

## Two-stage feature of Hartmann-Hahn cross relaxation in magic-angle sample spinning

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In magic-angle sample-spinning experiments the Hartmann-Hahn cross relaxation between protonated  $^{13}\text{C}$  and protons usually proceeds in two stages, except in fast internal rotating  $^{13}\text{CH}_3$ . The protonated  $^{13}\text{C}$  magnetization of powder samples changes very rapidly during the first tens of microseconds due to the fast energy exchange between each protonated  $^{13}\text{C}$  and its directly bonded  $^1\text{H}$  spins; then it approaches at a much slower rate to a quasiequilibrium value via the energy exchange between these  $^{13}\text{CH}_n$  subsystems and the remaining  $^1\text{H}$  spins. This fact means that the whole  $^1\text{H}$  spin system is not in a quasiequilibrium state and is not describable by a single spin temperature at least during the first stage of the cross relaxation. The two-stage feature has been obviously revealed by the depolarization experiment for  $^{13}\text{C}$  magnetization. The expression for protonated  $^{13}\text{C}$  magnetization as a function of depolarization time has been deduced, which reaches agreement with the experiments semiquantitatively. The depolarization experiment offers a reliable approach to distinguishing between  $^{13}\text{CH}$  and  $^{13}\text{CH}_2$  signals in organic solids.

It is generally accepted that the time dependence of Hartmann-Hahn cross relaxation<sup>1</sup> between rare and abundant spins in NMR may be described by a single exponential function.<sup>2,3</sup> The theory most commonly put forward<sup>4-6</sup> describes the spin system as separate  $I$ - and  $S$ -spin reservoirs which are free to exchange energy in arbitrarily small units, and assigns a temperature to each of the two spin reservoirs so that the cross-relaxation process is describable by a change of spin temperature.

However, non-single-exponential phenomena have also been noted and analyzed. Hahn and his co-workers<sup>1,7,8</sup> have pointed out in their detailed analyses on cross relaxation that the exponential decay law is not valid for a short period of time at the beginning, when the relaxation is diffusion limited. Later Müller, Kumar, Baumann, and Ernst<sup>3</sup> observed strong transient oscillations in the polarized  $^{13}\text{C}$  magnetization as a function of the cross-polarization time in a static ferrocene single crystal. The model adopted to explain the transient oscillation is essentially a modification of that proposed by Slusher and Hahn.<sup>7</sup> It is assumed that the dipolar interactions of the isolated  $S$  spin with the  $I$  spins are neglected except for the coupling to the directly bonded  $I$  spin in the  $SI$  group.

It is our intention in this paper to report that in cross-polarization magic-angle sample-spinning (CP-MASS)<sup>1,9-11</sup> experiments protonated  $^{13}\text{C}$  magnetization of powder samples usually undergoes a two-stage change during the cross polarization except in fast internal rotating  $^{13}\text{CH}_3$ : it changes very rapidly during the first tens of  $\mu\text{sec}$ , then approaches at a much slower rate to a quasiequilibrium value exponentially. This two-stage feature is very common, not being restricted to those samples with fast internal molecular motion like ferrocene.

This two-stage feature was first observed in a selective polarization inversion experiment.<sup>12</sup> However, it is more advisable to reveal the feature with the depolarization method, where, instead of the polarized  $S$  magnetization in the standard CP-MASS experiment, the depolarized  $S$  magnetization is monitored. The pulse sequence utilized

is shown in Fig. 1, which is a simple modification of the standard CP pulse sequence. Two additional intervals  $t_i$  and  $t_{c2}$  are inserted between the usual polarization time  $t_{c1}$  and detection time  $t_d$ . Both irradiations  $H_{1I}$  and  $H_{1S}$  are turned on during  $t_{c2}$  as they are during  $t_{c1}$ , while only the irradiation  $H_{1S}$  is turned on during  $t_i$ . As usual, during  $t_{c1}$ ,  $I$  magnetization is locked along  $H_{1I}$ , while a polarized  $S$  magnetization is created along  $H_{1S}$ . During  $t_i$  (which is about 1 msec in our experiment) however, the  $I$  magnetization will gradually disappear due to spin-spin relaxation, while  $S$  magnetization is still locked along  $H_{1S}$  with only a little loss in intensity. During  $t_{c2}$ , when  $H_{1I}$  is turned on again, the polarization transfer happens again after an interruption during  $t_i$ , but in the reversed direction with respect to that during  $t_{c1}$ , that is, the  $S$  magnetization is now depolarized. The time dependence of the  $S$  magnetization as a function of depolarization time  $t_{c2}$  is recorded and analyzed.

