Shake-up excitations and core-hole screening in Eu systems

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X-ray photoemission (XPS) spectra of the Eu 3d and Eu 4d core levels in Eu metal and in divalent and trivalent Eu compounds exhibit splittings into two components, caused by the occurrence of $4f^6$ and $4f^7$ final-state configurations. The observed phenomena are described in terms of final-state shake-up and shake-down processes accompanied by a decrease, respectively, increase of the 4f occupation number. An extension of the Kotani-Toyozawa model to fand d states gives a consistent description of both processes.

Quite frequently, core-level photoemission (PE) spectra of light rare-earth (RE) solids exhibit—in addition to well-known multiplet splittings—a characteristic satellite structure, which has been attributed to final-state effects.¹⁻⁴ Mainly on relative intensity arguments high-binding-energy (shake-up) and lowbinding-energy (shake-down) satellites were identified. Recently, the underlying mechanism for these processes became better understood: The effective increase of the core potential during photoionization causes a deformation of the local density of states, which in turn can lead to a change of the 4f occupation number in the final state.^{5,6}

The case usually encountered is characterized by an increase in the 4f occupation number in the final state. Depending on the final-state position of the unoccupied 4f state relative to the valence-band-edge energy-gain (shake-down) or energy-loss (shake-up) satellites have been discussed.² Assuming an increase in the 4f occupation number to be energetically favored Fuggle et al.⁴ attributed, e.g., the observed 3d structures of Ce compounds to well-screened (low-binding-energy) and poorly screened (highbinding-energy) states. A second process for shake up was proposed by Wertheim and Campagna⁷ according to which the number of 4f electrons remains constant, but an excitation to higher multiplet levels within the 4f shell takes place. Such a shake-up process has not been identified so far. A third possibility for a shake-up process is given by a decrease in the 4foccupation number connected with the transition of a localized 4f electron to an unoccupied valence state. This process has not yet been considered in interpreting RE XPS (x-ray photoemission) spectra. In this Communication we show the occurrence of such a shake-up excitation and a corresponding shake-down process in integral-valent Eu compounds. On the basis of split $3d_{5/2}$ and 4d core levels the presence of both $4f^6$ and $4f^7$ final-state configurations are identified in Eu^{2+} and Eu^{3+} compounds. Support for this identification is given by theory: Calculations by Johansson⁸ and by Herbst and Wilkins⁹ agree well with the observed splittings. It is shown that an extension of the Kotani-Toyozawa (KT) model⁵ to fand d states gives a consistent description of shake-up and shake-down processes in core-level photoemission. Finally, implications of this model for the interpretation of XPS spectra of Ce compounds are discussed.

The XPS measurements were performed with a VG-ESCA-3 spectrometer with a total resolution of $\simeq 1$ eV, which is mainly due to the linewidth of the Al $K\alpha$ x rays. The sample surfaces were mechanically cleaned in a vacuum of 4×10^{-10} Torr, and this process was frequently repeated to keep the oxygen contamination below the limit of detectability (by monitoring the O1s signal). The samples of EuAl₂, $EuPd_3$, and $EuRh_6B_4$ were prepared by arc melting in Ar atmosphere and were characterized as described in Refs. 10 and 11. Eu_2O_3 surfaces were prepared by oxidation of the EuPd₃ sample and independently by using a pressed-powder sample of Eu₂O₃. In order to determine the initial-state valence of the studied Eu solids, ¹⁵¹Eu Mössbauer spectra and magnetization data were taken. These measurements clearly show that the studied materials have integral valences in the ground state (Eu²⁺:Eu, EuAl₂; Eu³⁺:EuPd₃, $EuRh_6B_4$, Eu_2O_3).

