

Emergence of near room-temperature superconductivity in hydrides with H₂ molecular unitsZhao Liu,¹ Junda Li,¹ Eva Zurek,² Quan Zhuang,³ Yanhui Liu,^{1,*} Jincheng Yue,¹ Siqi Guo,¹ Ao Zhang,¹ Zhenhua Chi,¹ Xiaoli Huang,⁴ and Tian Cui^{1,4,†}¹*Institute of High Pressure Physics, School of Physical Science and Technology, Ningbo University, Ningbo 315211, People's Republic of China*²*Department of Chemistry, State University of New York at Buffalo, Buffalo, New York 14260-3000, USA*³*Inner Mongolia Key Laboratory of Carbon Nanomaterials, Nano Innovation Institute (NII), College of Chemistry and Materials Science, Inner Mongolia Minzu University, Tongliao 028000, People's Republic of China*⁴*State Key Laboratory of Superhard Materials, College of Physics, Jilin University, Changchun 130012, People's Republic of China*

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The achievement of high-temperature superconductivity in compressed hydrides with extended lattices, e.g., H₃S and LaH₁₀, has become a milestone in the quest for room-temperature superconductivity. For realizing room-temperature superconductivity, lattices where hydrogen adopts multicentered bonds are deemed as indispensable, while hydrides containing H₂ molecular units are believed to be unfavorable. Here, we report H₂ molecular type hydrides with an exceptional near room-temperature superconductivity of 270 K in compressed NaH₁₀ and a T_c of 152 K in NaH₁₂, where H atoms solely constitute H₂ units, and Na-H forms ionic bonds. Our first-principles calculations unveil that the high T_c is mainly attributed to strong electron-phonon coupling stemming from the large electron-phonon matrix element driven by medium-frequency interatomic interactions and high-frequency H-derived phonon softening caused by Fermi surface nesting, thus scattering itinerant electrons to form Cooper pairs. Of particular note, we reveal that the unique delocalized background charges cooperate with other electrons occupying the pressure-induced *sp*-hybridized antibonding bands of molecular H₂ units, acting as itinerant electrons to mediate metallic interactions and participate in electron-phonon coupling. This observation reshapes the understanding of superconductivity dominated by molecular H₂ units, provides insights for elucidating phonon-mediated superconductivity, and raises broad prospects of realizing room-temperature superconductivity in molecular hydrogen-based superconductors.

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The quest for new superconducting materials and the elucidation of their microscopic mechanism of superconductivity has been a grand challenge [1,2] ever since the first ground breaking experiments of Onnes. Ashcroft proposed that the lightest element, hydrogen, would evolve into a metallic solid phase at high pressures, and its high phonon frequencies and robust electron-phonon coupling (EPC) would be a key ingredient required to achieve high- T_c phonon-mediated superconductivity [3]. Despite great experimental strides, it is still debated if a high-pressure phase of solid metallic hydrogen has been made in diamond anvil cells, and the huge pressures required to metallize hydrogen in the solid have challenged the capability of current experiments [4,5]. Hydrides, subsequently, were selected as alternative candidate materials for exploring high- T_c superconductivity ever since Ashcroft proposed that chemical precompression [3,6] can lower the pressure required for metallization. Extensive theoretical studies have predicted a number of binary hydrides with high critical superconducting temperatures (T_c 's), finding two general classes of superconducting materials [7–12]. The first includes *p*-block elements that bond with hydrogen

covalently in novel proportions, for example, H₃S, which was the first hydride found experimentally and theoretically nearly simultaneously to attain a record-breaking T_c of 203 K at 155 GPa [7,13]. The second includes electropositive metal elements whose valence electrons are transferred to hydrogenic lattices. In particular, experiments have verified predictions of high- T_c superconductivity in phases with clathratelike lattices such as YH₉ (243 K at 200 GPa), LaH₁₀ (260 K at 180 GPa) [14–19], and CaH₆ (215 K at 172 GPa) [20,21]. These three-dimensional clathratelike hydrogen lattices, with multi-centered hydrogen bonds, are deemed as being indispensable to exceptional superconductivity [2].

