PHYSICAL REVIEW B

CONDENSED MATTER

THIRD SERIES, VOLUME 51, NUMBER 8

15 FEBRUARY 1995-II

Effects of Ce-Lu substitution on the d valence states of Laves phases

L. Duò, P. Vavassori, and L. Braicovich

Dipartimento di Fisica, Politecnico di Milano, piazza Leonardo da Vinci 32, 20133 Milano, Italy

G. L. Olcese

Istituto di Chimica Fisica, Università di Genova, Corso Europa 26, 16132 Genova, Italy (Received 19 April 1994; revised manuscript received 17 October 1994)

The legitimacy of the extraction procedure of the 4f spectral function for Ce intermetallic compounds by modeling the non-f states in terms of isostructural substitutions of the rare-earth species is analyzed by combined ultraviolet direct and inverse photoemission spectroscopy on CeCo₂, LuCo₂, CeRh₂, and LuRh₂. We find that the occupied states show a strong sensitivity to the R substitution when the Ratoms interact strongly with the metal as in the extended 4d M wave functions (Rh) while the variations are negligible in the case of 3d M partner (Co). The more delocalized empty states are sensitive to the Ce-Lu substitution both in RCo_2 and RRh_2 systems. The results are discussed in terms of the different spatial extent of Ce and Lu 5d wave functions and of the degree of intermixing with the M partner dvalence states.

I. INTRODUCTION

In recent years the spectroscopic study of Ce and Cetransition metal (M) compounds has attracted considerable attention, with particular emphasis on their electronic and magnetic properties.¹ In these compounds the energy overlap between the f states and the more delocalized d states, which are very important in the Ce-Mchemical bond, makes it difficult to disentangle the two contributions both in photoemission spectroscopy (PES) and inverse photoemission (IPES). In many cases² by implicitly assuming that the rare-earth (R) substitution does not alter the d-partial density of states (d-PDOS), the Ce f-related spectral functions have been obtained by difference spectra with similar compounds containing a completely empty (e.g., Y and La) or completely filled (e.g., Lu) f shell.

The aim of this work is to assess the limits of applicability of this approach through a quantitative analysis of the substitution effects in R compounds. This is achieved by a systematic study of PES and IPES (in the ultraviolet) on the Ce-Lu substitution in a number of Laves phase intermetallic compounds³ with the same crystal structure and similar lattice parameters (upon R substitution), namely, CeCo₂, LuCo₂, CeRh₂, and LuRh₂. These cubic Laves phases are interesting because, besides allowing isostructural R substitutions, they are characterized by a large R-M coordination number and represent, therefore, a good test to evaluate the sensitivity of the electron states to the variation of R partner. The reason of the choice of Lu as a substituting element of Ce is twofold: on one hand, in order to be able to study the whole dPDOS, it is important to avoid the presence of the fstates in a wide energy range spanning both filled and empty states; this rules out the possibility of using La,⁴

which has a strong 4f emission in the region between ~ 4 and $\sim 7 \text{ eV}$ above E_F ;^{5,6} on the other and, the differences in the shape of the d wave functions should be minimized, in order to avoid a severe distortion of the charge distribution around the R sites; from this point of view Lu appears more appropriate than Y.⁷

We find that in the occupied states of $CeCo_2$, the Ce-Lu substitution has a negligible effect, while a sizable modification is present in empty states; on the other hand, $CeRh_2$ is more sensitive to substitution, both in occupied and in empty states, i.e., the substitution effects are larger in the compounds having the greater R-M hybridization. The experimental trends are rationalized in terms of the interplay between (i) the different spatial extent of Ce vs Lu 5d wave functions, (ii) the different spatial extent of Co 3d vs Rh 4d, and (iii) the lattice parameters variation upon substitution. This explains also the greater sensitivity of empty states to substitution due to the greater wave functions extent.

II. EXPERIMENT

The IPES spectrograph⁸ is based on an ultraviolet grating dispersing photons on a flat field. Spectra are collected by scanning the electron energy (isochromat mode); at various photon energies hv, the spectra are acquired simultaneously so that they are strictly comparable. The full width at half maximum (FWHM) is ~0.4 eV for hv=12 eV (Ref. 8). Pt spectra have been taken for the energy calibration⁹ during the same experiment. The spectra are normalized on line, to the current impinging onto the sample.

