

Dense H₂ and proton NMR in *a*-Si:H

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A model is proposed which resolves an apparent paradox in the results and interpretation of various experiments detecting H and H₂ in *a*-Si:H. A number of measurements indicate that the H₂ in *a*-Si microvoids is at densities of the order of the density of solid H₂. However, at these densities, the electric quadrupole-quadrupole (EQQ) interactions between molecules in bulk fluid or solid H₂ dominate the nuclear spin relaxation $T_1(\text{H}_2)$ and its temperature dependence in such a way that it cannot be used to explain the proton NMR $T_1(\text{H})$ relaxation results for Si-bonded H. In order to fit the *a*-Si:H relaxation data it is necessary that the effective EQQ interaction varies widely among H₂ in *a*-Si:H. The $T_1(\text{H})$ relaxation in *a*-Si:H arises from interaction with that part of the adsorbed H₂ fraction which has a small effective EQQ.

There is an apparent paradox associated with the molecular hydrogen trapped in microvoids in hydrogenated amorphous silicon. The dense fluid and solid H₂(D₂) which has been detected in *a*-Si:H(D) in recent calorimetric,¹⁻³ infrared,^{4,5} and NMR⁶⁻⁹ measurements cannot, for the most part, be the molecular hydrogen responsible for the observed nuclear spin-lattice relaxation $T_1(\text{H})$ of silicon-bonded atomic hydrogen.¹⁰⁻¹² The temperature dependence, magnetic-field dependence, and magnitude of the relaxation all disagree with those which would follow from relaxation via dense fluid and solid H₂.

The original reason for proposing that H₂ exists in *a*-Si:H was that the temperature dependence and magnetic-field dependence of the proton $T_1(\text{H})$ closely resembled those for the phonon-dominated and much faster proton $T_1(\text{H}_2)$ of isolated H₂ molecules in nonmagnetic hosts.^{13,14} $T_1(\text{H}_2)$ in the normal solid or dense fluid phases of H₂ is very different from that for dilute H₂ because of the EQQ interactions between *o*-H₂ molecules in the dense material. The *a*-Si:H NMR $T_1(\text{H})$ results require relaxation via effectively isolated *o*-H₂ molecules with no significant EQQ interactions between them, while the ortho-to-para conversion results, infrared results, and NMR spin counts, indicate H₂ at roughly solid densities. Our proposed resolution to this paradox is that the relaxation-center *o*-H₂ molecules are a part of those dominated by contact with the silicon surfaces, have their intermolecular EQQ interactions quenched, and thus are effectively dilute insofar as nuclear relaxation is concerned.

It now is generally accepted that for most samples of plasma-deposited *a*-Si:H, ≤ 0.1 at. % H₂ located in microvoids act as the dominant nuclear spin-lattice relaxation centers⁹⁻¹¹ (between 3-300 K) for protons bonded into the *a*-Si structure. The proton magnetization diffuses to rapidly relaxing effectively dilute *o*-H₂ molecules, where it then relaxes to the lattice. If this dipolar spin diffusion is fast enough, the temperature variation of $T_1(\text{H})$ is proportional to that which would have been anticipated^{10,13} for $T_1(\text{H}_2)$ of the uncoupled *o*-H₂ relaxation centers. The resulting observed $T_1(\text{H})$ characteristically shows a phonon-related power-law minimum located between 25-55 K. The $T_1(\text{H})$ minimum is deeper^{9,12} for samples known to have microstructure and hence more adsorbed H₂.

The open symbols in the upper part of Fig. 1 show such $T_1(\text{H})$ results reported by Boyce and co-workers^{15,16} for

lattice-bound protons in a high-quality unannealed *a*-Si:H sample deposited from 100% SiH₄ at low-power density (0.025 W/cm²) onto an Al substrate heated to 230 °C. This is the same unannealed sample in which Boyce and Stutzmann⁸ have observed the Pake-doublet spectrum for oriented *o*-H₂ molecules in solid H₂ in the voids below 20 K. The open triangles show $T_1(\text{H})$ for 92.5 MHz and the open circles show $T_1(\text{H})$ for 46 MHz. The 92.5-MHz data in fact extend¹⁶ to 2 K at which temperature $T_1(\text{H})$ has in-

