Total energy differences between SiC polytypes revisited

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The total energy differences between various SiC polytypes (3*C*, 6*H*, 4*H*, 2*H*, 15*R*, and 9*R*) were calculated using the full-potential linear muffin-tin orbital method using the Perdew-Wang generalized gradient approximation [I. P. Perdew, in *Electronic Structure of Solids '91*, edited by P. Ziesche and H. Eschrich (Akademie-Verlag, Berlin, 1991), p. 11] to the exchange-correlation functional in the density-functional method. Numerical convergence versus **k**-point sampling and basis-set completeness are demonstrated to be better than 0.5 meV/atom. The parameters of several generalized anisotropic next-nearest-neighbor Ising models are extracted and their significance and consequences for epitaxial growth are discussed. $[$ S0163-1829(98)01019-4]

I. INTRODUCTION

Despite many years of study, the origin of polytypism in SiC is still not completely understood. A much debated question is whether polytypism is a manifestation of kinetic factors during growth or whether polytypes should be viewed as distinct (possibly metastable) thermodynamic phases with a specific stability range of external parameters (such as pressure and temperature). In a thermodynamic approach to the problem, the most important quantities are the total freeenergy differences between the various polytypes. A major contribution to the latter is the energy difference at zero temperature. Vibrational entropy contributions at higher temperature were discussed by Heine *et al.*^{1,2} and Zywietz *et al.*³ Several groups have performed first-principles local-densityfunctional calculations of these energy differences using the norm-conserving pseudopotential plane-wave method. $4-8$ However, there are significant discrepancies between the results of various calculations for these energy differences, which are of order of a few meV/atom or less. More seriously, the three more recent calculations appear to invalidate some of the important conclusions drawn from these calculations by the early work of Heine *et al.*¹

Heine *et al.*¹ discussed the relative energy of polytypes in terms of a generalized anisotropic next-nearest-neighbor Ising (ANNNI) spin model in which the energy of a given polytype (per atom) is written as

$$
E = E_0 - \frac{1}{N} \sum_{i,n} J_n \sigma_i \sigma_{i+n}, \qquad (1)
$$

in which *N* is the number of layers in the system, a ''spin'' $\sigma_i = \pm 1$ is associated with each (close-packed) SiC double layer such that parallel spins represent a locally cubic stacking, and antiparallel spins represent a locally hexagonal stacking. The parameters J_n represent the interlayer interaction between succesively farther removed layers and E_0 is a common energy reference. In terms of this model truncated beyond $n=3$, the energies of some of the polytypes of interest are given in column 2 of Table I. According to Heine *et al.*, ¹ what distinguishes SiC from other semiconductors and leads to the multitude of stacking arrangements consti-

tuting polytypism is that $J_1 \cong -2J_2 > 0$ (with $J_n \ll J_{1,2}$ for *n* $>$ 2). For this special ratio of J_1 / J_2 , a multiphase degeneracy point occurs in the ANNNI model corresponding to all phases consisting of successive bands of two or three parallel spins (which in the following we will call 2-3 banded polytypes). This would explain the relatively frequent occurrence of polytypes such as $6H$ (which is $\langle 3 \rangle$ in Zhdanov notation,⁹ indicating that it consists of bands of three parallel spins), $4H$ or $\langle 2 \rangle$, and 15*R* or $\langle 32 \rangle$, in contrast to polytypes such as 8*H* or $\langle 4 \rangle$, 10*H* or $\langle 5 \rangle$, and 2*H* or $\langle 1 \rangle$, which are rather rare. Surprisingly, recent calculations^{6–8} found that $J_1 \leq |J_2|$, a condition very far away from the multiphase degeneracy point. Furthermore, in some of these results, $6,7$ the energy difference $E_{2H} - E_{3C}$ is found to be smaller than the energy difference between 3*C* and any of the other low-energy polytypes. This makes it difficult to understand why 2*H* is such a rare polytype.

The purpose of the present paper is to systematically reevaluate these energy differences of polytypes including some not previously calculated ones and to discuss the meaning of the ANNNI model parameters in the light of these results. Since the accuracy is a crucial matter here, we next discuss the computational method and associated convergence parameters in some detail.

