- 29, 948 (1963). ¹¹G. A. Candela and R. E. Mundy, J. Chem. Phys. <u>46</u>, 47 (1967).
- ¹²R. Orbach, Proc. Roy. Soc. (London) <u>A264</u>, 485 (1961).
- ¹³D. L. Huber, Phys. Rev. <u>131</u>, 190 (1963).
- ¹⁴M. B. Walker and F. I. B. Williams, Can. J. Phys. <u>46</u>, 1347 (1968); <u>48</u>, 355 (1970).
- ¹⁵J. M. Baker and N. C. Ford, Jr., Phys. Rev. <u>136</u>, A1692 (1964).
- ¹⁶W. J. Brya and P. E. Wagner, Phys. Rev. <u>147</u>, 239 (1966).
- 17 C. B. P. Finn, thesis (Oxford University, 1961) (un-published), quoted in Ref. 12.
- ¹⁸R. C. Sapp, B. G. Aldridge, and A. Singh, Bull. Am. Phys. Soc. 14, 195 (1969).
- ¹⁹J. T. Hoffman and R. C. Sapp, J. Appl. Phys. <u>39</u>, 837 (1968).
- ²⁰J. B. Gruber and R. A. Satten, J. Chem. Phys. <u>39</u>, 1455 (1963).
- ²¹J. H. van Vleck, Phys. Rev. <u>59</u>, 724 (1941).
- $^{22}\mathrm{R.}$ H. Ruby, H. Benoit, and C. D. Jeffries, Phys.
- Rev. <u>127</u>, 51 (1962).
 ²³B. W. Faughnan and M. W. P. Strandberg, J. Phys.
 Chem. Solids <u>19</u>, 155 (1961).
- ²⁴A. M. Stoneham, Proc. Phys. Soc. (London) <u>86</u>, 1163 (1965).
- ²⁵A. Kiel and W. B. Mims, Phys. Rev. <u>161</u>, 386 (1967).
 ²⁶D. W. Preston, Ph.D. thesis (University of Kansas,
- 1970) (unpublished).
- ²⁷J. M. Daniels, Proc. Phys. Soc. (London) <u>A66</u>, 673 (1953).
- ²⁸J. T. Hoffman, Ph.D. thesis (University of Kansas, 1970) (unpublished).
- ²⁹Prepared by A. D. Mackay Co., 198 Broadway, New York, N. Y. 10038.
- ³⁰F. G. Brickwedde, H. van Dijk, M. Durieux, J. R. Clement, and J. K. Logan, J. Res. Natl. Bur. Std. <u>64</u>,

- 1 (1960).
- $^{31}\mathrm{H.}$ B. G. Casimir and F. K. du Pré, Physica 5, 507 (1938).
 - ³²H. Meyer, Phil. Mag. <u>2</u>, 673 (1953).
- ³³L. D. Roberts, C. C. Sartain, and B. Borie, Rev. Mod. Phys. 25, 170 (1950).
- ³⁴I. Svare and G. Seidel, Phys. Rev. <u>134</u>, A172 (1964).
 ³⁵J. A. A. Ketelaar, Physica <u>4</u>, 619 (1937).
- ³⁶D. R. Fitzwater and R. E. Rundle, Z. Krist. <u>112</u>,
- 362 (1959).
- ³⁷M. T. Hutchings, in *Solid State Physics*, edited by F. Seitz and D. Turnbull (Academic, New York, 1964),
- Vol. 16, p. 227.
- ³⁸K. W. H. Stevens, Proc. Phys. Soc. (London) <u>A65</u>, 209 (1952).
- ³⁹R. J. Elliott and K. W. H. Stevens, Proc. Roy. Soc. (London) <u>A215</u>, 437 (1952); <u>A219</u>, 387 (1953).
 - ⁴⁰S. Hüfner, Z. Physik <u>169</u>, 417 (1962).
 - ⁴¹G. Burns, Phys. Rev. <u>128</u>, 2121 (1962).
 - ⁴²G. Burns, J. Chem. Phys. 42, 377 (1965).
- ⁴³A. J. Freeman and R. E. Watson, Phys. Rev. <u>127</u>,
- 2058 (1962).
- 44 C. Y. Huang, Phys. Rev. <u>139</u>, A241 (1965). 45 A. Singh, Ph.D. thesis (University of Kansas, 1971) (unpublished).
 - ⁴⁶R. J. Birgeneau, Can. J. Phys. <u>45</u>, 3761 (1967).
 - ⁴⁷H. A. Buckmaster, Can. J. Phys. <u>40</u>, 1670 (1962).
- ⁴⁸R. J. Birgeneau, J. Phys. Chem. Solids <u>28</u>, 2429
- (1967). 49 We are much indebted to Professor H. J. Stapleton for

