Correlation effects in the electronic structure and photoemission spectra of mixed-valence cerium compounds

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The electronic ground states and core-hole and valence-hole excited states of so-called tetravalent and mixed-valence Ce compounds are studied taking into account strong correlations among 4felectrons. By an analysis of the core-level photoemission satellites, CeO₂, which has been thought to be tetravalent, is identified as a mixed-valence system with ~0.6 4f electron, where the magnetic moment of the $4f^1$ configuration is quenched via singlet coupling between the 4f electron and the O 2p hole. 4f-derived satellite features in the resonant photoemission spectrum of CeO₂ are calculated, which qualitatively explain resonant photoemission spectra of tetravalent Ce intermetallic compounds. The core-level and 4f-derived photoemission spectra of mixed-valence CeN are also studied by considering only the 4f-electron—N 2p-hole coupling as in CeO₂ and neglecting the 4f-5d mixing. It is also suggested that for the α phase of Ce metal the absence of a localized magnetic moment, volume collapse, and a $4f^0$ signal in the core-level photoemission spectra are explained by this type of configuration-mixing description.

I. INTRODUCTION

There has been much controversy on the nature of the 4f state in Ce compounds, to which a variety of anomalous physical and chemical properties have been ascribed.¹ A central issue is the γ - to α -phase transition in Ce metal, where the localized magnetic moment of the Ce $4f^1$ configuration disappears and the lattice volume decreases by ~15% in going from the γ to α phase. Absence of a localized magnetic moment and small lattice volume are also seen in so-called tetravalent Ce compounds and at least at low temperatures in mixed-valence Ce compounds. This fact might suggest that common mechanisms are responsible for the above properties in all these compounds, although quite different 4f-electron numbers have generally been supposed: α -Ce has nearly one 4f electron²⁻⁸ as in γ -Ce, mixed-valence Ce compounds have a nonintegral 4f number between zero and 1. and tetravalent compounds have no 4f electron.

Recently, fundamental questions have been raised against the existence of purely tetravalent $(4f^0)$ Ce compounds by x-ray absorption and photoemission spectroscopies. The x-ray absorption edge of all tetravalent Ce compounds studied by Bauchspiess *et al.*⁹ including CeO₂ has shown both trivalent and tetravalent features. Some authors^{10,11} have attributed these features to final-state effects due to strong perturbation by a core hole, namely $4f^0 \rightarrow 4f^1$ transition. However, for CeRu₂, CeCo₂, CeRh₃, etc., which are nonmagnetic and have also been thought to be tetravalent, 4f-derived emission was observed by resonant photoemission.¹²⁻¹⁴ Core-level x-ray photoemission spectroscopy (XPS) of these compounds has shown both $4f^1$ and $4f^0$ lines.¹⁰ Therefore, if the $4f^1$ features in the core-level spectra are largely due to the initial-state $4f^1$ component rather than to the final-state effects, the tetravalent compounds should be regarded as mixed valence and certain properties of mixed-valence and tetra-valent compounds would be explained on the same basis.

In the present paper, photoemission spectra of some nonmagnetic Ce compounds are studied theoretically in order to quantitatively investigate the electronic structure of these compounds, e.g., the 4f-electron number. Electron correlations among the 4f electrons are explicitly taken into account in the initial state as well as in the final state of photoemission. A mechanism is proposed in which a localized magnetic moment of Ce 4f can be quenched via hybridization with neighboring orbitals. This hybridization can give any 4f number between 0 and 1 depending on materials, and it is strongly suggested that a significant amount of the 4f electron is present in all tetravalent Ce compounds.

First, we consider a typical tetravalent oxide CeO₂ which is nonmagnetic¹⁵ and insulating, and can be most simply understood as the $4f^0$ configuration and the completely filled O 2p valence band. We shall show, however, in the following sections that the ground state of CeO₂ is a mixture of the $4f^0$ and $4f^1$ configurations. The core-level XPS of CeO₂ as well as other tetravalent compounds¹⁶ [Ce(tmhd)₄ (tmhd = tetramethylheptane dionato), BaCeO₃] exhibits quite different features from those of trivalent ($4f^1$) compounds. The core-level spectra of trivalent compounds show main $4f^1$ and weaker $4f^2$ shake-down peaks, both originating from the stable $4f^1$ initial-state configuration. The energy separation between the $4f^1$ and $4f^2$ peaks is 2–6 eV, and in general the satellite intensity decreases rapidly with increasing peak separation.¹⁷ In CeO₂, on the other hand, two intense peaks separated by as much as ~16 eV are observed together with weaker peaks at the higher-binding-energy side of the low-energy intense peak (Fig. 1). Burroughs *et al.*, ¹⁶ based on the $4f^0$ ground state, have identified the highest-binding-energy peak with the main line ($4f^0$ final state) and the lowest-binding-energy peak with a ligandto-4f shake-down satellite ($4f^1$ final state), but according to Ref. 17 (hereafter referred to as I) such a high intensity would not be expected for that large a main-satellite separation. Furthermore, the weak lines could not be given a reasonable assignment.

A recent theoretical study using the scattered-wave- $X\alpha$ $(SW-X\alpha)$ method for a CeO₈ cluster¹⁸ has identified the lowest-binding-energy peak as the main peak and the weak lines as ligand-to-4f shake-up satellites. These authors, however, could not give a reasonable explanation for the intense highest-binding-energy peak, and tentatively attributed it to the Ce 5p-to-np transition. The latter assignment is hardly acceptable because one would expect similar transitions in Ce₂O₃ or in La₂O₃ which have not been observed. Basic difficulty probably lies in an assignment based on one-electron molecular-orbital energy diagrams in terms of which one cannot properly take into account electron correlations among the 4f electrons, in particular in the final state of photoemission. It has been found in some cases even in correlated systems that one-electron theories give correct ground-state properties such as total energies, electron distribution, magnetization, etc.^{8,19}; in fact, in the SW- $X\alpha$ study¹⁸ the 4f occupancy in the ground state has been calculated to be ~ 0.5 , which is in fair agreement with our present results. However, the success of one-electron theories only means that the above quantities are well approximated by expectation values of a single Slater determinant consisting of molecular orbi-



FIG. 1. Experimental (Ref. 16) and calculated Ce 3*d* corelevel XPS spectra of CeO₂. *v*, *v'*, *v''*, and *v'''* belong to the Ce $3d_{5/2}$ spin-orbit component and *u*, *u'*, *u''*, and *u'''* to the Ce $3d_{3/2}$ component. The calculated spectrum is shown by vertical bars. $4f^0$, $4f^1$, and $4f^2$ represent main components of the final states. $E_1 = E'(4f^0 \rightarrow 4f^1L^{-1})$ is an energy required to move an electron from O 2*p* to Ce 4*f* in the presence of the core hole and Δ_{so} is the spin-orbit splitting between the $3d_{5/2}$ and $3d_{3/2}$ levels.

tals, but does not necessarily mean that the real groundstate wave function itself is properly represented by the single Slater determinant.

