Prediction of a new phase in solid N_2 at high pressure

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A pressure-induced structural phase transition from the γ structure to the $R \ 3m$ structure (ϵ -N₂) of solid N₂ is predicted. Lattice-energy calculations using an exp-6-1 potential form fitted to the *ab initio* results of Berns and van der Avoird show that the transition is located at $P \sim 40$ kbar at zero temperature with a volume change of 0.2 cm³/mole. The calculated P-V relation and the lattice parameters are presented.

I. INTRODUCTION

In addition to the three well-known α , β , and γ phases of solid N₂, a new phase δ has recently been discovered above P = 49 kbar and room temperature.¹ The crystal structure of δ -N₂ has been determined² to be cubic Pm 3n with eight molecules per unit cell, where the molecules are orientationally disordered and occupy T_d (two molecules) and D_{2d} (six molecules) symmetry sites. This structure is identical to that of γ -O₂ and β -F₂ which exist at normal pressure. The appearance of the orientationally disordered Pm 3n structure, instead of the β -N₂ structure (hexagonal P6₃/mmc), is attributed to the small quadrupole moments of O₂ and F₂, Q = -0.39 $\times 10^{-26}$ and 0.9×10^{-26} esu cm², respectively.³ By comparison, $Q = -1.4 \times 10^{-26}$ esu cm² for N₂.³ The existence of the Pm 3n, δ -N₂ structure in solid N₂ implies that the anisotropy of the electrostatic quadrupole-quadrupole (EQQ) potential is dominated by the repulsive overlap potential at high pressures. This is not surprising since the EQQ energy has a relatively weak volume dependence. The above information and a comparison of the phase diagram with those of O_2 and F_2 suggest that, in solid N_2 at low temperature and high pressure, there may exist an orientationally ordered phase whose structure is identical to one of the low-temperature phases of O_2 or F_2 .^{4–7} Since α -O₂ is stabilized by the magnetic interaction and α -F₂ is apparently stabilized by chargetransfer interactions, the only relevant crystal struc-

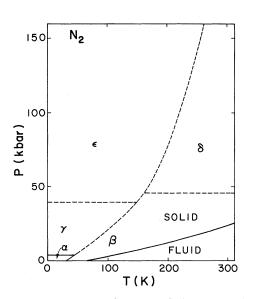


FIG. 1. Phase diagram of solid N_2 . Solid curves are experimentally established and dashed curves are tentative.

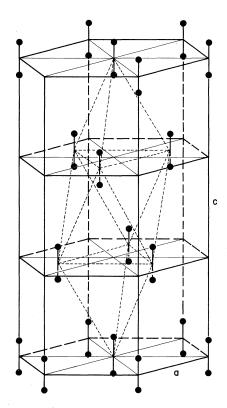


FIG. 2. $R \bar{3}m$ crystal structure proposed for ϵ -N₂.

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ture for N₂ is that of β -O₂.

Based upon this information and because β -O₂ is a very closed packed structure for rodlike molecules,⁴ we predict⁸ that a phase transition like the β - γ transition in solid O_2 will also occur for solid N_2 at low temperature and high pressure, from δ -N₂ into an orientationally ordered state with a β -O₂ structure. We call this new phase ϵ -N₂. The purpose of this article is to show that a transition from γ -N₂ into ϵ -N₂ at high pressure and zero temperature is predicted on the basis of a simple calculation. Figure 1 shows the phase diagram of N₂ including the proposed region of stability for the new ϵ phase. The solid curves are experimentally established and the dashed curves represent tentative boundaries. The proposed β -O₂ crystal structure for ϵ -N₂ is shown in Fig. 2. It is rohomboliedral $R \overline{3}m$ with one molecule per unit cell where the molecules are oriented parallel to the c axis.

II. METHOD

Lattice-energy calculations on the α , γ , and the predicted $R \overline{3}m$, ϵ -N₂ structure of solid N₂ have been made using an atom-atom exp-6-1-type representation⁹ of *ab initio* molecular orbital results obtained for the interaction between a pair of N₂ molecules by Berns and van der Avoird.¹⁰ In this model, the EQQ potential is represented by point charges distributed linearly along this molecular axis. Calculated results for the sublimation energy and lattice vibrational frequencies in α - and γ -N₂ are in good agreement with experiment,⁹ considering that the potential parameters are constrained to best fit the ab initio data and are otherwise not adjustable. In our calculation, the lattice energy is minimized with respect to the lattice parameters and the intramolecular bond length at each volume using a pattern search optimization method that has been described elsewhere.¹¹ Lattice sums are taken out to 12 Å. Zeropoint vibrational energies are neglected. Pressures were determined by taking the derivative of energyvolume results.

III. RESULTS AND DISCUSSION

Figure 3 shows the calculated lattice energies for γ and ϵ -N₂ versus molar volume V. The two curves intersect at $V = 19.0 \text{ cm}^3/\text{mole}$, indicating the existence of a γ - ϵ phase transition. The P-V relations for γ and ϵ -N₂ are shown in Fig. 4, and the calculated lattice parameters are plotted in Fig. 5. The γ - ϵ transition takes place at $P_t = 39.6$ kbar and the volume change associated with the transition is about $0.2 \pm 0.05 \text{ cm}^3/\text{mole}$. The volume change obtained from the common tangent of the E-V curves and from the P-V curves agree to within 0.05 cm³/mole.

