Valence instability in $YbPd_2Si_2$: Magnetic susceptibility, x-ray absorption, and photoemission studies

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(Received 14 December 1983)

The bulk and surface valence states of Yb were determined for the intermetallic compound YbPd₂SI₂ by magnetic susceptibility (χ) , L_{III}-edge x-ray absorption, and valence-band photoemission measurements. In the susceptibihty data this compound exhibits the characteristic features of homogeneously mixed-valent systems: enhanced Pauli paramagnetism at low temperatures, a maximum in χ at about 30 K, and Curie-Weiss behavior at high temperatures. The observed temperature dependence of χ is discussed in light of interconfigurational fluctuation (ICF) as well as Fermiliquid models. The mixed-valent state of Yb is also borne out in the L_{III} -edge x-ray absorption spectra, which reveal a weak temperature dependence of the Yb mean valence \bar{v} , in agreement with the results from an analysis of the χ data in light of the ICF model. Valence-band photoemission spectra support our view of a homogeneously mixed-valent state of Yb; they further identify a valence transition to the divalent state in the outermost surface layer. 4f binding energies are found to be 0.80 ± 0.05 eV higher at the surface as compared to the bulk. All three methods suggest that within the limits of accuracy—the room-temperature mean valence of Yb in YbPd₂Si₂ is close to the high-temperature limit given by the ionic ICF model.

I. INTRODUCTION

Ever since the structure of $ThCr₂Si₂$ was determined (BaAl₄-type, space group $I4/mmm$), a large number of isotypic rare-earth (RE) compounds of the form RX_2Y_2 $(R = RE, X = transition element, Y = B, Ga, Si, Ge, Sn, P,$ As, and Sb) have been reported.² These ternary compounds have attracted considerable attention in the recent past because of the interesting Kondo, valencefluctuation, $5-5$ superconducting, $5-8$ and magnetic properties $9-10$ which some of these compounds possess. Particularly, the recent characterization of $EuPd_2Si_2$ as a rather unique intermediate-valent system, $\frac{11}{1}$ with a precipitous temperature- and pressure-induced valence transition, $11-15$ aroused increased interest in the field of valencefluctuation phenomena in these ternary compounds.

Systematic studies² of the lattice volume of these ternary RE compounds suggest a correlation between the valence state of a given RE ion (Ce, Eu, and Yb) and the position of the elements X and Y in the Periodic Table. In addition, any abnormality in the valence of Eu in a given compound seems to be indicative of the occurrence of a similar anomaly in the isotypic Yb compound. This prompted us to investigate the compound $YbPd_2Si_2$ for the occurrence of valence instabilities, although no anomaly for YbPd₂Si₂ is visible in the lattice volume data of the RPd_2Si_2 series.² L_{III} -edge x-ray absorption (XA) and 4d core-level x-ray photoemission (PE) studies have been reported previously for YbPd₂Si₂ at room temperature;¹⁶ it was not possible, however, to decide from those results whether the observed valence mixing of Yb is static (inhomogeneous) or dynamic (homogeneous) in nature.

In this paper we report magnetic susceptibility (in the temperature range from 2.4 to 300 K), L_{III} -edge XA (16 to 300 K), and valence-band PE (at 300 K) studies of YbPd₂Si₂. These results conclusively prove the dynamic nature of the valence mixing in $YbPd_2Si_2$. We find a weak temperature dependence of the mean valence \bar{v} of Yb, contrary to what is known for the mean valence of Eu in EuPd₂Si₂. In addition, we observe a shift of the $4f$ binding energy in the outermost surface layer by 0.80 ± 0.05 eV to higher values as compared to bulk Yb ons. This allows a separation of the $4f¹³$ final-state spectral features originating from bulk and surface Yb^{2+} ions. As in other homogeneously mixed-valent compounds, $17-20$ the observed surface and bulk intensities indicate that the outermost surface layer of $YbPd_2Si_2$ is in the divalent state.