Photoemission spectroscopy data has been measured with a 150-mm hemispherical analyzer using HeI radiation ($h\nu$ =21.2 eV) with FWHM=0.15 eV. The rather broad angular definitions both in IPES and in PES, together with the polycrystalline nature of the samples, ensure that all measurements are \mathbf{k} integrated.

Polycrystalline samples have been prepared by induction melting from stoichiometric amounts of the components in Ta crucibles after Ar purging. They were annealed at 800-900 °C for several days to obtain homogeneous polycrystals; the quality has been checked by xray diffraction and microprobe analysis. All crystals are cubic Laves phases with the MgCu₂ structure type (Ref. 4). The number of nearest neighbors is, for the R (M) site, 12 (6) M atoms and 4 (6) R atoms. In Table I the atomic distances, d[R-R], d[R-M], and d[M-M], are given.

All the samples have been mechanically scraped in situ with a diamond file at a pressure lower than 5×10^{-10} mbar. The base pressure during the data collection was lower than 2×10^{-10} mbar. The main contribution to surface contamination at room temperature comes from surface segregation of impurities from the interior of the sample; this has been avoided by keeping the samples at ~ 100 K during the scraping and the measurements, without any temperature increase. Sample cleanliness and surface stoichiometry¹⁰ have been accurately checked by x-ray core-level spectroscopy.

III. RESULTS

Figure 1 presents the PES results on $CeCo_2$ (dots) and $LuCo_2$ (line) [Fig. 1(a)] and on $CeRh_2$ (dots) and $LuRh_2$ (line) [Fig. 1(b)] at $h\nu = 21.2$ eV, normalized to their maximum intensity. While there are strong similarities between the $CeCo_2$ and $LuCo_2$ profiles with the maximum intensity (A) reached at 0.35 eV below E_F , substantial line-shape variations upon R substitution are present in RRh₂: the dominant peak (B), at ~1.15 eV below E_F in LuRh₂, moves to a larger binding energy (BE~1.65 eV), resulting in a centroid shift for CeRh₂; furthermore, the weight of the shoulder A close to E_F becomes lower compared to the maximum intensity.

Figure 2 gives the IPES results at $h\nu = 10.4$ and 13.4 eV for CeCo₂ (dots) compared with the spectrum of LuCo₂ (line, $h\nu = 13.4$ eV). In the spectra four features can be identified in CeCo₂ (LuCo₂) at about 0.8 eV (0.8 eV, A), 2.2 eV (1.8 eV, B), 3.8 eV (3.2 eV, C), and 5.4 eV (4.6 eV, D) eV above E_F . Finally, Fig. 3 displays IPES results at $h\nu = 11.7$ and 13.4 eV in CeRh₂ and LuRh₂ (line, $h\nu = 13.4$ eV). Two structures are visible in the spectra located in CeRh₂ (LuRh₂) at ~0.8 eV (0.8 eV, A) and 4.8 eV (3.2 eV, B) eV above E_F . Despite some intensity variations, the energy, of the IPES features shown in Figs. 2

TABLE I. Summary of the *R-R*, *R-M*, and *M-M* nearestneighbors distances d of CeCo₂, LuCo₂, CeRh₂, and LuRh₂, respectively. All the compounds have the MgCu₂ cubic Laves phases structure type.

Compound	d[R-R] (Å)	<i>d</i> [<i>R-M</i>] (Å)	<i>d</i> [<i>M</i> - <i>M</i>] (Å)
CeCo ₂	3.10	2.97	2.53
LuCo ₂	3.08	2.95	2.51
CeRh ₂	3.27	3.13	2.67
LuRh ₂	3.21	3.08	2.62

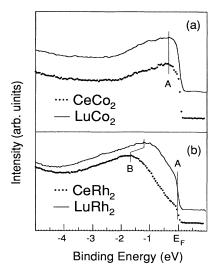


FIG. 1. (a) Intensity normalized PES results for $CeCo_2$ (dots) and $LuCo_2$ (line), with $h\nu = 21.2 \text{ eV}$. The maximum intensity A lies at 0.35 eV below E_F . (b) Same as (a) but for $CeRh_2$ (dots) and $LuRh_2$ (line). The peak intensity B lies at 1.15 (1.65) eV for $LuRh_2$ (CeRh₂).

and 3, is constant within a large hv range (Refs. 3 and 11).