We therefore use a basis set with the definite numbers of the 4f electrons, which is suited for strongly correlated systems. When the initial state is a mixture of the 4f⁰ and 4f¹ configurations, the core-level spectra show characteristics of valence mixing, namely 4f⁰, 4f¹, and 4f² final-state peaks. In this case the 4f⁰ peak as well as the 4f¹ peak mainly arises from a direct transition, not by a shake-up or shake-down process, and therefore can be intense irrespective of the 4f⁰-4f¹ line separation. This configuration mixing in CeO₂ is of a local type $4f^0 \leftrightarrow 4f^1L^{-1}$, where L^{-1} is a ligand (O 2p) hole, and does not necessarily imply the position of the 4f level located at the Fermi level E_F as in the usual valence-mixing mechanism involving conduction electrons $4f^0(5d6s)^n \leftrightarrow 4f^1(5d6s)^{n-1}$, where $(5d6s)^n$ represents n electrons in the 5d 6s conduction band. The 4f occupancy is thus obtained to be as large as ~0.6.

We also present the 4*f*-derived emission spectrum of CeO₂ which can be observed in resonant photoemission spectroscopy. It is shown that gross features of the 4*f* spectrum of tetravalent Ce compounds are not very different from those of trivalent compounds studied in I. This would account for the 4*f* emission from tetravalent CeRu₂, CeCo₂, CeRh₃, etc.,¹²⁻¹⁴ similar to those of trivalent CeAl₂, γ -Ce, etc.^{20,21}

The local valence mixing $4f^0 \leftrightarrow 4f^1L^{-1}$ as well as extended $4f^0(5d \, 6s)^n \leftrightarrow 4f^1(5d \, 6s)^{n-1}$ might be important in metallic mixed-valence systems such as CeN,²² CePd₃,^{23,24} etc. Thus we have applied the above procedure to CeN by neglecting interactions between the 4f level and the 5d conduction band. The results reproduce well the corelevel XPS (Ref. 25) and resonant photoemission²⁶ spectra with the 4f occupancy of ~0.8, which is considerably larger than previously accepted values estimated from the lattice volume,^{22,27} although the $4f^0(5d \, 6s)^n \rightarrow 4f^1(5d \, 6s)^{n-1}$ mixing is also suggested to be significant.

The same mixed-valence mechanism as in CeO₂ could also be responsible for the α phase of Ce metal, where the localized magnetic moment of Ce 4f disappears. The core-level XPS spectra of α -Ce have clearly shown 4f⁰ final-state signals.^{25,28} This fact indicates that the 4f⁰ component, though much smaller than in CeO₂, is present in the ground state of α -Ce. It should be noted that an essential point is the symmetry of the 4f state but is not the 4f occupancy. A simplified model is presented for the γ -to- α transition and discussed with relation to the resonant photoemission below.

The organization of this paper is as follows: In Sec. II general discussion of core-level and 4f-derived resonant photoemission spectra are given for stable trivalent and tetravalent or mixed-valence Ce compounds. In Sec. III formulation based on a cluster model is developed with CeO₂ as an example. Results for CeO₂ and CeN are given in Secs. IV and V, respectively. A model for the γ -to- α transition of Ce metal is presented in Sec. VI. Finally, in Sec. VII, relation of our model to other mixed-valence descriptions are discussed. A summary of the results for CeO₂ has already been published elsewhere.²⁹

II. GENERAL DESCRIPTION OF CORE-LEVEL AND 4f-DERIVED PHOTOEMISSION

Photoemission spectroscopy is a powerful tool in studying mixed-valence systems: Although a photoinduced hole produces strong perturbations in the system studied, spectroscopic selection rules would give unique opportunity to characterize the electronic structure of the ground state. Before proceeding with mathematical formulation, we describe a general picture of the core-level and resonant 4f photoemission for mixed-valence Ce compounds. In the sudden approximation the intensity of an XPS core level is given by

$$I_F \propto |\langle \Psi'_I | \Psi'_F \rangle|^2, \qquad (1)$$

where Ψ'_F is a core-hole final state and Ψ'_I is a core-hole state with all other orbitals frozen in the initial state. Equation (1) means that the intensity of a final state is determined by a projection of the initial state onto the final state. As has been shown in I, for stable-valence (Ce $4f^{1}$) compounds there is hybridization among different final-state configurations which give rise to so-called shake-up or shake-down transitions. For mixed-valence compounds hybridization both in the initial and final states should be considered. The degree of hybridization is generally large between those two configurations which are close to each other in energies. Figure 2 illustrates transitions associated with a core-level ionization. There intensities of transitions are designated by thickness of arrows and resulting spectra are also schematically shown: In the $4f^1$ compound the ground state is almost purely



FIG. 2. Energy diagrams for core-level photoemission from trivalent (Ce^{3+}) and tetravalent (Ce^{4+}) or mixed-valence Ce. Intensities of the transitions are given by thickness of arrows. Resulting spectra are shown on the right-hand side.

 $4f^1$, while in the mixed-valence compound the initial state is a mixture of the $4f^0$ and $4f^1$ configurations and the $4f^0$ signal is intense owing to the direct transition from the initial-state $4f^0$ component.

In resonant photoemission a photon is resonantly absorbed, $4d^{10}4f^1 + h\nu \rightarrow 4d^94f^2$, and then the $4d^94f^2$ state decays via a super-Coster-Kronig (SCK) decay into $4d^{10}4f^0 + \epsilon l$. As the intermediate state is localized due to the strong core-hole potential, a localized $4f^0$ hole is left in the final state. Thus the $4d \rightarrow 4f$ resonant photoemission can be described simply as enhanced 4f emission of the 4f electron, that is, the intensity is effectively given by

$$I_F \propto |\langle \Psi_I | e \vec{\mathbf{r}} | \Psi_F \rangle|^2, \qquad (2)$$

where dipole matrix elements for the 4*f*-emission $(4f^1 \rightarrow 4f^0)$ component are predominant. As can be seen in Fig. 3, $4f^1$ and mixed-valence Ce compounds show grossly similar spectra: Two features corresponding to the $4f^0$ and $4f^1$ final states are expected, the latter being allowed by hybridization with the former state.

III. GROUND STATE, CORE-HOLE, AND VALENCE-HOLE STATES OF THE Ce-LIGAND CLUSTER

We describe in this section theoretical details by using CeO₂ as an example. CeO₂ forms in the CaF₂ structure as in Fig. 4. As we are interested only in localized excitations, a CeO₈¹²⁻ cluster as in the figure is considered. The present formulation is closely related to that in I, where we have studied stable-valence cerium with the ground state having a localized magnetic moment of $4f^{1}(^{2}F_{5/2})$, while here the ground state is nonmagnetic and therefore should be fully symmetric (singlet) with respect to both orbitals and spins. In addition to an obvious singlet configuration, $4f^{0}$ configuration with the com-



FIG. 3. Energy diagrams for 4f-derived photoemission from trivalent (Ce³⁺) and tetravalent (Ce⁴⁺) or mixed-valence Ce. Resulting spectra are shown on the right-hand side.



