## Magnetic and crystallographic structure of $Y_6Mn_{23}D_{23}$

Kay Hardman-Rhyne, J. J. Rhyne, and E. Prince

National Measurement Laboratory, National Bureau of Standards, Washington, D.C. 20234

C. Crowder\* and W. J. James

Graduate Center for Materials Research, University of Missouri-Rolla, Rolla, Missouri 65401

(Received 6 June 1983)

The magnetic behavior of  $Y_6Mn_{23}$  is dramatically altered upon hydrogenation (or deuteration). In this study it has been found, by means of high-resolution powder diffraction and Rietveld refinement techniques, that the crystallographic structure is distorted from face-centered cubic (*Fm3m*) at 295 K to a primitive tetragonal structure at 4 K in which deuterium atoms are atomically ordered.  $Y_6Mn_{23}$  is a ferromagnetic compound with  $T_c = 486$  K, and bulk magnetization of  $13.2\mu_B/f.u.$  (formula unit). After deuteration of  $Y_6Mn_{23}$  to the composition  $Y_6Mn_{23}D_{23}$ , low-temperature scattering data (T < 180 K) show that the *b* and  $f_2$  sites in the *Fm3m* structure are antiferromagnetic and the *d* and  $f_1$  sites have no spontaneous magnetic moment.

#### INTRODUCTION

Y<sub>6</sub>Mn<sub>23</sub> is a face-centered-cubic (fcc) compound isostructural with Th<sub>6</sub>Mn<sub>23</sub>, which has the space group Fm3m. The magnetic and crystallographic properties of these materials are greatly altered by hydrogenation (deuteration).  $Th_6Mn_{23}$  is a Pauli paramagnet, while the deuteride Th<sub>6</sub>Mn<sub>23</sub>D<sub>30</sub> is almost ferromagnetic (one Mn spin coupled antiparallel to the other 22 parallel Mn spins), with  $T_c = 355$  K, and a bulk magnetization value of 18.4 $\mu_B$ /f.u. (formula unit).<sup>1,2</sup> In contrast, from bulk studies, Y<sub>6</sub>Mn<sub>23</sub>D<sub>23</sub> exhibits no net spontaneous magnetization,<sup>3,4</sup> while  $Y_6Mn_{23}$  is a ferromagnetic compound with  $T_c = 486$  K and a bulk magnetization of  $13.2\mu_B/f.u.^5$  The magnetic structure of Y<sub>6</sub>Mn<sub>23</sub> is collinear, with spins along the [111] direction with the b site (one Mn atom), and the d site (six Mn atoms) coupled antiparallel to the two f sites (eight Mn atoms each). All magnetically ordered spins within each of the sites are coupled in the same direction. The interatomic distances of the Mn atoms vary significantly between different sites which partly accounts for the wide variations of Mn magnetic moments observed in the magnetically ordered compounds of this class. The magnetic properties of these hydrides are also strongly affected by changes in the electronic band structure.

 $R_6 Mn_{23} H_x$  compounds are of interest in developing models for predicting the potential occupancy and location of hydrogen in rare-earth intermetallic compounds.<sup>6,7</sup> These compounds are very complex (e.g.,  $Y_6 Mn_{23} D_{23}$  has four formula units or 208 total atoms per unit cell). A geometric model developed by Westlake shows nine different sites are possible for the D atoms, but in many of the combinations not all the sites can be simultaneously occupied.<sup>6</sup>

Neutron diffraction has proven to be a valuable tool in understanding the magnetism and the deuterium occupancy and location in these compounds. In this paper we will discuss the changes occurring in both the crystallographic structure and the magnetic ordering of  $Y_6Mn_{23}D_{23}$  at low temperatures.

