Nuclear-magnetic-resonance study of $Li⁺$ motion in lithium aluminates and LiOH

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We report T_1 and T_2 measurements of ⁷Li resonances, which demonstrate thermally activated motion of Li⁺ in the ionic conductors α - and β -Li₅AlO₄. At high temperatures our data give $T_1/T_2 \approx (I + 1/2)^2 = 4$, which demonstrates that dipole interactions between magnetic impurities and ⁷Li produce the relaxation, even for our nominally pure samples, which are shown by ESR measurements to contain \sim 15-ppm magnetic impurities. A liquidlike spatial diffusion model is used to analyze the relaxation times. At high temperatures the relaxation is determined by the Li⁺ jump time for which our analysis gives an activation energy $E_a = 0.83(2)$ eV for a, and $E_a = 0.56(4)$ eV for β , with an attempt frequency $v_0 = 2 \times 10^{12}$ sec⁻¹ for both phases. Near room temperature, T_1 is apparently determined by the impurity relaxation time, for which we phases. Ivear from temperature, T_1 is apparently determined by the impurity relaxation time, for which we estimate $T_{1e} \sim 10^{-5}$ sec. T_2 measurements in LiOH show motional narrowing of the ⁷Li resonance above 320°C but not the ¹H resonance, and thus identify Li^+ as the conducting ion in LiOH. Limited T_1 and T_2 measurements are also reported for ⁷Li in γ -LiAlO₂.

I. INTRODUCTION

Solid-state ionic conductors are of technological interest for a variety of possible applications such as high-energy density storage batteries. Li' conductors are receiving special attention because of the low atomic weight of Li and its large reduction potential. Among the most promising candidates for a Li⁺ conductor at \sim 450 °C is Li₅AlO₄, for which Raistrick *et al.*¹ obtained conductivities as high as $0.3 \ (\Omega \text{ cm})^{-1}$. To learn more about its conduction mechanism, we began work about its conduction inechanism, we began work
on Li_5AlO_4 . Johnson *et al.*² have since shown that the reported high conductivities can be obtained only by exposing pure samples to water vapor, or by initially incorporating excess LiOH in the samples. This paper reports our work on pure α and β -Li₅A1O₄ and a limited investigation of LiOH. Further work on composite samples is underway at our laboratory. A discussion of the crystal structure and thermal properties of $Li₅AIO₄$ is given in Sec. II.

We have used pulsed NMR to probe the diffusion of Li' in our samples. Nuclear relaxation occurs by fluctuations in the local environment of the nucleus, and hence can be sensitive to diffusion processes. We have measured the transverse relaxation rate $(T₂⁻¹)$ and the longitudinal or spinlattice relaxation rate (T_1^{-1}) of the ⁷Li nuclei in our samples. In Sec. III we define these relaxation rates more precisely and discuss their experimental determination. Because the 'Li nucleus has spin $I(^{7}Li) = \frac{3}{2}$, it possesses an electric quadrupole moment and interacts with electric field gradients in its vicinity. The consequences of this interaction on T_1^{-1} and T_2^{-1} are also given in Sec. III.

Our initial attempts to prepare $Li₅AIO₄$ resulted

in the β phase being quenched into a metastable state. Since the data from this phase demonstrate diffusion-governed relaxation rates, we present the T_1^{-1} and T_2^{-1} results in Sec. IVA. The ratio T_1/T_2 at high temperatures is found to agree with that calculated for dipolar interactions with magnetic impurities, while disagreeing with the calculated ratio for electric quadrupole interactions. We present a liquid like diffusion model which accounts for our β -Li₅AlO₄ data quite well, and extract the activation energy and attempt frequency of the Li' jump responsible for 'Li relaxation. We also identify a temperature regime where fluctuations of the impurity moment determine ⁷Li relaxation, instead of Li' jumps.

Our initial β -Li₅AlO₄ samples were found to contain γ -LiAlO₂. To separate the signals of the two components in relaxation curves, 'Li data were obtained on pure γ -LiAlO₂. The limited results are presented in the Appendix since they also demonstrate some of the diffusive features found in $Li₅AlO₄$.

 α -phase data similar to β are presented in Sec. V. Although T_1^{-1} does not exhibit the dependence on the nuclear-resonance frequency expected from the diffusion model, we use it to extract the jump attempt frequency and activation energy.

To investigate the conduction mechanism of LiOH, whose presence so radically alters the conductivity of $Li₅AIO₄$ samples, we have measured its 'Li and 'H relaxation rates. Unfortunate- μ , T_1^{-1} is much less experimentally accessible than in the above compounds. Thus we report only T_2^{-1} data in Sec. VI. These data clearly identify Li' as the conducting ion.

In Sec. VII, we discuss possible explanations for deviations of the data from the theoretical model given in Sec. IVA. None are found to be

18 5928 0 1978 The American Physical Society

5929

entirely satisfactory. We examine the jump parameters obtained for α - and β -Li₅A1O₄. Finally, we discuss the distinguishing features of relaxation by magnetic impurities and examine nuclear relaxation in other ionic conductors where such relaxation may be dominant.

II. PROPERTIES OF Li_s AlO₄

A. Structural description

Li_sAlO₄ exists in either a low-temperature (α) or high-temperature (β) structure with the α to β transition occurring at \sim 780 °C. Detailed structural determinations have been carried out on both α - and β -Li₅AlO₄.³⁻⁵ The structure of isotopic α -Li₅GaO₄ has also been reported.⁶ These structures are ordered derivatives of the antifluorite (Li,0) structure with vacancies occupying distinct lattice positions and can be formulated as $Li_sV₂AIO₄$ where V represents a vacancy.