II. COMPUTATIONAL METHOD AND CONVERGENCE TESTS

The computational method employed is the full-potential linear muffin-tin orbital method as implemented by Methfessel¹⁰ and van Schilfgaarde.¹¹ The total energy is calculated using the density-functional method using the generalized gradient approximation (GGA) for the exchangecorrelation energy of Perdew and Wang.¹² For the E_{2H} $-E_{3C}$ energy difference, which is of particular concern below, we verified that other exchange-correlation functionals, such as the Langreth-Mehl GGA (Ref. 13) and the Ceperley-Alder¹⁴ and Hedin-Lundqvist¹⁵ parametrizations of the local-density approximation, yield results that do not differ by more than 1 meV/atom from those for the Perdew-Wang GGA adopted in the rest of this paper.

In all results presented below, we used the ideal struc-

P	ANNNI	FP LMTO	$n_{max} = 2^{\alpha}$	$n_{max} = 3^{\rm b}$	K term	$n_{max} = 3 + K^{c}$
2Н	$2J_1 + 2J_3$	2.7	2.7	2.7	θ	2.7
4H	$J_1 + 2J_2 + J_3$	-1.2	-1.2	-1.2	Ω	-1.2
6H	$\frac{2}{3}J_1+\frac{4}{3}J_2+2J_3$	-1.05	-0.08	-1.05	$-\frac{4}{3}K$	-1.05
9R	$rac{4}{3}(J_1+J_2)$	1.0	0.1	0.3	$-\frac{4}{3}K$	1.0
15R	$\frac{4}{5}(J_1+2J_2+2J_3)$	-1.5	-1.0	-1.1	$-\frac{4}{5}K$	-1.1
15R'	$\frac{4}{5}(2J_1+J_2+J_3)$		1.1	1.3	$-\frac{4}{5}K$	1.6
8H	$\frac{1}{2}(J_1+2J_2+3J_3)$		-0.6	-0.8	$-K$	-0.9
10H	$\frac{2}{5}(J_1+2J_2+3J_3)$		-0.5	-0.6	$-\frac{4}{5}K$	-0.7

TABLE I. Energy difference $\Delta E_P = E_P - E_{3C}$ for various polytypes *P* in meV/atom.

^aUsing $J_1 = 1.350$, $J_2 = -1.285$, and $J_3 = 0$, extracted from the first two polytypes.

^bUsing J_1 = 1.528, J_2 = -1.285, and J_3 = -0.177 meV/atom, extracted from the first three polytypes.

^cUsing J_1 = 1.781, J_2 = -1.275, J_3 = -0.431, and $K = -0.244$ meV/atom, extracted from the first four polytypes.

tures, but relaxed the total energies with respect to volume. All polytypes were found to closely obey the expected relation $a_h = a_c / \sqrt{2}$ and a_c was found to be 4.33 Å, within 1% of the experimental value. To check the uncertainties introduced by using ideal structures, we performed relaxations for 2*H* SiC. We found $c/a = 1.644$, which is slightly larger than the ideal ratio $c/a = \sqrt{8/3} = 1.633$, in good agreement with experiment,¹⁶ which gives $c/a = 1.641$. We obtain *u* $=0.3745$, which is very close to the ideal value of $3/8$. The important point is that the total energy per atom in 2*H* was reduced by only 0.6 meV/atom by relaxation of the structure. In the above calculation, an intracell parameter *u* relaxation was performed for each c/a . This energy lowering is consistent with the value estimated from the elastic constants for a distortion from the minimum energy $c/a = \eta$ to the ideal c/a , given by $\Delta E = (1/9)(\delta \eta/\eta)^2 \Omega [C_{33}-2C_{13}+(C_{11}+C_{12})/2],$ in which Ω is the volume per Si-C pair. This expression equals 0.56 meV/atom using the elastic constants given in Ref. 17. Since other polytypes of type 2*nH* are found experimentally to have c/a values closer to the ideal value of $n\sqrt{8}/3$, 2*H* is the extreme case and places an upper limit on the errors introduced by using ideal structures. The effect of internal cell structural parameters is even an order of magnitude lower. From the TO-phonon frequency in SiC of 23 THz,¹⁷ we can estimate the force constant k for the Si-C bond-length distortions to be 18 eV/ \AA ². Thus the change in energy per bond related to a bond-length change of δl $= (\delta u)c$ is $\Delta E = (1/2)k(\delta l)^2$. This gives only 0.03 meV/ atom for our calculated δu , in good agreement with our direct calculation. High-precision theoretical determinations of the atomic relaxations were reported by Kackel *et al.*⁷ They confirm that the bond lengths differ by less than 0.3% from the ideal bond length, which according to the above estimate would give at most 0.14 meV/atom for the relaxation energy. As far as atomic relaxation effects on the total energies are concerned, our results differ substantially from theirs. In their results without atomic relaxations or cell-shape relaxations, 2*H* lies about 8 meV/atom above 3*C* and the energy of the polytypes increases monotonically with hexagonality. They find the internal cell atomic relaxations to have a marked effect on all hexagonal polytypes resulting in a lower energy than 3*C* for 6*H* and 4*H* and a substantial reduction of the 2*H* to 3*C* energy difference to only 1 meV/atom. This implies that the relaxations would produce relative changes of 3, 4, and 7 meV/atom for 6*H*, 4*H*, and 2*H*, respectively. This is inconsistent with the above estimates based on elastic and force constants and with our explicit calculations. The origin of this discrepancy is not entirely clear. However, the comparison between their relaxed and unrelaxed energies is complicated by their use of different **k**-point sets for the two calculations. As will be shown below, converged Brillouin zone integrations are an important requirement for drawing conclusions about polytype energy differences. As we will show below, we find 6*H* and 4*H* to have lower energy than 3*C* without cell shape or internal position relaxations.