However, with the exception of molecular hydrogen, hydrides that contain H₂ units are not known to possess the highest T_c under pressure. The predicted T_c of molecular H-III phase (*Cmca*-12) is 242 K at 450 GPa [22]. The H₂ unit constituted *Ama*2-GeH₄ and *P6*₃/*mmc*-SnH₄ phases are predicted to possess T_c 's of 57 and 62 K, and H₂ units are considered to play a dominant role in improving the EPC [23,24]. In addition, SiH₄(H₂)₂ and GeH₄(H₂)₂ containing more H₂ molecule units are synthesized with theoretically predicted T_c 's of 107 and 90 K [25,26]. Nevertheless, the near room-temperature superconductivity is never obtained in H₂ molecular hydrides and, instead, hydrides containing H₂ units have recently been regarded as an unfavorable factor

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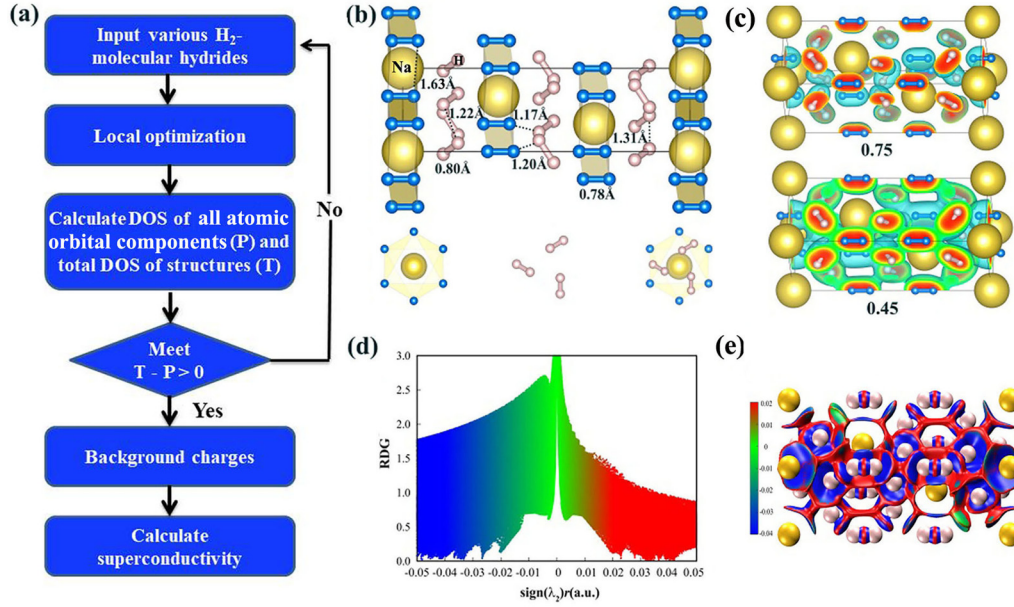


FIG. 1. (a) Schematic illustrating the methodology for identifying potential H₂ molecular type hydrides with high T_c . (b) The R32-NaH₁₀ phase at 300 GPa and (c) three-dimensional isosurface of the electron localization function (ELF). (d), (e) Plots of the reduced density gradient (RDG) vs electron density (ρ) in terms of the sign of the second Hessian eigenvalue (λ_2) [47].

for superconductivity [2,27]. Whether the H₂ molecular structure can induce a robust EPC and (near) room-temperature superconductivity in hydrides has not yet been determined and remains challenging.

The metallization of phases containing H₂ units can be understood via considering the knowledge of compressed alkali metal polyhydrides [28]. Being attributed to the transfer of valence electrons from the electropositive metal to H₂- σ^* orbitals thus partially filling the H₂- σ^* bands, the *R*-3*m*-LiH₆ phase generated a high density of states (DOS) at the Fermi level (ε_F). Meanwhile, “Madelung precompression” generated a smaller volume than what would be expected at relatively low pressures [28]. The pressure-induced broadening and overlap of H⁻ and H₂- σ^* bands metallize the *Pm*-NaH₉ phase [29,30]. In KH₅ and RbH₅ species with H₃⁻ counteranions and H₂ bystander molecules, metallization occurred via pressure-induced widening and the final overlapping of the filled H₃⁻ nonbonded band with the metal *d* bands [6,31,32]. The potential T_c of phonon-mediated H₂ molecular type hydrides is markedly influenced by the DOS at ε_F . Intriguingly, we noticed an additive inconsistency between the total DOS and the sum of discrete DOS for different elements in the electronic structure of SbH₄ [33], which are composed of H₂ units (see Fig. S1 in the Supplemental Material [34]). This discrepancy results in the emergence of novel background charges [33,46]. To investigate whether the background charge as a free electron participates in the EPC and induces high-temperature superconductivity holds significant research value in elucidating phonon-mediated superconductivity for H₂ molecular type hydrides. Herein, we proposed a high-throughput structural search approach to screen H₂ molecular type hydrides to quickly identify potential high- T_c superconductors based on the presence of background charges near the ε_F [see Fig. 1(a)]. Application of the proposed method identified and confirmed H₂ molecular type hydrides NaH₁₀, TeH₄, and SnH₁₄

