sum reduction $8A_g \oplus 9B_{1g} \oplus 10B_{2g} \oplus 9B_{3g}$ $\oplus 8A_u \oplus 8B_{1u} \oplus 9B_{2u} \oplus 8B_{3u}$ for the 69 optical normal modes, of which 36 are Raman active.

The space group of the cubic β phase of SnWO₄ is the nonsymmorphic T^4 with the point group *T*. The character table of the point group *T* together with the characters of the mechanical representation $\Delta^{(\Gamma)}$, calculated from Eq. (4), are shown in Table II. The direct-sum reduction of the mechanical representation is $6\Gamma_1 \oplus 6\Gamma_2 \oplus 6\Gamma_3 \oplus 18\Gamma_4$, as calculated from the characters in Table II with Eq. (3) . The group operations given in Table II are all performed with respect to the same origin (Fig. 2). Optical modes Γ_1 (or *A* in Mulliken notation), Γ_2 and Γ_3 are Raman active and the modes Γ_4 are only IR active. Three acoustic modes correspond to a representation Γ_4 (or *T* in Mulliken notation). The complexconjugate representations Γ_2 and Γ_3 are bracketed together and labeled as a two-dimensional representation E (Mulliken notation) in Table II, because the time-reversal symmetry makes them degenerate. In all, we have the direct-sum reduction $6\Gamma_1 \oplus 6\Gamma_2 \oplus 6\Gamma_3 \oplus 17\Gamma_4$ for the 69 optical normal modes, of which 18 are Raman active.

C. Mo¨ssbauer experiments

Figures $6(b)$ and $6(c)$ show characteristic Mössbauer spectra of the α -SnWO₄ powders fused in vacuum and in the argon atmosphere, respectively. For each spectrum, the M ossbauer data analysis program¹⁵ was used to analyze the data. The small peak at around 0 mm s^{-1} was treated as a doublet and is called here the $SnO₂$ component, although other phases such as $Sn₂O₃$ and $Sn₃O₄$ containing $Sn⁴⁺$ ions may also result from the oxidation and decomposition of SnWO4. The larger peaks in Fig. 6 were also treated as a doublet with an intensity difference in the two peaks, this ratio being called $G-K$ (see below). The full width at the half maximum (FWHM) was set to be equal for the two doublets. The initial values for the doublet splitting ΔE_Q , the chemical isomer shift δ , the FWHM, and the intensities were given and the fittings made. The results are given in Table III.

TABLE II. Character table together with normal modes at $k=0$ and selection rules for β -SnWO₄. Bases of the irreducible representations of the point group *T* are also given. R and IR indicate Raman-active and IR-active modes, respectively. $\omega = \exp(-2\pi i/3)$; t_0 $= (000); t_1 = (a/2 a/2 0); t_2 = (a/2 0 a/2); t_3 = (0 a/2 a/2).$

Space group T^4				Group operation		Modes at $k=0$					
Irreps (and $\Delta^{(\Gamma)}$		Basis	$\{E t_0\}$	$\{C_{2z} t_0\},\$ $\{C_{2y} t_2\},\$ ${C_{2x} t_0}$	${c_3^{xyz} t_0}, {c_3^{xyz} t_2}$ ${c_3^{\bar{xy}\bar{z}} t_1}, {c_3^{\bar{xy}z} t_3}$	${c_3^{2xyz}} t_0\}, {c_3^{2x\overline{yz}}} t_3\}$ ${c_3^{2\bar{x}y\bar{z}} t_2}, {c_3^{2\bar{x}yz} t_1}$	Acoustic	Optic	R/IR		
A								6	R		
\boldsymbol{E}	Γ_2 Γ_3				ω ω^2	ω^2 ω		6 6	R R		
\boldsymbol{T} $\Delta^{(\Gamma)}$	Γ_{4}	x, y, z	72	$-$ $\overline{0}$	Ω θ	θ $\overline{0}$		17	$_{\rm IR}$		

The linewidth in the spectrum of the α -SnWO₄ powder obtained by fusion in vacuum was narrower than the one in the spectrum of the powder fused in the argon atmosphere. No significant differences were detected in the Mossbauer spectra in Fig. 7 measured in three edge directions of the cube (see Experiment). The differences were inside the margin of error, i.e., the peak asymmetry for the doublet of α -SnWO₄ was not due to texture effects. The fitted results for the three spectra in Fig. 7 together with fitted results for a spectrum of the thin sample (see Experiment) are also presented in Table III. The FWHM was, as expected, smaller in the spectrum of the thin sample than in the ordinary one.

