

Theoretical study of As₂ desorption from the Ga dangling-bond site

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The cluster model As₂-(GaH₂)₄ is used for *ab initio* complete active space self-consistent-field (CASSCF) molecular-orbital calculations, to investigate the electronic mechanism of As₂ desorption from the fourfold-hollow Ga dangling-bond (DB) site. The electronic state of the adsorbed As₂ is revealed to be the $\sigma_g^2 \pi_u^2 \pi_g^2$ doubly excited configuration, where the free As₂ molecule is characterized by $\sigma_g^2 \pi_u^4$ with the triple bond. Namely, the four π orbitals of the As₂ unit are used to form the four Ga-As covalent bonds with the Ga-DB site and a weak σ bond remains between the two As atoms. The calculated CASSCF wave functions indicate that the desorption process could be just the chemical reaction of multiple bond conversions having strong electron correlations due to the near degeneracy. The single-configurational picture is completely invalid in understanding the As₂ desorption. The transition state is found to be of the "late-barrier" type, and from this the "cold" vibrational distribution of the desorbed As₂ is predicted.

I. INTRODUCTION

Very recently, a series of *ab initio* molecular-orbital (MO) calculations have been performed to investigate the Ga layer growth mechanism in the chloride atomic layer epitaxy (ALE) process.¹⁻³ In the chloride ALE, to grow the Ga monolayer, the GaCl molecules are supplied together with the carrier H₂ gas at the As dangling-bond (DB) site. The fundamental electronic mechanism of Ga layer growth has been revealed as follows: First, the GaCl is absorbed by forming Ga-As covalent bonds through configuration mixing.¹ Nondynamical electron correlations are essential to describe properly the GaCl/As-DB adsorption process because of the near degeneracy, as in the case of SiCl₂/Si-DB.⁴ It is notable here that the electronic structure of the adsorbed GaCl does not correlate with the ground state $^1\Sigma^+$ but does correlate with the valence excited state $^3\Pi$. At the second stage, the H₂ removes the Cl atom from Ga as HCl by the concerted reaction.^{2,3} The Cl removal should be followed by the formation of the Ga-Ga reactive "dimer." Cluster models have been used in Refs. 1-4 and the results show that the electrons concerned with surface reactions seem to be reasonably localized around the active reaction site. Therefore, the cluster approach can be a useful tool as long as discussions are focused on the nature of reaction mechanisms.

In this paper, the electronic mechanism of As₂ desorption from the Ga-DB site will be analyzed in detail by using the MO-cluster approach, taking electron correlations into account. Since the control of As stoichiometry during growth is very important to obtaining defectless GaAs crystals, the surface chemistry of As layers is of special interest for general epitaxial growth processes including ALE or molecular-beam epitaxy (MBE). Thus, there have been extensive experimental and theoretical

studies on the As surface.⁵⁻³² The loss of As atoms crucially affects the surface stoichiometry and this leads to a variety of reconstructions. In other words, the stable surface phase depends on the As coverage Θ_{As} . The most probable loss route is the As₂ desorption (or sublimation). The energetics of As₂ desorption are investigated mostly in the temperature-programmed desorption (TPD) experiments.⁵⁻¹¹ The stable structures of the As layer are measured by methods such as scanning tunneling microscopy (STM), reflection high-energy electron diffraction (RHEED), and x-ray diffraction.¹²⁻²⁴ The associated theoretical calculations are based on the density-functional (DF) and tight-binding schemes.²⁵⁻³² For example, in the case of the (100) surface of $\Theta_{As}=0.75$, the stable phase seems to be settled as $2 \times 4-c(2 \times 8)$ which contains the three As dimer rows and the one "missing-dimer" row in the unit cell. The experimental TPD values for this structure are reported to be 1.95,⁵ 2.25,⁸ and 2.42 eV.⁹ Ohno has reported a corresponding estimate of 2.8 eV by using the DF-slab calculations.³² This value compares reasonably with experimental values 1.95-2.42 eV (Refs. 5 and 8-9). However, to our knowledge, there have been no studies involving the electronic mechanism of the As₂ desorption process. The As dimer adsorbed on the fourfold-hollow Ga site of the (100) direction should have an As-As σ bond. Namely, each As atom of the dimer has two Ga-As bonds, one As-As σ bond, and one 4s-like lone pair. On the contrary, the free As dimer or the As₂ molecule has the triple bond of $\sigma, \pi \times 2$, whose main configuration is isovalent with N₂ and is symbolically given by

$$X \ ^1\Sigma_g^+ : \sigma_g^2 \pi_u^4 . \quad (1)$$

Except for some changes of the hybridization, this formally predicts that all four π orbitals of As₂ should be

used to form the four Ga-As bonds through the excitation of $\pi_u^2 \rightarrow \pi_g^2$. When the As₂ completely desorbs from the Ga-DB site, the Ga-Ga "dimerization" is also expected. This naturally suggests that the As₂ desorption process could be understood as a chemical reaction of multiple bond conversions, where the four Ga-As bonds are being broken and the two As-As π bonds of As₂ and the two Ga-Ga bonds of the DB site are being created. From the viewpoint of chemical reactions, such multiple bond conversions would be the common feature of As₂ desorption independent of various surface phases. Therefore, the present MO-cluster approach, which will illuminate desorption mechanisms such as configuration mixing, should be informative for understanding the As surface chemistry.

The complete active space self-consistent-field³³ (CASSCF) scheme is applied in the present theoretical investigation, since the nondynamical correlation due to the near degeneracy of many electronic configurations is highly expected for the multiple bond conversions. In fact, the system will show strong configuration mixing. The CASSCF calculations are performed on the cluster model As₂-(GaH₂)₄ which mimics the As dimer on the fourfold-hollow Ga-DB site. According to previous studies,¹⁻⁴ although the present cluster is the simplest model, it is still sufficient to qualitatively understand the bond-conversion process during the As₂ desorption. The two-dimensional potential-energy surface (PES) is calculated, where the geometrical parameters are the distance between As₂ and the Ga-DB surface and the As-As distance. The structures of the stable surface As dimer and the transition state (TS) are determined, and the characterizations of the PES of desorption are also made.