II. EXPERIMENTAL METHODS

The $YbPd_2Si_2$ compound used for the present studies was prepared by melting the constituent elements in an arc furnace in an atmosphere of purified argon. By adding a slight excess of Yb, the loss due to vaporization of Yb while melting was compensated. The ingot thus obtained was homogenized by vacuum annealing at 700'C for four days. From x-ray diffraction, the single-phase nature of the sample was confirmed, and the obtained values for the lattice constants were found to be in good agreement with those reported in the literature.²

The magnetic susceptibility measurements were performed in an external magnetic field of S kOe in the temperature range from 2.4 to 300 K using a vibrating sample magnetometer. For the XA experiments in the region of the Yb- L_{III} edges, performed in transmission geometry, the highly monochromatic tunable x-ray beam from the EXAFS-II spectrometer at the Hamburg Synchrotron-

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strahlungslabor (HASYLAB, DESY, Hamburg) was employed. [The rocking-curve width is ≈ 2 eV (full width at half maximum) at 8.9 keV.] XA coefficients were obtained by monitoring the intensities of the incident and transmitted x-ray beams with nitrogen-filled ionization chambers. A homogeneous sample was prepared by pressing the finely powdered material into paraffin. The valence-band PE experiments were performed at the Berliner Elektronenspeicherring-Gesellschaft fur Synchrotronstrahlung (BESSY, Berlin) using the photon beam of the SX-700 monochromator. At 120-eV photon energy the combined-system resolution of the double-pass cylindrical-mirror analyzer and monochromator was about 0.5 eV. Clean surfaces were prepared by scraping the sample in a vacuum of 1×10^{-10} Torr with a metalbonded diamond file; the cleanliness was checked via the 0 2p PE signal.

III. RESULTS AND DISCUSSION

A. Magnetic susceptibility

The temperature dependence of the magnetic susceptibility (Y) measured in an external field of 5 kOe is shown in Fig. 1. χ attains a peak at $T_{\text{max}} \approx 30$ K and tends to a constant value below 10 K. Above 120 K, χ^{-1} is a linear function of T , with a slope corresponding to an effective magnetic moment of $\mu_{\text{eff}}=4.5\pm0.1\mu_B$; this is very close to the moment expected for the free Yb^{3+} ion $(\mu_{\text{eff}}=4.54\mu_B)$. From an extrapolation of the hightemperature behavior of χ^{-1} , we obtain a paramagnetic Curie temperature, $\Theta_p = -70$ K. Such a large negative Θ_n is characteristic of homogeneously mixed-valent compounds; its magnitude is much larger than the Θ_n values known²¹ for other stable-valent RPd_2Si_2 compound (GdPd₂Si₂: $\Theta_p = 2$ K; TbPd₂Si₂: -1 K; DyPd₂Si₂: -4 K; HoPd₂Si₂, $\frac{1}{2}$, $\frac{1}{2}$ **K**; ErPd₂Si₂: -1 **K**, Byrd₂Si₂: -3 **K**). We have also studied the magnetization of YbPd₂Si₂ at 4.2 K in external fields up to 70 kOe and found it to be a linear function of the applied field. In the temperature range of investigation (2.4 to 300 K), there is no indication for magnetic order-

FIG. 1. Magnetic susceptibility of $YbPd_2Si_2$ as a function of temperature. The solid line is obtained from theory as described in the text. The inset shows the linear relationship between χ and T^2 for $T \ll T_{\text{max}}$.

ing. These features in the susceptibility data—high negative Θ_p , a maximum in the temperature dependence of X, and nonmagnetic behavior at low temperatures conclusively prove the homogeneously mixed-valent state of Yb in $YbPd_2Si_2$.