With respect to self-consistency, all total energies were converged to better than 0.1 meV/atom. The contributions to the total charge density from each angular-momentum component were converged to a root-mean-square error less than 10^{-4} electron. Within the full-potential (FP) linear muffin-tin orbital method (LMTO), the wave functions are expanded in an extended basis set of muffin-tin orbitals with different spatial decay constants (i.e., spherical Hankel envelope function exponents κ). Figure 1 shows results for different basis sets for the $E_{2H} - E_{3C}$ energy difference and the individual cohesive energies of E_{2H} and E_{3C} . The notation for the basis set is illustrated as follows: *dps* means up to *d* orbitals for the first $\kappa = -0.05$ Ry, up to *p* for the second $\kappa = -1$ Ry, and one *s* orbital for the third $\kappa = -2.3$ Ry. The unfilled bars in the bottom graph give E_{2H} , the filled ones E_{3C} . The top graph gives their energy difference in meV/atom. The dashed lines indicate the corresponding information for the same basis sets with f orbitals added for the first κ . We can see that the contributions of each orbital to the total cohesive energies are several 10 meV and that increasing the basis set decreases the energy. Adding the *f* orbitals makes about a -30 -meV contribution independently of which basis set they are added to. The third κ , *d* orbital contributes only about -7 meV to the total energy. The most important point is that the polytype energy difference is stable at 2.4 ± 0.3 meV/ atom for the four most complete basis sets considered. Adding empty sphere orbitals s and p and the second κ , s to the ddp basis set changed the energies by only -8 meV and is thus also considered ineffective. For polytypes with many atoms per unit cell, the calculations with the basis sets larger than *fdp* tend to become unstable. If the basis set is very

FIG. 1. Basis-set convergence of $E_{2H} - E_{3C}$. The bottom graph shows cohesive energies of 2*H* and 3*C* as open and filled bar graphs with various basis sets as indicated. The top graph shows $E_{2H} - E_{3C}$. Dashed (full) lines are results for which *f* orbitals are (not) included.

close to completeness, slight numerical errors can make the basis set appear overcomplete or linearly dependent. The optimal basis set is thus considered to be *fdp* and used systematically for the other polytypes.