II. COMPUTATIONAL DETAILS

A. Basis set

In the present calculations, the model potential (MP) technique³⁴ is applied to replace the 28 1s-3d core atomic orbitals (AO's) of the Ga and As atoms. Namely, the 4s4p valence electrons of Ga and As are explicitly treated using the basis-set expansion. The valence *Gaussian* basis set is 5s5p, whose functions are contracted to (4,1/3,1,1). The core AO potentials are also constructed by the *Gaussian* sets of 12s9p4d. These numerical parameters were determined by two of the authors (Y.S. and E.M.).³⁵ For these Ga and As valence basis functions, the sets of diffuse-*s* and polarization-*d* functions are augmented. The exponents for Ga are 0.023 384 9 for diffuse *s* and 0.207 (Ref. 36) for polarization *d*. The corresponding values for As are 0.039 249 9 and 0.293.³⁶ The extra *s* exponent is derived from the two outermost exponents of 5s in the geometric mean manner. The contracted sets for Ga and As can be written as [3s3p1d]. For H in the GaH₂ unit, Van Duijneveldt's 5s primitives³⁷ are contracted to (4,1). The total number of basis functions is 118 for the cluster model.

B. Cluster model

The cluster model As₂-(GaH₂)₄ illustrated in Fig. 1 is used to calculate the CASSCF PES of the As₂ desorption.

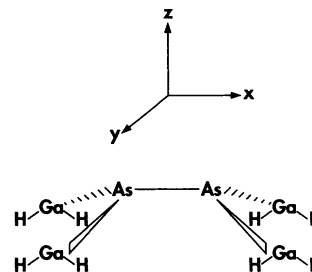


FIG. 1. Structure of the As₂-(GaH₂)₄ cluster.

This locally "extracts" the single As dimer unit on the fourfold-hollow Ga-DB site of the (100) surface. Since the symmetry of this cluster is C_{2v} , the MO's are classified into the four subgroups of C_{2v} , i.e., a_1 , a_2 , b_1 and b_2 . The C_{2v} framework is kept throughout the calculations. Namely, the As₂ is imposed to be desorbing along the *z* axis without any tilting and twisting. The C_{2v} restriction could make the orbital phasings between As₂ and Ga-DB clearest. The lower symmetry cases will be considered in Sec. III B. Note that the As₂ $^1\Sigma_g^+$ state (1) of the $D_{\infty h}$ symmetry correlates with the 1A_1 state under the C_{2v} symmetry group currently used.

When the desorption is completed, the free Ga-DB site should be dimerized with some reconstructions. However, the reconstruction at the dimerized Ga site is neglected in the present model, and all the Ga-Ga distances are fixed at 4 Å (=7.55 a.u.) of the GaAs bulk value throughout the calculations, as done in the previous studies.^{1,4} The effect of this constraint will be discussed in Sec. III B. The inner Ga-As bonds, which are directed toward the As sublayer under the Ga-DB site, are simply replaced by the isopolar Ga-H bonds, where the positive polarity is on the Ga atom.^{2,3} Thus, the fourfold-hollow Ga-DB site is mimicked by the (GaH₂)₄ cluster in the current simulation. The local symmetry of the GaH₂ fragment is also the C_{2v} point group and the ground state is 2A_1 with the one open-shell MO (a_1 symmetry). The bond angle of H-Ga-H is set at 109.5° by assuming sp^3 hybridization. With the fixed angle of 109.5°, the Ga-H bond length is optimized to be 1.55 Å (=2.93 a.u.) by using the five-electron SCF calculations because of the MP replacement. The GaH₂ unit has an empty Ga 4p(b_1) orbital which is perpendicular to the H-Ga-H plane. The Ga sp^3 lobe to be connected with the As dimer should be formed by mixing the open-shell a_1 MO and the empty b_1 MO of each GaH₂ and the four open-shell electrons are active, or are concerned with, the Ga-As bonds. At the desorption limit, the dimerized Ga-DB site with the two Ga-Ga bonds is formally represented by the lowest 1A_1 state of (GaH₂)₄, although there is the multiconfigurational character, as pointed out in Refs. 1 and 4. Note that the other Ga sp^3 lobes, where are directed away from the As₂ side, are kept vacant throughout the present computational scheme.

C. CASSCF calculations

When the MP is used to replace the 28 1s-3d core electrons, the total number of electrons of the As₂ unit is 10.

In the free As_2 case, the six electrons are actively involved in the triple bond of configuration (1) and the remaining four electrons are of the semicore $4s$ type. Similarly, for $(\text{GaH}_2)_4$, the total number of electrons is 20 ($=5 \times 4$), where the four electrons are of the active DB type and the other 16 electrons belong to the terminating Ga-H bonds. Thus, the total number of electrons is 30 for the As_2 - $(\text{GaH}_2)_4$ cluster.

According to the prediction in Sec. I the four Ga-As bonding MO's could be schematically decomposed into components of As_2 and Ga-DB, as illustrated in Fig. 2. Note that a similar figure was used by Panas, Siegbahn, and Wahlgren who analyzed the O_2 adsorption onto the fourfold-hollow Ni (100) surface by using the MO-cluster approach.³⁸ Hoffmann's "isolobal analogy" concept³⁹ relates the present MO decomposition of $\text{As}_2/\text{Ga-DB}$ with that of O_2/Ni . The z and y components of π orbitals of As_2 fragments are perpendicular and parallel to the Ga surface, respectively. The As_2 π orbitals are labeled by a_1 for π_{uz} , b_2 for π_{uy} , b_1 for π_{gz} , and a_2 for π_{gy} , under the present C_{2v} coordinate system. The four possible combinations of active sp^3 lobes of each GaH_2 also present symmetric orbitals, $\text{DB}a_1$, $\text{DB}a_2$, and $\text{DB}b_1$, and $\text{DB}b_2$. Therefore, as shown in Fig. 2, the four Ga-As bonding orbitals result from the in-phase mixings between As_2 and Ga-DB sides.