FIG. 4. Crystal structure of CeO₂. Open and shaded circles represent O and Ce atoms, respectively. The CeO₈ cluster used in the present study is indicated by thick lines.

pletely filled O 2p levels, we can also consider a $4f^1$ -ligand-hole $(4f^1L^{-1})$ complex having the full symmetry of the cluster. As the $4f^1L^{-1}$ configuration is expected to lie not too high in energy as compared to the 4f-L hybridization (~1 eV), the mixing $4f^0 \leftrightarrow 4f^1L^{-1}$ must be present. The ground state may thus be written as

$$\Psi_I = c_{I0} | 0 \rangle + \sum_{l} c_{Il} | l \rangle , \qquad (3)$$

$$|l\rangle = \alpha_{l}^{\dagger}|0\rangle = \sum_{i,j} \Gamma_{ij}^{l} \sum_{\sigma} L_{i\sigma} f_{j\sigma}^{\dagger}|0\rangle , \qquad (4)$$

where $L_{i\sigma}$ and $f_{j\sigma}^{\dagger}$ are operators which annihilate an electron in a ligand level L_i and create an electron in a 4f crystal-field level f_j with spin σ , respectively. The first term $|0\rangle$ represents the $4f^0$ configuration with the closed O 2p shell (to be exact, the completely filled Ce 5d-O 2p bonding levels) and the second term represents the $4f^1$ state where the 4f electron and ligand hole form a singlet state. The coefficients Γ_{ij}^l 's are to make the ground state fully symmetric. Within the CeO₈ cluster model we have four *l*'s in (3), two of which involve t_{1u} -like molecular orbitals, one of which is t_{2u} -like and the rest are a_{2u} -like orbitals.

The ground state is obtained by solving a secular equation

$$\sum_{m} H_{lm} c_{Im} = E_I c_{Il} \ . \tag{5}$$

Nonzero matrix elements of the Hamiltonian in (5) have the form of $H_{0l} = \langle 0 | H | l \rangle = V_l$ and $H_{ll} = \langle l | H | l \rangle$ $= E(4f^0 \rightarrow 4f_i L_j^{-1})$, which is an energy required to move an electron from L_j to $4f_i$ in the orbital set *l*. According to the SW-X α results¹⁸ two V_l 's out of four are significant, one involving a_{2u} -like orbitals and the other involving t_{1u} -like orbitals, and we retain only the two V_l 's. Moreover, we set $E(4f^0 \rightarrow 4f_i L_j^{-1})$'s for a_{2u} and t_{1u} to be the same (equal to E_0) since the ligand orbitals a_{2u} and t_{1u} have the same energies within 0.01 Ry.¹⁸ Thus the Hamiltonian is

$$\begin{bmatrix} 0 & V_1 & V_2 \\ V_1 & E_0 & 0 \\ V_2 & 0 & E_0 \end{bmatrix}.$$
 (6)

Transfer integrals $\langle f_{xyz} | h | L_{xyz} \rangle$ for a_{2u} and $\langle f_x | h | L_x \rangle$ for t_{1u} , where h is the one-electron Hamiltonian, are estimated from the energy-level diagram and the 4f populations of the CeO₈ cluster¹⁸ to be 1.05 and 0.70 eV, respectively. Then using relations

$$|t_{1u}\rangle = (1/\sqrt{2})[(1/\sqrt{3})(L_{x\uparrow}f_{x\uparrow}^{\dagger} + L_{y\uparrow}f_{y\uparrow}^{\dagger} + L_{z\uparrow}f_{z\uparrow}^{\dagger}) + (\downarrow \text{ terms})]|0\rangle ,$$

$$|a_{2u}\rangle = (1/\sqrt{2})[L_{xyz\uparrow}f_{xyz\uparrow}^{\dagger} + (\downarrow \text{ term})]|0\rangle ,$$

we obtain $V_1 = \sqrt{2} \langle f_{xyz} | h | L_{xyz} \rangle = 1.48$ eV and $V_2 = \sqrt{6} \langle f_x | h | L_x \rangle = 1.71$ eV.

Valence electrons in a core-hole state are also in a singlet state if one can neglect the core-hole-4f-multiplet coupling. This would be a good approximation for the 3dcore levels which we are going to treat here. Thus the core-hole state is given by

$$\Psi_{F}^{\prime} = c_{F0} \mid 0^{\prime} \rangle + \sum_{l} c_{Fl} \mid l^{\prime} \rangle + \sum_{l,m} c_{Flm} \mid lm^{\prime} \rangle , \qquad (7)$$

$$|l'\rangle = \alpha_l^{\dagger} |0'\rangle , \qquad (8)$$

$$|l,m'\rangle = \alpha_l^{\dagger} \alpha_m^{\dagger} |0'\rangle , \qquad (9)$$

where primes represent the presence of a core hole on Ce. The first, second, and last terms in (7) represent, respectively, the $4f^0$, $4f^1$, and $4f^2$ final states. Within the same approximation as that for the initial state we use two V_i 's in the second term and one $(l = t_{1u} \text{ and } m = a_{2u})$ in the last term of (7). Thus the final-state Hamiltonian is

$$\begin{pmatrix} 0 & V_1 & V_2 & 0 \\ V_1 & E_1 & 0 & V_2 \\ V_2 & 0 & E_1 & V_1 \\ 0 & V_2 & V_1 & E_1 + E_2 \end{pmatrix} .$$
 (10)

Here $E_1 = E'(4f^0 \rightarrow 4f^1L^{-1})$ is an energy required to move an electron from the ligand to 4f level, and $E_2 = E'(4f^1L^{-1} \rightarrow 4f^2L^{-2})$ is that required to further move one more electron in the presence of the core hole. The intensity of the final state Ψ'_F is given within the sudden approximation by

$$I_F \propto \left| c_{F0} c_{I0} + \sum_{l} c_{Fl} c_{Il} \right|^2.$$
(11)

The valence-band spectrum is obtained by considering a valence-hole state

$$I_{fi\sigma} = c_{F0}L_{i\sigma} \mid 0 \rangle + \sum_{l} c_{Fl}L_{i\sigma} \mid l \rangle .$$
(12)