## EXPERIMENTAL OBSERVATIONS

An ingot of  $Y_6Mn_{23}$  was prepared by induction melting of 99.9% pure elements in a water-cooled Cu boat under an argon atmosphere. X-ray diffraction photographs showed that the specimen was single phase. The ingot was deuterated with 99.5% pure deuterium in a stainlesssteel vessel at 20 atm and 110 °C for 2 h. Pressure-volume measurements indicated approximately 23 atoms of deuterium absorbed per formula unit. Deuterium was used in lieu of hydrogen because hydrogen has a large incoherent scattering cross section for neutrons. Figure 1 shows a deuterium absorption isotherm for Y<sub>6</sub>Mn<sub>23</sub>, demonstrating



FIG. 1. Absorption isotherm for deuterium in  $Y_6Mn_{23}$ . Note that a plateau does not exist in this range (see Ref. 8).

Atom	Site	N/2 f.u. (max)	x	у	Z	В	
Y	е	12 (12)	0.205	0	0	0.82	
Mn	b	2 (2)	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	0.47	
Mn	d	12 (12)	0	$\frac{1}{4}$	$\frac{1}{4}$	0.47	
Mn	$f_1$	16 (16)	0.179	0.179	0.179	0.47	
Mn	$f_2$	16 (16)	0.372	0.372	0.372	0.47	
D	а	2 (2)	0	0	0	6.35	
D	$f_3$	10 (16)	0.100	0.100	0.100	1.37	
D	$j_1$	20 (48)	0	0.169	0.373	4.95	
D	$k_1$	14 (48)	0.161	0.161	0.049	2.28	
$a_0 = 12.805$ Å Z = 4		$R_{wp} = 10.64$ $R_e = 6.29$	$R_{wp} = 10.64$ $R_e = 6.29$		χ=1.69		

TABLE I. Atomic parameters for  $Y_6Mn_{23}D_{23}$  at 295 K in the fcc (Fm 3m) structure. Refined occupancy is N (maximum occupancy of that site), position is (x, y, z), and B is the Debye-Waller temperature factor.

the low pressures required to obtain deuteration up to 24 atoms per formula unit.<sup>8</sup> No absorption plateau is found in this region. This is in sharp contrast to  $Th_6Mn_{23}D_x$ ,<sup>2</sup> which has a broad absorption plateau from 17 to 24 atoms of deuterium.

 $Y_6Mn_{23}$  samples were taken at the National Bureau of Standards Reactor utilizing a high-resolution five-detector powder diffractometer. Neutrons of wavelength 1.542 Å were obtained from a (220) reflection copper monochromator, with a graphite filter to eliminate second-order contamination. Soller slit collimators of angular diver-

Neutron-diffraction data on pure and deuterated



FIG. 2. Neutron-diffraction data on  $Y_6Mn_{23}D_{23}$  at 295 K which is above the crystallographic phase transition and magnetic ordering temperature. The top half of the figure shows data and residuals from the first two of the five detectors (note the difference in scale of intensities). The solid curve is the calculated profile fit to the data and the vertical lines represent the calculated relative intensity and position of the nuclear reflections.

gences 10'-20'-10' were placed before and after the monochromator and in front of the detector, respectively.

The powder data from the five detectors, displaced in angle by 20° from each other, covered the  $2\theta$  range from 10° to 115° in increments of 0.05° and were analyzed by a modified Rietveld total pattern refinement technique which allowed background parameters to be refined in addition to the other structural and instrumental variables.<sup>9</sup> Standard deviations were generally less than  $\pm 2$  in the last digit listed, except for the magnetic moments which were  $\pm 0.1\mu_B/$ atom. The standard deviations of the thermal factors (B) for deuterium atoms in some cases were quite large (order of 30%).