 α -Li_sAlO₄ crystallizes in the orthorhombic space group *Pbca* with $a = 9.087$, $b = 8.947$, and $c = 9.120$ \AA and with eight formula units in the unit cell.⁶ The Li and Al atoms and the vacancies occupy the tetrahedral sites in the nearly fcc array of 0 atoms. Their ordered arrangement and their slight displacements from the ideal antifluorite positions give rise to the orthorhombic rather than cubic structure. These atoms and the vacancies occupy the general $(8c)$ positions. There are five crystallographically distinct Li atom positions giving rise to a total of 40 Li atoms in the unit cell. Although each Li atom has identical nearest neighbors, each being tetrahedrally coordinated by four oxygen atoms, they have different nextnearest and further-distant neighbors. These differences can in principle cause the five distinct types of Li atoms to have different activation energies or potential barriers for motion. Also, there are three distinct types of empty octahedral positions in the fcc oxygen atom array. If the lithium ion motion were to take place through the octahedral positions, there could again be different activation energies for motion.

 β -Li₅AlO₄ crystallizes in the orthorhombic space
"oup *Pmmn* with $a = 6.420$, $b = 6.302$, and c
4.620 Å with two formula units in the unit cell.^{4,5} group Pmmn with $a=6.420$, $b=6.302$, and c $=4.620$ Å with two formula units in the unit cell.^{4,5} Again the Li and Al atoms and the vacancies occupy tetrahedral sites and their ordered arrangement causes the orthorhombic distortion from cubic symmetry. There are two distinct Li atom positions, one in the general $(8g)$ position and one in the special (2b) position. Also, there are two slightly different octahedral positions. Either of these differences could cause more than one activation energy for motion of the Li atoms.

B. Materials preparation

Reagent grade $Li₂O$ and $Al₂O₃$ were used to prepare both α - and β -Li_sAlO₄. The supplier's specifications give \sim 10-ppm Fe content, a significant point since we shall show that the 'Li relaxation is due to dilute magnetic impurities. To prepare α -Li₅AlO₄ from these materials, they were heated at 650 \degree C in alumina crucibles in air or O_2 for 2-6 days. Samples of β -Li₅AlO₄ were prepared by melting the α -Li₅AlO₄ powder in an alumina crucible at \sim 1100 °C and then annealing at 900 °C for \sim 24 h. After the annealing period at 900 °C the high-temperature structure, β -Li₅AlO₄, was retained by quenching the sample. Both α and β -Li_sAlO₄ samples were stored in an argon filled glove box to prevent hydration. The NMR measurements were carried out on samples sealed in quartz tubes under a partial atmosphere of argon or vacuum to avoid contamination by water vapor.

III. NUCLEAR RELAXATION FOR $I=\frac{3}{2}$

To more clearly identify T^{-1}_{1} and T^{-1}_{2} , we need to discuss the properties of the 'Li nuclear resonance and the manner in which the relaxation rates are measured. An important feature of the ⁷Li resonance is that the nucleus has spin I (⁷Li) $=\frac{3}{2}$ and has an electric quadrupole moment. The four evenly spaced energy levels of an isolated nucleus in a magnetic field are perturbed by the electric field gradient of the surrounding ions in 'electric field gradient of the surrounding ions
the solid. The $\pm \frac{5}{2} \rightarrow \pm \frac{1}{2}$ nuclear transitions are shifted in frequency from the unshifted (in a first shitted in frequency from the unshifted (in a first order perturbation treatment) $+\frac{1}{2} \leftrightarrow -\frac{1}{2}$ transition which alone is observed in our experiments. This has important consequences for T_1^{-1} and T_2^{-1} measurements.

 T_1^{-1} is measured as follows. A "comb" of 8-10 saturating pulses is applied at the resonance frequency to destroy the net nuclear magnetization of the resonance. After a delay time t_d from the last comb pulse, the nuclear magnetization is measured by the spin-echo amplitude. The magnetization $M(t_d)$ for simple $I=\frac{1}{2}$ systems recovers as

$$
M(t_d) = M(\infty) (1 - e^{-t_d/T_1}). \tag{1}
$$

Thus, one expects a linear plot for the quantity $\ln[1 - M(t_d)/M(\infty)]$ vs t_d , with slope T_1^{-1} .

However because of the quadrupole splitting of the 7 Li nuclear transitions, the magnetization recovery can relax as a sum of several exponential terms. This can occur even when the quantum mechanical process causing the nuclear relaxation terms. This can occur even when the quantum
mechanical process causing the nuclear relaxat
is a $\Delta m = \pm 1$ process,^{7,8} as is the case for magnetic impurities. For $I(^{7}Li) = \frac{3}{2}$, we would then

expect the plotted quantity to then decay as

$$
ae^{-t_d/T_1} + be^{-6t_d/T_1}, \t\t(2)
$$

where $a + b = 1$. In such systems, one finds that the b term can be minimized by optimizing the comb pulse width, adding more pulses, and spacing the comb pulses T_2 apart. Experimentally we find that the slow component can be made to exceed $a > 0.8$ and often 0.9, and take its slope to be T^{-1} .

By T_2^{-1} we mean the phase-memory decay rate. In most cases, we observe the resonance by using two equal width rf pulses separated by time t to generate a spin echo at time $2t$ after the first pulse. T_2^{-1} is then obtained by measuring the decay of the echo amplitude with pulse spacing, which
goes as e^{-2t/T_2} . T_2^{-1} thus measured correspond to the homogeneous broadening of the nuclear resonance linewidth and not the full width with inhomogeneous broadening.

mogeneous broadening.
Effects due to observing only the $\frac{1}{2} \leftrightarrow -\frac{1}{2}$ transi tion can also be observed in the phase-memory decay rate when nonsecular (or lifetime, T_1^{-1}) contributions to T_2^{-1} become significant. These contributions can be. enhanced above the value of $\frac{1}{2}T_1^{-1}$, which applies for $I=\frac{1}{2}$ and for cases where all nuclear transitions occur at the same frequency. However, the enhancement for $+\frac{1}{2} \rightarrow -\frac{1}{2}$ depends upon the detailed form of the interaction producing the relaxation. In isotropic limits (where T_{1}/T_{2} =1 for I = $\frac{1}{2}$ for a dipolar interaction with a time-dependent local field

$$
H_{\text{int}} = -\gamma_n \vec{I} \cdot \vec{H}_{\text{loc}}(t) , \qquad (3)
$$

the ratio becomes'

$$
T_1/T_2 = (I + \frac{1}{2})^2,
$$
\n(4)

which equals 4 for $I=\frac{3}{2}$. However for relaxation by the quadrupole interaction with a time-dependent electric field gradient, the same ratio has
been calculated to be 1 for $I = \frac{3}{2}$.¹⁰ Thus the two been calculated to be 1 for $I = \frac{3}{2}$.¹⁰ Thus the two interactions can be distinguished in the isotropic limit.

