PHYSICAL REVIEW B

VOLUME 48, NUMBER 5

Rapid Communications

Rapid Communications are intended for the accelerated publication of important new results and are therefore given priority treatment both in the editorial office and in production. A Rapid Communication in Physical Review B should be no longer than four printed pages and must be accompanied by an abstract. Page proofs are sent to authors.

Influence of configurational atomic order on the relative stability of bcc and close-packed structures in Cu-based alloys

Mariàngels Jurado, Lluís Mañosa, and Antoni Planes

Departament d'Estructura i Constituents de la Maièria, Facultat de Física, Universitat de Barcelona, Diagonal, 647, 08028 Barcelona, Catalonia, Spain

(Received 22 April 1993)

We present a study of the influence of atomic order on the relative stability of the bcc and the 18R martensitic structures in a Cu_{2.96}Al_{0.92}Be_{0.12} crystal. Calorimetric measurements have shown that disorder increases the stability of the 18R phase, contrary to what happens in Cu-Zn-Al alloys for which it is the bcc phase that is stabilized by disordering the system. This different behavior has been explained in terms of a model recently reported. We have also proved that the entropy change at the martensitic transition is independent of the state of atomic order of the crystal, as predicted theoretically. Our results suggest that differences in the vibrational spectrum of the crystal due to different states of atomic order must be equal in the bcc and in the close-packed phases.

The stability of the bcc phases exhibited at high temperatures by some metals and alloys stems from the fact that they have an entropy larger than the entropy of the close-packed structures that would be the stable phases from purely energetic considerations.¹ Most of these metals and alloys undergo a structural transition towards a close-packed structure at a temperature M_s . This is the so-called martensitic transition which is first order, diffusionless, and is mainly accomplished by a homogeneous (110) $\langle 1\overline{10} \rangle$ shear.²

Phonon softening has been observed on approaching the transition: both the frequencies of the low-energy T_2 transverse [$\xi\xi 0$] phonons³ and the shear elastic constant $C' [=(C_{11}-C_{12})/2]$ (Ref. 4) decrease with temperature. An entropy change occurs at the transition, related to the change in the vibrational spectrum of the crystal; it is in this sense that these transitions have been acknowledged to be vibrational-entropy driven. Indeed, extensive research dealing with the role played by vibrational entropy in driving the martensitic transition is being undertaken from both experimental⁴⁻⁶ and theoretical⁷ points of view.

On the other hand, most of the alloys undergoing martensitic transitions exhibit ordered structures below a certain temperature T_c ($T_c > M_s$). There has been rapid progress towards understanding the thermodynamics of atomic order in metals and alloys, but there is only very weak experimental evidence for the importance of vibrational entropy in order-disorder transformations. For example, only very recently has this effect been estimated for Ni-Al alloys.⁸ Since it is established that the degree of atomic order modifies the relative stability of the bcc and the closepacked structures,⁹ an investigation of the influence of atomic order on the martensitic transition will provide valuable information about order-disorder processes and martensitic transitions, both topics of great interest nowadays.

In the present work we have studied a shape-memory Cu-Al-Be alloy close to a stoichiometric Cu₃Al composition. This alloy undergoes an order-disorder transition from a disordered (Im 3m) to an $L2_1$ ordered (Fm 3m)structure and a martensitic transformation from a bcc to an orthorhombic (18R) structure. It has been established that for this alloy, the martensitic transition is purely vibrational-entropy driven.⁶ The sample investigated is a single crystal $(Cu_{2.96}Al_{0.92}Be_{0.12})$ grown by a modified Bridgman method (0.689 g) with transition temperatures $T_c \approx 800$ (Ref. 10) and $M_s \approx 240$ K (Ref. 4) for the orderdisorder and martensitic transitions, respectively. We have established different degrees of atomic order in the crystal by means of the following heat treatments. First the sample is maintained at 973 K for 300 s, then it is air cooled down to a temperature T_q (characteristic cooling time about 300 s) and when this temperature is reached, it is guenched into a mixture of ice and water. This heat treatment is suitable to have the sample at T_q with its equilibrium ordering degree.¹¹ Just after the quench, the martensitic transition has been investigated using a highsensitivity microcalorimeter.¹² A second calorimetric run has been performed 23 h after the first run. Both sets of data are compared to those obtained after a reference

0163-1829/93/48(5)/3540(4)/\$06.00

48

3540

heat treatment consisting of an air cooling from 973 K down to room temperature, for which a maximum degree of atomic order is expected.

