

## Two-dimensional electron gas on the surface of alkali-earth metal based electrides: Assistance to overcome tunneling barriers in ohmic contacts

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van der Waals (vdW) stacking of two-dimensional (2D) metals and 2D semiconductors has attracted significant interest in metal-semiconductor junctions (MSJs). Unfortunately, the vdW gap always leads to large tunneling barriers even in ohmic contacts. Herein, by constructing 2D electrides possessing sufficient electron gas at the surface, the formation of quasibonds at MSJ interface is expected to overcome the challenge of contact resistance induced by vdW gap. Specifically, 2D  $\text{Ca}_2\text{XY}_2$  ( $X = \text{Ti, Zr, Hf}; Y = \text{N, P}$ ) electrides possess ultralow work functions ranging from 3.28 to 3.90 eV, accompanied by nearly free electrons on the surface, rendering them efficient electron donors. Taking typical 2D semiconductor  $\text{MoS}_2$  to contact  $\text{Ca}_2\text{XY}_2$ , the ohmic contact and complete tunneling effect can be achieved. Application of a modest bias voltage yields a noticeable current density of about  $0.6 \mu\text{A}/\text{\AA}^2$ . Moreover, these MSJs exhibit superior environmental stability with bromine terminated. Our work not only offers a series of promising 2D electrides, but also paves the way for advancing the progress of 2D electronic and optoelectronic devices.

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### I. INTRODUCTION

With the increasing demand for miniaturization of devices, two-dimensional (2D) metals provide a promising platform for developing nanoscale electronic devices [1–4]. As the crucial components of electronic and photoelectronic devices, metal-semiconductor junctions (MSJs) with ohmic contacts are essential to the device's performance. Compared with the bulk metals, 2D metals lack dangling bonds on the surface, which interact with 2D semiconductors via van der Waals (vdW) forces, significantly suppressing the chemical disorder and releasing the lattice mismatch. This largely avoids metal-induced gap states and disorderly induced gap states. Therefore, the Fermi-level pinning effect is weak and the Schottky-Mott limit can be approached in 2D MSJs [5–9]. However, the intrinsic vdW gap leads to a large tunneling barrier, limiting the tunneling probability [10,11]. In previous studies, transition-metal dichalcogenides  $AB_2$  ( $A = \text{Nb, Ta}; B = \text{S, Se}$ ), MXenes, graphene, germanene, etc., have been utilized as electrodes in 2D MSJs [12–17]. Although some approaches such as hydrogenated interface [18] and polarization reversal [19,20] have been employed to meet the above two requirements, these strategies are not satisfactory and bring inconvenience in practical operation. Therefore, it is urgent to

explore high-effect contact modes for overcoming vdW gap in 2D ohmic contacts.

Interestingly, the intrinsic features of 2D electrides, such as  $\text{Ca}_2\text{N}$  [21], provide a potential platform for highly efficient carrier injection/extraction. On the one hand, 2D electrides possess ultralow work functions, facilitating the formation of ohmic contacts. On the other hand, freely diffused anionic electrons, that is, free 2D electron gas [22,23], make them effective electron donors. For example, the interface of 2D  $\text{Ca}_2\text{N}$  and  $\text{ZrS}_2$  dominates by the quasibond interaction, and the vdW gap can be filled with the massive free electrons [24]. Inserting 2D  $\text{Ca}_2\text{N}$  between  $\text{MoS}_2$  and bulk Au can effectively eliminate both the tunneling and Schottky barriers due to the charge transfer from the surface of  $\text{Ca}_2\text{N}$  [25]. However, harnessing these exceptional properties to improve contact characteristics in 2D MSJs is also accompanied with certain drawbacks. For example, the environmental instability of  $\text{Ca}_2\text{N}$  presents a significant obstacle to further development [26]. Up to now, except 2D  $\text{Ca}_2\text{N}$  and its derivatives predicted as  $M_2K$  ( $M = \text{Ca, Sr, Ba}; K = \text{N, P, As, Sb}$ ) [27], obtaining workable 2D electrides still faces significant challenges to high-performance 2D MSJs.

Herein, enlightened by  $P_{n+1}Q_n$  configurations of MXenes ( $n = 1, 2, \dots$ ) [28], rearranging atomic layers of existing 2D electrides may have great potential to improve the performance of 2D MSJs. Through redesigning the atomic layer arrangement of 2D  $\text{Ca}_2\text{N}$ , the layers' arrangement from top to bottom is set as  $\text{Ca-Y-X-Y-Ca}$  ( $X = \text{Ti, Zr, Hf}; Y = \text{N, P}$ ). Then, 2D  $\text{Ca}_2\text{XY}_2$  electrides obtained possess ultralow work functions ranging from 3.28 to 3.90 eV and nearly

