Superconductivity in uranium compounds with Cu₃Au structure

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Superconductivity has been observed in uranium compounds with partly disordered Cu_3Au -type crystal structure and interatomic U–U distances of more than 4 Å. Low-temperature specific-heat experiments reveal no anomalous enhancement of the electronic specific heat thus distinguishing the present materials from the unconventional superconductors UBe₁₃ and UPt₃.

Until very recently, only a few uranium compounds were known to be superconducting at low temperatures, namely, $U_6 X$ compounds with X = Mn, Fe, Co, and Ni,^{1,2} UCo,¹ and U_2PtC_2 .³ For elemental uranium, superconductivity is claimed for its γ (Ref. 3) and its α modification, but the latter under external pressure only.⁴ In these materials superconductivity was ascribed to the formation of *f*-electron-derived energy bands due to direct 5*f*-wavefunction overlap between adjacent nearest-neighbor U atoms which are less than about 3.6 Å apart.⁵

Newer work on uranium intermetallics has shown that superconductivity is also possible when the U atoms are farther apart than 4 Å. So far, the most extreme example in this respect is UBe_{13} ,⁶ where the nearest-neighbor U-U distance is 5.13 Å. In UPt₃ (Ref. 7) this distance is 4.13 Å, considerably larger than that found in the substances mentioned above. In these latter two cases, however, superconductivity develops out of a rather anomalous normal state and involves electrons with unusually heavy effective masses, giving rise to enormous electronic specific heats at low temperatures. For both UBe₁₃ (Refs. 8 and 9) and UPt₃ (Ref. 9) it has been reasoned that superconductivity is not primarily due to conventional electron-phonon interaction, and therefore the superconducting state may be characterized by $l \neq 0$ paired electrons as opposed to the conventional l = 0 BCS superconductors, where l is the electron-pair angularmomentum quantum number.

With these observations in mind, it seemed of interest to check whether the mentioned interatomic U–U distance is a crucial parameter in determining the lowtemperature behavior of metallic uranium compounds and we also simply aimed at finding additional superconducting U compounds. In the course of this work we discovered superconductivity in U compounds crystallizing in the cubic Cu₃Au structure. Superconductivity was probed by measuring the magnetic susceptibility between 0.02 and 4 K, using a low-frequency ac mutualinductance system with a primary-field amplitude of less than 0.1 Oe. For some of the compounds the magnetic susceptibility was also measured between 1.5 and 300 K, using a sample-moving magnetometer. In order to obtain values for the electronic specific-heat parameter γ , we also measured the temperature dependence of the specific heat between 1.5 and 12 K for two typical compounds. For further characterization we also measured the electrical resistivity ρ of three compounds between 1.5 and 300 K.

The samples were synthesized by arc melting under purified argon. The purity of the starting elements Al, Ga, Si, and Ge were 99.99% or better, that of Ru and Rh was 99.9% (Johnson-Matthey Co., London), and that of U was 99.7% with carbon as the main impurity. Of the materials studied (see Table I), UAl₂Si₂ and UAl₂Ge₂ had been reported by Zygmunt¹⁰ to crystallize with primitive cubic cells and a = 4.145(2) and 4.217(5) Å, respectively. From the experimental density we derive one formula unit per unit cell. This leads to $Pm\overline{3}m$ as the probable space group, i.e., a partly disordered perovskite or a Cu₂Au structure. The former structure, however, would require a much higher density. Thus we are left with the possibility of partially disordered Cu₃Au structures as met in ThPd₄ and UPd₄. In the case of UAl₂Si₂ the Guinier photographs indicated a slight variation of the lattice constant (see Table I). All UA12Si2 powder patterns contained some foreign lines, though no lines from unreacted Si. The Guinier pattern of UAl₂Ge₂ revealed traces of free Ge in addition to the pure Cu₃Au-type phase. The same holds for the UGa₂Ge₂ sample which surprisingly had almost the same lattice constant as the corresponding Al compound.

Based on the lattice constants a_0 of the Cu₃Au end members $UX_3(X=AI, Ga, Si, and Ge)$ and assuming the validity of Végard's rule, we would expect the observed lattice constants at the following compositions (with the 1a positions (0,0,0) fully occupied by U atoms only, and the X atoms statistically distributed on the 3c positions $(\frac{1}{2}, \frac{1}{2}, 0; \frac{1}{2}, 0, \frac{1}{2}; 0, \frac{1}{2}, \frac{1}{2})$: UA1_{1.68}Si_{1.32} (4.145 Å), UA1_{1.47}Si_{1.53} (4.163 Å), UA1_{2.5}Ge_{0.5} (4.219 Å), and UGa_{2.1}Ge_{0.9} (4.218 Å). In the case of the U-Al-Si compound one might accept this interpretation, but not for the U-Al-Ge and U-Ga-Ge phases. Since according to the x-ray photographs these latter phases only contained Ge as a second phase we measured their density δ with a pycnometer. For UA1₂Ge₂ we obtained δ =7.73(6) g/cm³. Assuming the sample to consist of a mixed

