Two-center transitions in the antiferromagnetic salt FeCO₃

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The optical-absorption spectrum of the antiferromagnetic salt $FeCO_3$ has been measured. As many as 34 bands have been observed. In order to give a reasonable explanation of the observed spectrum, an *ab initio*—like calculation including a trigonal crystal field has been made. Three stronger bands at 24 390, 24 875, and 25 126 cm⁻¹ are interpreted in terms of two-center transitions of Fe²⁺-ion pairs. The essential idea is the application of an analytical approximation of Watson's self-consistent-field procedure to obtain the spin-orbit interaction and the electrostatic parameters for the isolated ions. Crystal-field energy levels are then calculated using the nearest-neighbor point-ion approximation. Despite these approximations, the results are quite good.

I. INTRODUCTION

Iron carbonate (occurring naturally as the mineral siderite) has a rhombohedral structure belonging to the space group $D_{3d}^6(R\bar{3}c)$. The Fe²⁺ ions order antiferromagnetically at 38 K in alternating (001) ferromagnetic sheets with the spin directed along the trigonal (c) axis. Neutron inelastic scattering studies demonstrate that the antiferromagnetic FeCO₃ is accurately described by the Ising model.¹ The Ising nature of the magnetic structure of FeCO₃ is a result of the crystal-field splitting of the Fe²⁺ ion as described by Kanamori and Okiji.² However, at the present time there is no satisfactory theory of a d-d transition in transition-metal ions which relates its spectrum to its crystal structure; consequently, a theoretical prediction of crystal-field splitting of a complex ion is not available. Recently an attempt was made using an *ab initio*-like calculation for the spectra of MnF_2 (Refs. 3–6) and ruby. In the present work we will investigate the quantitative relationship between spectra and crystalline parameters on the basis of an analytical approximation of Watson's self-consistent-field (SCF) *d* orbit. A specimen from the Sichuan siderite deposit was used in the present study.

II. d ORBIT FOR THE Fe^{2+} ION

Following the previous work,³⁻⁶ and using three mathematical conditions governing overlap: $R_d(r) \approx R'_d(r)$ (Watson) (r < 0.4 a.u.), $\langle R_d(r) | R'_d \rangle \approx 1$, and $\langle R_d | r^{-3} | R_d \rangle \approx \langle R'_d | r^{-3} | R'_d$ (Watson), then the analytical approximation of Watson's SCF *d* orbit is approximately given by

$$R_d(r) = 0.5692 \left[\frac{11.03806^7}{6!} \right]^{1/2} r^2 \exp(-5.51903r) + 0.6500 \left[\frac{3.54794^7}{6!} \right]^{1/2} r^2 \exp(-1.77397r) .$$

Here the d orbit $R_d(r)$ is a solution of the Schrödinger equation,

$$\left| -\frac{1}{2} \frac{d^2}{dr^2} + \frac{3}{r^2} - v(r) - E \right| r R_d(r) = 0,$$

with

$$v(r) = \left[\frac{3}{r}\right] \frac{a_1 N_1 \zeta_1 \exp(-\zeta_1 r) + a_2 N_2 \zeta_2 \exp(-\zeta_2 r)}{a_1 N_1 \exp(-\zeta_1 r) + a_2 N_2 \exp(-\zeta_2 r)} - \frac{1}{2} \frac{a_1 N_1 \zeta_1^2 \exp(-\zeta_1 r) + a_2 N_2 \zeta_2^2 \exp(-\zeta_2 r)}{a_1 N_1 \exp(-\zeta_1 r) + a_2 N_2 \exp(-\zeta_2 r)} - E ,$$

$$N_j = \left[\frac{(2\zeta_j)^7}{6!}\right]^{1/2}, \quad j = 1, 2$$

$$a_1 = 0.5692, \quad a_2 = 0.6500, \quad \zeta_1 = 5.519\,03, \quad \zeta_2 = 1.773\,97 ,$$

where the potential v(r) is an analytical approximation of the Hartree-Fock SCF central potential by Watson.⁷ In the wave equation, the effects of s, p, d hybridization are neglected on the basis of the justification by Gray and Ballhausen.⁸