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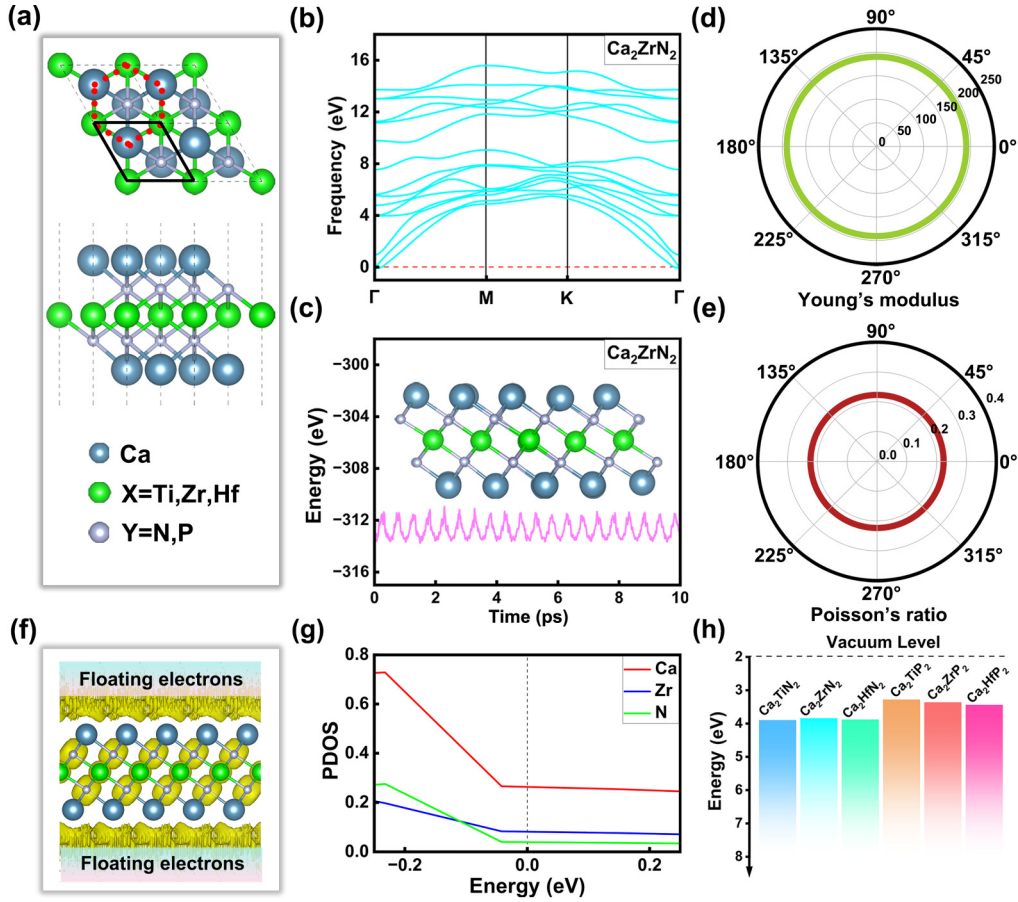


FIG. 1. (a) Top and side views of 2D  $\text{Ca}_2\text{XY}_2$ . (b) The phonon spectrum of 2D  $\text{Ca}_2\text{ZrN}_2$ . (c) Energy fluctuation from AIMD simulations for 2D  $\text{Ca}_2\text{ZrN}_2$  at 300 K. Insets show the simulation's snapshot after 10 ps. In-plane (d) Young's modulus and (e) Poisson's ratio for 2D  $\text{Ca}_2\text{ZrN}_2$  as a function of angle  $\theta$ . (f) ELF of 2D  $\text{Ca}_2\text{ZrN}_2$ . (g) PDOS of 2D  $\text{Ca}_2\text{ZrN}_2$ ; the red, blue, and green lines represent the contributions of Ca, Zr and N atoms, respectively. (h) Work functions of 2D  $\text{Ca}_2\text{XY}_2$ .

free 2D electrons gas on the surface, making them effective electron donors. Given the significance of  $\text{MoS}_2$  in the field of 2D semiconductors, we employ  $2H$ - $\text{MoS}_2$  to interface with 2D  $\text{Ca}_2\text{XY}_2$ , unveiling their contact properties. Excitingly, all  $\text{MoS}_2/\text{Ca}_2\text{XY}_2$  junctions exhibit ohmic contacts with a 100% tunneling probability, and structural stability can be maintained through *ab initio* molecular dynamics (AIMD) simulations. The transmission spectra and  $I$ - $V$  curves demonstrate the excellent transport properties of  $\text{MoS}_2/\text{Ca}_2\text{HfY}_2$  junctions. Additionally, bromination treatment on the uncontacted surface remarkably enhances the device's stability in the air.

## II. COMPUTATIONAL METHODS

First-principles calculations were performed with the Vienna *Ab initio* Simulation Package (VASP) under the framework of the Perdew-Burke-Ernzerhof functional and the generalized gradient approximation [29–31]. The energy cut-off for plane-wave basis was set to 450 eV. The convergence criteria of electron and ion relaxation were set as  $10^{-6}$  eV per atom and 0.01 eV/Å, respectively. The  $k$ -point sampling in the Brillouin zone was performed by employing the Monkhorst-Pack  $k$ -point meshes of  $10 \times 10 \times 1$  [32]. To eliminate the interactions between adjacent layers, a vacuum space of more

than  $15 \text{ \AA}$  along the  $z$  direction was adopted. DFT-D3 method of Grimme with zero-damping function was used to describe van der Waals interactions [33]. Phonon dispersion spectrums were calculated with the  $4 \times 4 \times 1$  supercell by the PHONOPY code within the density-functional perturbation theory to evaluate the dynamical stability [34,35]. Thermodynamic stability was checked via AIMD by the Nosé thermostat method in the  $NVT$  ensemble at 300 K [36,37]. The simulation time was 10 ps with a time step of 1 fs. Bader analysis was performed to evaluate the number of charge transfers [38]. To match the lattice constant between  $\text{Ca}_2\text{XY}_2$  and  $\text{MoS}_2$  as much as possible, the  $1 \times 1 \times 1 \text{ Ca}_2\text{XN}_2$  and  $1 \times 1 \times 1 \text{ MoS}_2$  and  $\sqrt{3} \times \sqrt{3} \times 1 \text{ Ca}_2\text{XP}_2$  and  $2 \times 2 \times 1 \text{ MoS}_2$  supercell were adopted. The lattice constant mismatching rate was controlled within 5% (cf. Table S2 in the Supplemental Material (SM) [39]). Integrating local density of states (ILDOS) was utilized to visualize the surface states in real space:

$$Q(\mathbf{r}) = \int_{E_0}^{E_F} \sum_{\{k\},n} |\Psi(\mathbf{r}, E)|^2 \delta(E - E_{\{k\},n}) dE, \quad (1)$$

in which  $E_F$ ,  $E_0$ ,  $n$ , and  $E_{\{k\},n}$  represent the Fermi level, bottom edge of the surface band, band index, and eigenenergies, respectively. An example of surface band is the band branches highlighted in bold sky-blue lines near the Fermi level [see