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 $^{50}\mathrm{E.}$ Borchi, S. de Gennaro, and M. Mancini, Nuovo Cimento <u>61B</u>, 241 (1969).

⁵¹M. M. Curtis, D. J. Newman, and G. E. Stedman, J. Chem. Phys. 50, 1077 (1969).

⁵²J. Levy, Phys. Rev. B <u>1</u>, 4261 (1970).

PHYSICAL REVIEW B

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Hyperfine Interactions of Fe²⁺ in Ilmenite

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Hyperfine interactions at the Fe²⁺ sites in ilmenite have been determined by Mössbauer spectroscopy. At 5 °K the internal magnetic field is (-43 ± 3) kOe and the quadrupole-coupling constant $\frac{1}{2}e^2 qQ$ is $(+1.44 \pm 0.01)$ mm/sec. Theoretical interpretation of the internal magnetic field in terms of dipolar-, orbital-, and core-polarization contributions is shown to depend critically upon the lattice contribution to $\frac{1}{2}e^2 qQ$, which is still unknown.

I. INTRODUCTION

The hyperfine structure of Fe^{+2} in ilmenite (FeTiO₃) was studied by low-temperature Mössbauer spectroscopy several years ago.¹ A relatively small internal magnetic field H_{int} was observed in the antiferromagnetic state and only partial resolution of the hyperfine spectrum was obtained. Theoretical calculations suggested that the sign of $H_{\rm int}$ was negative and that the core-polarization contribution to $H_{\rm int}$ was low.²

In conjunction with an investigation of the $Fe_{2-x}Ti_xO_3$ system³ with compositions similar to lunar ilmenite specimens, we have synthesized stoichio-





metric FeTiO₃. The liquid-helium-temperature Mössbauer spectrum of this material had narrow linewidths and hence showed a well-resolved hyperfine structure. An analysis of this spectrum and data on a single-crystal mineral sample is presented.

II. EXPERIMENTAL

Synthetic ilmenite was prepared by solid-state reaction of appropriate amounts of high-purity Fe, Fe₂O₃, and TiO₂; the iron powder was prepared from Johnson, Matthey and Co. iron sponge by reduction in H₂ for 1 h at 1175 °C followed by rapid cooling. The starting material was finely ground, compressed, and fired at 850 °C (2 h) and 1200 °C (2 h) in high vacuum ($\approx 10^{-8}$ mm of Hg). Powderdiffraction x-ray photography was used to establish that the material was single phase. The lattice constants measured for this material based on a hexagonal unit cell are a = 5.085 Å and c = 14.087 Å.

Mössbauer measurements were made with an automated mechanical constant-velocity spectrometer operating in the transmission mode; the spectrometer is described briefly elsewhere.⁴ The source used was Co^{57} in Cu at room temperature (≈ 22 °C).

The single-crystal mineral specimen of ilmenite (from Quebec) contained a considerable amount of exsolved hematite α -Fe₂O₃ and also about 10% Fe³⁺

in solution. Nonetheless the Fe^{2*} spectrum was sufficiently sharp and well resolved from the Fe^{3*} absorption lines to be clearly interpretable. The sample was oriented by back-reflection Laue xray-diffraction photography and mechanically polished \perp to the *c* axis to ≈ 1.8 -mil thickness.