Using an IBM-130 computer and the formulas in Refs. 3-6, we find

 $\langle r^2 \rangle = 2.2949$, $\langle r^{-3} \rangle = 4.6072$, $\langle r^4 \rangle = 14$, $\zeta_d = 416$, $A = 160\,830$, B = 947, C = 3375,

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(2)

Term	j	Theoretical frequency (cm^{-1}) $\alpha = 86 \text{ cm}^{-1}$ (Refs. 13 and 14)	Experimental frequency (cm ⁻¹) (Ref. 15)
$^{5}D_{j}$	4	0	0
	3	416	436
	2	728	738.9
	1	936	932.4
	0	1 040	1 027.3
${}^{3}P_{j}$	2	19 580	19 404.8
	1	19788	20 688.4
	0	19 892	21 208.5
${}^{3}H_{j}$	6	19811	20051.1
	5	20 335	20 300.8
	4	21 010	20481.9
${}^{3}F_{j}$	4	19 764	21 462.2
	3	20 180	21 699.9
	2	20 490	21 857.2
${}^{3}G_{j}$	5	22 810	24 558.8
	4	23 331	24 940.9
	3	23 747	25 142.4

TABLE I. The energy levels of the free Fe^{2+} ion.

where $\langle r^2 \rangle$, $\langle r^{-3} \rangle$, and $\langle r^4 \rangle$ are in atomic units, and ζ_d , A, B, and C are in units of cm⁻¹. A, B, and C are the Racah electrostatic parameters, and ζ_d is the spin-orbit coefficient including the Blume-Watson⁹ correction. A comparison of the theory with experiments is presented in Table I.

III. HAMILTONIAN MATRICES OF A $d^{4,6}$ CONFIGURATION POSSESSING C_{3v} SYMMETRY

In the work by Phillips,¹⁰ a new method is presented for calculating crystal-field splittings. The method includes the effect of orthogonalization to ligand orbitals by grouping the latter terms into an effective repulsive potential. It is shown by a very general argument that this repulsive potential should cancel to high accuracy the excess attractive potential obtained when point-ion ligands are replaced by distributed-charge ligands. This result therefore provides a qualitative justification for the empirical success of the Van Vleck point-ion approximation for crystal fields. In this work we are utilizing the Van Vleck point-ion approximation for crystal-field energy levels.

In the case of C_{3v} symmetry, the crystal-field potential is given by

$$\begin{split} v &= \sum_{i=1}^{N} \left[-3 \left[\frac{\pi}{5} \right]^{1/2} eqr_i^2 \left[\frac{1}{R^3} (3\cos^2\Theta - 1) + \frac{1}{R'^3} (3\cos^2\Theta' - 1) \right] Z_{20}(\theta_i, \varphi_i) \\ &- \frac{\sqrt{\pi}}{4} eqr_i^4 \left[\frac{1}{R^5} (35\cos^4\Theta - 30\cos^2\Theta + 3) + \frac{1}{R^5} (35\cos^4\Theta' - 30\cos^2\Theta' + 3) \right] Z_{40}(\theta_i, \varphi_i) \\ &- \frac{\sqrt{70\pi}}{2} eqr_i^4 \left[\frac{1}{R^5} \sin^3\Theta \cos\Theta - \frac{1}{R^5} \sin^3\Theta' \cos\Theta' \right] Z_{43}^c(\theta_i, \varphi_i) \right] \quad (C_3 ||Z) , \end{split}$$

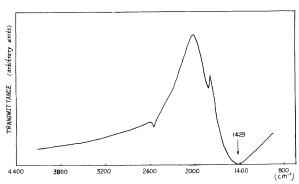


FIG. 1. Optical-absorption spectrum of siderite $(800-4400 \text{ cm}^{-1})$.

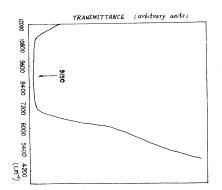


FIG. 2. Optical-absorption spectrum of siderite $(4200-12\ 000\ cm^{-1})$.

