Brief Reports

Brief Reports are accounts of completed research which, while meeting the usual Physical Review standards of scientific quality, do not warrant regular articles. A Brief Report may be no longer than four printed pages and must be accompanied by an abstract. The same publication schedule as for regular articles is followed, and page proofs are sent to authors.

Optical spectra of U²⁺, U³⁺, and U⁴⁺ ions in calcium fluoride crystals

W. A. Hargreaves

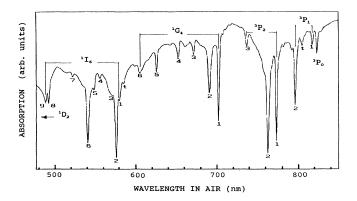
1301 Poitras Drive, Vero Beach, Florida 32963 (Received 8 January 1991; revised manuscript received 19 March 1991)

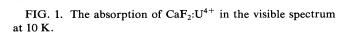
We report on optical spectra of single-valence U^{2+}, U^{3+}, U^{4+} ions in CaF_2 crystals at 10 K. The report supplements two previous papers by the author.

The physical and chemical properties of uranium ions in CaF_2 crystals have been discussed in two previously published papers of the present author.^{1,2} A review of those papers indicates that no significant changes are required. However, the desirability for publication of more detailed optical transmission characteristics of the various crystal materials is apparent. This Brief Report is made to remove that deficiency (see Figs. 1–8). It should be noted that the optical-absorption spectra presented are of almost valence-pure species of each uranium ion, U^{2+} , U^{3+} , and U^{4+} . The addition of these spectral data com-

pletes the presentation of the papers, 1,2 may assist in the resolution of an old controversy, 3 and clarifies the status of many published papers involving particularly the U^{4+} ion. $^{4-10}$

Interest in the $\text{CaF}_2: \text{U}^{3+}$ laser has been inactive for a long time, mainly because cryogenic cooling is required for efficient operation, but it is an excellent system with very low pumping threshold. It is a likely candidate for renewed interest, because of improved cryogenic technology, and new interest in the 2.5 to 2.6 μ m laser wavelengths.





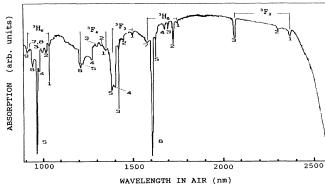


FIG. 2. The absorption of $\text{CaF}_2{:}\text{U}^{4+}$ in the infrared spectrum at 10 K.

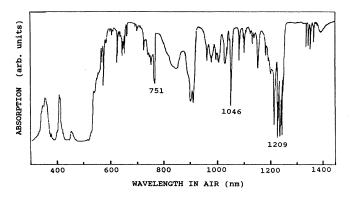


FIG. 3. The absorption of $\text{CaF}_2{:}\text{U}^{3+}$ in the visible spectrum at 10 K.

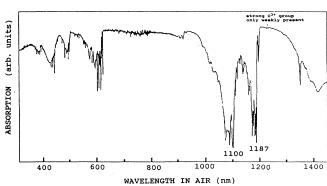


FIG. 4. The absorption of $\text{CaF}_2{:}U^{2+}$ in the visible spectrum at 10 K.

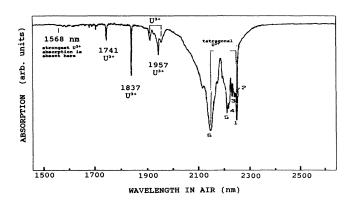


FIG. 5. The absorption of $CaF_2:U^{3+}$ in the infrared spectrum at 10 K.

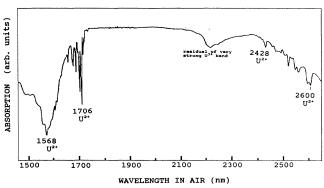


FIG. 6. The absorption of $\text{CaF}_2 \cdot \text{U}^{2+}$ in the infrared spectrum at 10 K.

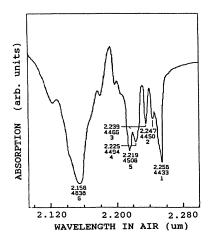


FIG. 7. The absorption of the six tetragonally sited $^4I_{11/2}$ levels of CaF₂:U³⁺ at 10 K.

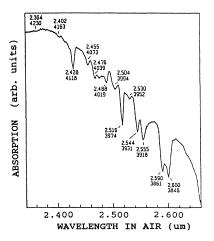


FIG. 8. The absorption of the 5I_5 levels of $CaF_2:U^{2+}$ at 10 K. These levels are diagrammed in Fig. 3 of Ref. 2.

- ¹W. A. Hargreaves, Phys. Rev. **156**, 331 (1967).
- ²W. A. Hargreaves, Phys. Rev. B 2, 2273 (1970).
- ³R. M. McGlaughlin, U. Abed, J. G. Conway, N. Edelstein, and E. H. Huffam, J. Chem. Phys. **53**, 2031 (1970).
- ⁴P. F. McDonald, E. L. Wilkinson, and R. A. Jensen, J. Phys. Chem. Solids **28**, 1629 (1967).
- ⁵H. C. Meyer, P. F. McDonald, J. D. Stettler, and P. L. Donoho, Phys. Lett. 24A, 569 (1967).
- ⁶G. C. Wetsel, Jr., J. Appl. Phys. **39**, 692 (1968).
- ⁷G. C. Wetsel, Jr. and E. L. Kitts, Jr., Phys. Rev. Lett. **18**, 841 (1967).
- 8C. M. Bowden, H. C. Meyer, P. F. McDonald, and J. D. Stettler, J. Phys. Chem. Solids 30, 1535 (1969).
- ⁹C. M. Bowden, H. C. Meyer, and P. F. McDonald, Phys. Rev. Lett. 22, 224 (1969).
- ¹⁰P. F. McDonald, Phys. Rev. 177, 447 (1969).