Epitaxy of CoSi_x (1<x<2) silicides on Si(111) studied by photoemission and extended x-ray-absorption fine-structure techniques

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Electronic and structural properties of epitaxial CoSi_x layers have been investigated by means of core-level and valence-band photoemission, x-ray photoelectron diffraction, and extended x-ray-absorption fine-structure (EXAFS) experiments. CoSi_x layers of various x compositions have been grown on silicon by low rate Co and Si co-deposition onto room-temperature Si(111) substrates, with film thicknesses ranging from 30 to 100 Å. Photoemission shows substantial differences in valence and core-level spectra with respect to those of stable fluorite-type CoSi_2 and ε -CoSi and indicate that well-defined metastable phases are formed. In particular, core-level photoemission experiments performed with a monochromatized x-ray source show large Si 2pbinding-energy shifts (~0.4 eV) in the room-temperature deposited CoSi_x (1<x<2), with respect to stable ε -CoSi and CaF₂-type CoSi₂. X-ray photoelectron diffraction as well as extended x-ray-absorption finestructure measurements suggest that these pseudomorphic phases have a cubic structure, over a wide composition range. EXAFS measurements reveal that Co atoms are coordinated with eight Si atoms with a bond length of ~2.33 Å and with Co atoms with bond lengths in the 2.67–2.68 Å range. Such short Co-Co bond lengths show that the structure is definitively different from the stable CaF₂-type CoSi₂, even at the CoSi₂ composition. All experimental data indicate that CoSi_x silicides crystallize in a cubic lattice close to that of pseudomorphic FeSi_x silicides, namely, a CsCl-type derived structure.

I. INTRODUCTION

The growth of ordered metastable phases on metallic or semiconducting substrates is of great importance, since they could exhibit physical properties, not encountered in bulk materials. Recent accounts for the growth of metastable materials on Si(111) have been given for the Fe silicide-Si system. The Fe silicide-silicon system is a prototypical system with respect to these considerations, since several unusual Fe silicide phases, i.e., pseudomorphic phases, can be stabilized by epitaxy on Si(111). In particular, the formation of a metallic FeSi phase, with a CsCl-type structure using the molecular-beam technique, has been reported (the stable ε -FeSi phase crystallizes in a simple cubic B20-type lattice).¹ Thin FeSi₂ layers were found to crystallize in three metastable structures: a cubic CsCl-type FeSi derived structure,^{2–5} an α -FeSi₂ derived structure,^{6,7} and a fluorite (CaF₂-type) structure.^{8–10} Furthermore, it was shown that Fe silicides with a composition between FeSi and FeSi2 can also be grown on Si(111) with a CsCl-type structure.^{1-3,5} The stable ε -FeSi phase has a lattice misfit of -4.7%, with respect to Si. The FeSi₂ phase exists in two forms in the bulk phase diagram: the orthorhombic β -FeSi₂ phase, stable up to 920 °C and the quadratic α -FeSi₂, which exists only above 920 °C. β-FeSi₂ crystallizes in an orthorhombic structure with lattice parameters a=9.863 Å, b=7.791 Å, and c = 7.833 Å.¹¹ Epitaxy of β -FeSi₂ on Si(111) has been observed either with β -FeSi₂(100)||Si(111) or with β -FeSi₂(110)||Si(111). In both cases, the lattice misfit is of \sim 5–6%. It seems that the rather large mismatch between

bulk Fe silicides (ε -FeSi and β -FeSi₂) lattices as well as the very different atomic arrangement in the (111) planes of stable Fe silicide and Si make thin silicide layers adopt a crystal structure, which is better lattice matched to the substrate. The formation of a CsCl-type CoSi, epitaxially grown on Si(111) at room temperature has been reported more recently.¹² The stable CoSi phase, namely, ε -CoSi, also crystallizes in a simple cubic B20-type structure, with a lattice misfit of -5.6%, with respect to Si.¹¹ As opposed to bulk ε -CoSi, ε -FeSi, and β -FeSi₂ silicides, the mismatch between stable CaF_2 -type $CoSi_2$ and Si is rather small (-1.2%) and the atomic arrangement in the $CoSi_2(111)$ and Si(111) planes is very similar. Nevertheless, we show in this paper, by means of x-ray photoelectron diffraction (XPD), extended x-ray-absorption fine structure (EXAFS), and valence-band and core-level photoemission, that a metastable CoSi₂ phase forms indeed upon room-temperature co-deposition of Co and Si on Si(111). Epitaxy of CoSi₂ on Si(111) at room temperature has already been reported, but no investigation of its crystallographic structure was done at that time.^{13,14} This phase is stable up to $\sim 400-450$ °C. Furthermore, we have found that epitaxial $CoSi_x$ silicides, with $1 \le x \le 2$, can be also grown on Si(111). Experimental data show that they have a crystallographic, as well as electronic structure very similar to that of the CsCl-type FeSi derived FeSi_r silicides grown on Si(111).