One can see that (12) consists of one-ligand-hole and twoligand-hole—one-4f-electron parts. Again we consider only valence-hole states with the a_{2u} and t_{1u} symmetry because here we are interested only in 4f-derived photoemission which arises from the second term of the initial state (3) involving only ligand holes with a_{2u} and t_{1u} characters. Emission from ligand orbitals which do not couple with Ce $4fa_{2u}$ or t_{1u} gives simply one-electron density-of-states (DOS) features of occupied molecular orbitals. For the a_{2u} -hole state or the t_{1u} -hole state we solve the secular equation of the Hamiltonian

$$\begin{bmatrix} \epsilon_L & \overline{V}_1 & \overline{V}_2 \\ \overline{V}_1 & \epsilon_L + E_a & 0 \\ \overline{V}_2 & 0 & \epsilon_L + E_a \end{bmatrix},$$
(13)

where ϵ_L is the O 2*p* level energy and $E_a = E(4f^0L^{-1} \rightarrow 4f^1L^{-2})$. For the a_{2u} -like hole state [i.e., $i = a_{2u}$ in (12)] $\overline{V_1} = (1\sqrt{2})V_1$ because of the Pauli principle and $\overline{V_2} = V_2$. For the t_{1u} -like hole state [i.e., $i = t_{1u}x$, *y*, or *z* in (12)] $\overline{V_1} = V_1$ and $\overline{V_2} = \sqrt{5/6}V_2$. Note that the energy zero of (13) is the initial ground state. Thus the intensity of resonant photoemission appearing at a binding energy E_B , an eigenvalue of (13), is given by

$$I_{Fi} \propto |c_{F0}c_{Ii}|^2, \qquad (14)$$

for the $i\sigma$ -hole final state, where $i = a_{2u}$, $t_{1u}x$, y, or z.

IV. CeO₂

The XPS spectrum of the Ce 3d core level in CeO₂ consists of two spin-orbit components $3d_{5/2}$ and $3d_{3/2}$ separated by ~ 20 eV, each having satellite structures as are described in Sec. I (Fig. 1). The Ce 4d spectrum shows also similar satellite structures,¹⁶ though spin-orbit splitting is much smaller and multiplet splittings are significant. The parameters E_0 , E_1 , and E_2 have been adjusted to reproduce the experimental spectrum: $E_0 = 0.1$, $E_1 = -14.2$, and $E_2 = 1.1$ eV; $\Delta_{so} = 19.5$ eV has been used. The signal u'''(v''') corresponds to the $4f^0$ final state arising mostly from the initial-state $4f^0$ component (see Fig. 2). u(v) is the $4f^1$ final state arising from the $4f^1$ initial state, and u' and u'' (v' and v'') are the $4f^2$ final states resulting from L-to-4f shake-up transitions from the $4f^1$ final states. The 4f occupancy thus obtained is ~ 0.6 (Table I), which agrees well with that obtained by the x-ray absorption edge, 0.68.⁹

4f-derived emission using the above parameters are shown as a function of E_a in Fig. 5. Stronger lines are due to 4f emission leaving a $4f^{0}L^{-1}$ state, while weaker features are due to $4f^{1}L^{-2}$ -dominant final states arising from 4f emission followed by the $L \rightarrow 4f$ transition. So far resonant photoemission of CeO₂ has not been reported. The valence-band XPS spectrum of CeO₂ (Ref. 30) seems to be consistent with 4f emission overlapping the O 2p emission, as the O 2p-band width appears a little larger than that of La₂O₃. This might correspond to the calculated spectrum of $E_a = 1-2$ eV in Fig. 5, where 4f emission overlaps the O 2p band.

TABLE I. Energies (in eV) and $4f^0$ and $4f^1$ population of the ground state and singlet excited states of the CeO₈¹²⁻ cluster.

Energy ^a	$4f^0$	$4f^1(a_{2u})$	$4f^{1}(t_{1u})$
-2.21	0.39	0.44	0.17
0.10	0.28	0.56	0.16
2.31	0.32	0.00	0.68

^aPure $4f^0$ state is taken as the energy zero.



FIG. 5. Calculated 4f-derived photoemission spectra of CeO₂ as a function of $E_a = E(4f^0L^{-1} \rightarrow 4f^1L^{-2})$. The O 2p (a_{2u} and t_{1u}) level (equal to ϵ_L) is shown by arrows. The top panel shows the one-electron density of states of the CeO₈ cluster (Ref. 18).

The 4f spectra in Fig. 4 would explain gross features of the 4f-derived emission in the tetravalent compounds CeRu₂, CeCo₂, and CeRh₃.¹²⁻¹⁴ This may give evidence that they are mixed valence rather than $4f^0$. Furthermore, the intensities of the $4f^0$, $4f^1$, and $4f^2$ core-level lines¹⁰ suggest that the 4f occupancy in these compounds are closer to 1 rather than to 0.

We compare here the present results with those of the SW-X α calculation.¹⁸ In SW-X α cluster calculations the ground state of a core-ionized state is assigned to the main line and excited states to shake-up states. This has sometimes caused confusion, particularly in cases where the ground state of a core-hole state is reached by a shakedown transition. It should be noted, however, that both in SW-X α and other theories the core hole is generally most effectively screened in the lowest-binding-energy core-hole state.³¹ In the case of CeO_2 , the core-hole ground state in the SW-X α calculation corresponds to one $(4f^1)$ of the two $(4f^0 \text{ and } 4f^1)$ main lines, and the shake-up satellites in the SW-X α results also correspond to the $4f^2$ final states arising from the L-to-4f shake-up transitions. In the SW-X α ground state the 4f occupancy has been calculated to be about 0.45 within the Ce muffin-tin sphere, which is well compared with our result ~ 0.6 . It seems that one-electron local-density calculations usually give reasonable 4*f*-electron numbers in the ground state if the calculation is performed to self-consistency.^{8,19} In fact, in the 3d core-hole ground state the 4f number increases to about 0.9 and in the shake-up states further to ~ 1.3 , which corresponds well to our $4f^1$ and $4f^2$ final states. The SW-X α shake-up energies, however, are too small $(\sim 2 \text{ eV})$ in the SW-X α results as compared to the experimental ones ($\sim 5 \text{ eV}$). This discrepancy is inherent in the SW-X α method itself, and is not discussed further here. The basic difficulty in the SW-X α calculation is that it could not explain the $4f^0$ final-state lines (u''' and v''').

These final states would probably not be reached by a one-electron excitation from the core-hole ground state but would essentially need a many-electron excitation in the molecular-orbital picture. The use of the basis set with the definite numbers of the 4f electron instead of molecular-orbital bases is therefore quite suitable in describing such a strongly correlated system.