The observed temperature dependence of the magnetic susceptibility can be interpreted within the framework of an ionic interconfigurational fluctuation (ICF) model²² as well as Fermi liquid models.²³⁻²⁵ In the ICF model, expectation values of the measured physical quantities (valence, Mössbauer isomer shift, magnetic susceptibility, etc.) are calculated by averaging over the occupied $4f^n$ and $4f^{n-1}$ -multiplet states. The energy difference beween the lowest-lying ionic states of the $4f^{n-1}$ and $4f^n$ configurations is denoted by E_x and all ionic levels are assumed to have the same finite mixing width $\Gamma = k_B T_f$; T_f is called the characteristic fluctuation temperature. Applying this model to the case of Yb, we obtain for the normalized occupation probability of the Yb^{3+} state

$$
p_3 = 8/[8 + \exp(E_x/T^*)],
$$
 (1)

where E_x is expressed in degrees Kelvin and $T^* = (T^2 + T_f^2)^{1/2}$. T^* is called the effective temperature, 26 and it takes both quantum and thermal fluctuations into account. This model qualitatively explains the observed maximum in X if the Yb^{3+} ionic state is assumed at higher energy than the Yb^{2+} state. When the temperature is lowered, the Curie-Weiss increase in χ is then overcompensated by a decrease in χ due to increased population of the lower-lying nonmagnetic Yb^{2+} state. This ICF model has successfully explained the susceptibility behavior of a number of intermediate-valent Yb sys $tems.²²$ Quantitatively, the magnetic susceptibility of an intermediate-valent Yb system, neglecting crystal-field effects, is given by

$$
\chi(T) = \frac{8N(4.54\mu_B)^2}{3k_B T^*[8 + \exp(E_x/T^*)]}.
$$
 (2)

All symbols in Eq. (2) have their usual meanings. In this model, the high-temperature limit of \bar{v} for Yb is obtained as 2.89 (assuming positive values for E_r).

We have least-squares fitted the theoretical form given by Eq. (2) to the measured χ data. For the temperature range above 33 K we obtain $E_x = 70 \pm 5$ K and $T_f = 100 \pm 5$ K. A good fit of the data measured in the temperature range 10 K $< T < 33$ K is obtained only if either E_x or T_f is assumed to be weakly temperature dependent. With a temperature-independent T_f , E_x takes the form $E_x = 154.6 - 4.64T + 0.07T^2$ (for $T < 33$ K); thus, at 10 K, $E_x = 115$ K. These parameters lead to a slightly temperature-dependent mean valence of Yb $(\bar{v} \approx 2.86, 2.82, \text{ and } 2.75 \text{ at } 300, 85, \text{ and } 16 \text{ K}, \text{ respective-}$ \mathbf{I} y).

Recently it was shown that the maximum in the susceptibility of several mixed-valent Yb and Ce compounds may also be interpreted in a Fermi-liquid model.²³ This interpretation is based on the very similar scaling behavior of X in these $4f$ systems and the strongly interacting Fermi-liquid ³He. At low temperatures ($\overline{T} \ll T_{\text{max}}$), the magnetic susceptibility should then have the form²

$$
\chi(T) = \chi(0)(1 + \alpha T^2) \tag{3}
$$

For $T>T_{\text{max}}$ free-spin behavior is expected. This leads to a maximum in χ for positive values of α , as observed in several Ce and Yb compounds.

We have analyzed the χ data for YbPd₂Si₂ within this model, with the result that both a T^2 behavior of χ for $T < 16$ K as well as the scaling behavior characteristic of Fermi-liquid systems are observed. The first point is obvious from the inset in Fig. 1, where it is shown that χ varies linearly with T^2 at low temperatures. The scaling behavior, on the other hand, leads to a proportionality bebehavior, on the other hand, leads to a proportionality be-
tween $\chi(0)$, T_{max} , and Θ_p , which for Yb systems is given by the relation²³ $C/X(0)T_{\text{max}} \le 5$, where C is the free-ion Curie constant. Our results for $YbPd_2Si_2$ are in agreement with such a ratio. In summary, we can say that $YbPd_2Si_2$ exhibits all the characteristic features of strongly interacting Fermi-liquid systems, like enhanced Pauli paramagnetism, T^2 behavior of χ at low temperatures, crossover to free-spin behavior at high temperatures, and scaling. It is therefore of interest to perform specific-heat measurements on $YbPd_2Si_2$ and verify the relation between $\chi(0)$ and the linear coefficient of specific heat known for such systems.²⁷