IV. METASTABLE β -Li₅AlO₄ DATA AND ANALYSIS

 T^{-1}_1 and T^{-1}_2 data are shown for β -Li₅AlO₄ in Fig. 1 as a function of T^{-1} (K⁻¹) for temperature from room temperature $(\sim 24 \degree C)$ to 500 $\degree C$ for frequencies of 7.2, 12.7, and 21.5 MHz. We attempted to take data above 500 \degree C, but samples cycled to higher temperatures would not reproduce T , values initially obtained below 500 °C, and also exhibited deterioration of their signal amplitudes. X-ray analysis of such samples showed the presence of the α phase. The β phase can be "quenched in" from samples. prepared at high temperatures

FIG. 1. Experimental values of T_1^{-1} (0: 21.5 MHz, \blacksquare : 12.7 MHz, and A: 7.2 MHz) and T_2^{-1} (\bullet : 21.5 MHz) are given for the ⁷Li resonance in metastable β -Li₅AlO₄. Representative error limits are shown for selected points. The solid curves show experimental trends and are not theoretical fits to the data.

(>800 °C) where β is the equilibrium phase, but on heating the sample to the range $500-800$ °C, it evidently changes structure to the equilibrium phase of that temperature range α . Thus data were not accepted for temperatures greater than 500 'C, nor from samples which had been cycled above this temperature. Since the data shown in Fig. 1 were reproducible after thermal cycling below 500 \degree C, and since they show a continuous temperature dependence, we accept them as valid data on a unique structure, metastable β -Li₅AlO₄.

Our initial β samples showed two signal components in both transverse and longitudinal relaxation curves. We were able to identify the long T_1 component with the short T_2 component, and vice versa. X-ray examination of these samples detected the presence of γ -LiAlO₂. This led to our brief investigation of γ -LiAlO₂, which is presented in the Appendix. The γ -LiAlO₂ data are consistent with the long- T_1 -short- T_2 values observed in the two-phase samples. Subsequently,

we were able to prepare samples consisting entirely of β -Li₅AlO₄, with T_1 and T_2 values consistent with the short- T_1 -long- T_2 data mentioned above. Most of the data in Fig. 1 were obtained with these single-phase samples.

A. Diffusive relaxation

 T_1^{-1} data above 146 °C and T_2^{-1} above 236 °C show diffusive behavior. On the semilog plot in Fig. 1, diffusive behavior. On the semilog plot in Fig. T_1^{-1} goes through a maximum at ~450 °C, with strong frequency dependence $(T_1^{-1} \propto \omega^{-1.8})$ on the low-temperature side of the maximum. T_2^{-1} shows motional narrowing, with approximately the same activation energy $[0.63(5)$ eV as found in the T_1^{-1} data $[0.56(4) \text{ eV}]$. These features show that a Bloembergen-Purcell-Pound-type model¹¹ for relaxation by thermally activated diffusion (which gives $T_1^{-1} \propto \omega^{-2}$ at low temperatures) is relevant.

To identify the interaction responsible for the relaxation, we examine the magnitude of T_1^{-1} at the maxima. Consider first the nuclear dipoledipole interaction between neighboring 'Li nuclei. The strength of this interaction is measured by the The strength of this interaction is measured by the Van Vleck second moment $(\Delta \omega_d^2)^{12}$ Summing over the four nearest ⁷Li nuclei, we obtain $(\Delta \omega_d^2)^{1/2}$ $\approx 2.5 \times 10^4$ sec⁻¹. Estimating the maximum relaxation at frequency ω from this interaction by $T_1^{-1}|_{max} \sim \Delta \omega_d^2/\omega$, we obtain ~5 sec⁻¹ at 21.5 MHz. This is too weak to explain the experimental value \sim 5 \times 10² sec⁻¹, which requires a stronger interaction. A time-varying component of the quadaction. A time-varying component of the quad-
rupole interaction of magnitude $\Delta \omega_q \sim 2 \times 10^5 \text{ sec}^{-1}$ would be sufficiently strong, but requires a hightemperature ratio $T_1/T_2 = 1$. The high-temperature data in Fig. 1 are in good agreement with a ratio of 4 and in clear disagreement with 1. Thus they support the dipolar interaction of Eq. (3). A nonzero time-averaged component of the quadrupole interaction is also required by the experiment interaction is also required by the experiment
ratio so that only the $\frac{1}{2} \leftarrow -\frac{1}{2}$ transition be observed, but evidently the quadrupole fluctuations $\Delta\omega$, are less than the above estimate. The source of the local field is evidently the dipole field of impurities with a magnetic moment, provided they are abundant enough to produce the observed magnitude of nuclear relaxation.

To check this proposed explanation, ESR signals have been searched for and found in our $Li₅AlO₄$ have been searched for and found in our Li_5AlO_4
samples.¹³ Resonances were observed at $g=2.0$, 3.3, and 4.9. The $g = 2.0$ resonance is strongest, and is likely to be that of $\mathbf{F}e^{3+}$ ions in the Al^{3+} positions. When the signal intensities were compared with that from a known amount of $CuCl₂ \cdot 2NC₅H₅$, the magnetic impurity concentration is estimated to be $~150$ ppm per formula unit and hence per Al^{3+} . This corresponds to $15-20$

ppm per cation site.