II. EXPERIMENTAL PROCEDURE

Epitaxial CoSi_x silicides with composition $1 \le x \le 2$ were prepared on Si(111) in a UHV system, with a base pressure

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of 7×10^{-11} mbar. Co and Si were evaporated in the 1–2 Å/min rate range from boron nitride and carbon crucibles, respectively. The silicide composition was controlled by means of calibrated quartz microbalances, which measured the Co and Si fluxes independently, during the silicide growth. Prior to $CoSi_x$ deposition, the Si(111) substrates were cleaned by heating the sample up to $\sim 900-950$ °C. This procedure leads to a well-ordered Si(111) surface, as attested by a sharp 7×7 low-energy electron-diffraction (LEED) reconstruction. Absence of oxygen or carbon on the surface was checked by x-ray photoelectron spectroscopy. The $CoSi_x$ layers were grown in two steps. First, a 9-Å-thick CoSi₂ was co-deposited onto the clean Si(111) surface and subsequently annealed at 550 °C. A well-ordered CoSi₂ layer is achieved in this way, as attested by a well-defined 1×1 LEED pattern and sharp confinement states observed at ~ 0.5 eV on the ultraviolet photoemission spectroscopy spectrum.^{15,16} Second, 30-100-Å-thick CoSi_x layers were deposited onto such substrates maintained at a temperature lower than 100 °C. In this paper, such silicides films are called "room-temperature grown," as opposed to stable Cosilicides phases achieved by anneals at temperatures above 300 °C, typically.

Photoemission measurements were carried out using a Leybold Heraeus EA 200 spectrometer, equipped with a high power (1600 W) unmonochromatized x-ray source (photon energy $\hbar\omega$ =1486.6 eV) used for XPD measurements and a monochromatized x-ray source (photon energy $\hbar\omega$ =1486.6 eV) used for higher-resolution valence-band and core-level measurements. The photoelectrons were analyzed using a hemispherical energy analyzer (150 mm in radius), with a 50-meV energy resolution and an 18-channel multidetection assembly. The overall energetic resolution, using the monochromatized Al $K\alpha$ source, was about 0.5 eV. The two-stage electron lens at the entry of the electron analyzer allowed a tunable angular resolution in the $\pm 1^{\circ}$ and $\pm 8^{\circ}$ range. The acceptance angle of the analyzer was set to $\pm 8^{\circ}$ for highenergy-resolution core-level measurements and to $\pm 1^{\circ}$ for XPD measurements. Co $2p_{3/2}$ and Si 2p core lines were recorded at kinetic energy $E_c = 706$ and 1388 eV, respectively.

After photoemission measurements, a 20-Å-thick protective amorphous Si capping layer was deposited onto the silicide films. EXAFS measurements were performed at the Laboratoire pour l'Utilisation du Rayonnement Electromagnétique at Orsay on the XAS 2 beam line of the DCI storage ring. The incident radiation was monochromatized using a Si(111) plane two-crystal spectrometer. The beam line was equipped with a mirror at glancing incidence for the harmonic rejection. The variation of the x-ray-absorption coefficient was measured above the *K* edge of cobalt (7709 eV) at RT in the fluorescence mode. A detailed description of the data collection is given in Ref. 5.

III. RESULTS AND DISCUSSION

The CoSi_x silicides are epitaxial after deposition, as attested by a 1×1 LEED pattern. For compositions close to CoSi, the LEED pattern reveals additional spots arising from electron diffraction on [101] oriented facets. The presence of

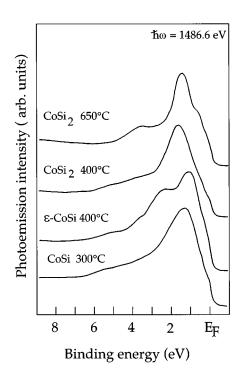


FIG. 1. Valence-band spectra collected at normal electron emergence for stable CaF_2 -type $CoSi_2$, ε -CoSi and pseudomorphic CoSi and $CoSi_2$ silicides. Spectra were recorded using a monochromatized Al $K\alpha$ source at a photon energy of 1486.6 eV.

these additional LEED spots depends critically on the film thickness. Their observation begins above 30 Å and they dominate the LEED pattern for thicknesses typically higher than 100 Å. In any case, the 1×1 LEED pattern is slightly improved by mild annealing up to temperatures ranging from 300 °C (Co-rich silicide) to \sim 400 °C (Si-rich silicides). Anneals at higher temperatures induce phase transitions towards stable phases. Pseudomorphic CoSi is completely transformed into the stable ε -CoSi phase upon annealing at 350– 400 °C. At this stage, the 1×1 LEED pattern disappears, leaving only a strong diffuse background. Pseudomorphic CoSi₂ is transformed into the stable epitaxial CaF₂-type $CoSi_2$ phase upon annealing at ~500 °C. This latter phase also exhibits a 1×1 LEED pattern. These phase transitions are clearly evidenced by valence-band spectra measurements. Figure 1 shows valence-band spectra taken with the monochromatized x-ray source on the Co silicides below and above the phase transition. Pseudomorphic CoSi and CoSi₂ valence bands are mainly composed of a broad peak centered at ~ 1.5 and 1.7 eV binding energy, respectively. Additional weak structures are observed at higher binding energies, which extend to about 5-eV binding energy. The CoSi spectrum is in good agreement with that published in Ref. 12. It evolves into a spectrum dominated by a double peak at 1.1and 2.2-eV binding energy, upon annealing at 400 °C. This double peaked shape is characteristic of monosilicides crystallizing in the cubic B20 structure, such as ɛ-FeSi and ε -CoSi.^{17–19} The CaF₂-type CoSi₂ valence-band spectrum of Fig. 1 is characterized by two main structures located at \sim 1.5- and 3.5-eV binding energy, arising from nonbonding and bonding states, respectively.^{20,21} For compositions in between, CoSi, evolves into a mixed phase composed of