We note that some characteristic features of mixedvalence Ce compounds, e.g., disappearance of a localized magnetic moment, small lattice volume could be understood within the present model. As the ground state (3) is a bonding state between the $4f^0$ and $4f^1L^{-1}$ configurations, it is stabilized for a smaller lattice volume through increased 4f-ligand hybridization as compared to the nonbonding localized $4f^1$ state. Stabilization energy in CeO₂ relative to the pure $4f^0$ configuration is 2.2 eV as in Table I. We may infer that the anomalous magnetic behavior of mixed-valence Ce are largely due to low-lying magnetic excited states of the 4f ion rather than from the ground state itself. In the case of CeO₂, $4f^1L^{-1}$ states, which could be magnetic, have been calculated to be ~2.3 eV above the ground state and in fact no anomalous magnetism has been observed experimentally.¹⁵

The parameters for the core-hole state obtained here, $E'(4f^0 \rightarrow 4f^1L^{-1})$ and $E'(4f^1L^{-1} \rightarrow 4f^2L^{-2})$, are somewhat different from those in metallic mixed-valence Ce systems: In CePd₃, $E'(4f^0 \rightarrow 4f^1L^{-1}) \sim -10 \text{ eV}$,^{32,33} as compared to ~ -14 eV in CeO₂. This would largely be due to efficient screening of the core hole in metallic systems. On the other hand, in the $L_{\rm III}$ x-ray absorption edge of CeO₂, two features, probably corresponding to the $4f^0$ and $4f^1L^{-1}$ final states, are separated by only ~10 eV.^{9-11,34} As the final state of the x-ray absorption edge is the same as that of XPS except for the presence of an excited 5d electron just above E_F , the smaller separation in the x-ray absorption edge may be ascribed to screening of the core hole by the excited electron. Such a difference between XPS and x-ray absorption is expected to be small for metallic systems, and in fact the same $4f^{0}-4f^{1}$ separation has been observed for CePd₃ (Ref. 11) in XPS and xray absorption. The positive value of $E'(4f^{1}L^{-1})$ $\rightarrow 4f^2L^{-2}$ ~ 1.1 eV is due to a higher Ce $4f^2$ level in CeO₂ as compared to trivalent compounds where $E'(4f^1L^{-1} \rightarrow 4f^2L^{-2})$ is negative and the $4f^2$ satellites are due to shake-down transitions. A main discrepancy between the calculated and experimental spectra is the width of the $4f^2$ signal. [In a deconvoluted spectrum in Ref. 16, the $4f^2$ signal is resolved into two peaks, u'' and u' (v'' and v'), though not very clearly.] This is probably due to multiplet splittings of the $3d^{9}4f^{2}$ final state,³⁵ which is significant but is neglected in the present calculation.

Our conclusion that CeO₂ has ~0.6 of a 4f electron appears to conflict with the simple $4d^{9}4f^{1}$ -like multiplet structures observed in the $4d \rightarrow 4f$ photon absorption.^{35,36} Multiplet structures of the $4f^{9}4f^{2}$ final state clearly seen in γ -Ce and $4f^{1}$ compounds become broad and weak, and $4d^{9}4f^{1}$ -like features develop in going to mixed-valence or tetravalent Ce.^{12,14} This, however, does not necessarily mean that CeO₂ is mainly $4f^{0}$, because the $4d^{9}4f^{2}$ multiplet structures could disappear not only by complete ab-

sence of the 4f electron but also by absence of certain orbital components of Ce 4f: In the ground state of CeO₂ there is essentially only $4f a_{2u}$ electrons (Table I), and therefore the final state is $4d^{9}4f^{1}+4f^{9}4f(a_{2u})^{1}4f^{1}$. The multiplicity of the $4d^{9}4f(a_{2u})^{1}4f^{1}$ configuration is only twice that of $4d^{9}4f^{1}$ resulting in the structure as simple as $4d^{9}4f^{1}$. As for α -Ce, which will be discussed in Sec. VI, although the local symmetry of the $4f^{1}$ configuration is the same as CeO₂, other components of Ce 4f, t_{2u} , and/or t_{1u} would be also present and more complete $4d^{9}4f^{2}$ features have been observed.⁶ Thus we think that the $4d \rightarrow 4f$ absorption structure cannot be used directly as a tool for monitoring the 4f occupancy.

V. CeN

CeN is a mixed-valence material, where $4f^{0}5d^{n}$ $\leftrightarrow 4f^{1}5d^{n-1}$ configuration mixing is thought to give rise to anomalous magnetism and lattice volume²² [6s is omitted in $(5d 6s)^n$ as the Ce 6s states in CeN are well above the Ce 5d states]. However, as can be seen from the CeO_2 results, it would be important to consider also the local configuration mixing $4f^0 \leftrightarrow 4f^1L^{-1}$ (5dⁿ is omitted for brevity) in CeN, as it has an electronic structure similar to that of CeO₂ with the closed N 2p shell apart from the presence of the 5d conduction electrons. In order to assess the importance of the $4f^0 \leftrightarrow 4f^1L^{-1}$ mixing, we try to explain the core-level XPS (Ref. 25) and resonant photoemis $sion^{26}$ spectra by taking into account only the 4*f*-L mixing and by neglecting effects of 5d conduction electrons. (Note that the 4f-5d mixing on a single Ce atom is not allowed by symmetry.) We consider the Ce 4f and N 2p orbitals in the CeN₆¹²⁻ cluster and $l = t_{1u}$ and t_{2u} in Eq. (3). V_2 are V_1 obtained by assuming and $(fp\sigma) = -2(fp\pi) = 0.4$ eV, which is a little smaller than that for La pnictides $[(fp\sigma) = 0.5 \text{ eV}]$ used in Ref. 37. Thus we use $V_1 = 1.38$ eV and $V_2 = -0.98$ eV. In (13) $\overline{V}_1 = V_2$ and $\overline{V}_2 = \sqrt{5/6}V_2$ for the t_{1u} -like hole and $\overline{V}_1 = \sqrt{5/6}V_1$ and $\overline{V}_2 = V_2$ for the t_{2u} -like hole. The XPS spectrum of the Ce 3d core level²⁵ in Fig. 6

The XPS spectrum of the Ce 3d core level²⁵ in Fig. 6 shows $4f^0$ signals as well as the $4f^1$ main lines and the $4f^2$ shake-down satellites, which is characteristic of mixed valency. The valence-band photoemission spectra for photon energy just below and on the $4d \rightarrow 4f$ resonance²⁶ are also shown in the figure and the difference between the two spectra may be thought of as the 4f-derived emission. In the original assignment²⁶ the splitting of the peaks Aand B (C and D) has been attributed to surface effects: B(D) has been attributed to emission from the surfacestabilized $4f^1$ initial state.

The 3d core-level spectrum has been reproduced with the use of parameters $E_0 = -1.6$, $E_1 = -11.4$, and $E_2 = -4.0$ eV as shown in Fig. 6. The 4f occupancy is thus given to be ~0.8. The 4f spectra calculated with the above E_0 , E_1 , and E_2 and $\epsilon_L = -2.2$ eV are shown as a function of E_a in Fig. 7, where as in the case of CeO₂ stronger and weaker lines are, respectively, the main 4f lines and the $L \rightarrow 4f$ satellites. The 4f spectrum with $E_a = 1.4$ eV seems to reproduce well the three peaks B, C, and D and is superposed in Fig. 6. The peak A is not reproduced in the calculated spectrum and would be due to mixing of 4f into the 5d conduction band which has



FIG. 6. Experimental (Refs. 25 and 26) and calculated Ce 3d core-level x-ray photoemission (left-hand side) and valence-band photoemission (right-hand side) spectra of CeN. Valence-band spectra for photon energies just below and on the $4d \rightarrow 4f$ resonance are given by dashed and solid curves, respectively. Calculated spectra are indicated by vertical bars. $E_1 = E'(4f^0 \rightarrow 5f^1L^{-1})$. Δ_{so} is the spin-orbit splitting of Ce 3d. The calculated 4*f*-derived spectrum that is for $E(4f^{0}L^{-1} \rightarrow 4f^{1}L^{-2}) = 1.4 \text{ eV} \text{ and } \epsilon_{L} = -2.2 \text{ eV}.$

not been included in our model. Surface states have not been completely ruled out, however, as we could identify the peaks A, C, and D with the bulk emission and the peak B with the surface 4f level which is not visible in the XPS valence-band spectrum²⁵ due to the low surface sensitivity of XPS. In this case, the 4f-5d hybridized band may not be enhanced very much or may overlap the peak A or B.