In the stable As dimer structure, the single closed-shell 1A_1 configuration

$$^1A_1: (1a_1 - 5a_1)^{10} (1a_2 - 3a_2)^5 (1b_1 - 4b_1)^8 (1b_2 - 3b_2)^6 \quad (2)$$

dominates the CASSCF wave function of the cluster system, where its weight will be shown to be 86% in Sec. III B. The $5a_1$ MO, which consists of As $4p_x$ AO's, describes the As-As σ bonding orbital. The $4a_1$, $3b_2$, $4b_1$, and $3a_2$ of the MO correspond, respectively, to the Ga-As bonding MO's illustrated in Fig. 2, $\pi_{uz} + \text{DB}a_1$, $\pi_{uy} + \text{DB}b_2$, $\pi_{gz} + \text{DB}b_1$, and $\pi_{gy} + \text{DB}a_2$. However, if the system even slightly proceeds to the desorption from the stable dimer structure, the single-configurational description immediately becomes a poor approximation. Namely, the multiconfigurational description is needed because of the near degeneracy among excited configurations. Such multiconfigurational characters

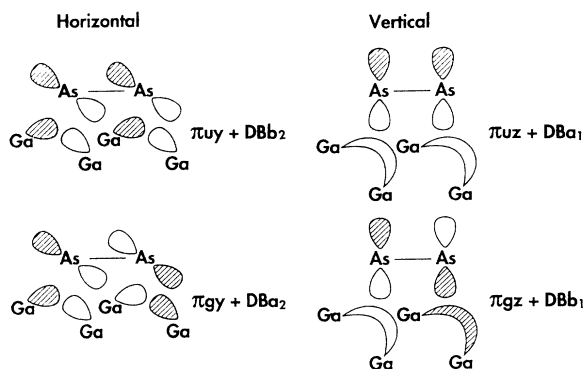


FIG. 2. Schematic illustration of the four Ga-As bonds.

should decrease the occupations of Ga-As bonding MO's and increase those of antibonding MO's that have the out-of-phase combinations between As_2 and Ga-DB. Finally, at the desorption limit, the total of eight bonding and antibonding Ga-As MO's have to separate into the four As_2 π and four Ga-DB fragmental orbitals, where the state of free As_2 is $^1\Sigma_g^+$ ($=^1A_1$) and that of dimerized $(\text{GaH}_2)_4$ is 1A_1 . The changes of π -electron configurations should affect the σ bond length of the As_2 unit, and the proper description of σ space requires the pairing of $\sigma_g(a_1)$ and $\sigma_u(b_1)$. Therefore, the CASSCF active orbital space consists of the σ_g , π_u , π_g , and σ_u of As_2 and the $\text{DB}a_1$, $\text{DB}a_2$, $\text{DB}b_1$, and $\text{DB}b_2$ of $(\text{GaH}_2)_4$. These ten orbitals correlate with the MO's of $4a_1-6a_1$, $3a_2-4a_2$, $4b_1-6b_1$, and $3b_2-4b_2$ in the As_2 - $(\text{GaH}_2)_4$ cluster. Under the constraint of the 1A_1 total symmetry, the ten [six from As_2 and the remaining from $(\text{GaH}_2)_4$] active electrons are distributed among the ten active MO's in all possible ways. Thus, the complete active space configuration-interaction (CASCI) dimension is 4936. The remaining ten occupied MO's are defined as inactive and are composed of two $4s$ -like MO's of As_2 and the eight Ga-H bonding orbitals. Note that the two lone pairs of the stable As dimer could be described by the As $4s$ inactive MO's through some hybridizations with $4p$.

Unfortunately, the further expansion of the active orbital space, e.g., inclusion of the four vacant Ga sp^3 lobes, is impossible due to the dimensional explosion. The successive multireference singly and doubly excited CI (MRSDCI) calculations, which incorporate dynamical correlations in the system, are definitely desired but are similarly intractable on the current computational resources. In other words, only nondynamical correlations are taken into account by CASSCF in the present computational scheme.

By changing the two geometrical parameters, the distance between As_2 and the Ga surface and the As-As distance, more than 100 grid points are calculated. The stable structure of the As dimer is determined by the global fitting into the sixth-order polynomial. In contrast, the TS structure is obtained by second-order local fitting.

III. RESULTS AND DISCUSSION

A. Electronic structure of As_2

The electronic structure of the As_2 unit must be changing during the desorption process. Therefore, before discussing the mechanism of As_2 desorption, the electronic structure of As_2 itself should be reviewed and the reliability of the present computational scheme should be proved. Table I summarizes the CASSCF spectroscopic constants and important configurations for the $X^1\Sigma_g^+$ As_2 molecule. To assess the present results, the Table also contains the experimental data⁴⁰ and the other theoretical estimates⁴¹⁻⁴³ published previously. In Refs. 41 and 42, the effective core potential (ECP) technique was used to replace the $1s-3d$ AO's, as in the present MP calculations. The quality of the valence basis could be virtually the same for the calculations listed in Table I.

First, Table I shows that the noncorrelated SCF

TABLE I. CASSCF results for the $X^1\Sigma_g^+$ As₂.