XPS spectra of the Eu $3d_{5/2}$ core-level region are shown in Fig. 1 for the five studied Eu solids. It is striking that both for Eu^{2+} and Eu^{3+} systems the $3d_{5/2}$ line is split into two compounds (by 7.8 \pm 0.2 eV in Eu metal and by 9.0 \pm 0.2 eV in Eu₂O₃), with varying relative intensities. The shapes and widths of the two components are somewhat different, which can be expected due to multiplet splitting for two different final-state 4 f configurations. Recently, Crecelius et al.² and Herbst et al.³ have shown that due to the stability of the half-filled 4f shell in Eu²⁺ the $4f^7$ and $4f^8$ final states coincide energetically within the linewidth of the Eu $3d_{5/2}$ core level. Therefore, the observed splittings of the core-level XPS lines in Eu^{2+} and Eu^{3+} compounds must be interpreted as a consequence of Eu $4f^6$ and Eu $4f^7$ final-state configurations. A further confirmation for the correctness of this identification is given by the magnitude of the

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FIG. 1. XPS spectra of the $\text{Eu} 3d_{5/2}$ region in EuAl_2 , Eu metal (Ref. 19), EuPd₃, EuRh₆B₄, and Eu₂O₃, excited with Al $K\alpha$ x rays. The $K\alpha_{3,4}$ satellites were numerically removed. The structure in the EuAl₂ spectrum at a binding energy of 1138 eV is due to plasmon loss. The Eu₂O₃ spectrum shown was obtained with a pressed-powder sample.

observed splittings. Using an equivalent-core argument, in which trivalent core-ionized Eu is represented by tetravalent Gd, the energy separation between the final-state Eu^{3+} and Eu^{2+} components should be given in first order by the energy difference between Gd^{4+} and Gd^{3+} for the fully relaxed state. Johansson⁸ calculated this splitting as 7.8 eV, in excellent agreement with the present results. Herbst and Wilkins⁹ using a renormalized-atom scheme obtained for the splitting of the Eu 3d core level a value of 9.6 eV, which is slightly larger than the splittings observed for the metallic Eu solids. The somewhat larger splitting found for Eu₂O₃ indicates the absence of final-state conduction-electron screening in this insulator. Since in metallic Eu³⁺ systems a larger screening shift is expected for the $4f^6$, as compared to the $4f^7$ configuration, the splitting should be reduced in metallic systems as compared to insulating solids.

Since the initial-state configuration of divalent Eu is $4f^7$ the XPS spectra of divalent Eu metal and to a less extent of EuAl₂ show the existence of a *shake-up* process characterized by a *decrease* of the 4f occupation number. In the trivalent Eu compounds studied (EuPd₃, EuRh₆B₄, Eu₂O₃), on the other hand, *shake-down* satellites are observed, which are due to an *increase* of the 4f occupation number.

The shake-down satellites observed for the trivalent Eu compounds may contain contributions from an initial-state surface-induced valence transition to the divalent state.¹² We have studied such a possibility by two methods, (i) by varying the electron take-off angle relative to the surface normal (which changes the effective electron escape depth) and (ii) by taking into account the different electron escape depths λ of the photoelectrons from 3*d*- and 4*d*-core levels (Fig. 2) (a ratio of $\lambda_{4d}/\lambda_{3d} \simeq 3$ can be expected for the kinetic energies resulting from excitation with Al $K\alpha$ x rays).¹³ While on the basis of this analysis contributions from surface-induced valence transitions can be neglected for Eu₂O₃ and $EuRh_{6}B_{4}$ a possible contribution could not be excluded for $EuPd_3$. (A detailed analysis where both, a shake-up process in the divalent surface layer and a shake-down process in the trivalent bulk are considered, will be published elsewhere.) For the shake-up satellites observed for divalent Eu metal (and EuAl₂), however, such contributions due to a surface-induced valence change mechanism can be excluded, since the change in coordination at the surface would always favor the lower-valent state of the rare-earth ion.14

Both the shake-up and the shake-down processes can be consistently described within an extension of the KT model to f and d states. For simplicity let us consider a RE system with energetically degenerate fand d states (this situation is realized in an intermediate-valent system¹⁵). Then we have the following



