Electronic band-structure hybridization in the heavy-fermion superconductor $UPt₃$

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Many physical properties, such as electrical resistivity and specific heat, of UPt₃ are different from the other two heavy-fermion superconductors, UBe_{13} and $CeCu₂Si₂$. The electronic structure of UPt₃ also differs from the other two materials because of the presence of the platinum d band, which is close in energy to the uranium $5f$ band that is responsible for the exotic heavy-fermion ground state. In this paper various contributions to linear muffin-tin orbital atomic-sphere approximation electronic band-structure calculations are examined to show that platinum p hybridization is the dominant platinum hybridization mediating the formation of the uranium 5f band. The main effect of uranium 5f and platinum 5d hybridization is instead to induce platinum d character at the Fermi energy that would otherwise be absent, while having little effect on the uranium 5f band itself. Any experiment, such as soft-x-ray emission, resonant photoemission, or nuclear magnetic resonance that can probe this platinum d character could yield an important confirmation of itinerant uranium f electrons.

Heavy-fermion systems exhibit a variety of unusual and exotic ground states.¹ These states are almost surely dominated by the strong electron-electron correlations that are not included in normal density-functional electronic band-structure calculations. Nonetheless, these exotic correlated ground states must form out of a typical metallic one-electron-like state as the temperature is lowered, and it is therefore important to examine and understand the physics of this "normal" state. Also, since there are many possible low-temperature ground states for heavyfermion materials, some peculiarity in the normal-state properties could cause electron-electron correlations to favor one ground state over another.

The focus of this paper is on one-electron hybridization and especially its effects on the uranium 5f electron-band in UPt₃, because this band is almost certainly responsible for the exotic low-temperature properties. The UPt_3 system itself is of special interest among heavy-fermion materials because of the possible coexistence of spin fluctuations and superconductivity² as well as the general possibility that UPt_3 is a *p*-wave superconductor.³

Of the three known heavy-fermion superconductors (UBe₁₃ and CeCu₂Si₂ are the other examples), UPt₃ is different. For example, the UPt_3 electrical resistivity decreases monotonically as the temperature is lowered rather like a normal metal, whereas the other two systems show a far more complex temperature dependence, which are quite similar to each other. Other differences are reviewed in Ref. 1. In terms of one-electron structure, one might readily assume that these differences are due to the platinum 5d bands, since UBe_{13} and $CeCu_2Si_2$ have no d bands anywhere near the Fermi energy E_F (the fully occupied Cu d bands in CeCu₂Si₂ are well below the Fermi energy⁴). In platinum metal E_F cuts across the top of the 5d band, which therefore plays a significant role in the physical properties of platinum. Because platinum makes up 75% of UPt₃, one might expect a similar situation. However, while explicit calculation (see below) shows a filled platinum $5d$ band, nonetheless, these bands are still sufficiently close to the Fermi energy that their effects on the uranium 5f band (which is pinned to E_F by Coulombic energy considerations) demand careful investigation.

Figure ¹ shows the total density of states (DOS) of UPt_3 as calculated by the linear-muffin-tin-orbital⁵ (LMTO) scalar-relativistic electronic band-structure method in the atomic-sphere approximation (ASA) with a Barth-Hedin exchange-correlation potential. The density of states was calculated by the tetrahedron method 6 with 175 k points in the irreducible part of the Brillouin zone for the final converged iterations and for the plots shown in this paper. Equal atomic sphere radii $(S=3.0488)$ bohrs) were used for the uranium and platinum sites in order to reduce overlap of the atomic spheres. The calculations were done for the experimentally observed hexagonal SnNi₃ structure (eight atoms per unit cell, two equivalent uranium and six equivalent platinum sites).

We have also done more detailed calculations^{7,8} includng the combined correction term⁵ as well as fully relativstic (including spin-orbit) calculations, as have others.^{9,10} While these additional effects are important for obtaining an accurate one-electron picture of the UPt₃ electronic structure, our interest here is solely to understand the important influences within the one-electron structure, for which the ASA approximation is sufficiently accurate.