		Theoretical frequency (cm ⁻¹) ($\alpha = 85 \text{ cm}^{-1}$) ^a ($R = 2.18 \text{ Å}, \Theta = 53.036^{\circ}$)	Experimental frequency (cm^{-1})	
Transition		($\mathbf{R} = 2.18 \ \mathbf{A}, \ \mathbf{O} = 55.050$) (Refs. 16 and 17)	This work	Refs. 1 and 2
${}^{5}E(1) \rightarrow$	${}^{5}A_{1}(1)$	1358	1429	1410
	${}^{5}E(2)$	9178	9150	10 000
	${}^{3}E(1)$	12 102	12 195	
	${}^{3}A_{2}(1)$	12 553	12 468	
	${}^{3}A_{1}(1)$	15 425	15 576	
	$^{3}E(2)$	16 383	16 393	
	${}^{3}A_{2}(2)$	19 726	19 531	
	${}^{3}E(3)$	20 292	20 534, 20 833	
	${}^{3}E(4)$	21 591	21 186, 21 367	
	${}^{3}A_{2}(3)$	21 902	21 930	
	${}^{3}A_{1}(2)$	22 640	22 222	
	${}^{3}E(5)$	23 501	23 474	
	$^{3}E(6)$	23 927	24 038	
	${}^{3}E(7)$	26 2 56	26 246	
	${}^{3}A_{1}(3)$	26 865	26 882	
	${}^{3}A_{2}(4)$	27 099	27 472, 27 777	
	${}^{3}E(8)$	29 029	28736, 29070	
	${}^{3}E(9)$	29 849	29 762	
	${}^{3}A_{2}(5)$	30 035	30 4 8 8	
	${}^{3}A_{1}(4)$	31 699	31 447	
	${}^{3}A_{1}(5)$	32 428	32 467	
	${}^{3}E(10)$	32 871		
	${}^{3}E(11)$	34 823		
	${}^{3}A_{2}(6)$	35 187	35 700	
	${}^{3}E(12)$	36 739	38 168	
	${}^{3}A_{2}(7)$	50415	44 642?48 544?	
	${}^{3}E(13)$	50 446	50 000?	
	${}^{3}A_{2}(8)$	50 998	51 020?	
	${}^{3}E(14)$	53 532		
	${}^{3}A_{1}(6)$	54 870		
	${}^{3}A_{2}(9)$	55 671		
	${}^{3}E(15)$	56918		
${}^{5}E(1) + {}^{5}E(1)$	$\rightarrow {}^{3}E(1) + {}^{3}E(1)$	24 204	24 390	
	${}^{3}E(1) + {}^{3}A_{2}(1)$	24 655	24 875	

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TABLE II. d-d transition in FeCO3.

^a α indicates Tree's correction.

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where e is the electronic charge, q is the electric charge of ligand, $(r_i, \theta_i, \varphi_i)$ are the coordinates of the *i*th electron, (R_k, Θ_k, Φ_k) are the coordinates of the *k*th ligand, and $R = R', \Theta' = \pi - \Theta$ for D_{3d} symmetry.

 ${}^{3}A_{2}(1) + {}^{3}A_{2}(1)$

Until now, there were no complete ligand-field matrices of $d^{4,6}$ configuration possessing C_{3v} symmetry and including the Trees¹¹ correction. In this work, through elaborate mathematical derivations, the C_{3v} matrices of $d^{4,6}$ are derived using the weak-field scheme. The results are given in an unpublished paper.

IV. d-d TRANSITIONS OF FeCO3

The optical-absorption spectrum of FeCO₃ is shown in Figs. 1 and 2; as many as 34 absorption bands have been observed in the Sichuan siderite. The stronger bands of 24 390, 24 875, and 25 126 cm⁻¹ are interpreted in terms of the two-center transition of the Fe²⁺-ion pairs (see Ref. 2). The comparison of theory with experiments is listed in

Table II (see Figs. 1–3). In the calculation, the delocalization of the d orbitals are ignored on the basis of the experiments by Spiering *et al.*¹²

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It follows from Table II that there is rather good numerical agreement between theory and experiment. This means that the errors of quite different sorts in the crystal-field model are compensated by the errors of the d orbit.