Figure $6(a)$ shows a characteristic Mössbauer spectrum of the β -SnWO₄ powder. It also consists of two doublets where

FIG. 6. ¹¹⁹Sn Mössbauer spectra from (a) metastable β -SnWO₄ powder fused in vacuum and from α -SnWO₄ powders fused (b) in vacuum and (c) in argon.

the small peak at around 0 mm s^{-1} was treated as a doublet corresponded to the $SnO₂$ component and the larger peaks were also treated as a doublet with some asymmetry in their intensities $(G-K$ ratio) belonging to the β -SnWO₄ phase. The FWHM was set to be equal for the two doublets and the fitted results are presented in Table III.

1. Central doublet

For all Mössbauer spectra we interpret the small doublet at around 0 mm s^{-1} to originate in some extra phase(s) containing Sn^{4+} ions. The chemical isomer shift δ is about 0.01 mm s⁻¹ and its electric quadrupole splitting ΔE_Q is about 0.54 mm s^{-1} in Table III. These values are in good agreement with the published data for the crystalline $SnO₂$ $\left[\delta = 0.02(2) \text{ mm s}^{-1} \text{ and } \Delta E_{Q} = 0.50(1) \text{ mm s}^{-1} \text{ (Ref. 16)} \right]$. The intensity of this $SnO₂$ component is roughly around 18% in Table III, but the amount of this phase component is not necessarily 18%. The intensity is a product of the amount and the relevant Mössbauer recoil-free fraction *f* (see below). For polycrystalline SnO₂, the *f* factor at room temperature is reported to be 0.56^{16} . The values of the *f* factor for the $SnWO₄$ phases are unknown in the literature, but the calculations done here using reported x-ray temperature factors (see below) gave values of around $0.16-0.20$ for the f factors. Using these values and 0.56 for the $SnO₂$ component gives around 7% for the amount of the extra phase (s) .

In the case of very small grains of the extra $phase(s)$ at the surface of the $SnWO₄$ matrix, it is not possible to see their reflections in the XRD patterns. However, they may have a strong influence on the chemical surface properties. In conversion electron Mössbauer spectra (CEMS) from α -SnWO₄ thin films, made by sputtering, the central doublet was much higher than the doublet from the α -SnWO₄ phase after annealing the films at 400 $^{\circ}$ C.³ However, in the XRD patterns of the films there were high reflections only from the α phase and very small reflections from $SnO₂$ and some other phases. The reason may relate to the fact that CEMS gives information which is more surface sensitive and sputtered surfaces are more sensitive to decompose during annealing. Large differences were found in the gas-response properties between α -SnWO₄ thin films, made by sputtering, and thickfilms, made by screen printing from fused powders, with similar XRD patterns but very different Mössbauer spectra.³

TABLE III. Results from the fitting of Mössbauer spectra for Sn^{4+} and Sn^{2+} cations in various α - and β -SnWO₄ powder samples (see Experiment). FWHM is for the linewidth, *I* is for the fractional intensity between Sn^{4+} and Sn^{2+} doublets, δ is for the chemical isomer shift, ΔE_Q is for the electric quadrupole splitting, and $G-K$ is the intensity ratio of the Sn^{2+} doublet.

			Parameters for Sn ⁴⁺		Parameters for Sn^{2+}			
Sample	FWHM $(mm s^{-1})$ ± 0.01	\boldsymbol{I} (%) ±1.0	δ $\rm (mm \; s^{-1})$ ± 0.02	ΔE_Q $(mm s^{-1})$ ± 0.03	Ι (%) ±1.0	δ $\rm (mm \ s^{-1})$ ± 0.01	ΔE_O $(mm s^{-1})$ ± 0.01	$G-K$ ±0.02
β -SnWO ₄	0.47	14.0	-0.01	0.49	86.0	3.50	1.30	1.08
α -SnWO ₄ (argon)	0.66	19.5	0.03	0.52	80.5	3.42	1.18	1.08
α -SnWO ₄ (vacuum)	0.58	19.7	0.02	0.55	80.3	3.42	1.18	1.06
α -SnWO ₄ (cube x)	0.60	18.8	0.03	0.59	81.2	3.43	1.18	1.08
α -SnWO ₄ (cube y)	0.63	18.6	0.01	0.56	81.4	3.43	1.18	1.09
α -SnWO ₄ (cube z)	0.63	18.6	0.02	0.50	81.4	3.43	1.18	1.06
α -SnWO ₄ (thin sample)	0.46	18.8	0.01	0.54	81.2	3.43	1.16	1.08