1. Magnetic-impurity contributions to T_I^{-1}

Abragam¹² has discussed nuclear relaxation by magnetic impurities, and has shown that the dipolar interaction between the resonant nucleus and the impurity moment is more effective in producing relaxation than other interactions. The relaxation is determined by the time dependence of the impurity dipolar fields as seen by the nuclei, called the "local field." This time dependence is described by a correlation function which relates the local field at a given time with the local field at a time t later. This is related to the component of the impurity moment in the direction of the magnetic field S_z and is assumed to decay exponentially:

$$
\langle \langle H_{\text{loc}}(0) | H_{\text{loc}}(t) \rangle \rangle \propto \langle \langle S_z(0) | S_z(t) \rangle \rangle \tag{5}
$$

$\propto S(S+1)e^{-t/t_c}$

For static lattices, the decay is due to longitudinal fluctuations of the impurity moment, and the correlation time t_c is equal to the impurity T_1 : t_c $=T_{1e}$. However if there is rapid motion of the Li⁺ ions the decay will be due to Li' jumps, which occur within a time τ , and $t_c = \tau \ll T_{1e}$. Implicit in our treatment which describes all the nuclei by a single correlation function (and neglects spin diffusion) is that the $Li⁺$ diffusion is rapid enough that all the nuclei sample the same time-averaged environment within the experimental times T_1 and $T, \{ \langle \langle \rangle \rangle \}$ in Eq. (5) denotes a spatial and temporal average over all 'Li nuclei].

For this liquidlike model, Abragam's expression¹² [Eq. (40), p. 380], is slightly modified. For a powder, one obtains 14

$$
T_1^{-1} = \frac{2}{5} c \hbar^2 \gamma_e^2 \gamma_N^2 S(S+1) \frac{\tau}{1 + \omega^2 \tau^2} \sum_n \frac{1}{r_n^6},
$$
 (6)

where c is the impurity concentration, γ_e and γ_n are the impurity and nuclear gyromagnetic ratios, S the impurity spin, and the sum is over neighboring ⁷Li nuclei at distances r_n from the impurity. Assuming that the impurity is Fe^{3+} (S = $\frac{5}{2}$, g = 2.0) and that it substitutes in an Al^{3+} site, we calculate the sum (out to fourth neighbors) to be 3×10^{46} cm⁻⁶ for β -Li₅AlO₄. With 15-ppm Fe³⁺, we calculate the maximum relaxation $[\omega \tau = 1$ in Eq. (6)] to be $T_1^{-1}|_{\text{max}} \approx 6 \times 10^2 \text{ sec}^{-1}$ at 7.2 MHz. The nearly precise agreement with the experimental value is fortuitous, but demonstrates that the concentration of magnetic impurities observed with ESR is sufficient to explain. our observed nuclear relaxation rates.

For thermally activated diffusion we expect

$$
\tau = \tau_0 e^{B_a / kT} \,, \tag{7}
$$

where E_a is the thermal activation energy for a jump to occur and τ_{0}^{-1} \equiv ν_{0} is usually identified as the jump attempt frequency. With this temperature dependence imposed on τ in Eq. (6), we expect three regimes of T_1^{-1} . At low temperatures $\omega^2 \tau^2$ 1, T_1^{-1} increases with increasing temperature as T_1^{-1} $\propto \omega^2 \tau^{-1}$. At $\omega \tau$ = 1, T_1^{-1} goes through a maximum. Finally at high temperatures $\omega^2 \tau^2 \ll 1$, $T_{\:\:1}^{-1}$ decreases with increasing temperature as $T^{-1}_{1} \propto \tau.$

The T^{-1} data in Fig. 1 above 136 °C show this behavior but with two minor differences. First, the frequency dependence below the maxima is slightly weaker than ω^{-2} , being better described as $\sim \omega^{-1.8}$. Whether this is due to a minor frequency-independent contribution which we are unable to clearly identify, or whether the frequency dependence of the motion governed T_1^{-1} is actually weaker than ω^{-2} , we cannot say. The second deviation requires examination of the curves for each frequency at the maxima. The 7.2-MHz curve is more rounded than the 21.5- MHz curve, whereas Eq. (6) gives identically shaped curves displaced from each other. The rounding also results in a weaker than ω^{-1} frequency dependence of T_1^{-1} at the maxima of the curves as predicted by Eq. (6).

Equation (7) can be used with the observed temperature of the T_1^{-1} maxima to extract the attempt frequency. We take the observed activation energy to be the barrier height E_a in (7). Using $E_a = 0.56$ eV, the 21.5-MHz data can be fit quite well with $v_0 = 2 \times 10^{12}$ sec⁻¹. The other frequencies and the uncertainty in E_a give a factor of 2 uncertainty in this value.

2. Magnetic-impurity contributions to T_2 ⁻¹

The motional narrowing of T_2^{-1} between 236 and 330 \degree C can be accounted for by the same model. In this temperature range, T_2^{-1} is determined by low-frequency longitudinal fluctuations of the local field. We then expect Eq. (6) to also describe T₂¹ if we let ω + 0 and divide by two (there is only one longitudinal local-field direction, whereas there are two independent transverse directions):

$$
T_2^{-1}\big|_{\parallel} = \frac{1}{5}c\hbar^2\gamma_e^2\gamma_N^2S(S+1)\tau\sum_n\frac{1}{\gamma_n^6}.
$$
 (8)

If the T_2^{-1} data in the above temperature range are extrapolated into the high-temperature region where $T_1^{-1} \propto \tau$, we find $T_1^{-1} \approx 1.7T_2^{-1}|_{\text{extrap}}$, in fair agreement with the expected factor of 2; Above 330 'C, transverse fluctuations contribute significantly to T_2^{-1} , enhanced by a factor η (= 7 for miticantly to T_2^* , enhanced by a flactor η (= 7 for $I = \frac{3}{2}$) because we are only observing the $\frac{1}{2} \rightarrow -\frac{1}{2}$ nuclear transition

$$
T_{2}^{-1} = T_{2}^{-1} \big|_{\parallel} + \frac{1}{2} \eta T_{1}^{-1} \,. \tag{9}
$$

For isotropic relaxation in the high-temperature regime, i.e., for $T_2^{-1}|_{\parallel} = \frac{1}{2}T_1^{-1}$, the ratio T_1/T_1 $=(I+\frac{1}{2})^2$ is obtained as discussed in Sec. III.