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Substance	Sample No.	a_0 (Å)	T_c (K)	$\gamma \ (mJ/mol K^2)$	$\beta (\mathrm{mJ/mol}\mathrm{K}^4)$	Comments
UAl ₂ Si ₂	91	4.145(5)	1.34 ±0.05	27.9	0.435	
	77	4.163(4)	not checked			
	99	4.176(6)	not checked			
UAl ₂ Ge ₂		4.219(3)	1.60 ± 0.10			
UGa ₂ Ge ₂		4.218(3)	0.87 ± 0.08			. ·
URu ₃		3.977(3)	0.145 ± 0.005	12.4	0.413	partly disordered
URu ₂ Rh		3.977(4)	<20 mK			purity association
USi ₃		4.0348(8)	< 20 mK			
UGa ₃		4.256(3)	<20 mK			orders antiferromagnetically at 67 K

TABLE I. Relevant physical parameters of some U compounds with Cu₃Au-type crystal structure.

phase Cu₃Au-type free germanium, and $(U_x Al_{1-x})Al_{3x-1}Ge_{4-3x} + (5x-4)Ge$, the above experimental value for the density suggests a composition close to $(U_{0.82}Al_{0.18})Al_{1.46}Ge_{1.54}+0.1$ Ge. Unfortunately, the quality of the Guinier photograph was not sufficient to allow a clear choice based on the intensity of the diffraction lines. The pycnometric density for the U-Ga-Ge sample came out as 9.13(6) g/cm³. For a Cu₃Au-type one-quarter of the Au sites empty, cell with $(U_{0.75}\square_{0.25})Ga_{1.5}Ge_{1.5}$, the calculated density would be $\delta = 8.706$ g/cm³, which is much too low, while for a perovskite-type distribution UGa(Ge₂Ga) the calculated value $\delta = 11.61$ g/cm³ would be much too high. The observed density corresponds to a sample composition $(U_{0.87}Ga_{0.13})Ga_{1.6}Ge_{1.4}+(0.35)Ge$. However, we do not depend heavily on the correct composition of our Cu₃Autype phases; this view is even more valid since we cannot completely exclude a small fractional occupation of the 1b position $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$. Whatever the exact stoichiometry may be, the U–U distances are clearly greater than 4 Å which usually favors localized behavior of the uranium 5felectrons. Characterization of the binary Cu₃Au compounds was, of course, much less problematic than for the ternaries.

From measurements of the ac magnetic susceptibility, UAl₂Si₂, UAl₂Ge₂, UGa₂Ge₂, and URu₃ were found to be superconductors around and below 1 K. The experimental values for the superconducting transition temperature T_c are listed in Table I. We also investigated UGa₃, USi₃, and URu₂Rh which are, however, not superconducting above 0.02 K. The special case of UGa₃ will be further discussed below. The true composition of URu₂Rh may slightly deviate from the given stoichiometry (weight of the starting elements, 2.535 g; button after arc melting, 2.515 g; no chemical analysis). On the other hand, URu₂Rh may be a further example of an unexpected lattice-constant variation in a series of uranium alloys, certainly correlated with the electronic structure. It may also be noteworthy that URu₃ and URu₂Rh are both extremely hard, whereas the other compounds break rather easily.

In Fig. 1 we show the temperature dependence of the magnetic susceptibilities of URu_3 and UAl_2Si_2 between 4 and 280 K. Apart from these compounds which may be regarded as typical representatives of our binary and ternary Cu_3Au -type compounds, we also investigated and

show the results of UGa₃ and USi₃. For all compounds we find an almost temperature-independent χ in this temperature range, indicating the absence of well-defined magnetic moments on the U ions in these materials, contrary to what one might expect from the lattice constant values. A somewhat unusual temperature dependence of χ for UGa₃, which has previously also been observed by Buschow and van Daal¹¹ with about 5% higher absolute values than ours, might suggest but certainly not indicate a magnetic phase transition below 100 K in this compound (see below). A comparison of the data in Ref. 11 with ours for USi₃ reveals that the former were probably obscured by impurities at lower temperatures since we do not observe the previously reported upturn below 150 K. The contribution of the free silicon in our sample can safely be neglected.¹²