2. Goldanskii-Karyagin effect in α -SnWO₄

The peak-intensity asymmetry is most probably due to the Goldanskii-Karyagin effect.¹⁷ The Mössbauer spectra for the α phase in Figs. 6(b) and 6(c) are very similar irrespectively if the samples were fused in vacuum or in argon, consequently we can deal with them together. From the anisotropic temperature-factor coefficients β_{ii} [Eq. (3)] in Ref. 5 for the Sn atom in α -SnWO₄ it is possible to calculate the mean-square displacement matrix B [Eq. (1)] and diagonalize it to get *B'* with $B'_{11} = \langle u_{x'}^2 \rangle$, $B'_{22} = \langle u_{y'}^2 \rangle$, and B'_{33} $= \langle u_z^2 \rangle$ to represent the principal axes components of the vibrational ellipsoid in the new coordinate system (x', y', z') :

$$
B' = \begin{pmatrix} 1735.0 & 0 & 0 \\ 0 & 1006.96 & 0 \\ 0 & 0 & 712.28 \end{pmatrix} 10^{-5} \text{\AA}^2, \quad (6)
$$

The diagonalization in Eq. (6) means a rotation around the unit-cell **c** axis of the α -SnWO₄ structure and z' and **c** axes are parallel. Figure 8 shows a projection of the distorted $SnO₆$ octahedron in the unit-cell coordinate system of α -SnWO₄. The lone-pair orbital of Sn is in the direction of the *z* axis. Since the lone pair has the strongest effect in the electric-field-gradient (EFG) tensor it can be considered that the principal axis of the EFG tensor coincides with the **c** axis of the unit cell.

Since a principal axis of the mean-square displacement matrix B' is now parallel to the principal axis of the EFG tensor, a general formula for the ratio between the intensities in the two peaks of the Mössbauer doublet for a nonsymmetric electric-field gradient is¹⁸

$$
\frac{I_l}{I_2} = \frac{\int \int P_l(\theta, \varphi) h(\theta) f(\theta, \varphi) \sin \theta \, d\theta d\varphi}{\int \int P_2(\theta, \varphi) h(\theta) f(\theta, \varphi) \sin \theta \, d\theta d\varphi},\tag{7}
$$

where $h(\theta)=1$ in the case of no texture.¹⁹ The quantities $P_1(\theta,\varphi)$ and $P_2(\theta,\varphi)$ are the relative angular-dependent absorption probabilities for the transitions $(1/2, \pm 1/2) \rightarrow (3/2, \pm 1/2)$ \pm 3/2) and $(1/2, \pm 1/2) \rightarrow (3/2, \pm 1/2)$, respectively. In the

FIG. 7. ¹¹⁹Sn Mössbauer spectra from a cubic α -SnWO₄ powder (fused in vacuum) sample measured in the three cube directions *x*, *y*, and *z*.

FIG. 8. Projection of the distorted $SnO₆$ octahedron around the lone-pair tin cation in α -SnWO₄. The lone pair is in the *z* direction (parallel to the $z[′]$ and **c** axes) and is marked by two dots.

case of a randomly oriented crystal the transition probabilities are

$$
P_1(\theta, \varphi) = 4 \sqrt{1 + \frac{\eta^2}{3}} + (3 \cos^2 \theta - 1 + \eta \sin^2 \theta \cos 2\varphi),
$$

$$
P_2(\theta, \varphi) = 4 \sqrt{1 + \frac{\eta^2}{3}} - (3 \cos^2 \theta - 1 + \eta \sin^2 \theta \cos 2\varphi),
$$
 (8)

where η is the asymmetry parameter. For a positive value of the electric quadrupole splitting $\Delta E_Q = 1/2eQV_{zz}$, P_1 corresponds to a higher energy and thus to a larger velocity in the spectrum. For the γ -photon direction (θ, φ) the recoil-free fraction is given as

$$
f(\theta, \varphi) = \exp\{-k^2 [(\langle u_{x'}^2 \rangle \cos^2 \varphi + \langle u_{y'}^2 \rangle \sin^2 \varphi) \sin^2 \theta + \langle u_{z'}^2 \rangle \cos^2 \theta] \},
$$
 (9)