B. L_{III} -edge x-ray absorption

In order to support the conclusions obtained from the susceptibility measurements regarding the temperature dependence of \bar{v} of Yb in YbPd₂Si₂, L_{III} -edge measurements were performed at various temperatures. Figure 2

FIG. 2. Yb L_{III} -edge x-ray absorption of YbPd₂Si₂ at 16 and 300 K. The solid lines represent the results of least-squares-fit analyses as described in the text. The dash-dotted and dashed subspectra are assigned to Yb^{2+} and Yb^{3+} initial-ionic states, respectively.

shows the L_{III} edge of Yb at the lowest and the highest temperature points studied (16 and 300 K). To derive values for the mean valence, a superposition of two identical L_{III} -edge profiles, separated by $\Delta E(2p)=6.5 \text{ eV}$, was least-squares fitted to the experimental data points. The L_{III} -edge profiles were analytically described as the sum of a Lorentzian line and an arctangent function convoluted with a Gaussian, representing the finite-energy resolution of the spectrometer. In agreement with similar work on other mixed-valent Yb compounds, the L_{III} -edge Feature at lower excitation energy is assigned to $2p^54f^{14}$ and the intense feature to $2p^54f^{13}$ final core states, 28 originating from Yb^{2+} and Yb^{3+} ions, respectively.^{29,30} The difference in the 2p excitation energies, $\Delta E(2p)$, used in the present analysis is the same as the one found for the mixed-valent high-pressure phase of Yb metal.²⁹ The best fits were obtained for $\Delta E(2p) = 6.5$ eV, but the resulting valences (given below) are the same within the given error bars for slightly different values of $\Delta E(2p)$.

From the relative intensities of the Yb²⁺- and Yb³⁺derived L_{III} -edge features, the following values for the mean valence \bar{v} were obtained: $\bar{v}=2.89\pm0.08$ at 300 K and $\bar{v} = 2.82 \pm 0.08$ at 16 K. The absolute error bars are quoted as ± 0.08 corresponding to about five times the statistical errors in the data analysis. With such a conservative estimate of error bars we believe that possible systematic errors due to uncertainties in the line shape and possible differences in $2p \rightarrow 5d$ transition probabilities for the Yb^{2+} and Yb^{3+} ions, as well as possible effects from core-hole screening, are accounted for. Despite the relatively large absolute error bars, however, the achieved accuracy in the relative variation of \overline{v} with temperature is estimated to be better than ± 0.02 .

 L_{III} -edge measurements were performed at five different temperatures. The results are presented in Fig. 3, where the difference $\Delta \bar{v}$ of the mean valence at a given temperature relative to the value at 300 K is plotted as a function of temperature. The data clearly show a small

FIG. 3. Temperature dependence of the Yb mean valence \bar{v} as obtained from L_{III} -edge measurements (solid points); plotted is the difference between $\overline{v}(T)$ and the value at 300 K. The solid line through the data points serves as a guide to the eye. The dashed curve is obtained from an analysis of the χ data in the light of the ionic ICF model, as described in the text.

but distinct decrease of \bar{v} by $\Delta \bar{v} = 0.07$ when the temperature is lowered from 300 to 16 K. For comparison, we have also included in Fig. 3 (dashed curve) the variation of \bar{v} as obtained with the ionic ICF model, using the parameters $E_x(T)$ and T_f as determined from an analysis of the temperature dependence of the magnetic susceptibility (see above).

The close agreement between the directly determined temperature variation of \bar{v} from L_{III} -edge measurements and the one obtained from the ionic ICF model deserves special emphasis. It shows that the ICF model, in the form employed here, is suitable in describing the temperature variation of \overline{v} . In addition, the consistency of the \overline{v} results from an analysis of the susceptibility and L_{III} -edge data suggests rather small final-state effects in the L_{III} edge XA spectra of YbPd₂Si₂. Finally, we may say that the clearly observed variation of \bar{v} with temperature confirms our view of a homogeneously mixed-valent state of Yb in $YbPd_2Si_2$.