As is well known, high speed spinning at magic angle causes a chemical shift powder pattern to become a single narrow line or a centerband flanked by sideband.<sup>13,14</sup> The intensity of the single line or the centerband is now taken

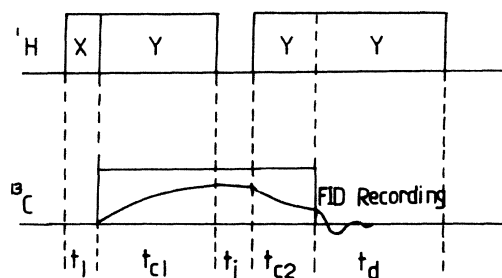


FIG. 1. Pulse sequence utilized in the depolarization experiment, where  $t_{c1}$  is the polarization time for  $S$  spins,  $t_i$  is the depolarization time for  $I$  spins, and  $t_{c2}$  is the depolarization time for  $S$  spins. The development of the  $S$  magnetization is schematized by a solid line.

as the measure of the  $S$  magnetization.

Figure 2 shows the time dependence of intensity of  $^{13}\text{COOH}$  and  $^{13}\text{CH}_2$  signals in a powder sample glycine, keeping  $t_{c1}$  at 5 msec and  $t_i$  at 1 msec. While the  $^{13}\text{COOH}$  signal decreases single exponentially, the  $^{13}\text{CH}_2$  obviously decays in two stages with a turning point at  $t_{c2} \approx 40 \mu\text{sec}$ . Figure 3 shows the similar situation of  $^{13}\text{CH}$  signal in a powder sample terephthalic acid with a turning point at  $t_{c2} \approx 80 \mu\text{sec}$ . The data shown have been obtained on the Bruker MSL-300 spectrometer with a spinning rate of 4 kHz.

The two-stage feature can be explained with the following model, which is essentially the same as that proposed by Müller *et al.*:<sup>3</sup> each isolated  $S$  spin and its directly bonded  $I$  spins make up a quasi-isolate  $SI_n$  subsystem, which is in thermal contact with a reservoir made up of remaining  $I$  spins. The energy exchange in each  $SI_n$  subsystem is very fast due to the strong direct dipolar interaction and determines the rapid change of  $S$  magnetization in the first stage of cross polarization. The much slower one between these subsystems and the remaining  $I$  spins via spin diffusion determines the second stage.

The two-stage feature signifies that the cross relaxation between protonated  $^{13}\text{C}$  and protons belongs to the diffusion bottleneck limit,<sup>1</sup> so that the proton spin system is not always describable by a single temperature, at least so during the first stage.

In addition, it is also noted that the turning point of  $^{13}\text{C}$  signal in  $^{13}\text{CH}_n$  group is just at about  $(n+1)^{-1}$  of its initial value. This fact is by no means occasional and can be well understood if it is taken into account that the ratio of heat capacities<sup>8,15</sup> of the  $^{13}\text{C}$  and  $^1\text{H}$  in the subsystems  $^{13}\text{CH}_n$  is just  $n^{-1}$  under the Hartmann-Hahn matching condition.<sup>1</sup> Besides, this fact offers a reliable approach to distinguishing between  $^{13}\text{CH}$  and  $^{13}\text{CH}_2$  signals in organic solids, because the relative intensity at the turning point is independent of the relaxation rate to a certain extent.

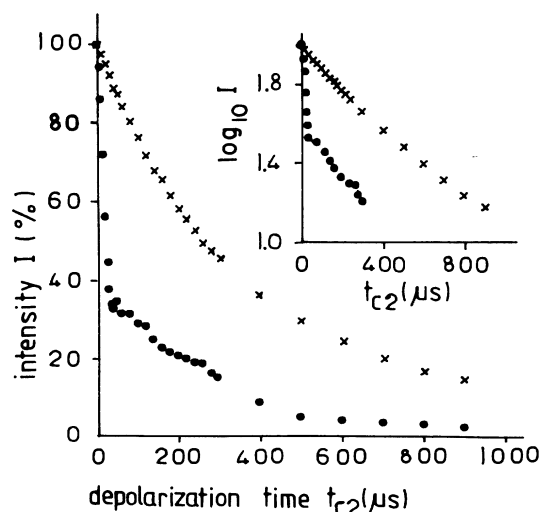


FIG. 2. The time dependence of intensity of  $^{13}\text{C}$  signals on  $t_{c2}$  in a powder sample glycine in the depolarization experiment, keeping  $t_{c1}$  at 5 msec and  $t_i$  at 1 msec.  $\times$  stands for  $^{13}\text{COOH}$  signal, and  $\bullet$  stands for  $^{13}\text{CH}_2$  signal.