Experimental	R_e (a.u.)	ω_e (cm ⁻¹)	D_e (eV)
Experimental Ref. 37	3.97	430	4.01
Theoretical, present MP [3s3p1d]/CASSCF ^a	4.10	393	2.74
Theoretical, Ref. 41, ECP [3s3p1d]/MRSCI	4.09	394	2.71
Theoretical, Ref. 42, ECP [3s3p1d]/SCF	3.95	508	-0.72
	/MRSDCI	429	2.54
Theoretical, Ref. 43, all-electron [7s5p2d]/SCF	3.91	514	0.43
	/CCSD	437	2.67

Configuration	Excitation	Weight (%) ^b
$\sigma_g^2\pi_u^4$		87.17
$\sigma_g^2\pi_u^2\pi_g^2$	$\pi_u^2 \rightarrow \pi_g^2$	9.30
$\sigma_g\pi_u^3\pi_g\sigma_u$	$\sigma_g\pi_u \rightarrow \pi_g\sigma_u$	2.03

^aThe R_e is determined by the fourth-order polynomial fitting. The total energy at R_e is -12.1509 a.u.

^b Configurations whose weights are greater than 0.8% are shown.

values^{42,43} for ω_e and D_e are quite poor even for the free As₂ molecule. The reason for this is apparently the near degeneracy, which is demonstrated by the weights in the present CASSCF wave function. Its artificially enhanced single-configurational character is responsible for the seemingly close correspondence between the SCR R_e of Ref. 42 and the experimental value of 3.97 a.u. Hereafter, our discussions will focus on the correlated results.

The types of CI calculations used in Refs. 41 and 42 are MRSCI [or first-order CI (FOCI)] and MRSDCI, respectively. Note that the MRSCI hardly takes dynamical correlations into account. In contrast, the coupled-cluster single and double excitations (CCSD) theory⁴⁴ in Ref. 43 is started on the SCF single reference, but the multireference nature is partly introduced by the incorporation of higher-order multiple (more than two-body) excitations. Thus, the CCSD method is applicable for a system having some near degeneracies⁴⁵ such as As₂ around R_e .

The As-As equilibrium distance R_e is currently calculated to be 4.10 a.u. which is appreciably longer than the experimental length of 3.97 a.u. The other theoretical R_e 's show the same trend. In Ref. 43, Scuseria also reported the very accurate CCSD value of 3.98 a.u. which was obtained by using the more extended basis of [7s6p4d1f] and by further correlating the outer-core 3d electrons. Namely, the extensively correlated treatment, which takes the higher-order excitations, the up-to-f polarizations, and the additional correlations from outer-core electrons into account, should be required to quantitatively estimate the R_e of As₂. Such calculation is unable to perform the As₂-(GaH₂)₄ cluster system. Therefore, a certain correction should be considered, if the calculated As-As bond length for the stable As dimer on the Ga-DB site is compared with the experimental data in Sec. III B.

The present ω_e of 393 cm⁻¹ is in reasonable agreement with the experiment (430 cm⁻¹). The D_e of 2.74 eV is similar with the other theoretical values,⁴¹⁻⁴³ although there is still an error more of than 1 eV in comparison with the experimental data of 4.01 eV.⁴⁰ As a whole, the

present CASSCF procedure is expected to have an acceptable reliability for the purpose of investigating the electronic mechanism of As₂ desorption from the Ga-DB site.

The near degeneracy in the π space is notable for the As₂ molecule, as seen in Table I. Namely, the nondynamical correlation is important even at the R_e . The $\pi_u^2 \rightarrow \pi_g^2$ doubly excited configuration $\sigma_g^2\pi_u^2\pi_g^2$ should describe the left-right correlation of the π electrons, where its weight in the CASSCF wave function is 9.3%. Balasubramanian also pointed out the importance of the $\pi_u^2 \rightarrow \pi_g^2$ excited configuration⁴¹ for the As₂ molecule. The mixing of $\sigma_g^2\pi_u^2\pi_g^2$ substantially reduces the formal bond order of 3 given by the $\sigma_g^2\pi_u^4$ ground configuration (1). The effective configuration of $X^1\Sigma_g^+$ As₂ at R_e can be represented as

$$\sigma_g^{1.96}\pi_u^{3.76}\pi_g^{0.24}\sigma_u^{0.04} \quad (3)$$

by the occupation numbers of present CASSCF active orbitals.

In Sec. I of this paper, the configuration of the stable As dimer on surface was formally predicted to be $\sigma_g^2\pi_u^2\pi_g^2$, which results from the double excitation and has the single σ bond. Thus, the excited states should be of interest. Balasubramanian already surveyed the singlet and triplet excited states.⁴¹ The lowest one was calculated to be the singly excited $c^3\Sigma_u^+$ state whose main configuration is $\sigma_g^2\pi_u^3\pi_g$, where the estimated excitation energy of 1.47 eV is comparable to the experimental value of 1.80 eV. Balasubramanian⁴¹ also showed that the doubly excited states, which are characterized by $\sigma_g^2\pi_u^2\pi_g^2$, are quasibound relative to the ground-state dissociation, but are low lying, where the excitation energies are 4.06 eV for $^1\Sigma_g^+$ (II) and 3.80 eV for $^3\Delta_g$. The quintet excited state $^5\Sigma_g^+$, which was not calculated in Ref. 41, would be lower than these singlet and triplet states, due to the exchange interaction. According to the intramolecular Hund's rule, the contribution of the quintet $\sigma_g^2\pi_u^2\pi_g^2$ might be mostly favored to form the four Ga-As bonds with the fourfold-hollow Ga-DB site, as seen in the previous study of GaCl/As-DB.¹ The CASSCF results

for the doubly excited ${}^5\Sigma_g^+$ will be shown in Sec. III B, and compared with the character of the stable As dimer on the Ga surface.

B. Nature of As₂ desorption from Ga-DB

Figure 3 illustrates the schematic view of CASSCF PES for the As₂ desorption of the 1A_1 state. The energetic summary of PES is given in Table II. Prior to focusing the discussions on the electronic mechanism of desorption, the calculated energetics and the stable dimer length are compared with the available experimental data, checking the reliability of the present investigation.