possessing background charges near the ε_F [30,33,48]. We take the compressed NaH₁₀ as representatives to investigate superconductivity, elucidate the influence of background charges on superconducting behavior, and tackle the challenge of achieving room-temperature superconductivity in H₂ molecular type hydrides.

In this Letter, we confirm a pressure-induced H₂ molecular type superconductor, NaH₁₀, with a T_c of 270 K at 4 Mbar, possessing an unexpected near room-temperature superconductivity. Our first-principles calculations illustrate that large electron-phonon matrix elements are driven by the diverse medium-frequency interatomic interactions, aided via high-frequency H-derived phonon softening caused by FS nesting that scatters itinerant electrons to form Cooper pairs. Notably, these itinerant electrons partially act as delocalized background charges and partially occupy the pressure-induced *sp*-hybridized antibonding bands of H₂ units to participate in EPC. Further testing the presence of H₂ molecular type hydrides on the resulting superconductivity, we doped sodium into the H-III phase. Our computational experiment yielded a dynamically stable *R*-3*m*-NaH₁₂ structure with a T_c of 154 K at 150 GPa, where itinerant electrons occupied *sp*-hybridized orbitals showing delocalized characteristics. These results demonstrate that near room-temperature superconductivity is indeed possible in H₂ molecular type hydrides. Our study reshapes the traditional understanding of superconductivity dominated by H₂ units and provides insights for elucidating phonon-mediated superconductivity in molecular hydrogen based superconductors.

Our enthalpy calculation identified that molecular hydride NaH₁₀ is stabilized above 200 GPa with dynamical stability and has a relatively higher enthalpy than NaH₉ + 1/2H₂ compounds [49] (see Fig. S2 and Scheme S2 in the Supplemental Material [34]). Notably, NaH₁₀ lives on the convex hull when the pressure is above 360 GPa. At 300 GPa, the