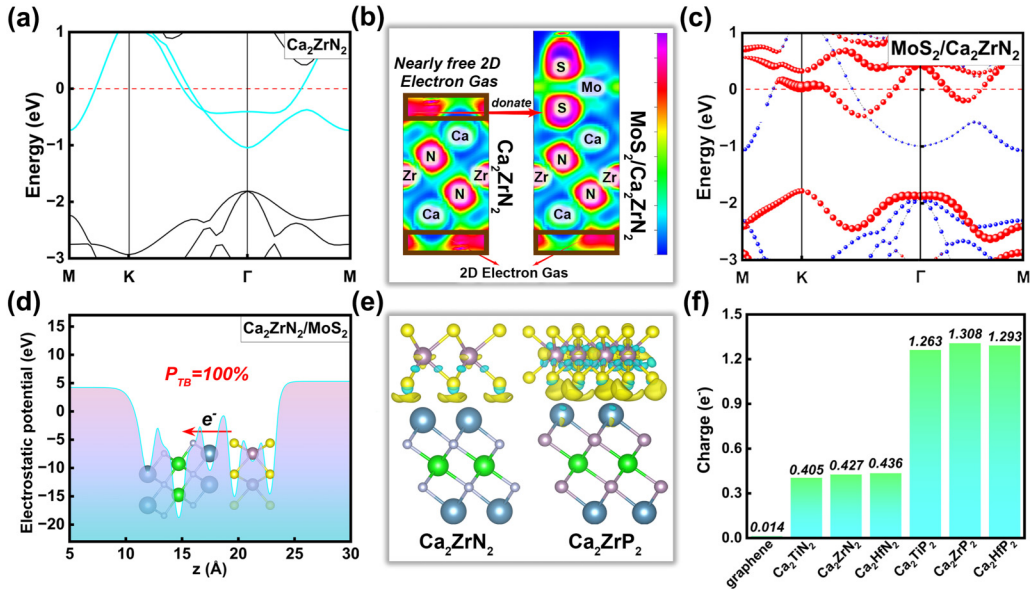


FIG. 2. (a) The band structure of 2D Ca<sub>2</sub>ZrN<sub>2</sub> in the  $M$ - $K$ - $\Gamma$ - $M$  path, and the Fermi level is set to zero. (b) ELF of 2D Ca<sub>2</sub>ZrN<sub>2</sub> and MoS<sub>2</sub>/Ca<sub>2</sub>ZrN<sub>2</sub> junction. (c) The projected band structure of MoS<sub>2</sub>/Ca<sub>2</sub>ZrN<sub>2</sub> junction with the Fermi level set to zero. The red and blue dots represent the contributions of MoS<sub>2</sub> and Ca<sub>2</sub>ZrN<sub>2</sub>, respectively. (d) The electrostatic potential of MoS<sub>2</sub>/Ca<sub>2</sub>ZrN<sub>2</sub> junction. (e) Plane-averaged charge-density difference between MoS<sub>2</sub> and Ca<sub>2</sub>ZrN<sub>2</sub>. (isovalue: 0.004 e/Å<sup>3</sup>; yellow: electron accumulation; blue: electron depletion) (f) Quantity of electrons transfer from Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub> (graphene) to MoS<sub>2</sub> layer.

Fig. 2(a)]. By integrating the planar average of ILDOS,  $Q(z)$  from the Ca atoms to the vacuum region, the surface charges ( $Q_{SC}$ ) of 2D Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub> can be evaluated as

$$Q_{SC} = \int_{z_1}^{z_2} Q(z) dz, \quad (2)$$

where  $z_1$  and  $z_2$  represent the position of the Ca atoms and the vacuum region, respectively.

The quantum transport properties of MoS<sub>2</sub>/Ca<sub>2</sub>HfY<sub>2</sub> junctions were calculated by using nonequilibrium Green's function in the NANODCAL package [40]. The convergence criterion of density matrix was set as 10<sup>-6</sup> hartree and a 1 × 9 × 1  $k$ -point mesh was employed to perform the self-consistent calculations. A 1 × 100 × 1  $k$ -point mesh was used to calculate the current-voltage curve and transmission spectrum for the central region. The transmission coefficient was calculated by the following formula:

$$T(E) = \text{Tr}[\Gamma_L(E)G^r(E)\Gamma_R(E)G^a(E)], \quad (3)$$

where  $\Gamma_L$  and  $\Gamma_R$  represent a linewidth function for describing the coupling between the channel and electrode. The advanced Green function  $G^a$  is the Hermitian conjugate of the retarded Green function  $G^r$ . The electric current is determined by the Landauer-Buttiker formula [41]:

$$I(V_b) = \int_{-V_b/2}^{V_b/2} I(E) dE = -\frac{2q}{h} \int_{-V_b/2}^{V_b/2} T(E) [f_L(E, V_L) - f_R(E, V_R)] dE, \quad (4)$$

where  $I(E)$  refers to the spectral current, and  $T(E)$  is the transmission coefficient under external bias voltage  $V_b = V_L - V_R$  in which  $V_L$  and  $V_R$  represents the potential of the left and right electrodes, and  $f_L/f_R$  is the Fermi-Dirac distribution function of the left/right electrodes.

### III. RESULTS AND DISCUSSION

#### A. Geometric structures of 2D Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub>

The crystal structure of 2D Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub> is shown in Fig. 1(a), in which the  $Y$  atoms are located as the central layer of Ca and  $X$  layers with strong bonding to make structures stable. The red dotted lines underline the region of cavity composed of Ca and  $X$  cations, satisfying the basic rule for electrides, namely, the metallic atoms comprising the coordination around the anionic electrons must exhibit electropositive characteristics [42]. The lattice constants of 2D Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub> range from 3.23 to 3.88 Å, comparable to the lattice constant of 2D Ca<sub>2</sub>N (3.57 Å) [43]. Additionally, bond length of Ca-N is 2.42 Å in 2D Ca<sub>2</sub>N, which barely alters in 2D Ca<sub>2</sub>XN<sub>2</sub> (cf. Table I), although the bond length of Ca-P becomes larger due to the smaller ionic radius of N<sup>3-</sup> compared to P<sup>3-</sup> [44], which is also comparable to the Ca-N bond length. Hence, this design strategy does not cause significant lattice variation, which is favorable to the structural stability. In the SM [39], Fig. S1 shows the x-ray-diffraction (XRD) patterns of 2D Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub>, providing the reference for the future experimental synthesis.