The overall features of the electronic structure are clear from Fig. 1. From approximately -8 to -1 eV there is a large broad peak in the DOS due to the platinum $5d$ bands. Above these bands are broad platinum 6s-6p bands and uranium 6d-7s bands. While the uranium 6d is mainly above E_F it hybridizes with the platinum d bands causing a hybridization gap or relatively low DOS near E_F . The uranium 7p and platinum 5f sit very high in energy above E_F and play no role. The dominant feature of the electronic structure is, of course, the enor-

FIG. 1. Fully hybridized density of states for UPt_3 in the SnNi₃ lattice structure. The calculations were done by the LMTO ASA electronic band-structure method. The zero of energy in the figure is the Fermi energy of the fully hybridized calculations. The density of states (DOS) is the complete density of states including all uranium and platinum contributions and all angular momentum components.

mous DOS of the narrow uranium ⁵f band, which is pinned to E_F . Because the U-U atomic separation is large 11 and hence the overlap of the uranium wave functions is correspondingly small, it is necessary to study the hybridization with the surrounding platinum atoms in order to understand the formation of an uranium 5f-bandlike state.

Within the ASA approximation the influence of the platinum atoms on the uranium Sf band can be studied by selectively turning on uranium Sf hybridization with different platinum angular momentum components. The results of such calculations (more specific details about this procedure are given below) are shown in Figs. 2 (a) and (b), which focus on the relevant uranium f -projected DOS. As expected, the unhybridized 5f band is very narbook. As expected, the uniformized $3f$ band is very har-
row, with a bandwidth $W_f \sim 0.2$ eV compared with the fully hybridized result of $W_f \sim 0.8$ eV. Turning on hybridization with platinum d states shifts the center of the f band to higher energies and produces an additional high energy tail in the projected f DOS. The width of the central peak remains at about 0.2 eV. The hybridization with platinum p states, in contrast, causes a dramatic broadening of the uranium Sf band. While the remaining hybridizations are necessary to get the details of the band structure exactly right (the platinum d hybridization, e.g., is necessary to center the band at higher energies), Fig. 2 (a) suggests that the platinum p hybridization is the dominant effect controlling the band formation of the uranium $5f$ itinerant states.

Two major consequences follow from this: (i) $UPt₃$ is similar to other heavy-fermion systems in that the band formation of the f states is mediated by broad s - p bands of neighboring atoms. (ii) The platinum $5d$ bands seem less likely as a candidate to explain differences between $UPt₃$ behavior and other heavy-fermion superconductors and suggests that more attention be given to other possible candidates, such as the closer uranium spacing of $UPt₃$.

In addition to the results of Fig. 2, platinum s hybridization, which has even less effect than platinum d hybridization in broadening the uranium $5f$ band, as well as various conbinations of different kinds of hybridization such as simultaneous platinum s and d hybridization have

FIG. 2. (a) Uranium f projected density of states for fully hybridized calculations (solid line) and for uranium f and platinum p hybridized calculations (dashed line). The zero of energy is the same as Fig. 1. (b) Uranium f projected density of states for unhybridized calculations (solid line) and for uranium f and platinum d hybridized calculations (dashed line). The zero of energy is the same as Fig. 1.

also been examined. Hybridization between uranium f and uranium s, p, and d states are of course ineffective because of the large U-U atomic separation. In all these cases, the results are similar to Fig. 2, i.e., a combination involving platinum p hybridization is necessary to sufficiently broaden the central f-DOS peak to compare favorably with the fully hybridized result. Similar calculations for a hypothetical cubic Cu₃Au crystal structure (which differs from the observed hexagonal $Ni₃Sn$ structure by the usual fcc-versus-hcp stacking of atomic planes) resembled those of Fig. 2, except for a slight enhancement of the effects of the platinum p hybridization.

Because of the nearness of the filled platinum d band to the uranium f band and to the Fermi energy, it is also important to understand how this d band hybridizes and is otherwise influenced by the uranium atoms. Figures 3(a) and (b) show the effects of uranium-6d—platinum-5d hybridization, where all other hybridization is turned off. Unlike pure uranium metal, in UPt₃ the uranium $6d$ band sits relatively high in energy (about 2 eV above the Fermi energy). For this reason the uranium and platinum d hybridization is relatively weak, each inducing a small density of states in the region of the other band. The main part of the density of states for each band is almost identical in position and magnitude with the corresponding unhybridized bands. A hybridization gap with essentially zero density of states occurs between the platinum and uranium d bands