Equation (8) and the above T_2^{-1} data correspond to the high-temperature regime of the magnetitive the magnetic theorem is the magnetic theorem in the set of H and α do do do do do do do the t all. "tagging" experiments of Hogg et $al.,$ ¹⁵ who doped the F^- conductor PbF_2 with Mn^{2+} impurities. Their data show a peak in T_2^{-1} , with the low-temperature regime being described by

$$
T_2^{-1} \approx z c / \tau \,, \tag{10}
$$

where z is the number of neighboring 7 Li sites which couple to the magnetic impurity. Equation (10) can be physically interpreted as being the rate at which nuclei encounter a strong local field which does not permit them to refocus and contribute to the spin echo. This mechanism should be appropriate for the magnetic impurity contribution to T_2^{-1} in some region below 236 °C. However, we are not able to separate its contribution to the experimental value from that of the nuclear dipole-dipole interaction. If we assume that the correlation time for the nuclear-dipole interaction is the same as that obtained from T_1^{-1} , we can is the same as that obtained from T_1^* , we can
estimate $T_2^{-1}|_{d} \approx \Delta \omega_d^2 \tau_c \approx 10^3$ sec⁻¹ at 100 °C. Thus while $T_2^{-1}|_d$ might obscure the magnetic contribution at lower temperatures, it is very unlikely to be influencing the experimental values above the 236 °C point, where τ_c has further decrease by two orders of magnitude. Since $T_2^{-1}|_d$ decreases with increasing temperature and Eq. (10) increases, adding the two may account for the nearly temperature-independent experimental values between 20 and 236 \degree C.

To find the approximate conditions of the maximum in the magnetic contribution to T_2^{-1} , we can equate (8) and (10) to obtain (approximated to first neighbors)

$$
\frac{1}{5}\tau\hslash\gamma_e\gamma_N[S(S+1)/\gamma_n^3] \sim 1,
$$
 (11)

and

$$
T_{2}^{-1}\Big|_{\max} \sim \frac{1}{5} z c \hbar \gamma_e \gamma_N S(S+1) \,. \tag{12}
$$

These expressions can be checked for consistency with the jump times obtained with the T_1^{-1} analysis. At the onset of the exponential temperature dependence at 236 °C, we obtain a value of \sim 1.9 for the left-hand side of (11) and calculate $T_2^{-1}|_{max}$ $\sim 4 \times 10^3$ sec⁻¹, in fair agreement with the slightly peaked experimental value of $~10^4$ sec⁻¹. A complete theoretical treatment of T_2^{-1} over the entire
temperature range has recently been obtained.¹⁴ temperature range has recently been obtained.¹⁴

B. Relaxation by magnetic fluctuations

At 86 °C and below, T_1^{-1} deviates dramatically from its extrapolated temperature dependence in the range $150-250$ °C. The frequency dependence is still observed, but the correlation time t_c appears to be shorter than expected by assuming that (7) describes t_c over the entire experimental temperature range. The temperature dependence of t_c is also very weak.

A ready explanation of these features is provided with the magnetic impurity model by assuming that at ~100 °C, τ $\sim T_{1e}$ so that at lower temperatures the correlation function decays by the more rapid fluctuations of the magnetic moment rather than Li^+ jumps. The T_1^{-1} analyses and data below 100 °C thus allow us to estimate $T_{1e} \sim 1$ $\times 10^{-5}$ sec⁻¹ at room temperature. The roomtemperature linewidth of the $g=2$ ESR resonance is ~8 Oe, which places an upper limit $T_{\infty}^{-1} \lesssim 7$ $\times 10^7$ sec⁻¹. Since $T_{1e} \ge T_{2e}$, we obtain $T_{1e} \ge 1.4$ \times 10 Sec, which our value from the NMR analysis \times 10⁻⁸ sec, which our value from the NMR analysis satisfies.

V. α -Li₅AlO₄ DATA AND ANALYSIS

'Li nuclear relaxation rates were measured in α -Li₅AlO₄ from 24 to 750 °C. Since differential thermal analysis show the onset of the $\alpha \rightarrow \beta$ transition at 780 °C, we limited ourselves to the above range of temperatures in order not to be troubled by the irreproducibility of results such as observed when metastable β -Li₅AlO₄ was heated too high. We checked and found that samples cycled to 750 °C and back to lower temperatures gave the same T_1 and T_2 values as previously obtained. T_1^{-1} values obtained at 21.5, 12.7, and 7.2 MHz and T_2^{-1} values at 21.5 MHz are shown in Fig. 2. Only a single component was detected in the nuclear relaxation curves, in agreement with x-ray analysis which showed only α -Li₅AlO₄ to be present. Throughout the entire study, samples sealed as above showed no evidence of deterioration (such as discoloration or chemical attack of the quarts tube). With the known effects of water vapor on the conductivity in mind, we stress that our results are for pure α -Li₅AlO₄ without water or LiOH contamination.

A. Diffusive relaxation

In many respects the data in Fig. 2 resemble those of the metastable β phase in Fig. 1. A diffusive regime is observed between 300-750 C. The high-temperature ratio T_1/T_2 [~] 4 again requires relaxation by magnetic impurities. In this case, 'however, $T_1^{-1}\Big|_{\text{max}} \propto \omega^{-1}$ as expected from (6), though the temperature position of the maximum changes

FIG. 2. Experimental values of T_1^{-1} (\bullet : 21.5 MHz, ■: 12.7 MHz, and ▲: 7.2 MHz) and T_2^{-1} (●: 21.5 MHz) are given for the ⁷Li resonance in α -Li_sAlO₄ for temperatures between 250-750 °C in the large graph, with 20-250 C in the inset. Representative error limits are shown for selected points. The solid curves show experimental trends and are not theoretical fits to the data.

little with frequency, and the 750 \degree C values are not frequency independent. Qualitatively, in the α phase we apparently must go higher in temperature (650 \degree C) to obtain the same jump times at the maxima as with the metastable β phase (400 °C). In the α phase, however, the frequency dependence below the maximum is approximately $\omega^{-1.0}$ vs $\omega^{-1.8}$ for the β phase and $\omega^{-2.0}$ as expected from Eq. (6) .