IV. DISCUSSION

The chemical bonds in the compounds under study involve a low d occupancy atom (R) and a much higher doccupancy atom as Co and Rh. On the basis of the present knowledge on the electron states of these compounds (Refs. 3, 11, and 12), we can qualitatively state that the occupied d DOS's have a dominant Co (Rh) character with a redistribution of the small Ce (or Lu) dcontribution to higher BE's than in pure R. On the other hand, the empty states region well above E_F is dominated by the R d-derived states and in the intermediate region,

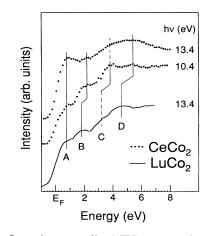


FIG. 2. Intensity normalized IPES results for $CeCo_2$ (dots) and $LuCo_2$ (line), measured in the isochromat mode at $h\nu = 10.4$ and 13.4 eV. Features A, B, C, and D on $CeCo_2$ ($LuCo_2$) lie at 0.8 (0.8), 2.2 (1.8), 3.8 (3.2), and 5.4 (4.6) eV above E_F , respectively. The weaker features are represented as dashed lines.

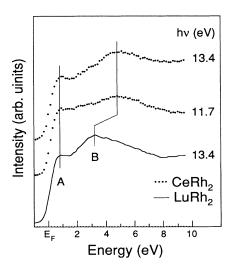


FIG. 3. Same as Fig. 2 but for CeRh₂ (dots) and LuRh₂ (line) at hv = 11.7 and 13.4 eV. Features A and B on CeRh₂ (LuRh₂) lie at 0.8 (0.8) and 4.8 (3.2) eV above E_F , respectively.

i.e., just above E_F , a contribution from empty Co (Rh) d states is still present.¹¹

As far as the relative sensitivity of the measurements to d vs. f states is concerned, we have shown in a previous work⁵ that in the $h\nu = 10-30$ eV range, IPES changes from a mostly d to a f sensitive regime due to the $h\nu$ dependence of cross sections. This happens also in R-M compounds.¹¹ Thus, at low $h\nu$ ($< \sim 15$ eV), a direct comparison between d-like states in Ce- and Lu-based compounds can be performed. In direct photoemission, due to the low f occupation the He I spectrum is d sensitive. Within this framework, one can, therefore, address the sensitivity of the d PDOS with respect to the variation of R partner in different chemical environments.

A. Occupied d states

Since the PES profiles are mainly related to the M dPDOS, their possible variations, upon Ce-Lu substitution, must be understood in terms of M-M and M-R chemical interaction. The former depends only on the M-Mnearest-neighbors distance, which decreases by only 0.02 (0.05) Å for M=Co (Rh) upon Ce-Lu substitution, because of the smaller Lu atomic size. The R-M overlap is influenced by the different Ce vs Lu 5d wave-function width as well: actually, this effect is definitely larger than the tiny lattice contraction and results in a lower Lu 5d-M d than Ce 5d-M d hybridization. Therefore, in the case of CeM a shift to larger BE may occur, due to the stronger R-M hybridization, as compared to LuM. Indeed this happens in R-Rh₂ and not in RCo₂.

The marked similarity between the PES results of $CeCo_2$ and $LuCo_2$ [Fig. 1(a)] can be understood in terms of the pronounced localization of the Co 3d wave function on the atomic site, so that both Co 3d-Co 3d and R 5d-Co 3d overlap integrals do not undergo remarkable variations upon substitution. This is further supported by a recent work¹³ on YCo₂ and a number of RCo_2 compounds (R = Ce, Pr, Nd, and Sm) showing a strong

overall line-shape similarity among the d-sensitive PES results despite different combinations of wave-function widths and lattice parameters.

The strong difference between the PES results of $CeRh_2$ and $LuRh_2$ [Fig. 1(b)] has, thus, to be attributed to the much wider radial distribution of the Rh 4d orbitals, generating a stronger sensitivity to the Ce-Lu substitution. The shift of the valence-band peak [feature B in Fig. 1(b)] to higher BE and the consequent reduction of the DOS close to E_F (feature A) as compared to LuRh₂, thus, represent the experimental evidence that the increased R-Rh hybridization pulls down the bonding states depleting the region close to E_F .