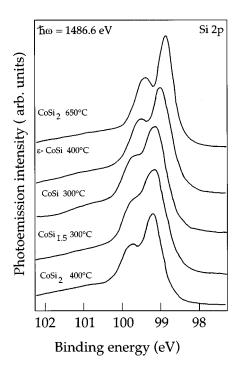


FIG. 2. Si 2*p* spectra collected at normal electron emergence for stable CaF₂-type CoSi₂, ε -CoSi and pseudomorphic CoSi, CoSi_{1.5}, and CoSi₂ silicides. Spectra were recorded using a monochromatized Al $K\alpha$ source at a photon energy of 1486.6 eV.

 ε -CoSi and fluorite-type CoSi₂, upon annealing above 500 °C.

At compositions close to CoSi, the distinction between pseudomorphic and stable phases is clear, on the basis of LEED observations. For compositions close to CoSi₂, LEED exhibits a 1×1 pattern for both pseudomorphic and stable phases. Hence, it must be shown that these samples are composed of a single phase and not a mixture of phases that include an appreciable amount of CaF₂-type CoSi₂. Information about this crucial point can be gained by core-level Si 2p and Co $2p_{3/2}$ measurements. Figure 2 shows the Si 2plines measured on the room-temperature grown $CoSi_x$, subsequently annealed in the 300-400 °C range. These spectra are presented along with that of stable CaF₂-type CoSi₂. Co $2p_{3/2}$ lines measured on CoSi_x and CaF₂-type CoSi₂ silicides are shown in Fig. 3. The photoemission spectra of Figs. 2 and 3 were obtained with monochromatized Al $K\alpha$ radiation $\hbar\omega$ =1486.6 eV and were collected at normal electron emergence. The silicide thickness is large enough (90 Å) to avoid substrate contribution to the signal and to minimize surface effects. Figure 2 shows that metastable silicide Si 2p lines are substantially shifted towards higher binding energies, with respect to clean Si(111), ε -CoSi or fluorite-type CoSi₂. Si 2p lines are shifted by about 0.35 eV, with respect to CaF₂-type CoSi₂ and by about 0.20 eV, with respect to ε -CoSi. At the CoSi₂ and CoSi_{1.5} compositions, the Co $2p_{3/2}$ lines lie at the same binding energy as the fluorite CoSi₂ or ε -CoSi ones. Improvement of the silicide crystallinity upon annealing is characterized by an appreciable decrease of their full width at half maximum. The stability of the as-deposited CoSi₂ phase versus annealing temperature has been examined. Figure 4 shows the Si 2p line evolution of a roomtemperature deposited CoSi₂ layer, as a function of annealing temperature (15-min anneals). After co-deposition the Si 2pline is broader and shifted by ~ 0.40 eV, with respect to that

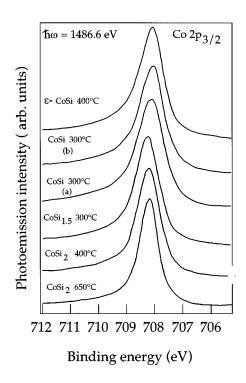


FIG. 3. Co $2p_{3/2}$ spectra collected at normal electron emergence for stable CaF₂-type CoSi₂, ε -CoSi and pseudomorphic CoSi, CoSi_{1.5}, and CoSi₂ silicides. Spectra were recorded using a monochromatized Al $K\alpha$ source at a photon energy of 1486.6 eV.

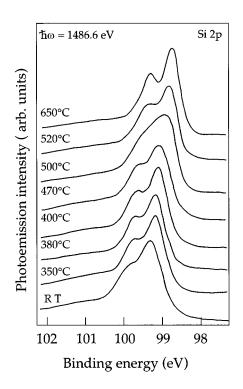


FIG. 4. Si 2*p* spectra collected at normal electron emergence for 90-Å-thick room-temperature co-deposited CoSi_2 silicide versus annealing temperature. Spectra were recorded using a monochromatized Al $K\alpha$ source at a photon energy of 1486.6 eV.