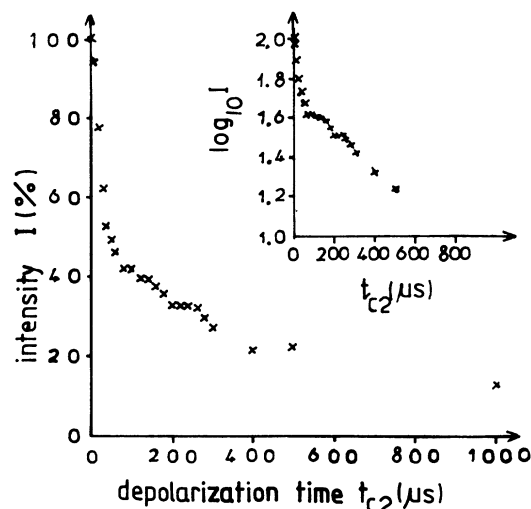


FIG. 3. The time dependence of  $^{13}\text{CH}$  signal on  $t_{c2}$  in a powder sample terephthalic acid in the depolarization experiment, keeping  $t_{c1}$  at 5 msec and  $t_i$  at 1 msec.

It is worthwhile here to emphasize that the depolarization experiment is superior to the standard CP-MASS in revealing the details of cross relaxation process. With the latter it is difficult to know the real maximum polarized  $S$  signal and the exact time dependence of  $S$  signal to be determined by cross polarization alone, because the creation of  $S$  polarization depends not only on the cross-relaxation process itself, but also on some other factors, such as  $T_{1\rho}$  of  $I$  spin, and Hartmann-Hahn matching condition. In the depolarization scheme, however, the finite initial  $S$  magnetization at the beginning of  $t_{c2}$  will certainly reach zero at last and the turning point occurs during the first tens of  $\mu\text{sec}$ . Therefore, the depolarization experiment adequately exposes the two-stage feature, and exactly reveals the relative intensity of  $S$  signal at turning point.

The quantitative expression for  $^{13}\text{C}$  magnetization in high-speed spinning powder samples as a function of depolarization time can be deduced, starting from the formula for static single crystal obtained by Müller *et al.*:<sup>3</sup>

$$M_{sx} = M_{\infty} \left( 1 - \frac{1}{2} e^{-Rt} - \frac{1}{2} e^{-3Rt/2} \cos \frac{1}{2} bt \right), \quad (1)$$

where  $M_{sx}$  is the polarized  $^{13}\text{C}$  magnetization for  $^{13}\text{CH}$  group,  $b$  is dipolar coupling between  $^{13}\text{C}$  and its directly bonded proton,  $R$  is the spin diffusion rate among protons, and  $t$  is the polarization time.

It is easy to obtain an expression for depolarized  $^{13}\text{C}$  magnetization under the otherwise same condition:

$$M_{sx} = M_0 \left( \frac{1}{2} e^{-Rt} + \frac{1}{2} e^{-3Rt/2} \cos \frac{1}{2} bt \right). \quad (2)$$

For a static powder sample, we will get a broadened powder pattern due to the anisotropic chemical shift. Each spot in the line corresponds to certain molecular orientations and may oscillate with a characterized frequency as the function of depolarization time because of the orientation dependence of  $b$ .

As the sample turns, the contributions from all the

magnetically equivalent  $^{13}\text{C}$  spins in the molecules with different orientations will be added to form a single narrow line or a center band flanked by side bands. The oscillations with different frequencies destructively interfere with each other, resulting in a rapid exponential decay of the  $^{13}\text{C}$  magnetization. The time dependence of the dipolar coupling due to the sample spinning will also have an effect on the behavior of the  $^{13}\text{C}$  magnetization, but it can be neglected to a first approximation, because the decay occurs during tens of  $\mu\text{sec}$  which is only tens of a percent of the rotational period of the sample. The expression for  $^{13}\text{C}$  magnetization can thus be obtained from Eq. (2), after summing over all orientations of the crystallites. Instead of  $\cos \frac{1}{2}bt$  in Eq. (2), one gets  $\sum_i (I_i \cos \frac{1}{2}b_i t) / (\sum_i I_i)$ , which can be approximated to  $\exp(-t^2/2T_2^2)$ ,<sup>16</sup> where

$$(1/T_2^2) = \sum_i I_i (\frac{1}{2}b_i)^2 / \left( \sum_i I_i \right)^2 .$$

After the replacement, the depolarized  $^{13}\text{C}$  magnetization

in the  $^{13}\text{CH}$  group in a high-speed spinning powder sample at magic angle will be obtained:

$$M_{sx} = M_{s0} \left( \frac{1}{2} e^{-Rt} + \frac{1}{2} e^{-3Rt/2} e^{-t^2/2T_2^2} \right) , \quad (3)$$

where the second term represents the initial rapid drop. Roughly speaking,  $^{13}\text{C}$  in  $^{13}\text{CH}_2$  turns out to be

$$M_{sx} = M_{s0} \left( \frac{1}{3} e^{-Rt} + \frac{2}{3} e^{-3Rt/2} e^{-t^2/2T_2^2} \right) . \quad (4)$$

Equations (3) and (4) predict not only the basic two-stage feature of protonated  $^{13}\text{C}$  signals, but also the correct turning point of them in the depolarization experiment, reaching agreement with the experiments semi-quantitatively.

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