Two-photon spectroscopy in GdAlO₃

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(Received 27 June 1989)

Two-photon absorption spectroscopy is used to study $f \rightarrow f$ transitions of the Gd³⁺ ion in pure GdAlO₃ crystals. This technique allows us to observe the electronic transitions far in the ultraviolet without background as opposed to conventional spectroscopy. We observed several transitions that violated the angular-momentum selection rules $\Delta J \leq 2$.

Two-photon spectroscopy has been shown to be useful in obtaining complementary information to that obtained from the conventional one-photon spectroscopy due to the difference in the selection rules.¹ With the development of tunable dve lasers, two-photon spectroscopy became a powerful tool for the investigation of sharp $4f \rightarrow 4f$ transitions of rare-earth ions in solids in the ultraviolet spectra.² In the present paper, the observation of direct twophoton $f \rightarrow f$ transitions of the Gd³⁺ ion in pure GdAlO₃ crystals is reported. To our knowledge this is the first observation of sharp lines in two-photon spectra in pure crystals. This ion has a very special level structure in relation to all other rare-earth ions because its first excited state is 32000 cm⁻¹ above the ground state.³ Two-photon transitions were induced between the ${}^{8}S_{7/2}$ ground state and the ^{6}P and ^{6}I excited states of the f configuration. As a result of the shielding of 4f levels by 5s and 5d the former do not form band structures and very sharp lines are measured even in pure crystals. Gadolinium orthoaluminate belongs to the family of rare-earth aluminum oxides that crystallize in a slightly distorted orthorhombic perovskite structure $(Pb_{nm}-D_{2h}^{16})$. Its structure can be regarded as a perfect cube with eight distorted octahedra of oxygen in each vertex, and a Gd^{3+} ion displaced slightly within the xy plane in the center. The local symmetry of the Gd^{3+} ion in the gadolinium aluminate is C_s .⁴ These crystals were grown from high-temperature solution technique.⁵ During the growing of these crystals, color centers can be formed.⁶ We verified the existence of these centers in our samples by using one-photon spectroscopy techniques. The blue fluorescence of color centers was used to monitor the two-photon absorption of the Gd^{3+} energy levels. as indicated in the energy-level diagram in Fig. 1.

The laser used in two-photon spectroscopy was a homemade pulsed dye laser of the modified Littman type,⁷ pumped by the second harmonic of a Nd:YAG laser (YAG is yttrium aluminum garnet). Typical characteristics of this dye laser are 500-kW peak power, 0.1-cm⁻¹ linewidth,⁷ and 10-Hz repetition rate. The luminescence of the color center was collected at 90° from excitation. The completed rejection of laser radiation was done by employing color filters with a transmission range of 24000 ± 1000 cm⁻¹. A homemade data-acquisition system based on a PC-XT microcomputer was used to control the experiment. The details of our experimental apparatus have been described elsewhere.⁸ Conventional

luminescence spectroscopy was employed to measure the emission spectra of the color centers. By using conventional optical absorption the presence of the color centers was not detected. Its presence was only observed when the luminescence spectrum was measured by exciting the ${}^{6}P$ levels of the Gd³⁺ ion (Fig. 2). The luminescence occurs through the color centers showing the coupling between it and the Gd³⁺ ion. Direct fluorescence from the ${}^{6}P$ was not observed. The lifetime of the color centers band was measured by means of temporal resolution spectroscopy. The lifetime is about 10 μ sec.

Although the Gd^{3+} ion transitions are allowed for onephoton spectroscopy,⁹ two-photon absorption spectroscopy is the ideal technique to study this ion. Besides the facility of using visible tunable dye lasers, the other advantage of this technique is that two-photon absorption is only allowed for the ion levels, being forbidden for the crystal excitonic band. As a consequence of it the electronic transitions in the ultraviolet of gadolinium can be seen in the two-photon spectroscopy without the background as op-



FIG. 1. Schematic diagram of the energy level of our sample of GdAlO₃ showing one- and two-photon excitation of ${}^{6}P_{J}$ and subsequent decay by fluorescence of color centers.