1. Magnetic-impurity contributions to T_I^{-1}

To check the effectiveness of relaxation by the magnetic impurities, we apply Eq. (6) to β -Li₅AlO₄ at 7.2 MHz. The sum over neighboring sites is

 4×10^{46} cm⁻⁶, giving $T_1^{-1}|_{max} \approx 9 \times 10^2$ sec⁻¹, which is comparable to the experimental value 1.2 $\times 10^3$ sec⁻¹. Thus magnetic-impurity relaxation is sufficiently rapid to explain our data, and the following analyses closely parallel those of β -Li₅AlO₄.

Turning to the data between 300 and 750 'C, we find that T_1^{-1} data for each frequency shown and activation energy $E_a=0.83(2)$ eV. The temperature of each maximum can then be fit with the $\omega \tau \approx 1$ condition to obtain an attempt frequency. When the results from each frequency along with the uncertainty in E_a are considered, we conclude ν_0 $=2\times10^{12}$ sec⁻¹ to within a factor of 2. We have ignored the weaker frequency dependence of the data, but we still expect the condition $\omega \tau \approx 1$ to hold in the vicinity of the maxima.

2. Magnetic-impurity contributions to T_2 ⁻¹

The activation energy of the T_2^{-1} data in the motional narrowing regime 400-550 'C is approximately the same [0.78(8) eV] as that of the T_1^{-1} data. Extrapolating these data into the high-temperature T_1^{-1} regime gives $T_1^{-1} \approx 1.4T_2^{-1}|_{\text{extrap}}$, in fair agreement with Eqs. (6) and (8). The data show a slight peak at 300-350 'C whose magnitude is within a factor of 3 of that calculated above from Eq. (12). However, we calculate a value of \sim 40 for the left-hand side of Eq. (11). Thus the α -phase data are not as consistent with the model for magnetic-impurity relaxation given in Sec. IVA as are the β data. The poor agreement with Eq. (11) suggests that our values of τ (and hence v_0 and E_a) extracted from the T_1^{-1} data are less reliable.

B. Relaxation by magnetic fluctuations

A break from the diffusive regime to a much more slowly temperature-dependent T_1^{-1} regime is seen again in Fig. ² below 300'C. As before, if we assume that at 300 °C $\tau \approx T_{1e}$, we estimate $T_{\mu} \sim 2 \times 10^{-5}$ sec at room temperature. This again satisfies the ESR upper limit discussed in Sec. IVB, but is subject to the reliability of the τ value.

VI. LiOH DATA AND ANALYSIS

Powdered samples of reagent grade LiOH were sealed in quartz tubes under a partial argon atmosphere. We attempted to work with both the 'Li and 'H resonances but were hampered by extremely long T , values, which limited how fast we could repeat the pulse sequence and accumulate data. For the 'Li resonance, the optimum repetition period at room temperature was ~200 sec, de-

FIG. 3. T_2^{-1} data for the ⁷Li resonance (\bullet) and the ¹H resonance (A) in LiOH. Note that the horizontal (temperature) scale is broken at the far right to include roomtemperature data.

creasing to \sim 1 sec at 430 °C. The corresponding values for the 'H resonance were 60 and ² sec. With these unfavorable T_1 values, we chose to measure only T_2^{-1} at 21.5 MHz. Shown in Fig. 3 are T_2^{-1} values for both the ¹H and ⁷Li resonances at room temperature and in the range 300-435 C.

The proton value evidently remains constant from 24 to 300 \degree C. It decreases only slightly at 430 'C, which is just below the melting temperature of 470 'C. The absence of motional narrowing shows that the protons are evidently static until just prior to melting, and are not responsible for the conductivity of LiOH. Proton relaxation is due to the nuclear dipole interaction, for which the onset of motional narrowing is described by 12

$$
\omega_d \tau_H \approx 1 \,, \tag{13}
$$

where ω_d is the static lattice dipole interaction and τ_H the proton jump time. Using the static lattice T_2^{-1} value for ω_d , the absence of motional marrowing then requires $\tau_H \ge 3 \times 10^{-5}$ sec at 430 °C.

The 'Li data show clear evidence of Li' motion by, the decrease in T_2^{-1} above 320 °C. These data show more scatter than those of the previously discussed compounds but are in fair agreement

with a thermally activated mechanism with E_a = 1.3(2) eV. T_2^{-1} evidently exhibits a peak at 320 \degree C with the start of motional narrowing. This suggests that the interaction responsible for the diffusive relaxation is again magnetic impurities. While further data, especially T_1^{-1} , are needed for a more complete understanding of the nuclear relaxation processes in LiOH, our results definitely identify Li' as the diffusing ion.

VII. DISCUSSIONS AND CONCLUSIONS

We turn now to exploring explanations for the deviations from the relaxation by magnetic impurities as discussed in Sec. IVA. The strongest deviations occur with the α -Li_sAlO₄ data, especially the weak frequency dependence at tem-'pecially the weak frequency dependence at tem-
peratures below the T_1^{-1} maxima $(T_1^{-1} \propto \omega^{-1}$ instea of ω^{-2}). As a first explanation, we might try to separate the experimental rate into two contributions:

$$
T_1^{-1} = A + B \omega^{-2}, \qquad (14)
$$

where $B\,\omega^{-2}$ is the magnetic impurity contribution to T_1^{-1} and A is a second frequency-independ contribution. The data between 300 and 600 'C can be fitted with such an expression. However, the resulting A terms are strongly temperature dependent with an apparent activation energy which is about the same as that of the $B\omega^{-2}$ terms. Forcing the data to fit Eq. (14) therefore seems inappropriate since the frequency-independent contribution would also be apparently determined by the same Li' motion. The net effect of such a fit on the 7.2-MHz data is to increase E_a slightly and double ν_{0} .