In Fig. 1 we present some typical curves obtained for the bcc $\rightarrow 18R$ transition just after a quench from different T_q temperatures (a) and after a 23 h roomtemperature aging (b). It is apparent from the curves that the martensitic transition takes place at different temperatures for different T_q . In addition the kinetics of the transition is also dependent on T_q : the jerky character of the transition depends on T_q and premonitory effects are detected above the bulk of the transformation. The magnitude of these phenomena is maximum for T_q around 820 K. After 23 h room-temperature aging there is a slight change in the transition temperatures, the transformation is less jerky and the permonitory effects are almost undetectable.

In order to quantify the change in the transition temperature with T_q we have adopted the temperature of the maximum of the thermal curve (T_M) as a signature for the equilibrium temperature between bcc and 18*R* phases. T_M is plotted in Fig. 2 as a function of T_q . After 23 h room-temperature aging, T_M is always slightly lower than just after the quench but it exhibits the same dependence with T_q , indicating that changes in T_M have a different origin than the jerky character of the transition together with the presence of premonitory effects. These



FIG. 1. Typical thermal curves obtained during the $bcc \rightarrow 18R$ transition of $Cu_{2.96}Al_{0.92}Be_{0.12}$ just after the quench (a) and after 23 h room-temperature aging (b) for different quenching temperatures T_q .



FIG. 2. Transition temperature (T_M) as a function of quenching temperature T_q for Cu_{2.96}Al_{0.92}Be_{0.12} (\bigcirc): Just after a quench; (\blacksquare): after 23 h room-temperature aging. The open circle corresponds to a standard heat treatment.

two last phenomena are associated to a large number of quenched-in vacancies that act as pinning centers for the motion of the interfaces. After 23 h room-temperature aging most of the vacancies are eliminated and as a consequence these effects have considerably decreased in magnitude.

It is worth mentioning that premonitory effects of tweed nature have been detected in a number of systems undergoing martensitic transitions¹³ and have been attributed to compositional random disorder of thermal origin.¹⁴ We believe that the premonitory effects observed here are also associated to random disorder, nevertheless optical microscopy observations have shown that these effects, detected calorimetrically, are related to the transition of domains with length scales comparable to those transforming at the bulk of the transformation.

Changes in T_M with T_q are due to different degrees of atomic order reached by the system. Recently a model has been proposed¹⁵ that accounts for changes in the martensitic transition temperature in terms of the degree of atomic order and the pair ordering energies of the different atomic elements. This model also predicts the entropy change at the martensitic transition to be independent of the degree of atomic order.

The dependence of the transition temperature with the long-range atomic order parameter s can be expressed as¹⁵

$$\delta M_s \simeq (1/\Delta S) K[s^2(T_q) - s_0^2], \qquad (1)$$

where s_0 and $s(T_q)$ are the equilibrium values of the long-range atomic order parameter at room temperature and T_q , respectively. ΔS is the entropy change at the martensitic transition. The parameter K is related to pair ordering energies as

3542

(2)

$$K = \frac{2}{3} W^{(1)} - V^{(2)} \; .$$

For our Cu-Al-Be alloy, the composition is close to stoichiometric Cu₃Al and $W^{(1)}$ is taken to be the nearest-neighbor ordering energy for Cu-Al pairs in the martensitic phase and $V^{(2)}$ to be the next-nearest-neighbor ordering energy for Cu-Al pairs in the bcc phase.

Using the pair ordering energies given in Refs. 9 and 11, $W^{(1)}=1250$ K and $V^{(2)}=825$ K, and taking $\Delta S=0.16k_B$ we obtain a maximum shift in the martensitic transformation temperature of 50 K.