#### B. Stability of 2D Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub>

The cohesive energy serves as a thermodynamic foundation for the robust stability of 2D Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub>, which can be defined as

$$E_{co} = \frac{2E_{Ca} + E_X + 2E_Y - E_{Ca_2XY_2}}{5}, \quad (5)$$

where  $E_{Ca_2XY_2}$ ,  $E_{Ca}$ ,  $E_X$ , and  $E_Y$  are the total energies of Ca<sub>2</sub>X<sub>2</sub>Y<sub>2</sub>, single Ca,  $X$ , and  $Y$  atoms, respectively. The cohesive energies of Ca<sub>2</sub>XN<sub>2</sub> and Ca<sub>2</sub>XP<sub>2</sub> are about 5 and 3 eV per atom, respectively (cf. Table I). These values are

TABLE I. The lattice constants, bond lengths, cohesive energies, work functions, and Young's modulus of 2D  $\text{Ca}_2\text{XY}_2$ .

2D electrides	Lattice constant (Å)	Ca-Y bond length (Å)	X-Y bond length (Å)	Cohesive energy (eV per atom)	Work function (eV)	Young's modulus (N/m)
$\text{Ca}_2\text{TiN}_2$	3.230	2.338	2.200	4.859	3.895	164.106
$\text{Ca}_2\text{ZrN}_2$	3.330	2.400	2.299	4.907	3.837	190.273
$\text{Ca}_2\text{HfN}_2$	3.310	2.411	2.270	5.022	3.877	199.680
$\text{Ca}_2\text{TiP}_2$	3.790	2.745	2.568	2.840	3.279	82.240
$\text{Ca}_2\text{ZrP}_2$	3.880	2.759	2.694	2.894	3.359	92.264
$\text{Ca}_2\text{HfP}_2$	3.840	2.760	2.658	2.941	3.438	93.573

comparable with  $\text{MoS}_2$ 's (4.89 eV per atom) and silicene's (3.91 eV per atom) [45,46], implying the strongly bonded networks. Compared with 2D  $\text{Ca}_2\text{XP}_2$ , the larger cohesive energies of  $\text{Ca}_2\text{XN}_2$  reflect the stronger interactions between atoms, which is consistent with the smaller bond lengths in 2D  $\text{Ca}_2\text{XN}_2$ . The dynamics and thermodynamics stabilities further support the feasibility of the experimental synthesis. In the phonon dispersion of 2D  $\text{Ca}_2\text{XY}_2$ , the absence of imaginary frequency indicates the dynamical stability (see Figs. 1(b) and S2 in the SM [39]). For 2D  $\text{Ca}_2\text{XP}_2$ , their larger bond lengths mean the looser interaction with the neighboring atoms, explaining the flatness of the highest phonon mode in the high-energy region [47]. During the AIMD simulations, the energy oscillates around the equilibrium state and any structural reorganization is invisible (see Figs. 1(c) and S3 in the SM [39]).

Under the Born criteria [48], namely  $C_{11} > C_{12}$  and  $C_{11}, C_{22}, C_{66} > 0$ , the mechanical stability of 2D  $\text{Ca}_2\text{XY}_2$  is confirmed. Due to hexagonal symmetry, elastic constants  $C_{11}$  equal to  $C_{22}$ , and  $C_{66}$  can be written as  $\frac{C_{11}-C_{12}}{2}$ . Thus, 2D  $\text{Ca}_2\text{XY}_2$  is mechanically stable (cf. Table S1 in the SM [39]). Further, Young's modulus  $Y(\theta)$  and Poisson's ratio  $\nu(\theta)$  of 2D  $\text{Ca}_2\text{XY}_2$  can be obtained as follows [49]:

$$Y(\theta) = \frac{C_{11}C_{22} - C_{12}^2}{C_{11}\sin^4\theta + A\sin^2\theta\cos^2\theta + C_{22}\cos^4\theta}, \quad (6)$$

$$\nu(\theta) = \frac{C_{12}\sin^4\theta - B\sin^2\theta\cos^2\theta + C_{12}\cos^4\theta}{C_{11}\sin^4\theta + A\sin^2\theta\cos^2\theta + C_{22}\cos^4\theta}, \quad (7)$$

where  $A = \frac{C_{11}C_{22}-C_{12}^2}{C_{66}} - 2C_{12}$ , and  $B = C_{11} + C_{12} - \frac{C_{11}C_{22}-C_{12}^2}{C_{66}}$ . All these 2D  $\text{Ca}_2\text{XY}_2$  show large Young's modulus and strong mechanical isotropy (see Figs. 1(d) and S4 in the SM [39]). As illustrated in Figs. 1(e) and S5 [39], Poisson's ratio of 2D  $\text{Ca}_2\text{XY}_2$  is also isotropic, indicating their compressibility. The specific data are summarized in Table S1 [39]. These good intrinsic features establish a foundation for the application in 2D MSJs.

### C. Electronic structures of 2D $\text{Ca}_2\text{XY}_2$

Before exploring 2D  $\text{Ca}_2\text{XY}_2$ -based MSJs, it is essential to establish a primary understanding of the electronic structures of 2D  $\text{Ca}_2\text{XY}_2$ . Using 2D  $\text{Ca}_2\text{ZrN}_2$  as an illustration, electron-localization function (ELF) reveals that massive electrons are floating on the surface [see Fig. 1(f)], aligning with the behavior observed in typical electrides such as 2D  $\text{Ca}_2\text{N}$  [50]. As shown in Fig. 1(g), through the analysis of projected