If the temperature-independent χ 's are interpreted as Pauli-type electronic magnetic susceptibilities, their values are rather large, however. It was therefore of interest to check whether the electronic specific heat might be equally enhanced at low temperatures, and we show the temperature dependence of the specific heats again of URu₃ and of UAl₂Si₂ between 1.5 and 12 K in Fig. 2. The c_p/T versus T^2 plot indicates that only at the lowest temperatures $c_p(T)$ may satisfactorily be represented by

$$c_p = \gamma T + \beta T^3 , \qquad (1)$$

where γ is the electronic specific-heat parameter and β







FIG. 2. Low-temperature specific heat of UAl₂Si₂ and URu₃ between 1.5 and 12 K.

determines the lowest-order contribution to the lattice. Obviously this T^3 term is almost identical for both compounds (see Table I). At higher temperatures we observe deviations from this simple behavior which are very similar to those recently reported for U_6Fe ,¹³ but have also been observed in Chevrel phases.¹⁴ In Ref. 14 this temperature dependence of c_p/T was successfully reproduced with calculations, taking into account details of the electron-lattice interactions. We note that this temperature dependence of c_p is characteristically different from that reported earlier for binary U compounds with Cu₃Au structure.¹⁵ The resulting γ values (see Table I) are obviously much smaller than what is obtained for UBe₁₃ or UPt₃ (although in the latter case, the U-U distance is similar to our values), but of the same order of magnitude as those found in cubic NaCl-type U monopnictides and monochalcogenides¹⁶ which also have similar nearest



A considerable difference is also observed in the general temperature dependence of the electrical resistivity between USi₃ and UAl₂Si₂. We show the results of our $\rho(T)$ measurements between 1.5 and 300 K for both these compounds in Figs. 3 and 4, respectively. As in ordinary metals, $\rho(T)$ of USi₃ decreases steadily on cooling and a residual resistivity ratio $\rho_{300}/\rho_{4.2}$ of about 100 is observed, much higher than what may be estimated from the data given in Ref. 11. Below 50 K, the observed temperature dependence of ρ is rather unusual, as may be seen from the expanded plot contained in Fig. 3. The data do not



FIG. 3. Temperature dependence of the electrical resistivity of single-crystalline USi₃ between 1.6 and 300 K. $\rho(300 \text{ K})$ is of the order of 20 $\mu\Omega$ cm.



FIG. 4. Temperature dependence of the electrical resistivity of UAl_2Si_2 between 1.6 and 300 K.



FIG. 5. Temperature dependence of the electrical resistivity of single-crystalline UGa₃ between 1.6 and 300 K.

yield a T^2 dependence of ρ in this temperature range as was reported in Ref. 11. These differences may be due to the fact that our sample was a flux-grown single crystal.

Turning to UAl₂Si₂, the absolute value of ρ is already higher at room temperature, as may be guessed from $\rho(T)$ of UAl₃.¹¹ As is expected for a disordered compound, ρ is not very much reduced upon cooling. In this case, the low-temperature resistivity follows a T^3 law rather than a T^2 law below 40 K, again characteristically different from the behavior of the binary compounds reported in Ref. 11.

Again in contrast to previously published data¹¹ is the T dependence we observe for ρ of UGa₃ below room temperature, measured on a single-crystal specimen. Starting from a room-temperature resistivity of the order of 100 $\mu\Omega$ cm, $\rho(T)$ decreases very slowly but with increasing positive $\partial\rho/\partial T$ upon cooling. At 67 K, ρ starts increasing again with decreasing temperature, passes through a maximum around 62 K, and finally reaches a still rather large residual resistivity at liquid-helium temperatures. The anomaly of the electrical resistivity below 67 K clearly indicates a phase transition which is, as established by

neutron-diffraction experiments^{18,19} of antiferromagnetic nature. The observation of this anomaly hence corroborates the claim of magnetic ordering in UGa₃ from previous neutron studies,^{18,19} although the temperature dependence of χ is not indicative of that at all. Further studies with regard to details of the magnetically ordered state of UGa₃ might be very instructive.

In conclusion, we have found superconductivity in U compounds with Cu₃Au structure and nearest U–U distances of more than 4 Å. The normal-state properties resemble those of "enhanced" d transition metals, indicating that the f electrons in some way take part in the formation of itinerant electron states but do not show the anomalous behavior at low temperatures as, for example, in UBe₁₃ and UPt₃. In this context it is interesting to note that the ratios of the experimental values of χ and γ are much larger for our superconducting substances than those found for UBe₁₃ and UPt₃.

We find it remarkable that although the binary compounds UAl_3 , USi_3 , and UGe_3 are not superconducting, mixtures resulting in ternary compounds with disordered Cu_3Au structure are superconductors with critical temperatures close to or even above 1 K. Even more surprising is the fact that by starting from magnetically ordering UGa_3 , a superconductor is obtained by replacing Ga with Ge, although, as mentioned above, the other binary end member is again not superconducting. Finally, our work implies that the atomic separation alone is not a reliable parameter for predicting the low-temperature behavior of U intermetallics.

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