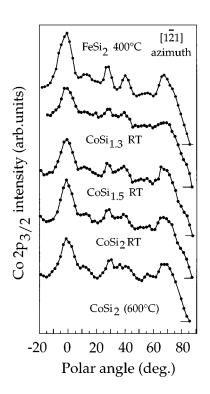


FIG. 5. Experimental polar-angle scans of Co $2p_{3/2}$ emission for 90-Å-thick pseudomorphic CoSi_{1.3}, CoSi_{1.5}, and CoSi₂, CaF₂-type CoSi₂ and pseudomorphic FeSi₂ silicides, epitaxially grown on Si(111), along the [121] azimuth of the Si(111) substrate. The polar angle $\theta=0^{\circ}$ corresponds to emission normal to the sample.

of the CaF₂-type CoSi₂ (topmost spectrum), where the Co occupies a single well-defined crystallographic site. Upon annealing at 350 °C, a slight binding energy decrease is observed along with better resolved Si $2p_{1/2}$ -Si $2p_{3/2}$ spin-orbit split components. The Si 2p line binding energy remains rather unchanged up to ~450 °C. At this stage, the Si 2p line has a full width at half maximum comparable to that of fluorite CoSi₂. The transition towards stable CaF₂-type CoSi₂ mainly starts at ~500 °C, as evidenced in Fig. 4 by a drastic broadening of the Si 2p line arising from the co-existence of at least two types of silicide phases. For annealing temperatures of 550 °C and above, one observes only the single Si 2p contribution relevant of the CaF₂-type CoSi₂ phase.

Co $2p_{3/2}$ core line intensity has been measured versus polar angle Θ for several 90-Å-thick CoSi_r silicide layers, in the $1 \le x \le 2$ composition range, and for a fluorite-type CoSi₂ layer. This latter 90-Å-thick layer was co-deposited onto a room-temperature substrate and annealed at 600 °C. The polar XPD scans have been measured in the (101) plane along the $[1\overline{2}1]_{Si}$ and $[\overline{1}2\overline{1}]_{Si}$ azimuthal directions of the Si(111) substrate and compared in Figs. 5 and 6. Also shown in Figs. 5 and 6 is the angular distribution of Fe $2p_{3/2}$ intensity measured on a pseudomorphic CsCl-type derived FeSi₂ layer epitaxially grown on Si(111), along the same azimuthal directions of the substrate. This 90-Å-thick FeSi₂ silicide layer was grown by Fe and Si co-deposition onto a roomtemperature Si(111) substrate.⁵ A Co $2p_{3/2}$ core line intensity variation measured on ε -CoSi versus polar angle (not shown) exhibits only a featureless profile, as expected for a polycrystalline layer. The most striking feature is that, along a given azimuthal direction, the main intensity maxima are observed

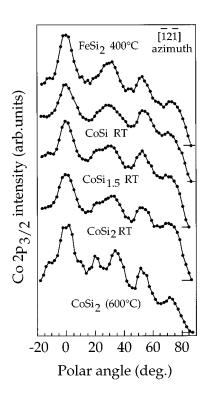


FIG. 6. Experimental polar-angle scans of Co $2p_{3/2}$ emission for 90-Å-thick pseudomorphic CoSi_{1.5} and CoSi₂, CaF₂-type CoSi₂ and pseudomorphic FeSi₂ silicides, epitaxially grown on Si(111), along the [121] azimuth of the Si(111) substrate. The polar angle θ =0° corresponds to emission normal to the sample.

at the same polar angles for all silicides presented in Figs. 5 and 6. It is well known that pseudomorphic CsCl-type derived FeSi_2 and fluorite-type CoSi_2 mainly grow on Si(111) with 180° (*B*-type) orientation.^{2,22–24} Thus, data of Figs. 5 and 6 indicate that CoSi, layers also grow with B-type orientation on Si(111). For all silicides, the $[121]_{Si}$ direction is parallel to the [121]_{Silicide} one and vice versa. For sake of clarity, the azimuthal directions specified in the following discussion are those of silicides. Co $2p_{3/2}$ profiles measured on the CaF2-type CoSi2 layer exhibit prominent structures centered at polar angles $\theta = 0^{\circ}$, $\theta \sim 30^{\circ}$, and $\theta \sim 55^{\circ}$, in the [121] direction and $\theta \sim 35^{\circ}$ and 70° in the opposite [121] azimuthal direction, i.e., at polar angles aligned with dense atomic rows in the $(10\overline{1})$ plane of the CaF₂ structure. CaF₂-type CoSi₂ and CsCl-type FeSi structures are shown in Fig. 7, along with their cut through the (101) plane. Peaks located at $\theta = 0^{\circ}$, $\theta \sim 55^{\circ}$, and $\theta \sim 70^{\circ}$ appear as single peaks, while more complex features are observed at $\theta \sim 30^\circ$ in the [121] direction and $\theta \sim 35^{\circ}$ in the $[1\overline{2}1]$ direction.²⁴ The structure centered at $\theta \sim 30^\circ$ is composed of two sharp peaks at $\theta \sim 20^{\circ}$ and 35° and a deep valley in between. In the opposite [121] direction, the structure centered at $\theta \sim 35^{\circ}$ exhibits three well-defined maxima. As noted above, along a given azimuthal direction, the Co $2p_{3/2}$ profile measured on the room-temperature grown CoSi, exhibits intensity maxima at essentially the same polar angles as for the CaF₂-type CoSi₂. This means that the room-temperature-grown CoSi_y silicides crystallize in a cubic structure similar to that of fluorite-type CoSi₂ one. Yet, the marked differences between these profiles reside in intensity changes in the broad structures centered at $\theta \sim 30^{\circ}$ and 35° . In the 121 direction, the deep