The wave vector is here $k = E_{ph}(\hbar c) = 12.0908 \text{ Å}^{-1}$ using the value $E_{\text{ph}}(^{119}\text{Sn}) = 23.875 \text{ keV}$ of the γ -ray transition for ¹¹⁹Sn. The average recoil-free fraction f can be calculated by numerical integration of Eq. (9) , and the use of the diagonal elements of the matrix B' in Eq. (6) gave $f = 0.20$.

The integrals in Eq. (7) were calculated for different values of the asymmetry parameter η between 0 and 1 and the ratios I_1/I_2 (Fig. 9) were obtained as a function of η . The experimental asymmetry of about 1.08 for α -SnWO₄ in Table III corresponds well with the calculated values that ranges from 1.06 to 1.14 in Fig. 9. The experimental average value of 1.08 corresponds to a value of 0.75 for η in Fig. 9. Furthermore, we can conclude that ΔE_Q is positive. Since the nuclear quadrupole moment Q is negative for 119 Sn the electric-field gradient V_{zz} is also negative.

3. Goldanskii-Karyagin effect in b**-***SnWO***⁴**

The Mössbauer doublet of β -SnWO₄ in Fig. 6(a) is also asymmetric due to the Goldanskii-Karyagin effect. The tin environment is also now a distorted octahedron as is shown in Fig. 2. From the anisotropic temperature-factor coeffi-

FIG. 9. The intensity ratio I_1/I_2 for the two peaks in the Mössbauer doublet of α -SnWO₄, calculated as a function of the asymmetry parameter η from Eq. (7).

cients β_{ij} in Ref. 1 for the Sn atom it is possible to calculate the mean-square displacement matrix *B* and diagonalize it into the form

$$
B' = \begin{pmatrix} 1408.82 & 0 & 0 \\ 0 & 1408.82 & 0 \\ 0 & 0 & 955.4 \end{pmatrix} 10^{-5} \text{\AA}^2.
$$
 (10)

The principal axis $z⁶$ of the vibrational ellipsoid is in the |111| direction in the unit-cell cube. The average recoil-free fraction *f* can be calculated by numerical integration of Eq. (9) and the diagonal elements of the matrix B' in Eq. (10) gave $f = 0.16$.

Figure 10 shows a projection of the distorted oxygen octahedron surrounding the lone-pair tin cation in the β -SnWO₄ structure. The lone-pair orbital of the Sn cation is along the z' ([111]) direction, and as in the α phase the principal axis of the EFG tensor coincides with the principal z' axis of the vibration ellipsoid. Through the Sn position on

FIG. 10. Projection of the distorted $SnO₆$ octahedron around the lone-pair tin cation in β -SnWO₄. The lone pair is in the *z*^{*'*} direction which is parallel to the $[111]$ direction in the unit cell and is marked by two dots.

the cube diagonal in the β -SnWO₄ structure, the cylindrical symmetry gives the value zero for the asymmetry parameter η .

The intensities of the two peaks of the asymmetric doublet can now be calculated by the numerical integration of the integrals in Eq. (7) and the results are $I_1 = 8.547$ and I_2 =7.785, so the ratio I_1 / I_2 =1.098. The experimental asymmetry of 1.08 reported for β -SnWO₄ in Table III corresponds well with the calculated value. Also here are ΔE_Q >0 and V_{zz} < 0 as for α -SnWO₄.

