Polaronic conduction in *n*-type BaTiO₃ doped with La_2O_3 or Gd_2O_3

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(Received 30 July 1990; revised manuscript received 26 November 1990)

The dielectric properties and conductivities (σ) of BaTiO₃ doped with 0.03 mol % of La₂O₃ or Gd₂O₃ were investigated in the temperature range of 10–300 K. A speculation over the experimental results, such as the temperature independence of conductivities below 30 K, linear Arrhenius relations between $\sigma T^{3/2}$ and 1/T with an activation energy of 0.070 ± 0.002 eV above 50 K, and relaxation absorptions in the loss tangents with an activation energy of 0.068 ± 0.002 eV leads to the conclusion that the dominant conduction in *n*-type BaTiO₃ is due to hopping motions of nonadiabatic small polarons. This conclusion is confirmed theoretically. Based upon polaron energies obtained experimentally, several parameters relevant to a nonadiabatic small polaron in this material are estimated.

The nature of electrical conduction in semiconducting polycrystalline n-type BaTiO₃ is still a subject of discussion as to whether the charge carrier is a small polaron or a conduction electron, in spite of its well-known characteristic, the PTC effect, the function of which is used in many practical applications. Most analyses as to the conduction in this material pertain to charge-carrier mobilities around the Curie point.¹⁻⁶ Although optical and thermopower results on this material,^{7,8} suggest a high possibility of the conduction due to a hopping process of small polarons, there is no direct evidence, unfortunately. If such conduction takes place, an electron moves by hopping from one localized state to another and, consequently, an activation energy is required for each step because the localized states have energies extended over a certain range. Then, we can expect some dielectric relaxation behavior due to these motions in alternating fields. If we can observe such a relaxation experimentally, it would be direct evidence for the conduction due to a hopping process of small polarons. In fact, the dielectric relaxation related to the hopping motions of polarons is confirmed experimentally in WO₃ (Ref. 9) and Sm₂O₃-P₂O₅ glass.^{10, 11} From this point of view, we have carried out dielectric measurements of BaTiO₃ doped with La₂O₃ or Gd₂O₃, along with dc-conductivity measurements in the temperature range of ~ 10 to 300 K.

From x-ray measurements on related perovskite crystals such as $Sr_{1-x}La_xTiO_3$ and $Pb_{1-x}La_xTiO_3$ (Refs. 12 and 13) and chemical analysis,⁶ most of the La³⁺ ions are expected to substitute only for Ba²⁺ ions. Then, the Ladoped perovskite BaTiO₃:La corresponds to the chemical formula Ba_{1-y}La³⁺_y(Ti⁴⁺_{1-y}Ti³⁺_y)O₃. The ionic radius of Gd³⁺ compared with that of La³⁺ suggests similar substitutions in BaTiO₃ doped with Gd₂O₃.¹⁴

The BaTiO₃ ceramic specimens doped with La_2O_3 or Gd_2O_3 (0.03 mol %) were prepared by the conventional solid state synthesis technique.¹⁵ Powders of BaCO₃, La_2O_3 , Gd_2O_3 (Johnson Matthey, 5N grade) and TiO₂ (Aldrich Chemical, 4N grade) were used. Powder x-ray

measurements (Rigakudenki RAD-III B) indicate no other phases except $BaTiO_3$, being independent of dopants. Flat surfaces of the specimens were coated with In-Ga alloy in 7:3 ratio by a rubbing technique for the electrode which gives a good ohmic contact. YHP 4272A and 4275A LRC meters were used for measurements of capacitances, while Keithley 619 Resistance Bridge or Advantest TR 6871 with a high-ohm unit TR 68704 were used for dc-conductivity measurements. The copperconstantan thermocouple precalibrated at 4.2, 77, and 273 K was employed for the temperature measurements.

As is well known, there are three phase transition points in BaTiO₃; the first one is from the rhombohedral structure to monoclinic at ~ 200 K, the second one from monoclinic to tetragonal at ~ 300 K, and the third one from tetragonal to cubic at ~ 400 K associated with the ferroelectric-paraelectric transition (Curie point) which induces the PTC effect. In our specimens, there appear the anomalies due to the phase transition at ~ 200 K in both of dc-conductivities and dielectric properties. In the temperature range below 30 K, the conductivities contain nearly temperature-independent regions, which are predicted theoretically by Austin and Mott¹⁶ for the small polaron conduction. On the other hand, the conductivities start to increase as the temperature rises above 30 K. Ihrig⁵ suggests the electric conduction due to a hopping process of nonadiabatic small polarons between Ti sites in n-type BaTiO₃ in the temperature region where the PTC effect appears. If the conduction like this takes place also at low temperatures, the polaron theories¹⁶⁻¹⁸ characterize the following temperature dependence of the conductivity:

$$\sigma = T^{-3/2} e n_0 \exp[-(W_H + W_D/2 + W_0/2)/k_B T], \quad (1)$$

where W_H is the hopping energy of small polarons, W_D is the disorder energy, $n_0 \exp(-W_0/2k_BT)$ represents the density of Ti³⁺, in the thermal equilibrium, nonadjacent to La³⁺ or Gd³⁺ substituted for Ba²⁺ with the concentration of n_0 , and we denote the potential energy of Ti³⁺