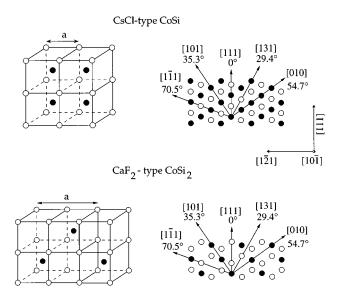


FIG. 7. CaF_2 -type $CoSi_2$ and CsCl-type CoSi structures along with their cut through the $(10\overline{1})$ plane.

valley between peaks at 20° and 35° is filled for roomtemperature co-deposited silicides. In the opposite [121] direction the central peak of the broad structure at $\theta \sim 35^\circ$, aligned with the [101] crystallographic direction, has disappeared when compared to CaF2-type CoSi2. In this respect, let us consider the Fe $2p_{3/2}$ profile versus the polar angle measured on a metastable FeSi2 layer, which also crystallizes in a cubic structure (upper curves of Figs. 5 and 6). This FeSi₂ structure is derived from the CsCl-type FeSi one shown in Fig. 7 by random removal of Fe atoms in order to reach the 1:2 composition. The CsCl-type FeSi structure is close to the CaF_2 one, with respect to XPD in the sense that the dense atomic rows appear at the same polar angle in both structures. The difference between profiles measured on the CsCl-type FeSi₂ and CaF₂-type CoSi₂ disilicides is also mainly reflected in intensity variations of the broad structures at $\theta \sim 30^\circ$ and 35°. These structures are very sensitive to small diffraction condition changes, such as changes of the wavelength or changes in the nature of the scatterers, as shown in recent works.²⁴ However, the difference between wavelengths associated with Co $2p_{3/2}$ (Ec=706 eV) and Fe $2p_{3/2}$ (Ec=779 eV) photoelectrons is rather small. The difference between the shape of the structures at $\sim 30^{\circ}$ ([121] azimuth) and $\theta \sim 35^{\circ}$ ([121] azimuth) in CaF₂-type CoSi₂ and $CoSi_x$ is, therefore, due to a change in crystallographic structure. At this stage, XPD data indicate that the roomtemperature-grown CoSi, layers have a cubic structure and that it closely resembles that of CsCl-type derived FeSi2. A clear-cut distinction between CaF2-type and CsCl-type structures has been made by using a local probe as EXAFS. For instance, in a CsCl-type structure, Co would be surrounded by Co scatterers at distances of $\sim 2.6-2.7$ Å, while the closest Co scatterers are located at ~ 3.8 Å from the emitter in the CaF₂-type structure. EXAFS measurements at the Co Kedge have been performed on $CoSi_x$ and fluorite-type $CoSi_2$ layers.

The EXAFS data have been analyzed using the software of Ref. 25. Figure 8 shows the experimental $k\chi(k)$ data recorded at the Co K edge on two epitaxial $CoSi_x$ and fluorite-

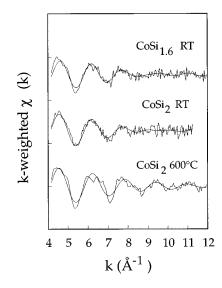


FIG. 8. *k*-weighted $\chi(k)$ EXAFS data recorded on 90-Å-thick fluorite-type CoSi₂ and room-temperature co-deposited CoSi_{1.6} and CoSi₂, at the Co *K* edge. Also shown are the Fourier-filtered contribution of Si NN and Co NNN.

type CoSi_2 layers. CoSi_x with compositions $\text{CoSi}_{1.6}$ and CoSi_2 have been analyzed. These data are presented after background subtraction and normalization to the edge jump. Figure 9 displays the Fourier transforms (FT) between 3 and 12 Å⁻¹ of the k^2 -weighted data. Bulk CoSi_2 crystallizes in the CaF₂ structure, with a unit-cell length of 5.368 Å.¹¹ Co atoms are surrounded by 8 first-shell Si neighbors at a distance of 2.324 Å and 12 second-shell Co neighbors at 3.794 Å. These two neighbor shells are clearly visible in the FT in Fig. 9 (bottom spectrum), which is relevant to high-temperature annealed CoSi_2 . In contrast with CaF_2 -type CoSi_2 , the first neighbor's contribution in CoSi_x silicide is mainly composed of two subshells, as shown in the Fourier-transform spectra of Fig. 9. These subshells are attributed to

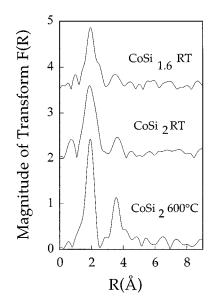


FIG. 9. Fourier-transform magnitude F(R) of Co EXAFS for a 90-Å-thick fluorite-type CoSi_2 and room-temperature co-deposited $\text{CoSi}_{1.6}$ and CoSi_2 .