A dependence of M_s with ordering has already been reported for Cu-Zn-Al alloys.¹⁶ It has been shown that for this system, the configurational atomic order tends to stabilize the bcc phase. In particular we have examined a Cu_{2.72}Zn_{0.64}Al_{0.64} single crystal under the same experimental conditions as for Cu_{2.96}Al_{0.92}Be_{0.12}. Results are shown in Fig. 3. They are in agreement with those reported previously.¹¹ In this case, the absence of detectable premonitory effects enables an accurate determination of M_s .

We have estimated the maximum shift predicted theoretically for $Cu_{2.72}Zn_{0.64}Al_{0.64}$ using the pair ordering energies given in Refs. 9 and 11 and $\Delta S = 0.16k_B$. For this alloy $W^{(1)}$ and $W^{(2)}$ are given by

$$W^{(1)} = (x_{Zn} W^{(1)}_{Cu-Zn} + x_{Al} W^{(1)}_{Cu-Al}) / (x_{Al} + x_{Zn}), \qquad (3)$$

$$V^{(2)} = (x_{Zn} V^{(2)}_{Cu-Zn} + x_{Al} V^{(2)}_{Cu-Al}) / (x_{Al} + x_{Zn})$$
(4)

with $V_{i-j}^{(2)}$ and $W_{i-j}^{(1)}$ the ordering energies for i-j pairs and x_i , the atomic fraction of the *i* component (i, j=Cu, Zn, Al). We have obtained a maximum shift in M_s of -68 K.

It is worth noting that the model predicts an increase



FIG. 3. Transition temperature (M_s) as a function of quenching temperature T_q for Cu_{2.72}Zn_{0.64}Al_{0.64}, measured just after a quench. The open circle corresponds to a standard heat treatment.

in the transition temperature for $Cu_{2.96}Al_{0.92}Be_{0.12}$ and a decrease for $Cu_{2.72}Zn_{0.64}Al_{0.64}$, in perfect agreement with experimental findings. The predicted shifts are larger than the measured ones; in fact, the model only gives an upper limit for the transition temperature shift owing to the fact that it cannot account for the partial reordering of the system that takes place during the quench, which is experimentally unavoidable because real quenches are performed at high but finite cooling rates.

As displayed in Fig. 2, T_M (or M_s) does not exhibit a monotonous dependence with T_q , the T_M (or M_s) versus T_q curve has a minimum for Cu_{2.96}Al_{0.92}Be_{0.12} and a maximum for Cu_{2.72}Zn_{0.64}Al_{0.64} at $T_q \approx 820$ K. These extrema are associated with the fact that the order-disorder transition is around this value for the two alloys and therefore maximum reordering is expected for quenches from around this temperature.

From the recorded thermal curves, we have computed the entropy change (ΔS) for the transition undergone after 23 h room-temperature aging. The reason for choosing the second cycle is that the effect of atomic order is separated from other spurious effects such as interface pinning at vacancies that may result in irreversible energy dissipation, leading to entropy production. The results obtained for ΔS are plotted in Fig. 4 as a function of T_a ; the constancy of ΔS is remarkable (scatter less than 3%), with an average value of $0.154 \pm 0.003 k_B$. It has been shown that for displacive transitions undergone by systems with a high elastic anisotropy, the equivalence between a shear in a (110) plane along the $\langle 1\overline{10} \rangle$ and along the $\langle \overline{1}10 \rangle$ directions, leads to an entropy change between the bcc and the close-packed phases independent of the degree of atomic order.¹⁵ This theoretical prediction is confirmed by present results in $Cu_{2.96}Al_{0.92}Be_{0.12}$.

Recent neutron-scattering measurements in bcc Fe₃Al (Ref. 17) have shown that only the soft T_2 transverse $[\xi\xi 0]$ mode is affected by the state of order in the crystal. A decrease of 0.50 THz in the zone boundary frequency from the disordered to the ordered (DO₃) structures has been measured. This value enables computation of the entropy difference between disordered and ordered structures to be around $0.4k_B$. On the other hand, the excess vibrational entropy between disordered and ordered ($L1_2$) structures has only very recently been measured for



FIG. 4. Entropy change computed for the martensitic transition undergone by $Cu_{2.96}Al_{0.92}Be_{0.12}$ after 23 h room-temperature aging.