As a check on our use of the liquidlike Li' diffusion model, we can require the ions to have adequate diffusion to encounter at least one impurity during T_1 . This can be expressed mathematically $as¹⁴$

$$
(6DT_1/a^2)c > 1 , \t\t(15)
$$

where a is the Li-Li jump distance, c the impurity concentration, and D the diffusion constant. We can estimate $D \approx \nu a^2$, where ν is the Li⁺ jump rate, and then obtain

$$
6\nu T_1 c > 1. \tag{16}
$$

At the minimum $T_1 \approx 10^{-3}$ sec, $\nu \approx \omega = 4.5 \times 10^7$ sec⁻¹ (7.2 MHz), and for $c = 15 \times 10^{-6}$ we obtain $6\nu T_1c = 4.0 > 1$. Approximately this value is appropriate over the entire diffusion regime since on the low-temperature side $T_1 \propto \nu^{-1}$. However in the magnetic fluctuation regime $T₁$ is shorter than predicted by ν^{-1} . At room temperature for β -Li₅A1O₄ we obtain $6\nu T_1c \sim 0.1$, while an even less staisfactory value is obtained for α : $6\nu T_1 c \approx 10^{-4}$, thus questioning the Li' diffusion model in the magnetic fluctuation regime, especially for α .

If Li' diffusion is too slow, we might expect that a spin-diffusion model is appropriate. In this case, the diffusion rate is given by $6D/a^2 \approx T_2^{-1}$ $\approx 10^4$ sec⁻¹, where the rigid lattice T_2 is used to estimate the time for mutual spin flips. If this value satisfies Eq. (15), then Eq. (6) still remains valid. However at room temperature T_1 [~] 1 sec, and we obtain $6DT_1c/a^2 \approx 0.2$, which questions the use of (6). For $6DT_1c/a^2 \ll 1$, the slow diffusion treatment of Abragam¹² gives (for $\omega T_{1e} \gg 1$)

$$
T_{1}^{-1} \propto \omega^{-1/2} T_{1e}^{-1/4} \ . \tag{17}
$$

This gives a weak temperature dependence (through T_{1e} , and the magnitude calculated with Abragam's full expression [Eq. (44), p. 382] is in fair agreement with our room-temperature values. However, the frequency dependence is weaker than our observed $T_1^{-1} \sim \omega^{-1}$. It is possible that we are in an intermediate diffusion regime where neither (6) nor (17) is appropriate. A further complication is the spin-diffusion barrier¹⁶ which has not been accounted for in (17) .

Two modifications of relaxation equations similar to (6) have been proposed to account for weaker frequency dependences. First, we conweaker frequency dependences. First, we consider that of Walstedt $et al.,¹⁷$ used to explain T_1 data for the ²³Na resonance in Na β -alumina. Their explanation proposes that there is not just a simple single-valued energy barrier to be overcome in the diffusion process, but a distribution of activation energies due to imperfections with the crystal lattice. This model changes the frequency dependence at and just below the T maximum essentially by spreading the ω^{-1} dependence of the maximum over a range of temperatures in which different components of the energy distribution are going through the $\omega \tau \approx 1$ condition. Such a distribution might explain the frequency dependence of the T_1^{-1} maxima in β -Li₅AlO₄, but not the ω^{-1} dependence in α -Li₅AlO₄ away from the maximum.

The second possible explanation assumes that there are more than one type of crystographically different sites for Li' occupation, with different energy barriers to be overcome and different thermal equilibrium populations. The discussion in Sec. II shows this could readily be the case for Li,A10,. Several workers have modeled relaxation Li_sAlO₄. Several workers have modeled relaxat
for such cases.^{18,19} In such models, the therma average occupation of the higher-energy site (P $=pe^{-\Delta/kT}$) multiplies an expression like Eq. (6) to give T_1^{-1} . One then obtains an activation energy $E_a + \Delta$ on the low-temperature side of the T_1^{-1} maximum and $E_a - \Delta$ on the high-temperature side.

Equation (8) for T_2^{-1} would also be multiplied by P and have a net activation energy $E_a - \Delta$. However, our T_2^{-1} and T_1^{-1} data show essentially the same activation energy, and thus do not fit this proposed explanation.

None of the possible explanations considered above is able to explain deviations from (6) observed in our data. The data thus remain as tests for theories of magnetic impurity relaxation in ionic conductors and of detailed Li' motion in Li₅AlO₄. The frequency dependence of T_1^{-1} and the diffusivity estimates suggest that the simple rapiddiffusion approach is less appropriate for α -Li_sAlO₄, and that the jump parameters we have obtained for it are less reliable. A more general treatment utilizing both particle and spin diffusion for all values of the ratio τ/T_{1g} is currently in progress.¹⁴ progress.

Our T_1^{-1} and T_2^{-1} measurements have shown thermally activated Li' motion in all the materials studied. The activation energy of α -Li_sAlO₄ $[0.83(2)$ eV is comparable to values obtained from recent conductivity measurements [0.94(4) eV] on samples without LiOH contamination which on samples without LiOH contamination which
were either in vacuum or under argon.²⁰ We note that in principle our NMR activation energy need not agree with that of conductivity measurements since the correlation function [Eq. (5)] involves Li' jumps into and away from neighboring sites to magnetic impurities, while the conductivity is determined by jumps in the bulk material. The impurities could change the energy of neighboring sites and produce a different activation energy for nuclear relaxation. The attempt frequency, 2 $\times 10^{12}(\times/2)$ sec⁻¹, is close to a typical phonon frequency $(0.10^{13} \text{ sec}^{-1})$ and is not anomalously low like values observed in several other ionic conductors.²¹ $ductors.^{21}$