density of states (PDOS), the dominating contribution of states near the Fermi level comes from Ca elements, facilitating electrons to escape to the vacuum level easily. Further, using ILDOS, we evaluate the  $Q_{\text{SC}}$  for 2D  $\text{Ca}_2\text{XY}_2$ . For 2D  $\text{Ca}_2\text{N}$ , its  $Q_{\text{SC}}$  we calculated is  $3.74 \times 10^{14} e^- \text{cm}^{-2}$ , which is in agreement with the existing result  $3.72 \times 10^{14} e^- \text{cm}^{-2}$  [51]. Figure S6 in the SM [39] shows the planar-averaged ILDOS of 2D  $\text{Ca}_2\text{XY}_2$ , their charge distribution in real space, and the specified values of  $Q_{\text{SC}}$ . As the atomic number of  $X$  increases, its peak intensity also gradually decreases, which is similar to the trend of ILDOS for  $\text{M}_2\text{K}$  [51]. Taking 2D  $\text{Ca}_2\text{ZrN}_2$  and  $\text{Ca}_2\text{HfN}_2$  as examples, according to PDOS (see Fig. S7 in the SM [39]), the state contributed by Zr/Hf (Ca) is weak (strong), corresponding to the valley (peak) in their planar-averaged ILDOS. Thus, the difference of  $Q_{\text{SC}}$  for 2D  $\text{Ca}_2\text{XY}_2$  should be attributed to the different lattice constants and the ability of  $X$  to bind electrons. All 2D  $\text{Ca}_2\text{XY}_2$  can exhibit ultralow work functions, ranging from 3.28 to 3.90 eV [see Fig. 1(h)]. Meanwhile, the fine metallicity of 2D  $\text{Ca}_2\text{XY}_2$  ensures them to serve effectively as an electrode for interfacing with 2D semiconductors (see Fig. S8 in the SM [39]). Therefore, in the 2D  $\text{Ca}_2\text{XY}_2$ -based interfaces, accompanied with the vdW gap filled by plentiful electrons, the tunneling barriers have a chance to be eliminated in ohmic contacts.

### D. Contact properties of $\text{MoS}_2/\text{Ca}_2\text{XY}_2$ junctions

A typical feature of 2D electrides is the nearly free 2D electron gas on the surface. Taking 2D  $\text{Ca}_2\text{ZrN}_2$  as an example, its band branches highlighted in bold sky-blue lines exhibit a parabolic pocket near the Fermi level, deemed as characteristic of electrides [see Fig. 2(a)] [50,52,53]. In the left panel of Fig. 2(b), 2D electron gas emerges on the surface of 2D  $\text{Ca}_2\text{ZrN}_2$ , rendering it an effective electron donor. Due to the importance of  $\text{MoS}_2$  in the field of MSJs [54], we choose 2H- $\text{MoS}_2$  to contact with 2D  $\text{Ca}_2\text{XY}_2$  for further exploration. As depicted in Fig. 2(b), when 2D  $\text{Ca}_2\text{ZrN}_2$  contacts with  $\text{MoS}_2$ , 2D electron gas is readily transferred to the  $\text{MoS}_2$  layer, thereby significantly impacting the electron density of  $\text{MoS}_2$ . The Fermi level enters the conduction band of  $\text{MoS}_2$ , indicating the formation of ohmic contacts (see Figs. 2(c) and S9 in the SM [39]). In addition, compared with 2D  $\text{Ca}_2\text{XN}_2$ -based MSJs, the band structure of  $\text{MoS}_2$  undergoes a more significant alteration in  $\text{MoS}_2/\text{Ca}_2\text{XP}_2$  MSJs. To clearly understand this behavior, taking  $\text{MoS}_2/\text{Ca}_2\text{XP}_2$  MSJ as an example, we compare the band structures of isolated  $\text{MoS}_2$  before and after contact. As shown in Fig. S10 [39], before



and after contact, the band structure of  $\text{Ca}_2\text{HfP}_2$  almost has no variation. Upon formation of the interface, two band branches are introduced to the gap of isolated  $\text{MoS}_2$ \_contact compared to the original band gap of isolated  $\text{MoS}_2$ \_no contact. It is well known that states in  $\text{MoS}_2$  close to the Fermi level are contributed by Mo instead of S atoms. However, charge density of these two bands is distributed not only on Mo, but also partially on S atoms. Thus, the slight structure distortion caused by enhanced interlayer interaction effectively reduces conduction-band minimum, and thus induces the formation of ohmic contact. The similar phenomenon can be observed in  $\text{MoS}_2/\text{Ca}_2\text{N}$  systems [55]. Note that in order to avoid affecting the intrinsic properties of  $\text{MoS}_2$ , it is reasonable to stretch 2D  $\text{Ca}_2\text{XY}_2$  in a small scale to match  $\text{MoS}_2$  [56–58]. Figure S11 in the SM [39] exhibits the band structures of 2D  $\text{Ca}_2\text{XY}_2$  under biaxial strain from  $-6$  to  $6\%$ . Although partial band branches slightly alter, the metallicity of 2D  $\text{Ca}_2\text{XY}_2$  remains fine and the pocketlike structures across the Fermi level do not change; thus, 2D electron gas cannot be affected by little strain.

Referring to the electrostatic potential profile depicted in Fig. 2(d), electrons can transport from the surface of  $\text{MoS}_2$  to 2D  $\text{Ca}_2\text{XY}_2$  without energy loss. The tunneling probability is defined as

$$P_{TB} = \exp\left(-\frac{2W_{TB}}{\hbar}\sqrt{2m\phi_{TB}}\right), \quad (8)$$

where  $\hbar$ ,  $m$ ,  $W_{TB}$ , and  $\phi_{TB}$  represent the reduced Planck constant, the mass of the free electron, and the width and height of the tunneling barriers (TB), respectively. According to Eq. (8), a square potential has been assumed to simplify the irregular barriers into square barriers, and the  $P_{TB}$  of all  $\text{MoS}_2/\text{Ca}_2\text{XY}_2$  junctions is 100% (see Fig. S12 in the SM [39]). In detail, the  $\text{MoS}_2$  obtains abundant electrons from 2D electron gas, forming a robust negative space-charge region on its surface. Subsequently, the complete tunneling effect can be achieved at the  $\text{MoS}_2$ - $\text{Ca}_2\text{XY}_2$  interfaces. Furthermore, according to Simmon's model under the square-barrier approximation, tunneling-specific resistivity ( $\rho_t$ ) can be calculated as [59,60]

$$\rho_t \approx \frac{4\pi^2\hbar W_{TB}^2 \exp\left(\frac{2(2m)^{1/2}}{\hbar}W_{TB}\phi_{TB}^{1/2}\right)}{q^2 \frac{(2m)^{1/2}}{\hbar}W_{TB}\phi_{TB}^{1/2} - 1}. \quad (9)$$

When  $W_{TB}$  is extremely close to zero, this equation will cause an unphysical negative value. Based on the low-bias approximation, the corrective equation is provided as follows:

$$\rho_t^* \approx \frac{8\pi^2\hbar W_{TB}}{3q^2(3m\phi_{TB})^{1.2}} \exp\left(\frac{2W_{TB}(2m\phi_{TB})^{1/2}}{\hbar}\right). \quad (10)$$

