

Molecular H₂: Nuclear-spin-relaxation centers for protons in *a*-Si:H

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It is proposed that dilute H₂ molecules are the source of proton spin-lattice relaxation in those *a*-Si:H samples which show a T_1 minimum. Two-phonon Raman processes relax the H₂ molecular rotational states, modulating the intramolecular dipolar interactions and relaxing the protons of the H₂. The majority of protons in the sample then relax to the H₂ by spin diffusion. The model explains the reported temperature and frequency variations of T_1 as well as the change in T_1 upon dilution of H with D.

Proton NMR observations in *a*-Si:H have many puzzling aspects. Some samples exhibit a T_1 minimum below 30 K; others do not. In those that do show a minimum, the T_1 variation is not accompanied by any motional narrowing of the two-component NMR line.¹ Such behavior would not be unusual for proton relaxation by paramagnetic centers. However, there is no apparent connection in good-quality *a*-Si:H material between the proton T_1 and the electron-spin density determined by ESR.² Carlos and Taylor¹ have suggested that local disorder modes serve as relaxation centers for the protons. Motion in such disorder modes is assumed to modulate dipolar fields, relaxing nearby protons and producing a sharply asymmetric T_1 minimum, on a T^{-1} scale. However, Moghavar and Schweitzer³ have pointed out that the depth of the observed T_1 minima would require that an unreasonably large fraction (10%) of the protons participate in disorder modes. Reimer, Vaughan, and Knights² have reported that dilution of the protons by replacement with deuterons increases the proton T_1 , suggesting that spin diffusion limits the relaxation. However, such a spin-diffusion limited rate ought to be temperature independent, contrary to experiment. Two more aspects of the observed proton relaxation which must be explained are the frequency independence^{1,4} of the minimum T_1 and the lack of a single apparent energy of activation.¹

Figure 1 shows the 42.3- and 12.3-MHz proton T_1 data reported for a typical sample by Carlos and Taylor¹ replotted as $\ln T_1$ vs $\ln T$. It is apparent that the 42.3-MHz data approach a T^2 temperature dependence (indicated by the dashed lines at high temperatures) above the temperature of the minimum and below 200 K. The temperature variation of the 42.3-MHz T_1 at temperatures below the T_1 minimum

is more rapid than T^{-2} . The overall temperature dependence below 200 K is suggestive of two-phonon Raman relaxation.⁵ We propose that the proton relaxation of Fig. 1 reflects the presence of a dilute system of fast-relaxing H₂ molecules which serve as relaxation centers. The protons of an ortho-H₂

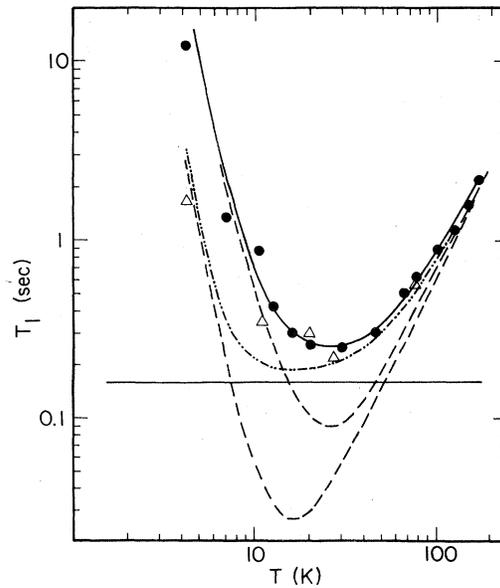


FIG. 1. Proton relaxation times below 200 K reported by Carlos and Taylor (Ref. 1) for *a*-Si:H at 42.3 MHz (solid dots) and at 12.3 MHz (triangles). The solid and dot-dashed lines are fits of a proposed relaxation model to the data. The horizontal line indicates a spin-diffusion bottleneck $T_1=0.16$ sec and the dashed lines indicate the quantity T_1^\dagger calculated for the two frequencies and the case of rapid spin diffusion.