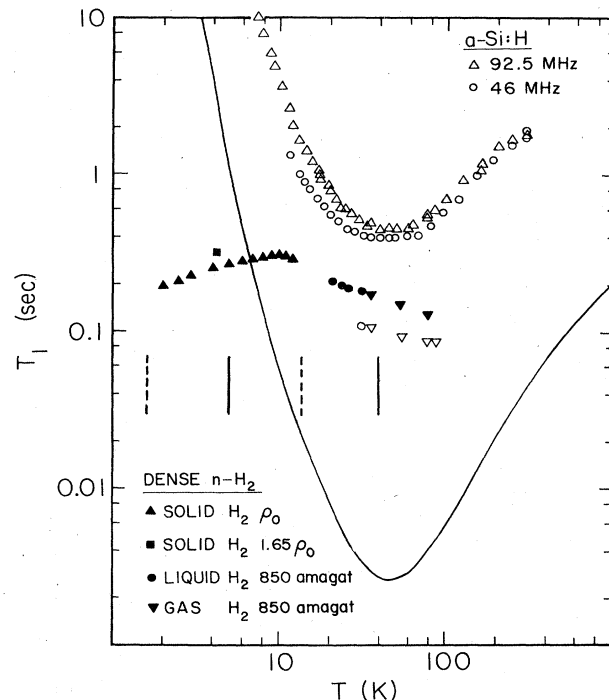


FIG. 1. Proton-relaxation times observed for bonded H in a plasma-deposited sample of *a*-Si:H are compared with those reported for isochoric solid H₂ and dense fluid H₂. The curved line indicates $T_1(\text{H}_2)$ calculated for effectively isolated *o*-H₂ at 92.5 MHz and under conditions described in the text. The dashed vertical lines indicate the spin-ordering transition for *o*-H₂ and the melting temperature for bulk H₂ at zero pressure. The solid vertical lines indicate corresponding transitions in the *a*-Si:H sample.

creased to 600 sec. Analysis of the temperature dependence of $T_1(H)$ above 150 K in this sample has led to the conclusion^{17,18} that the relaxation-center o -H₂ molecules in this case are located in sites with large static electric-field gradients (EFG) of low symmetry. The curved solid line shows the relaxation time $T_1(H_2)$ calculated¹³ for effectively dilute o -H₂ at 92.5 MHz and with a T_1 minimum at 48 K. The calculation has assumed large static electric-field gradients of no symmetry with parameters¹⁴ $r=0$ and $\Theta_c=40$ K. The deviation from a T^2 temperature variation of $T_1(H_2)$ above 150 K is that characteristic^{17,18} for EFG of no symmetry. The minimum value of $T_1(H_2)$, here determined only by ω_0 and the EFG symmetry, is 2.6 msec.

The solid symbols in Fig. 1 show, for comparison, EQQ-dominated spin-lattice relaxation times $T_1(H_2)$ observed in bulk solid n -H₂ and in dense fluid n -H₂ at 850 amagat. The solid triangles indicate the frequency-independent 6.8- and 12.8-MHz results of Amstutz, Meyer, Myers, and Mills¹⁹ for solid n -H₂ at the usual solid density $\rho_0=790$ amagat. The solid square indicates the 4.2 K $T_1(H_2)$ for solid H₂ with $x=0.67$ compressed²⁰ to $1.65\rho_0$ (at some 7 kbar). The solid circles and inverted triangles indicate the Lipsicas and Hartland^{21,22} 30-MHz $T_1(H_2)$ observed for liquid and gas n -H₂ compressed to 850 amagat. The lower open circles and inverted triangles indicate corresponding $T_1(H_2)$ results for fluid n -H₂ at 650 amagat. The dashed vertical lines indicate for saturated vapor pressure (SVP) bulk n -H₂ the usual $T_c=1.6$ K temperature of o -H₂ orientational ordering and the usual melting point, $T_m=13.6$ K. Below 1.6 K the relaxation time $T_1(H_2)$ increases^{23,24} by a factor of 10^3 into an 0.8 K interval as the o -H₂ molecules freeze out into the $m_f=0$ state. The solid vertical lines indicate the much-increased corresponding approximate transition temperatures observed⁸ for H₂ in the unannealed a -Si:H sample of Fig. 1. T_c for much of the H₂ increases to ≈ 10 K and melting occurs over an interval from about 20 K up to the 40 K indicated. Even with the bulk H₂ data shifted and averaged in temperature, it is apparent that there is no similarity between the temperature variations of the proton-relaxation times $T_1(H)$ for the a -Si:H and those of $T_1(H_2)$ for the isochoric solid and dense fluid n -H₂ at densities near those proposed for H₂ in a -Si:H. The bulk H₂ results exhibit no $T_1(H_2)$ minimum and are independent of Larmor frequency. Proton spin systems in thermal equilibrium with each other relax as a single entity with a common relaxation rate $\Gamma = T_1^{-1}$ given by

$$\Gamma = \frac{\sum_i C_i \Gamma_i}{\sum_i C_i} \quad (1)$$

where the C_i are the spin-heat capacities of the various proton systems (H or H₂) with relaxation rates Γ_i . Clearly, bulk dense H₂ in the voids cannot provide the necessary relaxation centers responsible for the observed $T_1(H)$ of the bonded H in amorphous silicon.