Since the value of  $W_{TB}$  is zero,  $\rho_t^* \approx 0$ , which is much smaller than that in  $\text{LSi}_2\text{N}_4$ -2D (3D) metals contacts ( $L = \text{Mo}, \text{W}$ ) [61,62]. Moreover, the interface charge redistribution between  $\text{MoS}_2$  and  $\text{Ca}_2\text{ZrY}_2$  can be demonstrated by the charge-density differences, which can be defined as

$$\Delta\rho = \rho_{\text{MSJ}} - \rho_{\text{Ca}_2\text{ZrY}_2} - \rho_{\text{MoS}_2}, \quad (11)$$

where  $\rho_{\text{MSJ}}$ ,  $\rho_{\text{Ca}_2\text{ZrY}_2}$ , and  $\rho_{\text{MoS}_2}$  represent the charge densities of the 2D MSJ,  $\text{Ca}_2\text{ZrY}_2$ , and  $\text{MoS}_2$ , respectively. Regardless

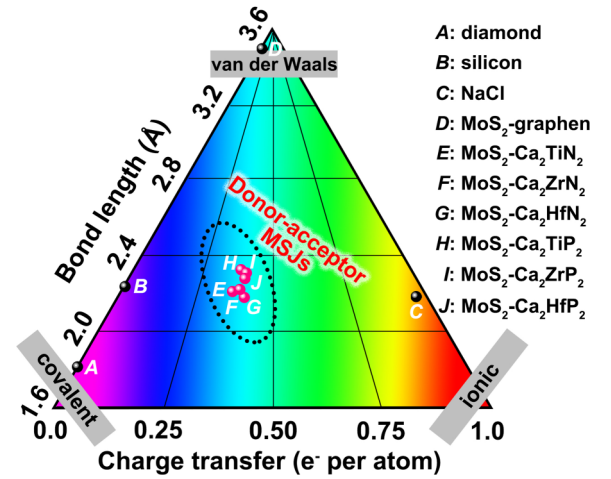


FIG. 3. Bond length of covalent, ionic, and vdW systems. The black balls represent some well-known examples such as diamond–silicon having no net electron transfer, typical ionic crystal NaCl, and the common  $\text{MoS}_2$ /graphene vdW junction. The pink balls give the information of bonding and charge transfer in  $\text{MoS}_2/\text{Ca}_2\text{XY}_2$  junctions. Here, charge transfer refers to the transfer from each Ca to  $\text{MoS}_2$  layer.

of whether they are N- or P-based electrides, the strength of charge rearrangement is remarkable [see Fig. 2(e)], which can be attributed to the large difference in work functions between  $\text{MoS}_2$  and  $\text{Ca}_2\text{ZrY}_2$ . In addition, Bader analysis indicates that the quantity of charges donated from 2D  $\text{Ca}_2\text{XY}_2$  to  $\text{MoS}_2$  significantly exceeds that observed in the graphene- $\text{MoS}_2$  contact [see Fig. 2(f)]. This behavior, different from the common 2D vdW system, implies the stronger interlayer interactions. Under this stronger coupling mode, the elimination of the tunneling barriers induced by the vdW gap is entirely comprehensible.

### E. Quasibond in donor-acceptor MSJs

In fact, the concept of quasibond provides a clearer understanding of this stronger coupling mode. Compared to the vdW forces, the systems dominated by quasibond show the shorter bonding distance and denser charge transfer [63,64]. As shown in Fig. 3, we choose 3.6 (1.6) Å as the bonding length to represent vdW (covalent) interaction, and the ultimate charge transfer in ionic bonding is  $1e^-$ . Within this framework,  $\text{MoS}_2/\text{Ca}_2\text{XY}_2$  systems are mainly concentrated in the central area of the “triangle,” behaving as the typical donor-acceptor MSJs. Specifically, in these donor-acceptor MSJs, the quasibond interaction causes the slight structural distortion of  $\text{MoS}_2$  (more apparent in  $\text{MoS}_2$ - $\text{Ca}_2\text{XP}_2$  contacts; see Fig. S13 in the SM [39]), thus further inducing the overlapped states [24]. Then, the moderately enhanced charge redistribution can effectively promote the contact properties. By introducing the concept of quasibond, it is clear that strengthened interlayer interaction is the key to eliminating vdW gap in 2D ohmic contacts.

With stronger interlayer interactions and enhanced charge transfer, the phenomenon of reconstruction may occur in 2D thin-layer structures. For example, 2D  $\text{Ca}_2\text{EN}_2$  ( $E = \text{Zr},$