The integrations over the interstitial region are done using an auxiliary set of spherical Hankel functions times spherical harmonics for the expansion of products of two Hankel functions. These expansions are cut off at $l_{max}=6$. We found that this cutoff is necessary to make the results stable and independent of the sphere radii choice. The empty spheres were chosen to be nearly touching with two empty spheres equal in size to the atomic spheres $(Si$ and C being chosen equal) in each cubic stacking double layer unit and a large $(1.134s_{atom})$ and small empty sphere $(0.666s_{atom})$ in each hexagonal unit. The large spheres occupy the empty channel in the wurtzite structure. That is, if atoms are taken to sit in *A* and *B* positions in the basal plane, the large empty spheres occupy the *C* positions in the plane at a height halfway between the bonding Si and C atoms in the *A* position. The small spheres occupy the sites halfway between the Si and C atoms opposite to the nearest-neighbor Si-C bond along the *c* axis. In cubic SiC, the spheres occupy about 68% of the unit cell volume. In 2*H* they occupy 63% of the volume and in other polytypes the filling is in between these values in proportion to the degree of hexagonality $[i.e.,$ the ratio of the number of hexagonally stacked layers *h* to the total (i.e., hexagonal and cubic *c*) number of layers $h/(h+c)$.

The next convergence issue to consider is the Brillouin zone integration. The Monkhorst-Pack¹⁸ special **k**-point sampling technique is used with the number of divisions along reciprocal lattice vectors in the basal plane equal to *N* and along the c axis equal to M . For longer polytypes (along the c axis), one needs fewer divisions M along the c axis. Rather than picking exactly equivalent sets for each polytype and thus counting on error cancellation, we picked *M* large

FIG. 2. Brillouin zone sampling convergence. *N* is the number of divisions along the two basal plane reciprocal lattice vectors. Circles, 2*H*; squares, 4*H*; filled diamonds, 6*H*; upward triangles, 9*R*; downward triangles, 15*R*. The full line curve is a power-law fit $e^{13}N^{-7.4}$. The values of $\Delta E_P(\infty)$ are given in Table I.

enough to ensure absolute convergence. For 2*H*, we used $M = N$ and for longer polytypes we reduced to $M = N/2$ for the larger *N* values. Figure 2 shows the results for various polytypes as a function of *N*. The quantity shown is $\Delta E_P(N) - \Delta E_P(\infty)$, where $\Delta E_P(N) = E_P(N) - E_{3C}$, the energy difference for a given polytype *P* from the absolutely converged value of E_{3C} calculated with $N=M=10$, and the value of $E_p(\infty)$ is estimated by extrapolation so as to ensure that all results fall on a universal curve. This clearly shows that the final values $\Delta E_p(\infty)$ are converged to better than 0.5 meV/atom.

III. RESULTS AND DISCUSSION

The converged energy differences of the polytypes with respect to 3*C*, i.e., $\Delta E_p(\infty)$ as defined in the preceding section, are given in column 3 of Table I. They are compared with those of previous calculations in the literature in Fig. 3.

Next, we extract the J_n parameters. Columns 4 and 5

FIG. 3. Energy differences between various polytypes: comparison with other calculations.

correspond respectively to truncation at $n_{max}=2$ and n_{max} = 3 using the energy differences $E_{2H} - E_{3C}$ and $E_{4H} - E_{3C}$ as input in the first case and additionally $E_{6H} - E_{3C}$ in the latter case. The other polytypes then allow for a check of the consistency of this model. We find the J_n parameters J_1 and *J*2, as listed in Table I, to be nearly independent of whether or not J_3 is included. Furthermore, we find $J_1 > |J_2|$.

Our results are somewhat closer to those of Heine *et al.*¹ than the other recent results, particularly that E_{2H} is higher above E_{3C} by an amount significantly larger than the other polytype energy differences. Also, we find the various 2-3 banded polytypes to be closer to each other than in the other calculations. Nevertheless, our results are far from the multiphase degeneracy point $J_1 = -2J_2$. As Heine *et al.* pointed out, the energy of a twin boundary, i.e., the energy cost of a boundary between all up-spin and all down-spin cubic half crystals, is given by

$$
E_{twin} = 2(J_1 + 2J_2). \tag{2}
$$

According to Heine *et al.*, this is nearly zero and hence explains why many twin boundaries in an otherwise cubically stacked crystal are likely to occur. With our present values of the J_n parameters, the energy cost of a twin is *negative*. This implies that twins are even more favorable than in the model of Heine *et al.* Hence there is no contradiction at all with the observation of a predominance of 2-3 banded polytypes.

