Two-center transitions in the antiferromagnetic salt $FeCO₃$

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The optical-absorption spectrum of the antiferromagnetic salt $FeCO₃$ has been measured. As many as 34 bands have been observed. In order to give a reasonable explanation of the observed spectrum, an *ab initio*—like calculation including a trigonal crystal field has been made. Three stronger bands at 24 390, 24 875, and 25 126 cm^{-1} are interpreted in terms of two-center transitions of $Fe²⁺$ -ion pairs. The essential idea is the application of an analytical approximation of Watson's self-consistent-field procedure to obtain the spin-orbit interaction and the electrostatic parameters for the isolated ions. Crystal-field energy levels are then calculated using the nearest-neighbor point-ion approximation. Despite these approximations, the results are quite good.

I. INTRODUCTION

Iron carbonate (occurring naturally as the mineral siderite) has a rhombohedral structure belonging to the space group D_{3d}^{6} ($R\overline{3}c$). The Fe²⁺ ions order antiferromagnetically at 38 K in alternating (001} ferromagnetic sheets with the spin directed along the trigonal (c) axis. Neutron inelastic scattering studies demonstrate that the antiferromagnetic $FeCO₃$ is accurately described by the Ising model.¹ The Ising nature of the magnetic structure of FeCO₃ is a result of the crystal-field splitting of the Fe^{2+} ion as described by Kanamori and Okiji.² However, at the present time there is no satisfactory theory of a d-d transition in transition-metal ions which relates its spectrum to its crystal structure; consequently, a theoretical prediction of crystal-field splitting of a complex ion is not available.

Recently an attempt was made using an ab initio-like calculation for the spectra of MnF_2 (Refs. 3–6) and ruby. In the present work we will investigate the quantitative relationship between spectra and crystalline parameters on the basis of an analytical approximation of Watson's selfconsistent-field (SCF) d orbit. A specimen from the Sichuan siderite deposit was used in the present study.

II. d ORBIT FOR THE $Fe²⁺$ ION

Following the previous work, $3-6$ and using three mathematical conditions governing overlap: $R_d(r) \approx R_d'(r)$
(Watson) $(r < 0.4$ a.u.), $\langle R_d(r) | R_d' \rangle \approx 1$, and (Watson) $(r < 0.4$ a.u.), $\langle R_d(r) | R'_d \rangle \approx 1$, and $\langle R_d | r^{-3} | R_d \rangle \approx \langle R'_d | r^{-3} | R'_d \text{ (Watson)} \rangle$, then the analytical approximation of Watson's SCF d orbit is approximately given by

$$
R_d(r) = 0.5692 \left[\frac{11.03806^7}{6!} \right]^{1/2} r^2 \exp(-5.51903r) + 0.6500 \left[\frac{3.54794^7}{6!} \right]^{1/2} r^2 \exp(-1.77397r).
$$

Here the d orbit $R_d(r)$ is a solution of the Schrödinger equation,

$$
\left(-\frac{1}{2}\frac{d^2}{dr^2}+\frac{3}{r^2}-v(r)-E\right)rR_d(r)=0,
$$

with

$$
\left(-\frac{1}{2}\frac{d^2}{dr^2} + \frac{3}{r^2} - v(r) - E\right) rR_d(r) = 0,
$$
\n
$$
v(r) = \left(\frac{3}{r}\right) \frac{a_1 N_1 \zeta_1 \exp(-\zeta_1 r) + a_2 N_2 \zeta_2 \exp(-\zeta_2 r)}{a_1 N_1 \exp(-\zeta_1 r) + a_2 N_2 \exp(-\zeta_2 r)} - \frac{1}{2} \frac{a_1 N_1 \zeta_1^2 \exp(-\zeta_1 r) + a_2 N_2 \zeta_2^2 \exp(-\zeta_2 r)}{a_1 N_1 \exp(-\zeta_1 r) + a_2 N_2 \exp(-\zeta_2 r)} - E,
$$
\n
$$
N_j = \left(\frac{(2\zeta_j)^7}{6!}\right)^{1/2}, \quad j = 1, 2
$$

 $a_1 = 0.5692$, $a_2 = 0.6500$, $\zeta_1 = 5.51903$, $\zeta_2 = 1.77397$,

where the potential $v(r)$ is an analytical approximation of the Hartree-Fock SCF central potential by Watson.⁷ In the wave equation, the effects of s,p,d hybridization are neglected on the basis of the justification by Gray and Ballhausen.⁸

Using an IBM-130 computer and the formulas in Refs. ³—6, we find

 $\langle r^2 \rangle = 2.2949$, $\langle r^{-3} \rangle = 4.6072$, $\langle r^4 \rangle = 14$, $\zeta_d = 416$, $A = 160830$, $B = 947$, $C=3375$,

 (2)

Term		Theoretical frequency $(cm-1)$ $\alpha = 86$ cm ⁻¹ (Refs. 13 and 14)	Experimental frequency $(cm-1)$ (Ref. 15)
5D_j	4	Ω	0
	3	416	436
		728	738.9
		936	932.4
		1040	1027.3
3P_j		19580	19404.8
		19788	20688.4
		19892	21 208.5
$^{3}H_{j}$	6	19811	20051.1
	5	20335	20 300.8
		21010	20481.9
3F_j	4	19764	21462.2
	3	20180	21 699.9
		20490	21857.2
3G_j	5	22810	24 5 5 8.8
	4	23 3 3 1	24 94 0.9
	3	23747	25 142.4

TABLE I. The energy levels of the free $Fe²⁺$ ion.

where $\langle r^2 \rangle$, $\langle r^{-3} \rangle$, and $\langle r^4 \rangle$ are in atomic units, and ζ_d , A, B, and C are in units of cm⁻¹. A, B, and C are the Racah electrostatic parameters, and ζ_d is the spin-orbit coefficient including the Blume-Watson⁹ correction. A comparison of the theory with experiments is presented in Table I.