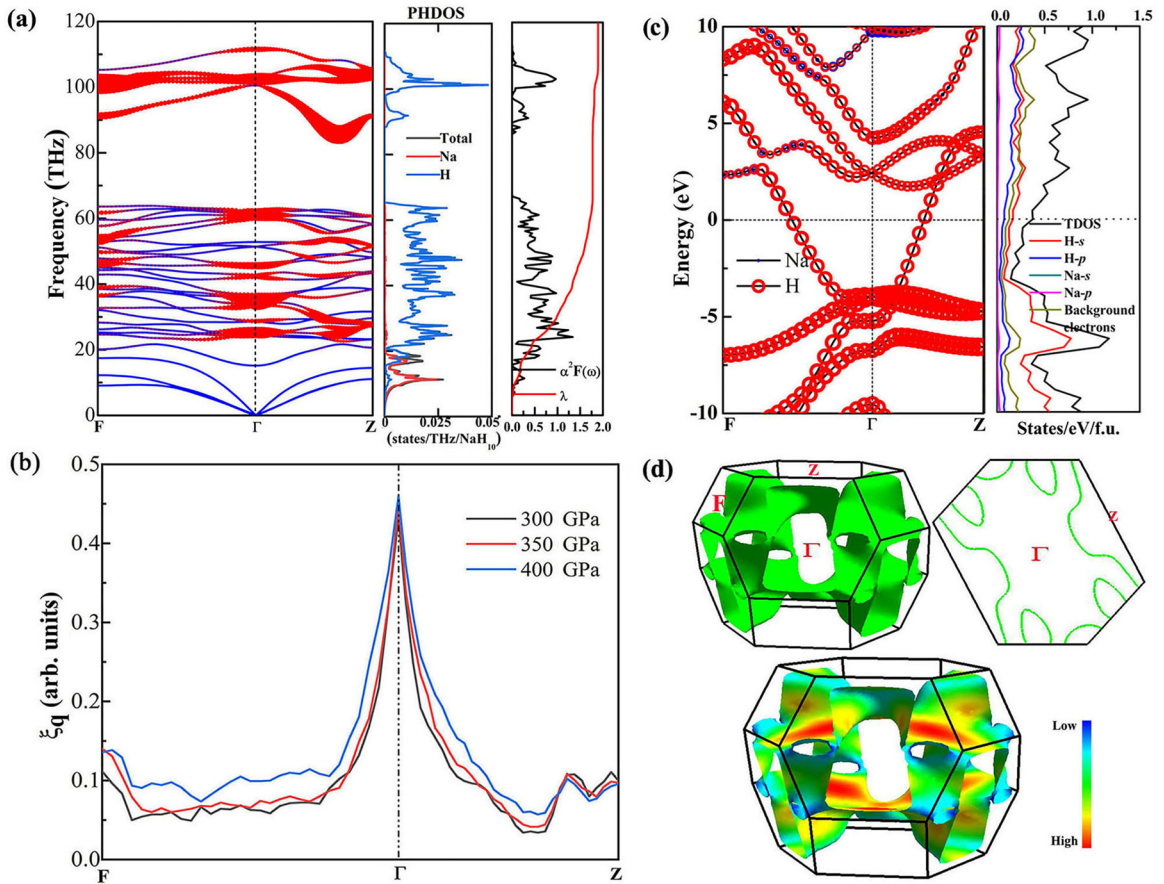


FIG. 2. (a) Phonon dispersion curves, phonon density of states (DOS), and Eliashberg spectral function $\alpha^2 F(\omega)$ together with the integral $\lambda(\omega)$ of the R32-NaH₁₀ phase at 300 GPa. The size of the solid dot on phonon spectra signifies the contribution to the linewidth. (b) The calculated nesting function ξ_q with pressure evolution along special q trajectories. (c) Electronic band structure and projected DOS. (d) The FS (upper) and corresponding Fermi velocity (bottom).

H–H bond lengths of 0.78–0.80 Å indicate typical H₂ molecular characteristics, as shown in Fig. 1(b) [23]. Along the c axis, this structure includes three alternate layers of sodium intercalated hydrogen molecules, three-molecule rings with unbound molecular hydrogen and elongated H₂ molecules. The distance between adjacent H₂ units, >1.17 Å, is much larger than the bond length of H₂ gas at ambient conditions, 0.75 Å. The Na–H distance falls between 1.67 and 2.72 Å, suggesting that NaH₁₀ is an ionic crystal. In Fig. 1(c), there is no electron localization function (ELF) value at the interval of the H₂ units when the isosurface of 0.75 is adopted, suggesting that there are no covalent interactions. However, metallic bonds have been formed in Fig. S3 [34], as evidenced by the distribution of the free electrons at an isovalue of 0.45.

To reveal the precise interactions between H₂ units, we calculated the key index, the reduced density gradient (RDG), defined as $1/6\pi^2 |\nabla \rho| / \rho^{4/3}$ in Fig. 1(d), where ρ represents the electron density. Abundant spikes have large negative values of $\text{sign}(\lambda_2)\rho$, and RDG trends to zero, characterizing the noncovalent interactions between H₂ units [50]. For the symmetrical distribution of weak chemical bonding, the vdW interactions located in a ρ range of -0.01 – 0.01 a.u. with negative and positive values of $\text{sign}(\lambda_2)$. The apparent spikes of RDG corresponding to $\text{sign}(\lambda_2)\rho$ associated with

positive $\text{sign}(\lambda_2)$ value greater than 0.03 a.u. indicate that steric effects, i.e., repulsive interactions, exist in intermolecular positions of H₂ units, implying that there are electrons occupying its antibonding molecular orbital. These results are also confirmed in Fig. 1(e), where the colors of dark blue and red correspond to attractively ionic and repulsively electrostatic interactions, respectively.