4. Electric quadrupole splittings and chemical isomer shifts

The obtained experimental values for the chemical isomer shift δ and electric quadrupole splitting ΔE_Q in the α - and β -SnWO₄ structures compares well with the values reported by Ballard and Birchall.²⁰ For the α phase there are two oxygen atoms at 2.82 Å [O(1)], two oxygen atoms at 2.18 Å $[O(2)]$ and two oxygen atoms at 2.39 Å $[O(2')]$ around the lone-pair tin cation in Fig. 8. The 5*s* electrons in Sn are thus not bare, but take part in the bonding. The electric quadrupole coupling in covalent tin systems is almost entirely made up by the imbalance in the population of the *p* orbitals and the electron density on the surrounding atoms and ions can be ignored. 21

The electric quadrupole moment *Q* is negative and the electric quadrupole splitting for Sn has been approximated with the expression 22

$$
\Delta E_Q = +4.0[n_z - 0.5(n_x + n_y)] \text{mm s}^{-1}, \quad (11)
$$

where n_x , n_y , and n_z are the populations of different *p* orbitals.

The chemical isomer shift depends on the electron density at the nucleus, and for Sn general formulas have been obtained.21,22 Flinn²³ has suggested the following formula for the chemical isomer shift with the absorber and source at 77 K:

$$
\delta(\text{vs BaSnO}_3) \approx [3.01n_s - 0.20n_s^2 - 0.17n_s n_p - 0.38] \text{m s}^{-1},
$$
\n(12)

where n_s and n_p are the 5*s*- and 5*p*-electron populations. In the case here, the δ versus CaSnO₃ and with the absorber and source at room temperature is reported. Assuming the same Debye temperature for the two tin oxides involved this formula can also be used up to room temperature. In this case stannous ions are assumed, i.e., $n_s + n_p = 2$. Furthermore, $n_p = (n_x + n_y) + n_z$. There are now four equations with four variables n_s , n_p , n_x+n_y , and n_z . The populations can be calculated and the results are $n_s = 1.45$, $n_p = 0.55$, $n_x + n_y$ $=0.17$, and $n_z=0.38$. The results are in agreement with the lone-pair drawing in Fig. 8.

For the β phase there are three oxygen atoms at 2.21 Å $[O(2)]$ and also three at 2.81 Å $[O(1)]$ around the lone-pair Sn cation in Fig. 10. Similarly, as in the α phase above, the populations for the 5s and 5p electrons in β -SnWO₄ can be calculated. The β phase has cylindrical symmetry around the Sn cations and $n_x = n_y$. The results for the population of the 5*s* and 5*p* electrons are $n_s = 1.48$, $n_p = 0.52$, $n_x = n_y$ $=0.065$, and $n_z=0.39$. Again, the results are in agreement with the lone-pair drawing in Fig. 10. The withdrawal of 5*s* electrons from Sn is larger for Sn in the α phase than in the β phase, as expected from the surrounding oxygen octahedra in Figs. 8 and 10.

V. CONCLUSIONS

¹¹⁹Sn Mössbauer and Raman spectroscopies together with x-ray-diffraction measurements were used to study the phase structure of $SnWO_4$ powders fused from SnO and WO_3 powders in vacuum and in argon into the α - and β -SnWO₄ phases. X-ray-diffraction patterns were incapable to reveal small extra phase(s) from the oxidation of Sn^{2+} ions into the Sn^{4+} form in the α - and β -phase powders. Raman spectroscopy as a local probe was also insensitive to expose the extra phase component (s) in the powders. The reason may be in the low intensities of Raman peaks in spectra from various tin oxides such as SnO and SnO₂. Mössbauer spectroscopy, on the other hand, is sensitive to changes in the valence state and bonding and revealed clearly the extra $phase(s)$ containing Sn^{4+} ions in the powders. An asymmetry was found between the two peaks in the Mössbauer doublet at both $SnWO₄$ phases and it was possible to relate the asymmetry to the Goldanskii-Karyagin effect in both α and β phases. The values 0.20 and 0.16 were calculated for the average recoilfree fraction f from published thermal parameters for α and β phases respectively, and a value of around 0.75 was obtained for the asymmetry parameter η in α -SnWO₄. The 5*s*and different 5*p*-electron populations in the stannous ions in the α - and β -phase structures were calculated from the obtained values for the electric quadrupole splitting and chemical isomer shift, and the results exposed the lone-pair character of the tin cations in both phases. Raman spectra are also given for both α - and β -phase powders together with peak frequencies from a curve fit. Normal-mode symmetries together with selection rules are also given for α - and β -SnWO₄ at **k**=0.

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