C. Valence-band photoemission

A typical photoemission spectrum of $YbPd_2Si_2$ obtained at room temperature using synchrotron radiation at BESSY is shown in Fig. 4; it is recorded at a photon energy of 120 eV, where emission from Pd 4d states is strongly suppressed due to the well-known Cooper minimum in the photoelectric cross section.³¹ The PE spectrum in Fig. 4 exhibits the $4f^{13}$ and $4f^{12}$ final-state multiplet structures originating from divalent $(4f^{14})$ and trivalent $(4f^{13})$ initial ionic Yb states, respectively.¹⁸ The lowestbinding-energy components of the two multiplet structures are separated by a Coulomb correlation energy of $U=$ 5.9 \pm 0.05 eV.

As in Yb metal, $32,33$ YbA1₂, 18 and other RE systems, 19,36 the divalent multiplet structure close to E_F is a superposition of two sets of bulk- and surface-derive spin-orbit-split $4f^{13}$ doublets (spin-orbit splitting 1.29 eV), shifted against each other by a surface core-level shift Δ_c . In agreement with observations in other RE systems, the structure close to E_F is assigned to emission from bulk Yb^{2+} ions, the other one to surface emission.

FIG. 4. Photoemission spectrum of $YbPd_2Si_2$ taken at $h\nu= 120$ eV. The solid curve is obtained by a least-squares fit analysis of the data as described in the text. The subspectra of bulk Yb^{2+} and Yb^{3+} (dashed curve) and surface Yb^{2+} (dotted curve) obtained from the fitting, along with the background (dash-dotted), are also shown (transposed from the measured spectrum).

In the least-squares-fit analysis of the spectrum, the theoretical function was set equal to the sum of the $4f^{13}$ and $4f¹²$ multiplet components with Doniach-Sunijc line shapes plus an inelastic background; this function was convoluted with a Gaussian for experimental resolution. The relative intensities of the $4f^{12}$ multiplet components were taken from the intermediate-coupling calculations of Gerken,³⁴ which are adjusted to the relative separations of the $4f^{12}$ multiplet lines known from optical absorption spectroscopy.³⁵ The results of the fit analysis are presented in Fig. 4. The values of the binding energy relative to E_F of the most weakly bound ${}^2F_{7/2}$ -4 f^{13} multiplet components are determined as 0.2 eV for the bulk Yb^{2+} ions and 1.0 eV for the surface Yb^{2+} ions, respectively. This corresponds to a surface core-level shift of $\Delta_c = 0.80 \pm 0.05$ eV for YbPd₂Si₂, and this value of Δ_c is quite similar in magnitude to those observed for Yb $_{\text{metal}}$, 32,33 YbAl₂, 18 and YbZn₂.³⁶

The fact that the least-bound bulk $4f^{13}$ multiplet component is observed in close vicinity of E_F is consistent with the homogeneously mixed-valent state of Yb in YbPd₂Si₂, and follows from the degeneracy of the $4f^{14}$ and $4f^{13}$ (plus conduction electron) states in such systems. The small 0.2-eV shift of the least-bound $4f^{13}$ multiplet component has been observed also in other homogeneously mixed-valent systems, and may be due to incomplete final-state screening and/or lattice relaxation effects.^{18,19} The most striking feature in the PE spectrum of Fig. 4 is the large intensity of the surface $4f^{13}$ final-state doublet (dotted) as compared to the bulk peaks $(4f¹³)$ part of the dashed curve). Such a strong surface enhancement of the $4f¹³$ emission cannot be explained on the basis of known electron-escape depths l for hot electrons with about 115 eV kinetic energy alone. Rather, one has to assume that a surface layer of thickness Δs undergoes a valence transition to the divalent state. Such surface valence transitions have been observed for a number of bulk-trivalent RE systems, like Sm metal,³⁷ EuPd_x,³⁸ YbAu₂,³⁹ TmS,¹⁹ and $SmAl₂$ ³⁶ they seem to be the rule for homogeneously mixed-valent RE systems, as TmSe, Tm_{1-x} YSe,¹⁹ EuPd₂Si₂,¹⁷ and YbAl₂,¹⁸