The numerical discrepancies associated with the disagreement are significant. More importantly, there are major qualitative discrepancies evident in Fig. 1 for the dependence of relaxation on magnetic field and temperature. The nuclear $T_1(H_2)$ of isolated H₂ molecules in a host lattice is a function of Γ_m , the molecular relaxation rate. The temperature dependence of the nuclear spin relaxation arises from the temperature dependence of Γ_m and the nuclear T_1 depends on the magnetic field when the frequency ω_0 of the NMR experiment is larger than Γ_m . For isolated H₂

molecules in a nonmagnetic host, Γ_m is dominated by the (molecular) spin-phonon interaction. However, with an appreciable density of o -H₂ molecules the intermolecular EQQ interactions dominate. Ebner and Myles²⁵ have performed extensive calculations for nuclear spin-lattice relaxation in solid H₂. Their results can be approximated by the expression

$$T_1(H_2) = 0.5(\rho/\rho_0)^{5/3} \text{ sec} \quad (2)$$

where ρ is the density of o -H₂ and ρ_0 is the 790-amagat density of solid H₂. For n -H₂ at ρ_0 , Eq. (2) yields $T_1(H_2) \approx 0.3$ sec in reasonable agreement with the solid and liquid $T_1(H_2)$ data in Fig. 1. Using the methods of Ref. 13, Eq. (2) yields an EQQ-induced temperature-independent molecular relaxation rate $\Gamma_m = \Gamma_Q$

$$\Gamma_Q = 2.6 \times 10^{12} (\rho/\rho_0)^{5/3} \text{ sec}^{-1} \quad (3)$$

At densities near that of the solid, $\Gamma_Q \gg \omega_0$ and $T_1(H_2)$ is nearly independent of magnetic field and temperature at any frequency employed in a reported NMR experiment. The small temperature dependence of the solid and liquid H₂ $T_1(H_2)$ data in Fig. 1 principally arises from motional narrowing of the Γ_Q EQQ contribution to Γ_m . At 36 K and 850 amagat the coefficient of self-diffusion²¹ in fluid H₂ is 2.1×10^{-4} cm²/sec. This corresponds to a molecular-diffusion motional-correlation frequency near 10^{12} sec⁻¹, of the order of the rigid lattice EQQ rate and more than 10^3 times larger than ω_0 . In addition, the observed $T_1(H)$ in typical a -Si:H samples requires the presence of o -H₂ relaxation centers at concentrations significantly smaller than the 0.5–5 at. % deduced^{1–9} for bulk H₂ in the voids. For the sample shown in Fig. 1 Boyce and Stutzmann⁸ have reported 0.1 at. % H₂ from an analysis of $T_1(H)$, but 0.25 at. % H₂ from the Pake-doublet oriented-H₂ signal intensity. In an annealed sample they found 0.2 and 1.0 at. % H₂, respectively.

The nuclear relaxation paradox is resolved if the wide distribution of characteristics for H₂ in a -Si:H voids includes a wide distribution of EQQ interactions, with small EQQ more probable for the H₂ fraction in contact⁵ with the Si void surfaces. For some of these molecules large surface-related static EFG much greater than EQQ can vary appreciably between nearby o -H₂. This effectively quenches the EQQ interaction and results in a small rate $\Gamma_Q \ll \omega_0$. For such o -H₂ the phonon-induced temperature-dependent molecular rate $\Gamma_m(T)$ can vary through ω_0 and produce a $T_1(H_2)$ minimum with a power-law temperature dependence similar to that characteristic^{10–14} of isolated o -H₂ molecules in a solid nonmagnetic host. These surface-relaxation-center H₂ usually are present at concentrations^{9,12} ≤ 0.1 at. %. Although the effective EQQ is small (because of the large surface EFG), these surface o -H₂ have significant numbers of near neighbors so that ortho-to-para conversion takes place via the bimolecular process at the rate observed²⁶ from proton NMR. In fact large EFG's have been observed for H₂ on zeolite surfaces,²⁷ in contrast with the much smaller EFG's inferred²⁸ for H₂ on grafoil surfaces. If an appreciable fraction of the H₂ is adsorbed on internal Si surfaces, then the pressure needed to hold it at large density is greatly reduced. The ortho-to-para conversion-rate constant observed²⁶ for the relaxation-center o -H₂ in one a -Si:H sample was 0.010 h^{-1} . This is in reasonable agreement with the 0.017 h^{-1} determined⁵ (in another sample) for infrared components associated with H₂ in contact with the Si sur-

faces of voids and in contrast with the 0.07 h^{-1} characteristic⁵ of H₂ not in contact with the void surfaces.

A second H₂ species from the point of view of proton relaxation, includes those surface H₂ for which the EFG do not vary appreciably between H₂ sites and the H₂ in the volumes of larger voids for which the same thing is true. In these cases EQQ remains large; there is a large corresponding molecular rate $\Gamma_Q \gg \omega_0$; there can be no $T_1(\text{H}_2)$ minimum, and the intrinsic $T_1(\text{H}_2)$ given by Eq. (2)

remains long ($\sim 0.3 \text{ sec}$) compared with that for the relaxation-center *o*-H₂ at all temperatures between a few and 300 K.

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