Electronic structure of LaBa₂Cu₂NbO₈ and related candidates for high- T_c superconductivity

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The nominally tetragonal, triple-perovskite compound LaBa₂Cu₂NbO₈ contains a pair of CuO₂ layers that are separated by an oxygen-free La layer as well as a BaO-NbO₂-BaO triple layer. The results of linear augmented-plane-wave band calculations for this (and a proposed isoelectronic Ti) compound exhibit electronic features near E_F (i.e., a pair of nearly degenerate, half-filled Cu $d_{x^2-y^2}$ -O $p_{x,y}$ σ antibonding subbands) that are characteristic of known cuprate superconductor parent compounds. This suggests that high- T_c superconductivity may be possible in appropriately doped LaBa₂Cu₂NbO₈-, LaBa₂Cu₂TaO₈-, or BaLa₂Cu₂TiO₈-type samples (or their chemical equivalents), although initial efforts have been unsuccessful thus far in the case of the group-V-type materials. PACS number(s): 74.10.+v, 71.25.Pi

Since the discovery of high-temperature superconductivity ($T_c \sim 30$ K) in the La-Ba-Cu-O system by Bednorz and Müller,¹ a wide variety of other cuprate superconductors has been synthesized, with T_c 's now extending above 100 K to the 110–125 K range.^{2,3} Although the crystal structures for the known cuprate superconductors differ in many details, they all share a common structural element, namely, the presence of one or more CuO₂ layers within a quasitetragonal unit cell. Within these electronically active CuO₂ layers, neighboring CuO₂ square complexes are joined by oxygen-sharing vertices along the [100] and [010] axes, thereby providing the strong Cu 3d - O 2p hybridization effects near E_F which are an important feature of the cuprate superconductors.

In general terms, the crystal structures for these cuprates consist of one or more perovskite-type cells stacked along the c axis. A systematic description of the various stacking constituents in terms of block layers has been provided by Tokura and Arima.⁴ In most cases, the crucial CuO₂ subunits in neighboring primitive cells are structurally and chemically isolated by semiconducting rocksalt-type (or fluorite-type) bilayers (trilayers), which also introduce a body-centering translation. The rocksalt bilayers can be formed either by the outermost faces of the perovskite subunits (as in La_2CuO_4) or by the insertion of extra monoxide layers (e.g., TlO, BiO). Since these rocksalt or fluorite buffer layers consist of oxides with relatively large cations, the obvious choices for these constituents (i.e., Ca, Sr, Ba, Y, La, Tl, Pb, Bi, and rare earths) have been thoroughly explored by materials scientists.

In searching for new classes of cuprate superconductors, it is necessary to explore alternative schemes for joining the CuO₂ layers of neighboring units cells while preserving their structural and electronic isolation. One interesting possibility is suggested by the Aurivillius phases,⁵ a series of tetragonal layer compounds (i.e., Bi-Ti-O, Bi-Nb-Ti-O, Bi-Ta-Ti-O, etc.) with Ti, Nb, and Ta constituents which have lattice parameters ($a \approx 3.84$ Å) and nearest-neighbor oxygen bond distances ($\sim a/2$) which are comparable to those of typical cuprate superconductors $(a \approx 3.80-3.85 \text{ Å})$. Since the *d* bands for these constituents fall 3.0-3.5 eV above the O 2*p* bands in related semiconducting perovskite compounds⁶ (i.e., SrTiO₃, KTaO₃), it is reasonable to expect that one might form new high- T_c candidate materials by interrupting the *c*-axis stacking of cuprate-type perovskite cells with structurally similar semiconducting perovskite layers. In contrast to the rocksalt or fluorite layers which are rotated by 45°, the present compounds would maintain the vertex-oxygen sharing feature in both the CuO₂ and semiconducting layers, thereby producing a simple-tetragonal structure.