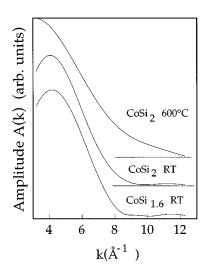


FIG. 10. Amplitude functions A(k) of the Fourier-filtered contribution of Si NN and Co NNN for the 90-Å-thick fluorite-type CoSi₂ and room-temperature co-deposited CoSi_{1.6} and CoSi₂.

Si nearest neighbors (NN) and Co next-nearest neighbors (NNN). The presence of two subshells can be clearly evidenced, using the beat node method.^{26,27} Figure 10 shows the Fourier filtered amplitude A(k) of the first main peak in the Fourier transform, as a function of the electron wave vector k. For CaF₂-type CoSi₂ (upper curve), A(k) decreases continuously, as expected for a single nearest-neighbor environment, while it exhibits a minimum at about 10 \AA^{-1} for RT grown CoSi_{1.6} and CoSi₂. This technique, described in Ref. 5, allows the determination of the difference between Co-Si and Co-Co bond lengths. This difference is found to be $\sim 0.33 \pm 0.04$ Å for CoSi_{1.6} and 0.34 ± 0.04 Å for CoSi₂. The most striking feature here is that these two subshells are observed in Fig. 9, even at the 1:2 composition. A further inspection of the Fourier-transform spectra reveals that the third-coordination shell (Co-Co-bonds) contribution is clearly visible at \sim 3.8 Å in CoSi_{1.6} and CoSi₂ spectra. This confirms the conclusions drawn from LEED and XPD measurements, i.e., that for these compositions, Co atoms reside in a well-defined cubic environment.

The Fourier-filtered contribution of NN and NNN peaks are superimposed on the experimental data in Fig. 8. They

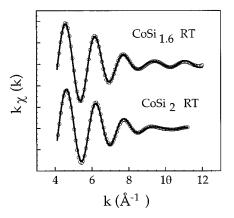


FIG. 11. Representative fits (full line) to Si NN and Co NNN Fourier-filtered signal (circles) for 90-Å-thick fluorite-type $CoSi_2$ and room-temperature co-deposited $CoSi_{1.6}$ and $CoSi_2$.

have been simulated in the single-scattering formalism, using experimental (CoSi₂ and Co metal) phase shifts and backscattering amplitudes. The phase shifts and backscattering amplitudes extracted from CoSi₂ have been used in the calculations for the Co-Si pair, while those extracted from Co metal data have been used for the Co-Co pair. The results of the simulations along with that of the beat node method are summarized in Table I. Figure 11 shows filtered EXAFS data (dots) and fits (full line) for room-temperature grown CoSi₂ and CoSi_{1.6}. CoSi_x Fourier-filtered spectra are well reproduced, assuming Si neighbors at $R_1 \sim 2.33 - 2.34$ Å and Co atoms at $R_2 \sim 2.67 - 2.68$ Å. $\Delta R = R_2 - R_1$ values of 0.33-0.34 Å given by the beat node method are in very good agreement with the value deduced from the simulations. $CoSi_2$ ($CoSi_{1.6}$) spectrum is well accounted for by 8 Si NN atoms and 3 (4) Co NNN atoms.

The NN Co-Si and NNN Co-Co bond lengths deduced from EXAFS measurements are consistent with a CoSi_x silicide structure close to that of pseudomorphic FeSi_x silicide, namely, a CsCl-derived structure. NNN Co-Co bond lengths of about 2.68 Å reveals that $\text{CoSi}_{1.6}$ and CoSi_2 lattice parameters ($a \sim 2.68$ Å) are smaller than that measured on CsCltype CoSi (a=2.74 Å).¹² Similar a lattice parameter and bond length increase versus a metal content increase was also found in FeSi_x silicides. Pseudomorphic FeSi and FeSi₂

TABLE I. Structural parameters deduced from the analysis of EXAFS spectra recorded at the Co K edge from co-deposited cobalt silicides epitaxially grown on Si(111). $\Delta\sigma$ is related to Co-Si and and Co-Co bonds in CoSi₂ and Co metal, respectively. In the last column, $\Delta R_{\rm BN}$ is the difference between the NNN Co-Co and NN Co-Si bond lengths, determined using the beat node method.