We add that strong modifications of the *d*-sensitive UV PES profiles has been previously observed¹⁴ in a compound similar to CeRh₂, namely, CeRu₂, upon substitution of Ce with La. All these findings seem to suggest that, at variance with the case of 3*d M*'s, when dealing with extended *d* valence states like those of 4*d M*'s (and *a fortiori* 5*d M*'s) *R* substitutions do perturb the non-f valence states below E_F , unless a proper matching between the effects of the lattice and the wave-function variation takes place as in the case of Ce₇Rh₃ vs Lu₇Rh₃.¹⁵

B. Unoccupied d states

Figure 2 shows some differences between the dsensitive IPES spectra for CeCo₂ and LuCo₂ at variance with the PES results. In both compounds there are four structures (A - D), in agreement with theoretical siteselective PDOS calculations (Refs. 3, 11). Apart from A, the other features are systematically found at larger energies in CeCo₂ with shifts of about 0.4 eV (B), 0.6 eV (C)and 0.8 eV (D). While feature A still contains a sizable contribution of Co 3d holes, and, similarly, to the occupied electronic states is unchanged upon R substitution, the structures **B** to **D** are mainly due to **R** 5d-like holes.¹¹ This region of the spectrum is, therefore, driven by the *R-R* 5*d* interaction, which is larger in the case of $CeCo_2$. The effect is a shift of the Ce 5d antibonding states, far from E_F in CeCo₂ compared to LuCo₂. It is worth noting that this shift seems to increase on going from B to D, i.e., increasing the antibonding character of the involved states.

A similar situation occurs in the low hv IPES comparison between CeRh₂ and LuRh₂. In fact, the results of Fig. 3 clearly show doubly peaked line shapes for both compounds with feature A (Rh-derived) at a fixed position and the broad structure B (R-derived) shifted to a higher energy in CeRh₂. Again, we interpret this experimental evidence as the effect of a stronger R-R 5d interaction in the case of CeRh₂. Furthermore, the intensity ratio between the features A and B of Fig. 3 is lower for CeRh₂ than LuRh₂. This is the empty states counterpart on CeRh₂ of the "hybridization gap"¹⁶ effect across E_F , shown by feature A in Fig. 1(b).

In the RCo_2 case, the more pronounced sensitivity of the empty *d* electron states to the Ce-Lu substitution, as compared to the occupied *d* states, should be attributed to the larger spatial extent of the R 5*d* wave functions with respect to the Co 3d, consistently with the PES results on RCo_2 vs RRh_2 . This seems to indicate that even in cases like RCo_2 , where R substitution effects are negligible in the occupied valence states, the states lying above E_F show a strong sensitivity to small variations of the radial functions.

V. CONCLUSIONS

We have presented a combined UV PES and IPES investigation in the *d* sensitive regime on CeCo₂, CeRh₂ and on the isostructural Lu substituted compounds (LuCo₂ and LuRh₂) in order to discuss the sensitivity of the partial *d*-DOS to the substitution effects. In the occupied electron states, where the *M d*-derived signal is predominant, the substitution effects are significant when the hybridization between the *R* and the *M* orbitals is large (M=Rh), and becomes negligible in the case of the more contracted Co 3*d* shell, whereas for the empty

states, whose spatial extent is larger, the effect is clearly present even in RCo_2 , although less pronounced than in RRh_2 . Moreover, we have presented a rationale on this effect based on the spatial extent of the involved wave functions.

These results may, therefore, represent a useful guideline to indicate in which cases a quantitative evaluation of the d PDOS in CeM compounds can be accurately performed, in terms of the d PDOS of an isostructural compound with a substituting R element, and in which cases, instead, these kinds of comparisons are not fully legitimate.

ACKNOWLEDGMENT

Financial support by the Ministero dell'Università e della Ricerca Scientifica e Tecnologica through the Istituto Nazionale di Fisica della Materia is acknowledged.