In order to get a rough estimate of the thickness Δs of the divalent surface layer on $YbPd_2Si_2$, we assume identical PE cross sections for bulk and surface Yb atoms. This leads¹⁸ to $I/I_B = \exp(\Delta s/I)$, where I is the total intensity of 4f emission (including surface multiplets) and I_R stands for the intensity of bulk emission $(4f^{12})$ and bulk- $4f¹³$ multiplets). From a least-squares-fit analysis of the 120-eV spectrum of YbPd₂Si₂ we obtain I/I_B $=2.00\pm0.05$ and hence $\Delta s / I = 0.70\pm0.03$. Assuming $l \approx 6$ Å for photoelectrons with about 115-eV kinetic energy,⁴⁰ we obtain $\Delta s \approx 4.2$ Å. If we compare this value for Δs with the shortest Yb-Yb distance in YbPd₂Si₂ (4.1 \AA) we arrive at the conclusion that the outermost surface layer is divalent in $YbPd_2Si_2$.

We may derive a value of the mean valence of bulk Yb ions from the relative intensities of the Yb^{3+} and bulk- Yb^{2+} emission, taking the different numbers of 4f electrons in the initial state into account, and assuming a divalent surface layer on $YbPd_2Si_2$. The result is $\overline{v} = 2.75 \pm 0.1$, which agrees reasonably well with the susceptibility and L_{III} -edge results at 300 K.

In Yb metal, the bulk- and surface- $F_{7/2}$ components are observed 1.2 and 1.8 eV below E_F , respectively, ^{32, 33} whereas in $YbPd_2Si_2$, the corresponding values are 0.2 and 1.0 eV. Therefore, the chemical shift of the $2F_{7/2}$ -binding energy between Yb metal and YbPd₂Si₂ is 1.0 eV for bulk Yb^{2+} ions, but only 0.8 eV for Yb^{2+} ions at the surface. The origin for the smaller chemical shift of surface atoms lies in differences in the cohesive energy, due to the reduced coordination of surface atoms. In general, the cohesive energy of the surface is only about 80% of that of the bulk.⁴¹ If one compares a series of compounds of, say, a particular RE element, any difference in the chemical bonding will result in different cohesive energies and hence in chemical shifts for the bulk core-level binding energy. The surface core-level binding energy in such compounds will also be proportional to the surface cohesive energy of the compound. It is therefore natural—as observed in $YbPd_2Si_2$ —that the surface chemical shift of a compound is about 20% smaller than that of the bulk.

IV. CONCLUSION

On the basis of magnetic susceptibility measurements, L_{III} -edge XA, and valence-band PE studies, we have shown that $YbPd_2Si_2$ is a homogeneously mixed-valent compound. The room-temperature mean valence of Yb in $YbPd_2Si_2$, determined in the present study, is close to the high-temperature limit of \bar{v} for Yb in the ICF model.²² The valence transition observed for the outermost surface layer of $YbPd_2Si_2$ is in accord with observations on other homogeneously mixed-valent materials, $17-19$ and actually seems to be the rule for such materials with 4f instabilities. It is therefore striking that surface-induced changes in the 4f occupancy have not been identified in mixedvalent Ce comyounds.

Both the L_{III} -edge XA data and the analysis of the magnetic susceptibility reveal a weak decrease of \bar{v} by about 0.07 as the temperature is lowered from 300 to 16 K. This is quite different from the situation found for K. This is quite different from the situation found for $EuCu_2Si_2,^{12}$ and particularly $EuPd_2Si_2,^{11-15}$ where strong valence changes as a function of temperature have been observed. In EuPd₂Si₂ the precipitous valence transition was explained in terms of cooperative effects arising from was explained in terms of cooperative effects arising from
ntersite coupling between the Eu ions.^{13,15} In homogeneously mixed-valent Yb systems, however, one should bear in mind that the lowering of the temperature causes two opposite effects on the lattice volume: a shrinkage due to the usual anharmonic lattice contraction and an expansive effect due to the increased population of the larger Yb^{2+} ionic states at lower temperatures. This may cause very unusual lattice expansion properties as actually observed for Yb systems.⁴³ In this way, Yb-Yb intersite-coupling effects—even if present in YbPd₂Si₂—may not strongly influence the temperature dependence of \bar{v} . The question of cooperative versus single-ion aspects of the valence changes in $YbPd_2Si_2$ can be addressed further by investigating the dilute system $Yb_{1-x}Lu_xPd_2Si_2$.