3543

the close-packed fcc Ni₃Al to be $0.3\pm0.1k_B$.⁸ The excess vibrational entropy of the disordered phase is equal in the two alloys within experimental uncertainties. Even if these results correspond to different alloy systems, they suggest that the entropy difference between the bcc and the close-packed structures phases should not be dependent on the state of order, thus the constancy in ΔS found in Cu_{2.96}Al_{0.92}Be_{0.12} may have a very general character.

In conclusion, we have investigated the effect of atomic order on the martensitic transition of Cu-based shape memory alloys. The degree of atomic order modifies the relative stability between bcc and 18R phases; the change terms of the pair ordering energies of the elements. The entropy change at the transition does not depend on the state of order of the system: should an entropy difference between ordered and disordered states exist in the bcc phase, this difference is maintained in the close-packed phase.

in the equilibrium temperature can be accounted for in

Samples were kindly provided by M. Morin and S. Belkahla, INSA, Lyon (France). M.J. acknowledges financial support from DGU (Catalonia). This work has been supported by CiCyT, Project No. MAT92-884.

- ¹J. Friedel, J. Phys. (Paris) Lett. **35**, 59 (1974).
- ²A. G. Kachaturyan, *Theory of Structural Transformations in Solids* (Wiley, New York, 1983).
- ³S. Hoshino, G. Shirane, M. Suezawa, and T. Kajitani, Jpn. J. Appl. Phys. **14**, 1233 (1975); G. Guénin, S. Hautecler, R. Pynn, P. F. Gobin, and L. Delaey, Scr. Metall. **13**, 429 (1979); A. Heiming, W. Petry, and J. Trampenau, J. Phys. (France) IV **1**, C4-83 (1991).
- ⁴A. Planes, Ll. Mañosa, D. Ríos-Jara, and J. Ortín, Phys. Rev. B **45**, 7633 (1992).
- ⁵A. Heimig, W. Petry, J. Trapenau, M. Alba, C. Herzig, H. R. Schober, and G. Vogl, Phys. Rev. B **43**, 10 933 (1991).
- ⁶Ll. Mañosa, A. Planes, J. Ortín, and B. Martínez, Phys. Rev. B (to be published).
- ⁷J. Morris and R. J. Gooding, Phys. Rev. Lett. **65**, 1769 (1990); Phys. Rev. B **43**, 6057 (1991).
- ⁸L. Anthony, J. K. Okamoto, and B. Fultz, Phys. Rev. Lett. 70, 1128 (1993).

- ⁹M. Ahlers, Prog. Mat. Sci. **30**, 135 (1986).
- ¹⁰M. A. Jurado, Ll. Mañosa, and A. Planes (unpublished).
- ¹¹A. Planes, R. Romero, and M. Ahlers, Acta Metall. Mater. 38, 757 (1990).
- ¹²G. Guénin et al., in Proceedings ICOMAT'86 (The Japan Institute of Metals, Nara, 1986), p. 794.
- ¹³I. M. Robertson and C. M. Wayman, Philos. Mag. 48, 421 (1983); 48, 443 (1983); 48, 629 (1983); S. M. Shapiro, J. Z. Larese, Y. Noda, S. C. Moss, and L. E. Tanner, Phys. Rev. Lett. 57, 3199 (1986).
- ¹⁴S. Kartha, T. Castán, J. A. Krumhansl, and J. Sethna, Phys. Rev. Lett. 67, 3630 (1991).
- ¹⁵A. Planes, E. Vives, and T. Castán, Phys. Rev. B **44**, 6715 (1991).
- ¹⁶A. Planes, J. L. Macqueron, R. Rapacioli, and G. Guénin, Philos. Mag. A 61, 221 (1990).
- ¹⁷I. M. Robertson, J. Phys. Condens. Matter 3, 8181 (1991).