Clearly, a basic problem with these proposed materials is that they must maintain an ordered layer structure despite the fact that the transition-metal constituents have similar ionic sizes. For example, recent studies' of the Y-Ba-Cu-W-O system reveal the formation of a three-dimensional superlattice with $YBa_2Cu_2WO_{9-\delta}$ composition and a fcc structure rather than the desired layer configuration. However, in studies involving the substitution of Ta for Cu (at the chain site) in the La analogue of the 1:2:3 phase, LaBa₂Cu_{3-x}Ta_xO_{7+δ}, Murayama et al.⁸ have succeeded in forming the ordered compound LaBa₂Cu₂TaO₈ in which the required layering of CuO_2 and TaO_2 planes is achieved, with no measurable intermixing. In particular, detailed structural studies have shown that TaO₂ replaces the chain CuO complex between the BaO layers of the 1:2:3 structure. Semiconducting properties and a simple-tetragonal structure have been determined from resistivity and powder x-raydiffraction measurements.8

Subsequent neutron-diffraction studies⁹ on a $LaBa_2Cu_2TaO_8$ as well as an analogous $LaBa_2Cu_2NbO_8$ sample have confirmed the CuO₂ and TaO₂ or NbO₂ layer ordering in both compounds. Efforts to observe superconductivity in appropriately doped samples (Ca,Sr \rightarrow La, or Ti \rightarrow Nb,Ta) have been unsuccessful thus far.⁹ A recent high-resolution neutron-diffraction study¹⁰ on $LaBa_2Cu_2NbO_8$ has revealed the presence of rotational distortions of the NbO₆ octahedra which are similar to those observed¹¹ below the 110 K cubic-to-tetragonal

The purpose of the present investigation is to evaluate the potential of LaBa₂Cu₂NbO₈ and a proposed isoelectronic prototype compound, BaLa₂Cu₂TiO₈, as parent compounds for possible new high- T_c cuprate superconductors. With this objective, self-consistent bandstructure calculations have been carried out for LaBa₂Cu₂NbO₈ and BaLa₂Cu₂TiO₈ in the local-density approximation (LDA) with the use of a scalar-relativistic version of the linear augmented-plane-wave (LAPW) method.¹² These calculations neglect the rotational distortions that have been reported by Rev et al.¹⁰ since these distortions are regarded as incidental to the occurrence of superconductivity. The calculations assume the undistorted simple-tetragonal LaBa₂Cu₂NbO₈ primitive unit cell that is shown in Fig. 1. The observed⁹ lattice parameters and atom position parameters for this phase are summarized in Table I.

According to Fig. 1, this simple-tetragonal structure consists of a triple stacking of perovskite-type cells along the c axis with an oxygen vacancy at the origin in the central La plane. The neighboring CuO₂ and BaO planes produce pyramidal coordination at the Cu sites, with an apical-oxygen bond distance (~ 2.27 Å) which is about 14% greater than the corresponding planar (~ 2.00 Å) values. The octahedrally coordinated Nb site is more symmetric, with apical- and planar-oxygen bond distances of 1.95 and 1.99 Å, respectively. The measured c/a ratio is 3.02. Assuming that Nb is 5+, a simple valence analysis suggests that the formal valence of Cu is 2+, the standard value for cuprate superconductor parent compounds.

A promising isoelectronic Ti compound would be obtained if the La and Ba constituents were interchanged and Ti is substituted for Nb. Because of differences in the ionic radii for La^{3+} (1.15 Å), Ba^{2+} (1.35 Å), Ti^{4+} (0.68 Å), and Nb⁵⁺ (0.70 Å), some modifications to the La-Ba-Cu-Nb-O structural parameters in Table I are expect-



FIG. 1. Simple-tetragonal primitive unit cell for $LaBa_2Cu_2NbO_8$. Oxygens in the Cu, Ba, and Nb layers are denoted by O(1), O(2), and O(3), respectively.

TABLE I. Atom-position parameters for tetragonal LaBa₂Cu₂NbO₈ (space group P4/mmm), as determined from neutron-diffraction measurements by Greaves and Slater (Ref. 9), where a = 3.970 Å and c = 11.989 Å, respectively.

Atom	Site	x /a	v/a	z/c
La	lc	1/2	1/2	0
Ba	$\frac{10}{2h}$	1/2	1/2	0.3082
Cu	2g	0	0	0.1479
Nb	1b	0	0	1/2
O(1)	4 <i>i</i>	1/2	0	0.1309
O(2)	2g	0	0	0.3375
O(3)	2e	1/2	0	1/2

ed for this proposed $BaLa_2Cu_2TiO_8$ phase. However, these variations have been neglected in the present study since they are expected to introduce only quantitative changes in the final electronic-structure results.