The nonmagnetic properties and the small lattice volume of CeN can be understood within the present mixed-valence description by an analogy to CeO₂. The Curie-Weiss behavior at high temperatures²² would be due to localized $4f^1$ excited states, which may be located closely to the ground state due to 4f-5d interactions. In the $4f^0 \leftrightarrow 4f^1L^{-1}$ mixing, the number of conduction electrons is one per Ce atom. The conduction-electron number obtained from the optical measurement²⁷ is a little



FIG. 7. Calculated 4*f*-derived photoemission spectra of CeN as a function of $E_a = E(4f^0L^{-1} \rightarrow 4f^1L^{-2})$. The N 2*p* (t_{1u} and t_{2u}) level (equal to ϵ_L) is shown by arrows.

smaller than this, but we note that this number has been obtained by assuming an effective mass of the 5*d* band and is dependent on the assumed mass. Depopulation of the conduction band at high temperatures has been observed in this optical study,²⁷ which is consistent with our model where localized $4f^1$ states are thermally populated, resulting in the reduced conduction-electron number.

Thus the satellite features in the 4*f*-derived spectrum of CeN are explained by the $4f^0 \leftrightarrow 4f^1L^{-1}$ mixing. However, the ground-state parameter $E_0 = E(4f^0 \rightarrow 4f^1L^{-1})$ = -1.6 eV may be inconsistent with the parameter $E_a = E(4f^0L^{-1} \rightarrow 4f^1L^{-2}) = 1.4$ eV in the valence-hole state, since E_a and E_0 should not differ by more than a Coulomb energy between two L holes, which would be at most ~ 1 eV. Here we note that the calculated 4f spectrum is not sensitive to E_0 (the spectral line positions were independent on E_0 and the intensities did not change very much with E_0 up to $E_0 \sim 1.5$ eV), and therefore the higher E_0 would be more appropriate. The higher E_0 , however, would lead to too large I_{4f^0}/I_{4f^1} core-level intensity ratio, and the extended $4f^{0}5d^n \leftrightarrow 4f^{1}5d^{n-1}$ process in addition to the local $4f^0 \leftrightarrow 4f^1 L^{-1}$ one should be considered in order to give the intense $4f^1$ core-level lines and consequently the large $4f^1$ component in the ground state. The present model could in principle be extended to include the two valence-mixing mechanisms by the use of a large cluster. Such a study would reveal relative importance of the two valence-mixing processes and also make unambiguous assignment of the 4f-derived spectrum in CeN.

VI. γ -to- α PHASE TRANSITION OF Ce METAL

So far several models have been proposed for the γ -to- α transition of the Ce metal: (i) promotion of the 4f electrons to the 5d 6s conduction band in going from the γ to- α phase,³⁸ that is, Ce changes from trivalent (4 f^1) to tetravalent $(4f^0)$, (ii) a Mott transition within the 4f band,³⁹ (iii) an increase of the 5d 6s-4f hybridization in the α phase,⁴⁰ and (iv) a Kondo model, according to which the α phase is identified as a Kondo singlet state.^{41,42} The promotion model is in contradiction to the almost unchanged 4f number across the transition revealed by positron annihilation,² Compton scattering,³ photoemission spectroscopy,⁴⁻⁶ and energy-band calculations,^{7,8,40} and also has been criticized by a thermodynamical argument.³ The Mott transition assumes the constant 4f number but the width of the 4f band W is of the order of 1 eV,^{7,8,40} while the Coulomb repulsion between two 4f electrons on one atom is $U \sim 5 \text{ eV}$,⁴³ which does not satisfy the condition $W \sim U$ for a Mott transition to occur. The Kondo model,^{41,42} on the other hand, is successful in explaining the disappearance of the magnetic moment and the volume collapse in the γ -to- α transition.

Here we would like to describe this transition as a change in the local electronic configuration of the 4f level: Nonmagnetic but essentially $4f^{1}$ -like α -Ce is described by the ground state of the forms (3) and (4). In the case of Ce metal, the smallest cluster relevant to the problem is Ce₁₃, where the 4f orbitals on the central atom and the 5d 6s orbitals on the 12 nearest-neighbor Ce atoms are considered. This is already considerably complicated as compared to the CeO₈- or CeN₆-cluster problem, and we give here only qualitative discussions using a simplified model. The ground state of γ -Ce may be described as

$$\Psi_{\gamma} = f^{\dagger} | 0 \rangle , \qquad (15)$$

where f^{\dagger} creates an electron in the ${}^{2}F_{5/2}$ state of the Ce $4f^{1}$ ion. For α -Ce,

$$\Psi_{\alpha} = c_0 \mid 0^* \rangle + \sum_l c_l \mid l^* \rangle , \qquad (16)$$

$$|l^*\rangle = \sum_{i,j} \Gamma^l_{ij} \sum_{\sigma} f^{\dagger}_{i\sigma} d_{j\sigma} |0^*\rangle , \qquad (17)$$

analogous to Eqs. (3) and (4). Here $|0^*\rangle$ is different from $|0\rangle$ in that there is one more non-4*f* electron in α -Ce added near E_F per Ce atom as compared to γ -Ce. $d_{i\sigma}$ annihilates a 5*d* 6*s* electron on the nearest-neighbor atoms of the central 4*f* ion. For α -Ce we expect $|c_0| \gg |c_l|$, that is, the 4*f* occupancy is close to 1. The energies of the γ and α phases may be roughly estimated as follows: for the γ phase,

$$E_{\gamma} = E(4f^1) , \qquad (18)$$

for the α phase, two configurations, $4f^{15}d^{-1}\epsilon_F$ and $4f^{0}\epsilon_F$, where ϵ_F stands for an extra electron added near E_F . For simplicity we consider only one state for each configuration. Then the energy of the α phase is given by

$$E_{\alpha} = \{E_A + E_B - [(E_A - E_B)^2 + 4V^2]^{1/2}\}/2 , \qquad (19)$$

where $E_A = E(4f^{1}5d^{-1}\epsilon_F)$, $E_B = E(4f^{0}\epsilon_F)$, and V is the matrix element of the Hamiltonian between the two states.