	Pair	<i>R</i> (Å)	Ν	$\Delta\sigma$ (Å)	$\Delta R_{\rm BN}$ (Å)
CoSi ₂					
(annealed at 600 °C)	Co-Si	2.33 ± 0.02	8 ±0.5	0	
90 Å					
CoSi ₂					
(RT deposited)	Co-Si	2.34 ± 0.02	8 ±0.5	0.04	0.34 ± 0.04
90 Å	Co-Co	2.68 ± 0.02	3.1 ± 1.0	0.05	
$\text{CoSi}_x (x \approx 1.6)$					
(RT deposited)	Co-Si	2.34 ± 0.02	8 ±0.5	0.06	0.33 ± 0.04
90 Å	Co-Co	2.67 ± 0.02	4.0 ± 1.0	0.05	

have lattice parameters of a = 2.77 and 2.70 Å, respectively.³ Fe-Fe bond lengths are significantly reduced in FeSi₂, with respect to FeSi. Such a decrease has been confirmed by EX-AFS measurements at the Fe K edge.⁵ NNN Fe-Fe bond lengths of 2.75 and 2.69 Å were measured in FeSi and FeSi₂, respectively. A first-neighbor Co-Co bond-length reduction of the same order of magnitude is observed in the present experiments. Furthermore, Si NN and Co NNN atom coordination numbers deduced from the simulations agree also very well with a structure derived from the CsCl-type CoSi one. CoSi_{1.6} and CoSi₂ structures are thus seen as a CoSi one in which 38% and 50% Co atoms are randomly removed, respectively. In a CsCl-type CoSi structure, Co atoms are bound to 8 Si NN and 6 Co NNN atoms. The number of Co NNN is reduced down to the mean value of 3 in a CoSi₂, as experimentally observed. Therefore, at a CoSi2 composition, the conservation of stoichiometry requires that the number of Co atoms at \sim 3.84 Å must be lower in this CsCl-derived structure than in the fluorite one.

Then, a further argument for the formation of CoSi_r with defected CsCl-type CoSi is given by the analysis of the thirdcoordination shell composed of Co atoms located at \sim 3.84 Å from the emitter. If, indeed, a $CoSi_x$ silicide structure is derived from a CsCl-type CoSi one, the number of Co atoms in the third-coordination shell would also decrease when the composition evolves from CoSi to CoSi₂. In this model, Co atoms are removed at random in all coordination shells. Fourier filtered spectra of Fig. 9 show a significant decrease of the peak at ~ 3.6 Å in CoSi_x, with respect to that in fluorite CoSi₂. This decrease could be attributed either to a loss of local order or to a lack of Co backscatterers. However, before any treatment of the Fourier-filtered contribution of the third-coordination shell, one must keep in mind that several multiple-scattering (MS) paths have a length close to that of the single-scattering (SS) Co-Co path and, thus, should be taken into account in the simulations. Thus, the ab initio multiple-scattering x-ray-absorption code FEFF (Refs. 28 and 29) was used to perform XAFS calculations including SS and MS paths around 3.8 Å. In a perfect CoSi₂ crystal, three double-scattering paths contribute to the peak located at \sim 3.6–3.8 Å in the FT: the first one called DS₁ (Co-Si-Si-Co) has a degeneracy of 24, the second one, DS_2 (Co-Si-Si-Co), has a degeneracy of 24 and the third one, DS_3 (Co-Co-Si-Co), has a degeneracy of 48. SS and DS paths are sketched in Fig. 12. These paths contribute as bond lengths of 3.66 Å (DS₁) and 4.22 Å (DS₂ and DS₃). The weighted magnitude of SS, DS₁, DS₂, and DS₃ are plotted versus wave vector k in Fig. 13. Similar calculations have been performed for a strained CoSi₂ layer. Indeed, when epitaxially grown on Si(111), CoSi₂ can be under lateral strain, due to the lattice mismatch of $\sim -1.2\%$. Only minor changes with respect to perfect cubic CoSi2 are detected, as far as scattering amplitudes are concerned. The MS process has generally a smaller contribution to the total signal than the SS one. Nevertheless, it strongly depends on k. The DS₁ and DS_2 paths have an amplitude close to zero for k values higher than 6 $Å^{-1}$ while DS₃ path still has a significant contribution at high k values. However, the damping due to the Debye-Waller factors is not taken into account in these curves. The Debye-Waller factor is significantly higher in the three-legs path making the DS₃ contribution also negligible

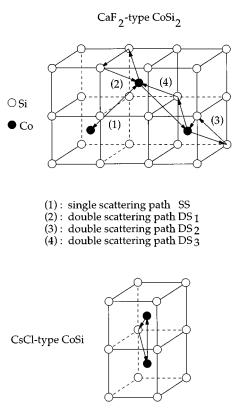


FIG. 12. A sketch of single-scattering and double-scattering paths in fluorite-type CoSi_2 and CsCl-type CoSi, which contribute to the peak at ~3.6–3.8 Å in the Fourier-transform spectra of Fig. 9.

above 6 Å⁻¹, as compared to the SS one. In the CsCl-type CoSi structure, an additional Co-Co-Si-Co (\sim 3.66 Å) twelvefold degenerated path must be considered, as shown in Fig. 12. Nevertheless, in CoSi_{1.6} and CoSi₂, this degeneracy decreases down to 8 or 6, making again these additional paths to contribute only weakly to the EXAFS signal. Therefore, the Fourier-filtered contribution of second- and third-coordination shells were simulated in the single-scattering formalism in the 6–13 Å⁻¹, for CaF₂-type CoSi₂ and room-

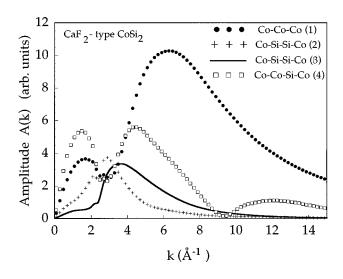


FIG. 13. Weighted magnitude of single-scattering and doublescattering paths shown in Fig. 12, as a function of wave vector k.