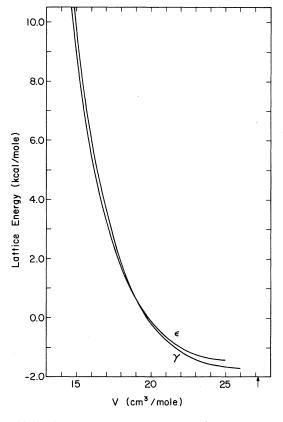


FIG. 3. Calculated static lattice energies for γ -N₂ and the $R\bar{3}m$, ϵ -N₂ structures vs molar volume. The arrow indicates the observed volume (Ref. 12) of α -N₂, 27.1 cm³/mole, at normal pressure.

Results obtained using the empirical 6-12 + EQQ potential by Thiéry and Chandrasekharan¹² also indicate a transition at a somewhat higher pressure.

The EQQ interaction appears to be important in stabilizing α - and γ -N₂ at low temperatures. For example, at $V = 19 \text{ cm}^3/\text{mole}$, the EQQ energy is -0.28kcal/mole for both α - and γ -N₂, whereas it is about +0.15 kcal/mole for the $R \overline{3}m$ structure of the proposed ϵ phase. As a consequence, the ϵ phase cannot become stable until a volume is reached for which the overlap energy associated with the ϵ structure is sufficiently less repulsive than in γ -N₂ that it overcomes the unfavorable EQQ energy difference. Thus it is the repulsive interaction that destabilizes the γ phase due to compression. To support this argument, we have used the exp-6 atom-atom potential of Kuan et al., ¹³ which does not include an EQQ interaction. In this model, the $R \overline{3}m \epsilon - N_2$ structure is more stable than both α - and γ -N₂ at all volumes considered. This is consistent with the conclusion of English and Venables.¹⁴ These results demonstrate the importance of the EQQ interaction to the stability of α - and γ -N₂. Similar conclusions

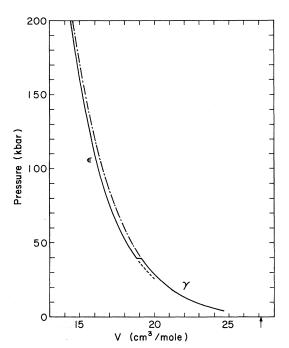


FIG. 4. Calculated P-V relation for γ - and ϵ -N₂. The arrow has the same meaning as in Fig. 3. The dash-dotted and dashed curves are the continuations for γ and ϵ phases, respectively.

have been obtained for β -N₂ using computer simulations.¹⁵

The atom-atom representation of the *ab initio* potential of Berns and van der Avoird^{9, 10} does not correctly resolve the stability of γ -N₂ over α -N₂ at $V \ge 24$ cm³/mole and $P \ge 3.6$ kbar,¹⁶ as is the case for all atom-atom-type N₂ potentials. Because of this widely recognized flaw in the potential form,¹⁷ the ϵ phase is predicted to become stable with respect to α -N₂ at $V \simeq 14.4$ cm³/mole and $P \simeq 200$ kbar. This transition, of course, is not physical but it does point out the ultimate stability of ϵ -N₂ at high pressures.

It is to be emphasized that, while the calculations show very clearly that a transition into ϵ -N₂ is expected, the quantitative reliability of the results depends both on the quality of the potential and on the calculational model. Except for the neglect of zero-point energy, which is of negligible importance compared to the static lattice energy at high pressures, the calculational model is nearly exact. The *ab initio* data¹⁰ and the atom-atom fit⁹ to that data are of high quality, but uncertainty still exists in the pair potential which makes numerical predictions of the phase tran-

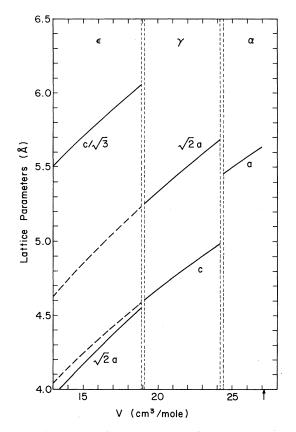


FIG. 5. Calculated lattice parameters for α -, γ -, and ϵ -N₂. The dashed curves are the parameters for γ -N₂. The arrow has the same meaning as in Fig. 3.

sition reliable to only about a factor of 2 in pressure. Experiments are in progress at our institution and at the Los Alamos Scientific Laboratory that should test the predictions made in this work.

Note added in proof. D. Schiferal and R. L. Mills of Los Alamos Scientific Laboratories have just found evidence from Raman scattering of a new phase in solid N₂ at low temperature above ~ 20 kbars. The spectrum is similar to that observed in the *Pm*3*n* structure. If this structure is confirmed, we predict that a transition into β -O₂ phase will occur at higher pressures.

ACKNOWLEDGMENTS

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