First, the TPD value is considered. Since the TS exists during the desorption, the theoretical TPD energy is estimated to be 2.65 eV by summing up the well depth (2.19 eV) and the barrier height of TS (0.46 eV). The value of 2.65 eV compares with the experimental TPD values of 1.95–2.42 eV.^{5,8,9} However, some problematic concerns should be pointed out for this direct comparison, as mentioned in Ref. 1. The first is the so-called “cluster size” problem which has been extensively studied in the adsorption energies on the transition-metal surfaces.^{46–51} Namely, the computational energetics depends on the sizes of model clusters, where their deviations are roughly

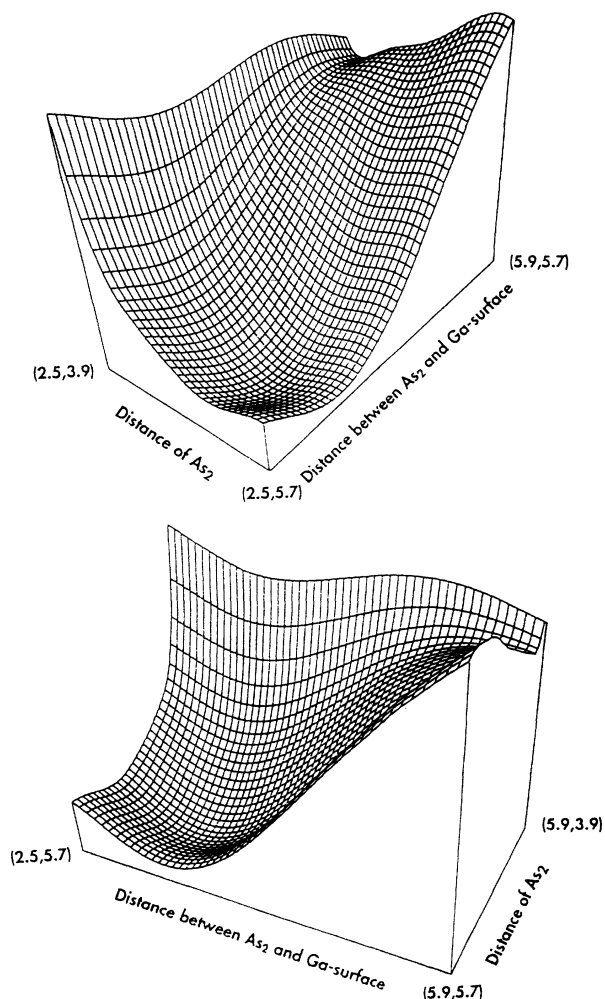


FIG. 3. CASSCF energy surface for the 1A_1 As₂ desorption. Two figures are drawn by changing the orientations.

TABLE II. Distance parameters and energies (relative energies to the desorption limit; the total energy at the desorption limit is -24.5582 a.u., where the As-As distance is 4.10 a.u. in Table I) for the stable surface dimer and the TS of desorption.

	As ₂ -Ga surface (a.u.)	As-As (a.u.)	Energies (eV)
Stable dimer	3.34	5.14	-2.19
TS	5.71	4.22	0.46

0.5–1.0 eV. A similar situation could also take place for semiconductor surfaces such as GaAs. To see the convergence trend for quantitative discussions, a series of systematically enlarged cluster models should be employed. The other two concerns, which should have some effects on the energetics, are the geometrical constraint noted in Sec. II B and the missing of dynamical correlations caused by the impossible execution of MRSDCI (Sec. II C). Recall that, in the present investigation, the CASSCF calculations are performed on the simplest cluster As₂-(GaH₂)₄ only to investigate the electronic mechanism of As₂ desorption such as configuration mixing. Although we mention above three problematic concerns, the first estimate by the *ab initio* MO-cluster approach was 2.65 eV, which is qualitatively acceptable for the present purpose. Definitely, future extensive *ab initio* MO calculations, in which the size dependence, the geometrical freedom, and the dynamical correlation are all taken into account, are required to quantitatively compare the theoretical As₂ TPD energies with the experimental value of 1.95–2.42 eV.^{5,8,9}

The second comparison with the experiment concerns the As-As distance of the stable dimer. The experimental dimer distance has been measured to be 2.56–2.62 Å (=4.84–4.95 a.u.) for the $c(4\times 4)$ structure by the x-ray diffraction.¹⁷ Even considering some geometrical constraints imposed on the present model cluster, the estimated value of 5.14 a.u. in Table II is appreciably longer than the experiment. The most probable reason for this disagreement could be the basis-set flexibility and the dynamical correlation effect involving further 3d electrons, as seen in Sec. III A for the free As₂ molecule. The calculated R_e for As₂ was longer than the experimental value⁴⁰ by 0.13 a.u. (see Table I). If the correction of 0.13 a.u. is adapted for the surface dimer, the corrected estimate of 5.01 a.u. (= 2.65 Å) seems to be acceptable relative to the experiment.¹⁷ This again implies that the present calculations are reliable enough to qualitatively reveal the electronic mechanism of desorption.

Now, we will discuss the nature of As₂ desorption. Figure 3 shows that the TS of desorption is located far from the stable As dimer structure. Namely, the TS is characterized to be of Polanyi’s “late-barrier” type,⁵² which is generally known for endothermic reactions as in the present case. The As-As distance of 4.22 a.u. at the TS is considerably shortened from that of the stable structure of 5.14 a.u. The close correspondence between 4.22 a.u. of TS and 4.10 a.u. of the free As₂ molecule is consistent with the “lateness.” This suggests that the As₂ π bond is “slowly” reproduced during the desorption.