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FIG. 2. Fluorescence spectra of color centers in GdAlO₃ at 300 K excited with a broadband excitation around $31\,900$ cm⁻¹.

posed to conventional spectroscopy.

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Two photons from a dye laser are simultaneously absorbed and produce a blue fluorescence in which intensity increases with the square of the laser power. This quadratic behavior is the most evident signature that a twophoton process takes place. Figure 3 shows the twophoton linear polarized absorption spectrum of the ${}^{8}S_{7/2} \rightarrow {}^{6}P_{7/2}$ transitions of the ${\rm Gd}^{3+}$ ion. All the transitions between the ground state and the four Kramers doublets were observed, as predicted by group theory for the C_{s} symmetry. These transitions can be explained in quantitative detail when third-order terms referring to spinorbit interactions are introduced.¹⁰ There are strong anomalies with respect to the predictions of a secondorder theory.

The ${}^8S_{7/2} \rightarrow {}^6P_J \ (J = \frac{5}{2}, \frac{3}{2}), {}^8S_{7/2} \rightarrow {}^6I_J \ (J = \frac{7}{2})$ transitions were observed in the two-photon absorption spectra. Second-order theory permits the transitions, but it is not able to explain the strong line intensity of the observed two-photon spectrum. Their number is according to the prediction of group theory for the site symmetry and ener-



FIG. 3. Two-photon absorption spectra of the electronic transition of Gd³⁺ in GdAlO₃. The absorption in this spectra corresponds to the ${}^{8}S_{7/2} \rightarrow {}^{6}P_{7/2}$ transition. The sample was at room temperature. The energy represents 2 times the laser frequency.



Multiplet	Observed components (cm ⁻¹)	
⁶ P _{5/2}	32450	
	32 500	
	32 540	
⁶ P _{3/2}	33060	
	33100	
⁶ I _{7/2}	35630	
	35650	
	35700	
	35750	

gies measured of these excited states are summarized in Table I.

The remaining ${}^{8}S_{7/2} \rightarrow {}^{6}I_{J}$ $(J = \frac{15}{2}, \frac{17}{2})$ transitions measured by us exhibit enormous discrepancies with the second-order perturbation theory: selection-rule violation and anomalously strong intensities. The second-order perturbation theory predicts zero intensity for ${}^{8}S_{7/2} \rightarrow {}^{6}I_{J}$ $(J = \frac{15}{2}, \frac{17}{2})$ because the selection rule allows only transitions with $\Delta J \leq 2$ for two-photon absorption. In Fig. 4 we show the two-photon absorption spectra of the electronic transitions from the ground state ${}^{8}S_{7/2}$ to the ${}^{6}I_{15/2}$. These transitions that should be forbidden for this theory are stronger than the allowed transitions. The twophoton-absorption-signal scale in both Figs. 3 and 4 are the same and are in arbitrary units. The discrepancies with second-order theory were explained previously with the introduction of third-order terms involving crystal interactions.¹⁰ The inclusion of a third-order term in the perturbation theory may allow transition with ΔJ up to six.¹¹ Therefore, these high-order terms explain our observation of the anomalous intensity of the ${}^{8}S_{7/2} \rightarrow {}^{6}I_{J}$ $(J = \frac{15}{2}, \frac{17}{2})$ two-photon absorption.



FIG. 4. Two-photon absorption spectra of the electronic transition of Gd³⁺ in GdAlO₃. The absorption in this spectra corresponds to the ${}^{8}S_{7/2} \rightarrow {}^{6}P_{15/2}$ transition. The sample was at room temperature. The energy represents 2 times the laser frequency.

We are grateful to Fundação de Amparo a Pesquisa do Estado de Sao Paulo and Conselho Nacional de Desenvolvimento Científico e Tecnológico for financial support.

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