The present LAPW calculations, which impose no shape approximations on either the charge density or the potential, treat exchange-correlation effects via the Wigner interpolation formula.¹³ They include an LAPW basis involving plane waves with an 11.5-Ry cutoff (\sim 840 LAPW's) and spherical-harmonic terms through l=8(l=6) for the metal (oxygen) constituents. The charge density and potential are expanded using ~ 9000 plane waves (55 Ry) in the interstitial region and by means of lattice-harmonic expansions with $l_{max} = 6$ ($l_{max} = 4$) within the metal-atom (oxygen) muffin-tin spheres. A ten-point k sample has been used for Brillouin-zone integrations. The electrons in the outermost shells of the atomic constituents (e.g., Ba $5p^{6}6s^{2}$, La $5p^{6}5d6s^{2}$, Cu $3d^{10}4s$, Nb $4d^{4}5s$, Ti $3d^{3}4s$, and O $2s^{2}2p^{4}$) are treated as valence electrons in these calculations. A rigid-core approximation is applied to treat the remaining inner-shell electrons.

The LAPW energy-band results for BaLa₂Cu₂TiO₈ and LaBa₂Cu₂NbO₈ are plotted along symmetry lines in the basal plane ($\Gamma XM\Gamma$) of the tetragonal Brillouin zone and along one *c*-axis direction (ΓZ) in Fig. 2. The nearly filled 34-band valence manifold evolves from the 10 Cu(3d) and the 24 O(2p) states. In both materials, these bands are occupied by a total of 66 valence electrons, leaving a pair of half-filled subbands. These valence bands are nearly two dimensional, exhibiting minimal (~0.2 eV) dispersion along *c*-axis directions. As expected, the Ti(3d) and Nb(4d) bands lie above E_F in each compound. The lowest portions of these bands are labeled with triangle symbols in Fig. 2. In both materials, the lowest unoccupied Ti or Nb *d*-band state at Γ has d_{xy} orbital symmetry.

For added clarity, the important bands within the valence-band manifold have been labeled with square symbols. These identify the σ -type bonding, nonbonding, and antibonding (σ^*) combinations of Cu $d_{x^2-y^2}$ and O $p_{x,y}$ orbitals which are a distinctive characteristic of all cuprate superconductors. In particular, the upper pair of σ^* bands at E_F , which have their maxima at M, contains exactly two electrons in perfectly stoichiometric samples; thus, each is half filled. The small splittings between

BaLa₂Cu₂TiO₈ LaBa2Cu2NbO8 2 0 EF ENERGY (eV) -2 -6 Х Γ Μ Г Ζ Г Х М ΓΖ

FIG. 2. LAPW energy-band results for BaLa₂Cu₂TiO₈ and LaBa₂Cu₂NbO₈ along symmetry lines in the tetragonal Brillouin zone. The square (triangle) symbols identify band states that have at least 30% Cu $d_{x^2-y^2}$ -O(1) $p_{x,y}$ (Ti 3d or Nb 4d) orbital weight within the corresponding muffin-tin spheres.

these nearly degenerate σ^* subbands reflect interactions between neighboring CuO₂ planes. These splittings vary from about 0.05 eV at *M* to ~0.10 eV near E_F .

In typical parent compounds such as La_2CuO_4 , the half-filled σ^* band produces an antiferromagnet with insulating properties. Metallic behavior and superconductivity are observed when the band filling is varied so that extra carriers (holes or possibly electrons) are added to the CuO₂ planes. Efforts to introduce holes into La-BaCu-Nb-O and La-Ba-Cu-Ta-O samples via chemical substitution (i.e., Ca,Sr \rightarrow La, or Ti \rightarrow Nb,Ta) have been unsuccessful thus far,⁹ producing impurity phases. The present results suggest that further attempts to dope these compounds should be pursued.

A comprehensive overview of the $BaLa_2Cu_2TiO_8$ electronic properties is provided by the density-of-states (DOS) results in Fig. 3. These have been calculated using tetrahedral interpolation based on LAPW results at 42

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uniformly distributed **k** points in the $\frac{1}{16}$ irreducible Brillouin-zone wedge. As expected, the Ba, La, and Ti constituents are electronically and chemically inactive, contributing minimal weight to the projected DOS in the valence-band energy range. The same is true for the oxygens in the Ba and Ti layers [i.e., O(2) and O(3), respectively]. The main contribution to the DOS near E_F originates from the CuO₂ planes. Above E_F , the La(5d) and Ti(3d) components become increasingly important.