Polanyi's rule⁵² could predict that the vibrational distributions of the products as "cold" for the reactions of the "late-barrier" type. Therefore, the vibrational distributions of the desorbed As₂ from the Ga-DB site are expected to be "cold." Recently, using the laser-induced-fluorescence (LIF) spectroscopy, Alstrin *et al.* have measured the vibrational distributions of the As₂ molecules which have just desorbed from the Si (100) and (111) surfaces.⁵³ They actually reported the "cold" vibrational distributions of As₂ for both (100) and (111) orientations and speculated that the desorption process, in which the Si-As covalent bonds should be broken, has a localized nature. Alstrin *et al.*⁵³ also pointed out that the covalent bonding to the Si surface would make the well depth of the As dimer bond more shallow. Although there should be some quantitative differences between the cases of Ga/As₂ and Si/As₂, their results are in accord with the present investigation. Reference 53 did not treat the rotational distributions. Note that the "cold" rotational distributions have been known for the recombinatively desorbed H₂ from Si surfaces.^{54,55} In the present study, to illustrate clearly the orbital interactions between As₂ and Ga-DB during the desorption, the C_{2v} symmetry restriction is imposed. Namely, the motion of As₂ is forced to be in the concerted manner and there is no torque to rotate the desorbing As₂ molecule. When the system has an As₂ tilting in the *xz* plane, the symmetry is lowered to C_s. The two Ga-As bonds associated with the one As atom should break first in the C_s desorption. In contrast, the desorption system belongs to the lowest C₁ symmetry, if the As₂ is twisted in the *xy* plane. The C₁ desorption may favor the first breaking of the two diagonal Ga-As bonds. Thus, the rotational motions could be induced for the routes of C_s (vertical to the surface) and C₁ (horizontal). According to the Woodward-Hoffmann rules, the C₁ desorption is expected to be the lowest route. Unfortunately, due to more geometrical parameters, the correlated calculations for the lower symmetry PES's are much more demanding than the present C_{2v} case. The experiments on rotational distributions will make the nature of As₂ desorption more clear. The rovibrational distributions should be experimentally measured for the Ga/As₂ system.

The four Ga-As bonds in Fig. 2 are symbolically described in the completely covalent fashion. However, as expected by the difference of electronegativities, a polarity of Ga^{δ+}-As^{δ-} should exist due to the contributions of ionic or dative types. In fact, the active bonding MO's have the larger components on the As₂ side than on the Ga-DB side. At the stable dimer structure, the Mulliken analyses on the active orbitals show that the weights of each As₂ component for a₁, b₂, b₁, and a₂ bonding MO's are 63% (π_{uz}), 56% (π_{uv}), 54% (π_{gz}), and 50% (π_{gy}), respectively. The corresponding antibonding active MO's which are weakly occupied at the stable structure must result from the counter mixings.

Table III shows the important configurations of the CASSCF wave functions and Table IV lists the resulting occupation numbers of active MO's. These tables impressively demonstrate that the significant multiconfig-

urational character appears when the system goes to desorption. The main configuration given by (2) dominates the CASSCF wave function by a weight of 86% at the stable dimer structure. However, the single-configurational picture is valid only around the bottom of PES. Toward the desorption, various excited configurations must mix to describe the multiple bond-conversion reaction in which the four Ga-As bonds are being broken and the two As₂ π bonds and the two Ga-Ga bonds are being created. In other words, nondynamical correlations are essential to trace properly the PES of the As₂ desorption. Reference 1 has also shown the multiconfigurational character during the GaCl adsorption onto the twofold-hollow As-DB site, but its degree is less significant than the present system, As₂/Ga-DB. The reason for this may be the difference of the number of bonds to be converted [currently four versus two previously (Ref. 1)].

The characteristics of active orbitals, which are sorted according to the occupation numbers in Table IV, are being changed by the progress of desorption. For example, the 4a₁ active MO corresponds to the π_{uz} + DBa₁ bonding orbital and the 5a₁ MO is of the As-As σ_g bonding type, at the stable As dimer structure. But the relation is reversed at the sample point of intermediate 1. The 3a₂ MO also changes from the π_{gy} + DBa₂ type (stable point) to essentially DBa₂ (TS). If one wishes to watch the effective configurations of the As₂ unit during the desorption process, this situation of active MO's is inconvenient. Thus, by the use of Mulliken populations of the active MO's, the effective occupation numbers of σ_g, π_u, π_g, and σ_u are estimated for the above object and the resulting numerical values are summarized in Table V. Note that the similar analysis was used by Panas, Siegbahn, and Wahlgren who calculated the O₂/Ni adsorption system.³⁸ The effective configuration of the As₂ fragment is shown to be

$$\sigma_g^{1.83} \pi_u^{2.41} \pi_g^{2.07} \sigma_u^{0.08} \quad (4)$$

at the stable structure. This is considered to essentially be the doubly excited configuration σ_g²π_u²π_g², being in exact agreement with the formal prediction in Sec. I. The π_u occupation increase of 0.4 should be attributed to the polarized Ga-As bonds in a₁ and b₂ symmetries, as discussed before. Intramolecular Hund's rule suggests that the quintet contribution of σ_g²π_u²π_g² doubly excited configuration could be mostly favored in the Ga-As bond formation, although the singlet and triplet contributions are also possible. To date, no quintet states have been calculated for As₂. Therefore, the CASSCF calculations are performed on the ⁵Σ_g⁺ state of σ_g²π_u²π_g². The calculations show that this state is bound with a very shallow well of 0.16 eV relative to the dissociation limit. The calculated spectroscopic constants for ⁵Σ_g⁺ are 5.16 a.u. for R_e, 166 cm⁻¹ for ω_e, and 2.58 eV for T_e. These values indicate that the remaining σ_g bond is weak. The ¹Σ_g⁺ → ⁵Σ_g⁺ excitation energy of 2.58 eV is actually lower than 4.06 eV for ¹Σ_g⁺(II) and 3.80 eV for ³Δ_g, which were calculated by Balasubramanian.⁴¹ Thus, the highest-spin

TABLE III. Important configurations (configurations with weights greater than 0.8% are listed) of the CASSCF wave functions of the $\text{As}_2\text{-(GaH}_2)_4$ cluster.