#### **RESULTS AND DISCUSSION**

#### Y<sub>6</sub>Mn<sub>23</sub>D<sub>23</sub> at 295 K

At 295 K,  $Y_6Mn_{23}D_{23}$  has no long-range magnetic ordering and is face-centered cubic, with space group  $Fm 3m(O_b^5)$ . Although previous experiments have suggested that it has a body-centered-tetragonal (bct) structure with space group  $I4/mmm(D_{4h}^{17})$ ,<sup>4</sup> the present studies conducted with higher-instrumental resolution and over a wider scattering angle range do not indicate a departure from the fcc symmetry. No peaks were found which could not be indexed on the Fm 3m cell, nor was there appreciable broadening or splitting of the peaks compared with the nondeuterated  $Y_6Mn_{23}$ . The structural parameters derived from the refinement are given in Table I. The deuterium atoms are found to prefer occupying the a,  $f_3$ ,  $j_1$ , and  $k_1$  sites. These results are in basic agreement with Westlake's geometric model,<sup>6</sup> but in significantly less agreement with the heat of formation model of Jacob<sup>7</sup> which does not include the possibility of deuterium in a  $j_1$  site. Both Jacob's and Westlake's models suggested the possibility of deuterium atoms located in the a,  $f_3$ , i, and  $j_2$  sites. Westlake also considered the h and l sites in his model. Our results found no deuterium atoms in the i, h,  $j_2$ , or l sites.

Based on the geometric model<sup>6</sup> criteria that hydrogen atoms cannot be closer together than 2.1 Å, the  $j_1$  site cannot have more than 12 deuterium atoms per formula unit or 48 deuterium atoms per unit cell. Our results (Table I) show ten D/f.u. present in the  $j_1$  site. The model also predicts that D atoms cannot occupy the  $k_1$  site unless some deuterium atoms are removed from the  $f_3$  site. This was indeed found to be the case in this study of  $Y_6Mn_{23}D_{23}$ , in contrast to that of Commandre et al.<sup>3</sup> They reported the  $f_3$  site filled (eight D/f.u.), and four D atoms in the  $k_1$  site, whereas we have five D atoms in the  $f_3$  site, and seven D atoms in the  $k_1$  site per formula unit. On careful examination of the positions of the deuterium atoms in the  $f_3$  and  $k_1$  sites with the use of the Westlake geometric model, we expect that a maximum number of 20 deuterium atoms could exist in the  $k_1$  site if no D atoms were present in the  $f_3$  site. Furthermore, one deuterium atom removed from the  $f_3$  site allows three D atoms to be added to the  $k_1$  site. Thus if there are five D



FIG. 3. Neutron-diffraction data and profile fit of  $Y_6Mn_{23}D_{23}$  at 4 K. Additional peaks result from antiferromagnetism as well as the structural distortion, which is best fit by the primitive tetragonal space group P4/mmm.

atoms in the  $f_3$  site, no more than nine D atoms can be present in the  $k_1$  site. The composition  $Y_6Mn_{23}D_{25}$ , as found by Malik *et al.*,<sup>1</sup> could then be obtained by having one D atom in the *a* site, ten D atoms in the  $j_1$  site, five D atoms in the  $f_3$  site, and nine D atoms in the  $k_1$  site; or by assigning twelve D atoms to the  $j_1$  site, five D atoms to the  $f_3$  site, and seven D atoms to the  $k_1$  site. However, if more than 25 total D atoms are present [e.g., Th<sub>6</sub>Mn<sub>23</sub>D<sub>30</sub> (Ref. 2)] then more D atoms may occupy the k site.

The temperature factors *B* for deuterium in the *a* and  $j_1$  sites are quite large. For deuterium located in the *a* site, the high thermal factor suggests little or no bonding of deuterium with the six Y atoms surrounding it. This volume is very large, and the deuterium atom is probably exhibiting large amplitude excursions. An alternate interpretation is that the *a* site is not actually occupied by the deuterium.<sup>10</sup> The high thermal factor for deuterium atoms in the  $j_1$  site probably arises from an averaging effect of D atoms disordered between two  $j_1$  sites that are only 0.90 Å from one another. The diffraction pattern, with the accompanying calculated profile fit, is shown in Fig. 2 for Y<sub>6</sub>Mn<sub>23</sub>D<sub>23</sub> at 295 K.