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at the normal site compared to that of Ti³⁺ adjacent to La³⁺ or Gd³⁺ as W_0 . Figure 1 demonstrates the linear relation between $\ln(\sigma T^{3/2})$ and 1/T in the temperature region of ~40 to ~200 K. Such a temperature dependence is likely to suggest conduction due to a hopping process of single polarons at low temperatures. The results in Fig. 1 yield a value, $W_H + W_D/2 + W_0/2 = 0.070 \pm 0.002$ eV, which looks independent of dopants within the experimental errors.

Figure 2 demonstrates frequency dependencies of the absorption peaks in the loss tangent at low temperatures below 100 K. These results have the following remarkable features. The peak temperature, T_m , is found to move higher as the applied frequency increases. This fact suggests the absorption peaks due to dielectric relaxation processes. In most cases, a single relaxation time usually does not describe the relaxation process; rather, a distribution of relaxation times is required. The theoretical description of physical models exhibiting a distribution of relaxation times is often very complex. It is sufficient,



however, to quote the results of a simple model^{19,20} in which the barrier heights involved for the polaron hopping are equally distributed between W_H and $W_H + W_D$. This model is also employed in Eq. (1). Then, the shift of the peak temperature, T_m , with increasing applied angular frequency, ω , has a form

$$\omega \propto \exp\left[-(W_H + W_D/2)/k_B T_m\right]. \tag{2}$$

There are the linear relationships between T_m and $\ln\omega$ and they yield a value, $W_H + W_D/2 = 0.068 \pm 0.001$ eV. From the difference between the activation energy obtained in the conductivities and that in the dielectric loss angle, we have a value, $W_0/2 \simeq 2 \times 10^{-3}$ eV. Our experiments cannot separate $(W_H + W_D/2)$ into individuals, but, in practice, the value of $W_D/2$ is smaller than the experimental error in the determination of W_H as Dominik et al. suggested²⁰ and thus this is neglected. Then, $W_H \simeq 0.068$ eV. It is of great interest that the hopping energy of nonadiabatic small polarons in reduced barium titanate (0.074 eV) (Ref. 21) is very close in value to ours.

Figure 2 shows an increase of the loss angle value at maximum, $(\tan \delta)_{max}$, with increasing frequency, i.e., the



FIG. 1. The Arrhenius relations between $\sigma T^{3/2}$ and 1/T of BaTiO₃ doped with 0.03 mol % La₂O₃ (solid circles) and 0.03 mol % Gd₂O₃ (open circles).

FIG. 2. The frequency dependencies of absorption peaks in loss tangent, $\tan \delta$, for the specimen doped with 0.03 mol % La_2O_3 (a) and that doped with 0.03 mol % Gd_2O_3 (b). Open circles represent the results at the applied angular frequency 10 kHz; squares, 40 kHz; and solid circles, 100 kHz.

increase in T_m . These behaviors are very similar to the results on TiO_{2-x} (Ref. 20) and NiO.²² This effect is not predictable from the Debye theory if a constant number of oscillators is assumed. Alternatively, an activation energy was introduced by Dominik *et al.*²⁰ which can explain the thermal activation of the number of the oscillating dipoles. In our case, the tanð value at maximum, $(\tan\delta)_{\max}$, is proportional to the polaron concentration, $n_0 \exp(-W_0/2k_BT)$. The linear Arrhenius relations of $(1.6\pm0.1)\times10^{-3}$ eV for the activation energy, $(W_0/2)$, which is in good agreement with the value obtained above, i.e., $W_0/2 \simeq 2.0 \times 10^{-3}$ eV.