III. HAMILTONIAN MATRICES OF A $d^{4,6}$ CONFIGURATION POSSESSING $C_{\rm 3v}$ SYMMETRY

In the work by Phillips,¹⁰ a new method is presented for calculating crystal-field splittings. The method includes the effect of orthogonalization to ligand orbitals by grouping the latter terms into an effective repulsive potential. It is shown by a very general argument that this repulsive potential should cancel to high accuracy the excess attractive potential obtained when point-ion ligands are replaced by distributed-charge ligands. This result therefore provides a qualitative justification for the empirical success of the Van Vleck point-ion approximation for crystal fields. In this work we are utilizing the Van Vleck point-ion approximation for crystal-field energy levels.

In the case of C_{3v} symmetry, the crystal-field potential is given by

$$
v = \sum_{i=1}^{N} \left[-3 \left(\frac{\pi}{5} \right)^{1/2} eqr_i^2 \left(\frac{1}{R^3} (3 \cos^2 \Theta - 1) + \frac{1}{R'^3} (3 \cos^2 \Theta' - 1) \right) Z_{20}(\theta_i, \varphi_i) - \frac{\sqrt{\pi}}{4} eqr_i^4 \left(\frac{1}{R^5} (35 \cos^4 \Theta - 30 \cos^2 \Theta + 3) + \frac{1}{R^5} (35 \cos^4 \Theta' - 30 \cos^2 \Theta' + 3) \right) Z_{40}(\theta_i, \varphi_i) - \frac{\sqrt{70\pi}}{2} eqr_i^4 \left(\frac{1}{R^5} \sin^3 \Theta \cos \Theta - \frac{1}{R^5} \sin^3 \Theta' \cos \Theta' \right) Z_{43}^c(\theta_i, \varphi_i) \right] (C_3 || Z) ,
$$

FIG. 1. Optical-absorption spectrum of siderite (800—⁴⁴⁰⁰ cm^{-1}).

FIG. 2. Optical-absorption spectrum of siderite (4200—¹²⁰⁰⁰ cm^{-1}).

TABLE II. d -d transition in $FeCO$

 a α indicates Tree's correction.

where e is the electronic charge, q is the electric charge of ligand, $(r_i, \theta_i, \varphi_i)$ are the coordinates of the *i*th electron, (R_k, Θ_k, Φ_k) are the coordinates of the kth ligand, and $R = R'$, $\Theta' = \pi - \Theta$ for D_{3d} symmetry.

 ${}^3E(14)$ $^{3}A_{1}(6)$ $^{3}A_{2}(9)$ ${}^3E(15)$ ${}^5E(1)+{}^5E(1) \rightarrow {}^3E(1)+{}^3E(1)$

 ${}^3E(1)+{}^3A_2(1)$ $^{3}A_{2}(1)+^{3}A_{2}(1)$

Until now, there were no complete ligand-field matrices of $d^{4,6}$ configuration possessing C_{3v} symmetry and including the $Trees¹¹ correction.$ In this work, through elaborate mathematical derivations, the C_{3v} matrices of $d^{4,6}$ are derived using the weak-field scheme. The results are given in an unpublished paper.

IV. d-d TRANSITIONS OF FeCO3

The optical-absorption spectrum of $FeCO₃$ is shown in Figs. ¹ and 2; as many as 34 absorption bands have been observed in the Sichuan siderite. The stronger bands of 24 390, 24 875, and 25 126 cm^{-1} are interpreted in terms of the two-center transition of the Fe^{2+} -ion pairs (see Ref. 2). The comparison of theory with experiments is listed in

Table II (see Figs. $1-3$). In the calculation, the delocalization of the d orbitals are ignored on the basis of the experiments by Spiering et al .¹²

It follows from Table II that there is rather good numerical agreement between theory and experiment. This means that the errors of quite different sorts in the crystal-field model are compensated by the errors of the d orbit.

V. CONCLUSION

(a) Quantitative relationships between crystal-structure parameters and optical spectra are established with the use of an analytical approximation of Watson's SCF d orbit.

(b) For the $d-d$ transitions in FeCO₃, there are no free parameters at all and no fitting of parameters to data has been done.

(c) This work presents an ab initio-like calculation of energy levels of the $Fe²⁺$ ion in FeCO₃. The essential idea

FIG. 3. Optical-absorption spectrum of siderite $(12000 - 52000 \text{ cm}^{-1})$.

is the application of an analytical approximation of Watson's SCF procedure to obtain the spin-orbit interaction and the electrostatic parameters for the isolated ions. Crystal-field energy levels are then calculated using the nearest-neighbor point-ion approximation. Despite this approximation, the results are quite good, indicating that the technique may be of general use as the starting point of more sophisticated calculations on d^n ions in crystals.

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