To estimate the superconductivity of the R32 phase, we have applied the Allen-Dynes modified McMillan equation based on harmonic approximation to calculate the T_c [51]. For pressure at 300 GPa, the phonon dispersion curves, partial phonon DOS, $\alpha^2 F(\omega)$, and EPC λ are displayed in Fig. 2(a). The high-frequency phonons corresponding to the vibration modes of H₂ units are confirmed in Fig. S4 [34], reflecting that this phase is mainly composed of H₂ units. The high-frequency vibrations are consistent with the vibrational frequencies of molecular hydrogen in the H-III phase, especially above 90 THz, which is vastly different from the $I4_1/amd$ -H-IV phase of metallic hydrogen containing atomic-H sublattices at 500 GPa [52,53]. Combined with the $\alpha^2 F(\omega)$ and the partial phonon DOS analysis, the most significant contributions, $\sim 80\%$ to the EPC, stem from diverse interatomic interactions identified in Fig. S4 [34], accumulated in medium-frequency vibration modes within 25–65 THz. On the other hand, the vibration modes above 90 THz only

contribute 6.36% to EPC. Medium-frequency optical phonon vibrations play a major role in scattering electrons forming Cooper pairs and enhance T_c , which can be proved by the peaks of $\alpha^2F(\omega)$, and integral $\lambda(\omega)$ and projection linewidth γ_{qv} . However, the contribution from the low-frequency translational vibrations constitutes 13.64% of the integral λ . The corresponding T_c of 238 K estimated with $\mu^* = 0.1$ is close to LaH₁₀ of 254 K. Particularly, the average phonon frequency ω_{\log} and λ reach 1443 K and 1.86, respectively. Compared with the vibrational characteristics and EPC of LaH₁₀ at 300 GPa, the ω_{\log} of NaH₁₀ is close to the 1488 K obtained for LaH₁₀, and $\lambda(\omega)$ is higher than the 1.78 calculated for LaH₁₀ [15], pointing out two important factors inducing the R32-NaH₁₀ phase to possess a high T_c . The ability of R32-NaH₁₀ to possess a high average phonon frequency is mainly attributing to the shortest H–H bond length of only 0.78–0.80 Å, severely shorter than atomic hydrogen type $Fm\bar{3}m$ -LaH₁₀, where $r_{H-H} > 1.0$ Å. The relatively large EPC of NaH₁₀ prompted us to investigate the reasons affecting the intensity of electrons scattered by phonons. Next, our research mainly focuses on the analysis of the electronic properties and phonon characteristics, traced back to the Hopfield expression, $\lambda = \frac{N(\varepsilon_F)\langle I^2 \rangle}{M\omega^2}$, to further unveil the features of EPC interaction and superconductivity [54].

Regarding the $N(\varepsilon_F)$ term arising from the electronic properties in Fig. 2(c), significantly, we revealed a unique mechanism of background charge participating in the EPC observed in the projected DOS, that is, valence electrons being ionized and delocalized caused by high pressure, forming a distributed environmental charge density to mediate metallic interactions. This result is completely different from the formation mechanism of interstitial quasiatoms in compressed lithium and sodium, which show insulating properties [55,56]. In fact, the increased kinetic energy of the electrons induced by pressure here exceeds the limit of the Coulomb potential field, which makes the partial charges redistribute in the lattice as background charges, and this analogous scenario also occurs in the atomic metallic H-IV phase ($I4_1/amd$) with a predicted T_c of 356 K at 500 GPa [52]. The projected DOS illustrates that high pressures induce the filling of H- p orbitals and form sp -hybridized states, which cooperate with the background charges to mainly constitute the FS. The hydrogen atoms in NaH₁₀ with p -electron occupied orbitals exhibit the characteristics of $2p$ -block elements, implying that multiple band pairings of s - and p -electronic states cross ε_F relative to s -block element H (see Fig. S6 [34]) [57]. In Fig. S7, a large EPC also benefited from the high DOS at ε_F caused by a Van Hove singularity, which is mainly constituted with H- sp hybridized states and background charges where they make the 58% and 42% contribution to the total DOS, respectively [57,58]. These sp -hybridized states are expected to have higher freedoms of orbital multiplicity and more easily form stable Cooper pairs than H- s electrons [59,60]. Figure S10 [34] further confirmed that the background charges mainly participate in EPC due to the total $N(\varepsilon_F)$ of background charges that increased with the pressurization.