ACKNOWLEDGMENTS

The authors acknowledge valuable discussions with C. Laubschat as well as the assistance of W. Wisny in the magnetic susceptibility measurements. The synchrotronradiation measurements would not have been possible without the technical assistance and cooperation of W. Malzfeldt and W. Niemann at HASYLAB (DESY Hamburg), and A. Puschmann and J. Haase at BESSY (Berlin). One of the authors (E.V.S.) is grateful for support from the Alexander von Humboldt Foundation. This work was supported by the Bundesministerium fiir Forschung und Technologie, Contracts Nos. 05-241-KA and 05-256-KA.

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- $1Z.$ Ban and M. Sikirica, Acta Crystallogr. $18, 594$ (1965).
- ²D. Rossi, R. Marazza, and R. Ferro, J. Less-Common Met. 55, 17 (1979), and references therein.
- 3 Valence Instabilities and Related Narrow Band Phenomena, edited by R. D. Parks (Plenum, New York, 1977), and references therein.
- Valence Fluctuations in Solids, edited by L. M. Falicov, W. Hanke, and M. B. Maple (North-Holland, Amsterdam, 1981), and references therein.
- 5Valence Instabilities, edited by P. Wachter and H. Boppart (North-Holland, Amsterdam, 1982), and references therein.
- F. Steglich, J. Aarts, C. D. Bredl, W. Lieke, D. Meschede, W. Franz, and H. Schafer, Phys. Rev. Lett. 43, 1892 (1979).
- 7G. W. Hull, J. H. Wernick, T. H. Geballe, J. V. Waszczak, and J. E. Bernardini, Phys. Rev. B 24, 6715 (1981).
- 8 H. F. Braun, N. Engel, and E. Parthe, Phys. Rev. B 28, 1389 (1983).
- ⁹K. Hiebel, C. Horvath, P. Rolg, and M. J. Sienko, J. Magn. Magn. Mater. 37, 287 (1983) and Solid State Commun. 48,

211 (1983).