- ¹For extensive reviews, see, e.g., J. W. Allen, S. J. Oh, O. Gunnarsson, K. Schönhammer, M. B. Maple, M. S. Torikachvili, and I. Lindau, Adv. Phys. **35**, 275 (1986); Y. Baer and W.-D. Schneider, in *Handbook on the Physics and Chemistry of Rare Earths*, edited by K. A. Gschneidner, Jr., L. Eyring, and S. Hüfner (Elsevier Science, Amsterdam, 1987), Vol. 10, p. 1; D. W. Lynch and J. H. Weaver *ibid.*, p. 231; F. U. Hillebrecht, and M. Campagna, *ibid.*, p. 425.
- ²See, e.g., M. Croft, J. H. Weaver, D. J. Peterman, and A. Franciosi, Phys. Rev. Lett. 46, 1104 (1981); F. U. Hillebrecht, J. C. Fuggle, G. A. Sawatzky, and R. Zeller, *ibid.* 51, 1187 (1983); D. Malterre, M. Grioni, P. Wiebel, B. Dardel, and Y. Baer, *ibid.* 68, 2656 (1992); P. Weibel, D. Malterre, M. Grioni, B. Dardel, and Y. Baer, Z. Phys. B 87, 165 (1992); E. Beaurepaire, J. P. Kappler, S. Lewonczuk, J. Ringeissen, M. A. Khan, J. C. Parlebas, Y. Iwamoto, and A. Kotani, J. Phys. Condens. Matter 5, 5841 (1993).
- ³A detailed spectroscopic analysis of Lu and Lu-*M* compounds can be found in L. Duò, P. Vavassori, M. Finazzi, L. Braicovich, and G. L. Olcese, Phys. Rev. B 49, 10159 (1994).
- ⁴Furthermore, in the La-Co phase diagram, the formation of LaCo₂ does not occur [see, A. Iandelli and A. Palenzona, in Handbook of the Physics and Chemistry of Rare Earths, edited by K. A. Gschneidner, Jr. and L. Eyring (Elsevier, Amsterdam, 1979), Vol. II, p. 1].
- ⁵L. Duò, M. Finazzi, and L. Braicovich, Phys. Rev. B **48**, 10728 (1993).
- ⁶J. K. Lang, Y. Baer, and P. A. Cox, J. Phys. F 11, 121 (1981);
 A. V. Fedorov, C. Laubschat, K. Starke, E. Weschke, K.-U. Barholz, and G. Kaindl, Phys. Rev. Lett. 70, 1719 (1993).
- ⁷An analysis on the atomic radial wave functions (Ψ) shows that the difference [Ψ (Ce 5d)]²-[Ψ (Lu 5d)]² integrated between 0.5 and 3 Å, i.e., in the bonding charge region, is about 50%

lower than that $[\Psi(\text{Ce } 5d)]^2 - [\Psi(Y \ 4d)]^2$ calculated in the same radial range.

- ⁸M. Sancrotti, L. Braicovich, C. Chemelli, F. Ciccacci, E. Puppin, G. Trezzi, and E. Vescovo, Rev. Sci. Instrum. 62, 639 (1991).
- ⁹Pt has been used as a reference because of its stronger cross section compared to noble metals; the energy calibration on Pt has been previously performed by a comparison with the Fermi edge of pure Au. [See, e.g., E. Puppin and P. Vavassori, J. Phys. Condens. Matter 4, 5551 (1992).]
- ¹⁰The accuracy of our x-ray analysis was greatly improved by the possibility of making a comparison among samples having the same elements but different stoichiometries (to this purpose, we have included Ce_7Rh_3 and Lu_7Rh_3 as well). This avoids any assumption on the variation of the instrumental efficiency vs photoelectron kinetic energy, on the photoionization cross sections, and on the electron escape depths.
- ¹¹L. Braicovich, E. Puppin, P. Vavassori, G. L. Olcese, L. Nordström, and B. Johansson, Solid State Commun. 89, 651 (1994).
- ¹²John C. Fuggle, F. U. Hillebrecht, R. Zeller, Zygmunt Zolnierek, Peter A. Bennet, and Ch. Freiburg, Phys. Rev. B 27, 2145 (1983).
- ¹³J.-S. Kang, J. H. Hong, J. I. Jeong, S. D. Choi, C. J. Yang, Y. P. Lee, C. G. Olson, B. I. Min, and J. W. Allen, Phys. Rev. B 46, 15689 (1992).
- ¹⁴D. J. Peterman, J. H. Weaver, M. Croft, and D. T. Peterson, Phys. Rev. B 27, 808 (1993).
- ¹⁵P. Vavassori, L. Duò, L. Braicovich, and G. L. Olcese, Phys. Rev. B **50**, 9561 (1994).
- ¹⁶J. F. van Acker, E. W. Lindeyer, and J. C. Fuggle, J. Phys. Condens. Matter 3, 9579 (1991).