molecule (para-H₂ has zero nuclear spin) are relaxed rapidly by modulation of their dipole-dipole interaction by the rotational motion of the molecule. The remaining great majority of protons relax by spin diffusion to these *o*-H₂ molecules. The temperature dependence of the observed T_1 thus is determined by the temperature dependence of the *o*-H₂ T_1 . However, when T_1 of the *o*-H₂ becomes very short, as it does near the observed T_1 minimum (and especially at the lower NMR frequency 12.3 MHz), the spin-diffusion process becomes a significant temperature-independent bottleneck. Hence, T_1 near the minimum is nearly independent of temperature and frequency. The proposed relaxation model is similar to one previously applied^{6,7} to the proton relaxation of solid HD in the presence of dilute *o*-H₂ impurities.

The temperature dependence of T_1 of dilute H₂ in the solid hosts Ne, Ar, Kr, and para-H₂ has been found⁸ to exhibit the power law temperature dependence characteristic of two-phonon Raman processes. Phonons induce transitions between the molecular rotation quantum sublevels at a rate τ^{-1} . The rotational transition rate τ^{-1} is expected on very general grounds to vary as T^2 for temperatures above or near the Debye temperature.⁹ The transition rate will be a stronger function of temperature for lower temperatures, eventually going as T^7 for low temperatures.⁹ The rate τ^{-1} is proportional to $E(T^*)T^{*2}$, where $E(T^*)$ is given as a tabulated function by Van Kranendonk.¹⁰ T^* is T/Θ_c , where Θ_c is a characteristic temperature for the local phonon dynamics as they enter the Raman process. The rotational transitions modulate the intramolecular dipolar interaction, leading to a relaxation rate¹¹:

$$\frac{1}{T_1(\text{H}_2)} = \frac{A\tau}{1 + \omega_0^2\tau^2} + \frac{B\tau}{1 + 4\omega_0^2\tau^2} \quad (1)$$

The spin-rotation coupling for H₂ does not appear above because crystalline fields are expected to quench the individual components of the rotational angular momentum.

The proposed relaxation model is a three-bath model (covalently bonded hydrogen, dilute molecular H₂, and phonons) with no direct link between the phonons and the covalent H. Hence, if one assumes spin diffusion to be rapid, the observed T_1^\dagger will be longer than $T_1(\text{H}_2)$ by a ratio of spin heat capacities:

$$T_1^\dagger = T_1(\text{H}_2) \frac{n(\text{H})I(I+1)}{\frac{3}{4}n(\text{H}_2)S(S+1)} \quad (2)$$

Here $n(\text{H})$ and $n(\text{H}_2)$ are the number densities of total hydrogen in the sample and of total H₂ molecules, respectively. The $\frac{3}{4}$ factor is the fraction of H₂ molecules which are *o*-H₂ and assumes that the

room-temperature ortho-para ratio is frozen in, as is common. The value of I above is the proton spin, $\frac{1}{2}$, and $S=1$ is the nuclear spin of *o*-H₂. The only temperature-dependent quantity in Eq. (2) is $T_1(\text{H}_2)$; so T_1^\dagger is expected to follow $T_1(\text{H}_2)$ as long as spin diffusion is sufficiently rapid.

The solid line in Fig. 1 is a fit of the proposed model to the 42.3-MHz T_1 data (solid dots). The fit has employed a spin-diffusion bottleneck T_1 (SD) of 0.16 sec, indicated by the horizontal line. The crystal fields for H₂ in *a*-Si:H are expected to be of low symmetry and the *o*-H₂ relaxation rate has been calculated using Fedders's no-symmetry, $r=0$ case¹¹ with the T_1 minimum taken to occur at 26 K and a characteristic temperature $\Theta_c=40$ K (the same value as was found appropriate⁸ for dilute *o*-H₂ in rare-gas solids). The calculated minimum $T_1(\text{H}_2)$ is 1.19 msec at 42.3 MHz. The upper dashed curve in Fig. 1 is $T_1^\dagger = 76T_1(\text{H}_2)$. The solid curve is the sum of the upper dashed curve and the spin-diffusion bottleneck: $76T_1(\text{H}_2) + 0.16$ sec. Thus the heat-capacity ratio in Eq. (2) is determined to be 76 and Eq. (2) yields $n(\text{H}_2) = 3.3 \times 10^{19}$ cm⁻³, assuming $n(\text{H}) = 5 \times 10^{21}$ cm⁻³ (for 10 at. %).