# Crystallographic structure of $Y_6Mn_{23}D_{23}$ at 78 and 4 K

A crystallographic distortion is seen from the 4-K diffraction pattern in Fig. 3. A similar structural distortion was also found in  $Th_6Mn_{23}D_{16}$  at 4 K (see Fig. 5 in Ref. 2). In the cases of  $Th_6Mn_{23}D_{16}$  at 4 K and  $Y_6Mn_{23}D_{23}$  at 78 and 4 K, the resulting structure is based on a primitive tetragonal cell. The lattice parameters of this cell are the same for  $Y_6Mn_{23}D_{23}$  at 4 and 78 K, and are  $a_0 = 9.030$  Å and  $c_0 = 12.722$  Å, giving a ratio of c to  $\sqrt{2a}$  of 0.996. The  $c_0$  value obtained for Y<sub>6</sub>Mn<sub>23</sub>D<sub>23</sub> at 78 K from a fcc structure refinement is 12.766 Å. An enlargement of a limited high-angle range of the diffraction pattern at both 295 and 4 K, with the fcc profile fit applied to both, is shown in Fig. 4. The calculated curve at 4 K has been shifted to account for the decrease in the lattice parameter with temperature. In the 4-K data at  $2\theta = 91.0^\circ$ , the (11,3,3) and (9,7,3) reflections have split and the intensities have changed. In the fcc lattice at 295 K there is only one peak in this region.

Of several possible space groups the one that provides the best fit to the nuclear structure of  $Y_6Mn_{23}D_{23}$  at low temperatures (T < 180 K) is  $P4/mmm(D_{4h}^{1})$ . As given in Table II, this structure splits the single Y e site into four sites (g, h, j, and k) and the Mn b site into b and c sites. The Mn d site is split into e, f, and  $r_1$  sites. The Mn f site and the D f site are split into s and t sites. The D asite is split into a and d sites, and the j site is split into p, q, and four r sites. Only half of the original j site can be filled. Therefore the p and q sites cannot each be more than half-full, and only two r sites (or no more than half of all four r sites) can be filled. Our results, given in Table III, show that the p site is half-full, the q site is almost half-full, and the  $r_2$  and  $r_4$  sites are almost completely filled. There are no D atoms in the  $r_3$  and  $r_5$  sites. The k site splits into  $s_4$ ,  $t_4$ , and two u sites. As mentioned before, no more than nine D atoms per formula unit can



FIG. 4. Expanded high-angle data for  $Y_6Mn_{23}D_{23}$  at 295 and 4 K, showing the peak broadening accompanying the tetragonal distortion below the crystal-structure transition temperature. Solid lines in both cases are the profile fit for the 295 K fcc data. Shift in the solid line represents the lattice contraction that has occurred at low temperatures.



FIG. 5. Magnetic susceptibility for  $Y_6Mn_{23}D_{23}$  (from Ref. 4).

fill these sites. Our results (Table III) show the  $t_4$  site to be empty of D atoms. To illustrate the small distortion in the unit cell, the fcc equivalent positions are given for the tetragonal cell in Table II. Modest shifts in positions (x,y,z) can be seen from the fcc equivalent positions (Table III) for some sites. The crystal-structure transformation at low temperatures may result from preferential occupation by the D atoms among the many sites that are available. In particular, the fact that the  $r_3$ ,  $r_5$ , and  $t_4$  sites are empty of D atoms could lead to a crystallographic transition.