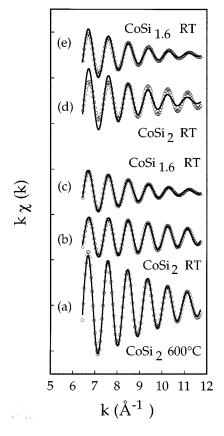


FIG. 14. Representative fits (full line) to the third-coordination shell Fourier-filtered signal (circles) for 90-Å-thick fluorite-type $CoSi_2$ (a) and room-temperature co-deposited $CoSi_2$ (b) and $CoSi_{1.6}$ (c). Also shown are the fits for room temperature co-deposited $CoSi_2$ (d) and $CoSi_{1.6}$ (e) performed assuming 12 Co backscatterers.

temperature grown CoSi1.6 and CoSi2, respectively. Experimental spectra were simulated using theoretical phase shifts and backscattering amplitudes.^{28,29} Figure 14 shows filtered EXAFS data (dots) and fits (full line) for CaF₂-type CoSi₂ and room-temperature grown CoSi2 and CoSi1.6. The results of the simulations are summarized in Table II. Simulations reveal that Co atoms are bound to 12 Co at $R_3 \sim 3.84 \pm 0.04$ Å in the fluorite-type CoSi2 annealed at 600 °C. Such a bond length, larger than that of a bulk CaF₂-type CoSi₂ structure, is very close to the translation periodicity (~ 3.84 Å) of the Si(111) surface. This would suggest that the hightemperature annealed CoSi₂ layer is rather well matched to the Si(111) substrate, i.e., it is laterally strained. This is surprising since it has been recently demonstrated that CaF₂-type CoSi₂ layer with comparable thickness annealed at 600-650 °C are fully relaxed instead.30 Thus, thirdcoordination shell bond length R_3 deduced from EXAFS seems overestimated by about 1%. The R_3 bond lengths, shown in Table II for RT CoSi₂ and CoSi_{1.6}, would have to

TABLE II. Structural parameters deduced from the analysis of the third-coordination shell Fourier-filtered EXAFS spectra recorded at the Co K edge from co-deposited cobalt silicides and CaF₂-type CoSi₂ layers epitaxially grown on Si(111).

	Pair	<i>R</i> (Å)	Ν	σ (Å)
90-Å CoSi ₂ (annealed at 600 °C)	Co-Co	3.84±0.04	12±3	0.08
90-Å CoSi ₂ (RT deposited)	Co-Co	3.84 ± 0.04	5±3	0.06
90-Å CoSi _{1.6} (RT deposited)	Co-Co	3.82±0.04	7±3	0.11

be reduced, accordingly. In doing so, these bond lengths would be of 3.80 Å (RT $CoSi_2$) and 3.78 Å (RT $CoSi_{1.6}$), as expected for cubic CsCl-type structure with secondcoordination shell bond length of 2.68 Å (RT CoSi₂) and 2.67 Å (RT CoSi_{1.6}). Furthermore, simulations of Fig. 14 clearly demonstrate that the important decrease of the FT peak at 3.6–3.8 Å in co-deposited silicides, with respect to that of fluorite-type CoSi₂, cannot be assigned to a loss of local order, only. Indeed, any attempt to simulate these spectra with 12 Co has failed. Thus, the damping of the FT peak is mainly due to a true change in the Co number of thirdshell neighbors, associated with a change in crystal structure. The coordination number deduced from the simulation is of 7 in CoSi_{1.6} and of 5 in CoSi₂. At CoSi_{1.6} and CoSi₂ compositions, third-shell mean coordination numbers of 8 and 6 are expected in a defected CsCl-type CoSi, in rather good agreement with experimental results summarized in Table II. This lends strong support to random (defected CsCl-type), as opposed to ordered (CaF_2 -type) occupation of the Co sites. However, because of the limited accuracy of third-shell coordination-number determinations in EXAFS, some deviation from perfect random distribution, with a preferential occupation of third-shell Co sites, cannot be ruled out and might be evidenced by x-ray-diffraction experiments.

IV. CONCLUSIONS

In summary, XPD profiles, Co-Si and Co-Co bond lengths deduced from EXAFS data as well as Si 2p core level, and valence-band photoemission demonstrate that pseudomorphic CoSi_x ($1 \le x \le 2$) silicides grown by room-temperature co-deposition crystallize basically in a CsCl-type structure with randomly distributed Co vacancies very similar to that of pseudomorphic FeSi_x silicides. In particular, it is shown that CoSi₂ can adopt a crystal structure different from that of the stable CaF₂-type CoSi₂, despite the small lattice mismatch (-1.2%) of this latter structure, with respect to Si. The growth of this CoSi₂ phase cannot be explained by its greater stability in the form of the films, as found for FeSi₂, but most likely results from kinetic factors.

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