In Fig. 8 we present, according to (18) and (19), the free energy H=E-TS per Ce atom for γ - and α -Ce as a function of the Ce-Ce interatomic distance at room temperature. The local symmetry of the magnetic $4f^1$ configuration in γ -Ce is ${}^2F_{5/2}$, while the $4f^0\epsilon_F$ and $4f^{15}d^{-1}\epsilon_F$ configurations have the ${}^{1}S_0$ (or ${}^{1}A_{1g}$) local symmetry and



FIG. 8. Free energies E - TS of α - and γ -Ce at room temperature as a function of the nearest-neighbor Ce-Ce distance. The $4f^1$ configuration (solid curve) corresponds to γ -Ce, and the bonding state (lower dashed curve) of the $4f^0\epsilon_F$ and $4f^15d^{-1}c_F$ configurations (dotted curves) corresponds to α -Ce. The equilibrium distances for α - and γ -Ce are shown by arrows.

are mixed with each other via V. V increases rapidly as the lattice parameter decreases. The energies of $4f^1$ and $4f^{1}5d^{-1}\epsilon_{F}$ are expected to have nearly the same lattice parameter dependence due to similar configurations with one 4f. The $4f^0 \epsilon_F$ energy is higher than $4f^1$ or $4f^{15}d^{-1}\epsilon_F$ by the $4f \rightarrow 5d$ 6s promotion energy (~2 eV) and has a minimum for a lattice parameter smaller than the $4f^1$ configurations. The $4f^{15}d^{-1}\epsilon_F$ energy is lowered by forming a 5d-4f bond through V according to Eq. (19) as is shown in Fig. 8, and produces a minimum at a smaller lattice volume, which we identify as the α phase. In the figure we also present an antibonding counterpart of the bonding $4f^{1}5d^{-1}\epsilon_{F} + 4f^{0}\epsilon_{F}$ state. One can see that α -Ce is almost $4f^{1}5d^{-1}\epsilon_{F}$ -like, and therefore the 4f occupancy is close to unity (in this simplified model the 4fnumber is calculated to be ~0.9). The γ -to- α transition is thus clearly a first-order phase transition accompanied by a change in the symmetry of electronic state and by a volume collapse. We have assumed volume dependence of the energies of the unhybridized configurations and V as

$$E(4f^{1}) = 4.43[(3.64/d)^{5.06} - 2(3.64/d)^{2.53}], \qquad (20)$$

$$E(4f^{1}5d^{-1}\epsilon_{F}) = E(4f^{1}) + a$$
, (21)

$$E(4f^{0}\epsilon_{F}) = 4.43\{[3.64/(d+0.2)]^{5.06}$$

$$-2[3.64/(d+0.2)]^{2.53}+b$$
, (22)

$$V = V_0 (3.64/d)^6 , (23)$$

where the Ce-Ce distance d is in Å and energies in eV. The exponents, prefactors, etc., in Eqs. (20)-(22) are so chosen as to give the elastic constant, cohesive energy, and equilibrium lattice parameter of the γ phase. The d^{-6} law of V is due to the interatomic distance dependence of the d-f transfer integrals.⁴⁴ We have chosen the parameters in Eqs. (21)-(23) as $V_0=0.6$, a=0.22, and b=2 eV, so that the α phase is more stable by ~ 1 kcal (Ref. 45) with the lattice parameter smaller by ~ 0.2 Å than the α phase. As the temperature is lowered, the $4f^1$ free energy rises owing to the entropy term TS ($S=k \ln 6$ for γ -Ce with a $J=\frac{5}{2}$ magnetic moment and $S \sim 0$ for α -Ce) and only the α phase is realized, which agrees with the phase diagram of Ce metal.³⁹

Resonant photoemission from the 4f level⁴⁻⁶ has shown two peaks (near E_F and at $\sim 2 \text{ eV}$) and their relative intensity changes across the γ -to- α transition. In the case of the localized $4f^1$ state (γ -Ce), resonant photoemission takes place via process $4d^{10}4f^1 + h\nu \rightarrow 4d^{9}4f^2$ $\rightarrow 4d^{10}4f^0 + \epsilon l$ followed by a satellite transition from the $4f^0$ -hole to a ligand- or 5d-hole state. For the mixedvalence case (α -Ce),

$$(4d^{10}4f^{0}\epsilon_{F} + 4d^{10}4f^{1}5d^{-1}\epsilon_{F}) + h\nu \rightarrow 4d^{9}4f^{1}\epsilon_{F} + 4d^{9}4f^{2}5d^{-1}\epsilon_{F} ,$$

then only the $4f^2$ part leads to a SCK decay

$$4d^94f^25d^{-1} \rightarrow 4d^{10}4f^05d^{-1}\epsilon_F + \epsilon l$$

For Ce metal, where the ligand level Ce 5d is not in a closed-shell configuration as the O 2p (N 2p) levels in CeO₂ (CeN), the above two cases make little difference, and the model for the 4f-derived photoemission presented in I may be applicable. According to this, the resonant photoemission spectra of γ - and α -Ce may be described as follows: The energy separation of the two 4f peaks is given by $(\Delta E^2 + 4V^2)^{1/2}$, where $\Delta E = E(4f^0 \rightarrow 4f^{15}d^{-1})$ for γ -Ce and $\Delta E = E(4f^{0}5d^{-1}\epsilon_F \rightarrow 4f^{15}d^{-2}\epsilon_F)$ for α -Ce. As $E(4f^0\epsilon_F \rightarrow 4f^{15}d^{-1}\epsilon_F) \sim -2$ eV in the ground state, we estimate $E(4f^0 \rightarrow 4f^{15}d^{-1}) \sim -2$ eV whose absolute value is larger than V, which explains why the 2-eV emission is stronger than the emission just below E_F in γ -Ce. On going from γ - to α -Ce, V increases according to (23), which results in an increase of the E_F -peak intensity with little change in the separation of the two peaks.

The present model for the 4f satellites is similar to a conceptual model by Hüfner and Steiner⁴⁶ or a model Hamiltonian study by Gunnarson and Schönhammer⁴⁷ in that the screening of the $4f^0$ hole takes place by the 4felectron. A recent model proposed by Liu and Ho⁴⁸ explains the 4f satellites in terms of the screening of the $4f^0$ hole by the 5d conduction electron. According to these authors, a localized 5d state splits off from the 5d 6s band due to the attractive potential of the $4f^0$ hole, which is either filled or empty in the final state giving rise to the two peaks. However, no such localized state is expected as can be seen from the core-level XPS spectra, where the core hole is screened by the $5d\,6s$ electron in the poorly screened peak and by the 4f electron in the well-screened peak, and no splitting into 5d 6s-screened and -unscreened peaks is observed, as would be expected from the model of Liu and Ho.