Sample ^a	$4a_1$	$5a_1$	$6a_1$	$3a_2$	$4a_2$	$4b_1$	$5b_1$	$6b_1$	$3b_2$	$4b_2$	Weight (%) ^c
Stable dimer											
(3.34,5.14)	2	2	0	2	0	2	0	0	2	0	86.32
	2	0	0	2	0	2	2	0	2	0	1.48
	2	2	0	1	1	2	0	0	1	1	0.94
	2	0	0	2	0	2	1	1	2	0	0.82
											Sum 89.56
Intermediate 1											
(4.00,4.90)	2	2	0	2	0	2	0	0	2	0	78.65
	2	2	0	1	1	2	0	0	1	1	1.72
	2	2	1	1	0	1	0	0	2	1	1.61
	2	1	1	2	0	1	1	0	2	0	1.49
	0	2	0	2	0	2	0	2	2	0	1.13
	2	2	0	2	0	1	1	0	1	1	0.93
	2	2	0	1	1	1	1	0	2	0	0.91
	2	1	1	1	1	2	0	0	2	0	0.85
	2	1	1	2	0	2	0	0	1	1	0.82
	2	2	1	2	1	1	0	0	1	0	0.80
											Sum 88.91
Intermediate 2											
(5.50,4.60)	2	2	0	2	0	2	0	0	2	0	69.01
	2	2	1	1	0	1	0	0	2	1	4.09
	2	2	0	1	1	2	0	0	1	1	2.54
	2	1	1	2	0	1	1	0	2	0	2.12
	2	2	0	0	0	2	0	2	2	2	1.76
	2	2	2	2	0	0	0	0	2	0	1.41
	2	2	0	2	0	1	1	0	1	1	1.28
	2	1	0	1	0	2	1	0	2	1	1.22
	2	2	1	2	1	1	0	0	1	0	1.19
	2	1	1	1	1	2	0	0	2	0	1.13
	2	2	0	1	1	1	1	0	2	0	0.94
	0	2	0	2	0	2	0	2	2	0	0.86
											Sum 87.55
Intermediate 3											
(5.00,4.30)	2	2	0	2	0	2	0	0	2	0	45.18
	2	2	1	1	0	1	0	0	2	1	13.00
	2	2	0	0	0	2	0	0	2	2	6.24
	2	2	2	2	0	0	0	0	2	0	4.64
	2	2	2	0	0	0	0	0	2	2	3.74
	2	2	0	1	1	2	0	0	1	1	2.79
	2	1	1	2	0	1	1	0	2	0	2.27
	2	2	2	0	0	2	0	0	2	0	1.71
	2	1	0	1	0	2	1	0	2	1	1.64
	2	2	1	2	1	1	0	0	1	0	1.53
	2	2	0	2	0	0	0	0	2	2	1.44
	2	2	0	2	0	1	1	0	1	1	1.02
	2	1	1	1	1	2	0	0	2	0	0.93
											Sum 86.13
TS											
(5.71,4.22)	2	2	1	1	0	1	0	0	2	1	20.00
	2	2	2	0	0	0	0	0	2	2	18.20
	2	2	0	2	0	2	0	0	2	0	17.70
	2	2	0	0	0	2	0	0	2	2	12.90
	2	2	2	2	0	0	0	0	2	0	10.35
	2	1	1	0	0	1	1	0	2	2	1.04
	2	1	1	2	0	1	1	0	2	0	0.91
	2	2	2	0	0	2	0	0	2	0	0.83
											Sum 81.93

^aThe two values in parentheses are the distance between the As_2 and Ga surface and the As-As distance, respectively. Units are in a.u.

TABLE IV. Occupation numbers of the CASSCF active orbitals of the As₂-(GaH₂)₄ cluster.

Sample	4a ₁	5a ₁	6a ₁	3a ₂	4a ₂	4b ₁	5b ₁	6b ₁	3b ₂	4b ₂
Stable dimer	1.95	1.93	0.06	1.94	0.05	1.95	0.08	0.05	1.94	0.05
Intermediate 1	1.94	1.92	0.11	1.88	0.09	1.89	0.10	0.06	1.91	0.11
Intermediate 2	1.95	1.88	0.20	1.78	0.13	1.80	0.13	0.04	1.87	0.21
Intermediate 3	1.96	1.86	0.49	1.46	0.15	1.51	0.15	0.04	1.85	0.53
TS	1.96	1.86	0.98	0.96	0.14	1.03	0.14	0.04	1.85	1.03

quintet contribution has an energetic advantage. The “cost” to prepare this valence excited state is paid by the formation of Ga-As covalent bonds. A similar favor was found in Ref. 1, where the triplet excited state ³Π [*T_e* = 3.68 eV (Ref. 40)] is more favorable than the quasibound singlet one ¹Π[~4.99 eV (Ref. 40)] for the covalently adsorbed GaCl. The ⁵Σ_g⁺ *R_e* of 5.16 a.u. is definitely comparable to 5.14 a.u. for the stable As dimer on the Ga surface (see Table II). The effective configuration of ⁵Σ_g⁺ is calculated to be

$$\sigma_g^{1.87} \pi_u^{2.02} \pi_g^{1.98} \sigma_u^{0.13} \quad (5)$$

and corresponds well to that of the stable dimer (4) except for the π_u occupation increase. Recall that Alstrin *et al.* pointed out the weakness of the dimer bond for the As₂/Si system.⁵³ The spectroscopic constants of ⁵Σ_g⁺ are in line with Ref. 53. As a whole, the main origin of the ¹A₁ stable dimer structure for As₂-(GaH₂)₄ could be the quintet-quintet combinations of ⁵Σ_g⁺ (= ⁵A₁) of As₂ and ⁵A₁ of (GaH₂)₄. Through nondynamical correlations, this combination should be dissolved toward the desorption limit.