Regarding the vibrational modes of phonons reflected by the $\frac{\langle I^2 \rangle}{M\omega^2}$ term as exhibited in Fig. 2(a), a striking feature of the phonon band is that various optical phonon

vibrational modes have differential degrees to participate in EPC corresponding to the same \mathbf{q} -vector. The underlying causations behind this physical connotation can be revealed by the linewidth γ_{qv} representation. The γ_{qv} is primarily subjected to two factors: the nesting function $\xi(\mathbf{q})$ and the electron-phonon matrix element $g_{mn}(\mathbf{k}, \mathbf{q})$ based on the formula $\gamma_{qv} = \pi\omega_{qv} \sum_{mn} \sum_{\mathbf{k}} |g_{mn}^v(\mathbf{k}, \mathbf{q})|^2 \delta(\varepsilon_{m,\mathbf{k}+\mathbf{q}} - \varepsilon_f) \times \delta(\varepsilon_{n,\mathbf{k}} - \varepsilon_f)$ [61]. Along the Γ to Z \mathbf{q} -route above 100 THz and at the F point of the phonon band above 80 THz (see Fig. S2 [34]), we can evidently find the softening dips. The corresponding phonon dips are attributed to the main causation induced by FS nesting, which can be verified by the nesting function $\xi(\mathbf{q}) = \frac{1}{N} \sum_{\mathbf{k}, i, j} \delta(\varepsilon_{\mathbf{k}, i} - \varepsilon_F) \delta(\varepsilon_{\mathbf{k}+\mathbf{q}, j} - \varepsilon_F)$ in Fig. 2(b). Two relatively high peaks correspond to a strong nesting strength along the Γ to Z \mathbf{q} -route and at the F point. Next, we calculate the FS to check for nesting behavior. Of particular note is that two tripodlike FSs are buckled upside down to form nesting, evidenced in Fig. 2(d). This result explains the softening dips on phonon modes caused by FS nesting. Moreover, the nature of the FS with differential Fermi velocity also affects the superconductivity. According to the relationship $\xi(\mathbf{q}) \propto \oint \frac{d\mathbf{k}}{|\mathbf{V}_{\mathbf{k}} \times \mathbf{V}_{\mathbf{k}+\mathbf{q}}|}$, the Fermi velocities of $\mathbf{V}_{\mathbf{k}}$ and $\mathbf{V}_{\mathbf{k}+\mathbf{q}}$ at ε_F are almost reverse collinear, indicating the $\xi(\mathbf{q})$ is stronger at the nesting points [46]. Nesting at the mentioned points certainly affects the phonon frequencies, while the nonsoftening phonon frequencies corresponding to the same \mathbf{q} -point contributing to EPC are interrelated with the $g_{mn}(\mathbf{k}, \mathbf{q})$, as reflected by the thickness of the linewidth in the phonon band [62]. Thus, the phonon softening caused by FS nesting and $g_{mn}(\mathbf{k}, \mathbf{q})$ are crucial factors causing a large T_c .

The properties of phonon band structures and energy band structures are closely related to the interactions between atoms, which, in turn, can be revealed by analyzing chemical bonds. We thus investigated the bonding characteristics of NaH₁₀ in comparison with clathratelike hydrides LaH₁₀ and YH₁₀, and insulating molecular phases of solid hydrogen at 300 GPa [63]. The bonding strengths were examined in depth through calculating the integral crystal orbital Hamilton population (iCOHP) and crystal orbital Hamilton population (COHP) of H–H pairs [64]. As Fig. 3(a) illustrates, the bonding strengths between H₂ units of NaH₁₀ in the solid red line region are less than the weak covalent bond strengths in extended clathrate hydrides LaH₁₀ and YH₁₀, and intramolecular bonding strengths in the solid light-blue line region are approximate or slightly less than that in pure H-III phases. These results reconfirm that the hydrogen framework is almost composed of molecular H₂ units. To explain the above-mentioned results, the COHP calculation in Fig. 3(b) indicates that the bonding molecular orbital (BMO, σ) of the H₂ molecule (0.80 Å) is completely occupied, and the antibonding molecular orbital (AMO, σ^*) is partially occupied due to the transferred electrons from sodium. Correspondingly, a small amount of electrons are distributed in the BMOs between H₂ units, which ensures the charge connectivity between them resulting in metallicity.