Consistently with other recent work we find 4*H* to have lower energy than 6*H*. In the FP LMTO calculations, we find 15*R* as the lowest energy polytype. In the ANNNI model we find 15*R* to lie in between 4*H* and 6*H* with 4*H* the lowestenergy polytype. This is slightly more expected since 15*R* is intermediate in character between 4*H* and 6*H*. This discrepancy, which is smaller than 0.5 meV, may be beyond the accuracy of our FP LMTO calculations in view of the fact that the computational convergence is most challenging for the largest polytype. As expected, the hypothetical^{19,20} 9*R* polytype with a high degree of hexagonality $(66%)$ is found to have higher energy than 3*C*, but lower than 2*H*.

The ANNNI model appears to somewhat underestimate the energy of 9*R*. This suggests that other terms in the effective Hamiltonian may be required. A term

$$
K \sum_{i} \sigma_{i} \sigma_{i+1} \sigma_{i+2}, \sigma_{i+3} \tag{3}
$$

was suggested by Cheng *et al.*⁴ The additional energy for each polytype due to this term is given in column 5 in Table I. Column 6 shows that this term allows us to fit 9*R* exactly without affecting the energy of 15*R* significantly.

We next consider the predictions of the model for a few other polytypes. Another polytype of high hexagonality $(80%)$ was recently considered²⁰ and labeled 15*R'* or $\langle 1112 \rangle$. Its energy within the ANNNI model is given in the bottom section of Table I. As expected, it is higher in energy than $9R$, but still lower than $2H$. We do not interpret this as an indication that these particular periodic stacking arrangements are more likely (because they seem excessively complicated), but rather as an indication that a high density of stacking faults is likely to occur in 2*H*. For any 2*nH* polytype with $n \geq 3$ the energy difference from 3*C* can be written as $(2/n)(J_1+2J_2+3J_3-2K)$. This shows that for $n \rightarrow \infty$, it will approach zero, as expected, since 3*C* corresponds to ∞ *H*, but only very slowly. In fact, the energies of 8*H* and 10*H* are seen in Table I to be still rather close to those of the 2-3 banded polytypes, consistent with the fact that these polytypes have indeed been observed.

As for the phonon contributions to the free energy (here denoted F_p), we note that Heine *et al*.¹ obtain a result that is the opposite of that found by Zywietz *et al.*,³ namely, F_{4H} $>F_{6H}$, and increasing with temperature. This tends to stabilize 6*H* at high temperatures, whereas Zywietz *et al.*³ find 4*H* to become even more stabilized at higher than at lower temperatures without affecting the polytype free-energy ordering. We note that with our calculated $E_{4H} - E_{6H}$ at zero temperature and the values of Heine *et al.* for the phonon contribution, the transition from 4*H* stability to 6*H* stability is predicted to occur above 8000 K, i.e., well above the melting temperature of SiC. With the phonon contributions of Zywietz *et al.*, no stabilization of 6*H* will ever occur. We conclude that either way, there is no substantial evidence from the calculations that the polytypes would have a welldefined temperature stability region. We think that it is much more likely that the slightly different tendencies for 4*H* and 6*H* growth in dependence on the growth temperature are due to kinetic factors. In fact, these experimental tendencies have not unequivocally been established.

Heine *et al.*²¹ also argued that the 3*C* dominance in epitaxial growth could be explained by assuming that only the surface layer stacking is determined by the equilibrium energy condition, but that the stacking is not subsequently rearranged after the layer is buried in the growing crystal. Since the energy difference for adding one surface layer to a substrate with opposite spin of the top layer as opposed to equal spin is $J_s = 2(J_1 \pm J_2)$, with \pm depending on whether the next layer down has equal or opposite spin, cubic stacking is always favored as long as $J_1 + J_2 > 0$. As in the results of Heine *et al.* and in contrast to other recent results, $6-8$ our present results satisfy this requirement, although only barely so. Of course, we caution that these interlayer interactions may change at a surface. If $J_1 + J_2 < 0$, on the other hand, a 4*H* stacking would always be preferred, as can easily be checked by following the same argument as given by Heine *et al.* The point is that second-layer interactions, which are ''antiferromagnetic,'' are then dominant. Thus, if we start from two equal spins in the top layers, the next growing layer must have opposite spin. The new surface then ends in two opposite spins and the following layer must have the same spin as the one last deposited, after which the cycle repeats. This is inconsistent with experimental observations. Independent nucleations on large terraces tend to have the 3*C* structure, which usually is accompanied by a large amount of so-called double positioning boundaries.