As previously mentioned, a bct structure with space group I4/mmm has been suggested<sup>4</sup> for  $Y_6Mn_{23}D_{23}$  at both room temperature and 78 K for the nuclear structure only. However, the present higher-resolution data at both 78 and 4 K show several structural details which clearly make the P4/mmm symmetry the better choice. This includes several predominantely nuclear peaks found at high angles (e.g., at  $2\theta$  about 56.3° and in the range  $2\theta$  from 69.4° to 69.9° and 84.9° to 85.6°) which are forbidden by the I4/mmm symmetry but are allowed and well represented by the P4/mmm structure. In addition, the unresolved asymmetric peak at 90.5° (Fig. 4) which arises from several reflections is fit considerably better by the composite structure factor of the primitive space group.

### Magnetic structure of $Y_6Mn_{23}D_{23}$ at 78 and 4 K

Malik *et al.*,<sup>1</sup> Commandre *et al.*,<sup>3</sup> Buschow,<sup>11</sup> and Crowder *et al.*<sup>4</sup> have discussed the drastic reduction and ultimate disappearance of magnetization of  $Y_6Mn_{23}$  upon hydrogenation or deuteration. There has been much speculation on these effects, with suggestions of either a spin glass or an antiferromagnetic state in  $Y_6Mn_{23}D_{23}$  and  $Y_6Mn_{23}D_{25}$ . A plot of the magnetic susceptibility of  $Y_6Mn_{23}D_{23}$  versus temperature<sup>12</sup> in Fig. 5 shows a broad anomaly suggestive of a Néel temperature  $T_N$ , approxi-

TABLE II. Equivalent positions in two crystallographic space groups. Z=4 for the Fm3m structure, and Z=2 for the P4/mmm structure.

Face-centered cubic (Fm 3m)					Primitive tetragonal (P4/mmm)					
atom	site	x, y, z	N/2 f.u.	site	<i>x</i> , <i>y</i> , <i>z</i>	x	у	Z	N/2 f.u.	
Y	е	x,0,0	12	g	0,0, <i>z</i>			0.205	2	
		x = 0.205		h	$\frac{1}{2}, \frac{1}{2}, z$			0.295	2	
				j	<i>x</i> , <i>x</i> ,0	0.205			4	
				k	$x, x, \frac{1}{2}$	0.295			4	
Mn	b	$\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$	2	b	$0, 0, \frac{1}{2}$				1	
				с	$\frac{1}{2}, \frac{1}{2}, 0$				1	
Mn	d	$0, \frac{1}{4}, \frac{1}{4}$	12	e	$0, \frac{1}{2}, \frac{1}{2}$				2	
				f	$0, \frac{1}{2}, 0$				2	
				<i>r</i> <sub>1</sub>	x, x, z	0.25		0.75	8	
Mn	$f_1$	<i>x</i> , <i>x</i> , <i>x</i>	16	<i>s</i> <sub>1</sub>	x,0,z	0.358		0.179	8	
		x = 0.179		$t_1$	$x, \frac{1}{2}, z$	0.142		0.321	8	
Mn	$f_2$	x, x, x	16	<i>s</i> <sub>2</sub>	x,0,z	0.256		0.372	8	
		x = 0.372		$t_2$	$x, \frac{1}{2}, z$	0.244		0.128	8	
D	а	0,0,0	2	a	0,0,0				1	
				d	$\frac{1}{2}, \frac{1}{2}, \frac{1}{2}, \frac{1}{2},$				1	
D	$f_3$	x, x, x	16	<b>s</b> <sub>3</sub>	x,0,z	0.200		0.100	8	
		x = 0.100		$t_3$	$x, \frac{1}{2}, z$	0.300		0.400	8	
D	$j_1$	0, <i>y</i> , <i>z</i>	48	р	<i>x</i> , <i>y</i> ,0	0.204	0.458		8	
		y = 0.169		$\boldsymbol{q}$	$x, y, \frac{1}{2}$	0.296	0.042		8	
		z = 0.373		<i>r</i> <sub>2</sub>	x, x, z	0.127		0.331	8	
				<b>r</b> <sub>3</sub>	x, x, z	0.169		0.373	8	
				r <sub>4</sub>	x, x, z	0.331		0.127	8	
				<i>r</i> <sub>5</sub>	x, x, z	0.373		0.169	8	
D	$k_1$	x, x, z	48	<i>s</i> <sub>4</sub>	x,0,z	0.322		0.049	8	
		x = 0.161		$t_4$	$x, \frac{1}{2}, z$	0.178		0.451	8	
		z = 0.049		<b>u</b> <sub>1</sub>	x, y, z	0.112	0.210	0.161	16	
				$u_2$	x, y, z	0.290	0.388	0.339	16	