III. RESULTS AND DISCUSSION

Ilmenite crystallizes in the trigonal space group $R\overline{3}$.⁵ The structure is derived from the α -Fe₂O₃ structure by replacing layers of Fe³⁺ which are perpendicular to the *c* axis by alternating layers of Fe²⁺ and Ti⁴⁺. All Fe²⁺ sites are crystallographically equivalent and have point symmetry 3. Neutron-diffraction, ⁶ magnetic, ⁷ and specific-heat measurements⁸ establish that the material orders antiferromagnetically below about 57 °K with spin directions $\|c$. The Fe²⁺ moments are ferromagnetically coupled within layers $\perp c$ and antiferromagnetically coupled between adjacent layers. The magnetic space group can be derived from this information and is $R_R\overline{3}$.

The Mössbauer spectrum of synthetic polycrystalline ilmenite at 5 °K is shown in Fig. 1. A leastsquares computer fit to six unconstrained Lorentz line shapes is shown through the data. The point symmetry of the Fe^{2*} site requires the nuclear electric field gradient to be axially symmetric with the principal axis z of the largest diagonalized ele-



FIG. 2. Dependence of core-polarization contribution to the internal magnetic field H_F in FeCO₃ and FeTiO₃ (calculated as described in text) on lattice contributions to the nuclear electric field gradient.

ment V_{zz} parallel to c. Thus H_{int} and z are \parallel and simple analytical expressions⁹ can be obtained for the transition energies in terms of $|H_{int}|$, $e^2 q Q$, and δ , where *e* is the proton charge, $q = V_{zz}/e$, Q is the nuclear quadrupole moment, and δ is the isomer shift. The typical six-line hyperfine pattern observed for nonmagnetized randomly oriented powders of magnetically ordered Fe⁵⁷ containing materials has relative absorption intensities 3: 2: 1: 1: 2: 3 in order of ascending energy. The small H_{int} and large $e^2 q Q$ observed in ilmenite at 5 °K shifts the $|-\frac{3}{2}\rangle - |-\frac{1}{2}\rangle$ transition to higher energy than the $|+\frac{1}{2}\rangle \rightarrow |+\frac{1}{2}\rangle$ transition and leads to relative intensities 2:1:1:2:3:3 in order of ascending energy. The experimental values of H_{int} , $\frac{1}{2}e^2 q Q$, and δ observed in the 5 °K spectrum of ilmenite are given in Table I.

To determine the sign of H_{int} , an external magnetic field H_{ext} of 55 kOe was applied to the singlecrystal mineral specimen of ilmenite. The absorber was at 87 °K (in the paramagnetic state) and $\vec{H}_{ext} \parallel c$. The internal field observed at the Fe²⁺ nuclei was (41±3) kOe, which experimentally established the negative sign of H_{int} predicted by theoretical arguments.² At 87 °K and $H_{\rm ext}$ = 55 kOe, the magnetization is far from saturated which accounts for the small reduction in $H_{\rm int}$.

Okiji and Kanamori² have calculated the orbital H_{orb} and dipolar H_{dip} contributions to H_{int} for the Fe²⁺ sites in the trigonal crystals FeTiO₃ and FeCO₃ (ferrous carbonate crystallizes in space group $R \ \bar{3}c$ and the Fe²⁺ site has $\bar{3}$ point symmetry). They express both H_{orb} and H_{dip} in terms of the valence contribution to the nuclear electric field gradient and calculate the core-polarization term H_F in conjunction with the experimental values of H_{int} . Based on previously reported FeTiO₃ results $(|H_{int}| = 70 \text{ kOe and } \frac{1}{2}e^2qQ = +1.14 \text{ mm/sec})^1$ they obtained $H_F = -455 \text{ kOe}$. This rather low value of

TABLE I. Hyperfine parameters of ${\rm Fe}^{2\star}$ in ${\rm FeTiO}_3$ at 5 °K.

| H _{int} (kOe) | -43 ± 3 |
|-----------------------------|------------------|
| $\frac{1}{2}e^2qQ$ (mm/sec) | $+1.44 \pm 0.01$ |
| $\delta (mm/sec)^2$ | $+1.22 \pm 0.01$ |

^aRelative to α -Fe at room temperature.