- ¹⁰I. Felner and I. Nowik, Solid State Commun. 47, 831 (1983).
- ¹¹E. V. Sampathkumaran, L. C. Gupta, R. Vijayaraghavan, K. V. Gopalakrishnan, R. G. Pillay, and H. G. Devare, J. Phys. C 14, L237 (1981).
- E. V. Sampathkumaran, R. Vijayaraghavan, K. V. Gopalakrishnan, R. G. Pillay, H. G. Devare, L. C. Gupta, B. Post, and R. D. Parks, in Valence Fluctuations in Solids, Ref. 4, p. 193.
- 13M. Croft, J. A. Hodges, E. Kemly, A. Krishnan, V. Murgai, and L. C. Gupta, Phys. Rev. Lett. 48, 826 (1982).
- ¹⁴G. Schmiester, B. Perscheid, G. Kaindl, and J. Zukrowsky, in Valence Instabilities, Ref. 5, p. 219.
- ¹⁵R. G. Pillay, E. V. Sampathkumaran, H. G. Devare, L. C. Gupta, and R. Vijayaraghavan, Proceedings of the Interna tional Conference on Mössbauer Effects (Indian National Science Academy, India, 1982), p. 592.
- 16C. N. R. Rao, D. D. Sarma, P. R. Sarode, E. V. Sampathkumaran, L. C. Gupta, and R. Vijayaraghavan, Chem. Phys. Lett. 76, 413 (1980).
- ¹⁷W. D. Schneider, N. Mårtensson, B. Reihl, V. Murgai, L. C. Gupta, and R. D. Parks, in Valence Instabilities, Ref. 5, p. 295.
- G. Kaindl, B. Reihl, D. E. Eastman, R. A. Pollak, N. Martensson, B. Barbara, T. Penney, and T. S. Plaskett, Solid State Commun. 41, 157 (1982).
- G. Kaindl, C. Laubschat, B. Reihl, R. A. Pollak, N. Martensson, F. Holtzberg, and D. E. Eastman, Phys. Rev. B 26, 1713 (1982).
- 20B. Reihl, G. Kaindl, F. Holtzberg, R. A. Pollak, G. Hollinger, and N. Martensson, in Valence Instabilities, Ref. 5, p. 28.
- ²¹J. K. Yakinthos and H. Gamari-Seale, Z. Phys. B 48, 251 (1982).
- $22B$. C. Sales and D. K. Wohlleben, Phys. Rev. Lett. 35 , 1240 (1975).
- ²³J. W. Lawrence and M. T. Beal-Monod, in *Valence Fluctua*tions in Solids, Ref. 4, p. 5.
- 24 M. T. Beal-Monod and J. M. Lawrence, Phys. Rev. B $21,5400$ (1980).
- 25J. W. Lawrence, P. S. Riseborough, and R. D. Parks, Rep. Prog. Phys. 44, 1 (1981), and references therein.
- W. Franz, F. Steglich, W. Zell, D. Wohlleben, and F. Pobell, Phys. Rev. Lett. 45, 64 (1980).
- ²⁷D. M. Newns and H. C. Hewson, in *Valence Fluctuations in* Solids, Ref. 4, p. 27.
- ²⁸J. F. Herbst, R. E. Watson, and J. W. Wilkins, Phys. Rev. B 17, 3089 (1978).
- ²⁹K. Syassen, G. Wortmann, J. Feldhaus, K. H. Frank, and G.

Kaindl, Phys. Rev. B 26, 4745 (1982).

- ³⁰K. R. Bauchspiess, W. Boksch, E. Holland-Moritz, H. Launois, R. Pott, and D. Wohlleben, in Valence Fluctuations in Solids, Ref. 4, p. 417.
- ³¹See, e.g., S. M. Goldberg, C. S. Fadley, and S. Kono, J. Electron Spectros. Relat. Phenom. 21, 285 (1981).
- 32S. F. Alvarado, M. Campagna, and W. Gudat, J. Electron Spectros. Relat. Phenom. 18, 43 (1980).
- W. D. Schneider, C. Laubschat, and B. Reihl, Phys. Rev. B 27, 6538 (1983).
- 34F. Gerken, J. Phys. F 13, 703 (1983).
- ~5W. T. Carnall, P. R. Fields, and K. Rajnak, J. Chem. Phys. 49, 4424 (1968).
- G. Kaindl, W. D. Schneider, C. Laubschat, B. Reihl, and N. Mårtensson, Surf. Sci. 126, 105 (1983).
- ³⁷G. K. Wertheim and G. Crecelius, Phys. Rev. Lett. 40, 813 (1978).
- 38V. Murgai, L. C. Gupta, R. D. Parks, N. Mårtensson, and B. Reihl, in Valence Instabilities, Ref. 5, p. 299.
- ³⁹G. K. Wertheim, J. H. Wernick, and G. Crecelius, Phys. Rev. B 18, 875 (1978).
- 40 B. Feuerbacher and R. F. Willis, J. Phys. C $\overline{2}$, 169 (1976).
- 41 B. Johansson, Phys. Rev. B 19 , 6615 (1979); B. Johansson and N. Mårtensson, *ibid.* 21, 4427 (1980), and references therein.
- ⁴²E. R. Bauminger, D. Froindlich, I. Nowik, S. Ofer, I. Felner, and I. Meyer, Phys. Rev. Lett. 30, 1053 (1973).
- 43D. K. Wohlleben, in Valence Fluctuations in Solids, Ref. 4, p. 1.