Using the $n(\text{H}_2)$ and $T_1(\text{SD})$ determined from the 42.3 data, the dot-dashed line shows T_1 predicted for 12.3 MHz. The lower dashed curve indicates $T_1^\dagger = 76T_1(\text{H}_2)$ calculated for 12.3 MHz. The calculated $T_1(\text{H}_2)$ minimum here occurs near 16.5 K and Θ_c again has been taken to be 40 K. The dot-dashed line represents $76T_1(\text{H}_2) + 0.16$ sec for 12.3 MHz. It is a reasonable fit to the limited data (triangles) and shows a substantial limitation by the spin-diffusion bottleneck.

An independent estimate of $n(\text{H}_2)$ can be determined from the spin-diffusion limited T_1 given by¹²

$$T_{1(\text{SD})}^{-1} = 4\pi D \frac{3}{4} n(\text{H}_2) b \quad (3)$$

The parameter b is the mean spacing between H₂ and its nearest H neighbors and is expected to be about 4 Å. The spin-diffusion coefficient D is approximately given by¹³

$$D = \frac{d^2}{30 \langle T_2 \rangle} \quad (4)$$

where d = the mean proton spacing (about 4 Å) and $\langle T_2 \rangle$ is an average T_2 of the multicomponent NMR line, taken as 50 μsec. Both the broad and narrow components of the NMR line show considerable clustering of H. The spin-diffusion coefficient we compute is an average over the various regions of H in the sample. It should be noted that samples with different total H content have similar NMR line shapes and hence similar spin-diffusion coefficients. With the above numbers Eq. (4) yields $D = 1 \times 10^{-12}$

cm²/sec in reasonable agreement with the value 2×10^{-12} cm²/sec previously estimated¹ for this *a*-Si:H sample. Equation (3) and $T_1(\text{SD}) = 0.16$ sec then yield $n(\text{H}_2) = 1.7 \times 10^{19}$ cm⁻³, a result within a factor of 2 of the $n(\text{H}_2) = 3.3 \times 10^{19}$ cm⁻³ obtained above from $T_1(\text{H}_2)$ and Eq. (2). Thus Eqs. (2) and (3) fit the T_1 and $T_1(\text{SD})$ data with about the same value of $n(\text{H}_2) \approx 2.5 \times 10^{19}$ cm⁻³ (an atomic fraction of 5×10^{-4}). The simultaneous fit to T_1^\dagger and $T_1(\text{SD})$ will not occur for proton-containing entities with a different proton-proton interaction than *o*-H₂. Because H₂ has such a small bond length, H₂ is unique in being able to fit the data. The NMR signal of the dilute H₂ should not be easily distinguishable, nor should any significant motional narrowing be associated with the T_1 minimum.

As indicated in Fig. 1 the temperature dependence of T_1 in *a*-Si:H is very similar to that of H₂ in Ne and Ar and the temperature of the minimum in *a*-Si:H (26 and 16.5 K at 42.3 and 12.3 MHz) is similar to those in Ar (13 K at 20 MHz and 17 K extrapolated to 42 MHz). It should be remarked that the power-law behavior of $T_1(T)$ is expected only for systems that can make transitions between states without surmounting an energy barrier. Such behavior is found for H₂ rotational levels (because of their quantum-mechanical nature arising from the small moment of inertia) and also for electron spins.

The observed T_1 is independent¹ of ω_0 on the hot side of the minimum, as expected from Eq. (1). This equation also predicts $T_1 \propto \omega_0^2$ on the cold side of the minimum, provided one is well away from the spin-diffusion bottleneck. The limited data in this range¹ agree with this prediction.