 H_F compared with the expected free-ion value^{2,10} was attributed by Okiji and Kanamori,² and more recently by Hazony,¹¹ to the covalent nature of Fe²⁺ in FeTiO₃. Our present low-temperature FeTiO₃ results (Table I) differ considerably from previous ones; in addition, improved liquid-helium-temperature data for $FeCO_3$ are now available $[H_{int} = (+185)]$ ± 3) kOe, $\frac{1}{2}e^2qQ = (+2.05 \pm 0.03)$ mm/sec].¹²⁻¹⁴ Thus we have recalculated the internal-field contributions in both $FeTiO_3$ and $FeCO_3$ making the same assumptions as were made by Okiji and Kanamori. The recalculated values for $FeTiO_3$ are $H_{orb} = +420$, $H_{dip} = +59$, and $H_F = -522$ kOe; for FeCO₃, H_{orb} =+579, $H_{dip}=+85$, and $H_F=-479$ kOe.

In general, the quadrupole-coupling constant in ionic Fe²⁺ compounds will have both lattice and valence contributions. Usually neglect of lattice contributions to q is a major source of error in relating H_F , as calculated above, to the degree of covalency. Because of the axial symmetry present at the Fe^{2+} sites in both $FeTiO_3$ and $FeCO_3$, the valence and lattice contributions to q may be ex-

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- ¹G. Shirane, D. E. Cox, W. J. Takei, and S. L. Ruby, J. Phys. Soc. Japan 17, 1598 (1962); G. Shirane and S.
- L. Ruby, ibid. Suppl. B1 17, 133 (1962).
- ²A. Okiji and J. Kanamori, J. Phys. Soc. Japan <u>19</u>, 908 (1964).
- ³R. W. Grant, R. M. Housley, and S. Geller (unpublished).
- ⁴R. W. Grant, R. M. Housley, and U. Gonser, Phys. Rev. 178, 523 (1969).
- ⁵T. F. N. Barth and E. Posnjak, Z. Krist. 88, 265 (1934).
- ⁶G. Shirane, S. J. Pickart, R. Nathans, and Y. Ishikawa, J. Phys. Chem. Solids 10, 35 (1959).

pressed simply as $q = q_{val} + q_{1at}$. The extreme sensitivity H_F to q_{1at} is shown in Fig. 2 where we have evaluated H_F by the method of Okiji and Kanamori but have considered the possible presence of a small q_{1at} term. In ionic Fe³⁺ compounds where only q_{1at} is present, $\frac{1}{2}e^2qQ$ values > 1 mm/sec are observed. In α -Fe₂O₃, which is structurally similar to FeTiO₃, the quadrupole splitting is $\frac{1}{2}e^2qQ$ = (+ 0. 440 ± 0. 012) mm/sec.¹⁵ If a q_{1at} term of this magnitude were present in FeTiO₃, the derived value of H_F would change by 150 kOe (Fig. 2). Thus, until lattice contributions to q can be determined with confidence, correlations of covalency and calculated H_F values must be rather uncertain.

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- ⁷H. Bizette and B. Tsai, Compt. Rend. <u>242</u>, 2124 (1956).
 - ⁸C. H. Shomate, J. Am. Chem. Soc. 68, 964 (1946).
- ⁹See, e.g., K. Ono and A. Ito, J. Phys. Soc. Japan <u>19</u>, 899 (1964).
- ¹⁰R. E. Watson and A. J. Freeman, Phys. Rev. <u>123</u>, 2027 (1961).

¹¹Y. Hazony, Phys. Rev. B 3, 711 (1971).

- ¹²U. Gonser, R. M. Housley, and R. W. Grant, Phys. Letters 29A, 36 (1969).
- ¹³D. W. Forester and N. C. Koon, J. Appl. Phys. <u>40</u>, 1316 (1969).

¹⁴H. N. Ok, Phys. Rev. 185, 472 (1969).

¹⁵J. O. Artman, A. H. Muir, Jr., and H. Wiedersich, Phys. Rev. 173, 337 (1968).