mately 170 K. This value is in good agreement with that found by Stewart *et al.*<sup>13</sup> ( $T_N = 180$  K) by Mössbauer studies on Y<sub>6</sub>( ${}^{57}$ Fe<sub>0.005</sub>Mn<sub>0.995</sub>)<sub>23</sub>H<sub>25</sub>. They found antiferromagnetic ordering of the Mn sites in this substituted compound.

Our neutron-diffraction measurements show new peaks emerging in the  $2\theta$  range from 20.0° to 21.5° at temperatures between 165 and 178 K. Analysis of the Y<sub>6</sub>Mn<sub>23</sub>D<sub>23</sub> data at 78 and 4 K reveals a weak antiferromagnetic structure with only two sites participating in the longrange order. The magnetic moments of the Mn atoms are given in Table IV, and involve only the b and  $f_2$  sites of the original fcc structure. The Mn moments are along the c axis. The splitting of the f sites allows the Mn atoms to be divided among two sites (s and t) with antiparallel moments, which represents antiferromagnetism on the equivalent fcc site. Surprisingly, the d and  $f_1$  sites of the fcc structure appear to have no magnetic moment. If any magnetic moment is forced onto these sites they prefer to couple ferromagnetically. However, the best profile fits were those with no magnetic moments on the d and  $f_1$  sites. Larger moments on the  $f_2$  (really  $s_2$  and  $t_2$  in the P4/mmm structure) sites were found for  $Y_6Mn_{23}D_{23}$  at 78 K rather than at 4 K. This corresponds to the broad minimum observed around 80 K in the magnetic susceptibility (Fig. 5).

#### CONCLUSIONS

The deuterium occupancy and location has been determined for  $Y_6Mn_{23}D_{23}$  at 295 K where the crystal structure is face-centered cubic with space group Fm3m. The deuterium atoms fill the *a* site and partially fill the  $f_3$ ,  $j_1$ , and  $k_1$  sites, which is in relatively good agreement with the Westlake model for  $A_6B_{23}H_x$  compounds.

There is a structural and accompanying magnetic tran-

TABLE III. Atomic parameters of Y<sub>6</sub>Mn<sub>23</sub>D<sub>23</sub> at 78 and 4 K in the P4/mmm tetragonal structure. All x,y,z, and B values were refined except the x,y,z values that are 0 or  $\frac{1}{2}$ . All deuterium occupancy values were also refined.