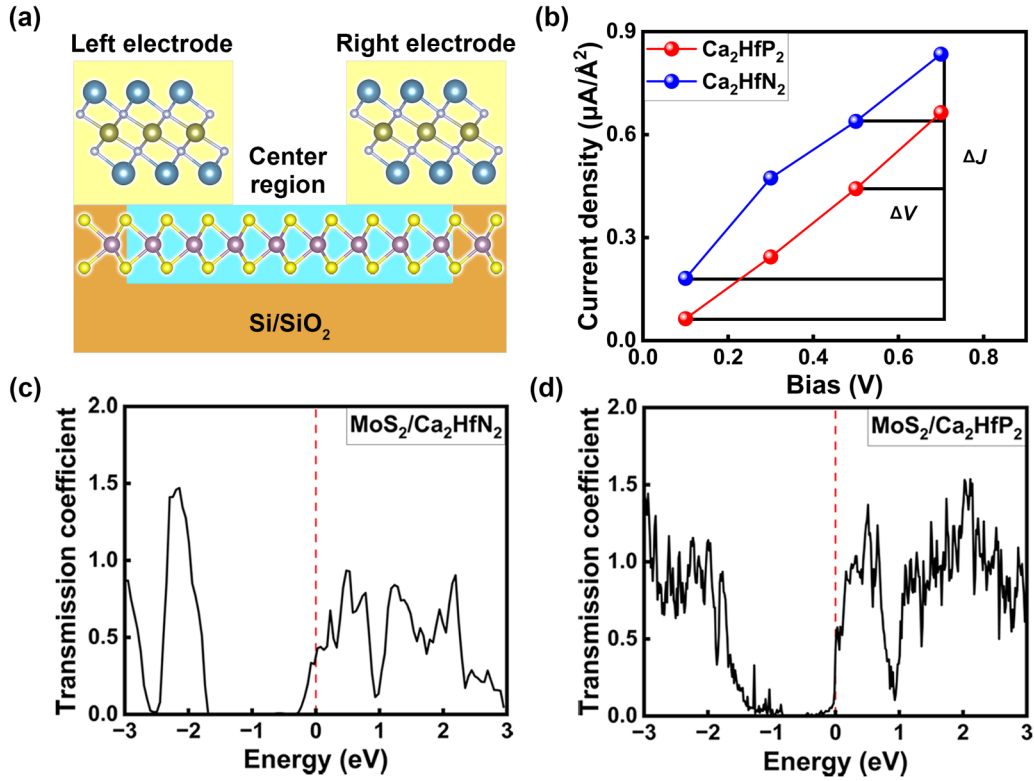


FIG. 4. (a) Schematic of a two-probe model of MoS<sub>2</sub>/Ca<sub>2</sub>HfN<sub>2</sub> junction. (b) The current-density–voltage curves for MoS<sub>2</sub>/Ca<sub>2</sub>HfN<sub>2</sub> and MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub> junctions. The transmission spectra with zero bias of (c) MoS<sub>2</sub>/Ca<sub>2</sub>HfN<sub>2</sub> and (d) MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub> junctions. The Fermi level is set to zero.

Hf) can effectively decrease the phase-transition barrier for 2*H*-MoTe<sub>2</sub> and stabilizes the 1*T'* phase. This phenomenon can be mainly attributed to the synergistic effect of the ultralow-energy difference (about 0.043 eV) between 2*H* and 1*T'*-MoTe<sub>2</sub> [65], as well as nearly free 2D electron gas on the surface. For MoS<sub>2</sub>, a large phase-transition energy of about 0.55 eV hardly allows similar behavior to be observed [66]. To confirm the structural collapse and phase transition do not exist in the MoS<sub>2</sub>/Ca<sub>2</sub>XY<sub>2</sub> systems, AIMD simulations are performed at 300 K for 10 ps. As shown in Fig. S14 [39], the energies smoothly fluctuate in a small range, and the structures of these 2D MSJs have no significant distortion or breakdown. The charge transfer from 2D Ca<sub>2</sub>XY<sub>2</sub> to MoS<sub>2</sub> is not sufficient to cause a phase transition. This point is in line with our original purpose of designing such electrides that can be utilized to realize stable and high-performance 2D MSJs.

#### E. Transport properties of MoS<sub>2</sub>/Ca<sub>2</sub>HfY<sub>2</sub> junctions

Furthermore, we take MoS<sub>2</sub>/Ca<sub>2</sub>HfY<sub>2</sub> junctions as instances to simulate their transport characteristics using the common two-probe model [see Fig. 4(a)]. In Figs. 4(c) and 4(d), the transmission spectra clearly show the ohmic contacts in MoS<sub>2</sub>/Ca<sub>2</sub>HfY<sub>2</sub> junctions. The transmission states crossed by the Fermi level allow the effective electron tunneling to form considerable interfacial currents. Figure 4(b) shows the roughly linear curves, and the current densities rapidly increase with the bias voltage added from 0.1 to 0.7 V, which

are typical features distinguished from Schottky contact. For nanoscale electronic devices, the impact of heating on their performance is also crucial [67]. Generally, the generated heat is proportional to the resistance. Resistance per unit area can be evaluated from *I*-*V* curves via the following equation:

$$R_s = \frac{\Delta V}{\Delta J}, \quad (12)$$

where *V* and *J* are voltage and current density, respectively. *R<sub>s</sub>* of MoS<sub>2</sub>/Ca<sub>2</sub>HfN<sub>2</sub> and MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub> junctions are 1.022 and 0.902 Å<sup>2</sup>V/μA, implying less heat loss in the MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub> junction. Thus, compared with 2D Ca<sub>2</sub>HfN<sub>2</sub>, 2D Ca<sub>2</sub>HfP<sub>2</sub> is the better potential candidate for the design of electronic devices based on MoS<sub>2</sub>.

#### G. Bromination of MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub> junctions

Like other electrides, the oxidation and hydrolysis also hinder the practical application of 2D Ca<sub>2</sub>XY<sub>2</sub>. The remaining 2D electron gas on the noncontact side can easily induce the chemical adsorption of oxygen (O<sub>2</sub>) and water (H<sub>2</sub>O) molecules in the air. Taking MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub> as an example, the adsorption energies indicate that O<sub>2</sub> and H<sub>2</sub>O can be adsorbed chemically on the 2D Ca<sub>2</sub>HfP<sub>2</sub> [see Fig. 5(a)]. Thus, an appropriate surface modification is necessary to stabilize the performance of devices. Halogenation on the surface of 2D materials has been considered an effective approach for modification [68,69]. Here, we employ bromination treatment to prevent the disintegration. Note that bromine-terminated surface will not remarkably alter the electronic structures,