We have shown that low concentrations of magnetic impurities in nominally pure ionic conductors can produce nuclear relaxation rapid enough to be 'the dominant relaxation mechanism. For $\tilde{I} \neq \frac{1}{2}$ nuclei, however, a secondpossible rapidrelaxation mechanism is the electric quadrupole interaction. We emphasize that the correct mechanism can be identified by the ratio of T_1/T_2 at high temperatures when only the central nuclear transition is being observed. The former gives $(I+\frac{1}{2})^2$, while the latter gives a ratio $\frac{1}{4}$ as large. For $I=\frac{3}{2}$, the quadrupole ratio has the value of 1, and thus cannot be distinguished from the case where all nuclear transitions are in communication with each other. The nuclear dipole-dipole interaction gives a ratio (= 1.4 for $I = \frac{3}{2}$) close to that of quadrupole interactions, 10 but can likely be removed from consideration by its weaker contribution to T_1^{-1} . However, an experimental ratio of 4 clearly identifies magnetic impurities. This appears to be the case for the 'Li resonances in the ionic conductors $Li_2Ti_3O_7$ [Ref. (21)] and β -LiAlSiO₄ [Ref. (22)]. The 63 Cu resonance in CuI [Ref. (23)] yields a ratio of 1, and its relaxation mechanism has been analyzed with the electric quadrupole interaction. Finally, the transition from a thermally activated T_1^{-1} at high temperatures to a weakly temperature-dependent T_1^{-1} at lower temperature (where we have argued $t_c \approx T_{1e}$) may be expected in other systems with magnetic impurities, provided other relaxation mechanisms (for example, phonons) do not obscure it.

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FIG. 4. T_1^{-1} and T_2^{-1} data for the ⁷Li resonance in γ -LiAlO₂ at 21.5 MHz (\bullet) with one datum at 12.7 MHz (+). Bepresentative error limits are shown for selected points. Note that the horizontal (temperature) scale is broken at the far right to include room-temperature data.

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APPENDIX

We report here the results of T^{-1}_1 and T^{-1}_2 measurements on γ -LiAlO₂. As with the other systems studied here, samples were sealed in quartz tubes under a partial argon atmosphere. With the exception of one datum, the results shown in Fig. 4 were obtained at 21.⁵ MHz.

 T_2^{-1} shows a slight peak just prior to the onset of motional narrowing at ~350 °C, again suggesting relaxation by magnetic impurities. On heating to $~500$ °C, two components are detected in the transverse magnetization decay curve.

T₁ begins to rapidly increase above 300 °C. At T_1^{-1} begins to rapidly increase above 300 °C. 550 and 650 °C two components were also found in the longitudinal magnetization relaxation curves. Thus at temperatures above 550 $^{\circ}$ C, both T_1^{-1} and T₂¹ show evidence of two distinct groups of Li⁺ ions in γ -LiAlO₂. While the motional narrowing indicates definite Li' motion, a given 'Li nucleus is evidently largely confined to one of two groups of sites. If a given nucleus samples all types of Li⁺ sites during the relevant experimental times,

- ¹I. D. Raistrick, C. Ho, and R. A. Huggins, Mater. Res. Bull. 11, 953 (1976).
- 2 R. T. Johnson, Jr., R. M. Biefeld, and J. D. Keck, Mater. Res. Bull. 12, 577 (1977).
- ³F. Stewner and R. Hoppe, Z. Anorg. Allg. Chem. 380, 241 (1971).
- ⁴F. Stewner and R. Hoppe, Z. Anorg. Allg. Chem. 381, 149 (1971).
- ⁵R. Hoppe and H. Konig, Z. Anorg. Allg. Chem. 430, 211 (1977).
- 6F. Stewner and R. Hoppe, Acta Crystallogr. B 27, 616 (1971).
- $^7E.$ R. Andrew and D. P. Tunstall, Proc. Phys. Soc. 78, 1 (1961).
- W . W. Simmons, W. J. O'Sullivan, and W. A. Robinson,
- Phys. Rev. 127, 1168 (1962).
⁹R. E. Walstedt, Phys. Rev. Lett. <u>19,</u> 146 (1967); <u>19,</u> 816(E) (1967).
- ^{10}P . A. Fedders, Phys. Rev. B 13, 2768 (1976).
- $11N$. Bloembergen, E. M. Purcell, and R. V. Pound,

we would expect a single-component relaxation curve, representing'a sample-averaged value of T_1^{-1} . The long T_1 value at 550-650 °C shows that motion between the two groups of sites is slower than \sim 3 sec. Our results do not mean that there are not two groups of sites at'lower temperatures, but if so the relaxation rates are too close to each other to separate. When they differ by $\neg \times 3$ as for the data above 550 \degree C, two distinct rates can be resolved.

The smaller T_2^{-1} and larger T_1^{-1} component appear to be associated with Li' diffusion with an activation energy of 0.60 eV. Expecting perhaps a maximum in T_1^{-1} , we measured the larger T_1^{-1} component up to 824 °C. Instead, we found a nearly component up to 621 C. Instead, we found a negotiation constant T_1^{-1} value from 550-824 °C. We cannot be absolutely certain that these results were not influenced by the second T_1^{-1} component. A frequency check of the larger T_1^{-1} component at 560 °C and 12.7 MHz showed little if any frequency dependence.

We emphasize that a more complete and careful study of both components of T_1^{-1} and T_2^{-1} is needed to better understand Li^+ diffusion in γ -LiAlO₂ and to confirm or dispell the above tentative analysis.

Phys. Rev. 73, 679 (1948).

- 12 A. Abragam, The Principles of Nuclear Magnetism (Oxford U.P., Oxford, England, 1961).
- 3E. Venturini (private communication).
- ¹⁴P. Richards (private communication).
- 15 R. D. Hogg, S. P. Vernon, and V. Jaccarino, Phys. Rev. Lett. 39, 481 (1977).
- 16 W. E. Blumberg, Phys. Rev. 119, 79 (1960).
- ¹⁷R. E. Walstedt, R. Dupree, J. P. Remeika, and A. Rodriguez, Phys. Rev. B 15, 3442 (1977).
- ¹⁸J. E. Andersen, J. Magn. Reson. 11, 398 (1973).
- 19 M. Polak and D. C. Ailion, J. Chem. Phys. 67, 3029 (1977).
- 20 R. M. Biefeld and R. T. Johnson (unpublished).
- ²¹B. A. Huberman and J. B. Boyce, Solid State Commun. 25, 759 (1978).
- $^{22}H.$ T. Weaver (private communication).
- ²³J. B. Boyce and B. A. Huberman, Solid State Commun. 21, 31 (1977).
-