Table V also contains the Mulliken overlap charges. The overlap charge could reflect the strength of the bond. Apparently, the overlap charges of Ga-As decrease when the desorption proceeds. In contrast, those of As-As increase. The changes of occupations in π_u and π_g consistently synchronize, describing the reproduction of the two π bonds during the desorption. Note again that the As-As overlap charge of 0.11 at the stable dimer compares with that of 0.07 for ⁵Σ_g⁺ As₂. The effective configuration and the As-As overlap charge at the TS are similar to those of the free *X* ¹Σ_g⁺ As₂ molecule. This also agrees with the “late-barrier”⁵² character of the TS.

At the TS, the active orbitals of 6a₁, 4b₁, 4b₂ and 3a₂ correspond, respectively, to DBa₁, DBb₁, DBb₂, and DBa₂ on the Ga-DB site. The reason why each occupation number is close to unity can probably be attributed to the neglect of the reconstruction for the Ga-DB site in the present model cluster. However, this may not be serious, since the Ga-Ga σ dimer bond on the surface is expected to be not so stable. The ground state of the Ga₂ molecule (or the free Ga dimer) is ³Π_u with the “formally single” bond given by $\sigma_g^1 \pi_u^1$ and its *D_e* is known to roughly be 1 eV.^{40,42,56,57} But this state cannot be directly referred. The usual σ bonded state for Ga₂ is ¹Σ_g⁺. Shim, Mandix, and Gingerich have theoretically estimated that the *T_e* for ³Π_u → ¹Σ_g⁺ is 0.36 eV and the *D_e* is only 0.78 eV.⁵⁷ Because of some distortions, the σ bond of the dimer on surface could be weaker than that of the free ¹Σ_g⁺ Ga₂. This leads to the dimer bond of Ga-DB being easy to break when some perturbations such as the interaction with As₂ take place. Therefore, even around the TS, the imposed fixation of (GaH₂)₄ geometries seems to be acceptable. Note that the all in-phase DBa₁ orbital has the smallest occupation at the TS. The largest repulsion with the π_{uz} electrons of the As₂ side should be responsible for it. Panas, Siegbahn, and Wahlgren also pointed out the similar effect due to the repulsion between O₂ π_{uz} and the Ni surface.³⁸

Finally, we will discuss the context of the As₂ adsorption onto the Ga-DB site. In the ALE and MBE methods, the active As carrier is considered to be mainly the As₄ molecule.⁵⁸ However, the As₂ molecule should coexist because of the thermochemical equilibrium:⁵⁹



Foxon and Joyce measured the sticking coefficients onto the Ga surface for both As₄ and As₂ and speculated that

TABLE V. Summary of Mulliken analyses: effective occupation numbers of the As₂ fragmental orbitals, overlap charges, and net charges. Occupations of As₂ orbitals are estimated by the Mulliken analyses on the active orbitals. The π_z components in a₁ and b₁ symmetries are obtained by summing up the triply divided 4p_z AO's and the polarization 4d_{zx} AO.

Sample	σ _g	π _u	π _g	σ _u	Ga-As ^a	As-As	q(As ₂)
Stable dimer	1.83	2.41	2.07	0.08	0.44	0.11	-0.49
Intermediate 1	1.82	2.53	2.02	0.02	0.36	0.12	-0.44
Intermediate 2	1.87	2.70	1.78	0.02	0.30	0.16	-0.37
Intermediate 3	1.92	3.12	1.16	0.03	0.15	0.42	-0.23
TS	1.96	3.58	0.41	0.11	-0.01	0.81	-0.06
Free As ₂	1.96	3.76	0.24	0.04		0.97	

^aValue for the single Ga-As bond.

the sticking mechanisms through adsorption are different from each other.^{60,61} The sticking coefficients are close to unity for As₂ but less than half for As₄. According to the detailed balance, the present study on As₂ desorption could be suggestive for As₂ adsorption as the reverse process. When the As₂ approaches the reactive Ga-DB site, the π bond should immediately start to break. Then, the As₂/Ga adsorption system reaches the TS of the "early-barrier" type.⁵² If the system passes through the "early" TS, the As₂ molecule is stuck onto the Ga surface by the attractive well which results from the Ga-As covalent-bond formation through significant configuration mixing. The experimental fact that the sticking coefficient of As₂ is nearly unity⁶⁰ could be consistent with the feature of presently calculated PES. Some trajectory calculations using this PES would be informative in understanding the behavior of incoming As₂ molecules.⁶² During the preparation for the present paper, Fukunishi and Nakatsuji have reported the SCF and the second-order Møller-Plesset perturbation (MP2) MO calculations on the As₂ adsorption process.⁶³ Unfortunately, their calculations were based on the single-configurational wave function. Note again that the multiconfigurational nature is essential to describe the interaction between the As₂ molecule and the Ga-DB site.

IV. SUMMARY

The As₂ desorption from the fourfold-hollow Ga-DB site was investigated by the *ab initio* CASSCF calculations on the As₂-(GaH₂)₄ cluster model. At the stable As dimer structure, the electronic configuration of the As₂ unit is revealed to be $\sigma_g^2 \pi_u^2 \pi_g^2$. In contrast, the desorbed As₂ molecule is triply bonded with the $\sigma_g^2 \pi_u^4$ configuration. Namely, the four π orbitals of As₂ are used to form the four Ga-As bonds by the double excitation. A remaining As-As σ bond is very weak. The calculations showed that the As₂ desorption process can be considered as just the chemical reaction of multiple bond conversions. Nondynamical correlations are essential to describe properly the desorption. The TS of As₂ desorption is found to be of the "late-barrier" type.

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