In contrast to this weak $d-d$ hybridization, Fig. 4 shows the effect of uranium $5f$ and platinum $6d$ hybridization on the platinum 6d band. While this f-d hybridization has little effect on the uranium 5f band as discussed

FIG. 3. (a) Uranium d projected density of states for calculations that only allow hybridization between uranium d and platinum d states. The zero of energy is the same as Fig. 1. (b) Platinum d projected density of states for calculations that only allow hybridization between uranium d and platinum d states. The zero of energy is the same as Fig. 1.

above, the effect on the platinum d character is dramatic. A relatively large peak in the platinum d density of states is induced in the region of the uranium 5f band. Since this density of states is not present in the absence of itinerant uranium 5f electrons (this has also been confirmed by other fully-relativistic calculations, not shown here,⁷ where three uranium f electrons have been forced into the core and only s-p-d orbital angular momentum components were included in the basis set of the bands), this suggests that any experiment that shows an enhanced platinum d character at the Fermi energy would provide strong additional evidence for itinerant f character in UPt₃. Soft-x-ray emission to the platinum $4f$ core states $(N VI-N VII$ emission), resonant photoemission, or nuclear magnetic resonance are some possible experiments where one might look for such effects.

The techniques used to generate Figs. ²—4, i.e., turning various hybridizations on and off as desired, demonstrate the usefulness of the ASA approximation of Andersen⁵

FIG. 4. Platinum d projected density of states for calculations that only allow hybridization between uranium f and platinum d states. The zero of energy is the same as Fig. 1.

for understanding electronic band-structure results. Instead of obtaining the final results in one complicated calculation, one can easily break up the calculation into components and look for the underlying physics in each. These same techniques of course apply to other compounds than just UPt₃. In principle, similar techniques could also be developed for other band structure methods such as Korringa-Kohn-Rostoker (KKR), for example.

Within the ASA approximation, the band-structure eigenvalues and wave functions for a Bloch wave vector k are determined by diagonalizing Hamiltonian H and overlap 0 matrices, which, for compounds, may be schematically indicated by

$$
\langle t \, lqm \mid (H - EO) \mid t'l'q'm' \rangle. \tag{1}
$$

Here t specifies the inequivalent atoms of the compound (in UPt₃ the uranium and platinum sites), q specifies a specific atom in the unit cell (e.g., which of two equivalent uranium sites or six equivalent platinum sites), and lm the standard angular momentum labels. For UPt₃ this is an 128×128 matrix for each k vector. The simplicity and elegance of the ASA is derived from the way that the dependence on the radial potential is confined to a set of five potential parameters for each inequivalent type of atom, whereas the geometry of the crystal lattice only enters the Hamiltonian and overlap matrices through a set of structure constants S_{LL}^{k} (where I have used L as an abbreviation for *tlqm*). Because of this one may easily turn off hybridization for different inequivalent atom types and angular momentum components by setting the structure constants to zero for whatever off-diagonal components $(L \neq L')$ one does not wish to couple. In principle this allows one to block diagonalize different pieces of the matrices separately. That is, instead of diagonalizing one huge 128×128 matrix, one may diagonalize matrices whose dimensions are determined by the number of l and t components that are coupled together.

In our case, the ASA program was modified to allow all diagonal components $t = t'$ and $l = l'$, while zeroing all off-diagonal components. This guarantees that the determinants of our matrices will always be nonzero and hence the single diagonalization of the ASA program works normally, in this case simultaneously generating all the unhybridized bands. Since all the different bands are decoupled from each other, one may focus on any band of interest, the uranium 5f band in our case. The different platinum hybridizations can then be turned on one by one and their effect on the uranium 5f band examined by zeroing all off-diagonal elements $L \neq L'$ of the structure constants, except for those couplings under investigation.

In conclusion, it has been shown that the dominant interaction controlling the formation of the normal-state uranium 5f bands in UPt₃ is the uranium 5f and platinum 6p hybridization and is not hybridization with the platinum $5d$ bands, as intuition would suggest. The main effect of the uranium $5f$ and platinum 6d hybridization is instead to induce a platinum d character at the Fermi energy that would otherwise be absent. Soft-x-ray emission, resonant photoemission, nuclear magnetic resonance, or any other experiment capable of probing this induced platinum d character would therefore be a useful test to

demonstrate itinerant f-electron character at the Fermi energy in UPt_3 .

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