		78 K						4 K		
atom	site	N (max)	x	У	Z	В	x	У	Z	В
Y	g	2 (2)	0	0	0.208	0.76	0	0	0.213	0.72
	h	2 (2)	$\frac{1}{2}$	$\frac{1}{2}$	0.287	0.76	$\frac{1}{2}$	$\frac{1}{2}$	0.285	0.72
	j	4 (4)	0.199	0.199	0	0.76	0.204	0.204	0	0.72
	k	4 (4)	0.291	0.291	$\frac{1}{2}$	0.76	0.297	0.297	$\frac{1}{2}$	0.72
Mn	b	1 (1)	0	0	$\frac{1}{2}$	0.25	0	0	$\frac{1}{2}$	0.01
	с	1 (1)	$\frac{1}{2}$	$\frac{1}{2}$	0	0.25	$\frac{1}{2}$	$\frac{1}{2}$	0	0.01
Mn	е	2 (2)	0	$\frac{1}{2}$	$\frac{1}{2}$	0.25	0	$\frac{1}{2}$	$\frac{1}{2}$	0.01
	f	2 (2)	0	$\frac{1}{2}$	0	0.25	0	$\frac{1}{2}$	0	0.01
	$\boldsymbol{r}_1$	8 (8)	0.259	0.259	0.748	0.25	0.255	0.255	0.747	0.01
Mn	<i>s</i> <sub>1</sub>	8 (8)	0.360	0	0.177	0.25	0.360	0	0.182	0.01
	$t_1$	8 (8)	0.135	$\frac{1}{2}$	0.318	0.25	0.145	$\frac{1}{2}$	0.317	0.01
Mn	<i>s</i> <sub>2</sub>	8 (8)	0.277	0	0.379	0.25	0.275	0	0.387	0.01
	$t_2$	8 (8)	0.250	$\frac{1}{2}$	0.129	0.25	0.249	$\frac{1}{2}$	0.137	0.01
D	a	1 (1)	0	0	0	6.12	0	0	0	7.80
	d	1 (1)	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	5.16	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	7.90
D	<b>s</b> <sub>3</sub>	4.7(8)	0.217	0	0.095	1.21	0.219	0	0.093	0.25
	$t_3$	5.3(8)	0.305	$\frac{1}{2}$	0.398	0.02	0.305	$\frac{1}{2}$	0.398	0.02
D	р	4 (8)	0.204	0.464	0	4.55	0.230	0.438	0	2.30
	q	3 (8)	0.291	0.072	$\frac{1}{2}$	1.48	0.304	0.045	$\frac{1}{2}$	1.80
	<b>r</b> <sub>2</sub>	6 (8)	0.138	0.138	0.338	3.04	0.141	0.141	0.331	2.83
	<b>r</b> <sub>3</sub>	(8)								
	<i>r</i> <sub>4</sub>	7 (8)	0.351	0.351	0.131	4.35	0.360	0.360	0.132	2.40
_	<b>r</b> <sub>5</sub>	(8)								
D	<i>S</i> <sub>4</sub>	2.5(8)	0.322	0	0.077	0.40	0.324	0	0.041	1.08
	$t_4$	(8)	0.072	0.174	0.144	2.20	0.044	0.155		
	<b>u</b> <sub>1</sub>	3.5(16)	0.073	0.174	0.166	3.30	0.066	0.177	0.141	2.67
	<i>u</i> <sub>2</sub>	8 (10)	0.299	0.385	0.333	0.16	0.296	0.392	0.327	0.20

P4/mmm structure						Fm3m structure				
	$Y_6Mn_{23}D_{23}$	78 K	4 K	N/2 f.u.	$Y_6$	Mn <sub>23</sub>	4 K	N/2 f.u.		
Mn	b	-3.60	-3.68	1	Mn	b	- 3.05	2		
	С	+3.60	+3.68	1						
Mn	е	0	0	2	Mn	d	-2.34	12		
	f	0	0	2						
	$r_1$	0	0	8						
Mn	<i>s</i> <sub>1</sub>	0	0	8	Mn	${f}_1$	+1.99	16		
	$t_1$	0	0	8						
Mn	<i>s</i> <sub>2</sub>	+2.08	+1.60	8	Mn	$f_2$	+1.80	16		
	$t_2$	-2.08	-1.60	8						
	$a_0$	9.031	9.030			12.402				
	$c_0$	12.723	12.722							
	Z = 2					Z = 4				
	$R_{wp}$	9.97	8.64			8.38				
	$R_e^{-1}$	6.06	5.81			4.34				
	$\chi = R_{wp}/R_e$	1.64	1.49			1.93				

TABLE IV. Magnetization values in  $\mu_B/Mn$  atom for Y<sub>6</sub>Mn<sub>23</sub> and Y<sub>6</sub>Mn<sub>23</sub>D<sub>23</sub>. Negative and positive signs indicate directions of the magnetic moments of the Mn atoms.