V. CONCLUSION

(a) Quantitative relationships between crystal-structure parameters and optical spectra are established with the use of an analytical approximation of Watson's SCF d orbit.

(b) For the d-d transitions in FeCO₃, there are no free parameters at all and no fitting of parameters to data has been done.

(c) This work presents an *ab initio*-like calculation of energy levels of the Fe^{2+} ion in FeCO₃. The essential idea

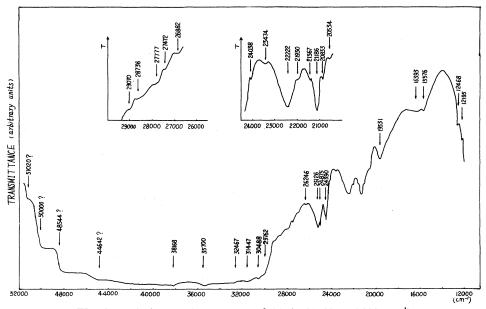


FIG. 3. Optical-absorption spectrum of siderite $(12\,000-52\,000\,\text{cm}^{-1})$.

is the application of an analytical approximation of Watson's SCF procedure to obtain the spin-orbit interaction and the electrostatic parameters for the isolated ions. Crystal-field energy levels are then calculated using the nearest-neighbor point-ion approximation. Despite this approximation, the results are quite good, indicating that the technique may be of general use as the starting point of more sophisticated calculations on d^n ions in crystals.

- ¹D. E. Wrege, S. Spooner, and H. A. Gersch, in *Magnetism and Magnetic Materials, Chicago, 1971, Proceedings of the 17th Annual Conference on Magnetism and Magnetic Materials, edited by D. C. Graham and J. J. Rhyne (AIP, New York, 1972), pp. 1334-1338.*
- ²A. Okiji and J. Kamamori, J. Phys. Soc. Jpn. <u>19</u>, 908 (1964).
- ³Zhao Min-Guang, Xu Ji-An, and Bai Gui-Ru, Sci. Sin. <u>25</u>, (10), 1066 (1982).
- ⁴Xu Ji-An and Zhao Min-Guang, Sci. Sin. <u>24</u>, (5), 721 (1981).
- ⁵Zhao Min-Guang, Xu Ji-an, Bai Gui-Ru, and Xie Huong-Sen, Phys. Rev. B <u>27</u>, 1516 (1983).
- ⁶Zhao Min-Guang, Bai Gui-Ru, and Jin Huei-Cheng, J. Phys. C. <u>15</u>, 5959 (1982).
- ⁷R. E. Watson, Massachusetts Institute of Technology Technical Report No. 12 (unpublished).

- ⁸H. B. Gray and C. J. Ballhanen, Acta Chem. Scand. <u>15</u>, 1327 (1961).
- ⁹M. Blume and R. E. Watson, Proc. R. Soc. London Ser. A <u>270</u>, 127 (1962).
- ¹⁰J. C. Philips, J. Phys. Chem. Solids <u>11</u>, 226 (1959).
- ¹¹R. E. Trees, Phys. Rev. <u>82</u>, 683 (1951); <u>84</u>, 1083 (1952).
- ¹²H. Spiering, D. L. Nagy, and R. Zimmerman, Chem. Phys. <u>18</u>, 243 (1976).
- ¹³R. H. Garstang, Mon. Not. R. Astron. Soc. <u>117</u>, 393 (1957).
- ¹⁴D. R. Rosseinsky and I. A. Dorrity, Coord. Chem. Rev. <u>25</u>, 31 (1978).
- ¹⁵J. Reoder and J. Sgar, J. Phys. Chem. Ref. Data <u>4</u>, 384 (1975).
- ¹⁶H. Spiering, D. L. Nagy, and R. Zimmerman, Chem. Phys. <u>18</u>, 243 (1976).
- ¹⁷P. P. Ewald and C. Herman, Strukturbericht, p. 194.