To further understand the superconducting behavior of the H₂ molecular type NaH₁₀ with pressure, we calculated the T_c 's, ω_{\log} , λ , and $N\varepsilon_F$ at different pressures in Fig. 4(a). The T_c 's increase as pressure increases from 168 K at

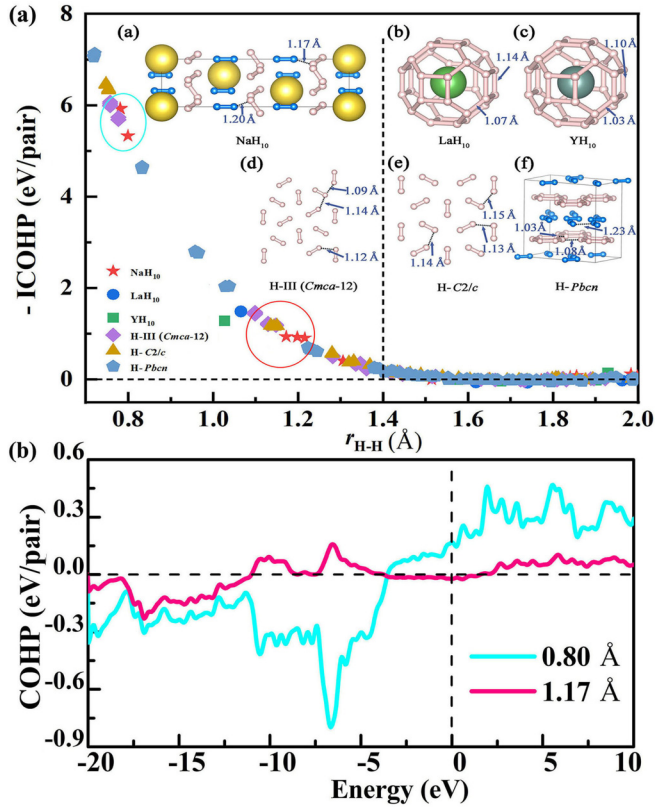


FIG. 3. (a) Calculated bond length $r_{\text{H-H}}$ and the corresponding $-i\text{COHP}$ values between the nearest-neighbor H atoms in different systems at 300 GPa. (b) Plot of COHP for H-H pairs separated by 0.80 Å (intramolecular distance) and 1.17 Å (intermolecular distance).

200 GPa to 270 K at 400 GPa. As the $R32$ phase is compressed to 400 GPa, the moderate increase of T_c is mainly brought about by the enhanced EPC, ω_{log} , and $N\mathcal{E}_F$. Especially in medium-frequency vibration modes (25–80 THz), the peaks of the $\alpha^2F(\omega)$ have increased (see Fig. S2 [34]). Note that the influence of medium-frequency vibrational modes on the integrated EPC has further expanded by nearly 8%, reflecting the electron-phonon matrix element driven by interatomic interactions contributing to λ continuously in a crucial manner. At 400 GPa, the abnormal performance of the optical phonon softening induced by FS nesting is increased at the F point corresponding to >80 THz, further enhancing EPC (see Fig. 2(b) and Fig. S2 [34]). Besides, $N\mathcal{E}_F$ increased with pressure enhancing the possibility of electrons being scattered by phonons, which increases the strength of EPC thus improving T_c . However, the effect of low-frequency modes on EPC become less significant with pressure, which suggests that T_c is very sensitive to the medium-frequency vibrational motions in this H_2 molecular type structure.