FIG. 2. XPS spectra of the Eu 4*d* region in EuPd₃, EuRh₆B₄, and Eu₂O₃ excited with Al $K\alpha$ x rays. The $K\alpha_{3,4}$ satellites were numerically removed. Note the similarity of these spectra of stable valent Eu³⁺ compounds with those presented in Ref. 20 for the mixed-valence compounds EuRh₂ and EuRh_{2-x}Pt_x.

possible initial-state configurations

 $R f^{n}(5d6s)^{m+1},$ (1)

 $R f^{n+1}(5d6s)^m$, (2)

where R denotes the Xe core, and n and m stand for f and d occupation numbers. Describing the PE process in the light of the KT model the potential of the

photohole leads to a deformation of the local density of states, due to lowering of states relative to the Fermi energy. If an unoccupied state is pulled down below the Fermi level in the final state, a finite occupation probability depending on the fds hybridization, the local density of states, and the position of initial and final states relative to the Fermi energy will exist. A core-level photoionization of initial state (1) yields the following final states (neglecting extra-atomic relaxation)

$$R^+4f^n(5d6s)^{m+1}h_f + e^-, (3)$$

$$R^+4f^{n+1}(5d6s)^m h_d + e^-, (4)$$

where R^+ denotes the ionized core and h_f , h_d denote the f and d holes, respectively. Here, under the influence of the photohole, an initially unoccupied 4f state (h_f) is stabilized below the Fermi energy (3) and populated in Eq. (4), leaving behind a conduction electron hole h_d . The transition from Eqs. (3) to (4) corresponds to final-state screening accompanied by an increase in f occupation number. However, the transition from Eqs. (3) to (4) is not characterized by a simple recombination but by an equilibrium between the occupation of 4f and 5d states given by definite transition probabilities.

Let us now consider a core-level photoionization of initial state (2). Here a d state is assumed to be slightly above the Fermi energy. This state can now be pulled down by the increased core potential and is coupled with the corresponding 4f state in the same way as the unoccupied 4f state is coupled with the occupied d state in Eq. (1). Thus we should arrive at the same final states (3) and (4) with the same occupation probabilities. The transition from Eqs. (2) to (3) corresponds then to a final-state shake-up process accompanied by a decrease in 4f occupation number.

In the case of a stable configuration, the states (1) and (2) are separated by different KT parameters (hybridization, local density of states, etc.). Thus depending on the initial-state configurations (1) or (2) different transition probabilities to states (3) or (4) are expected. For both $4f^6$ and $4f^7$ initial states, however, the two final-state configurations (3) and (4) are expected and are indeed observed.

In the picture of Fuggle *et al.*⁴ the peaks belonging to the $4f^6$ and $4f^7$ Eu configurations would be labeled as poorly screened and well-screened peaks. In the case of divalent Eu metal, however, this term is somewhat misleading, since the $4f^6$ final state in Eu metal (with both the 3*d* core hole and the hole in the 4f shell) may be better characterized as a two-hole final state. Then the connection of this shake-up phenomenon to the well-known Ni satellite¹⁶ and the recently discussed two-hole final states in U compounds¹⁷ is evident.

Let us finally consider a further unoccupied f state slightly above the Fermi level in the initial state.

Then, starting from state (2), an additional *f*-screened final state is possible

$$R^{+}4f^{n+2}(5d6s)^{m-1}, (5)$$

leading to three energetically distinct final states (3), (4), and (5) in the PE spectrum. This situation may be present in Ce³⁺ compounds where such a threefold splitting has been observed in the $3d_{5/2}$ corelevel spectra.¹⁸ Then, the f^0 component, observed, e.g., in CePd₃ may be interpreted as being due to a

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final-state shake-up process accompanied by a decrease in f occupation number rather than being indicative for Ce⁴⁺ in the initial state.

ACKNOWLEDGMENTS

The contribution of Dr. B. Perscheid in taking the Mössbauer data is gratefully acknowledged. This work was supported by the Sonderforschungsbereich 161 of the Deutsche Forschungsgemeinschaft.

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