Finally, we compare our α -to- γ transition model with other models. The promotion model³⁸ is evidently incompatible with our model, since we have shown that the 4foccupancy is close to unity for α -Ce. The Mott transition is also not compatible with our model as the 4f-5d 6s hybridization V is of the order of $\sim 1 \text{ eV}$, which may be sufficiently large to stabilize the nonmagnetic configuration, but is far too small to form an itinerant 4f band since $V \sim W \ll U$. As has been pointed out by the band calculations,^{7,8,40} the 4*f*-band width in the collapsed α phase of Ce or tetravalent Ce compounds is significantly increased, and in some tetravalent compounds possibility of the formation of the 4f band has been pointed out.¹⁴ It is noted, however, that electrons in such a 4f band, if it exists, are strongly correlated and are not appropriately described within a usual one-electron energy-band picture. In the Kondo mechanism^{41,42} also, the 4f-5d 6s hybridization is essential. The singlet state in α -Ce described here may be analogous to the Kondo singlet state. The stabilization mechanism of our singlet state is hybridization between two local configurations $4f^0$ and $4f^{15}d^{-1}$, while in the Kondo model⁴² the singlet state is stabilized by the 4f-5d 6s exchange interaction which is enhanced in the α phase by increased hybridization. Thus similar pictures seem to have emerged from the two quite different approaches. We note, however, a basic difference between the two approaches is that, in the Kondo model interactions of the 4f level with conduction electrons, particularly near E_F , are important while in our model interactions with localized (and , for CeO2 and CeN, closed-shell) orbitals are considered. Therefore, the DOS at E_F which is an

important parameter in the Kondo model does not enter our model, which might explain dense Kondo behaviors, e.g., in CeB₆ (Ref. 49), where a magnetic moment is quenched with relatively low concentration of electrons at E_F . A drawback of our model is obviously that it cannot treat properly itinerant (e.g., 5d 6s) electrons, thus we cannot study their contributions to electrical conduction, specific heats, etc., whereas it describes local correlation effects of the 4f ions in a realistic way.

VII. DISCUSSION

Generally in metallic mixed-valence Ce systems, the two configurations $4f^{0}(5d 6s)^{n}$ and $4f^{1}(5d 6s)^{n-1}$ have been thought to be degenerate. The present local mechanism $4f^0 \rightarrow 4f^1L^{-1}$ explains mixed valency in insulating compounds and also part of valence mixing in metallic compounds. This results from the large 4f-L hybridization in Ce compounds, where the 4f orbital is the most extended in the rare-earth series. Importance of the p-fmixing, where p corresponds to L here, as compared to the f-d mixing in stable-valence Ce pnictides has been pointed out by Kasuya and co-workers.^{37,50} In the $4f^0 \leftrightarrow 4f^1L^{-1}$ valence-mixing mechanism strong hybridization between 4f and localized orbital L quenches the magnetic moment. In the $4f^{0}(5d 6s)^{n} \leftrightarrow 4f^{1}(5d 6s)^{n-1}$ mixing, on the other hand, each matrix element for the 4f-5d mixing may be small, but interactions involving conduction-band levels near E_F need only low virtual excitation energies and therefore can be significant. In the case of metallic systems such as CePd₃, α -Ce, etc., we cannot separate unambiguously the two mechanisms, since the energy band formed by orbitals on neighboring atoms crosses E_F and shows either localized or extended characters. The local $4f^0 \leftrightarrow 4f^1L^{-1}$ process would particularly be important in the case of a narrow L band, particularly a closed-shell Lband, as interatomic Coulomb interactions lower the energy of a localized electron-hole pair $4f^{1}L^{-1}$ as compared to a well-separated pair. This interatomic Coulomb energy has not been incorporated in theories which use Anderson Hamiltonians.^{24,41,42,47,51}

In the present paper we have treated electron correlations within the small cluster. This is a good approximation for localized excitations, namely core-hole states, 4fhole states, etc., but might not be a good approximation for the ground state where equivalent rare-earth ions interact with each other. It is considerably difficult to treat correlation effects in a large system, however. On the other hand, there are a number of experiments showing that a single rare-earth ion exhibits mixed-valence behavior in a dilute system.^{52,53} This fact means that the valence mixing is essentially of the single-ion nature and that the present model which uses the small clusters is relevant to the valence-mixing phenomena.

Although the one-electron theory such as energy-band calculations and molecular-orbital calculations can predict certain ground-state properties of narrow band systems such as Ce compounds, physical properties associated with excited states have to be treated beyond the oneelectron picture. This is most clearly seen in Fig. 5, where the 4f-derived photoemission spectra are distributed in a wider energy range than the initial-state one-electron Ce 4f-like partial DOS. This can be understood as follows: While in the molecular-orbital levels Ce 4f have a width of the order of the Ce 4f-O 2p transfer integral W, the Ce 4f photoemission has a width of $\sim n^{1/2}W$, where n is the number of overlapping Ce 4f-O 2p pairs per Ce ion, including spin degeneracy. Thus information obtained by resonant photoemission is preferentially for local electronic structure of the Ce ion rather than for, if it exists, extended band states.

This local electronic structure is directly related to the $4f^0$ -hole state produced by the SCK decay of the $4d^94f^2$ state $(4d^94f^2 \rightarrow 4d^{10}4f^0)$ followed by satellite transitions, e.g., $L \rightarrow 4f$. A sum rule

$$\epsilon_{4f} = \int A_{4f}(\epsilon)\epsilon \,d\epsilon \tag{24}$$

gives the pure $4f^{0}$ -hole energy ϵ_{4f} in analogy to the sum rule for the core-level photoionization.⁵⁴ Here $A_{4f}(\epsilon)$ is the 4f spectral density which is the difference between the spectral intensity at the $4d \rightarrow 4f$ resonance and that at the antiresonance. Equation (24) combined with the resonant photoemission spectra⁴⁻⁶ suggests that in γ - and α -Ce ϵ_{4f} is 1–2 eV below E_F , with a slight shift to lower binding energies in going from γ -Ce to α -Ce. For trivalent $4f^1$ compounds as was studied in I, the 4f energy thus obtained bears connection to the ground-state 4f level. But in mixed-valence systems ϵ_{4f} is the energy of a localized $4f^0$ hole and it is not straightforward to relate ϵ_{4f} with a ground-state energy level.

In the mixed-valence system the 4f character might be present in delocalized (bandlike) states via hybridization between 4f and delocalized states. One-electron 4f-like partial DOS as obtained by band calculations might have some relevance to this 4f character, except for the high-4f DOS region.⁵⁵ For 4*f*-band states, one would expect to observe valence-band satellites as in Ni, namely localized and itinerant 4f holes.^{56,57} If this is the case, quite different resonant enhancement as a function of photon energy is expected for the two types of hole states, since resonance photoemission enhances localized final states selectively. [In localized $4f^1$ Ce pnictides (I), the same enhancement has been observed for resonant $(h\nu \sim 120 \text{ eV})$ and nonresonant $(h\nu \sim 50 \text{ eV})$ (Ref. 58) spectra, which is consistent with the 4f hole localized on one Ce ion.] On the other hand, we note that in either case of the localized, mixed-valence and bandlike 4f states, intensities of core levels $(4f^0, 4f^1, \text{ and } 4f^2)$ are expected to give useful quantitative information on the 4f number in the initial ground state, as the strong core-hole potential causes localization of the 4f electrons.

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