sition occurring around 175 K in  $Y_6Mn_{23}D_{23}$ . The crystal structure transforms into a primitive tetragonal structure which can be refined in the space group P4/mmm. This structure transformation probably results from the atomic ordering of the deuterium atoms at low temperatures and drives the c axis antiferromagnetic ordering on the Mn sites.  $Y_6Mn_{23}D_{23}$  is only weakly antiferromagnetic with the Mn atoms on the b and  $f_2$  sites having finite magnetic moments. In the P4/mmm structure, this corresponds to the Mn moments on the b and  $t_2$  sites coupled antiparallel to those on the c and  $s_2$  sites. No ordered moment was found on the Mn d and  $f_1$  sites in the Fm 3m structure.

-----

The comparison between these  $Y_6Mn_{23}D_x$  and the  $Th_6Mn_{23}D_x$  (Ref. 2) compounds suggests that the Mn-Mn atomic distances do not appear to play a dominant role in

determining the overall occurrence of magnetic order in these systems. Instead, deuterium composition and accompanying electronic band effects remain the important factors influencing the magnetic properties.

#### ACKNOWLEDGMENTS

The authors wish to thank A. Santoro for many discussions on crystal symmetry and possible crystal structures. B. Kebe performed the magnetization studies shown in Fig. 5, and H. Kevin Smith provided the deuterium absorption isotherm for  $Y_6Mn_{23}$ . Support from the U.S. Department of the Army under Grant No. DAAG-29-80-C-0084 for the work at the University of Missouri is gratefully acknowledged.

- \*Present address: Dow Chemical Company, Midland, MI 48640.
- <sup>1</sup>S. K. Malik, T. Takeshita, and W. E. Wallace, Solid State Commun. <u>23</u>, 599 (1977).
- <sup>2</sup>K. Hardman-Rhyne, H. K. Smith, and W. E. Wallace, J. Less-Common Met. (in press).
- <sup>3</sup>M. Commandre, D. Fruchart, A. Rouault, D. Sauvage, C. B. Shoemaker, and D. P. Shoemaker, J. Phys. (Paris) Lett. <u>40</u>, L639 (1979).
- <sup>4</sup>C. Crowder, B. Kebe, W. J. James, and W. B. Yellon, in *The Rare Earths in Modern Science and Technology*, edited by G. J. McCarthy, H. B. Silber, and J. J. Rhyne (Plenum, New York, 1982), Vol. III, p. 473.
- <sup>5</sup>A. Delapalme, J. Deportes, R. Lemaire, K. Hardman, and W. J. James, J. Appl. Phys. <u>50</u>, 1987 (1979).
- <sup>6</sup>D. G. Westlake, J. Mater. Sci. <u>18</u>, 605 (1983); Scr. Metall. <u>16</u>,

1049 (1982).

- <sup>7</sup>I. Jacob, Solid State Commun. <u>40</u>, 1015 (1981).
- <sup>8</sup>H. Kevin Smith (private communication). The deuterium absorption isotherm was obtained at the University of Pittsburgh, Pittsburgh, PA 15260.
- <sup>9</sup>R. A. Young, E. Prince, and R. A. Sparks, J. Appl. Crystallogr. <u>15</u>, 357 (1982).
- <sup>10</sup>J. J. Rhyne, Kay Hardman-Rhyne, H. Kevin Smith, and W. E. Wallace, J. Less-Common Met. <u>94</u>, 95 (1983).
- <sup>11</sup>K. H. J. Buschow, Solid State Commun. <u>21</u>, 1031 (1977).
- <sup>12</sup>The magnetic susceptibility is from Ref. 4, and was measured by the extraction method at the Service National des Champs Intense, Grenoble, France.
- <sup>13</sup>G. A. Stewart, J. Zukrowski, and G. Wortmann, J. Magn. Magn. Mater. <u>25</u>, 77 (1981).