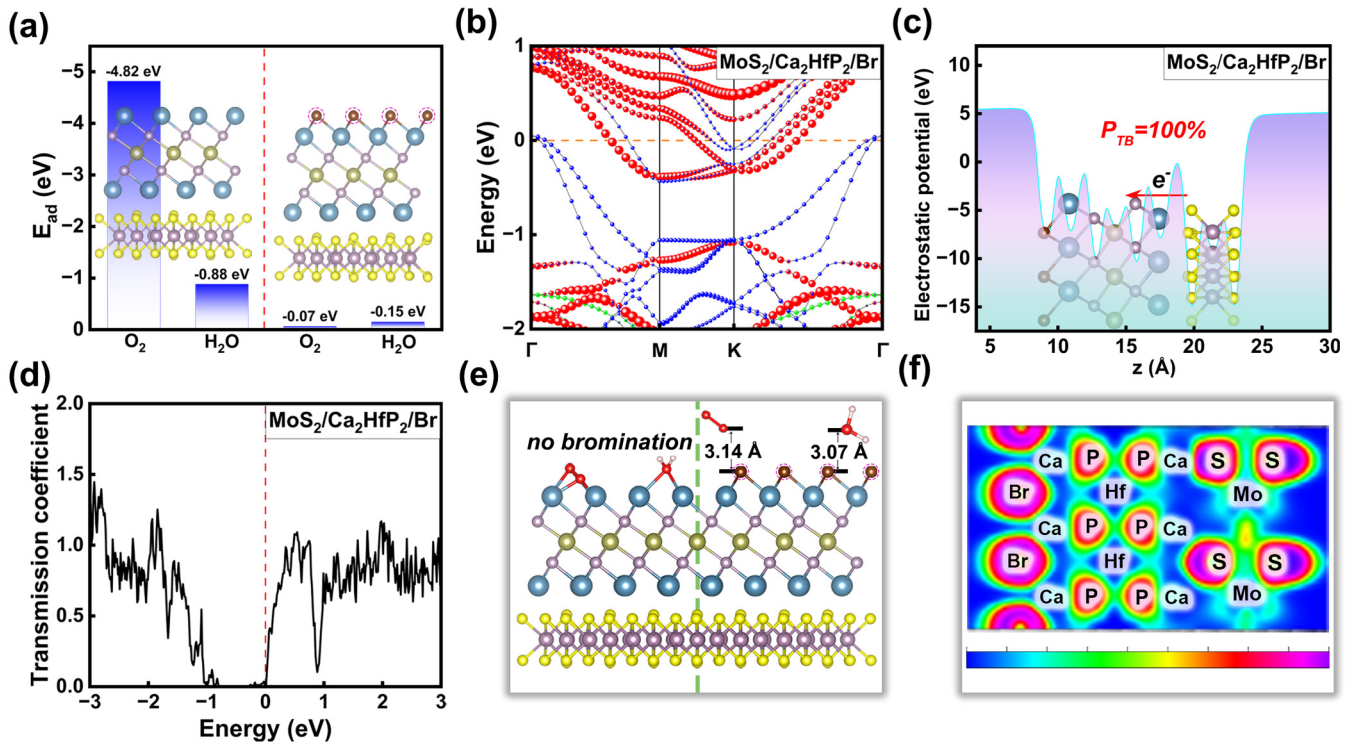


FIG. 5. (a) Adsorption energies of O<sub>2</sub> and H<sub>2</sub>O for MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub> (left panel) and MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub>/Br (right panel) junctions. (b) The projected band structure of MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub>/Br junction, and the Fermi level is set to zero. The red, blue, and green dots represent the contributions of MoS<sub>2</sub>, Ca<sub>2</sub>HfP<sub>2</sub>, and Br, respectively. (c) The transmission spectrum with zero bias, (d) electrostatic potential, and (f) ELF of MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub>/Br junction. (e) Optimized structures of MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub> and MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub>/Br junctions when O<sub>2</sub> and H<sub>2</sub>O float on the surface.

which can be illustrated by comparing band structures in Figs. 5(b) and S8(f). The contribution of Br to the band is slight and is mainly reflected in the energy region of  $-2$  eV, implying the little impact on the contact properties. As shown in Fig. 5(c), electrons transfer from MoS<sub>2</sub> to Ca<sub>2</sub>HfP<sub>2</sub>/Br without being hindered by tunneling barriers, implying that the filling of the vdW gap by massive electrons remains unaffected. Moreover, the transmission spectrum also indicates that the complete tunneling effect at the interface is uncompromised [see Fig. 5(d)]. Hence, the influence of additional Br on the electrical performance of devices can be disregarded.

On the brominated surface, O<sub>2</sub> and H<sub>2</sub>O can only be adsorbed physically due to the protection of Br atoms layer. As depicted in Fig. 5(e), the noncontact side of Ca<sub>2</sub>HfP<sub>2</sub> exhibits an inclination to bond with O<sub>2</sub> and H<sub>2</sub>O, while O<sub>2</sub> and H<sub>2</sub>O maintain approximately 3 Å from Br atoms in MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub>/Br junction. Moreover, ELF in Fig. 5(f) clearly illustrates that Br atoms can deplete the 2D electron gas on the non-contact surface, elucidating why O<sub>2</sub> and H<sub>2</sub>O no longer strongly bond with Ca<sub>2</sub>HfP<sub>2</sub>. Importantly, the AIMD simulation not only verifies the stability of MoS<sub>2</sub>/Ca<sub>2</sub>HfP<sub>2</sub>/Br junction (see Fig. S15 in the SM [39]), but also demonstrates the dynamical isolation effect of O<sub>2</sub> and H<sub>2</sub>O. The snapshots of the simulation for 2, 4, 6, and 8 ps can be searched in Fig. S16 [39], in which H<sub>2</sub>O throughout keeps about 3 Å away from the brominated interface while O<sub>2</sub> is repelled. This phenomenon is also in line with the fact that the adsorption energy of O<sub>2</sub> is almost zero, while that of H<sub>2</sub>O

is  $-0.15$  eV. This case of the surface modification allowing the device to be environmentally stable should also be practical for other 2D electrides.

#### IV. CONCLUSIONS

In summary, through redesigning atomic layers arrangement, we obtain a series of 2D Ca<sub>2</sub>XY<sub>2</sub> from the source material Ca<sub>2</sub>N. These 2D Ca<sub>2</sub>XY<sub>2</sub> not only behave steadily in dynamics and thermodynamics, but also possess typical electride properties. Since the bifunctional effect of the ultralow work functions and 2D electron gas on the surface, ohmic contacts can be formed in MoS<sub>2</sub>-Ca<sub>2</sub>XY<sub>2</sub> interfaces with the full tunneling effect. With a small bias voltage applied to the MoS<sub>2</sub>/Ca<sub>2</sub>HfY<sub>2</sub> junctions, the notable currents observed suggest the absence of Schottky barriers. Furthermore, the brominated surface serves as an effective barrier against hydrolysis and oxidation, and the excellent contact characteristics previously demonstrated are almost uncompromised. These findings provide a feasible case to design 2D electrides on the existing basis and stimulate more interest in eliminating vdWs gap in 2D MSJs.

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