Given that the preference for cubic stacking during growth is so small, the question arises whether this is really relevant. To address this question, we must consider size effects of the growing fragments. For a two-dimensional $(2D)$ island of N_i spins (or SiC units), the energy differences for being in a cubic or hexagonal stacking on top of a substrate should be of order N_iJ_s . This implies that up to N_iJ_s $\approx k_B T_G$, with T_G the growth temperature and k_B Boltzmann's constant, or for a typical growth temperature of 1500 K, and using $J_s = 0.3$ meV/SiC unit, up to $N_i \le 600$, there should be virtually no distinction in energy between either stacking. On the other hand, islands will definitely tend to be of a well-defined spin. This is because a lateral spin boundary corresponds ultimately to a defect such as an incoherent twin boundary. The energy of the latter is typically of the order of several eV /atom.²² This is because there are serious disruptions of the tetrahedral bonding associated with such boundaries, including wrong bonds (C-C or Si-Si) and possibly dangling bonds. Thus atoms migrating on the surface will have a strong tendency to adjust their spin (i.e., stacking with respect to the underlying layers) to that of the growing island to which they are attaching. This explains why welldefined polytype structures can evolve even if the growth does not occur in a strict layer-by-layer fashion in spite of the energy differences for different stacking for each atom being much smaller than the growth temperature. Only for islands of the above defined size, which corresponds to \sim 10 nm in diameter, one expects that the interactions with underlying layers become relevant. A predominace of cubic stackings with respect to the underlying layers assumes that such 2D islands can still adjust their stacking position by moving as a whole. Although this might seem to require overcoming a significant energy barrier, motion of islands might occur by a 2D dislocation motion. In the above estimate, we used J_s $=2(J_1+J_2)$ neglecting J_3 and *K* interactions. We also assumed growth on a cubic substrate and renormalized to energies per SiC unit rather than per atom. For growth on other polytype substrates or when including J_3 and/or K , the interaction J_s becomes somewhat larger and hence the critical island size somewhat smaller, but the general argument does not change. Even though a preference for cubic stacking can thus be rationalized, a certain number of double positioning boundaries are expected because some islands of opposite spin may become trapped in an initially unfavorable stacking due to the randomness of the initial nucleation events. A step-flow growth mechanism seems to be the only plausible mechanism for stabilizing other polytypes during epitaxial growth and depends crucially on the sizes of the terraces and the surface diffusion (hence growth temperature). 23

IV. CONCLUSIONS

In conclusion, we have carefully re-evaluated the zerotemperature energy differences between polytypes of SiC using well-converged all-electron density-functional calculations. We find that the ANNNI model with up to secondnearest-neighbor-layer interactions already provides a good description of the polytype energy differences with slight improvements being obtained by including a third-layer interaction and a four-spin term. Even though the values for J_1 , J_2 do not correspond to the multiphase degeneracy point, the predominance of polytypes of narrow bands of cubic stacking (typically 2-3 banded) can readily be explained by the fact that J_1 .0 and the twin boundary energy cost is negative. Our results agree closer with the work of Heine *et al.*¹ than other recent calculations in the sense that we obtain $J_1 > |J_2|$, the 2-3 banded polytype energies closer to each other, and the 2*H* energy significantly higher than that of 3*C*. We stress that this is not due to our neglect of relaxations because the latter were shown to be at most 0.6 meV/ atom. We nevertheless find the energies of 4*H* and 6*H* to differ substantially enough to preclude a well-defined temperature stability region for each polytype when using literaure data for the vibrational free-energy contributions. This suggests that polytypes are kinetically determined metastable phases rather than true thermodynanic phases. Some consequences for epitaxial growth were discussed. In particular, we extended the arguments of Heine *et al.* concerning the tendency for 3*C* growth to occur if only equilibrium of the top-surface layer is required by considering the island size effects. We also showed that for $J_1 \leq |J_2|$, 4*H* would always be stabilized, which is inconsistent with experiment.

Note added. Our values for the 6*H*, 4*H*, and 2*H* energy differences from 3*C* agree extremely well with the values -1.1 , -1.2 , and 3 meV/atom, respectively, obtained by Rutter and Heine 24 using pseudopotential calculations.

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