From a band-structure point of view, it is clear that the prototype compounds BaLa₂Cu₂TiO₈ and LaBa₂Cu₂NbO₈ both exhibit the characteristic band properties of a cuprate superconductor parent compound. However, these properties probably should be considered as a necessary but not sufficient requirement for the occurrence of cuprate superconductivity. For example, it has been shown¹⁴ that the halo-oxocuprate compounds $Ca_2CuO_2X_2$ (X = Cl,Br) exhibit the requisite band features, though superconductivity has not yet been observed in these materials or their $Sr_2CuO_2X_2$ counterparts.

The initial failure⁹ to form doped La-Ba-Cu-Nb-O and La-Ba-Cu-Ta-O samples which are homogeneous and superconducting could reflect the fact that these materials may be especially sensitive to preparation conditions. As in the case of the $La_{2-x}Sr_xCaCu_2O_6$ phase,¹⁵ one can anticipate that the successful synthesis of appropriately doped BaLa₂Cu₂TiO₈- and LaBa₂Cu₂NbO₈-type samples may require carefully selected preparation and annealing conditions. The details of the annealing process may be particularly important for these compounds in order to minimize the intermixing of constituents within the CuO₂ and TiO₂ or NbO₂ planes. Undoubtedly, such intermixing would be detrimental to the occurrence of superconductivity in these materials.

Because of the required layer configuration of these proposed high- T_c candidate materials, these compounds may be ideally suited for thin-film growth using molecular-beam-epitaxy procedures. Recent studies of high- T_c superconductors have shown that two-dimensional growth occurs in unit-cell-sized layers when all the constituents are supplied simultaneously.^{16,17} However, layer-by-layer growth can also be achieved by successively controlling the supply of the appropriate atomic constituents for each monolayer.¹⁷ Such procedures might be important for the present compounds if standard bulk-synthesis techniques are unsuccessful.

It should be emphasized that the constituents of the prototype compounds $BaLa_2Cu_2TiO_8$ and

LaBa₂Cu₂NbO₈ are to be regarded as illustrative rather than definitive. Clearly, it is important to investigate alternative combinations of trivalent (La,Y, rare earths), divalent (Ca, Sr, Ba), transition-element (Ti, Zr, Hf, V, Nb, Ta) constituents and dopants in the search for possible new superconducting phases. Analogous phases containing group-VI constituents (i.e., Cr, Mo, W) may also be possible high- T_c candidates, including materials of the type CaBa₂Cu₂WO₈, etc. Here, the expected lowering of the group-VI *d*-band energies could lead to partial *d*-band occupancy. This could produce self-doping effects analogous to those calculated¹⁸ for the Ca-Sr-Bi-Cu-O system.

Similarly, these triple-perovskite $BaLa_2Cu_2TiO_8$ and $LaBa_2Cu_2NbO_8$ phases should be regarded only as structural prototypes. It may be possible to form materials with fewer or greater numbers of CuO_2 layers. Using the Ba-La-Cu-Ti-O system as an example, one can regard the BaLa_2Cu_2TiO_8 compound as the n=2 member of the $La_2Ba_{n-1}Cu_nTiO_{2n+4}$ homologous series. The corresponding n=1 phase, La_2CuTiO_6 , is simply a two-layer perovskite. Properly doped, it should also possess electronic properties that are favorable for high- T_c superconductivity.

To summarize, the results of band calculations for two prototype cuprate compounds $BaLa_2Cu_2TiO_8$ and $LaBa_2Cu_2NbO_8$ show that the Ti and Nb constituents are 4+ and 5+, respectively. For the observed layer geometry of the Nb compound, both phases exhibit LDA band properties which are similar to those of typical cuprate superconductor parent compounds. These results suggest that appropriately doped samples of these (or chemically equivalent) phases should be explored as possible candidates for high- T_c superconductivity.

Note added in proof. Bulk superconductivity ($T_c \sim 28$ K) has been discovered recently by Cava et al.¹⁹ in cuprate compounds, which are similar to those discussed here in that they contain a NbO₂ or TaO₂ layer to separate the pair of CuO₂ planes in each primitive with overall composition cell. These phases, $Nd_{2-x}Ce_xSr_2Cu_2NbO_{10}$ or $Nd_{2-x}Ce_xSr_2Cu_2TaO_{10}$, are closely related to the simple-tetragonal structure that is shown in Fig. 1. The main difference is that the La monolayer in Fig. 1 is replaced by a fluorite-type $Nd_{2-x}Ce_{x}O_{2}$ trilayer. This introduces a body-centering translation and changes the space-group symmetry from P4/mmm to I4/mmm.

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