To propose more H_2 molecule type hydrides and investigate the correlation between molecular H_2 unit and high T_c , we further doped sodium into the H-III phase for computational simulation at 300 GPa. Subsequently, we obtained a dynamically stable $R\text{-}3m\text{-NaH}_{12}$ structure in Fig. 4(b), which was in agreement with the previously predicted result in NaH_{12} stoichiometry [30]. The molecular H_2 units

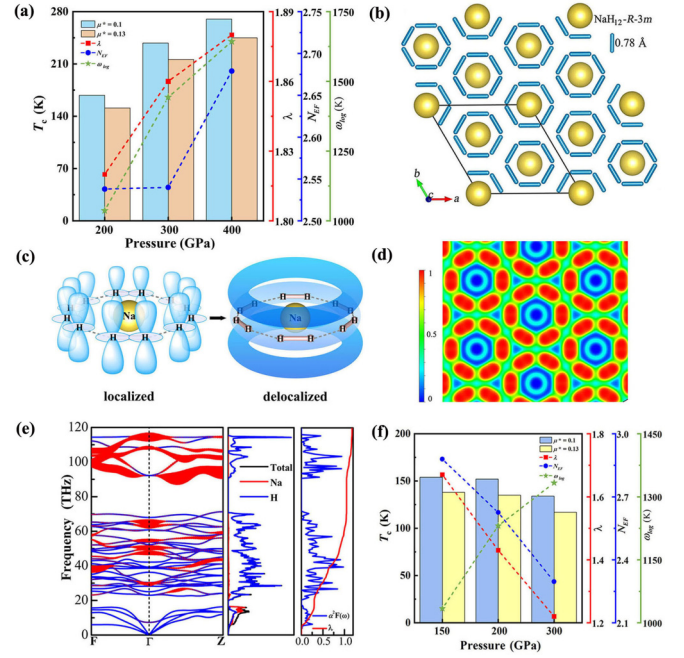


FIG. 4. (a) Calculated superconductivities for $R32\text{-NaH}_{10}$ as a function of pressures. (b) $R\text{-}3m\text{-NaH}_{12}$ phase at 300 GPa, (c) Energy distribution spectra of localized and delocalized sp -hybridized electrons in the Lewis structure for planar benzenelike H_{12} sublattice. (d) ELF of $R\text{-}3m\text{-NaH}_{12}$. (e) Phonon dispersion curves, $\alpha^2F(\omega)$ and integral $\lambda(\omega)$, and (f) superconductivities for $R\text{-}3m\text{-NaH}_{12}$ as a function of pressures.

with equal bond lengths of 0.78 Å are shaped into a planar benzenelike structure composed of 12 hydrogen atoms, but intermolecular interactions are also connected by noncovalent bonds and stacked along the c axis [see Fig. 4(d)]. In sharp contrast to the insulating molecular phase H-III of solid hydrogen, $R\text{-}3m\text{-NaH}_{12}$ is metallic as determined by the background charges and electrons occupying the sp -hybridized antibonding bands, which form delocalized characteristics in a planar benzenelike H_{12} sublattice. This result is supported by comparing the kinetic energy relationship in the cases of “localized or delocalized electronic states” as discussed in Fig. 4(c) and Scheme S1 [34]. Next, we conducted superconductivity and illustrated the T_c of this phase with the evolution of high pressure. The molecular-type hydride NaH_{12} also has a notable T_c of 154 K at a relatively low pressure of 150 GPa. Instructively, our observation of the emergence of H_2 molecular type hydrides with notable T_c highlights the correlation between H_2 units and novel superconductors.

In summary, we propose a high-throughput structural search approach to screen H_2 molecular type hydrides to quickly identify potential high T_c . We confirm a unique type of pressure-induced near room-temperature superconductors in solely H_2 molecule constituted NaH_{10} with T_c of 270 K at 400 GPa and NaH_{12} with high T_c of 154 K at 150 GPa, which is a prototype in the family of H_2 molecular type hydrides. Being subjected to extensive first-principles calculations, the electronic structures, bonding properties, and superconductivities of NaH_{10} and NaH_{12} are revealed. The potential near room-

temperature superconductivity originates from strong EPC to scatter itinerant electrons. The unique delocalized background charges and electrons occupying the pressure-induced *sp*-hybridized antibonding bands of molecular H₂ units act as itinerant electrons to participate in EPC. Our work reshapes the cognition of hydrogen molecular dominated superconductivity and will stimulate future high-pressure experimental work on synthesis of these H₂ molecular type structures to obtain room-temperature superconductors.

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The authors declare no conflict of interest.

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