Although the speculation over the experimental results suggests a high possibility of the polaron hopping, a theoretical reconfirmation is required. Referring to the discussion of Ihrig *et al.*⁶ that the conduction takes place by hopping of small polarons between Ti sites, we have the following conditions for the electron transfer integral between neighboring Ti ions in $\langle 110 \rangle$, $J_{\text{Ti-Ti}}$, from the nonadiabatic polaron theories of Holstein¹⁷ and Emin:²³

$$J_{\mathrm{Ti-Ti}} < 4W_H \tag{3}$$

and

$$J_{\text{Ti-Ti}} < (W_H k_B T / \pi)^{1/4} (\hbar \omega_{0L})^{1/2} .$$
(4)

There is another theoretical relation between the drift mobility and the electron transfer integral at sufficiently high temperatures subject to the "classical limit," i.e., $\hbar\omega_{0L}/k_BT \ll 1$:^{17,24}

$$\mu_{D} = \left[e \left(\sqrt{2}a_{0} \right)^{2} J_{\text{Ti-Ti}}^{2} \pi^{1/2} / 2\hbar W_{H}^{1/2} \right] \\ \times (k_{B}T)^{-3/2} \exp\left[- \left(W_{H} + W_{D} / 2 \right) / k_{B}T \right], \quad (5)$$

where a_0 is the cubic lattice constant (0.4007 nm) (Ref. 25) and ω_{0L} is the frequency of the longitudinal optical mode for which we employed the experimental value,²⁶ i.e., $\omega_{0L} = 7 \times 10^{13} \text{ s}^{-1}$. The "classical limit" indicates that Eq. (5) holds at temperatures $T \gg \hbar \omega_{0L} / k_B = 534 \text{ K}$. The drift mobilities and hall mobilities in BaTiO₃ were obtained experimentally.^{1-4,27} These mobilities and the hopping energy ($W_H = 0.068 \text{ eV}$) allow us to estimate the electron transfer integral. The mobilities, $\mu_D = 0.10 \text{ cm}^2 \text{ V}^{-2} \text{ s}^{-1}$ at 1050 K, 0.20 cm² V⁻¹ \text{ s}^{-1} at 1000 K, and 0.5 cm² V⁻¹ \text{ s}^{-1} at 400 K yield $J_{\text{Ti-Ti}} = 0.019$, 0.026, and

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0.037 eV at each temperature, respectively. Then, the right-hand side in Eq. (4) is 0.026 eV at 1050 K, 0.025 eV at 1000 K, and 0.020 eV at 400 K. Although the inequality in Eq. (3) holds at every temperature because the right hand side =0.27 eV, the inequality in Eq. (4) holds only at 1050 K and the lowest temperature subject to the "classical limit" looks to be about 1000 K in *n*-type Ba-TiO₃. Thus, Eq. (4) cannot hold at tempertures lower than 1000 K because of the "classical limit" employed. In view of the above, it is likely that the major carriers are nonadiabatic small polarons and our experimental results present the direct evidence for this conduction.

In order to understand the real feature of a nonadiabatic small polaron in n-type BaTiO₃, it is of interest to estimate several important physical parameters relevant to this. From the theoretical representation¹⁶ for the rigidlattice mass, m^* , corresponding to (110) direction connects neighboring Ti ions, $m^* = \hbar^2 / \hbar^2$ which $2J_{\text{Ti-Ti}}(\sqrt{2}a_0)^2$, along with $J_{\text{Ti-Ti}} = 0.019$ eV at 1050 K, we have a value, $m^* \simeq 6m_0$, which is nearly equal to the magnitude obtained by Ihrig⁵ ($m^* \simeq 3.5 \sim 5.4 m_0$), where m_0 is the electron rest mass. The theoretical expression for the relation between the polaron binding energy, W_P , and the coupling constant, α ,^{28,29} $W_P = 0.1 \alpha^2 \hbar \omega_{0L}$, yields that the coupling constant has a value of 5.4, being quite reasonable in comparison with NiO ($\alpha \simeq 6.3$) (Ref. 8) and some halides,³⁰ where we have used the approximation, $W_P \simeq 2W_H$, which holds in the case of nonadiabatic small polarons.¹⁶ There is a relationship between the polaron effective mass, m_P , and the coupling constant,¹⁶ i.e., $m_P = 0.02m^* \alpha^4$. Then, we have obtained a ratio of $(m_P/m_0) \simeq 105$, which is compatible with the ratio in TiO₂ (100~150),³¹⁻³³ and also with that in *p*-type NiO $(150).^{34}$

ACKNOWLEDGMENTS

This project was supported, in part, by a Grant-in-Aid for Science Research (No. 62550517) from the Ministry of Education, Japan, and by the foundation from Sumitomo Metal Industries Ltd. The authors are very grateful to Dr. S. Kawado (Sony Research Laboratory, Yokohama) for the x-ray measurements and also to Mr. K. Akashi, Mr. K. Yamauchi, and Mr. H. Dodoro for their assistance to this project.

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