The proposed model also is consistent with reported proton T_1 results on samples diluted with deuterium. Reimer, Vaughan, and Knights² have studied the T_1 of two samples of *a*-Si:H,D, comparing them with otherwise similar (same total hydrogenic content and preparation conditions) samples of *a*-Si:H. Taking C to be the relative concentration of H in the partly deuterated sample compared with that in the undeuterated sample, these workers found that T_1^{-1} was proportional to C . That is, one deuterated sample which contained only 40% as many protons as the otherwise identical nondeuterated sample had a proton relaxation rate 40% of the nondeuterated sample's rate. These experiments were carried out near room temperature, and T_1 was found to have some temperature dependence, verifying that spin diffusion was not the rate-determining process in the data. The partly deuterated samples were prepared by glow discharge of SiH₄-D₂ mixtures. If one reasonably assumes that the dilute H₂, HD, and D₂ molecules in *a*-Si:H,D are formed from atomic H and D, then statistically $n(\text{H}_2)$ should vary as C^2 while $n(\text{H})$ varies as C . Hence Eq. (2) predicts that T_1 will vary as C^{-1} away from the spin-diffusion limit, in

agreement with the experimental observations.²

It should be remarked that any nonexponential relaxation (which often occurs in the presence of spin diffusion to relaxation centers) will involve an exceedingly small fraction of the H in the *a*-Si sample. This is true because the relaxation centers have such a small effective radius (~ 4 Å) compared with the mean spacing between them (~ 60 Å). Even in the diffusion-limited regime, the observed decays would appear exponential, unless exceptional signal-to-noise ratios were attained. Indeed, the reported recoveries are exponential; furthermore, spin diffusion causes the entire line to relax at the same rate, as observed.^{1,4}

Carlos and Taylor¹ have reported that similar T_1 minima are observed (with some variation in the magnitude of T_1) in a number of glow-discharge samples of good quality from various laboratories. One sample that was sputtered in a fairly high pressure (4 mTorr) of H₂ also showed¹⁴ such a minimum. However, two samples sputtered with less H₂ (0.4 mTorr) showed quite different results, longer T_1 and the absence of a deep T_1 minimum. This suggests that the latter samples may have contained much less H₂. Results reported¹⁵ from the Ames group on sputtered material also fail to show the T_1 minimum.

Presumably, *a*-Si:H is permeable to H₂. Hence we assume that the small amount of H₂ is present in atomic size "cages," regions where the H₂ is trapped by geometrical considerations. Reimer, Vaughan, and Knights have observed¹⁶ motional narrowing of a small fraction of the protons at room temperature in *a*-C:H, *a*-Si:C:H, and *a*-Si:N:H. The corresponding resonance lines are narrow, so that these protons are accomplishing a good average of dipolar interactions. Such a good average probably is more typical of small symmetric molecules (H₂ or CH₄) than local disorder modes. Hence we feel Reimer *et al.* may have observed trapped small molecules in these amorphous semiconductors.

One test of the proposed relaxation model is to measure the H₂ content of *a*-Si:H. Different samples should contain different amounts of H₂ and the minimum T_1 of these samples should differ. Measurements of T_1 at frequencies higher than 42.3 MHz should show a minimum increasing with ω_0 as spin-diffusion bottlenecking becomes less significant. Also, an *a*-Si:H sample which shows a proton T_1 minimum should have that minimum destroyed and T_1 increased upon proton decoupling by magic-angle spinning. The most conclusive test, however, involves ortho-para conversion of the H₂. The H₂ should convert at reasonable rates because of the presence of electron spins. A striking confirmation of the model would be provided if the minimum T_1 were greatly increased by a prolonged waiting time at or below 20 K (where almost all of the H₂ will convert to para-H₂). Subsequent reheating to room tem-

perature for an appropriate interval should then restore the 3:1 ortho-to-para ratio and the T_1 minimum upon recoiling to ~ 20 K.

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